

## *Supporting Information*

### **Tailoring Co-Doping of Cobalt and Nitrogen in Fullerene-Based Carbon Composite and Its Effect for Supercapacitive Performance**

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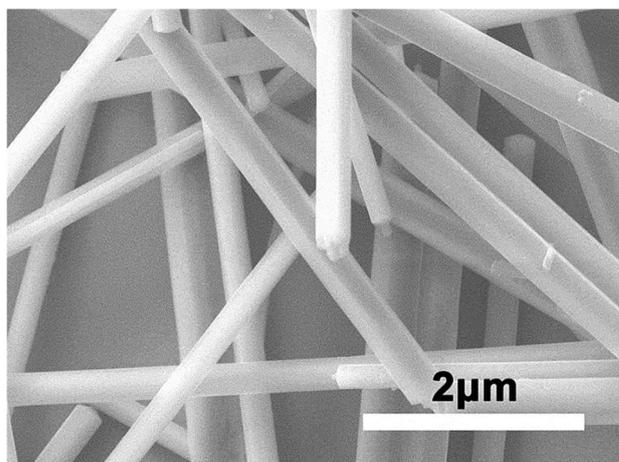
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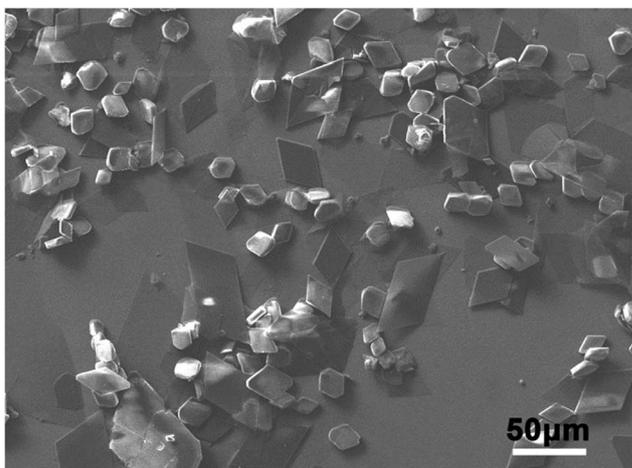
## Additional Data



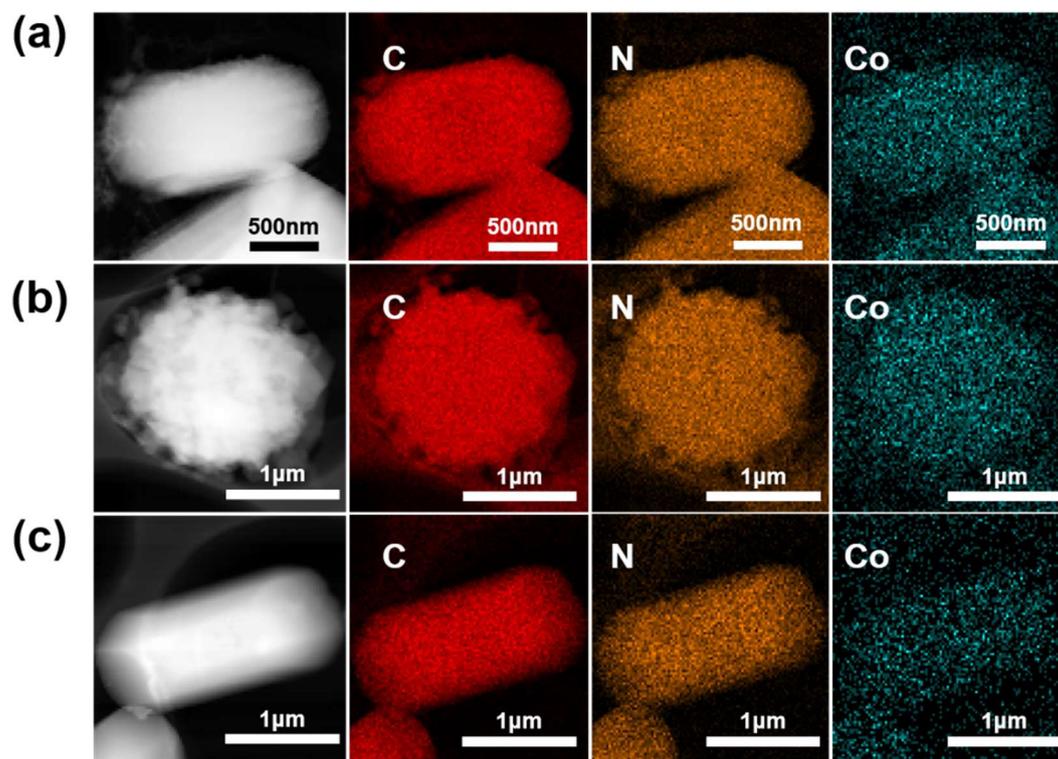
**Fig. S1.** SEM image of C<sub>60</sub>-rods by LLIP process of C<sub>60</sub>S in toluene-isopropyl alcohol solution.

