

Supporting Information

Highly conductive triple-layered hollow MnO₂@SnO₂@NHCS nanospheres with excellent lithium storage for high performance lithium-ion batteries

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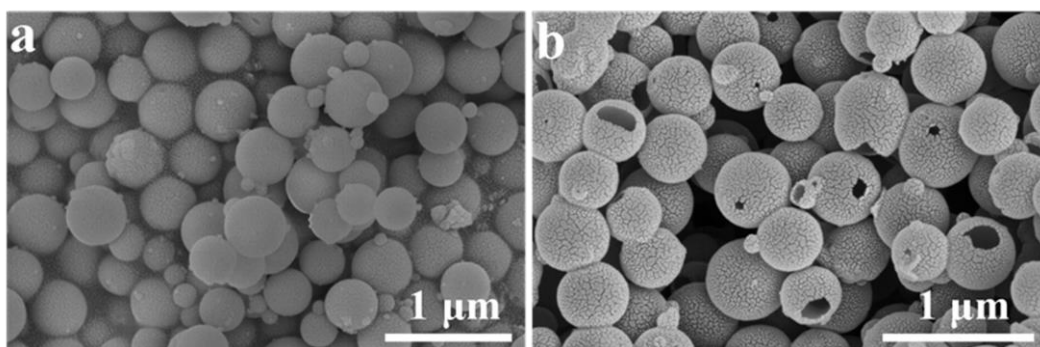


Fig. S1 FESEM images of (a) SiO₂ templates and (b) NHCS nanospheres.

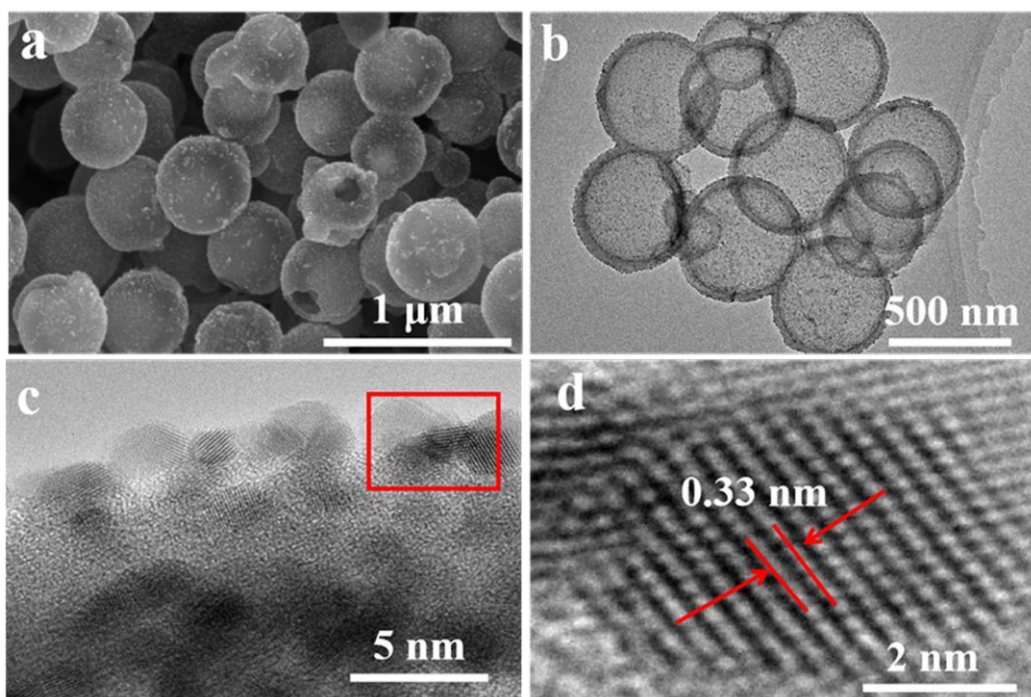


Fig. S2 FESEM(a), TEM(b) and HRTEM(c, d) images of SnO₂@NHCS.

