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# **Supporting information**

## Disposable electrode modified with metal orthovanadate and sulfur-reduced

### graphene oxide for electrochemical detection of sulfasalazine

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### **EXPERIMENTAL SECTION**

#### **Reagents and Chemicals**

Cerium (III) nitrate hexahydrate (Ce(NO<sub>3</sub>)<sub>3</sub>·6H<sub>2</sub>O, ACS reagent,  $\geq$ 99.0%), NH<sub>4</sub>VO<sub>3</sub>, and other chemicals were bought from Sigma-Aldrich, USA. Potassium ferricyanide (K<sub>3</sub>[Fe(CN)<sub>6</sub>]), potassium ferrocyanide (K<sub>4</sub>[Fe(CN)<sub>6</sub>]), and potassium chloride (KCl) were analar grade commercial products, used as received. For those experiments in which supporting electrolyte was present, KCl was added to remain a 5.0 mM concentration. The electrolyte, 0.05 M phosphate buffer saline (PBS, pH 7.0) was prepared by mixing, sodium mono-hydrogen phosphate (Na<sub>2</sub>HPO<sub>4</sub>) and sodium dihydrogen phosphate (NaH<sub>2</sub>PO<sub>4</sub>) in an aqueous medium. The pH of the electrolyte solution was adjusted by using aqueous H<sub>2</sub>SO<sub>4</sub> (0.5 M) and NaOH (2.0 M). The entire experiments were done with freshly prepared Milli-Q water (18.2 MQ cm<sup>-1</sup>, specific resistivity) collected from Millipore Water System.

### Characterizations

The nanostructure, composition and the morphological characterization of CVO@SRG composite was evaluated by the high resolution-transmission electron microscopy (HR-TEM) H-7600, Hitachi, (Japan). The chemical composition and elemental percentage were assessed through energy-dispersive X-ray (EDX) which affiliated with HR-TEM. The crystalline phase of as prepared samples was analyzed by X-ray diffraction (XRD) XPERT-PRO (PAN analytical B.V. The Netherlands) diffractometer with Cu M radiation (k = 1.54 Å). The oxidation state and the composition of the materials was scrutinized by X-ray photoelectron spectroscopy (XPS, Thermo ESCALAB 250 instrument).

#### **Electrochemical Apparatus**

The cyclic voltammetry (CV) and differential pulse voltammetry (DPV) experiments were carried out on a CHI 1205A analyzer electrochemical workstation (CH, USA). A conventional three-electrode system consisting of an Ag/AgCl (saturated KCl) electrode as the reference electrode, a platinum wire as the auxiliary electrode, and a CVO@SRG modified SPCE as the WE were used (Scheme 2).

#### SGR preparation

The graphene oxide (GO) was prepared by modified Hummers method from the graphite powder. Then, 10 mg of as-prepared GO was re-dispersed in 10 mL of de-ionized water and this suspension was ultrasonicated for 40 min. Further, the 5 mL of 1 M Na<sub>2</sub>S was added to the suspension. Then, the reaction mixture was ultrasonicated for 2 h at 100 °C. Now the graphene oxide turned to S doped reduced graphene oxide sheets (SRG) and the SRG was separated by centrifugation.

#### **CVO crystal structure information**

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Title  $Ce O_4 V$ 

Lattice type I Space group name I 41/a m d Space group number 141 Setting number 1

Lattice parameters

a b c alpha beta gamma 7.34000 7.34000 6.47000 90.0000 90.0000 90.0000 Unit-cell volume = 348.575135 Å^3

Structure parameters

	Х	у	Z	Occ.	U	Site	Sym.
10 01	0.00000	0.19000	0.35000	1.000	0.020	16h	.m.
2 Ce Ce1	0.00000	0.00000	0.00000	1.000	0.020	4a	-4m2
3 V V1	0.00000	0.00000	0.50000	1.000	0.020	4b	-4m2

Number of polygons and unique vertices on isosurface = 0(0) 107 atoms, 144 bonds, 23 polyhedra; CPU time = 3 ms



Fig. S1. Optimization of the electrochemical experiments (A): effect of concentrations of nanocomposite, (B): effect of catalyst.



Fig. S2. (A-B) Optimization of the electrochemical experiments effect of pH.

Table 1. Analytical parameters for the determination of SSZ at CVO/SRG/SPCE with previous reports [14,17,19,32-36].

SbFE	Antinomy film electrode
BiNP-CNT	Bismuth nanoparticle carbon nanotube
	-
MWCNTCOOH	Functionalized multiwalled carbon nanotube
GO/Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub>	Graphene oxide/magneticle nanoparticle at
	silicon dioxide
CD/RGO	Beta cyclodextrin –reduced graphene oxide
	composite
NiO	Nickel oxide nanoparticle

NiO/CNT	Nickel oxide-carbon nanotube nano		
	composite		
PtNPs/SWCNT	Platinum nanoparticles on single walled		
	carbon nanotube		