

## Supporting Information

### **Cobalt disulfide supported on porous carbon foam for high performance hydrogen evolution reaction**

Yujin Su<sup>a</sup>, Jinxin Liu<sup>a</sup>, Jinling Zhong<sup>b</sup>, Cuicui Zhang<sup>c</sup>, Qing Li<sup>a\*</sup>, Aijun Li<sup>a</sup>, Yantao Zhang<sup>a</sup>, Haichao Jiang<sup>a</sup>, Shanlin Qiao<sup>a</sup>

<sup>a</sup> *College of Chemistry and Pharmaceutical Engineering, Hebei University of Science and Technology, Shijiazhuang 050018, China*

<sup>b</sup> *Key Laboratory of Power Electronics for Energy Conservation and Motor Drive of Hebei Province, Department of Electrical Engineering, Yanshan University, Qinhuangdao 066004, China*

<sup>c</sup> *Shijiazhuang People's Medical College, Shijiazhuang 050000, China*

\*E-mail: [iccaslq@hebust.edu.cn](mailto:iccaslq@hebust.edu.cn)

## Contents

**Figure S1.** SEM images of pure CoS<sub>2</sub>.

**Figure S2.** TEM images of CoS<sub>2</sub>-CF-4.

**Figure S3.** SEM image of CoS<sub>2</sub>-CF-4 (a). Mapping images of (b) C, (c) Co, (d) N, (e) O, (f) P, and (g) S for CoS<sub>2</sub>-CF-4.

**Figure S4.** The N<sub>2</sub> adsorption/desorption isotherms of CoS<sub>2</sub>-CF-X (X=1, 2, 3, 4) composites (a), (c), (e) and (g). The calculated pore parameters of CoS<sub>2</sub>-CF-X (X=1, 2, 3, 4) composites (b), (d), (f) and (h).

**Figure S5.** The N<sub>2</sub> adsorption/desorption curves (a) and pore size distribution with CF-X (X=1, 2, 3, 4) composites (b).

**Figure S6.** The N<sub>2</sub> adsorption/desorption isotherms of CF-X (X=1, 2, 3, 4) composites (a), (c), (e) and (g). The calculated pore parameters of CF-X (X=1, 2, 3, 4) composites (b), (d), (f) and (h).

**Figure S7.** The polarization curves of CoS<sub>2</sub> (a) and its corresponding Tafel plot (b), the charge transport resistance of CoS<sub>2</sub> (c).

**Figure S8.** The polarization curves of CoS<sub>2</sub>-CF-5 (a) and its corresponding Tafel plots (b) the charge transport resistance of CoS<sub>2</sub>-CF-5 (c).

**Figure S9.** The LSV curves of CF-X (X=1, 2, 3, 4, 5), the corresponding Tafel plots (b) and charge transport resistance (c).

**Figure S10.** SEM images of CoS<sub>2</sub>-CF-4 composite after 4000 linear cycles.

**Figure S11.** Chronoamperometry test of CoS<sub>2</sub>-CF-4.

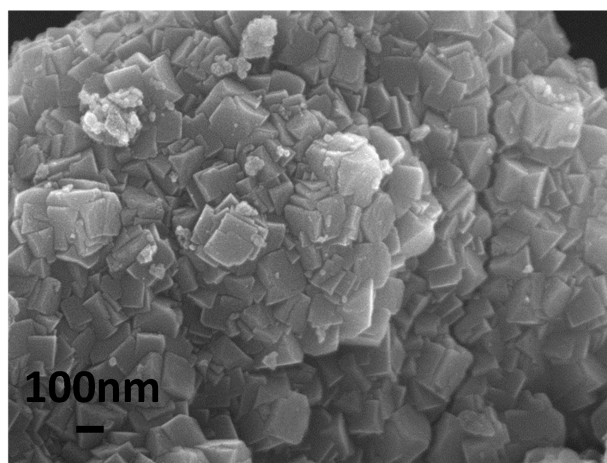
**Figure S12.** The stability of carbon rod electrode instead of platinum plate.

