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Supporting information

1,8-naphthalimide appended propiolate-based fluorescent sensor for selective detection of cysteine over glutathione and homocysteine in living cells

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Experimental Section

Synthesis of 4–Bromo–N–butyl–1,8–naphthalimide (2a). Compound **2a** (0.5 g, 1.80 mmol) and n–butylamine (0.69 g, 9.55 mmol) was refluxed for 6h in 20 mL glacial acetic acid. After cooling, the suspension was put into ice–water (250 mL), and then filtered to obtain a yellow solid. The precipitate was washed with excess H_2O and then re–crystallized from acetic acid gave a pale yellow crystal (yield: 79.3%).

¹H–NMR (600 MHz, CDCl₃): δ 8.61 (d, J = 7.2 Hz, 1H), 8.50(d, J = 8.4 Hz, 1H), 8.36 (d, J = 7.8 Hz, 1H), 7.99 (d, J = 7.8 Hz, 1H), 7.80 (t, J = 7.6 Hz, 1H), 4.17 (t, J = 7.6 Hz, 2H,), 1.76–1.68 (m, 2H), 1.44 (dt, 7.4 J = 2H), 0.97 (t, J = 7.4 Hz, 3H). ¹³C–NMR (150 MHz, CDCl₃): δ 163.66, 133.23, 132.02, 131.22, 131.09, 130.64, 130.20, 129.02, 128.08, 123.16, 122.30, 40.38, 30.16, 20.37, 13.84.

Synthesis of 4-hydrazine hydrate N-butyl-1,8-naphtic anhydride (3a). A mixture of compound 2a (8.0 g, 24.08 mmol) and hydrazine hydrate (7.62 g, 85%, v/v) in ethylene glycol mono methyl ether (93 mL) was refluxed for 24 h. After cooling, the mixture was put into 180 mL of ice-H₂O, the precipitate was filtered and dried in vacuum to obtain a red precipitate (yield: 80.3%).

¹H–NMR (600 MHz, CDCl₃) δ 9.08 (s, 1H), 8.57 (d, J = 8.3 Hz, 1H), 8.37 (d, J = 7.2 Hz, 1H), 8.25 (d, J = 8.5 Hz, 1H), 7.59 (t, J = 7.8 Hz, 1H), 7.21 (d, J = 8.6 Hz, 1H), 4.64 (s, 2H), 3.98 (t, J = 7.3 Hz, 2H), 1.58–1.52 (m, 2H), 1.34–1.27 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H). ¹³C–NMR (150 MHz, CDCl₃) δ 164.17, 163.32, 153.60, 134.61, 130.97, 129.70, 128.64, 124.52, 122.15, 118.84, 107.78, 104.41, 39.46, 30.28, 20.30, 14.20.

Synthesis of 2–formylphenyl propiolate (4a). The propiolate derivative **4a** was prepared by the DCC/DMAP esterification reaction of salicyl aldehyde with propiolic acid. Briefly, to an

ice cooled and stirred solution of propiolic acid (315.4 mg, 4.50 mmol) and salicyl aldehyde (500 mg, 4.09 mmol) in DCM (dry) was added dropwise to the solution of dimethylaminopyridine (DMAP, 3.75 mg, 0.03 mmol) and N,N'-dicyclohexylcarbodiimide (DCC, 929 mg, 4.05 mmol) in DCM (dry, 12 mL) during 1 h under N₂ atmosphere. Afterward, the mixture was stirred and kept at rt for 20 h, and then filtered to get rid of the insoluble byproduct N,N'-dicyclohexylurea. The combined filtrate was washed with 1.0 N HCl followed by washing with brine and dried utilizing anhydrous Na₂SO₄. The solvent was get rid of to give the crude product that was then purified by CC utilizing EtOAc: hexane (1:9, v/v) to give **4a** (90 %).

¹H NMR (600 MHz, CDCl₃) δ 9.90 (s, 1H), 7.58–7.51 (m, 2H), 7.05–6.97 (m, 2H), 3.03 (s, 1H). ¹³C–NMR (150 MHz, CDCl₃) δ 195.80, 160.56, 155.38, 136.15, 132.81, 119.56, 118.91, 116.60, 76.06, 72.96.

Preparation of 2-formylphenyl propiolate appended–N-butyl–1,8-naphthalimide (NASP).

An ethanolic solution of **4a** (0.23 g, 1.32 mmol) was carefully transferred to **3a** solution (0.37 g, 1.32 mmol) in EtOH (anhydrous, 3 mL). Acetic acid (2–3 drops) was transferred to the mixture, which was then refluxed at 80 °C for 24h. The orange solid formed was filtered and washed several times with cold EtOH. The product was dried under vacuum to give the probe **NASP** (71 %). Found (m/z): 437.810 and calculated (m/z): 439.15 for [**NASP**].

