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Supporting Information: Dilution and Photooxidation Driven Processes Explain the Evolution of Organic Aerosol in Wildfire Plumes

Ali Akherati^{1^}, Yicong He¹, Lauren A. Garofalo², Anna L. Hodshire², Delphine K. Farmer², Sonia M. Kreidenweis³, Wade Permar⁴, Lu Hu⁴, Emily V. Fischer³, Coty N. Jen⁵, Allen H. Goldstein⁶, Ezra J. T. Levin³, Paul J DeMott³, Teresa L. Campos⁷, Frank Flocke⁷, John M. Reeves⁷, Darin W. Toohey⁸, Jeffrey R. Pierce^{3*}, and Shantanu H. Jathar^{1*}

¹Department of Mechanical Engineering, Colorado State University, Fort Collins, CO, USA

²Department of Chemistry, Colorado State University, Fort Collins, CO, USA

³Department of Atmospheric Science, Colorado State University, Fort Collins, CO, USA

⁴Department of Chemistry and Biochemistry, University of Montana, Missoula, MT, USA

⁵Department of Chemical Engineering, Carnegie Mellon University, Pittsburgh, PA, USA

⁶Department of of Environmental Science, Policy & Management, University of California Berkeley, Berkeley, CA, USA

⁷Earth Observing Laboratory, National Center for Atmospheric Research, Boulder, CO, USA

⁸Department of Atmospheric and Oceanic Science, University of Colorado Boulder, Boulder, CO, USA ^{*}Correspondence to: Shantanu H. Jathar (<u>shantanu.jathar@colostate.edu</u>) and Jeffrey R. Pierce

(jeffrey.pierce@colostate.edu)

[^]Now at the Department of Civil and Environmental Engineering, University of California Davis, Davis, CA, USA

Precursor Class	VOC Surrogate	ΔLVP	<i>m</i> _{frag}	p ₁₀	p ₂₀	p ₃₀	p ₄₀	Reference
SVOCs	naphthalene	1.4922	0.7673	0.8138	0.0072	0.0635	0.1155	Zhang et al. (2014)
Linear alkanes	<i>n</i> -dodecane	1.4629	0.2627	0.9657	0.0010	0.0020	0.0314	Loza et al. (2014)
Benzene ^{&}	benzene	1.5495	0.7895	0.0743	0.0213	0.8963	0.0081	Ng et al. (2007)
Toluene ^{&}	toluene	1.4169	1.3064	0.5634	0.3413	0.0016	0.0937	Zhang et al. (2014)
C ₈₊ single-ring aromatics ^{&}	<i>m</i> -xylene	1.4601	0.0736	0.1418	0.2971	0.4571	0.1040	Ng et al. (2007)
Multi-ring aromatics	naphthalene	1.4922	0.7673	0.8138	0.0072	0.0635	0.1155	Zhang et al. (2014)
Isoprene	isoprene	1.8742	0.5207	0.9924	0.0003	0.0065	0.0009	Chhabra et al. (2011)
Terpene	α-pinene	1.9139	0.1312	0.5991	0.2923	0.1079	0.0007	Chhabra et al. (2011)
Oxygenated aromatics	phenol, guaiacol	2.023	0.315	0.109	0.048	0.439	0.404	Yee et al.(2013)
Oxygenated aromatics	syringol	1.629	0.148	0.394	0.121	0.071	0.414	Yee et al. (2013)
Heterocyclics	2-methylfuran, dimethylfuran	1.459	0.449	0.0005	0.0014	0.998	0.0001	He et al. (2020)

Table S1: Precursor classes, surrogates, and parameters used in this work.

[&]Together, aromatic hydrocarbons. [^]Together, biogenics.