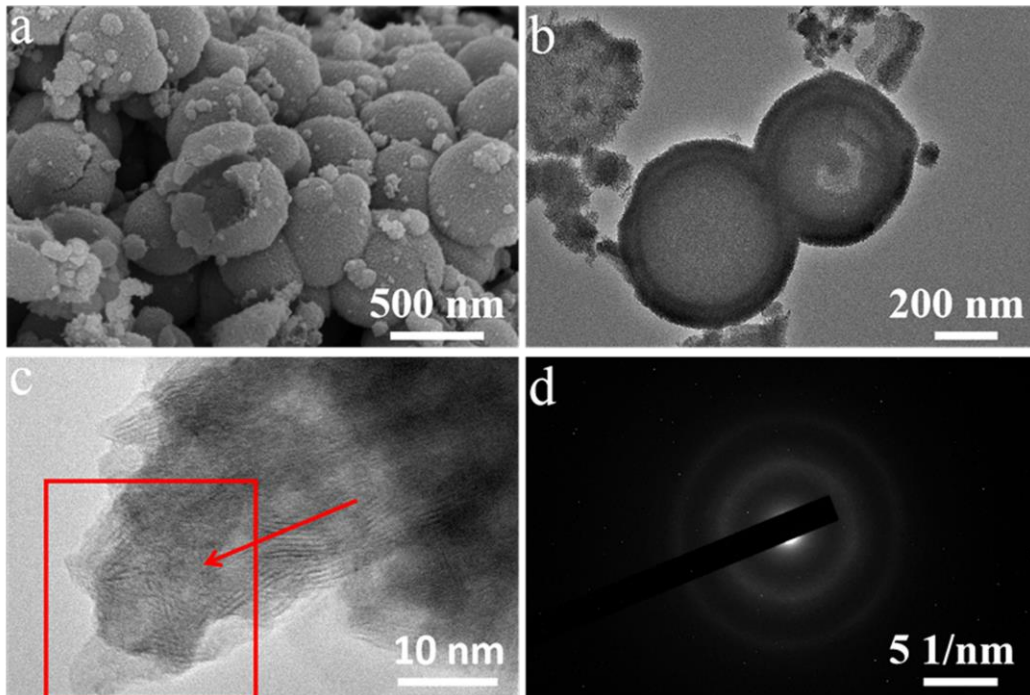


Fig. S3 FESEM(a), TEM(b), HRTEM(c) and SAED(d) images of $\text{MnO}_2@NHCS$.

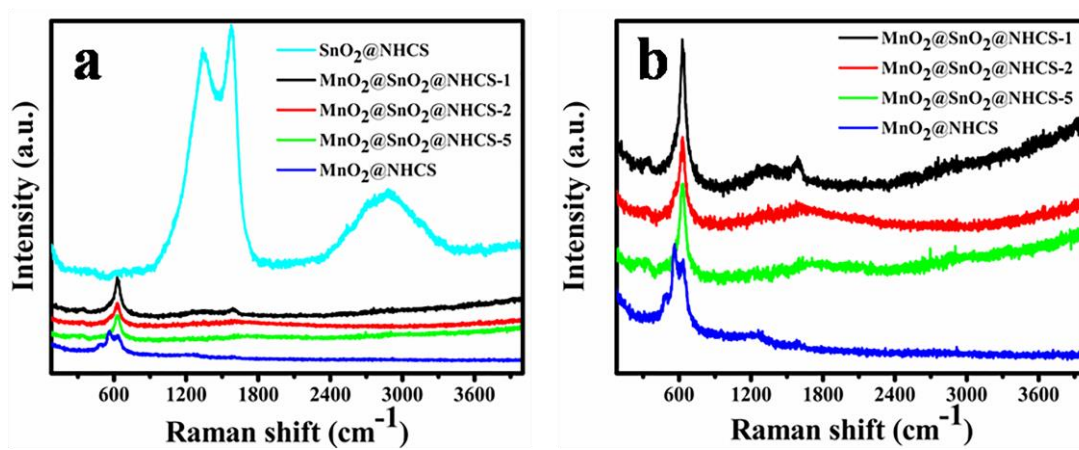


Fig. S4 Raman Spectra of $\text{SnO}_2@NHCS$, $\text{MnO}_2@SnO_2@NHCS-1$, $\text{MnO}_2@SnO_2@NHCS-2$, $\text{MnO}_2@SnO_2@NHCS-5$, and $\text{MnO}_2@NHCS$