¹H NMR (600 MHz, DMSO– d_6) δ 11.45 (s, 1H), 8.79 (s, 1H), 8.78 (s, 1H), 8.45 (d, J = 7.2 Hz, 1H), 8.35 (d, J = 8.4 Hz, 1H), 7.83 (d, J = 7.7 Hz, 1H), 7.75 (t, J = 7.8 Hz, 1H), 7.61 (d, J = 8.4 Hz, 1H), 7.24 (t, J = 7.6 Hz, 1H), 6.90 (dd, J = 16.6, 8.1 Hz, 2H), 4.35 (s, 1H), 4.00 (t, J = 7.4 Hz, 2H), 1.61–1.54 (m, 2H), 1.36–1.29 (m, 2H), 0.90 (t, J = 7.3 Hz, 3H). ¹³C NMR (150 MHz, DMSO– d_6) δ 164.08, 163.37, 156.68, 154.02, 146.76, 142.31, 134.03, 131.30, 129.59,

128.72, 126.77, 125.40, 122.40, 120.98, 119.98, 119.01, 116.58, 111.16, 106.87, 77.76, 76.54, 30.23, 20.29, 14.20. Found (m/z): 437.810 and calculated (m/z): 439.15 for [NASP].



Fig S1. ¹H–NMR spectrum of the compound 2a in CDCl₃- d_6



Fig S2. ¹³C–NMR spectrum of the compound **2a** in CDCl₃- d_6



Fig S3. ¹H–NMR spectrum of the compound **3a** in $CDCl_3$ - d_6



Fig S4. ¹³C–NMR spectrum of the compound **3a** in CDCl₃- d_6



Fig S5. ¹H–NMR spectrum of the compound **4a** in CDCl₃- d_6



Fig S6. ¹³C–NMR spectrum of the compound **4a** in CDCl₃- d_6



Fig S7. ¹H–NMR spectrum of the **NASP** in DMSO- d_6



Fig S8. 13 C–NMR spectrum of the **NASP** in DMSO- d_6



Fig S9.MALDI-TOF MS spectra of the NASP



Fig S10. The fluorescence emission spectra of the NASP, NASP-Cys and NASP-Hcy (a) in different solvents and (b) EtOH-water mixture study in different ratios for the NASP in the presence and absence of Cys or Hcy



Fig S11. Spectra of the **NASP** (5.0 μM) in the presence of Hcy (0.0–70.0 equiv) in EtOH:H₂O (90:10, v:v, 0.0670 M PBS buffer pH=7.0) media, LOD plot of the **NASP** versus Hcyconcentrations.



Fig S12. Pareto's diagram of the proposed method for Cys and Hcy



Fig S13. The emission intensity of the chemoprobe NASP (5.0 μ M) with competitive amino acids (*Asn, Trp, Phe, Ser, Thr, Met, Pro, Gly, Tyr, Val, Ala, Lys, Arg, His, Ile, GSH*)(**70 equiv., 0.350 mM**) (blue bar) and the emission intensity of NASP (5.0 μ M) in the presence of and **70 equiv. other competitive analytes** and **70 equiv. Hcy** (**0.350 mM**) in EtOH:H₂O (90:10, v:v, 0.0670 M PBS buffer pH=7.0) media



Fig S14. Response time experiments of the chemoprobe NASP (5.0 μ M) in the presence of Cys, and Hcy in EtOH:H₂O (90:10, v:v, 0.0670 M PBS buffer pH=7.0) media



Fig S15. pH study of the **NASP** with or without Cys/Hcy in EtOH:H₂O (90:10, v:v, 0.0670 M PBS buffer pH=7.0) media



Fig S16. In vitro cytotoxic effects of NASP on THLE2 (a) and HepG2 (b) cells for 24-h incubation. Data presented the mean of at least triplicate measurements and given as mean \pm standard error of the mean. The IC₅₀ value for the 24-h incubation was calculated as 105.6 \pm 1.04 µM for THLE2 cells and 109.3 \pm 1.03 µM for HepG2 cells.



