

Electronic Supplementary Information for:

Air-Grown Hybrid Copper (I) Halide Single Crystals: Structural Transformations and Ultraviolet-Pumped Photoluminescence Applications

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Fig. S1. Photos of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ single crystal at sunlight (A), and UV-254 nm (B) condition. (C) Powder and single-crystal X-ray diffraction (PXRD) patterns of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$. PLE and PL spectra of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ single crystal (D).

Fig. S2. Core level XPS spectra for (A) wide scan of $\text{Gua}_3\text{Cu}_2\text{I}_5$ single crystals, (B) C 1s, (C) N 1s, (D) I 3d, and (E) Cu 2p, fitted with peaks having an 80% Gaussian and 20% Lorentzian peak shape after applying background subtraction with Shirley function. Core level XPS spectra for (F) wide scan of $\text{Gua}_4\text{Cu}_4\text{Br}_8$ single crystals, (G) C 1s, (H) N 1s, (I) Cu 2p, and (J) Br 3d, fitted with peaks having an 80% Gaussian and 20% Lorentzian peak shape after applying background subtraction with Shirley function.

Fig. S3. Spatial distributions of VBM (A) and CBM (B) of $\text{Gua}_3\text{Cu}_2\text{I}_5$ at Γ point calculated by DFT.

Fig. S4. TGA/DSC diagrams of $\text{Gua}_4\text{Cu}_4\text{Br}_8$ and $\text{Gua}_3\text{Cu}_2\text{I}_5$.

Fig. S5. Photos of $\text{Gua}_3\text{Cu}_2\text{I}_5$ at room temperature and 100 °C under the irradiation of UV-254nm

Fig. S6. Absolute PLQY of $\text{Gua}_3\text{Cu}_2\text{I}_5$ powders based on BaSO_4 as standard material

Fig. S7. PL decay lifetime of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ powders.

Fig. S8. Absolute PLQY of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ powders based on BaSO_4 standard material

Table S1. Crystal structure data for $\text{Gua}_4\text{Cu}_4\text{Br}_8$, $\text{Gua}_3\text{Cu}_2\text{I}_5$ and $\text{Gua}_6\text{Cu}_4\text{I}_{10}$.

Movie S1. Video of UV-pumped LEDs based on $\text{Gua}_3\text{Cu}_2\text{I}_5$ single crystals.

