Electronic Supplementary Material (ESI) for Environmental Science: Processes & Impacts. This journal is © The Royal Society of Chemistry 2022

Supporting Information

$\label{eq:continuity} Elucidation of sorbent \ role \ in \ sorption \ thermodynamics \ of \ Uranium (VI) \ on \\$ goethite

Sumit Kumar^{1,2*}, Rama Mohana Rao Dumpala³, Annapurna Chandane¹, Jitendra Bahadur⁴

¹Radioanalytical Chemistry Division, Bhabha Atomic Research Centre, Mumbai - 400 085.

²Homi Bhabha National Institute, Anushaktinagar, Mumbai – 400 094.

³Radiochemistry Division, Bhabha Atomic Research Centre, Mumbai - 400 085.

³Solid State Physics Division, Bhabha Atomic Research Centre, Mumbai – 400 085.

*: Corresponding Author

E-mail id: sumitk@barc.gov.in, Phone: +91-22-2559 5958.Fax: +91-22-2550 5151.

Figure S1. Potentiometric titration of goethite suspension (1 g/L in 0.1 M NaCl) by 0.1 M HCl/NaOH under anaerobic condition at 25 °C. The solid line corresponding to the data is the fit line calculated as per the surface complexation modelling adopted in the FITEQL program. The WSOS/DF for this fit was 9.9.