	Species	к он [cm³	MW	Surrogate
Species name	formula	molecules ⁻¹ s ⁻¹	lg mot ¹]	
Pyrrole	C ₄ H ₅ N	1.45×10 ⁻¹⁰	67.09	Heterocyclics
Furan	C ₄ H ₄ O	4.00×10 ⁻¹¹	68.07	Heterocyclics
Isoprene	C ₅ H ₈	1.00×10 ⁻¹⁰	68.12	Biogenics
Dihydropyrrole	C ₄ H ₇ N	1.04×10 ⁻¹⁰	69.10	Heterocyclics
Tetrahydropyrrole	C ₄ H ₉ N	7.85×10 ⁻¹¹	71.12	Heterocyclics
Benzene	C ₆ H ₆	1.22×10 ⁻¹²	78.11	Reduced Aromatics
Pyridine	C ₅ H ₅ N	3.70×10 ⁻¹³	79.10	Heterocyclics
Methylpyrrole	C ₅ H ₇ N	1.10×10 ⁻¹⁰	81.12	Heterocyclics
MethylFuran	C ₅ H ₆ O	7.80×10 ⁻¹¹	82.10	Heterocyclics
Furanone	C ₄ H ₄ O ₂	5.66×10 ⁻¹¹	84.07	Heterocyclics
Toluene	C7H8	5.63×10 ⁻¹²	92.14	Reduced Aromatics
2-Furancarbonitrile	C ₅ H ₃ NO	7.15×10 ⁻¹²	93.08	Heterocyclics
MethylPyridine	C ₆ H ₇ N	1.10×10 ⁻¹²	93.13	Heterocyclics
Phenol	C ₆ H ₆ O	2.80×10 ⁻¹¹	94.11	Oxygenated Aromatics
Furfural	C ₅ H ₄ O ₂	3.55×10 ⁻¹¹	96.08	Heterocyclics
Dimethylfuran	C ₆ H ₈ O	2.00×10 ⁻¹⁰	96.13	Heterocyclics
2-Methanolfuran	C ₅ H ₆ O ₂	1.04×10 ⁻¹⁰	98.10	Heterocyclics
Dihydrofurandione	C ₄ H ₄ O ₃	8.56×10 ⁻¹³	100.07	Heterocyclics
Benzonitrile	C7H5N	3.44×10 ⁻¹³	103.12	Reduced Aromatics
Styrene	C ₈ H ₈	5.80×10 ⁻¹¹	104.15	Reduced Aromatics
Vinylpyridine	C ₇ H ₇ N	2.66×10 ⁻¹¹	105.14	Heterocyclics
Benzaldehyde	C7H6O	1.20×10 ⁻¹¹	106.12	Oxygenated Aromatics
C8 Aromatics	C_8H_{10}	1.32×10 ⁻¹¹	106.16	Reduced Aromatics
Benzoquinone	C ₆ H ₄ O ₂	4.51×10 ⁻¹²	108.09	Oxygenated Aromatics
Cresol	C ₇ H ₈ O	5.30×10 ⁻¹¹	108.14	Oxygenated Aromatics
Benzenediol	C ₆ H ₆ O ₂	1.04×10 ⁻¹⁰	110.11	Oxygenated Aromatics
Trimethylfuran	C7H10O	1.59×10 ⁻¹⁰	110.15	Heterocyclics
Dihydroxypyridine	C ₅ H ₅ NO ₂	4.55×10 ⁻¹¹	111.10	Heterocyclics
5-Hydroxy 2-furfural	C ₅ H ₄ O ₃	4.90×10 ⁻¹¹	112.08	Heterocyclics
Nitrofuran	C ₄ H ₃ NO ₃	5.06×10 ⁻¹²	113.07	Heterocyclics
5-Hydroxymethyl-2[3H]-furanone	C ₅ H ₆ O ₃	1.00×10 ⁻¹⁰	114.10	Heterocyclics
5-Hydroxytetrahydro2-furfural	C ₅ H ₈ O ₃	5.00×10 ⁻¹²	116.11	Heterocyclics
Benzeneacetonitrile	C ₈ H ₇ N	2.07×10 ⁻¹²	117.15	Reduced Aromatics
Benzofuran	C ₈ H ₆ O	3.70×10 ⁻¹¹	118.13	Heterocyclics
Methylstyrene	C ₉ H ₁₀	5.40×10 ⁻¹¹	118.18	Reduced Aromatics
Tolualdehyde	C ₈ H ₈ O	1.60×10 ⁻¹¹	120.15	Oxygenated Aromatics
C9 Aromatics	$C_{9}H_{12}$	2.20×10 ⁻¹¹	120.19	Reduced Aromatics
Salicylaldehyde	$C_7H_6O_2$	2.80×10 ⁻¹¹	122.12	Oxygenated Aromatics
Dimethylphenol	$C_8H_{10}O$	5.05×10 ⁻¹¹	122.16	Oxygenated Aromatics
Hydroxy benzoquinone	$C_6H_4O_3$	1.30×10 ⁻¹¹	124.09	Oxygenated Aromatics
Guaiacol	C7H8O2	7.53×10 ⁻¹¹	124.14	Oxygenated Aromatics
Hydroxymethylfurfural	$C_6H_6O_3$	1.00×10^{-10}	126.11	Heterocyclics
Methyl benzofuran	C ₉ H ₈ O	9.75×10 ⁻¹¹	132.16	Heterocyclics
Methyl propenyl benzene	$C_{10}H_{12}$	3.30×10 ⁻¹¹	132.20	Reduced Aromatics
3-Methylacetophenone	$C_9H_{10}O$	2.42×10 ⁻¹²	134.17	Oxygenated Aromatics