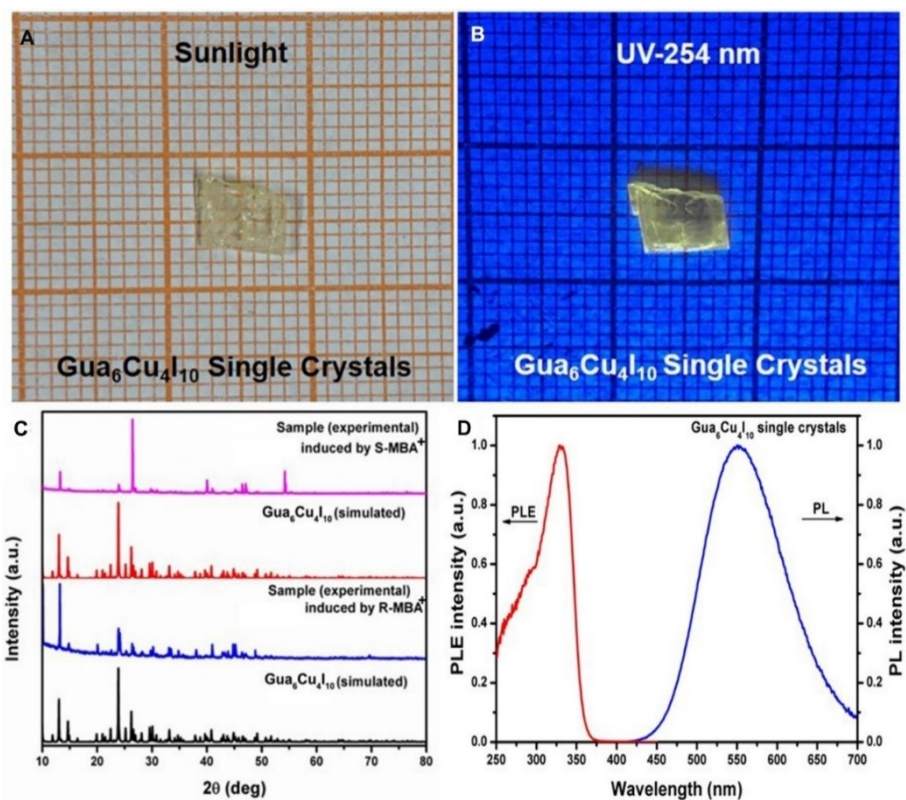


Fig. S1. Photos of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ single crystal at sunlight (A), and UV-254 nm (B) condition. (C) Powder and single-crystal X-ray diffraction (PXRD) patterns of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$. PLE and PL spectra of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ single crystal (D).