**Table S1.** Elemental analysis of CoS<sub>2</sub>-CF-4 composite.

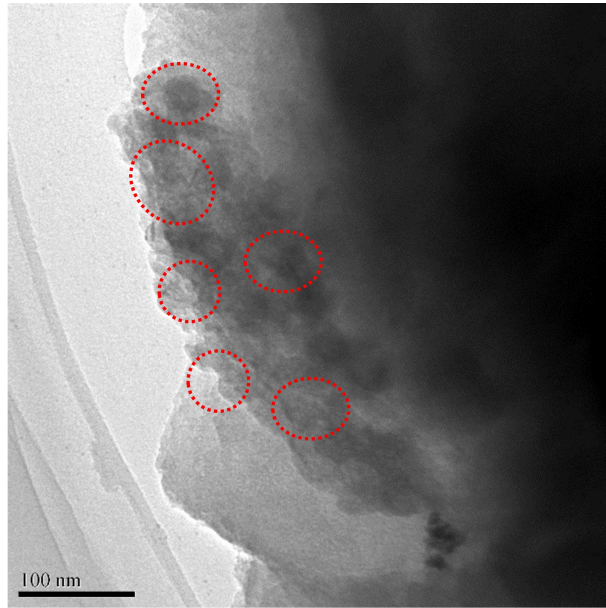
**Table S2.** XPS spectra fitting result of CoS<sub>2</sub>-CF-4 composite.

**Table S3.** The mechanism process in the electrochemical hydrogen evolution reaction (HER).

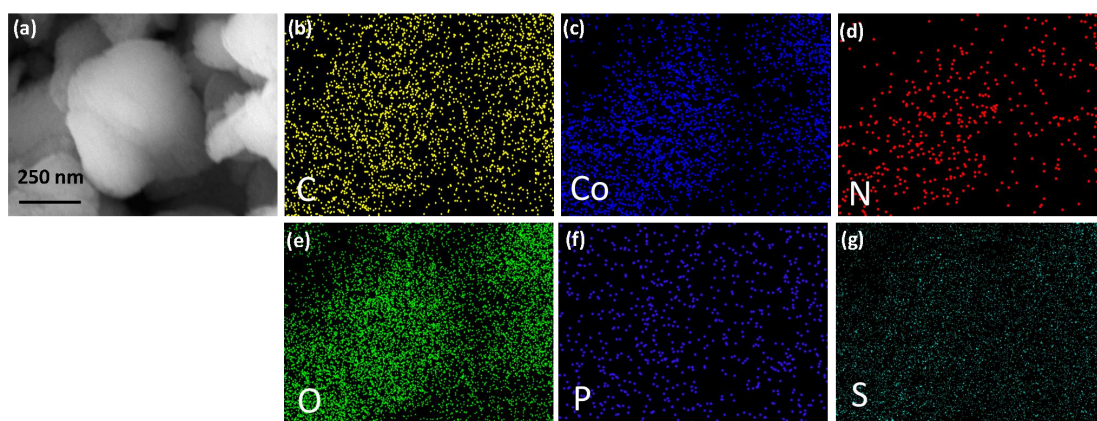
**Table S3.** HER activity of CoS<sub>2</sub>-CF-X and CF-X composite.



