

Supporting Information

Dehydrogenation of Formic Acid Using Iridium-NSi Species as Catalyst Precursors

Jefferson Guzmán,^a Asier Urriolabeitia,^b Víctor Polo,^b Marta Fernández-Buenestado,^a Manuel Iglesias^a and Francisco J. Fernández-Alvarez^{a,*}

^a*Departamento de Química Inorgánica – Instituto de Síntesis Química y Catálisis Homogénea (ISQCH).
Universidad de Zaragoza – CSIC, 50009, Zaragoza – Spain.
E-mail: paco@unizar.es*

^b*Departamento de Química Física – Instituto de Biocomputación y Física de Sistemas Complejos (BIFI).
Universidad de Zaragoza 50009, Zaragoza – Spain*

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General Procedure

Catalytic reactions were carried out on a microreactor (Man on the Moon series X102 Kit) with a total volume of 19 mL. Under an argon atmosphere, the reactor was filled with the desired amount of base (sodium formate or NEt_3) and of formic acid. The reactor was then closed and heated to the desired temperature in an oil bath. When the system is stabilized, the desired amount of complex, dissolved in a small amount of formic acid, is then injected with a microsyringe.

SF concentration effect: 500 μL of FA, 0.025 mol % of **1** (2.6 mg, $3.25 \cdot 10^{-3}$ mmol), and 2.5 mol % of SF (22.5 mg, 0.33 mmol), or 5 mol % of SF (45.0 mg, 0.66 mmol), or 10 mol % of SF (90.0 mg, 1.32 mmol) at 80 $^\circ\text{C}$.

Complex 1 concentration effect: 500 μL of FA, 5 mol % of sodium formate (45.0 mg, 0.66 mmol), 0.025 mol % of **1** (2.6 mg, $3.25 \cdot 10^{-3}$ mmol), or 0.05 mol % of **1** (5.2 mg, $6.50 \cdot 10^{-3}$ mmol), or 0.1 mol % (10.4 mg, 0.013 mmol) at 80 $^\circ\text{C}$.

Effect of the silicon substituents in the FA dehydrogenation with SF as base: 500 μL of FA, 5 mol % of sodium formate (45.0 mg, 0.66 mmol), 0.025 mol % of **1** (2.6 mg, $3.25 \cdot 10^{-3}$ mmol) or 0.025 mol % of **2** (2.25 mg, $3.25 \cdot 10^{-3}$ mmol) at 80 $^\circ\text{C}$.

Effect of the silicon substituents in the FA dehydrogenation with NEt_3 as base: 500 μL of FA, 10 mol % of NEt_3 (186 μL , 1.32 mmol), 0.1 mol % of **1** (10.4 mg, 0.013 mmol) or 0.1 mol % of **2** (9 mg, 0.013 mmol) at 80 $^\circ\text{C}$.

NEt_3 concentration effect: 500 μL of FA, 0.1 mol % of **2** (9 mg, 0.013 mmol) and 1.0 mol % (18.6 μL , 0.13 mmol), 2.5 mol % (46.5 μL , 0.32 mmol), 5.0 mol % (93 μL , 0.65 mmol) 10 mol % (186 μL , 1.32 mmol), 20 mol % (372 μL , 2.64 mmol) or 40 mol % (744 μL , 5.28 mmol) of NEt_3 at 80 $^\circ\text{C}$.

Temperature effect: 500 μL of FA, 0.1 mol % of **2** (9 mg, 0.013 mmol) and 10 mol % (186 μL , 1.32 mmol) of NEt_3 at different temperatures (from 40 to 100 $^\circ\text{C}$).

KIE Experiments: 250 μL of FA (HCOOH , HCOOD , DCOOH or DCOOD), 0.1 mol % of **2** (4.5 mg, $6.5 \cdot 10^{-3}$ mmol) and 40 mol % (372 μL , 2.64 mmol) of NEt_3 at 80 $^\circ\text{C}$.

Studies of the black precipitated: Black precipitate was separated from the reaction medium and washed with distilled water and dried in vacuo for its measurements in TEM spectroscopy.

FT-IR and GC of gaseous product: FT-IR: Gaseous products from the catalysis were collected in a rubber balloon that was attached to a Schlenk containing the reaction mixture: 500 μL of FA, 0.1 mol % of **2** (9 mg, 0.013 mmol) and 10 mol % of NEt_3 (186 μL , 1.32 mmol) at 80 $^\circ\text{C}$. The gas sample was transferred to the cell and the FT-IR spectrum was acquired. GC: The gaseous mixture was taken directly from the microreactor and injected into the cell.

Mercury drop test: A drop of mercury was added to the microreactor together with 450 μL of FA, and 10 mol % (186 μL , 1.32 mmol) of NEt_3 , the system was heated at 80 $^\circ\text{C}$. When the system is stabilized, a solution of 0.1 mol % of **2** (9 mg, 0.013 mmol) in 50 μL of FA was injected to the reaction mixture.

NMR Studies

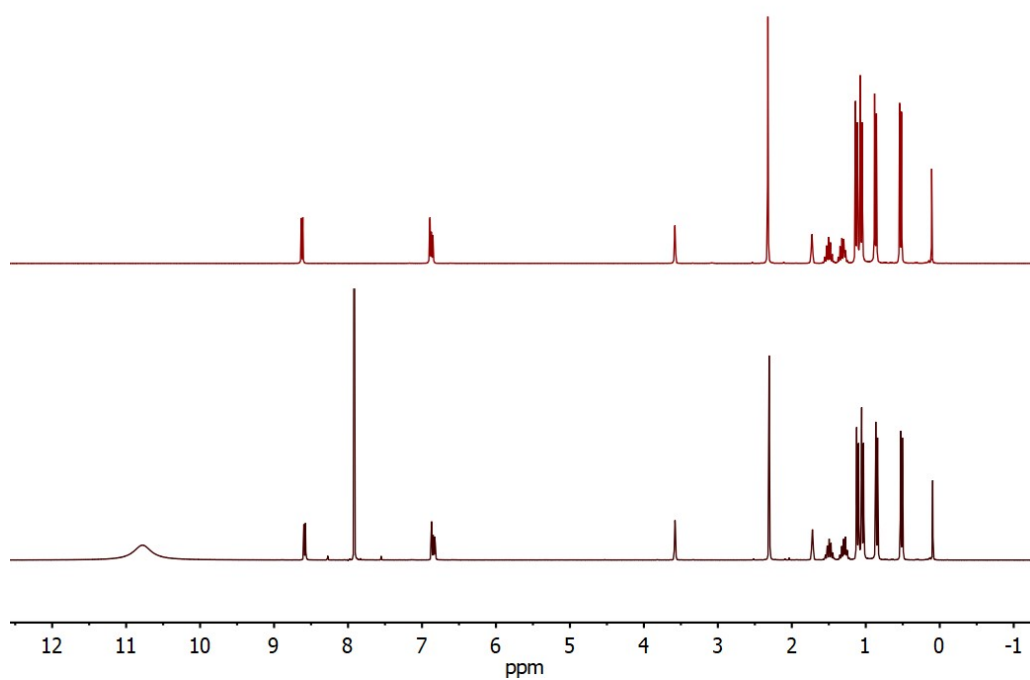


Figure S1. (up) ^1H NMR spectrum of **1** in thf-d_8 ; (down) ^1H NMR spectrum of **1** and HCOOH (20 eq.) in thf-d_8 after heating at 323 K for 3 h.

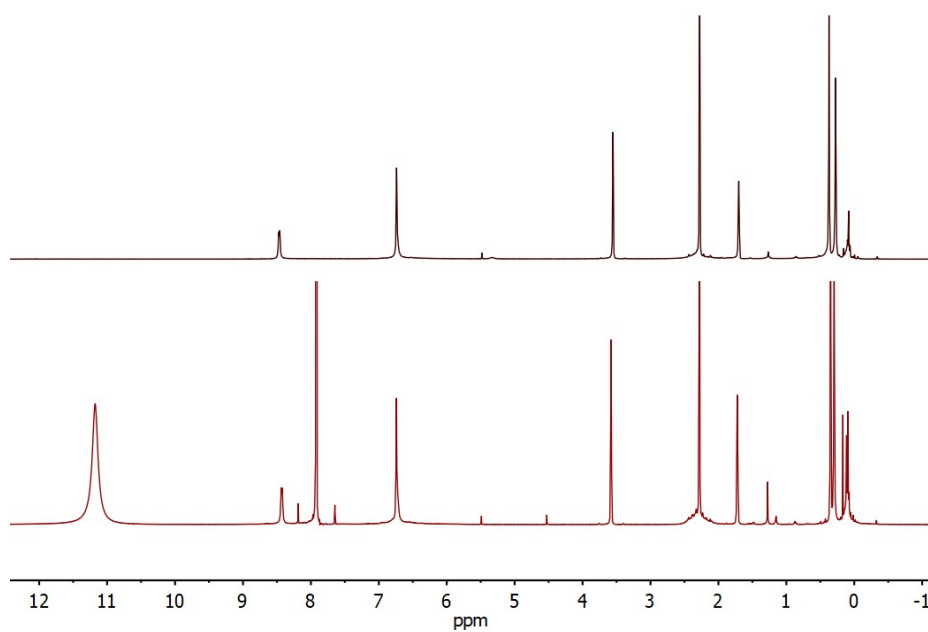


Figure S2. (up) ^1H NMR spectrum of **2** in thf-d_8 ; (down) ^1H NMR spectrum of **2** and HCOOH (20 eq.) in thf-d_8 after heating at 323 K for 3 h.

