

Supporting information

$A_2P_2S_6$ (A=Ba, Pb): A Good Platform to Study Polymorph Effect and Lone Pairs Effect to Form Acentric Structure

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1. Table S1. Refined crystallographic parameters of α -Ba₂P₂S₆, β -Ba₂P₂S₆, and Pb₂P₂S₆.

Compound	α -Ba ₂ P ₂ S ₆	β -Ba ₂ P ₂ S ₆	Pb ₂ P ₂ S ₆
Formula weight (g mol ⁻¹)	528.98	264.49	668.68
Crystal color	Pink/light orange	white	yellow orange
Temperature		298(2) K	
Radiation, wavelength		Mo-K α , 0.71073 Å	
Crystal system		Monoclinic	
Space group	<i>Pn</i> (no.7)	<i>P2₁/n</i> (no. 14)	<i>Pn</i> (no.7)
a(Å)	6.7411(2)	6.7459(6)	6.5961(6)
b(Å)	7.5818(2)	7.5871(6)	7.4542(7)
c(Å)	9.9870(3)	9.9983(9)	9.3854(10)
β (°)	91.2005(12)	91.204(3)	91.515(6)
V (Å ³)	510.32(3)	511.62(8)	461.31(8)
Z	2	4	2
D _c (g cm ⁻³)	3.443	3.434	4.814
μ (mm ⁻¹)	9.136	9.113	38.071
F(000)	476	476	580
R1, wR2 (I > 2 σ (I))	0.0162, 0.0405	0.0276, 0.0881	0.0402, 0.0588
R1, wR2 (all data)	0.0212, 0.0906	0.0384, 0.0939	0.0772, 0.0764

$$R_1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}; wR_2 = \frac{[\sum [w(F_o^2 - F_c^2)^2]]^{1/2}}{[\sum [w(F_o^2)^2]]^{1/2}}, \text{ and } w = 1/[\sigma^2 F_o^2 + (A \cdot P)^2 + B \cdot P], P = (F_o^2 + 2F_c^2)/3; A \text{ and } B \text{ are weight coefficients.}$$

2. Table S2. Atomic coordinate parameters for α -Ba₂P₂S₆, β -Ba₂P₂S₆, and Pb₂P₂S₆.

Atoms	Wyckoff	x	y	z	Occupancy	U_{eq} (Å ²)
α-Ba₂P₂S₆						
Ba1	2a	0.26582(18)	0.3419(4)	0.97379(10)	1	0.0137(6)
Ba2	2a	0.68861(19)	-0.1578(4)	0.97524(13)	1	0.0144(8)
P1	2a	0.184(3)	0.863(2)	0.7846(16)	1	0.009(3)
P2	2a	-0.225(2)	0.366(2)	1.1624(16)	1	0.012(3)
S1	2a	-0.211(2)	0.2753(19)	0.9756(17)	1	0.014(3)
S2	2a	0.164(2)	0.777(2)	0.9776(17)	1	0.018(3)
S3	2a	0.053(2)	0.457(2)	0.6824(13)	1	0.015(3)
S4	2a	0.032(2)	0.452(2)	1.2462(14)	1	0.013(3)
S5	2a	0.403(2)	0.044(2)	0.7580(14)	1	0.016(3)

S6	2a	0.424(2)	0.0481(19)	1.2039(15)	1	0.014(3)
β-Ba₂P₂S₆						
Ba1	4e	0.03858(5)	0.59197(5)	0.74938(3)	1	0.01486(15)
P1	4e	-0.0453(2)	0.11437(18)	0.56081(15)	1	0.0112(3)
S1	4e	0.1966(2)	0.2980(2)	0.97893(14)	1	0.0144(3)
S2	4e	0.1748(2)	0.29313(19)	0.53803(15)	1	0.0150(3)
S4	4e	-0.0620(2)	1.0265(2)	0.75068(14)	1	0.0169(3)
Pb₂P₂S₆						
Pb1	2a	0.09293(10)	0.1326(4)	0.54137(9)	1	0.0221(7)
Pb2	2a	0.67133(10)	-0.3670(4)	0.54250(9)	1	0.0228(7)
P1	2a	0.1932(17)	-0.3610(17)	0.7258(12)	1	0.009(2)
P2	2a	0.0703(16)	-0.1416(18)	0.8575(14)	1	0.013(2)
S1	2a	-0.1875(17)	-0.0556(19)	0.7555(15)	1	0.017(3)
S2	2a	-0.469(2)	0.2435(18)	0.5533(15)	1	0.016(3)
S3	2a	0.232(2)	-0.2546(18)	0.5299(14)	1	0.016(3)
S4	2a	0.9513(16)	-0.5551(17)	0.3261(12)	1	0.011(2)
S5	2a	-0.0255(18)	0.4463(17)	0.7370(14)	1	0.015(2)
S6	2a	0.2851(17)	0.0521(17)	0.8488(15)	1	0.014(2)