Table S2: SOA precursors measured by the PTR-ToF-MS and included in the SOM-TOMAS model. Only the most dominant isomer is listed in the first column.

C10_Aromatics	C ₁₀ H ₁₄	9.50×10 ⁻¹²	134.22	Reduced Aromatics
Methylbenzoic Acid	$C_8H_8O_2$	1.20×10 ⁻¹¹	136.15	Oxygenated Aromatics
Monoterpenes	$C_{10}H_{16}$	1.63×10 ⁻¹⁰	136.23	Biogenics
Nitrotoluene	C7H7NO2	7.72×10 ⁻¹³	137.13	Reduced Aromatics
Methylguaiacol	$C_8H_{10}O_2$	3.97×10 ⁻¹¹	138.16	Oxygenated Aromatics
Methylnaphthalene	$C_{11}H_{10}$	5.65×10 ⁻¹¹	142.20	Reduced Aromatics
Product of levoglucosan dehydration (pyrolysis)	$C_6H_8O_4$	5.28×10 ⁻¹¹	144.12	Alkanes
Dimethylbenzofuran	C ₁₀ H ₁₀ O	1.20×10 ⁻¹⁰	146.18	Heterocyclics
Methylchavicol	C ₁₀ H ₁₂ O	5.43×10 ⁻¹¹	148.20	Oxygenated Aromatics
C11_aromatics	$C_{11}H_{16}$	5.00×10 ⁻¹¹	148.24	Reduced Aromatics
Vinylguaiacol	$C_9H_{10}O_2$	5.44×10 ⁻¹¹	150.17	Oxygenated Aromatics
Vanillin	$C_8H_8O_3$	2.73×10 ⁻¹¹	152.15	Oxygenated Aromatics
Camphor	$C_{10}H_{16}O$	4.30×10^{-12}	152.23	Biogenics
Syringol	$C_8H_{10}O_3$	9.66×10 ⁻¹¹	154.16	Oxygenated Aromatics
Cineole	$C_{10}H_{18}O$	2.26×10 ⁻¹¹	154.25	Biogenics
1,3-Dimethylnaphthalene	$C_{12}H_{12}$	6.94×10 ⁻¹¹	156.22	Reduced Aromatics
Decanal	$C_{10}H_{20}O$	3.45×10 ⁻¹¹	156.26	Alkanes
C12_aromatics	$C_{12}H_{18}$	1.13×10 ⁻¹⁰	162.27	Reduced Aromatics
Isoeugenol	$C_{10}H_{12}O_2$	8.84×10^{-11}	188.22	Oxygenated Aromatics
C13_aromatics	$C_{13}H_{20}$	1.13×10 ⁻¹⁰	176.30	Reduced Aromatics
Sesquiterpenes	$C_{15}H_{24}$	3.00×10^{-10}	204.35	Biogenics
5-MethylFurfural	$C_6H_6O_2$	5.18×10 ⁻¹¹	110.11	Heterocyclics



Figure S1: Physical age plotted against the sampling time for the five transect sets and four wildfire plumes. Physical age was calculated by dividing the straight-line distance from the fire with the average wind speed measured within a transect. Sampling time was calculated as the time of the day minus the time when the first transect was sampled but offset by the physical age at the first transect.