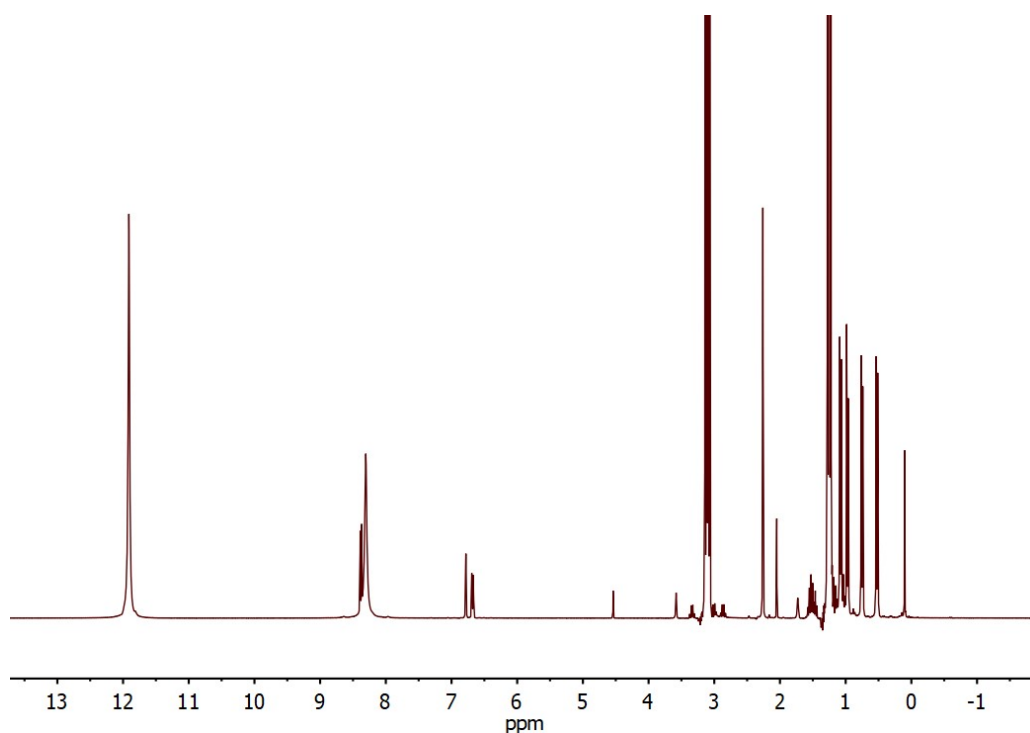


Figure S3. ^1H NMR spectrum of **1**, HCOOH (20 eq.), Et_3N (40 mol% to FA) in thf-d_8 , after heating at 323 K for 30 min.

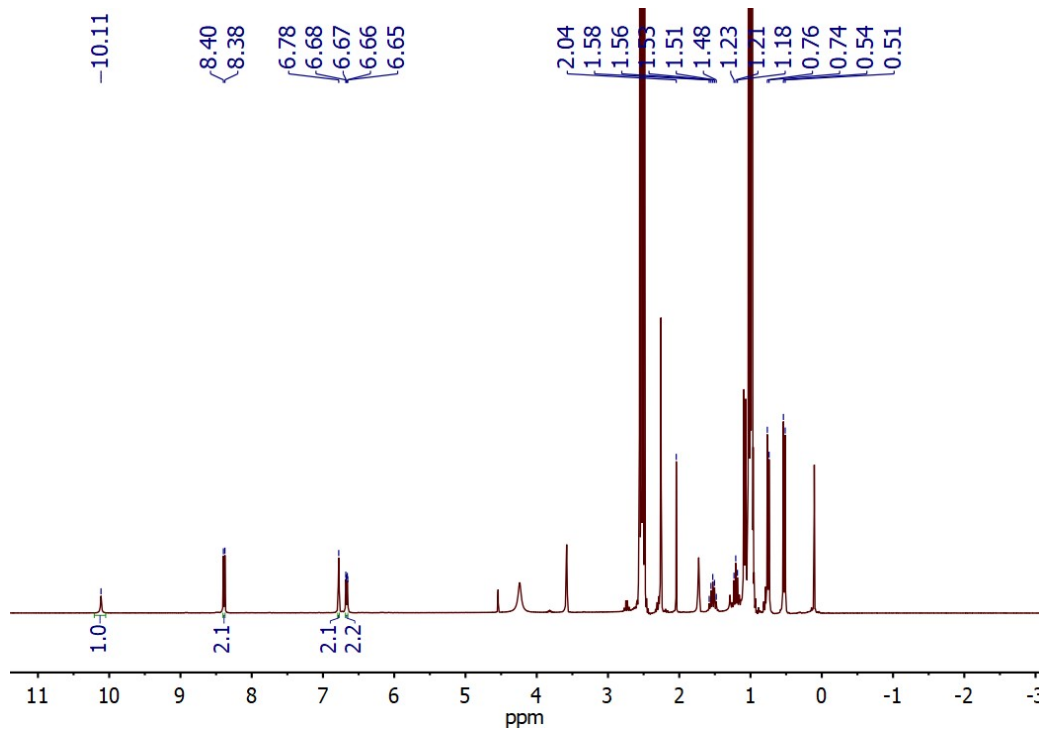


Figure S4. ^1H NMR spectrum of **1**-catalyzed FADH reaction in thf-d_8 once all the FA has been consumed. The only Ir-compound observed is species **3**.

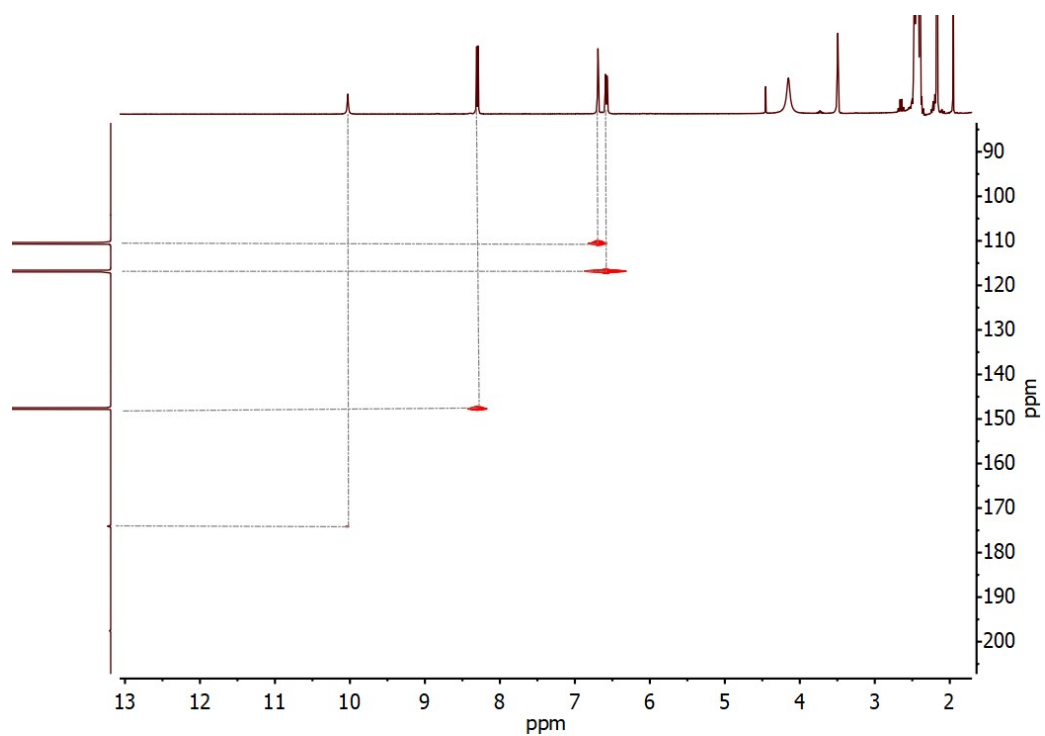


Figure S5. ^1H - ^{13}C HSQC NMR spectrum of **3** solutions in thf-d_8 .