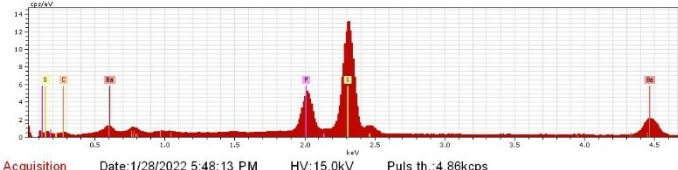
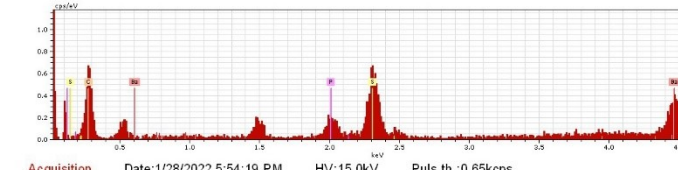
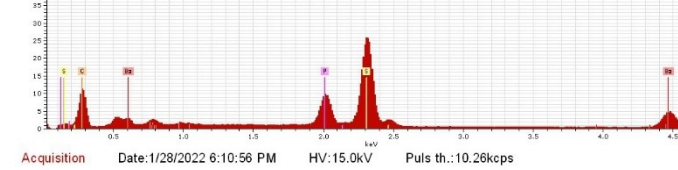
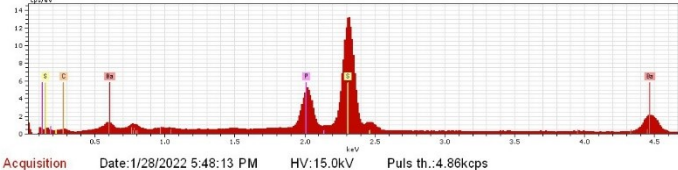
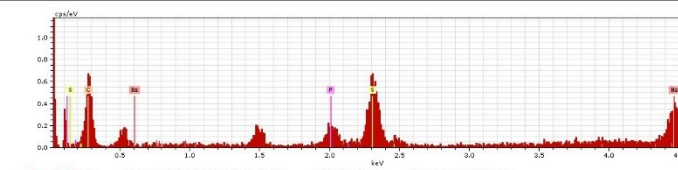
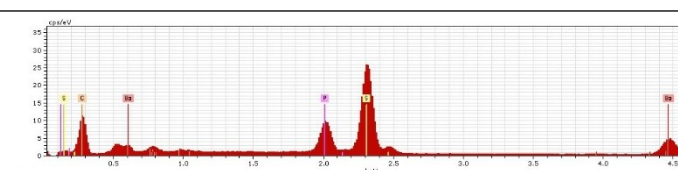
3. Table S3. Selected interatomic distances for α -Ba₂P₂S₆, β -Ba₂P₂S₆, and Pb₂P₂S₆.

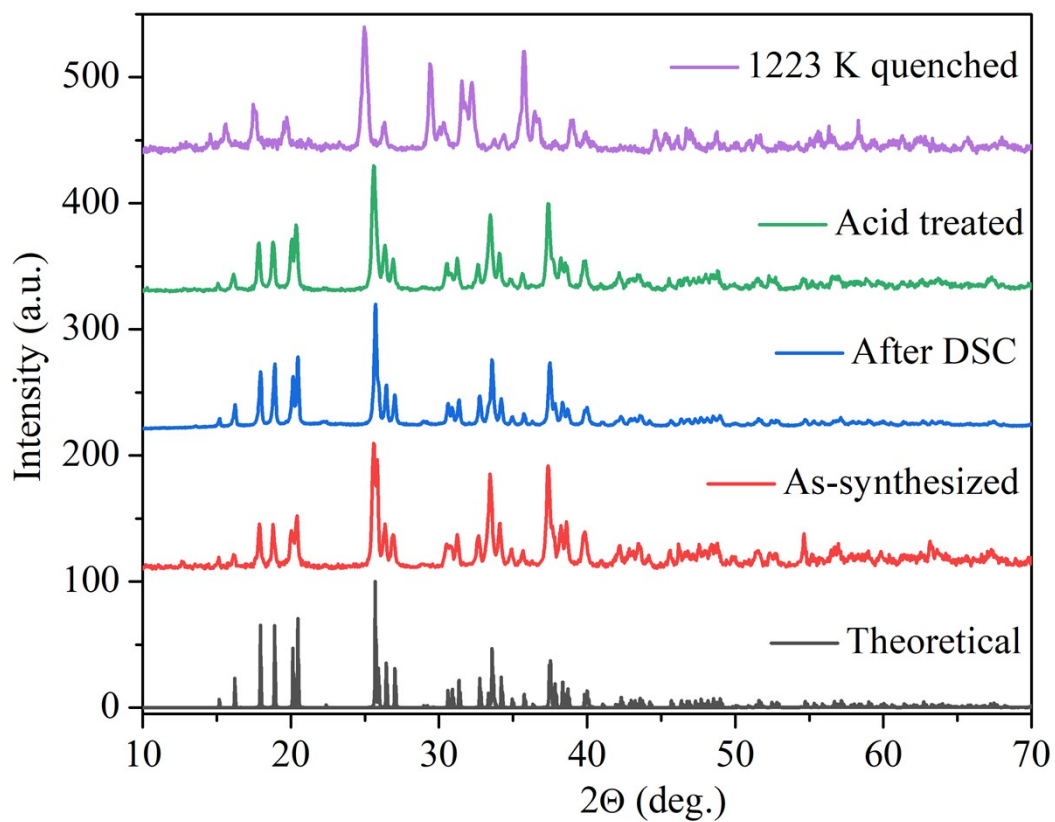
Atom Pairs		Distances (Å)	Atom Pairs		Distances (Å)
α-Ba₂P₂S₆					
Ba1	S1	3.254(17)	Ba2	S1	3.353(15)
	S1	3.562(14)		S2	3.568(17)
	S2	3.373(16)		S2	3.244(17)
	S3	3.200(16)		S3	3.218(16)
	S3	3.334(15)		S4	3.360(14)
	S4	3.283(14)		S5	3.253(17)
	S4	3.313(15)		S5	3.261(16)
	S5	3.267(16)		S6	3.277(15)
P1	S6	3.359(15)	S6	3.316(15)	
	S2	2.04(2)	P2	S1	1.99(2)
	S5	2.04(2)		S3	2.02(2)
S6	2.03(2)	S4		2.02(3)	
P1	P2	2.216(5)			
β-Ba₂P₂S₆					
Ba1	S1	3.2825(15)	P1	S1	2.021(2)