**Table S1.** Structural features of C<sub>60</sub> rods and C<sub>60</sub>CoTMPP superstructures.

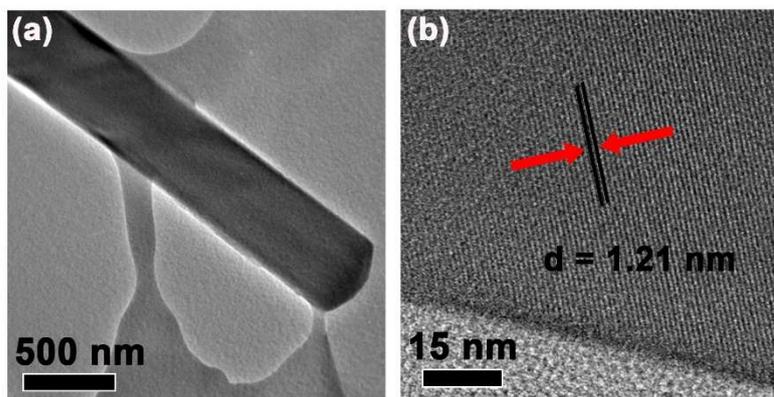
Sample	Morphology	Length (μm)	diameter (μm)
C <sub>60</sub> -rod (assembled structure from bare)	Long rod	10	0.15-0.40
C <sub>60</sub> CoTMPP-0.2	Short rods with polyprism structure	2.5	0.9
C <sub>60</sub> CoTMPP-0.1	Microsphere with lamellar structures		1.9
C <sub>60</sub> CoTMPP-0.05	Short rod	2.3	1.2



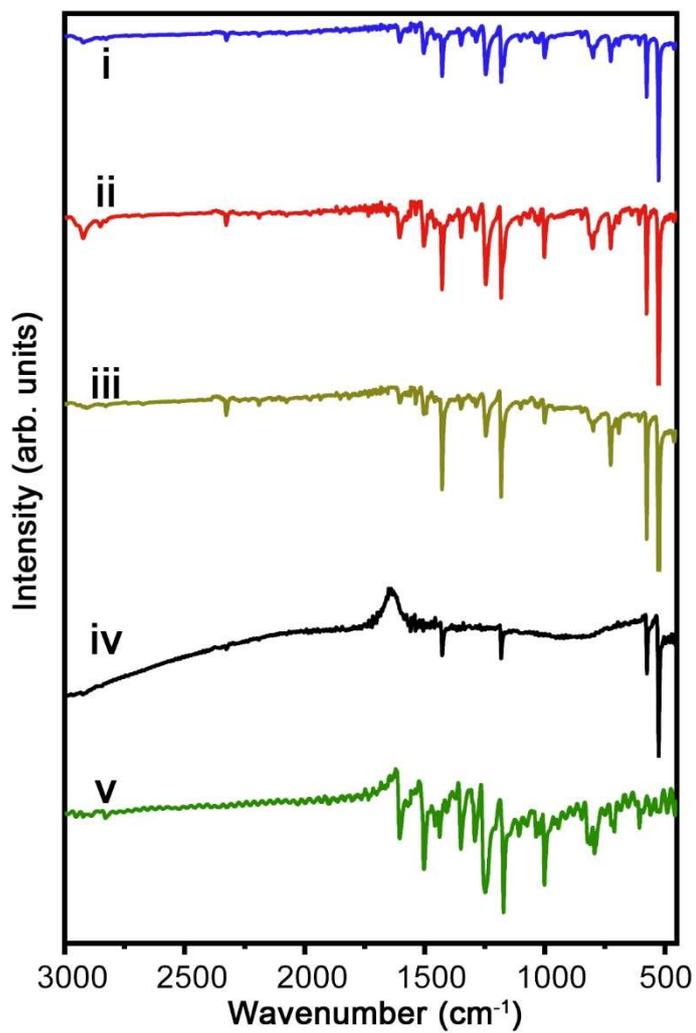
**Fig. S2.** SEM image of CoTMPP crystals formed in toluene-isopropyl alcohol solution.