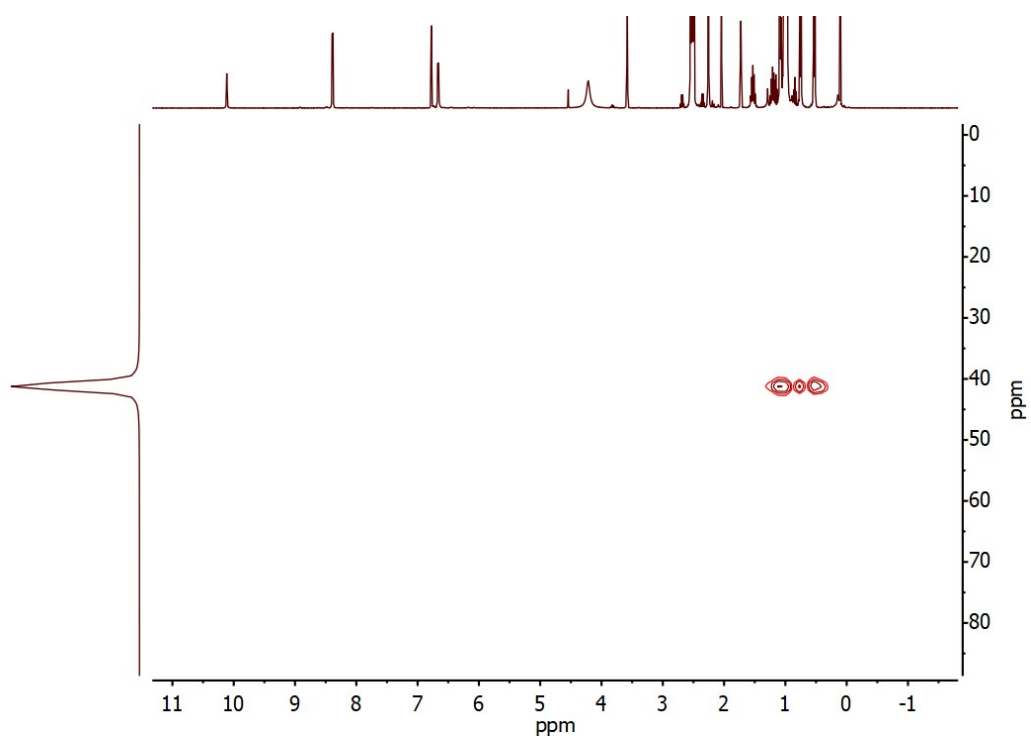


Figure S6. ^1H - ^{29}Si HSQC NMR spectrum of **3** solutions in thf-d_8 .

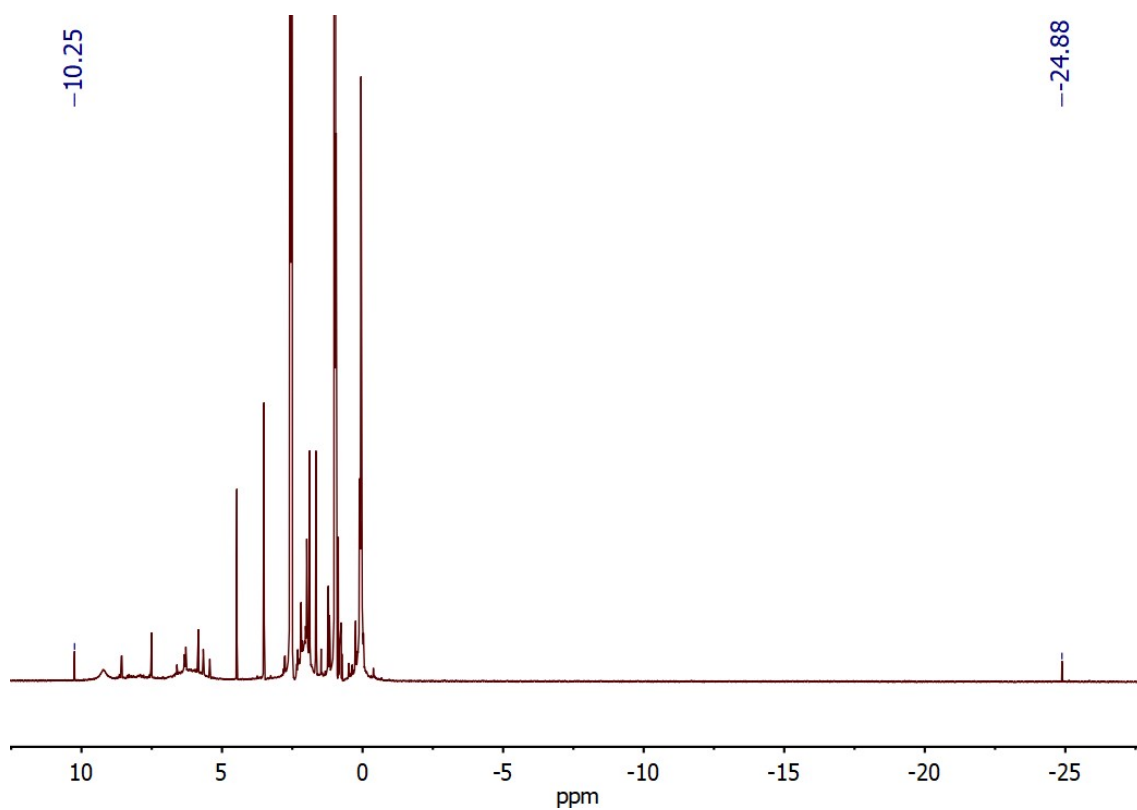


Figure S7. ^1H NMR spectrum of **2**-catalyzed FADH reaction in thf- d_8 once all the FA has been consumed.

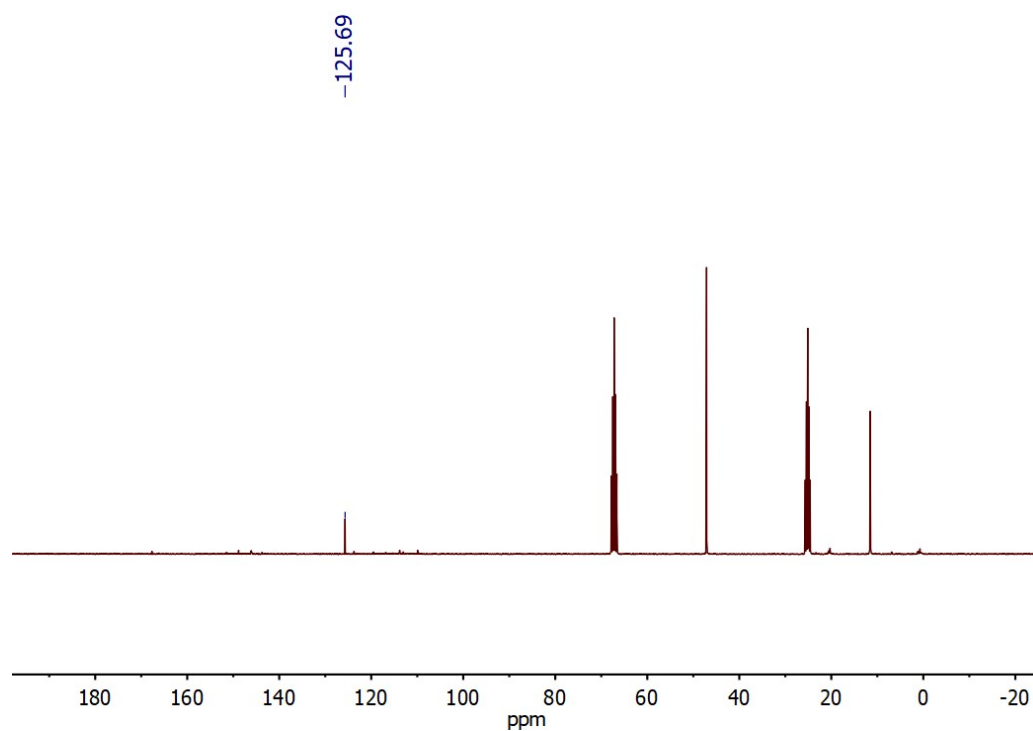


Figure S8. ^{13}C NMR spectrum of **2**-catalyzed FADH reaction in thf- d_8 once all the FA has been consumed.