	S1	3.3190(15)		S2	2.028(2)
	S1	3.3581(16)		S4	2.017(2)
	S2	3.2268(15)	P1	P1	2.213(3)
	S2	3.2453(16)			
	S2	3.3040(15)			
	S4	3.2530(16)			
	S4	3.3662(18)			
	S4	3.5658(16)			
Pb₂P₂S₆					
Pb1	S1	3.102(14)	Pb2	S1	3.192(12)
	S1	3.142(13)		S2	3.050(13)
	S2	3.004(13)		S3	3.014(13)
	S3	3.032(14)		S4	3.114(12)
	S4	3.214(13)		S4	3.120(12)
	S5	3.086(14)		S5	3.013(12)
	S6	3.015(12)		S5	3.165(12)
	S6	3.184(13)		S6	3.068(13)
P1	S3	2.025(18)	P2	S1	2.033(16)
	S4	2.023(15)		S2	2.011(19)
	S5	2.040(18)		S6	2.026(17)
P1	P2	2.216(6)			

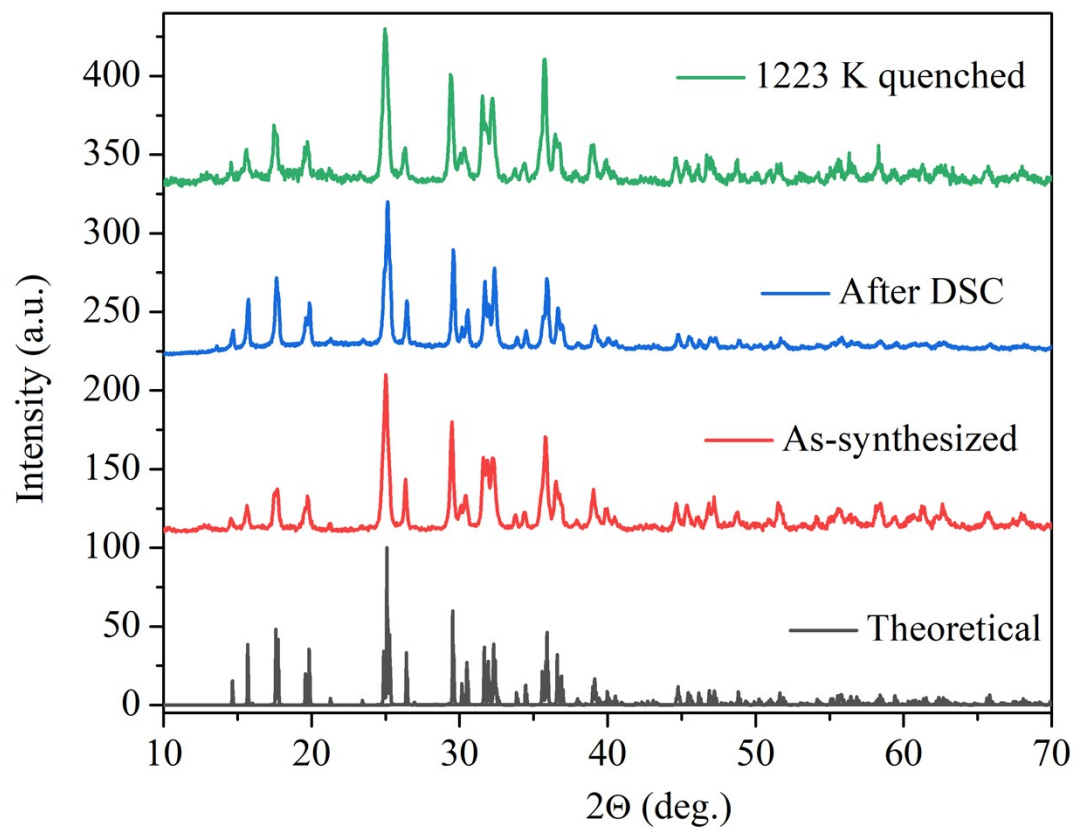
4. Table S4. Results of EDX analysis of selected crystals of α -Ba₂P₂S₆, β -Ba₂P₂S₆, and Pb₂P₂S₆.