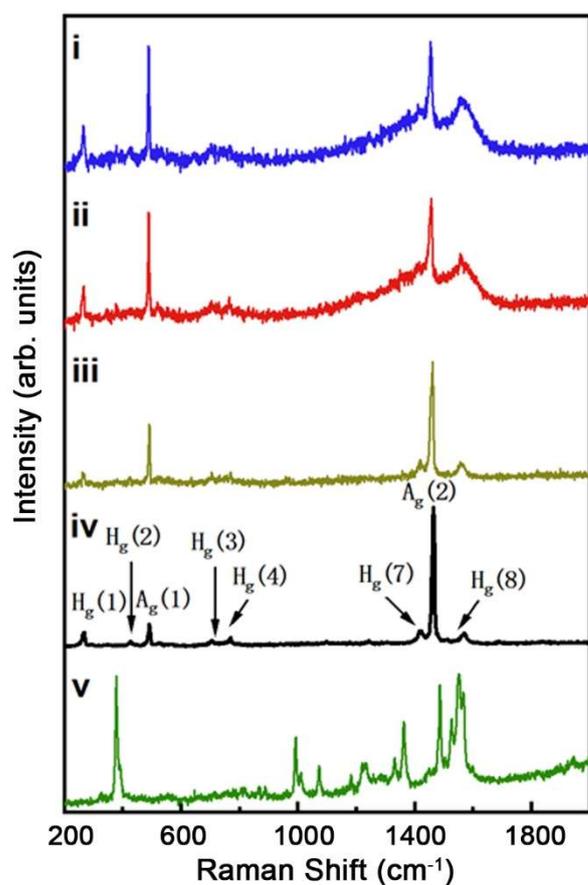
**Fig. S3.** STEM images and the corresponding elemental mapping images of C<sub>60</sub>CoTMPP crystals. (a) C<sub>60</sub>CoTMPP-0.2, (b) C<sub>60</sub>CoTMPP-0.1, (c) C<sub>60</sub>CoTMPP-0.05.



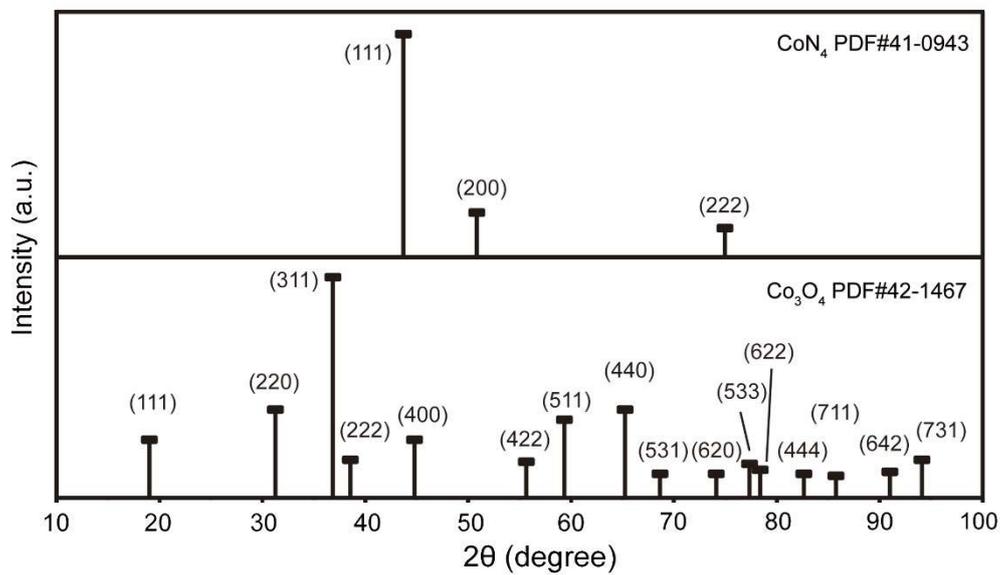
**Fig. S4.** TEM images of C<sub>60</sub> rod without CoTMPPs.



