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Supporting Information

Interface engineering in core-shell Co₉S₈@MoS₂ nanocrystals induces enhanced

hydrogen evolution in acidic and alkaline media

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Figure S1. SEM images of PAN/CoMo precursor nanofibers of $F-Co_9S_8@MoS_2/CNFs$ (a) and $S-Co_9S_8@MoS_2/CNFs$ (b). (c) XRD patterns of PAN/CoMo precursor nanofibers of F-Co_9S_8@MoS_2/CNFs and S-Co_9S_8@MoS_2/CNFs.



Figure S2. Electrochemical cyclic voltammograms curves of $S-Co_9S_8@MoS_2/CNFs$ and $F-Co_9S_8@MoS_2/CNFs$ in 0.5 M H₂SO₄ (a, b) and 1.0 M KOH (c, d) at scan rates of 10, 20, 30, 40, and 50 mV s⁻¹.



Figure S3. Calculation of C_{dl} by plotting Capacitive currents (Δj) against scan rates in 0.5 M H₂SO₄ (a) or 1.0 M KOH (b).



Figure S4. Polarization curves of $F-Co_9S_8@MoS_2/CNFs$ and $S-Co_9S_8@MoS_2/CNFs$ normalized by ECSA in 0.5 M H₂SO₄ (a) or 1.0 M KOH (b).

Table S1. Comparison of sulfide-based electrocatalysts.

Catalysts	Electrolyte	η_{10} (mA cm ⁻²)	η_{100} (mA cm ⁻²)	Tafel Slope (mV dec ⁻¹)	Refence
Co ₉ S ₈ /MoS _x nanotubes	0.5 M H ₂ SO ₄	161	/	78	[1]
Co ₉ S ₈ @MoS ₂ HNBs	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	106	/	51.8	[2]
$Co_9S_8@MoS_2$ hybrids	$0.5 \text{ M} \text{H}_2\text{SO}_4$	171	/	123	[3]
	1.0 M KOH	143	/	117	
Co ₉ S ₈ -30@MoS _x /CC	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	98	165	64.8	[4]
MoS ₂ /Ni ₃ S ₂	1.0 M KOH	110	/	83.1	[5]
CoMoS-2-C	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	135	/	50	[6]
Hollow CoS _x @MoS ₂ microcubes	$0.5 \text{ M} \text{H}_2\text{SO}_4$	239	/	103	[7]
Co ₉ S ₈ /NC@MoS ₂	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	117	/	68.8	
	1.0 M KOH	67	/	60.3	[8]
	1.0 M PBS	261	/	126.1	
Co ₉ S ₈ -MoS ₂ @3DC	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	230	/	111.7	
	1.0 M KOH	177	/	83.6	[9]
	1.0 M PBS	474	/	172	
Co ₉ S ₈ /1L MoS ₂	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	97	/	71	[10]
Co ₉ S ₈ /CNFs	$0.5 \text{ M} \text{H}_2\text{SO}_4$	165	/	83	[11]
Pd ₁₆ S ₇ /MoS ₂ /CNFs	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	83	/	113	[12]
Co ₉ S ₈ @MoS ₂ /CNFs	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	190	/	110	[13]
Co ₉ S ₈ HMs-140/C	0.1 M KOH	250	/	98	[14]
Co ₉ S ₈ /HWS ₂ /CNFs	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	83	235	56	[15]
	1.0 M KOH	87	375	72	
S-Co ₉ S ₈ @ MoS ₂ /CNFs	$0.5 \text{ M} \text{ H}_2\text{SO}_4$	77	236	83	This
	1.0 M KOH	122	322	66	work



Figure S5. The Nyquist plots of $F-Co_9S_8@MoS_2/CNFs$ and $S-Co_9S_8@MoS_2/CNFs$ in 0.5 M H₂SO₄ (a) and 1.0 M KOH (b) at $\eta = 10$ mV.

Table S2. R_{ct} values of F-Co₉S₈@MoS₂/CNFs and S-Co₉S₈@MoS₂/CNFs in 0.5 M H₂SO₄ and 1.0 M KOH at η = 10 mV.

R _{ct}	F-Co ₉ S ₈ @MoS ₂ /CNFs	S-Co ₉ S ₈ @MoS ₂ /CNFs
0.5 M H ₂ SO ₄	2.36	2.11
1.0 M KOH	3.83	2.26



Figure S6. XRD patterns of the S-Co $_9S_8@MoS_2/CNFs$ prepared at different temperatures.



Figure S7. HER LSV curves of S-Co₉S₈@MoS₂/CNFs prepared at different temperatures obtained in (a) 0.5 M H_2SO_4 and (b) 1.0 M KOH.