Spectrum	Measured results	Normalized Composition
α-Ba₂P₂S₆		
	(Atomic.%) C 47.40 Ba 10.90 P 10.12 S 31.58	Ba _{2.15} P ₂ S _{6.24}
	(Atomic.%) C 59.91 Ba 8.13 P 7.68 S 24.27	Ba _{2.11} P ₂ S _{6.32}
	(Atomic.%) C 42.16 Ba 12.38 P 10.89 S 34.56	Ba _{2.27} P ₂ S _{6.34}

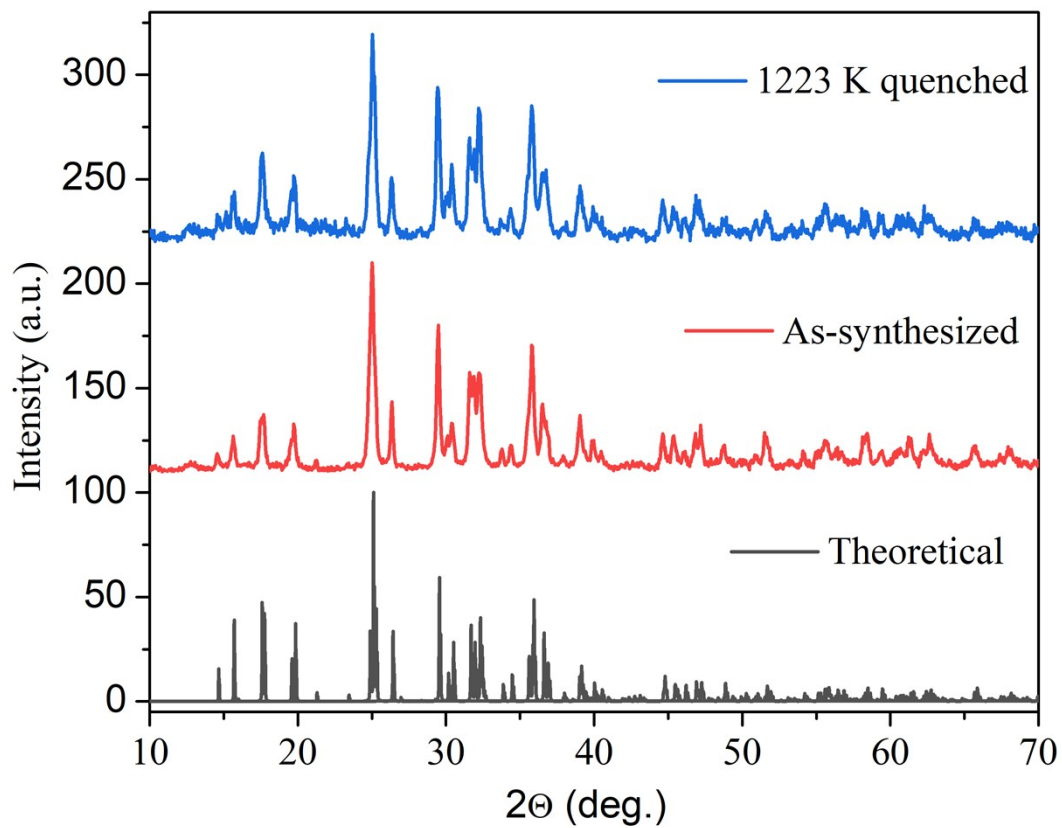
Averaged		Ba_{2.18}P₂S_{6.30}
β-Ba₂P₂S₆		
	(Atomic.%) C 45.23 Ba 9.18 P 11.70 S 33.90	Ba _{1.57} P ₂ S _{5.79}
	(Atomic.%) C 76.40 Ba 4.27 P 4.17 S 15.16	Ba _{2.05} P ₂ S _{7.27}
	(Atomic.%) C 82.49 Ba 3.21 P 3.56 S 10.75	Ba _{1.80} P ₂ S _{6.04}
Averaged		Ba_{1.81}P₂S_{6.37}
Pb₂P₂S₆		
	(Atomic.%) C 33.47 Pb 17.03 P 14.31 S 35.19 I 0.0	Pb _{2.38} P ₂ S _{4.91}
	(Atomic.%) C 35.82 Pb 16.52 P 13.91 S 33.62 I 0.13	Pb _{2.37} P ₂ S _{4.83}
	(Atomic.%) C 32.86 Pb 17.32 P 14.21 S 35.38 I 0.23	Pb _{2.43} P ₂ S _{5.00}
Averaged		Pb_{2.39}P₂S_{4.91}
Notes: the large deviation of S is due to the overlap between Pb and S		