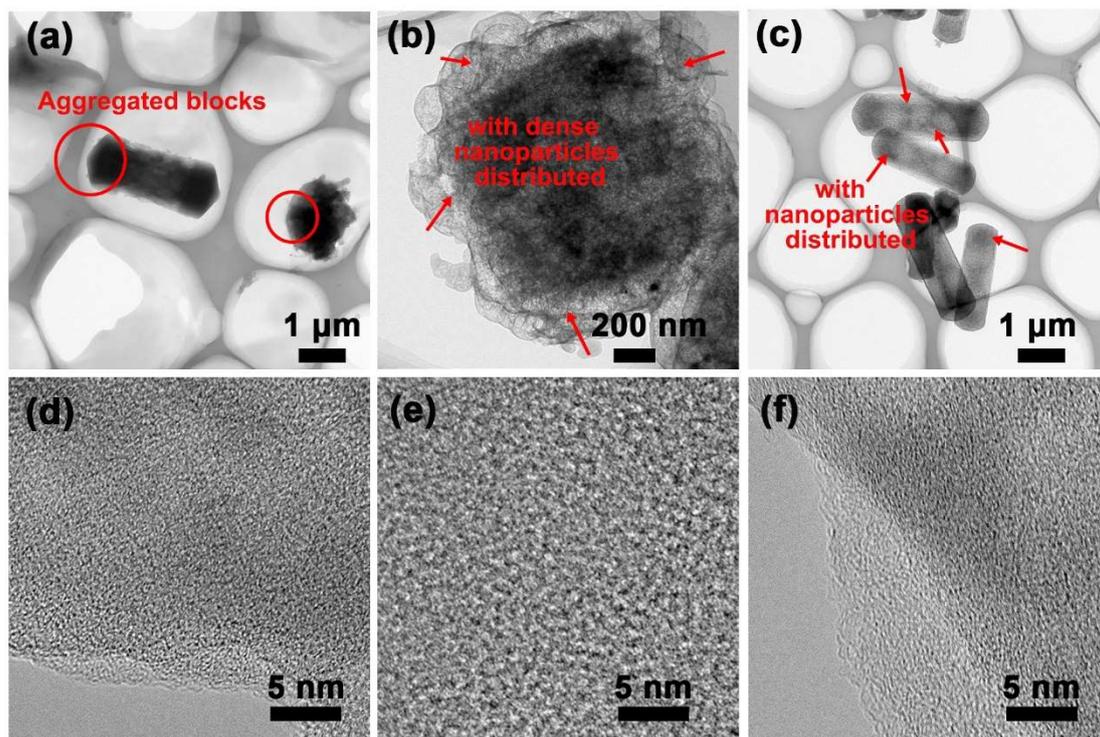
**Fig. S5.** FTIR spectra of C<sub>60</sub>CoTMPP crystals and CoTMPP. (i) C<sub>60</sub>CoTMPP-0.2, (ii) C<sub>60</sub>CoTMPP-0.1, (iii) C<sub>60</sub>CoTMPP-0.05, (iv) C<sub>60</sub>-rods, and (v) CoTMPP.



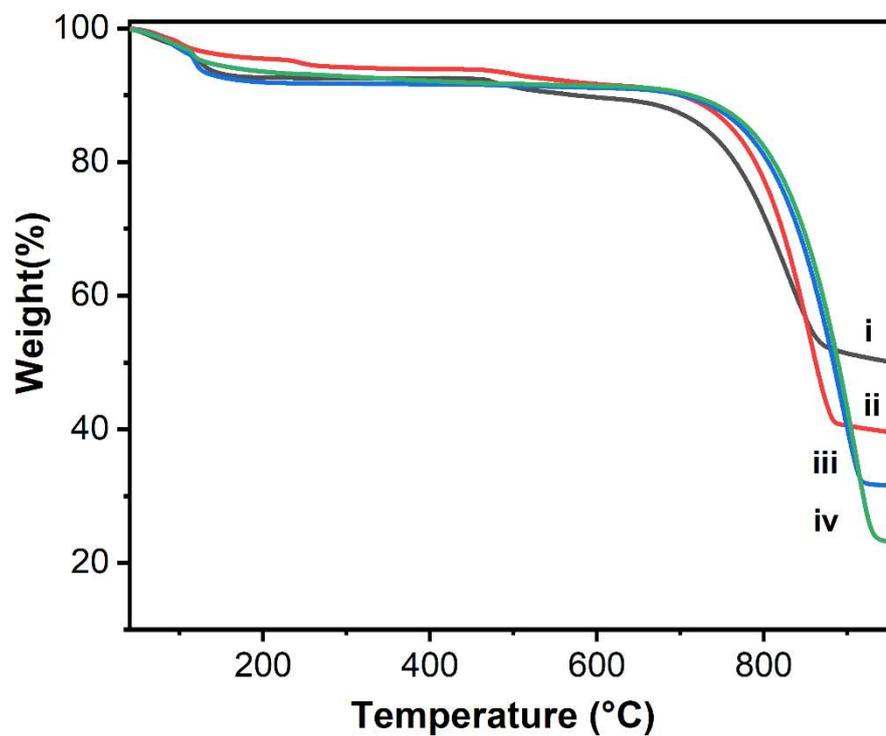
**Fig. S6.** Raman spectra of  $C_{60}CoTMPP$  superstructures and  $CoTMPPs$ . (i)  $C_{60}CoTMPP-0.2$ , (ii)  $C_{60}CoTMPP-0.1$ , (iii)  $C_{60}CoTMPP-0.05$ , (iv)  $C_{60}$ -rods, and (v)  $CoTMPP$ .



