

Supplementary Information

Amorphous NiS_x film as a robust cocatalyst for boosting photocatalytic hydrogen generation over ultrafine ZnCdS nanoparticles

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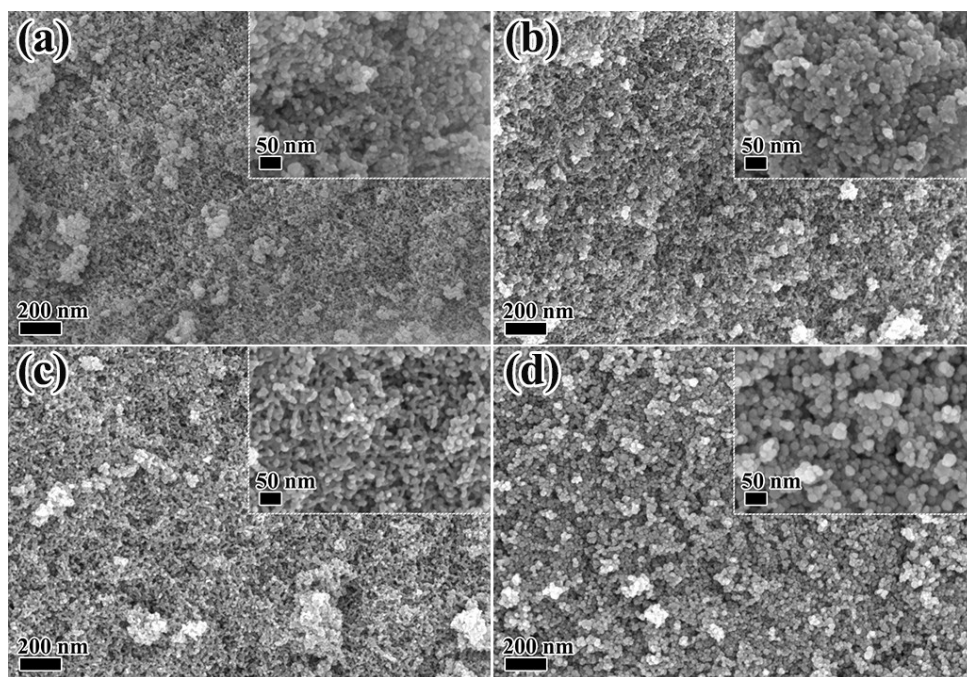


Fig. S1 SEM images: (a) ZCS, (b) ZCS-NS2, (c) CS-NS, and (d) ZS-NS nanocomposites.

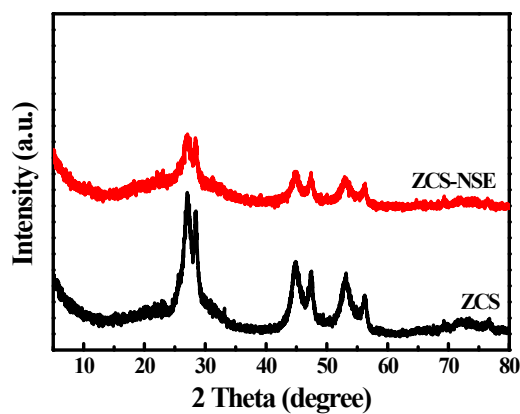


Fig. S2 The XRD patterns of ZCS and ZCS-NSE.

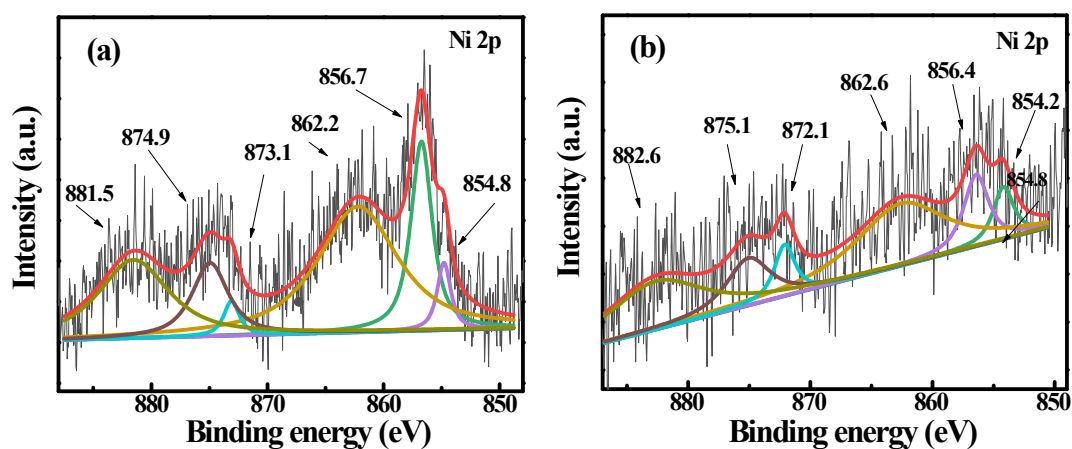


Fig. S3 High-resolution XPS spectra of Ni 2p of (a) CdS and (b) ZnS.