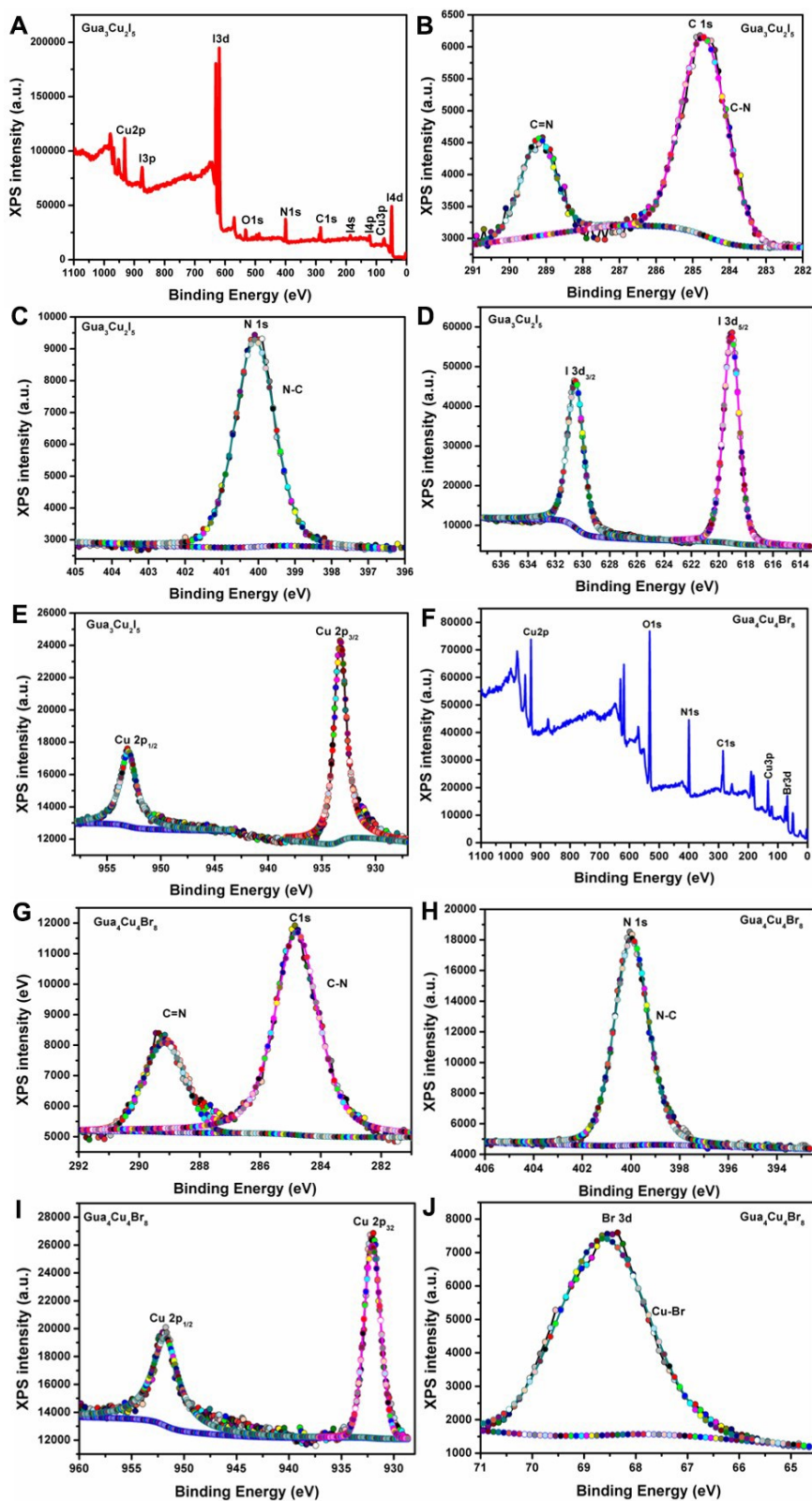


Fig. S2. Core level XPS spectra for (A) wide scan of $\text{Gua}_3\text{Cu}_2\text{I}_5$ single crystals, (B) C 1s, (C) N 1s, (D) I 3d, and (E) Cu 2p, fitted with peaks having an 80% Gaussian and 20% Lorentzian peak shape after applying background subtraction with Shirley function. Core level XPS spectra for (F) wide scan of $\text{Gua}_4\text{Cu}_4\text{Br}_8$ single crystals, (G) C 1s, (H) N 1s, (I) Cu 2p, and (J) Br 3d, fitted with peaks having an 80% Gaussian and 20% Lorentzian peak shape after applying background subtraction with Shirley function.

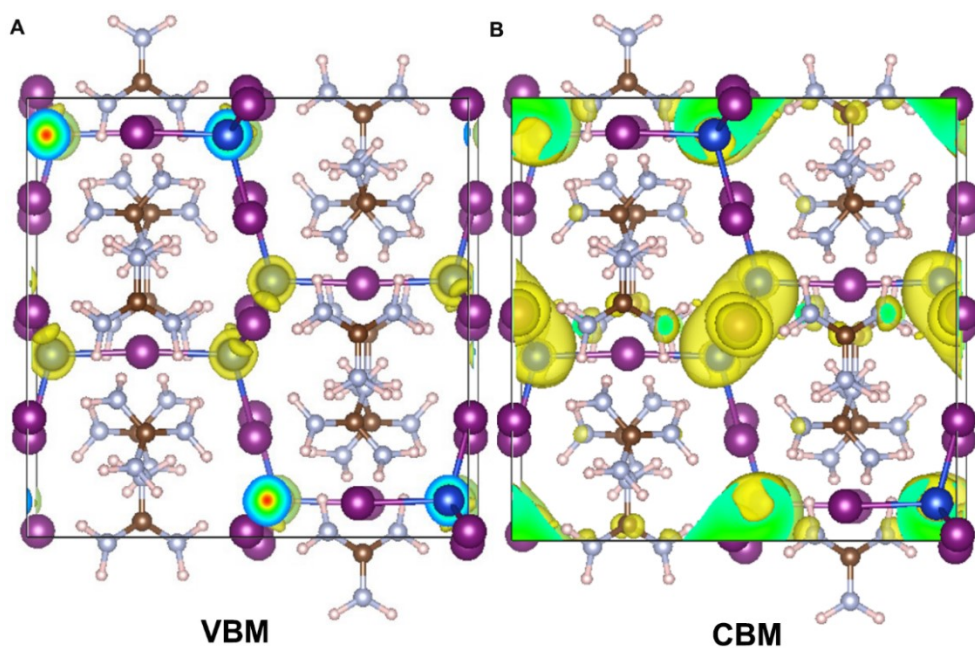


Fig. S3. Spatial distributions of VBM (A) and CBM (B) of $\text{Gua}_3\text{Cu}_2\text{I}_5$ at Γ point calculated by DFT.