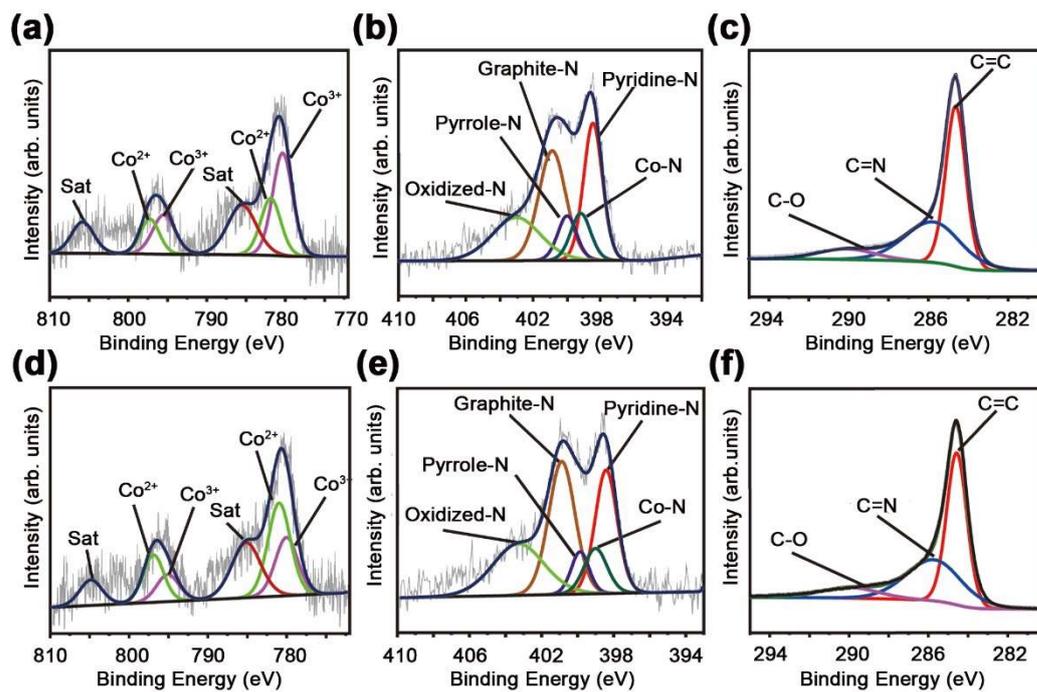
**Fig. S7.** XRD patterns of CoN<sub>4</sub> and Co<sub>3</sub>O<sub>4</sub> standard cards.



**Fig. S8.** TEM images of C<sub>60</sub>CoTMPP\_900s. (a), (d) C<sub>60</sub>CoTMPP-0.2\_900; (b), (e) C<sub>60</sub>CoTMPP-0.1\_900; (c), (f) C<sub>60</sub>CoTMPP-0.05\_900.



**Fig. S9.** The TGA weight loss curves of C<sub>60</sub>CoTMPPs. (i) C<sub>60</sub>CoTMPP-0.2, (ii) C<sub>60</sub>CoTMPP-0.1, (iii) C<sub>60</sub>CoTMPP-0.05, (iv) C<sub>60</sub>-rods.



