

Supporting Information

A single-band ratiometric luminescent thermometer based on tetrafluorides operating entirely in the infrared region

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KEYWORDS excited state absorption, SBR approach, luminescence thermometry, fluoride, neodymium

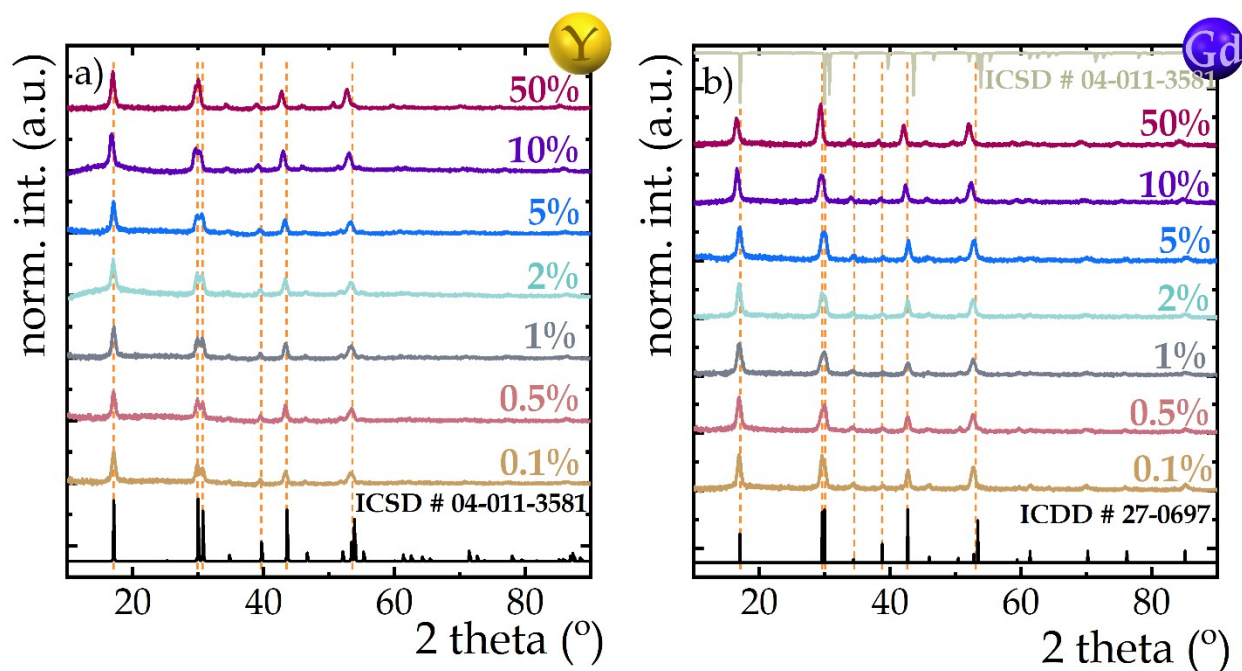


Figure S1. Shift of Bragg reflections towards lower angles for high concentrations of Nd³⁺ ions in NaYF₄ (a) and NaGdF₄ (b) host. At the top of figure b) a pattern for NaYF₄ is also shown,

allowing comparison of the shift of Bragg reflections in the NaGdF₄ matrix relative to those in the NaYF₄ counterpart.