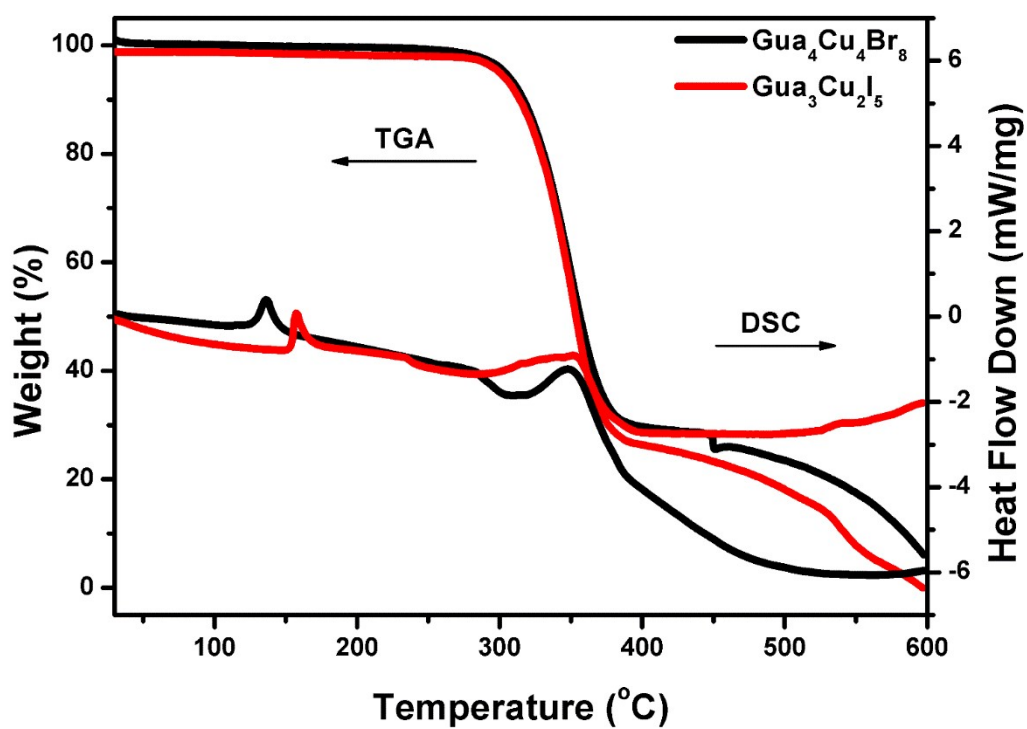


Fig. S4. TGA/DSC diagrams of $\text{Gua}_4\text{Cu}_4\text{Br}_8$ and $\text{Gua}_3\text{Cu}_2\text{I}_5$.

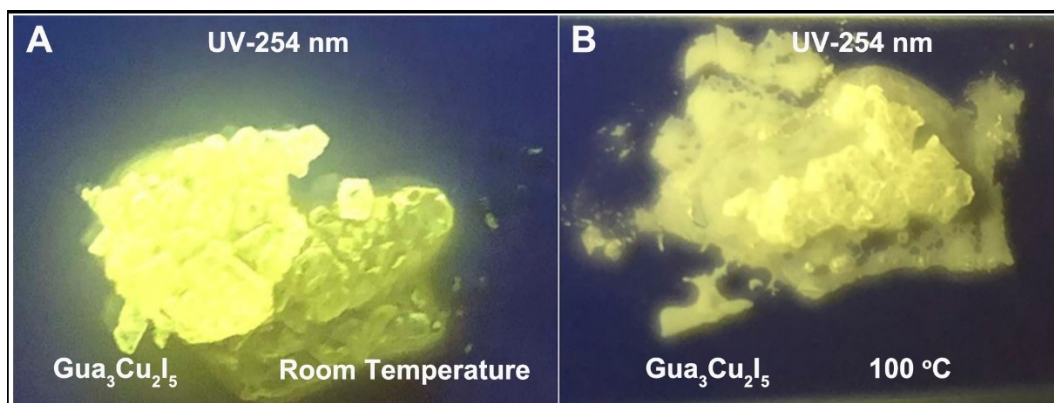


Fig. S5. Photos of $\text{Gua}_3\text{Cu}_2\text{I}_5$ at room temperature and 100 °C under the irradiation of UV-254nm