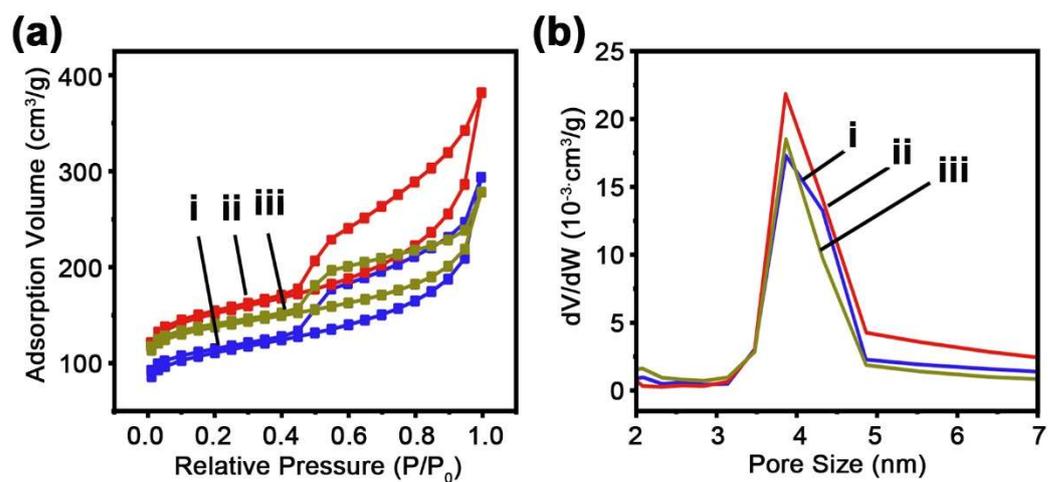
**Fig. S10.** (a) XPS Co 2p spectrum, (b) XPS N 1s spectrum, (c) XPS C 1s spectrum of C<sub>60</sub>CoTMPP-0.2\_900; (d) XPS Co 2p spectrum (e) XPS N 1s spectrum, (f) XPS C 1s spectrum of C<sub>60</sub>CoTMPP-0.05\_900.

**Table S2.** The comparison for the proportion of  $\text{Co}^{2+}$  and  $\text{Co}^{3+}$  in  $\text{C}_{60}\text{CoTMPP}_{900\text{s}}$  based on the relative area of each component peak in XPS Co 2p spectra, and the proportion of pyridine-N and Co-N in  $\text{C}_{60}\text{CoTMPP}_{900\text{s}}$  based on the relative area of each component peak in XPS N 1s spectra.

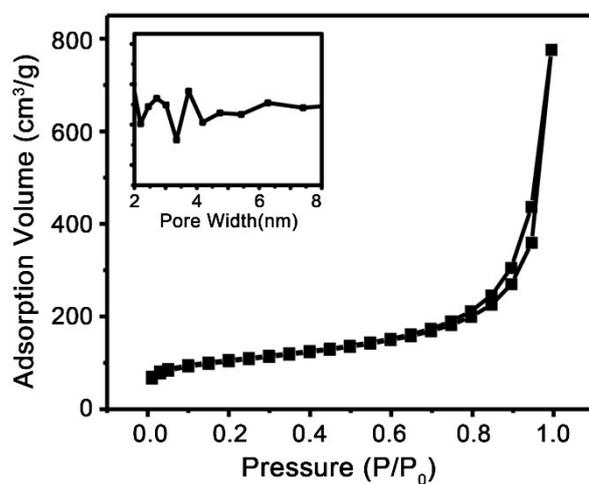
Samples	$\text{Co}^{3+}$	$\text{Co}^{2+}$	Pyridine-N	Co-N
$\text{C}_{60}\text{CoTMPP-0.2}_{900}$	44.24%	24.77%	27.35%	10.01%
$\text{C}_{60}\text{CoTMPP-0.1}_{900}$	24.51%	49.86%	26.49%	31.85%
$\text{C}_{60}\text{CoTMPP-0.05}_{900}$	26.24%	41.86%	25.53%	7.31%

**Table S3.** The Co amounts in  $\text{C}_{60}\text{CoTMPPs}$  and  $\text{C}_{60}\text{CoTMPP}_{900\text{s}}$  by ICP measurements.