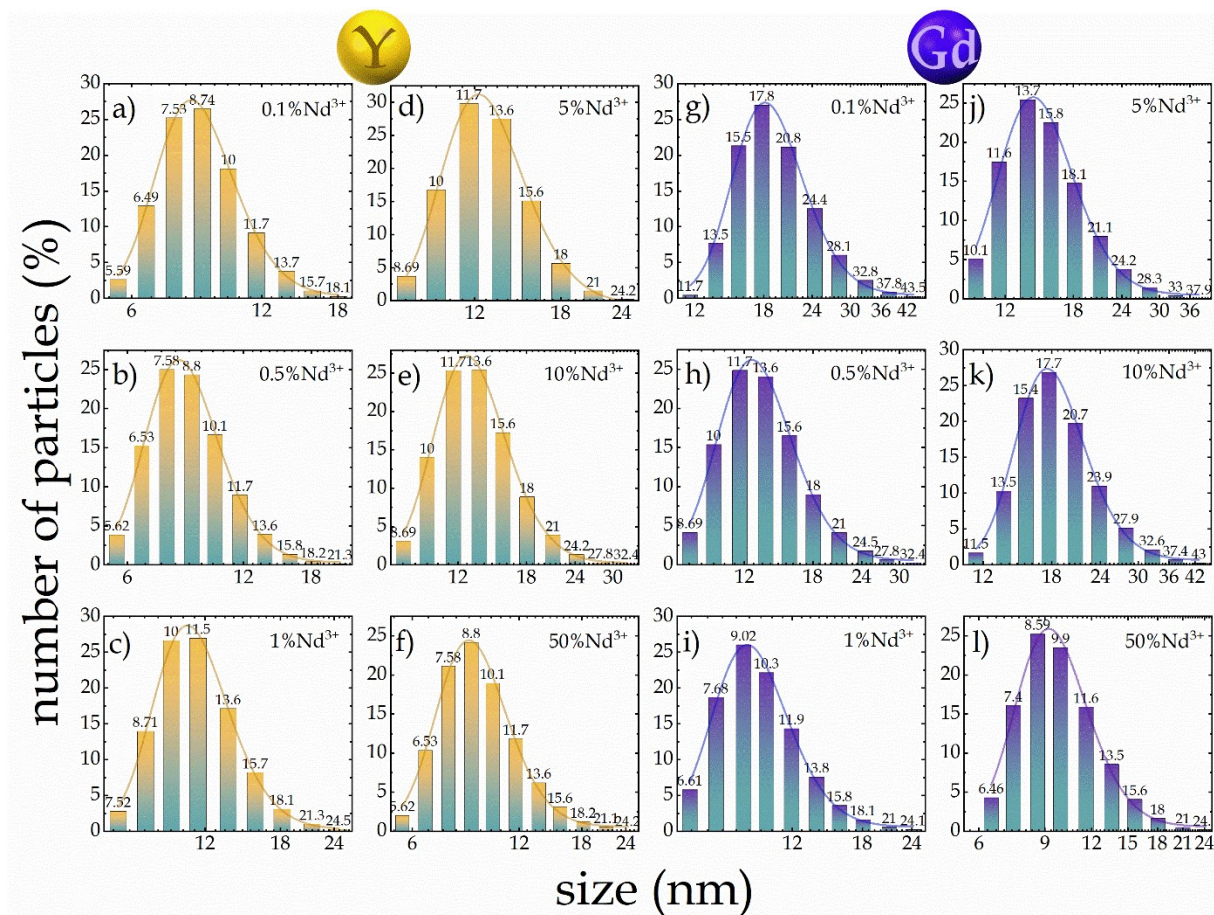


Figure S2. The hydrodynamic size distribution of representative NaYF₄ (a-f) and NaGdF₄: (g-l) doped with different Nd³⁺ concentration ranging from 0.1% to 50% determined from DLS measurement.