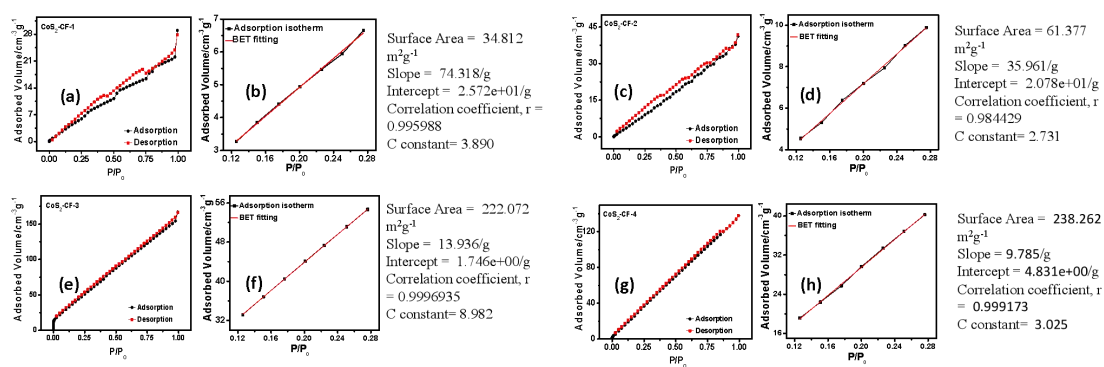
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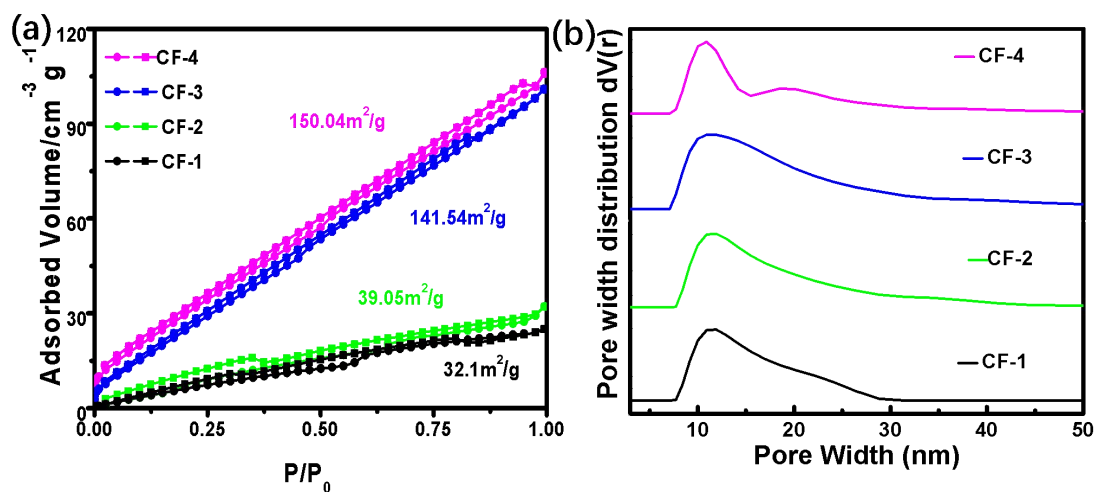
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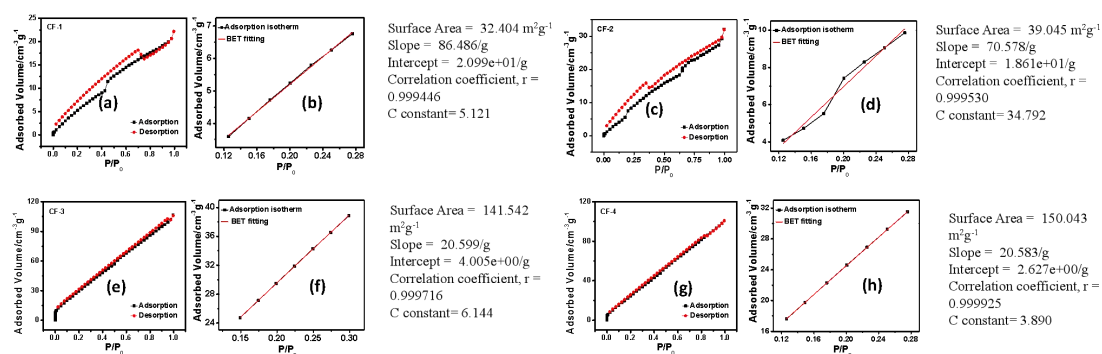
**Figure S3.** SEM image of  $\text{CoS}_2\text{-CF-4}$  (a). Mapping images of (b) C, (c) Co, (d) N, (e) O, (f) P, and (g) S for  $\text{CoS}_2\text{-CF-4}$ .



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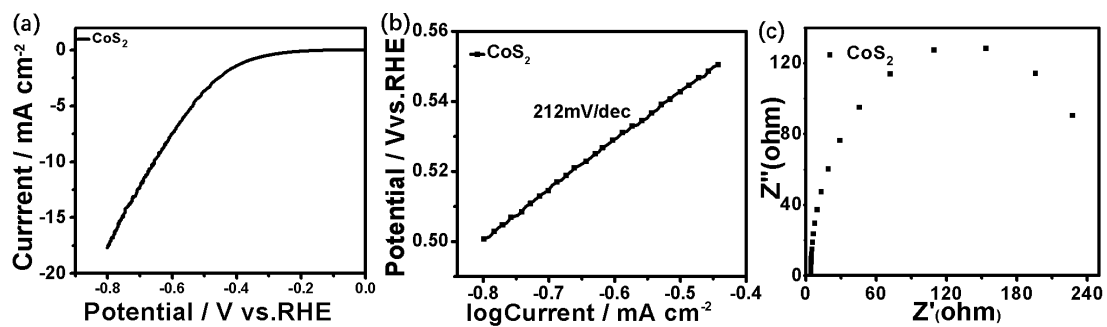


**Figure S5.** The  $N_2$  adsorption/desorption curves (a) and pore size distribution with CF-X (X=1, 2, 3, 4) composites (b).