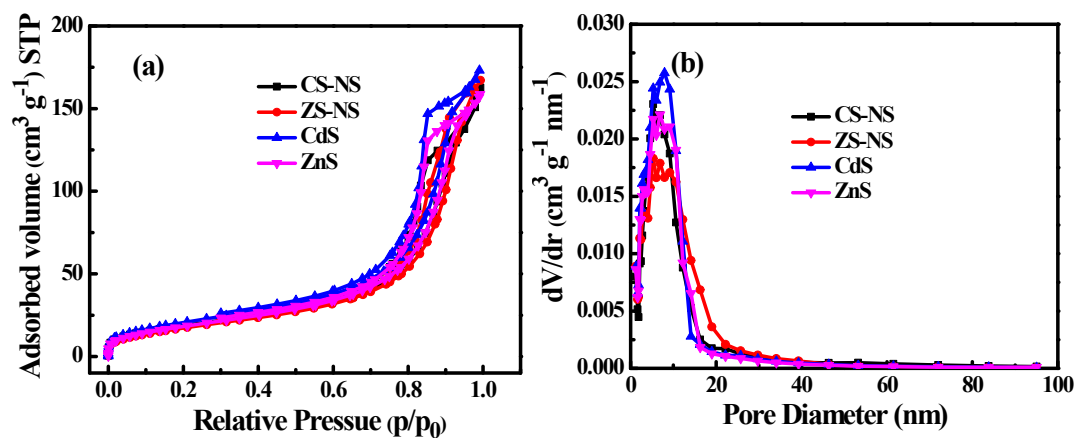


Fig. S4 Nitrogen adsorption/desorption isotherms and the corresponding pore-size distribution curves of the CS-NS, ZS-NS, CdS and ZnS.

As shown in Fig. S4 and Table S1, the present synthetic method can be extended to synthesize CS-NS, ZS-NS, CdS and ZnS with large specific surface areas, which is benefit to improve the photocatalytic performance.

Table S1 The corresponding textural properties of prepared samples.

Sample	S_{BET} ($\text{m}^2 \text{g}^{-1}$)	Mean pore diameter (nm)	Pore volume ($\text{cm}^3 \text{g}^{-1}$)
ZCS	102.23	11.75	0.30
ZCS-NS1	89.01	12.60	0.28
ZCS-NS2	96.30	11.90	0.29
ZCS-NS3	95.14	11.82	0.28
ZnS	70.66	13.77	0.247
ZS-NS	67.11	15.31	0.26
CdS	78.57	13.61	0.27
CS-NS	70.63	14.02	0.25

Table S2 Comparison of some nickel sulfide cocatalysts for photocatalytic hydrogen evolution.