TON and TOF determination

H₂ pressure:
$$P_{H_2} = \frac{P_{measured}}{2}$$

Amount of H₂ formed calculated with the Ideal Gas Law:
$$n_{H_2} = \frac{P_{H_2}V}{RT};$$

Total Volume = 0.019 L; R constant = 0.08314 bar L mol⁻¹ K⁻¹

$$TON = \frac{n_{H_2}}{n_{cat}}$$

$$TOF = \frac{TON}{t}$$

Effect of the silicon substituents with NEt₃ as base

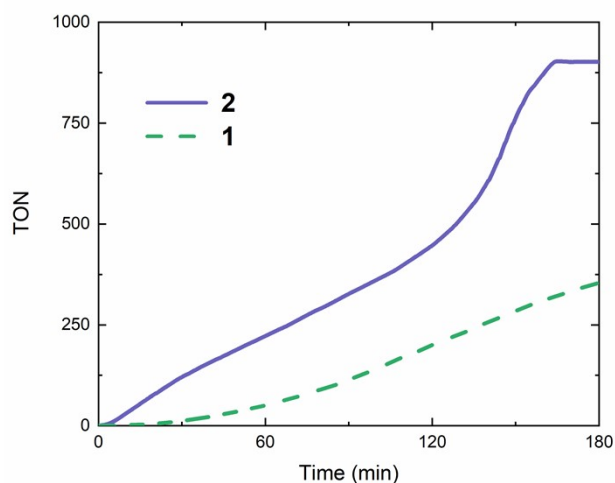


Figure S9. TON vs time comparison of the **1**- or **2**-catalyzed (0.1 mol %) solventless dehydrogenation of formic acid using triethylamine (10 mol %) at 80°C.

NEt₃ concentration effect

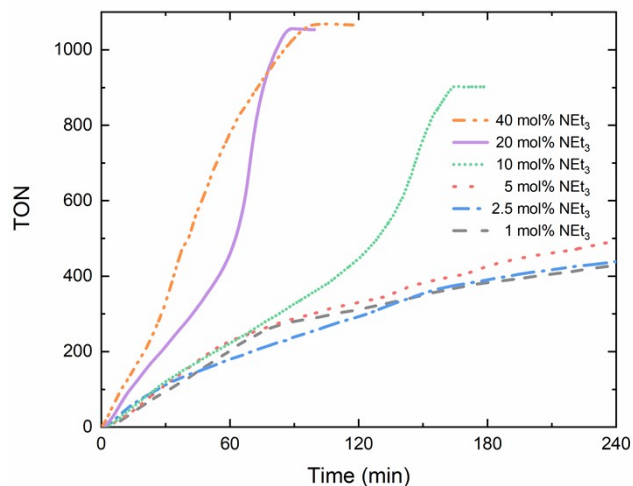


Figure S10. TON vs time representation of the **2**-catalyzed (0.1 mol %) solvent-free FA dehydrogenation with different concentrations of NEt₃ at 80°C

Temperature effect

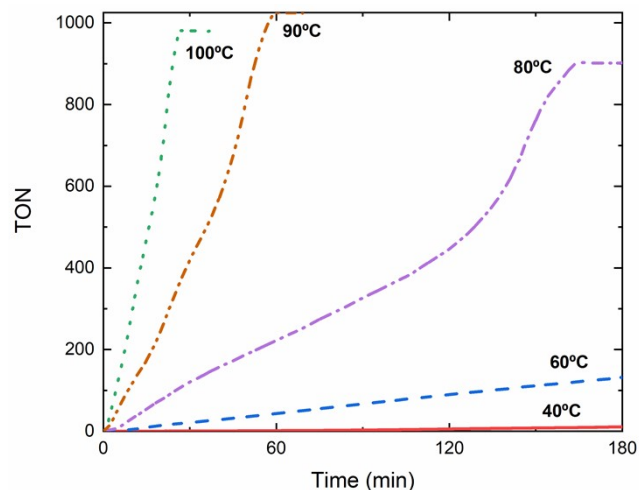


Figure S11. TON vs time representation of the **2**-catalyzed (0.1 mol %) solvent-free FA dehydrogenation with 10 mol % of NEt₃ at different temperatures.

2-Catalysed FA dehydrogenation with optimized conditions

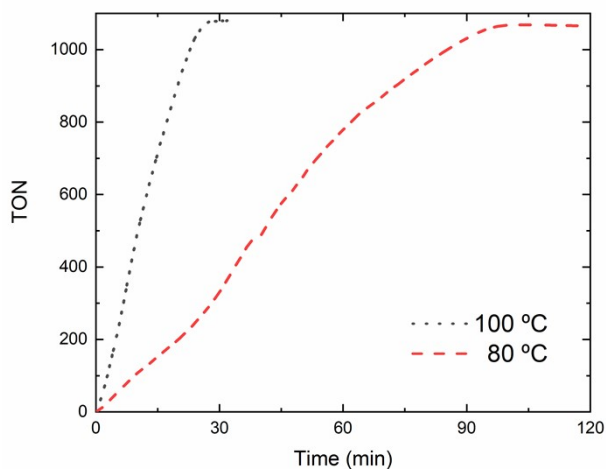


Figure S12. TON vs time representation of the 2-catalysed FA solvent-less dehydrogenation with NEt_3 (40 mol %) at 80 and 100 °C.

FT-IR and GC of gaseous product

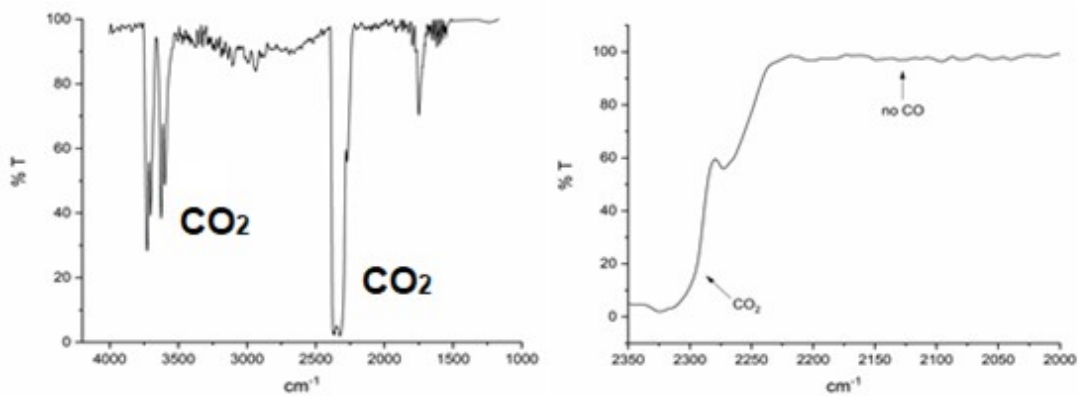


Figure S13. (Left) FT-IR spectrum of the gaseous product. (Right) Detail from the FT-IR spectrum showing the absence of CO signal.

File :C:\msdchem\1\data\20210608_JGV A.D
Operator :
Acquired : 8 Jun 21 2:46 pm using AcqMethod ANALISIS GASESCPLOTMOLSIEVE.M
Instrument : GCMSD
Sample Name :
Misc Info :
Vial Number: 1

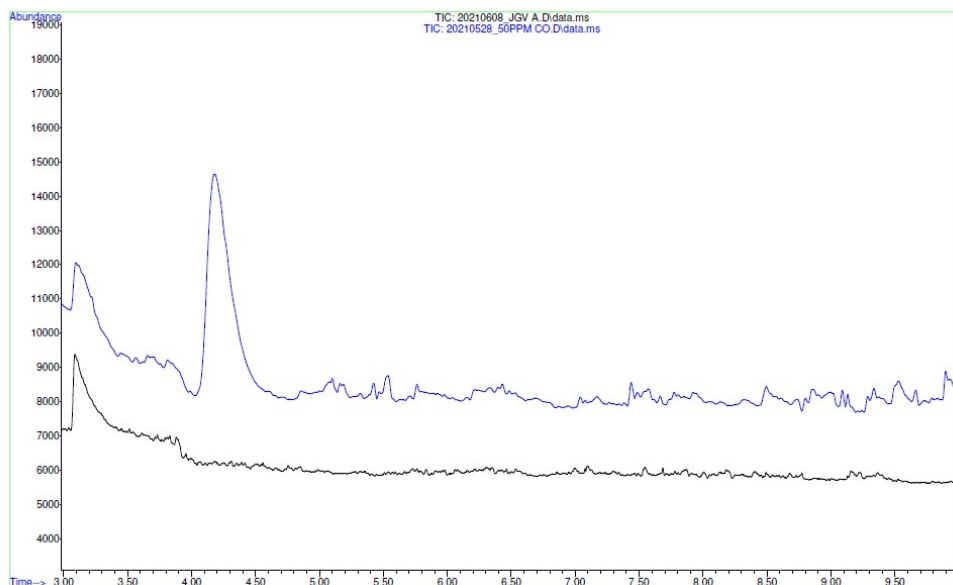


Figure S14. Gas chromatography of the gaseous mixture formed in the **2**-catalyzed (0.1 mol %) solvent-free FA dehydrogenation with 40 mol % of NEt_3 at 80°C

