Supporting information

Hemin-Based Conjugated Effect Synthesis of Fe-N/NCNT Catalysts for Enhanced Oxygen Reduction

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Figure S1. The synthetic pathway to synthesize the network structure.



Figure S2. (a) SEM and (b) TEM images of polypyrrole nanotubes (PPy).



Figure S3. (a) SEM and (b, c) TEM images of A-PPy.



Figure S4. EDS spectrum of the Fe-NCNT-800.



Figure S5. The FT-IR spectrum of Fe-NCNT-X (700-100) catalysts.



Figure S6. N_2 adsorption-desorption isotherm of Fe-NCNT-X (700-1000) and the corresponding pore size distribution of the BJH method.



Figure S7. XPS survey spectra of Fe-NCNT-X (700-1000) catalysts.



Figure S8. High-resolution XPS spectroscopy of N 1s of (a) Fe-NCNT-700, (b) Fe-NCNT-900 and (c) Fe-NCNT-1000.



Figure S9. High-resolution XPS spectroscopy of Fe 2p of (a) Fe-NCNT-700, (b) Fe-NCNT-900 and (c) Fe-NCNT-1000.



Figure S10. Cyclic voltammetry (CV) curves of Fe-NCNT synthesized at different ratios (mass ratio).



Figure S11. CV curve of Pt/C in O_2 -saturated 0.1 M KOH at a scan rate of 50 mV s⁻¹.



Figure S12. (a, b) SEM images of NCNT-800.



Figure S13. (a, b) SEM images of C-Fe-NCNT-800.



Figure S14. LSV before and after a 5000-cycle ADT of ORR on Pt/C in O_2 -saturated

0.1 M KOH.



Figure S15. The kinetic current density (J_k) at 0.51 V and the number of electron transfer n value of Fe-NCNT-X (X=700-1000).



Figure S16. High-resolution XPS spectroscopy of Fe 2p electrons of NCNT-800.



Figure S17. High-resolution XPS spectra of Fe 2p electrons with different Ar plasma etching times for Fe-NCNT-800 samples.

Sample	BET surface area (m ² g ⁻¹)	Pore size (nm)	Pore volume (cm ³ g ⁻¹)
A-PPy	39.80	11.34	0.1129
Fe-NCNT-700	52.46	3.826	0. 1894
Fe-NCNT-800	104.2	3.825	0.2865
Fe-NCNT-900	48.66	3.832	0.2351
Fe-NCNT-1000	55.20	3.832	0.2451

Table S1. Summary of surface properties of Fe-NCNT-X (700-1000) catalystsobtained from BET.

Sample	Pyridinc N	Fe-N	Pyrrolic	Graphitc	Oxidized
	(%)	(%)	N(%)	N(%)	N(%)
Fe-NCNT-700	32.7	21.7	24.8	13.3	8.3
Fe-NCNT-800	28.0	28.5	27.4	10.2	5.9
Fe-NCNT-900	21.4	21.2	27.4	22.4	7.6
Fe-NCNT-1000	19.8	18.7	20.9	26.4	14.2

Table S2. The relative content of different N sites obtained by high resolution XPS scanning of N1s.

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Sample	Fe(0)(%)	Fe(II)(%)	Fe(III)(%)
Fe-NCNT-700		81.57	18.43
Fe-NCNT-800		87.06	12.94
Fe-NCNT-900		82.78	17.22
Fe-NCNT-1000	24.11	57.50	18.39

Table S3. The relative content of different Fe sites obtained by high resolution XPS scanning of Fe 2p.

Fe-NCNT- 800	Pyridinic N(%)	Fe-N _x (%)	Pyrrolic N(%)	Graphitic N(%)	Oxidized N(%)
0 s	28.0	28.5	27.4	10.2	5.9
60 s	25.8	25.4	30.4	11.3	7.1

Table S4. The depth XPS spectra of N 1s with different Ar plasma etching times for Fe-NCNT-800.

Sample	Eonset(V vs RHE)	$E_{1/2}$ (V vs RHE)	J(mA cm ⁻²)
Fe-NCNT-800(5:30)	0.89	0.66	-2.86
Fe-NCNT-800(15:30)	0.91	0.73	-4.02
Fe-NCNT-800(30:30)	0.93	0.79	-5.02
Fe-NCNT-800(50:30)	0.93	0.76	-4.36
Pt/C	1.06	0.82	-4.64

Table S5. Summary of LSV data for Fe-NCNT catalysts prepared in differentsynthesis ratios (mass ratio).

	E _{onset} (V vs.RHE)	E _{1/2} (V vs.RHE)	J (mA cm ⁻²)
Fe-NCNT-700	0.90	0.72	-3.85
Fe-NCNT-800	0.93	0.79	-5.02
Fe-NCNT-900	0.89	0.71	-3.82
Fe-NCNT-1000	0.81	0.62	-3.2
Pt/C	1.06	0.82	-4.64
NCNT-800	0.82	0.66	-3.53
C-Fe-NCNT-800	0.87	0.75	-2.49

Table S6. Summary of LSV data for Fe-NCNT-X (X=700-1000) catalyst, NCNT-800,C-Fe-NCNT-800 and Pt/C.

Times	C Atomic (%)	N Atomic (%)	O Atomic (%)	Fe Atomic (%)
0s	82.46	7.17	9.65	0.72
30s	87.97	6.58	4.57	0.88
60s	89.24	6.92	3.10	0.74
120s	89.14	6.72	2.90	1.23
180s	88.91	7.22	2.48	1.39
240s	90.25	6.41	2.29	1.05
300s	89.71	6.45	2.54	1.30
360s	89.16	7.43	2.15	1.26

Table S7. The element summary table of high resolution XPS for different times ofAr plasma etching of Fe-NCNT-800 sample.

Electrocatalysts	E _{1/2} (V vs.RHE)	E _{onset} (V vs.RHE)	Electrolyte	Reference
Fe-NCNT-800	0.79	0.93	0.1 M KOH	This work
p-Fe-N-CNFs	0.82	0.91	0.1 M KOH	[1]
CoP-CMP800	0.79	0.84	0.1 M KOH	[2]
rGO/(Ni ²⁺ /THPP/Co ²⁺ / THPP)	0.71	0.84	0.1 M KOH	[3]
FNCT800-100	0.82	0.93	0.1 M KOH	[4]
Fe-N-C/HPC-NH ₃	0.803	0.945	0.1 M KOH	[5]
Fe-N-C	0.72	0.83	$0.5 \mathrm{M} \mathrm{H}_2 \mathrm{SO}_4$	[6]
Fe-N-C/H ₂ O ₂	0.79	0.93	0.1MHClO ₄	[7]
Fe-N-C / MXene	0.84	0.92	0.1 M KOH	[8]
C60-SWCNT5	0.82	0.88	0.1 M KOH	[9]
H-Mn ₃ O ₄ -TMSLs	0.84	0.91	0.1 M KOH	[10]
HT-NCT	0.76	0.89	0.1 M KOH	[11]
Mo ₂ C/NPCNFs	0.77	0.90	0.1 M KOH	[12]
Co–N–GN	0.80	0.87	0.1 M KOH	[13]
TPA-TPE-2		0.82	0.1 M KOH	[14]
rGO/(Ni ²⁺ /THPP/Co ²⁺ / THPP)	0.71	0.84	0.1 M KOH	[15]

Table S8. Comparison of ORR performance of Fe-NCNT-800 and other porphyrin-derived electrocatalysts and hollow structures.

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