Samples	$\text{C}_{60}\text{CoTMPP}_{-0.05}$	$\text{C}_{60}\text{CoTMPP}_{-0.1}$	$\text{C}_{60}\text{CoTMPP}_{-0.2}$	$\text{C}_{60}\text{CoTMPP}_{-0.05}_{900}$	$\text{C}_{60}\text{CoTMPP}_{-0.1}_{900}$	$\text{C}_{60}\text{CoTMPP}_{-0.2}_{900}$
Co mass% by ICP	0.26	0.57	1.26	0.30	0.71	1.52
Theoretical Co mass%	0.38	0.73	1.34	-	-	-



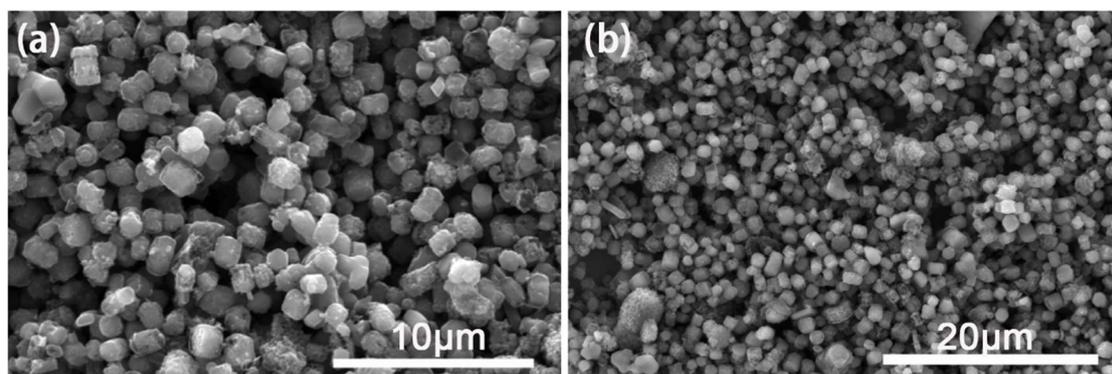
**Fig. S11.** (a) Nitrogen isotherms and (b) pore size distributions of the  $C_{60}CoTMPP\_900s$  (i)  $C_{60}CoTMPP-0.2\_900$ , (ii)  $C_{60}CoTMPP-0.1\_900$ , (iii)  $C_{60}CoTMPP-0.05\_900$ .



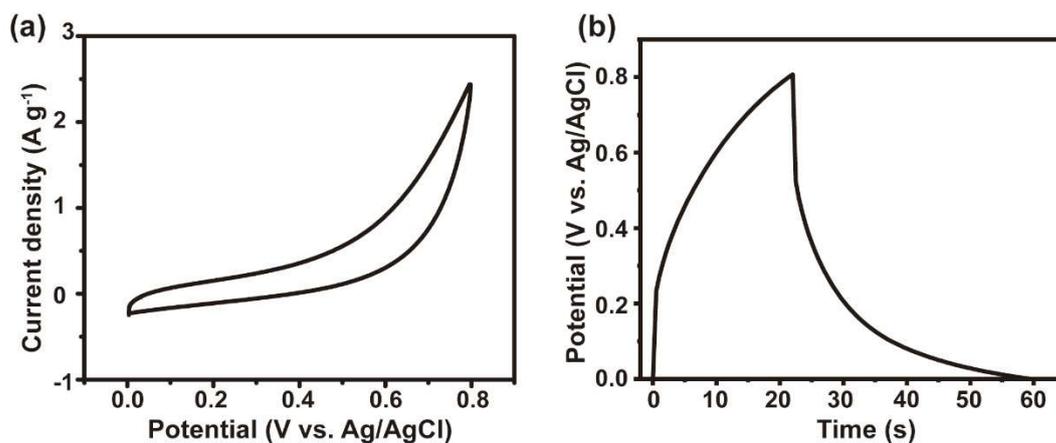
**Fig. S12.** Nitrogen isotherms and pore size distribution (inset) of the  $C_{60}\_900$  (rod morphology).

