Purification of quantum dots-based bioprobes with salting out

strategy

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Experimental Section

Materials

1-Octadecene (ODE, 90%), cadmium oxide (CdO, 99.99%), 1-octanethiol (98.5%), oleic acid (90%), trioctylphosphine (TOP), oleylamine (90%), 1-ethyl-3-(3-dimethylamino-propyl)carbodiimide (EDC), polyacrylic acid (MW 1800), and selenium powder (Se, 99.99%) were purchased from Aldrich. Amine-PEG-carboxyl (NH₂-PEG-COOH, MW=2000) was purchased from Laysan Bio. Anti-rabbit second antibody and anti-mouse second antibody (MW=150 KDa) were purchased from Bioss Inc (Massachusetts, US). Other chemicals including sodium chloride (NaCl), hexane, N, N-dimethylformamide (DMF), chloroform, ethanol, and octylamine were purchased from China National Pharmaceutical Group Corporation. Ultrapure water (18 M Ω . cm⁻¹) was prepared by a Millipore Milli-Q system.

Synthesis of CdSe/CdS QDs

The CdSe/CdS QDs was prepared according to previous reports with minimal change.^{1–3} Briefly, approximately 120 mg of CdO, 3 mL of oleic acid, 4 mL of oleylamine and 60 mL of ODE were added to a 250 mL three-necks flask. The resulting mixture was heated to 130 °C under vacuum for 20 min to remove oxygen and water. After purging 5 times with argon, the transparent solution were obtained and heated to 330 °C. Approximately 70 mg of Se powder dissolveed in 8 mL of TOP solution were injected into the three-necks flask swiftly. The resulting solution was rapidly cooled to 300 °C. Next, 1.0 mL of octanethiol dissolveed in 15 mL of ODE was injected into the three-necks flask at a flow rate of 1.2 mL/min. Finally, by removing heating source, the as-prepared CdSe/CdS QDs were cooled to room temperature. The CdSe/CdS were precipitated with the addition of ethanol and redispersed in hexane for water solubiliziation. CdSe/CdS QDs with green PL emission were prepared by controlling the amount of CdO and octanethiol.

Synthesis of OPA

Briefly, 1 g of polyacrylic acid and 1.5 g of EDC were dissolved in 50 mL of DMF. Approximately 700 μ L of octylamine was injected into the flask at a flow rate of 1.0 mL/20 min and stirred for overnight under the argon atmosphere. Next, DMF was removed by rotary evaporation, and 20 mL of hydrochloric acid (1 mol/L) was added, and a sticky white precipitate was obtained. After washed three times with deionized water, the resulted OPA were dissolved in ethanol and dried by rotary evaporation. Finally, the as-prepared OPA was stored in room temperature for the water-solubilization of CdSe/CdS QDs.

Preparation and OPA-QDs

Hydrophobic QDs (1 mg) and 5 mg of OPA were dispersed in 50 mL of chloroform. Subsequently, a OPA-QDs membrane layer was obtained by using a rotary evaporation to remove the chloroform. Adding 5 mL of borate saline buffer (pH 9, 50 mmol/L) to obtain OPA-QDs solution. Transmission electron microscope (TEM) images were obtained on a JEM2010FEF (UHR) microscope with an acceleration voltage of 200 kV. For the hydrophilic OPA-QDs samples, approximately 0.5 μ L of sample (1 × 10⁻¹¹ mol) mixed with 2 μ L of phosphotungstic acid (2%) were dropped onto copper grids for TEM characterization. For hydrophobic samples, the CdSe/CdS QDs with appropriate concentration (1×10⁻¹¹ mol/L) can be directly dropped onto the copper grids for the measurements. Agarose gel electrophoresis was performed on a DYY-6D gel electrophoresis apparatus (Beijing LiuYi instrument factory) with 1% (w/v) agarose gel, the gel picture was obtained by alphaimager HP system (Alpha Innotech). Dynamic light scattering measurements were carried out on a Malvern Nano instrument (ZS-ZEN 3600, Malvern Instruments, UK).

Salting-out of OPA-QDs solution

Approximately 1 mL of NaCl (2 mol/L) were added into 50 μ L of OPA-QDs (3 μ mol/L) solution, OPA-QDs solution change turbid aqueous phases. After 2 hours, the NaCl-treated OPA-QDs solution could be again recovered to transparent. The resulting samples were filtered by a centrifugal filter device with the MWCO of 50 kDa to remove NaCl, and the purified OPA-QDs was dispersed in borate saline buffer (pH 7.4, 20 mmol/L) and stored in room temperature for further characterization.