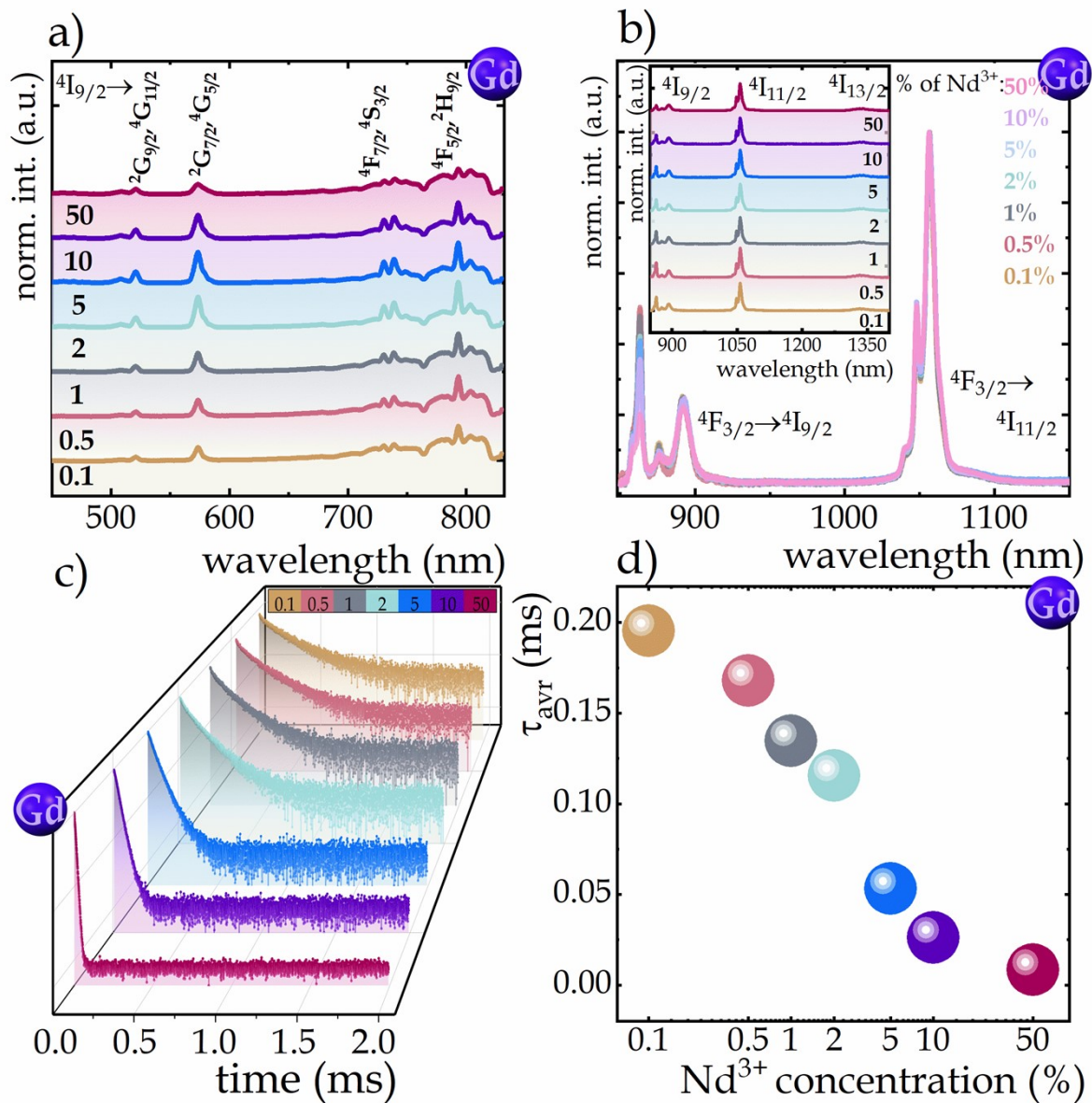


Figure S3. Nd³⁺ ions concentration impact on excitation spectra monitored at 1058 nm (a), emission spectra upon 793 nm excitation (b), the kinetics at 123 K of the excited $4F_{3/2}$ state (c) and average decay time of $4F_{3/2}$ level at 123K (d) in NaGdF₄ host.

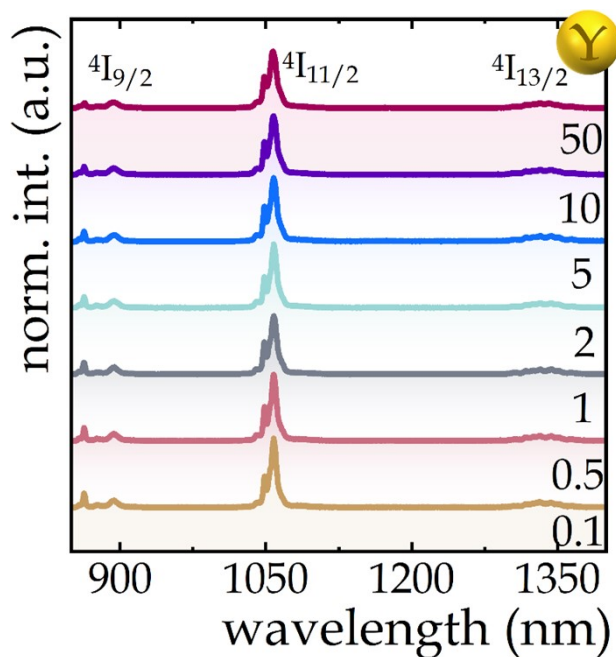


Figure S4. The comparison of the emission spectra upon 793 nm excitation for NaYF₄:Nd³⁺.