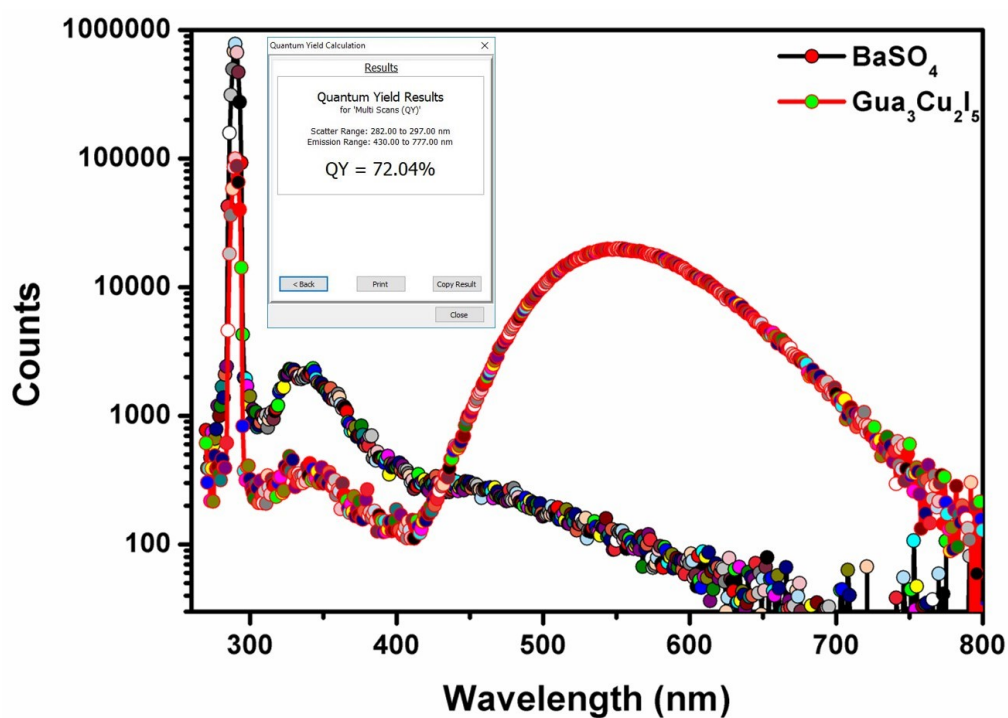


Fig. S6. Absolute PLQY of $\text{Gua}_3\text{Cu}_2\text{I}_5$ powders based on BaSO_4 standard material

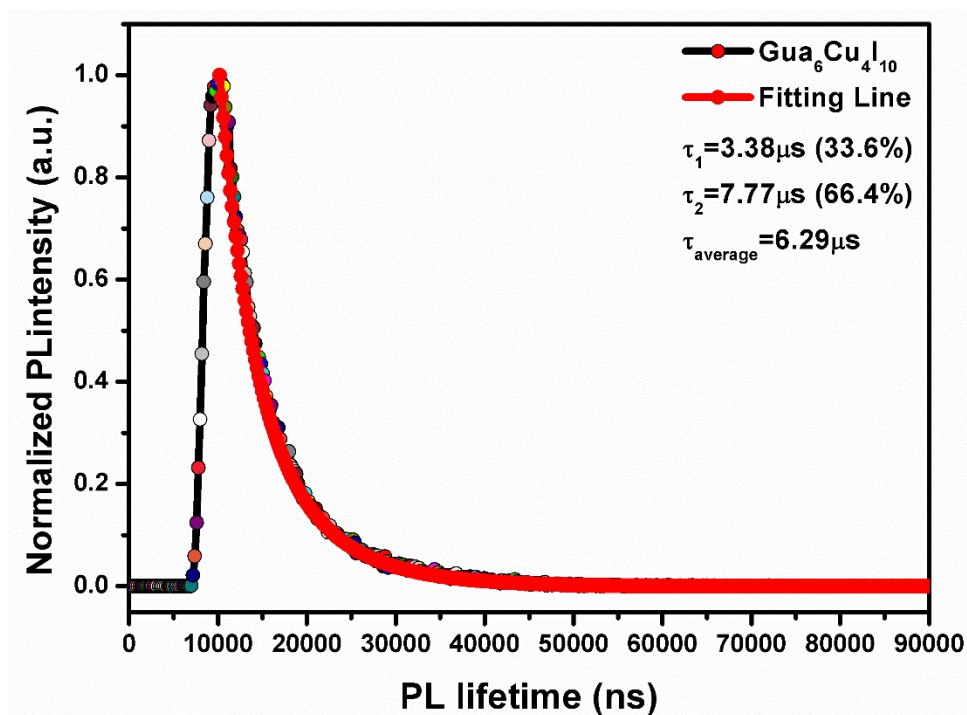


Fig. S7. PL decay lifetime of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ powders.