Arrhenius Plot

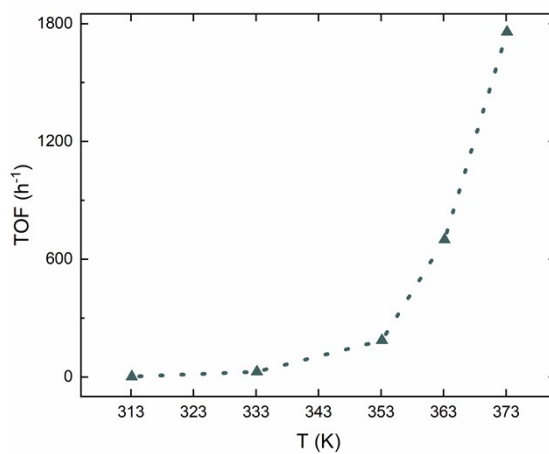


Figure S15. $\text{TOF}_{5\text{min}}$ (h^{-1}) versus T (K) from the **2**-catalyzed solvent-free FA dehydrogenation.

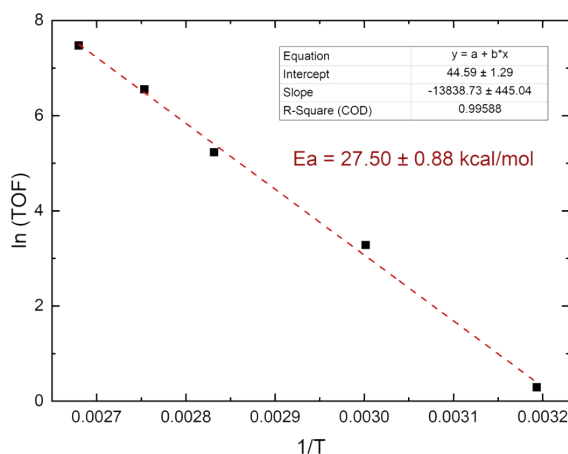
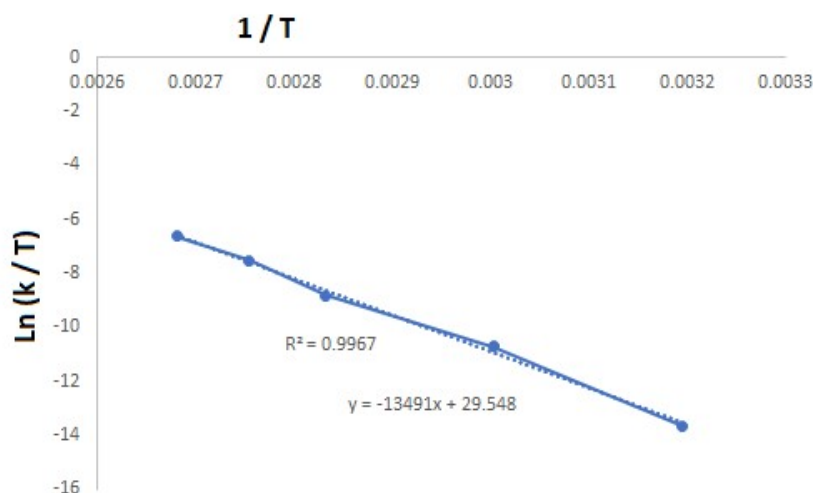


Figure S16. Arrhenius plot of the **2**-catalysed solvent-free FA dehydrogenation (apparent activation energy of 27.5 kcal mol⁻¹)

Eyring plot



Data:

| Point | T (°C) | T (K) | 1/T (K ⁻¹) | k (h ⁻¹) | k (s ⁻¹) | k/T | Ln (k/T) |
|-------|--------|--------|------------------------|----------------------|-----------------------|-----------------------|----------|
| 1 | 40 | 313.13 | 0.003194 | 1.3334 | $3.704 \cdot 10^{-4}$ | $1.183 \cdot 10^{-6}$ | -13.65 |
| 2 | 60 | 333.13 | 0.003002 | 26.5894 | $7.386 \cdot 10^{-3}$ | $2.217 \cdot 10^{-5}$ | -10.717 |
| 3 | 80 | 353.13 | 0.002832 | 186.3074 | $5.175 \cdot 10^{-2}$ | $1.465 \cdot 10^{-4}$ | -8.828 |
| 4 | 90 | 363.13 | 0.002754 | 699.4523 | $1.943 \cdot 10^{-1}$ | $5.351 \cdot 10^{-4}$ | -7.533 |
| 5 | 100 | 373.13 | 0.002680 | 1757.6207 | $4.882 \cdot 10^{-1}$ | $1.308 \cdot 10^{-3}$ | -6.639 |

Figure S17. Representation of LnK versus 1/T; $y = -13491x + 29.548$; $R^2 = 0.9967$; Activation Enthalpy = 112.2 ± 1.7 kJ mol⁻¹ (26.8 ± 0.4 kcal mol⁻¹) Activation Entropy = 47.7 ± 6 J K⁻¹ mol⁻¹ (11.4 ± 1.5 cal K⁻¹ mol⁻¹). The errors in the activation parameters were calculated following error propagation formulas (P. M. Morse, M. D. Spencer, S. R. Wilson, G. S. Girolami, Organometallics, 1994, 13, 1646-1655).

KIE Experiments

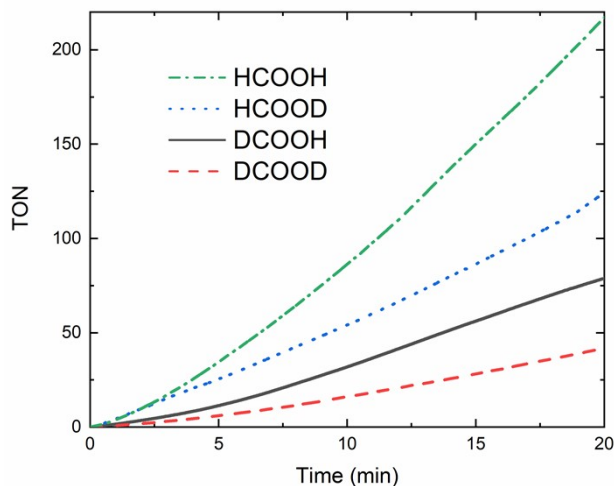


Figure S18. TON vs time representation of the **2**-catalyzed (0.1 mol %) solvent-free FA dehydrogenation with 40 mol % of NEt_3 at 80°C

Studies of the black precipitate

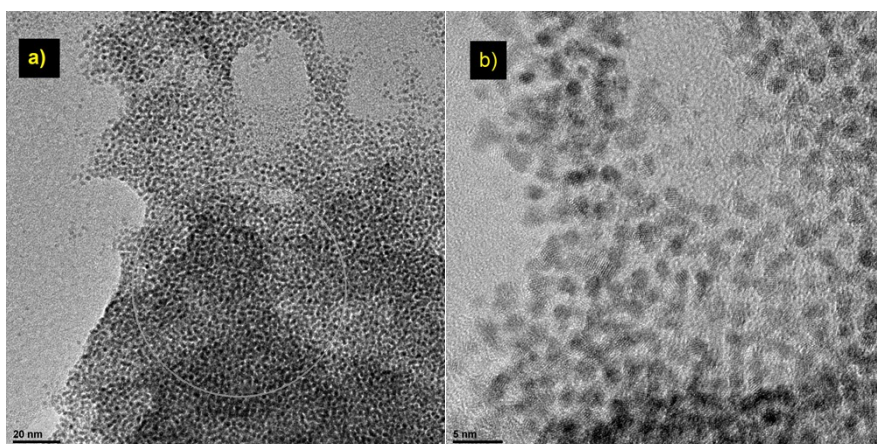


Figure S19. Transmission Electronic Microscopy images of the dark iridium nanoparticles. a) TEM image (20 nm). b) Closer image showing lattice fringes (5 nm).