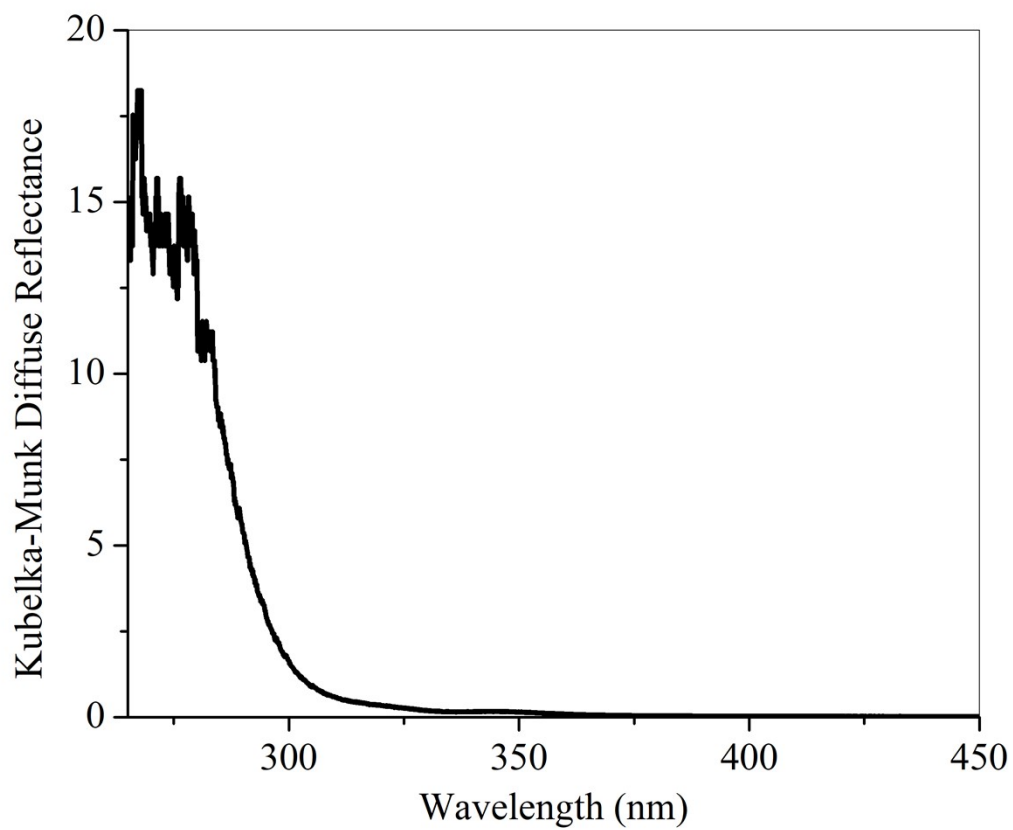
5. Figure S1. Powder X-ray diffraction results of theoretical, as-synthesized (stored in air for around 240 days), after DSC measurement, acid treated, and 1223 K quenched $\text{Pb}_2\text{P}_2\text{S}_6$.