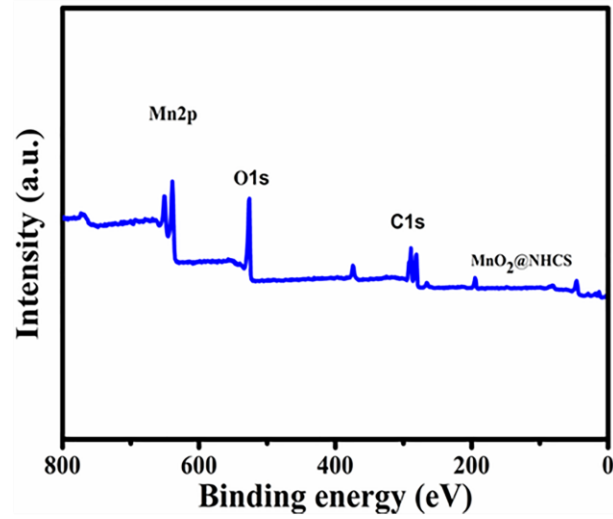


Fig. S5 XPS spectra of MnO₂@NHCS.

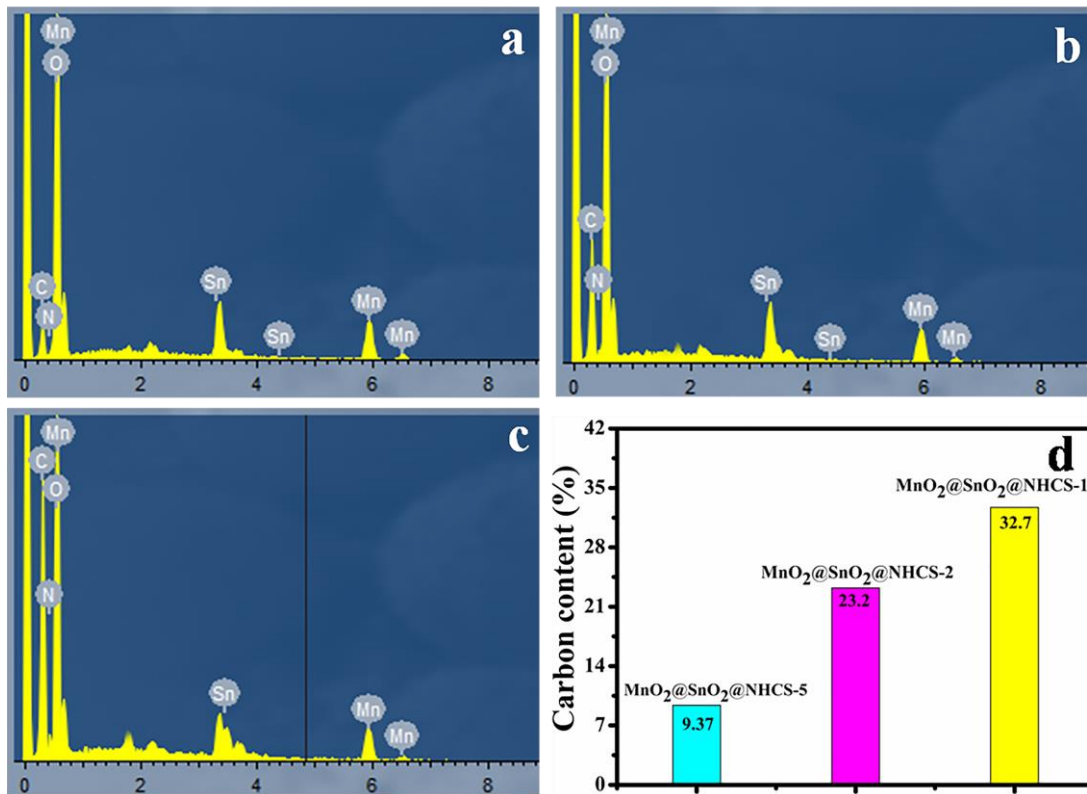


Fig. S6 EDX point spectra of (a) MnO₂@SnO₂@NHCS-5, (b) MnO₂@SnO₂@NHCS-2, (c) MnO₂@SnO₂@NHCS-1 and (d) carbon content of MnO₂@SnO₂@NHCS-5, MnO₂@SnO₂@NHCS-2, MnO₂@SnO₂@NHCS-1