**Table S4.** The porous features of C<sub>60</sub>\_900 and various C<sub>60</sub>CoTMPP\_900s.

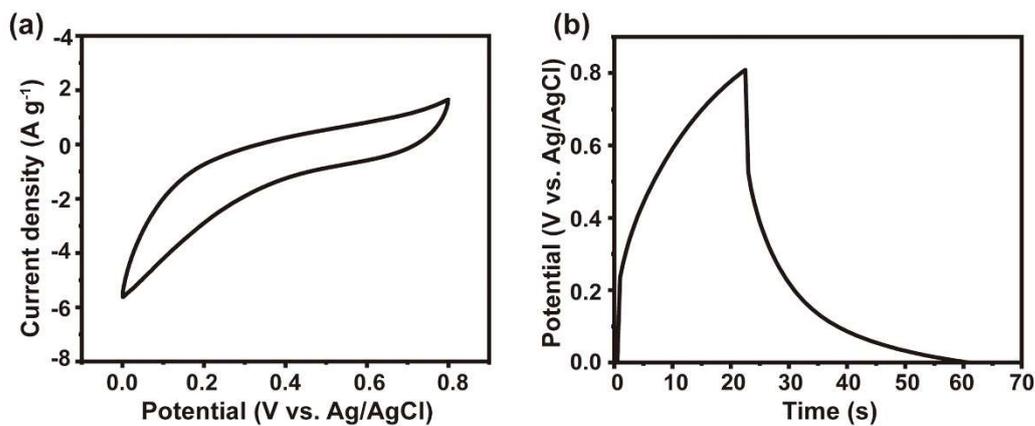
Sample	BET surface area (m <sup>2</sup> /g)	Average Pore size (nm)	Pore volume (cm <sup>3</sup> /g)
C <sub>60</sub> _900	287	2.3	0.059
C <sub>60</sub> CoTMPP-0.2_900	365	3.9	0.094
C <sub>60</sub> CoTMPP-0.1_900	496	3.9	0.143
C <sub>60</sub> CoTMPP-0.05_900	446	3.9	0.142



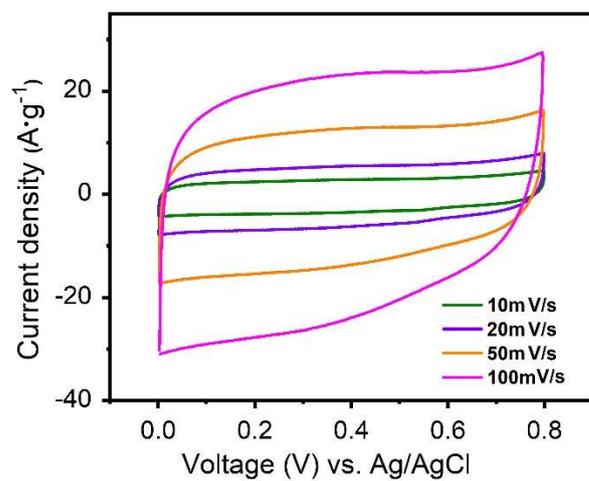
**Fig. S13.** SEM images of the C<sub>60</sub>CoTMPP-0.1\_900 on the electrode surface.



**Fig. S14.** (a) CV curve of  $C_{60\_900}$  at the scan rate of  $10 \text{ mV}\cdot\text{s}^{-1}$ , (b) CD curve of  $C_{60\_900}$  at  $1 \text{ A}\cdot\text{g}^{-1}$ .



**Fig. S15.** (a) CV curve of  $C_{60}\text{CoTMPP-0.1\_700}$  at the scan rate of  $10 \text{ mV}\cdot\text{s}^{-1}$ , (b) CD curve of  $C_{60}\text{CoTMPP-0.1\_700}$  at  $1 \text{ A}\cdot\text{g}^{-1}$ .



**Fig. S16.** CV curves of C<sub>60</sub>CoTMPP-0.1\_900 at different scan rates (10–100 mV·s<sup>-1</sup>).

**Table S5.** The comparison of the supercapacitor performance with various reported fullerene-derived carbon materials.

Materials	CD		CV		Ref.
	(F/g)	(A/g)	(F/g)	(mA/s)	
C <sub>60</sub> nanosheet	-	-	12.7	5	1
MF C <sub>60</sub>	141	0.5	-	-	2
HT-FNT_2000(C <sub>60</sub> )	-	-	145	5	3
HT-FNR_2000(C <sub>60</sub> )	-	-	132	5	3
Fe-MFC <sub>60</sub> -150	112.4	0.1	-	-	4
FCL700 (C <sub>60</sub> )	505.4	0.1	-	-	5
MCFC-900 (C <sub>70</sub> )	205	1	286	5	6
HTFT_2000(C <sub>70</sub> )	-	-	212	5	6
HTFT_900(C <sub>70</sub> )	-	-	26.4	5	7
MC <sub>60</sub> @C-1.33	213	0.5			8
200-HTC-800			2995	40	9
C <sub>60</sub> FcC <sub>60</sub> -8IPA_900	129	1	102.5	10	10
C <sub>60</sub> CoTMPP-0.1_900	416.31	1	296	10	this work

## References (in Table S5)

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