## Electronic Supplementary Information

# Incorporation of catechyl monomers into lignins: Lignification from the non-phenolic end via Diels-Alder cycloaddition? 

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Figure S1. Diels-Alder reactions and products from 4-O-methyl coniferyl alcohol 6 and the benzoquinone 9 from methyl 5-hydroxyvanillate 11.

Figure S2. Proton NMR spectrum of Diels-Alder product 13.
Figure S3. Proton NMR spectrum of Diels-Alder product 15.
Figure S4. Carbon NMR spectra of Diels-Alder products 13 and 15.
Figure S5. Proton NMR spectra of benzodioxane products 12 and $\mathbf{1 2}^{\prime}$.
Figure S6. Carbon NMR spectra of benzodioxane products 12 and 12'.
Figure S7. ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ correlation spectra of the benzodioxane 12.
Figure S8. ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ correlation spectra of the benzodioxane $\mathbf{1 2}^{\prime}$.

Figure S1. Diels-Alder reactions and products from 4-O-methylconiferyl alcohol 6 and the benzoquinone 9 from methyl 5-hydroxyvanillate 11. Two Diels-Alder reactions are possible; all possible orientations/regiochemistries are shown, with the usual ortho- and anti-designations (with respect to the OMe on 6 and the diketone on 9). A-D) The conventional Diels-Alder reaction in which the diene itself acts as the diene component. Because of the electronic requirements (see Figure 6 and its caption in the main text), there is really only one regiochemical possibility, producing $130 a$ (designated as simply 13 in the main text) and its optical isomer - see text. E-F) The 'hetero-Diels-Alder' reaction in which the o-benzoquinone acts as the diene. Two orientations can be envisioned but only one satisfies the electronic requirements (see Figure 6 and its caption in the main text), both producing strictly transbenzodioxanes (because of the concerted nature of the reaction), but in which the regiochemistry, if it were described as resulting from radical coupling, appears as either a $\beta-O-5$ product 12 or a $\beta-O-$ 4 product 12'. In fact only the $\beta-O-5$ product 12 was observed (see Figure 6 and its caption in the main text).



With the quinone as the diene



Figure S2. Proton NMR spectrum of Diels-Alder product 13 (in $\mathrm{CDCl}_{3}$ ) showing horizontal expansions and assignments. Color coding uses that established in Figure 3 and used throughout to designate moieties derived from the 4-O-methylconiferyl alcohol 6 and from benzoquinone 9.


Figure S3. Proton NMR spectrum of Diels-Alder product 15 (in acetone- $\mathrm{d}_{6}$ ) showing horizontal expansions and assignments; uses the same plot-regions as Figure S2, including expansions, plus an expansion of the methoxyl region to show how methoxyls can be assigned. Note that this product is a mixture of two closely related isomers (see text); many of the resonances can be resolved into the two isomers but no attempt is made to individually assign them. Color coding uses that established in Figure 3 and used throughout to designate moieties derived from the coniferyl alcohol $\mathbf{1}$ and from benzoquinone $\mathbf{9}$, and distinguishing between the A-ring and the B -ring

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Figure S4. Carbon NMR spectra of: A) Diels-Alder product 13 (in $\mathrm{CDCl}_{3}$ ), and B) Diels-Alder product $\mathbf{1 5}$ (in acetone-d ${ }_{6}$ ) showing horizontal expansions and assignments; each uses the same plot-regions, with the more complex mixture of the two isomers of 15 having additional expansions to show the split nature of peaks not included in the expansions above. Color coding uses that established in Figure 3 and used throughout to designate moieties derived from the cinnamyl alcohol (1, 6, or a preformed dimer 14, Figure 3) and from benzoquinone 9.

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Figure S5. Proton NMR spectra (in $\mathrm{CDCl}_{3}$ ) of benzodioxane products, A) compound 12 from a Diels-Alder reaction, and B) compound 12' from a radical coupling reaction followed by simple phenol-methylation, showing the same horizontal expansions and assignments. Color coding uses that established in Figure 3 and used throughout to designate, in green, moieties derived from the 4-O-methylconiferyl alcohol $\mathbf{6}$ and, in magenta, from catechol $\mathbf{1 1}$ or its derived benzoquinone $\mathbf{9}$.


Figure S4. Carbon NMR spectra of 13 and 15 . See full caption on previous page, page 4.


Figure S5. Proton NMR spectra of 12 and 12'. See full caption on page 4.




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Figure S7. ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ Correlation spectra (in $\mathrm{CDCl}_{3}$ ) of the benzodioxane 12 from the Diels-Alder reaction between 4-O-methylconiferyl alcohol 6 and the benzoquinone 9 from methyl 5-hydroxyvanillate 11. HSQC (red contours and assignments) is overlaid (from a separate spectrum/experiment). HMBC spectrum (black correlations) shows (with green highlighting) correlations of $\mathrm{H} \beta$ to C 5 ' and H $\alpha$ to C4' that define the regiochemistry. Some correlations required vertical expansion (by the factor indicated beside the box) to be seen. Other correlations aid in assignments. Of major significance is that this benzodioxane $\mathbf{1 2}$ is the product of apparent $\beta-0-5-c o u p l i n g$, but the mechanism is in fact [4+2] cycloaddition (which therefore produces a different regioisomer). As shown by the green shading, the $\alpha$-proton correlates with six carbons: $\gamma, \beta, 2,6,1$, and, importantly, 4 ' (in chemical shift order). Similarly, the $\beta$-proton correlates with three carbons: $\alpha, 1$, and, importantly, $5^{\prime}$. Contours colored light gray are from residual 1-bond correlations (split by the 1-bond ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ coupling constant) from some peaks due to deviation from the set average value ( 145 Hz ) or are from minor impurities.

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Figure S8. ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ Correlation spectra (in $\mathrm{CDCl}_{3}$ ) of the benzodioxane $\mathbf{1 2}^{\prime}$ from the radical coupling reaction between coniferyl alcohol 1 and methyl 5-hydroxyvanillate 11 followed by simple phenolic methylation to produce the compound analogous to 12 and allow the determination that these are isomers and are not the same products. HSQC (red contours and assignments) is overlaid (from a separate spectrum/experiment). HMBC spectrum (black correlations) shows (with green highlighting) correlations of H $\alpha$ to C 5 ' and $\mathrm{H} \beta$ to C 4 ' that define the regiochemistry. Some correlations required vertical expansion (by the factor indicated beside the box) to be seen. Other correlations aid in assignments. Note also the beautiful correlations from the cis-benzodioxane (at $\sim 5 \%$ of the trans); just $\mathrm{H} \alpha$ and $\mathrm{H} \beta$ correlations are colored dark blue to highlight them. Of major significance is that this radical coupling benzodioxane $\mathbf{1 2}^{\prime}$ is the product of $\beta-0-4$-coupling, as is usual for coupling of a monolignol with a phenolic. As shown by the green shading, the $\alpha$-proton correlates with six carbons: $\gamma, \beta, 2,6,1$, and, importantly, $5^{\prime}$ (in chemical shift order). Similarly, the $\beta$-proton correlates with three carbons: $\alpha, 1$, and, importantly, $4^{\prime}$, establishing the $\beta-0-4$-coupling. Contours colored light gray are from residual 1-bond correlations (split by the 1-bond ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ coupling constant) from some peaks due to deviation from the set average value ( 145 Hz ) or are from minor impurities.


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