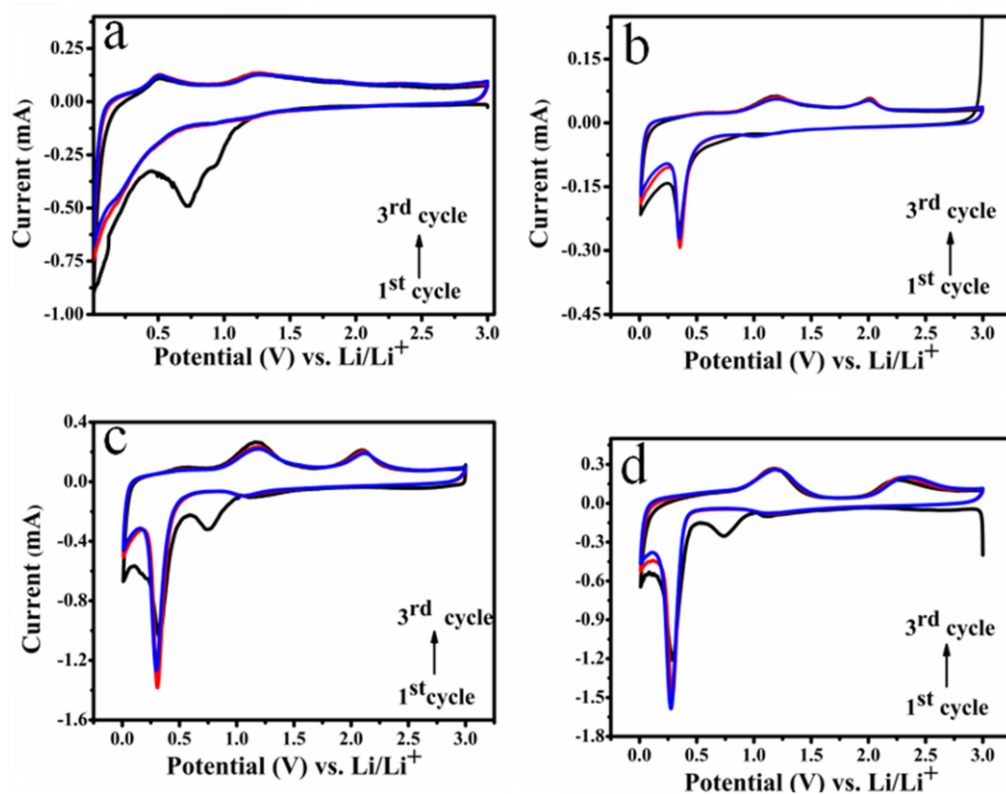


Fig. S7 CV curves of the (a) SnO₂@NHCS, (b) MnO₂@SnO₂@NHCS-1, (c) MnO₂@SnO₂@NHCS-2, (d) MnO₂@NHCS.

Table S1. Comparison of the electrochemical properties of the prepared MnO₂@SnO₂@NHCS-5 with previously reported anode materials for LIBs.

Materials	Current density (mA _g ⁻¹)	Cycle numbers	Capacity (mAh _g ⁻¹)	References
MnO₂@SnO₂@NHCS-5	100	100	1053.8	This work
SnO ₂ @C@VO ₂ CHNS	100	100	765.1	1
α-Fe ₂ O ₃ /MnO ₂	100	150	860	2
CF@MnO ₂	100	150	648	3
Fe ₂ O ₃ /Co ₃ O ₄	100	50	500	4
SnO ₂ -C	100	30	492.5	5
δ-MnO ₂	1000	100	320	6
C/MnO	100	100	943.6	7
N-MnO/GNS	100	90	772	8

References

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