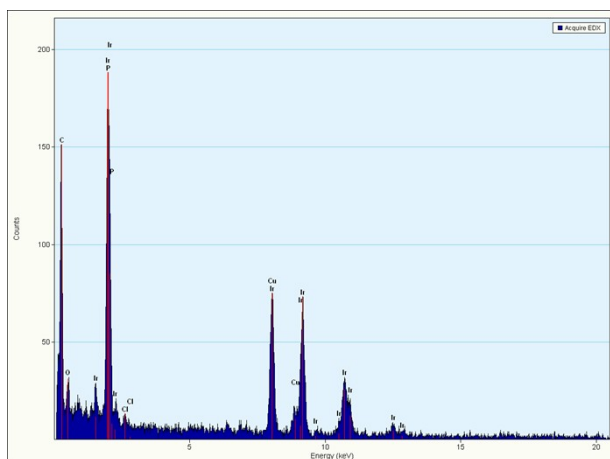


Figure S20. EDX spectrum of the iridium nanoparticles.

Mercury drop test

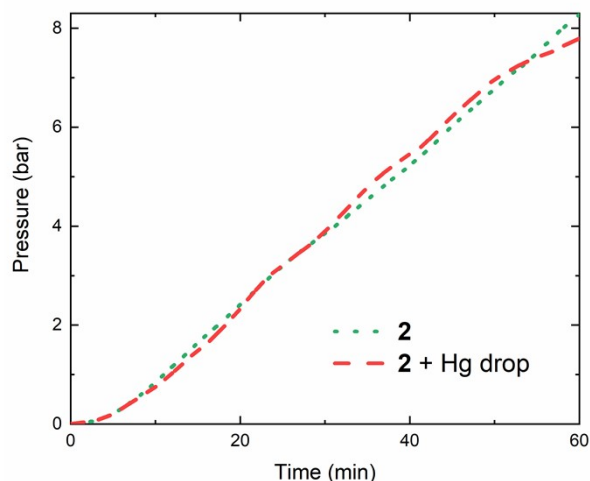


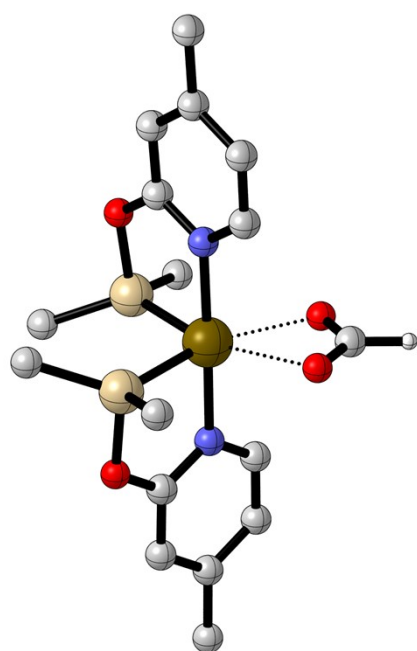
Figure S21. Pressure vs time representation of the **2**-catalysed (0.1 mol %) solvent-free dehydrogenation of FA with NEt_3 (10 mol %) at 80°C both in presence and absence of a mercury drop.

DFT Calculations

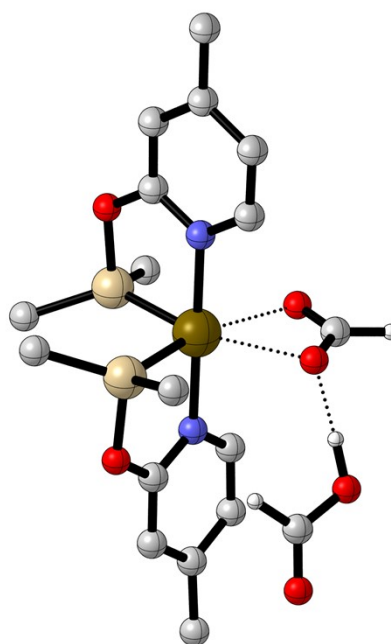
Table S1. Energetic values for all DFT calculated species. Geometrical optimizations performed at a B3LYP-D3(PCM)/def2-SVP level, E(DZ). Single point energies B3LYP-D3(BJ)/def2-TZVP, E(TZ). Frequencies were calculated at the same level than the optimization, G_{corr} , in the case of HCOOH the translational entropy contribution to the free energy was removed. Relative Gibbs free energies, ΔG , were calculated adding E(TZ) and G_{corr} and relative to **A** and isolated molecules. All absolute energies are given in a.u. and relative Gibbs free energies are given in kcal mol^{-1} .

| | E(DZ) | E(TZ) | G_{corr} | ΔG |
|---------------|------------|------------|-------------------|------------|
| A | -1756.1623 | -1757.5897 | 0.3124 | 0.0 |
| A + FA | -1945.8135 | -1947.4654 | 0.3399 | -9.8 |
| B | -1945.8198 | -1947.4656 | 0.3418 | -8.7 |
| TSAC | -1756.1136 | -1757.5458 | 0.3061 | 23.6 |
| TSBC | -1945.7719 | -1947.4232 | 0.3368 | 14.7 |
| C | -1945.7805 | -1947.4325 | 0.3334 | 6.8 |
| TSCA | -1945.7798 | -1947.4320 | 0.3302 | 5.1 |
| HCOOH | -189.6290 | -189.8587 | 0.0260 | |
| CO_2 | -188.4447 | -188.6740 | -0.0136 | |
| H_2 | -1.1741 | -1.1798 | -0.0044 | |

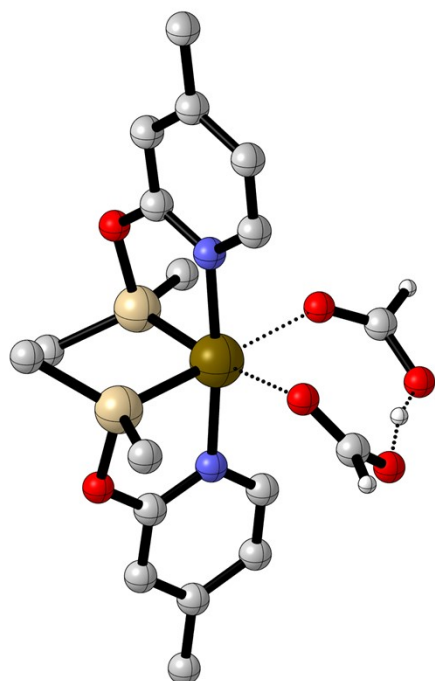
Figure S22. Geometrical representation for the DFT optimized structures. Only the most relevant hydrogen atoms are shown for clarity and distances are given in Å.