Figure S2: Box plots for (a) pressures and (b) temperatures encountered during the pseudo-Lagrangian evolution of smoke in the five transect sets and four wildfire plumes.



Figure S3: (a) POA volatility from May et al. (2013) and that used in this work expressed as the aerosol mass fraction against the OA mass concentration. The POA volatility based on the use of multi-ring aromatics was used with the base configuration of the SOM-TOMAS model while the POA volatility based on the use of other organic classes was used to perform sensitivity simulations. (b) Dependence of the OA O:C ratio as a function of the OA mass concentration. (c) Model species, shown in yellow, in the SOM grid used to represent the POA and SVOC mass. This distribution was informed based on the Monte-Carlo simulations performed and discussed in Section 3.2 in the main paper.



Figure S4: Averaged and normalized, (a) volatility distribution and (b) mass distribution in carbonoxygen space for the POA and SVOC emissions quantified by Jen et al. (2019). Volatility distribution from May et al. (2014) is also presented in panel (a). POA (c) O:C ratio and (d) aerosol mass fraction as a function of the OA mass concentration. Jen et al. (2019) only quantified 10 to 65% of the POA and SVOC mass collected on quartz filters. The volatility (c^{*}) was calculated for the POA and SVOC species either using EPISuite v4.11 (EPA, 2021) or the Statistical Oxidation Model (SOM) (Cappa and Wilson, 2012; Jathar et al., 2015).



Figure S5: Predictions of the (a) OA and (b) CO concentrations from the SOM-TOMAS model (solidcolored lines) compared against measurements (solid black circles) from the Taylor Creek Fire. Model predictions are shown for five different simulations that vary in their assumptions about POA volatility and oxidation chemistry.



Figure S6: Same as Figure 4 but with the standard error in the mean for the observations.



Figure S7: Predicted POA (dotted lines) and SOA (solid lines) ratios with OA for the base simulation results presented in Figures 4 and 5.



Figure S8: (a) POA volatility from May et al. (2013) and that based on 1000 Monte Carlo simulations. (b) OA O:C ratio as a function of the OA mass concentration for all 1000 Monte Carlo simulations. (c) SOM model species, shown in color, considered as options to represent the POA+SVOC mass.



Figure S9: Predictions of (a,b,c,d) OA NEMR and (e,f,g,h) OA O:C from the SOM-TOMAS model (solid colored lines) compared against measurements (solid black circles) from the four different transect sets. (i,j,k,l) Predictions of the fractional contributions of POA and SOA to OA. Model predictions are shown for sensitivity simulations performed with varying assumptions for the SVOC oxidation chemistry.



Figure S10: Predictions of (a,b,c,d) OA NEMR and (e,f,g,h) OA O:C from the SOM-TOMAS model (solid colored lines) compared against measurements (solid black circles) from the four different transect sets. (i,j,k,l) Predictions of the fractional contributions of POA and SOA to OA. Model predictions are shown for sensitivity simulations performed with varying assumptions for the POA volatility and SVOC oxidation chemistry.



Figure S11: Predictions of (a,b,c,d) OA NEMR and (e,f,g,h) OA O:C from the SOM-TOMAS model (solid colored lines) compared against measurements (solid black circles) from the four different transect sets. (i,j,k,l) Predictions of the fractional contributions of POA and SOA to OA. Model predictions are shown for sensitivity simulations performed with varying assumptions for OH.



Figure S12: Predictions of (a,b,c,d) OA NEMR and (e,f,g,h) OA O:C from the SOM-TOMAS model (solid colored lines) compared against measurements (solid black circles) from the four different transect sets. (i,j,k,l) Predictions of the fractional contributions of POA and SOA to OA. Model predictions are shown for sensitivity simulations performed by varying the 'first transect'.

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