Scheme S1. Proposed mechanism for the recognition of homocysteine

Table S1. Nominal parameters specified during the assessment of the proposed method for Cys and Hcy

slit of excitation	10 nm
slit of emission	10 nm
monitored wavelength	λ_{ex} =360 nm, λ_{em} =416 nm
photomultiplier tube (PMT) voltage	600 Volt
temperature	rt
pH	0.0670 M PBS buffer pH=7.0 (EtOH:H ₂ O (90:10, v:v)

		situation			
	parameters	nominal (+)	Changed (–)		
1	storage temperature (°C)	37.5	25		
2	source of water	ultrapure	distilled		
3	pH	7.0	5.0		
4	storage time before the analysis (h)	16	24		
5	nitrogen atmosphere	no	yes		
6	EtOH:H ₂ O (v:v)	90 / 10	80 / 20		
7	temperature of analysis (°C)	25	15		

Table S2.Parameters used in the robustness analysis of the proposed method for Cys and Hcy

Parameter \ C _i	<i>C</i> ₁	C_2	C_3	<i>C</i> ₄	C_5	C_6	C ₇	<i>C</i> ₈
1	+	+	+	+	–	–	–	–
2	+	+	_	-	+	+	_	_
3	+	-	+	-	+	_	+	_
4	+	+	_	_	_	_	+	+
5	+	_	+	_	_	+	_	+
6	+	_	_	+	+	_	_	+
7	+	_	_	+	_	+	+	_

Table S3.Factorial combinations used in the Youden test of robustness analysis of the proposed
method for Cys and Hcy

		average values (x)			highest		lowest	
	concentration	16h	20h	24h	value of emission	Q_{value}	value of emission	Q_{value}
Cys	1×10 ⁻⁵ M	91.60	99.40	111.24	111.24	0.60	91.60	0.40
	2×10 ⁻⁵ M	75.25	77.67	87.07	87.07	0.80	75.25	0.20
Нсу	1×10 ⁻⁵ M	75.29	75.31	78.11	78.11	0.07	75.29	0.93
	2×10 ⁻⁵ M	179.69	172.43	175.21	179.69	0.80	172.43	0.20

Table S4.Dixon's test utilized to the repeatability of the proposed method for Cys and Hcy

 $^{\dagger}Q_{critical}$ =0.970

			Cys			Нсу	
		analyst (1)	analyst (2)	F _{calcukated}	analyst (1)	analyst (2)	F _{calcukated}
_	Average (x)	91.60	92.49		75.30	78.86	
5 M	SD	0.13	0.12		0.15	0.12	
10-	RSD (%)	0.14	0.13	1.29	0.2	0.15	1.76
Ť	RSD Horwitz (%)	7.58	7.58	$F_{cal} < F_{critical}$	7.58	7.58	$F_{cal} < F_{critical}$
161	HorRat ratio	0.02	0.02		0.03	0.02	
	Average (x)	75.25	79.14		179.69	176.10	
-2 2	SD	0.14	0.12		0.13	0.19	
10	RSD (%)	0.19	0.15	1.56	0.07	0.11	2.34 $F_{cal} < F_{critical}$
h 2	RSD Horwitz (%)	6.83	6.83	$F_{cal} < F_{critical}$	6.83	6.83	
16	HorRat ratio	0.03	0.02		0.01	0.02	
4	Average (x)	99.40	91.94		75.30	79.20	1.56 F _{cal} < F _{critical}
l 2−1	SD	0.19	0.17		0.13	0.11	
×10	RSD (%)	0.19	0.18	1.08 $F_{cal} < F_{critical}$	0.18	0.14	
h 1:	RSD Horwitz (%)	7.58	7.58		7.58	7.58	
20	HorRat ratio	0.03	0.02		0.02	0.02	
	Average (x)	77.67	76.53		172.43	176.43	1.89 F _{cal} < F _{critical}
ا م ۴	SD	0.21	0.20		0.11	0.16	
$\dot{10}$	RSD (%)	0.27	0.27	1.02	0.07	0.09	
h 2>	RSD Horwitz (%)	6.83	6.83	$F_{cal} < F_{critical}$	6.83	6.83	
20	HorRat ratio	0.04	0.04		0.01	0.01	
	Average (x)	111.24	117.77		78.11	78.18	
-5 N	SD	0.15	0.10		0.09	0.07	1.46
$^{-10}$	RSD (%)	0.14	0.08	2.83 $F_{cal} < F_{critical}$	0.12	0.1	
h 1	RSD Horwitz (%)	7.58	7.58		7.58	7.58	$F_{cal} < F_{critical}$
24	HorRat ratio	0.02	0.01		0.02	0.01	
Z	Average (x)	108.4356	103.62		175.21	173.04	
l -5]	SD	0.13	0.14		0.14	0.14	
×10	RSD (%)	0.12	0.14	1.28	0.08	0.08	1.03
h 2:	RSD Horwitz (%)	6.83	6.83	$F_{cal} < F_{critical}$	6.83	6.83	$F_{cal} < F_{critical}$
24 h	HorRat ratio	0.02	0.02		0.01	0.01	

Table S5.Intermediate precision analysis of the proposed method for Cys and Hcy verified by
the HorRat ratio

[†]F_{critical}=2.98