Preparation of OPA-QDs-PEG and QDs-IgG nanoprobes

The preparation of OPA-QDs-PEG and QDs-IgG nanoprobes was carried out according to our previous literature.¹ Briefly, 10 mg of EDC and 10 mg of NH₂-PEG-COOH (MW 2000) dispersed in borate saline buffer (pH 7.4, 20 mmol/L) were mixed with 2 mL of purified OPA-QDs (3×10^{-9} mol). After stirred at 160 rpm for 1.5 h, the resulting OPA-QDs-PEG were purified by gel filtration on NAP-10 column (GE Healthcare) to remove the excess of NH₂-PEG-COOH. QDs-IgG nanoprobes were prepared by conjugating IgG to the surface of OPA-QDs-PEG. Briefly, 1 mg of IgG and 100 μ g of EDC dispersed in 1.5 mL borate saline buffer (pH 7.4, 20 mmol/L) containing OPA-QDs-PEG (1.0×10⁻⁹ mol). After stirred at 160 rpm for 2.0 h, the resulting QDs-IgG nanoprobes samples were stored in 5 °C. Based on the previous work,⁴ we speculate that the conjugation efficiency is 50 %.

Salting-out of IgG-QD bioprobes

Typically, approximately 1 mL of NaCl (2 mol/L) were added into 200 μ L of IgG-QD (3 μ mol/L). After 8 hours, the resulting samples were filtered by a centrifugal filter device with the MWCO of 50 kDa to remove NaCl, the purified IgG-QD can be dispersed in borate saline buffer (pH 7.4, 20 mmol/L) again and stored in room 5 °C for future uses. The SEC included A Shodex KW-804 size exclusion column (Showa Denko scientific instruments (Shanghai) Co., Ltd.) was utilized to monitor the purification results. Borate saline buffer (pH 7.2, 20 mmol/L) containing NaCl (0.2 mol/L) was utilized as the mobile phase with a flow rate of 0.5 mL/min. Typical injection volume was 100 μ L. An UV-Vis detector (SPD-20A, Shimadzu Corporation) at 210 nm was utilized to monitor the purification process.

QDs-Based Immunofluorescence Labeling

The QDs-Based immunofluorescence labelling procedure was carried out according to the instructions by Wuhan Jiayuan Quantum Dots Co., Ltd., Wuhan, China. The paraffin-embedded breast cancer tissues were provided as a gift from Hubei Cancer Hospital. Briefly, breast cancer tissues were dewaxed with xylene and dehydrated sequentially with 100%, 95%, 80%, and 70% alcohol. Two breast cancer tissues retrieve were performed in boiling EDTA (pH 9.0, 10 mM) solution for 3 min. Subsequently, one breast cancer tissues were incubated with rabbit antihuman *CK-8/18* monoclonal antibody, and the other breast cancer tissues were incubated with TBS-T (0.1 M Tris-base, 0.5% Tween, 0.8% NaCl, pH 7.6) to remove the excess primary antibodies. Then the as-prepared two kinds of QDs-IgG nanoprobes were added to incubate respectively with *CK8/18* and *P-63* on breast cancer tissues at 37 °C for 1 h, and washed three times with TBS-T to remove the excess QDs-IgG nanoprobes. The fluorescence signal of QDs was collected using the Caliper multispectral microscopy imaging system (Caliper Life Sciences, Hopkinton, MA, USA).



Fig. S1 Size distribution histograms of the as-prepared CdSe QDs, the corresponding size is 7.3 \pm 1.2 nm, the sizes are from measuring at least 150 individual QDs in TEM images



Fig. S2 Schematic diagrams for the synthesis of Octylamine-grafted poly-(acrylic acid) (OPA)



Fig. S3 picture of QDs before and after coated with OPA (left: QDs dispersed in hexane, right: OPA-QDs dispersed in water)



Fig. S4 Agarose gel electrophoresis image of OPA-QDs (1), OPA-QDs-PEG (2) and IgG-QD bioprobes (3)



Fig. S5 Agarose gel electrophoresis image of IgG-QD before (left, IgG-QD) and after (right, NaCl-treated IgG-QD) adding NaCl, the arrow represents the position of loading well.



Fig. S6 picture of IgG-QD before and after treated NaCl (left: IgG-QD dispersed in BR buffer; middle: IgG-QD treated NaCl; right: NaCl removed)



Fig. S7 the absorption and PL spectra of IgG-QD before and after adding NaCl.



Fig. S8 Quantum yield of QDs samples before and after adding NaCl. (1): hydrophobic CdSe/CdS QDs dispersed in hexane, (2):OPA-QDs before (green) and after (red) treated with NaCl, (3): IgG-QD bioprobes before (green) and after (red) treated with NaCl.

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