Photocatalyst	Light source	Sacrificial agent	Activity $\text{mmol h}^{-1} \text{g}^{-1}$	Ref.
NiS/CdS	300 W Xe-lamp $\lambda > 420 \text{ nm}$	Na_2S , Na_2SO_3	1.131	1 ¹
NiS/ $\text{Zn}_{0.5}\text{Cd}_{0.5}\text{S}$	300 W Xe-lamp $\lambda > 420 \text{ nm}$	Na_2S , Na_2SO_3	16.780	2 ²
NiS/ $\text{Zn}_{0.5}\text{Cd}_{0.5}\text{S}$ / RGO	solar simulator	Na_2S , Na_2SO_3	7.514	3 ³
NiS/ $\text{Cd}_{0.4}\text{Zn}_{0.6}\text{S}$	AM 1.5, 100 mW cm^{-2} 300 W Xe-lamp $\lambda > 420 \text{ nm}$	Na_2S , Na_2SO_3	1.2	4 ⁴
CdS/NiS/RGO	300 W Xe-lamp $\lambda > 420 \text{ nm}$	lactic acid	14.960	5 ⁵
NiS/ ZnIn_2S_4	320 W Xe-lamp $\lambda > 420 \text{ nm}$	lactic acid	3.333	6 ⁶
C_3N_4 - $\text{Zn}_{0.5}\text{Cd}_{0.5}\text{S}$ -NiS	300 W Xe-lamp $\lambda > 420 \text{ nm}$	Na_2S , Na_2SO_3	53.19	7 ⁷
$\text{Zn}_{0.2}\text{Cd}_{0.8}\text{S}/\text{MoS}_2/\text{NiS}$	300 W Xe-lamp solar light irradiation	Na_2S , Na_2SO_3	41.29	8 ⁸
NiSx/ $\text{Cd}_{0.8}\text{Zn}_{0.2}\text{S}/\text{rGO}$	300 W Xe-lamp $\lambda > 420 \text{ nm}$	lactic acid	7.84	9 ⁹
$\text{Cd}_{0.5}\text{Zn}_{0.5}\text{S}(\text{en})/\text{NiS}$	300 W Xe-lamp $\lambda > 420 \text{ nm}$	Na_2S , Na_2SO_3	38.187	10 ¹⁰
$\text{CdS}/\text{La}_2\text{Ti}_2\text{O}_7/\text{NiS}_2$	300 W Xe-lamp $\lambda > 400 \text{ nm}$	lactic acid	12.77	11 ¹¹
NiS/g- C_3N_4	300 W Xe lamp an AM 1.5 G filter	triethanolamine	16.4	12 ¹²
NiS/CdS	300 W Xe-lamp $\lambda > 420 \text{ nm}$	lactic acid	30.1	13 ¹³
NiSx/CdZnS	300 W Xe-lamp $\lambda > 420 \text{ nm}$	lactic acid	67.75	This Work

Table S3 The photocatalytic hydrogen evolution rates under single-wavelength light and the corresponding AQE of ZCS and ZCS-NS2.

Sample	420 nm		500 nm		520 nm	
	H ₂ evolution (μmol g ⁻¹ h ⁻¹)	AQE	H ₂ evolution (μmol g ⁻¹ h ⁻¹)	AQE	H ₂ evolution (μmol g ⁻¹ h ⁻¹)	AQE
ZCS	216.3	0.155%	23.7	0.015%	5.7	0.003%
ZCS-NS2	23991.0	17.073%	3286.8	1.965%	92.3	0.053%

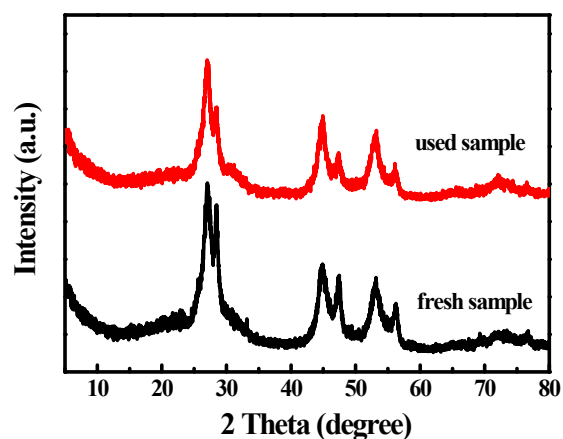


Fig. S5 The XRD patterns of the fresh and used ZCS-NS2 photocatalyst

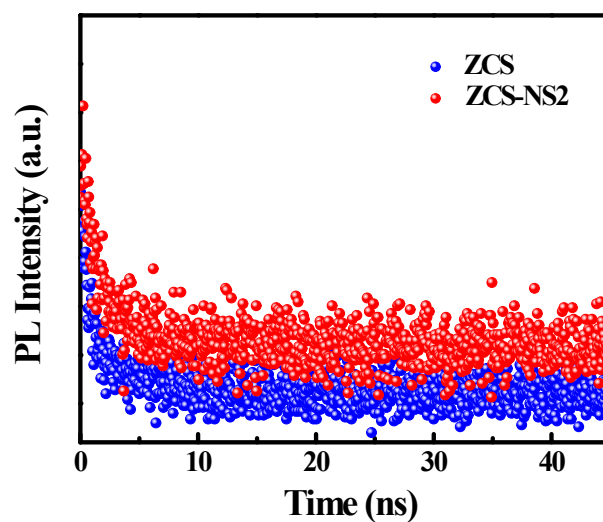


Fig. S6 Time-resolved PL decay under the excitation of 420 nm for ZCS and ZCS-NS2 samples.

Table S4 The radiative fluorescence lifetimes and relative percentages of the photoinduced charge carriers in the ZCS and ZCS-NS2

Sample	τ_1 (ns)	Rel. (%)	τ_2 (ns)	Rel. (%)	τ (ns)
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ZCS	1.02	43.27	4.82	56.73	3.18
ZCS-NS2	1.42	57.54	8.06	42.46	4.24

References

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