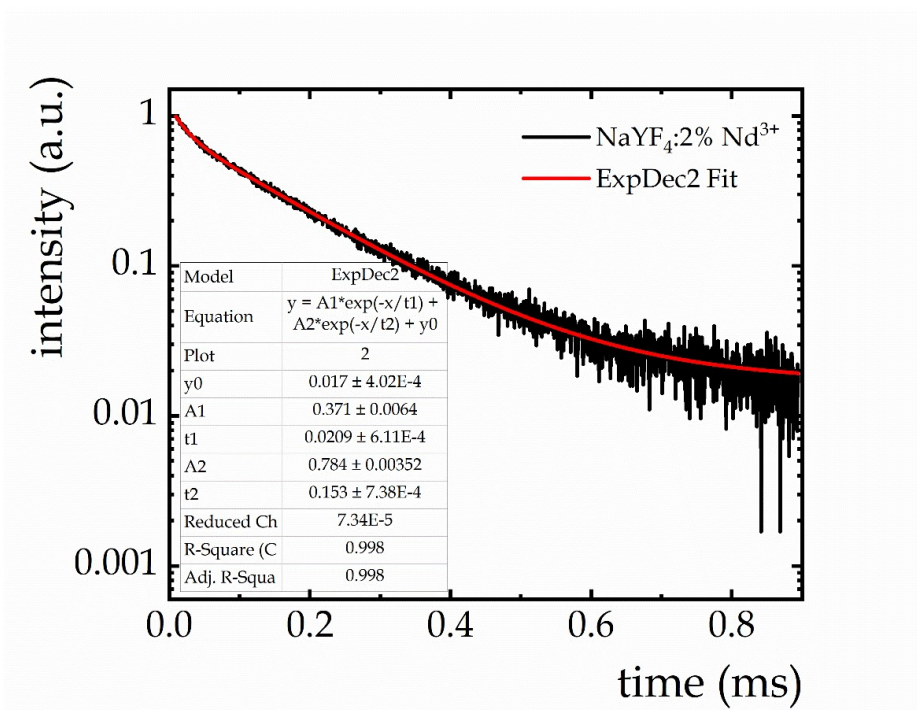


Figure S5. Decay curve and fit curve of the ExpDec2 function of an example NaYF₄:2%Nd³⁺ sample. The table shows the parameter values used to calculate the average time with the double-exponential function according to Equation 1.

Equation in the two-exponential ExpDec2 model:

$$y = A_1 \cdot e^{-x/t_1} + A_2 \cdot e^{-x/t_2} + y_0 \quad (\text{eq. S1})$$

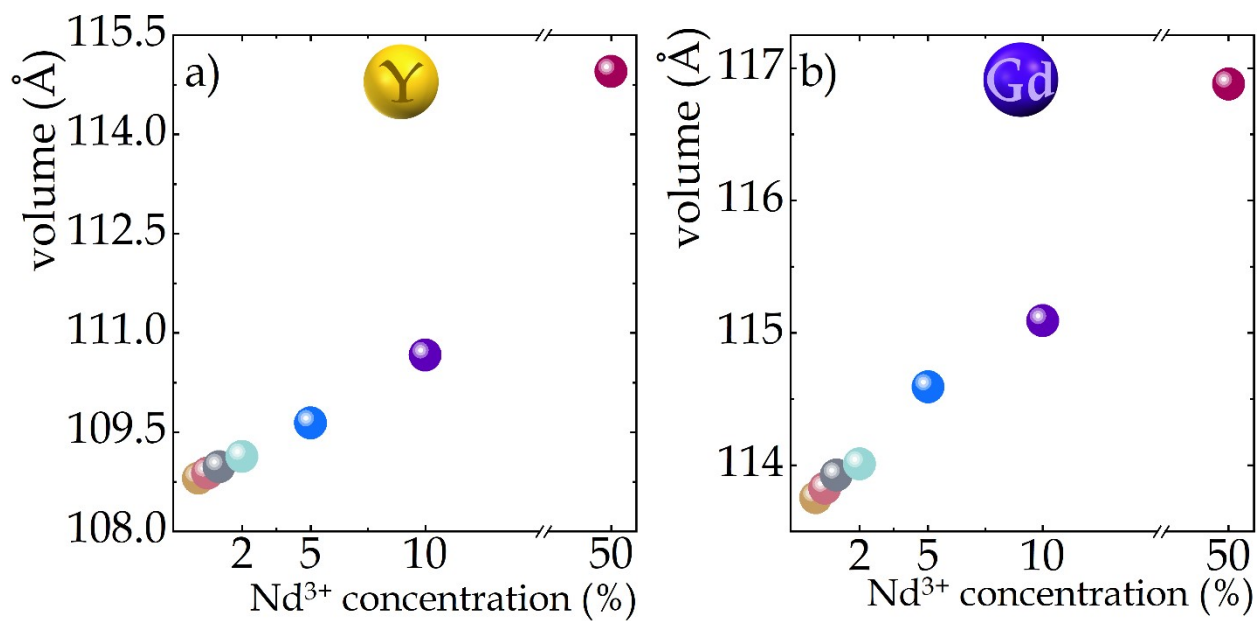


Figure S6. Dependence of unit cell volume on the concentration of Nd³⁺ ions in the host NaYF₄ (a) and NaGdF₄ (b)

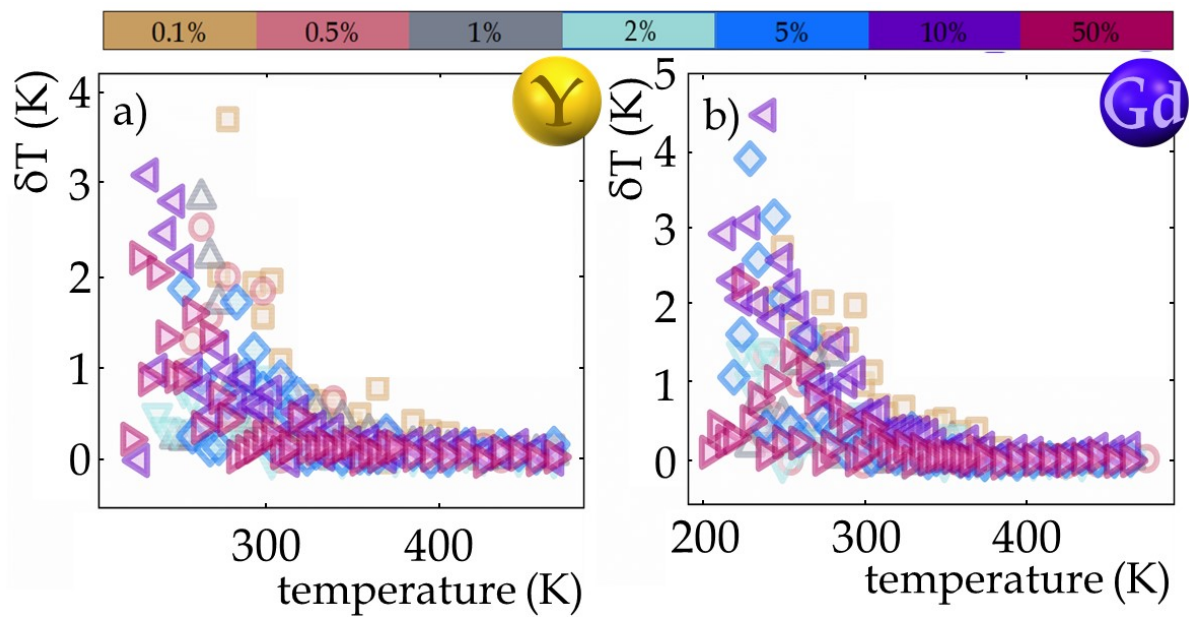


Figure S7. The temperature resolution of Nd³⁺-doped NaYF₄ and NaGdF₄ based LTs (a and b, respectively).