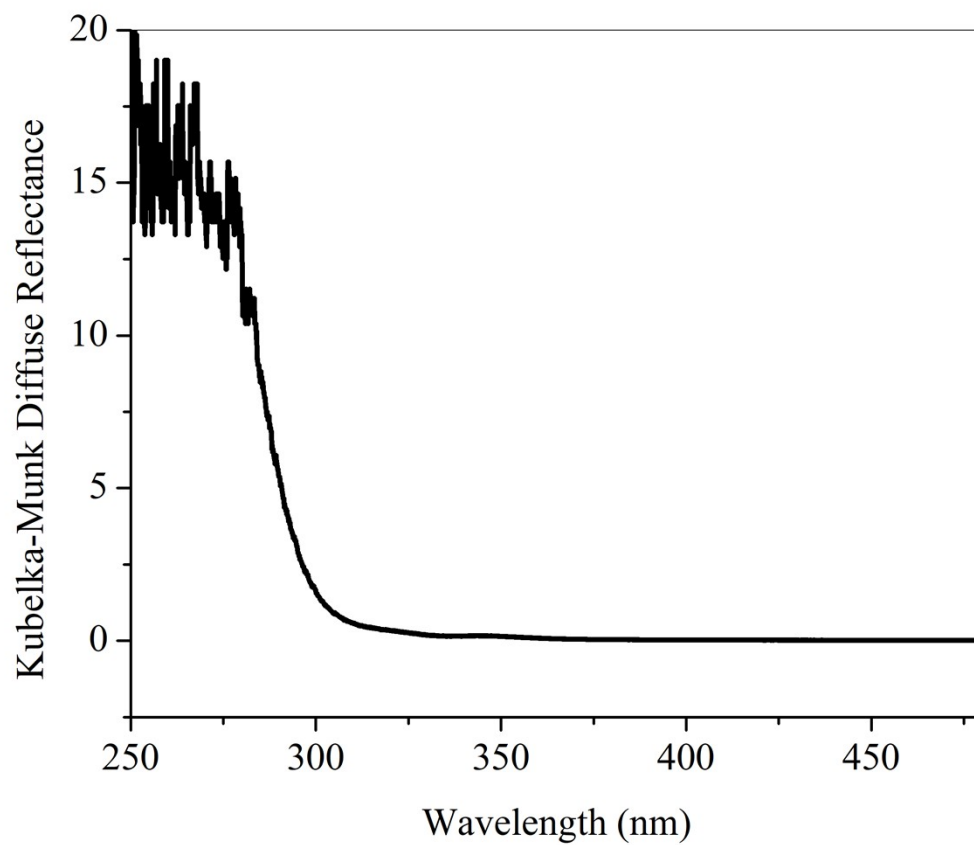
6. Figure S2. Powder X-ray diffraction results of theoretical, as-synthesized, after DSC measurement, and 1223 K quenched α -Ba₂P₂S₆.



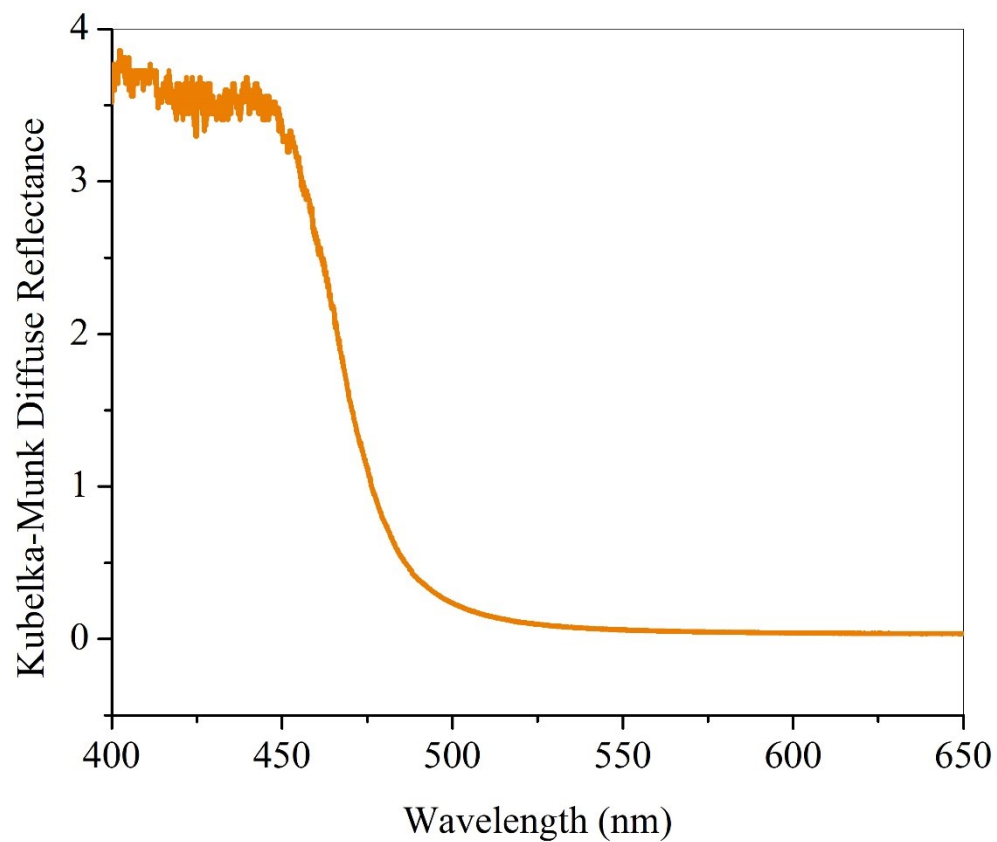
7. **Figure S3.** Powder X-ray diffraction results of theoretical, as-synthesized, and 1223 K quenched β -Ba₂P₂S₆.