References

[1] L. Wu, K. Zhang, T. Wang, X. Xu, Y. Zhao, Y. Sun, W. Zhong, Y. Du, Cobalt sulfide nanotubes (Co₉S₈) decorated with amorphous MoS_x as highly efficient hydrogen evolution electrocatalyst, ACS Appl. Nano Mater. 1 (2018) 1083–1093. https://doi.org/10.1021/acsanm.7b00271
[2] V. Ganesan, S. Lim, J. Kim, Hierarchical nanoboxes composed of Co₉S₈@MoS₂ nanosheets as efficient electrocatalysts for the hydrogen evolution reaction, Chem. Asian J. 13 (2018) 413-420. https://doi.org/10.1002/asia.201701536

[3] J. Bai, T. Meng, D. Guo, S. Wang, B. Mao, M. Cao, Co₉S₈@MoS₂ core-shell heterostructures as trifunctional electrocatalysts for overall water splitting and Zn−air batteries, ACS Appl. Mater. Interfaces 10 (2018) 1678–1689. https://doi.org/10.1021/acsami.7b14997

[4] X. Zhou, X. Yang, M. Hedhili, H. Li, S. Min, J. Ming, K. Huang, W. Zhang, L. Li, Symmetrical synergy of hybrid Co₉S₈-MoS_x electrocatalysts for hydrogen evolution reaction, Nano Energy 32 (2017) 470-478. https://doi.org/10.1016/j.nanoen.2017.01.011

[5] J. Zhang, T. Wang, D. Pohl, B. Rellinghaus, R. Dong, S. Liu, X.D. Zhuang, X.L. Feng, Interface engineering of MoS_2/Ni_3S_2 heterostructures for highly enhanced electrochemical overall-watersplitting activity, Angew. Chem. 55 (2016) 6702-6707. https://doi.org/10.1002/anie.201602237 [6] X. Dai, K. Du, Z. Li, M. Liu, Y. Ma, H. Sun, X. Zhang, Y. Yang, Co-doped MoS_2 nanosheets with the dominant CoMoS phase coated on carbon as an excellent electrocatalyst for hydrogen evolution, ACS Appl. Mater. Interfaces 7 (2015) 27242–27253.

https://doi.org/10.1021/acsami.5b08420

[7] L. Yang, L. Zhang, G. Xu, X. Ma, W. Wang, H. Song, D. Jia, Metal–organic-framework-derived hollow CoS_x@MoS₂ microcubes as superior bifunctional electrocatalysts for hydrogen evolution and oxygen evolution reactions, ACS Sustainable Chem. Eng. 6 (2018) 12961–12968. https://doi.org/10.1021/acssuschemeng.8b02428

[8] H. Li, X. Qian, C. Xu, S. Huang, C. Zhu, X. Jiang, L. Shao, L. Hou, Hierarchical porous Co₉S₈/nitrogen-doped carbon@MoS₂ polyhedrons as pH universal electrocatalysts for highly efficient hydrogen evolution reaction, ACS Appl. Mater. Interfaces 9 (2017) 28394–28405. https://doi.org/10.1021/acsami.7b06384

[9] L. Diao, B. Zhang, Q. Sun, N. Wang, N. Zhao, C. Shi, E. Liu, C. He, An in-plane $Co_9S_8@MoS_2$ heterostructure for the hydrogen evolution reaction in alkaline media, Nanoscale 11 (2019) 21479-21486. https://doi.org/10.1039/C9NR06609H

[10] H. Zhu, G. Gao, M. Du, J. Zhou, K. Wang, W. Wu, X. Chen, Y. Li, P. Ma, W. Dong, F. Duan, M. Chen, G. Wu, J. Wu, H. Yang, S. Guo, Atomic-scale core/shell structure engineering induces precise tensile strain to boost hydrogen evolution catalysis, Adv. Mater. 30 (2018) 1707301. https://doi.org/10.1002/adma.201707301

[11] L. Gu, H. Zhu, D. Yu, S. Zhang, J. Chen, J. Wang, M. Wan, M. Zhang, M. Du, A facile strategy to synthesize cobalt-based self-supported material for electrocatalytic water splitting, Part. Part. Syst. Charact. 34 (2017) 1700189. https://doi.org/10.1002/ppsc.201700189

[12] Y. Wen, H. Zhu, L. Zhang, J. Hao, C. Wang, S. Zhang, S. Lu, M. Zhang, M. Du, Beyond colloidal synthesis: Nanofiber reactor to design self–supported core–shell Pd₁₆S₇/MoS₂/CNFs electrode for efficient and durable hydrogen evolution catalysis, ACS Appl. Energy Mater. 2 (2019) 2013–2021. https://doi.org/10.1021/acsaem.8b02105

[13] H. Zhu, J. Zhang, R. Yanzhang, M. Du, Q. Wang, G. Gao, J. Wu, G. Wu, M. Zhang, B. Liu, J. Yao,
X. Zhang, When cubic cobalt sulfide meets layered molybdenum disulfide: A core–shell system toward synergetic electrocatalytic water splitting, Adv. Mater. 27 (2015) 4752-4759. https://doi.org/10.1002/adma.201501969

[14] Y. Zhang, S. Chao, X. Wang, H. Han, Z. Bai, L. Yang, Hierarchical Co₉S₈ hollow microspheres as multifunctional electrocatalysts for oxygen reduction, oxygen evolution and hydrogen evolution reactions, Electrochim. Acta 246 (2017) 380-390.

https://doi.org/10.1016/j.electacta.2017.06.058

[15] S. Zhang, Y. Li, H. Zhu, S. Lu, P. Ma, W. Dong, F. Duan, M. Chen, M. Du, Understanding the role of nanoscale heterointerfaces in core/shell structures for water splitting: Covalent bonding interaction boosts the activity of binary transition-metal sulfides, ACS Appl. Mater. Interfaces 12 (2020) 6250–6261. https://doi.org/10.1021/acsami.9b19382