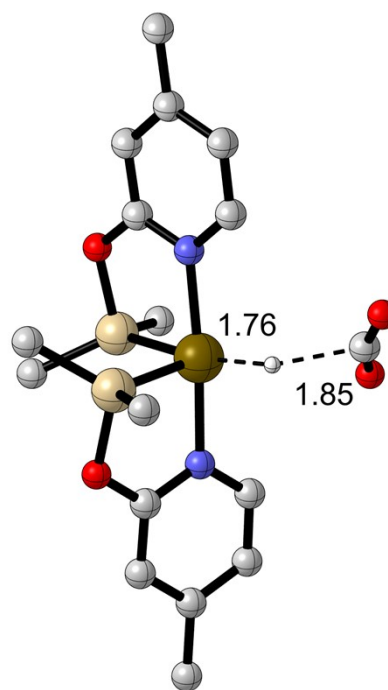
A



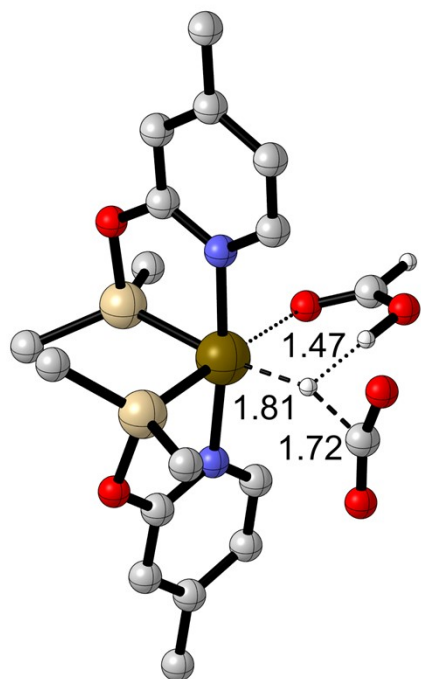
A + FA



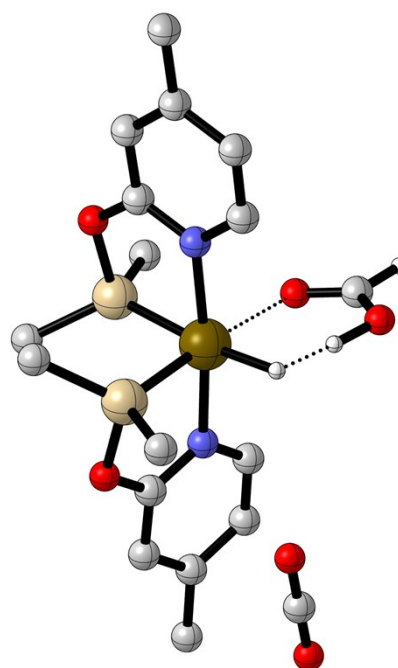
B



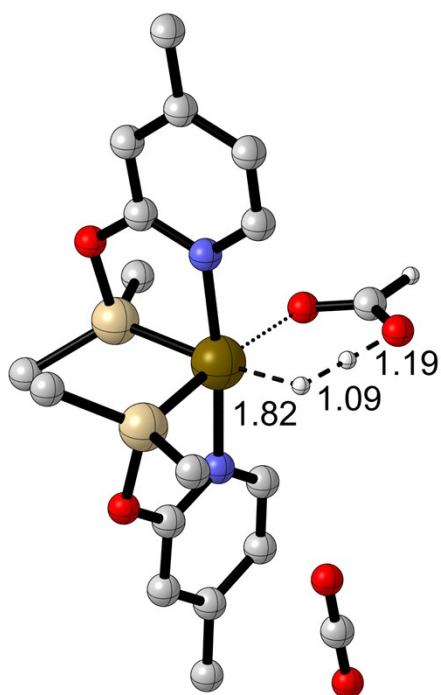
TSAC



TSBC



C



TSCA

Table S2. Cartesian coordinates (XYZ data) for all DFT calculated species

| 51 | | | | 56 | | | |
|----|-----------|-----------|-----------|------|-----------|-----------|-----------|
| A | | | | A+FA | | | |
| 77 | 0.000000 | 0.352515 | -0.000003 | 77 | 0.260001 | 0.002452 | -0.249494 |
| 7 | -2.080661 | 0.328582 | 0.036758 | 7 | 2.285406 | -0.446546 | -0.126522 |
| 6 | -2.784214 | 1.163413 | 0.840753 | 6 | 2.715503 | -1.724600 | 0.014248 |
| 6 | -2.755177 | -0.566472 | -0.733808 | 6 | 3.201301 | 0.557435 | -0.159253 |
| 6 | -4.162184 | 1.141334 | 0.914708 | 6 | 4.050942 | -2.051701 | 0.130584 |
| 1 | -2.180464 | 1.858477 | 1.425185 | 1 | 1.929077 | -2.480385 | 0.030397 |
| 6 | -4.160037 | -0.630614 | -0.699480 | 6 | 4.575565 | 0.281548 | -0.046452 |
| 6 | -4.886371 | 0.220338 | 0.126605 | 6 | 5.022580 | -1.027124 | 0.102098 |
| 1 | -4.675704 | 1.835387 | 1.581623 | 1 | 4.340546 | -3.097269 | 0.244229 |
| 1 | -4.651035 | -1.366142 | -1.337496 | 1 | 5.268150 | 1.123148 | -0.079060 |
| 7 | 2.080661 | 0.328582 | -0.036762 | 7 | -1.774703 | 0.432214 | -0.330962 |
| 6 | 2.755177 | -0.566462 | 0.733815 | 6 | -2.441205 | 0.737791 | 0.816064 |
| 6 | 2.784214 | 1.163405 | -0.840765 | 6 | -2.468824 | 0.366314 | -1.492878 |
| 6 | 4.160037 | -0.630604 | 0.699488 | 6 | -3.824566 | 0.975952 | 0.803236 |
| 6 | 4.162183 | 1.141326 | -0.914720 | 6 | -3.831053 | 0.578709 | -1.560123 |
| 1 | 2.180463 | 1.858462 | -1.425205 | 1 | -1.877964 | 0.121116 | -2.375776 |
| 6 | 4.886371 | 0.220338 | -0.126606 | 6 | -4.545794 | 0.882434 | -0.382441 |
| 1 | 4.651035 | -1.366124 | 1.337514 | 1 | -4.310924 | 1.200009 | 1.752920 |
| 1 | 4.675704 | 1.835373 | -1.581640 | 1 | -4.339728 | 0.501045 | -2.521815 |
| 8 | -2.072630 | -1.384300 | -1.510681 | 8 | 2.779543 | 1.800373 | -0.289915 |
| 8 | 2.072630 | -1.384282 | 1.510697 | 8 | -1.770700 | 0.784411 | 1.951622 |
| 14 | -0.308537 | -1.202831 | -1.635441 | 14 | 1.057535 | 2.109116 | -0.587408 |
| 14 | 0.308537 | -1.202808 | 1.635457 | 14 | -0.062793 | 0.302749 | 1.986187 |
| 6 | -0.029411 | -0.414006 | -3.327601 | 6 | 0.950069 | 2.523711 | -2.423666 |
| 1 | 1.054057 | -0.313230 | -3.509328 | 1 | 1.258471 | 1.668382 | -3.043146 |
| 1 | -0.455728 | -1.035681 | -4.132623 | 1 | -0.092425 | 2.773341 | -2.683168 |
| 1 | -0.474392 | 0.591255 | -3.376839 | 1 | 1.583170 | 3.392659 | -2.669105 |
| 6 | 0.337469 | -2.967048 | -1.718824 | 6 | 0.676768 | 3.667873 | 0.387370 |
| 1 | 0.228416 | -3.364612 | -2.740892 | 1 | -0.409062 | 3.755690 | 0.553658 |
| 1 | 1.407260 | -2.993064 | -1.454885 | 1 | 1.176467 | 3.670373 | 1.365201 |
| 1 | -0.201697 | -3.630525 | -1.029207 | 1 | 1.008756 | 4.554721 | -0.176347 |
| 6 | -0.337470 | -2.967023 | 1.718862 | 6 | 0.747251 | 1.541720 | 3.142376 |
| 1 | 0.201693 | -3.630505 | 1.029246 | 1 | 0.554226 | 1.259586 | 4.190083 |
| 1 | -0.228409 | -3.364580 | 2.740932 | 1 | 1.838149 | 1.552742 | 2.987183 |
| 1 | -1.407262 | -2.993043 | 1.454930 | 1 | 0.365296 | 2.558696 | 2.980464 |
| 6 | 0.029414 | -0.413958 | 3.327606 | 6 | -0.038736 | -1.375977 | 2.850716 |
| 1 | -1.054054 | -0.313174 | 3.509329 | 1 | -0.515886 | -2.159177 | 2.242796 |
| 1 | 0.455726 | -1.035622 | 4.132639 | 1 | 1.008879 | -1.676690 | 3.020816 |
| 1 | 0.474399 | 0.591303 | 3.376828 | 1 | -0.544393 | -1.325615 | 3.829330 |
| 8 | -0.065937 | 2.485206 | -1.106099 | 8 | -2.881251 | -2.956070 | -0.414228 |
| 6 | -0.000001 | 3.082773 | -0.000019 | 6 | -3.407745 | -2.408626 | 0.666439 |
| 1 | -0.000002 | 4.198157 | -0.000025 | 1 | -2.661158 | -1.949636 | 1.352706 |
| 8 | 0.065936 | 2.485218 | 1.106067 | 8 | -4.593076 | -2.405605 | 0.898795 |
| 6 | -6.386760 | 0.172752 | 0.185330 | 6 | 6.483851 | -1.352469 | 0.225049 |
| 1 | -6.815839 | 1.146319 | -0.101981 | 1 | 6.801482 | -2.016235 | -0.595376 |
| 1 | -6.796237 | -0.598612 | -0.480887 | 1 | 7.105995 | -0.447627 | 0.201753 |
| 1 | -6.727111 | -0.035280 | 1.212603 | 1 | 6.682302 | -1.890525 | 1.165939 |
| 6 | 6.386759 | 0.172747 | -0.185336 | 6 | -6.038372 | 1.039422 | -0.400061 |
| 1 | 6.815843 | 1.146341 | 0.101876 | 1 | -6.394047 | 1.636960 | 0.450888 |
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| | | | | | | | |
|----|-----------|-----------|-----------|------|-----------|-----------|-----------|
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| 7 | -2.014465 | -0.030834 | -0.012296 | TSAC | | | |
| 6 | -2.841263 | 0.843869 | 0.610131 | 77 | 0.016697 | 0.356339 | -0.082296 |
| 6 | -2.552874 | -1.166996 | -0.537132 | 8 | -0.979123 | 4.116454 | 0.359822 |
| 6 | -4.198820 | 0.634382 | 0.743134 | 8 | 0.503915 | 3.443484 | -1.287962 |
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| 7 | 2.156992 | 0.201979 | -0.055990 | 1 | 2.286622 | 1.782812 | -1.509614 |
| 6 | 2.819684 | -0.693359 | 0.726648 | 6 | 4.121258 | -0.735207 | 0.711898 |
| 6 | 2.870873 | 1.028070 | -0.858382 | 6 | 4.891401 | 0.016789 | -0.171033 |
| 6 | 4.224139 | -0.768019 | 0.699181 | 1 | 4.771070 | 1.574477 | -1.695876 |
| 6 | 4.249683 | 0.998372 | -0.920628 | 1 | 4.571349 | -1.465180 | 1.385627 |
| 1 | 2.272276 | 1.708461 | -1.464903 | 7 | -2.076535 | 0.219034 | -0.112723 |
| 6 | 4.961949 | 0.074535 | -0.125532 | 6 | -2.868589 | 1.206678 | 0.383954 |