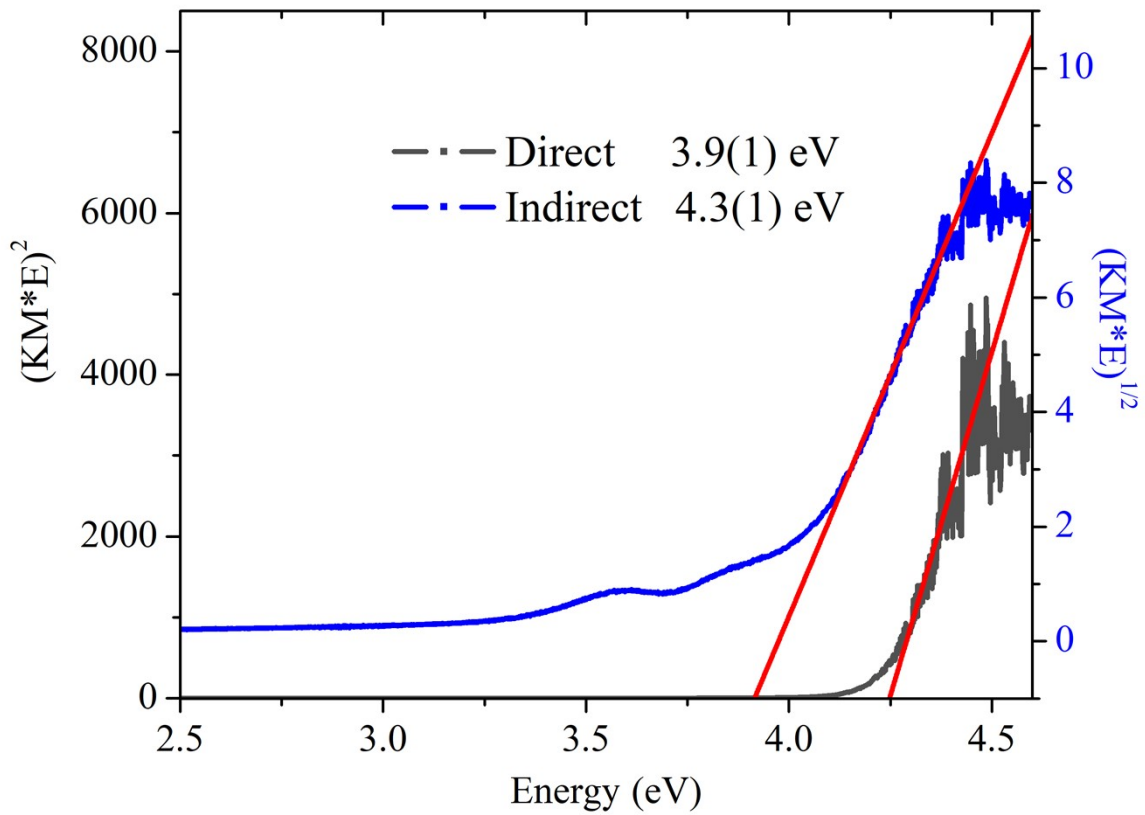
8. **Figure S4.** Kubelka-Munk Diffuse reflectance solid-state UV-Vis spectra of α -Ba₂P₂S₆.



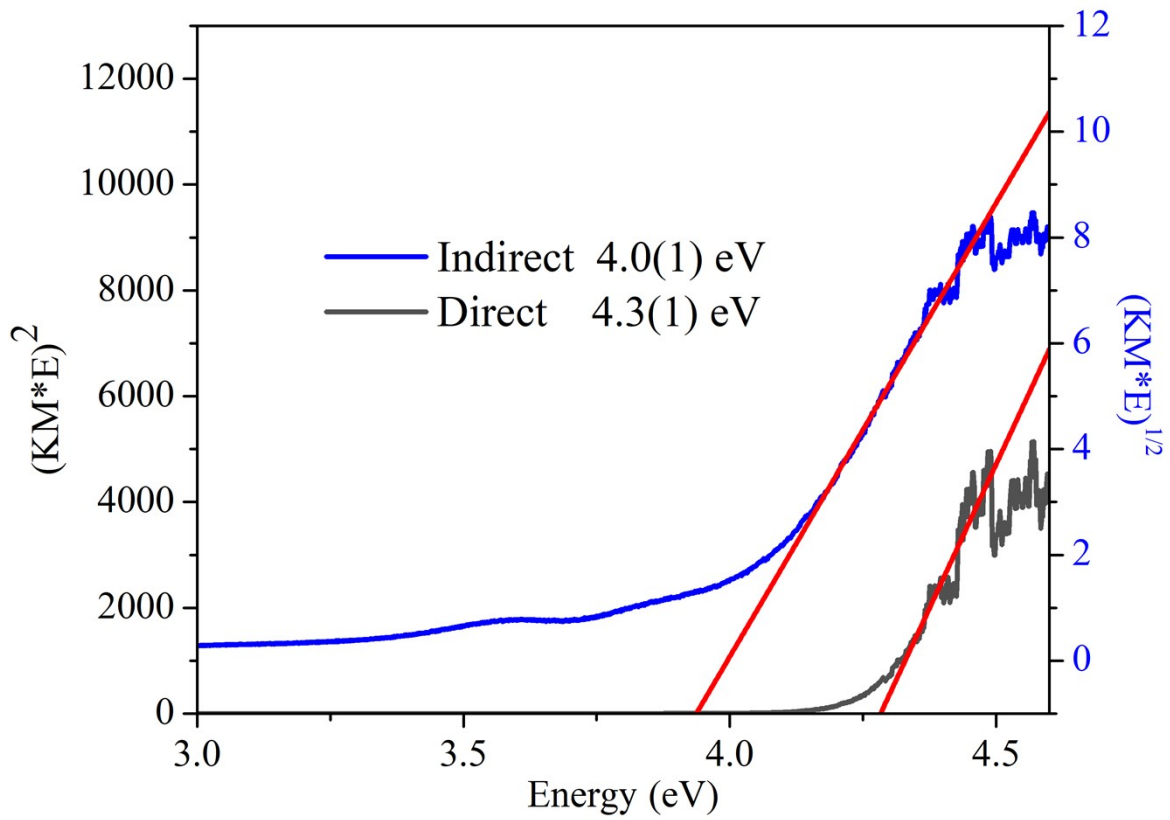
9. **Figure S5.** Kubelka-Munk Diffuse reflectance solid-state UV-Vis spectra of β -Ba₂P₂S₆.



