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Electronic supplementary information (ESI)

Novel synthesis and catalytic performance of hierarchical MOR

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Fig. S1²⁷ Al MAS NMR spectra of samples.

Catalysts	$\mathrm{Al}_{\mathrm{ef}}\left(\% ight)$	$\mathrm{Al}_{\mathrm{f}}\left(\% ight)$
M-0.02C-0.00S	38.1	61.9
M-0.02C-0.02S	30.6	69.4
M-0.02C-0.08S	22.9	77.1

Table S1 The content of framework and extra-framework Al.

Preparation of M-0.00C-0.00S, M-0.02C-0.00S and M-0.02C-0.08S samples crystallized for different time

In order to study the formation mechanism of hierarchical MOR, a series of samples were prepared under the same synthesis process of the M-0.02C-0.00S and M-0.02C-0.08S samples except for the crystallization time. After the crystallization, the products were recovered by centrifugation and washed with deionized water until the pH reached about 7. Then the samples were dried overnight at 120 °C and calcined at 550 °C in air for 6 h. The synthesized samples are denoted as M-0.02C-0.00S-zh and M-0.02C-0.08S-zh respectively, z represents the crystallization time (z=8, 16, 24, 32,40 and 48).

For comparison, a series of M-0.00C-0.00S-zh samples were prepared with the gel compositions of SiO₂: $0.0333Al_2O_3$: $0.3167Na_2O$: $27.5H_2O$ and the synthesis

process was the same as M-0.02C-0.08S-zh.



Fig. S2 XRD patterns of M-0.02C-0.08S-zh samples.



Fig. S3 SEM images of (a) M-0.02C-0.08S-8h; (b) M-0.02C-0.08S-16h; (c) M-0.02C-0.08S-24h;

⁽d) M-0.02C-0.08S-32h; (e) M-0.02C-0.08S-40h and (f) M-0.02C-0.08S-48h.



Fig. S4 XRD patterns of M-0.02C-0.00S-zh samples.



Fig. S5 SEM images of (a) M-0.02C-0.00S-8h; (b) M-0.02C-0.00S-16h; (c) M-0.02C-0.00S-24h;

(d) M-0.02C-0.00S-32h; (e) M-0.02C-0.00S-40h and (f) M-0.02C-0.00S-48h.



Fig. S6 XRD patterns of M-0.00C-0.00S-zh samples.



Fig. S7 SEM images of (a) M-0.00C-0.00S-8h; (b) M-0.00C-0.00S-16h; (c) M-0.00C-0.00S-24h;

(d) M-0.00C-0.00S-32h; (e) M-0.00C-0.00S-40h and (f) M-0.00C-0.00S-48h.



Fig. S8 XRD patterns of M-0.02C-0.12S, M-0.025C-0.12S and M-0.03C-0.12S samples



Fig. S9 (a) N₂ adsorption-desportion isotherms of C-MOR and (b) NH₃-TPD profiles of C-MOR

Table 32 Textulal properties of C-INC

Catalysta	S:0 /41 0	Surface area (m ² g ⁻¹)			Pore volume (cm ³ g ⁻¹)		
Catalysis	SIO_2/AI_2O_3	$\mathbf{S}_{\mathrm{BET}}$	Sext	$\mathbf{S}_{\mathrm{mico}}$	$\mathbf{V}_{\text{total}}$	V _{micro}	V _{meso}
C-MOR	15	521	99	422	0.220	0.196	0.024

Table S3 Acid amounts determined by NH ₃ -TPD of C-MOR.							
Catalysts	Total acidity (mmol g ⁻¹)	Weak acidity (mmol g ⁻¹)	Strong acidity (mmol g ⁻¹)				
C-MOR	1.093	0.912	0.181				