| 1 | 4.706110 | -1.503249 | 1.344376 | 6 | -2.675754 | -0.921747 | -0.565585 |
| 1 | 4.772783 | 1.685298 | -1.587571 | 6 | -4.242643 | 1.108852 | 0.455860 |
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| 14 | -0.098015 | -1.546018 | -1.510000 | 6 | -4.882645 | -0.063648 | -0.001237 |
| 14 | 0.387149 | -1.212088 | 1.726505 | 1 | -4.817764 | 1.939545 | 0.867584 |
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| 1 | 0.803001 | -0.696969 | -3.667862 | 14 | 0.300124 | -0.807206 | 1.831810 |
| 1 | -0.432503 | -1.959481 | -3.944182 | 8 | 1.993064 | -1.278140 | 1.600676 |
| 6 | 0.974609 | -3.086041 | -1.398096 | 8 | -1.942901 | -1.900839 | -1.050878 |
| 1 | 1.231540 | -3.379940 | -0.372723 | 6 | -0.106234 | -1.312589 | -3.185419 |
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| 8 | 0.330526 | 2.071203 | 1.445578 | 1 | 0.652127 | -0.372180 | 4.267648 |
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| 56 | | | | 1 | -0.806120 | 1.511408 | -4.193495 |
| TSBC | | | | 1 | -0.520949 | 2.280469 | -1.467120 |
| 77 | 0.014436 | 0.113236 | -0.222998 | 56 | | | |
| 8 | -0.811855 | 3.861689 | 0.694950 | C | | | |
| 8 | 1.476783 | 3.538037 | 0.587716 | 77 | 0.273565 | -0.260216 | -0.147470 |
| 6 | 0.296127 | 3.464555 | 0.542347 | 1 | 0.014882 | -1.892084 | 0.510735 |
| 7 | 2.097101 | 0.027776 | -0.218009 | 7 | 2.357697 | -0.423748 | 0.051507 |
| 6 | 2.840209 | 0.288515 | -1.319366 | 6 | 2.966281 | -1.579868 | 0.421508 |
| 6 | 2.726413 | -0.387538 | 0.915107 | 6 | 3.136583 | 0.680084 | -0.143736 |
| 6 | 4.213073 | 0.144704 | -1.347436 | 6 | 4.326735 | -1.691840 | 0.620725 |
| 1 | 2.278391 | 0.609008 | -2.196920 | 1 | 2.301373 | -2.430876 | 0.557138 |
| 6 | 4.122142 | -0.555112 | 0.942239 | 6 | 4.531113 | 0.622310 | 0.054162 |
| 6 | 4.889096 | -0.293661 | -0.188567 | 6 | 5.149913 | -0.559110 | 0.441033 |
| 1 | 4.758069 | 0.366898 | -2.266009 | 1 | 4.747896 | -2.652861 | 0.919599 |
| 1 | 4.575078 | -0.892450 | 1.875091 | 1 | 5.100120 | 1.537565 | -0.112666 |
| 7 | -2.075858 | 0.016034 | -0.103657 | 7 | -1.770244 | 0.136665 | -0.333003 |
| 6 | -2.845789 | 1.126816 | 0.034359 | 6 | -2.371672 | 0.915956 | 0.606674 |
| 6 | -2.685713 | -1.203085 | -0.073398 | 6 | -2.500752 | -0.286934 | -1.389716 |
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| 1 | -2.309455 | 2.075205 | 0.046417 | 6 | -3.832654 | 0.037154 | -1.561538 |
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| 1 | -4.514970 | -2.308223 | 0.080982 | 1 | -4.368365 | -0.336922 | -2.435007 |
| 14 | -0.221475 | -2.200729 | -0.530365 | 8 | 2.588054 | 1.815143 | -0.517527 |
| 14 | 0.310778 | -0.106946 | 2.026859 | 8 | -1.669488 | 1.325122 | 1.644915 |
| 8 | 2.006098 | -0.640764 | 1.992909 | 14 | 0.875759 | 1.830846 | -1.045311 |
| 8 | -1.963753 | -2.300577 | -0.165423 | 14 | -0.076342 | 0.583556 | 1.938546 |
| 6 | -0.128455 | -2.817837 | -2.313516 | 6 | 1.036026 | 1.943546 | -2.929426 |
| 1 | -0.478464 | -3.861279 | -2.386754 | 1 | 0.031732 | 1.992709 | -3.383831 |
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| 6 | 0.608990 | -3.542898 | 0.502140 | 6 | 0.214358 | 3.516224 | -0.505291 |
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| 1 | 1.630967 | -3.678509 | 0.108748 | 1 | 0.020621 | 3.586941 | 0.573488 |
| 6 | -0.666878 | -1.355625 | 3.040158 | 6 | 1.022554 | 1.921625 | 2.678260 |
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| 6 | 0.384651 | 1.455890 | 3.080838 | 6 | -0.438971 | -0.630627 | 3.336744 |
| 1 | -0.571314 | 2.000361 | 3.020767 | 1 | 0.459307 | -1.232931 | 3.551503 |
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| 6 | 3.138296 | 0.679815 | -0.145333 | 6 | 6.637050 | -0.651447 | 0.627856 |
| 6 | 4.331629 | -1.703412 | 0.574404 | 1 | 7.081461 | -1.410717 | -0.035627 |
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| 6 | 4.532392 | 0.621757 | 0.048382 | 1 | 6.860550 | -0.961086 | 1.661753 |
| 6 | 5.152679 | -0.566335 | 0.413785 | 6 | -5.928594 | 1.194680 | -0.736388 |
| 1 | 4.755360 | -2.669315 | 0.853120 | 1 | -6.540150 | 0.300657 | -0.531041 |
| 1 | 5.100646 | 1.539724 | -0.105389 | 1 | -6.228353 | 1.983097 | -0.032499 |
| 7 | -1.763494 | 0.146270 | -0.333379 | 1 | -6.164510 | 1.522746 | -1.760297 |
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| 6 | -2.493504 | -0.277344 | -1.390810 | 8 | -3.322197 | -2.049901 | 1.398374 |
| 6 | -3.717240 | 1.289546 | 0.482739 | 8 | -5.559361 | -1.905700 | 0.779831 |
| 6 | -3.826405 | 0.043214 | -1.559526 | 5 | | | |
| 1 | -1.953228 | -0.882435 | -2.118750 | HCOOH | | | |
| 6 | -4.472422 | 0.854092 | -0.601406 | 6 | -0.070484 | 0.432479 | 0.020809 |
| 1 | -4.148117 | 1.914988 | 1.265261 | 8 | -1.114535 | -0.269021 | -0.016950 |
| 1 | -4.362620 | -0.330321 | -2.432878 | 8 | 1.059472 | -0.116657 | -0.051433 |
| 8 | 2.582101 | 1.817742 | -0.504061 | 1 | -0.143528 | 1.495878 | 0.114341 |
| 8 | -1.661539 | 1.337264 | 1.643588 | 1 | 1.006941 | -1.005329 | 0.307875 |
| 14 | 0.879719 | 1.817996 | -1.048673 | 3 | | | |
| 14 | -0.062232 | 0.607018 | 1.933571 | CO ₂ | | | |
| 6 | 1.043433 | 1.903100 | -2.930654 | 6 | 0.000000 | 0.000000 | 0.000000 |
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| 1 | 1.564063 | 1.013942 | -3.316248 | 2 | | | |
| 6 | 0.179520 | 3.487402 | -0.519999 | H ₂ | | | |
| 1 | 0.856353 | 4.309341 | -0.807705 | 1 | 0.000000 | 0.000000 | 0.380606 |
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| 1 | -0.029567 | 3.557735 | 0.555633 | | | | |