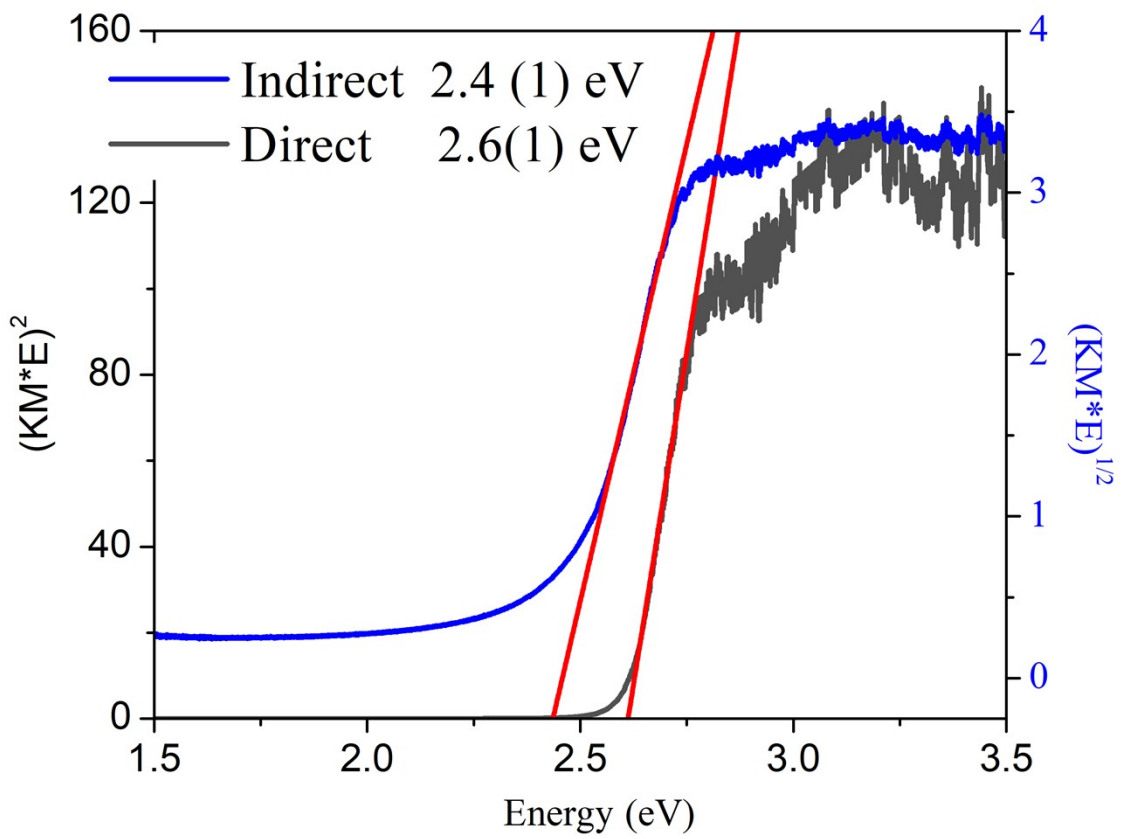
10. Figure S6. Kubelka-Munk Diffuse reflectance solid-state UV-Vis spectra of Pb₂P₂S₆.



11. Figure S7. Tauc plots for allowed direct and indirect transitions of α -Ba₂P₂S₆.



12. **Figure S8.** Tauc plots for allowed direct and indirect transitions of $\beta\text{-Ba}_2\text{P}_2\text{S}_6$.



13. **Figure S9.** Tauc plots for allowed direct and indirect transitions of $\text{Pb}_2\text{P}_2\text{S}_6$.