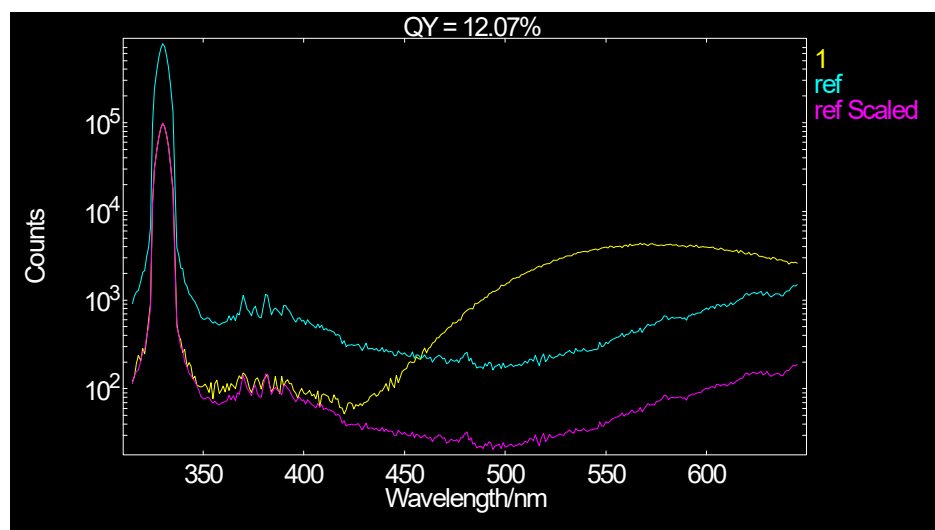


Fig. S8. Absolute PLQY of $\text{Gua}_6\text{Cu}_4\text{I}_{10}$ powders based on BaSO_4 standard material.

Table S1. Crystal structure data for Gua₄Cu₄Br₈, Gua₃Cu₂I₅ and Gua₆Cu₄I₁₀.

Identification code	Gua₄Cu₄Br₈	Gua₃Cu₂I₅	Gua₆Cu₄I₁₀
Empirical formula	C ₄ H ₂₄ Br ₈ Cu ₄ N ₁₂	C ₃ H ₁₈ Cu ₂ I ₅ N ₉	C ₆ Cu ₄ I ₁₀ N ₁₈
Formula weight/g·mol ⁻¹	1133.79	941.84	1847.40
Temperature/K	293(2)K	293(2)K	293(2)K
Wavelength/Å		0.71073	
Crystal color	Colorless	Colorless	Colorless
Crystal system	Monoclinic	Monoclinic	Orthorhombic
Space group	<i>P2₁/n</i> (no.14)	<i>C2/c</i> (no.15)	<i>Cmc2₁</i> (no.36)
a/Å	8.0772(2)	12.0317(9)	12.072(3)
b/Å	11.5386(4)	13.2491(10)	13.215(3)
c/Å	29.4561(11)	13.5695(10)	13.581(3)
α/°	90.00	90.00	90.00
β/°	96.1640(10)	90.985(7)	90.00
γ/°	90.00	90.00	90.00
Volume/Å ³	2729.42(15)	2162.8(3)	2166.6(9)
Crystal size (mm ³)	0.15 × 0.12 × 0.1	0.15 × 0.12 × 0.1	0.12×0.1×0.1
Z	4	4	2
Density/g·cm ⁻³	2.759	2.893	2.832
μ(mm ⁻¹)	14.804	9.105	9.087
F (000)	2112.0	1688.0	1616.0
GOF on F ²	1.079	1.111	1.087
Absolute Flack Factor	---	---	0.3(3)

Absorption correction	Semi-empirical from equivalents		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	6789/0/255	2204/0/91	2789/64/105
R ₁ , wR ₂ [I > 2σ (I)]	0.0397, 0.0855	0.0638, 0.1954	0.0593, 0.1295
R ₂ , wR ₂ (all data)	0.0670, 0.0970	0.0716, 0.1997	0.0797, 0.1416
Min/Max Δρ /eÅ ⁻³	-1.35/0.78	-2.48/3.17	-2.84/2.74
CCDC	2113135	2113134	2113136