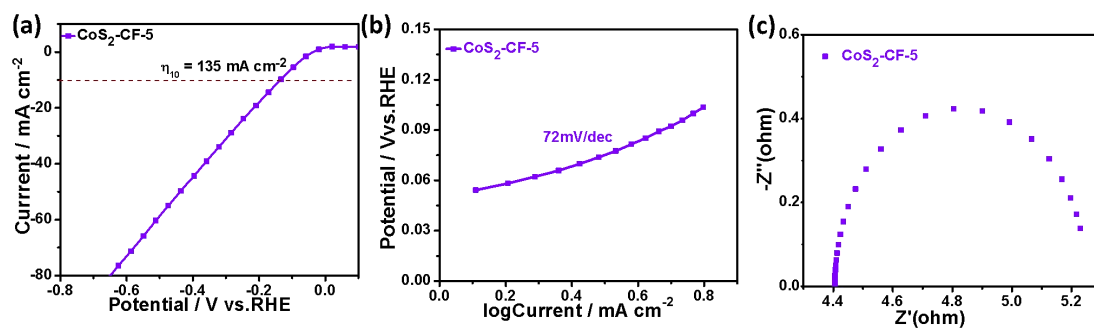


**Figure S6.** The  $N_2$  adsorption/desorption isotherms of CF-X (X=1, 2, 3, 4) composites (a), (c), (e) and (g). The calculated pore parameters of CF-X (X=1, 2, 3, 4) composites (b), (d), (f) and (h).

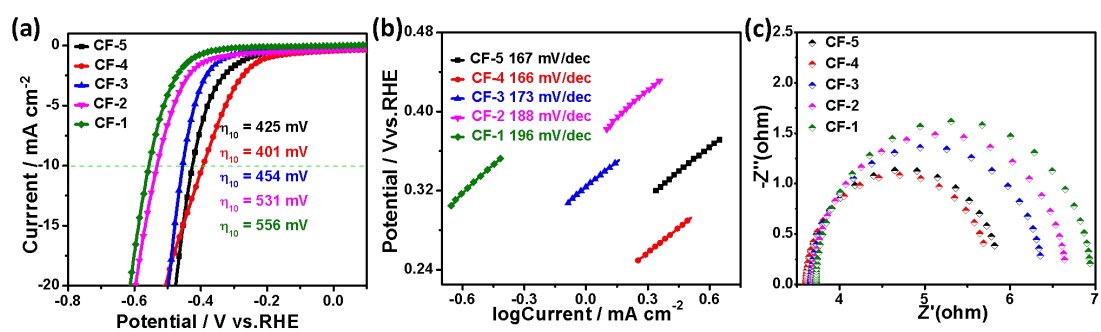




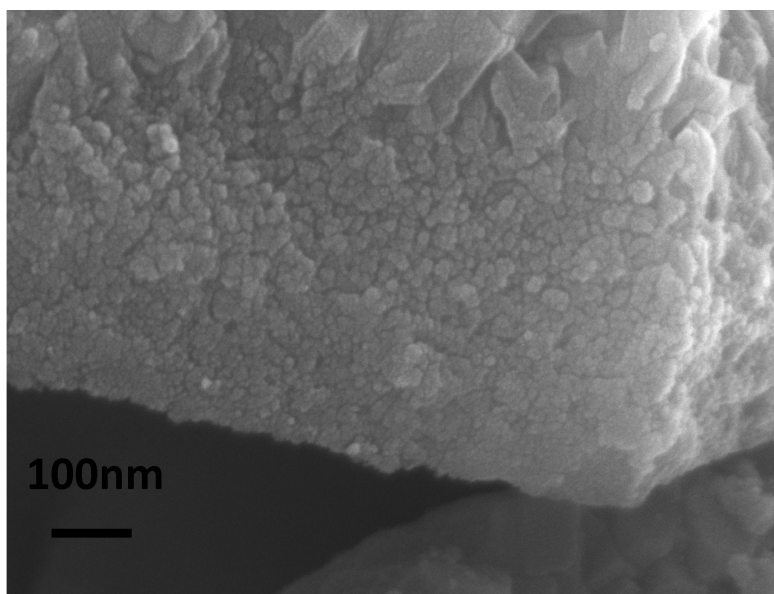
**Figure S7.** The polarization curves of CoS<sub>2</sub> (a) and its corresponding Tafel polts (b), the charge transport resistance of CoS<sub>2</sub> (c).



**Figure S8.** The polarization curves of CoS<sub>2</sub>-CF-5 (a) and its corresponding Tafel plots (b) the charge transport resistance of CoS<sub>2</sub>-CF-5 (c).



**Figure S9.** The LSV curves of CF-X (X=1, 2, 3, 4, 5), the corresponding Tafel plots (b) and charge transport resistance (c).



**Figure S10.** SEM images of CoS<sub>2</sub>-CF<sub>4</sub> composite after 4000 linear cycles.

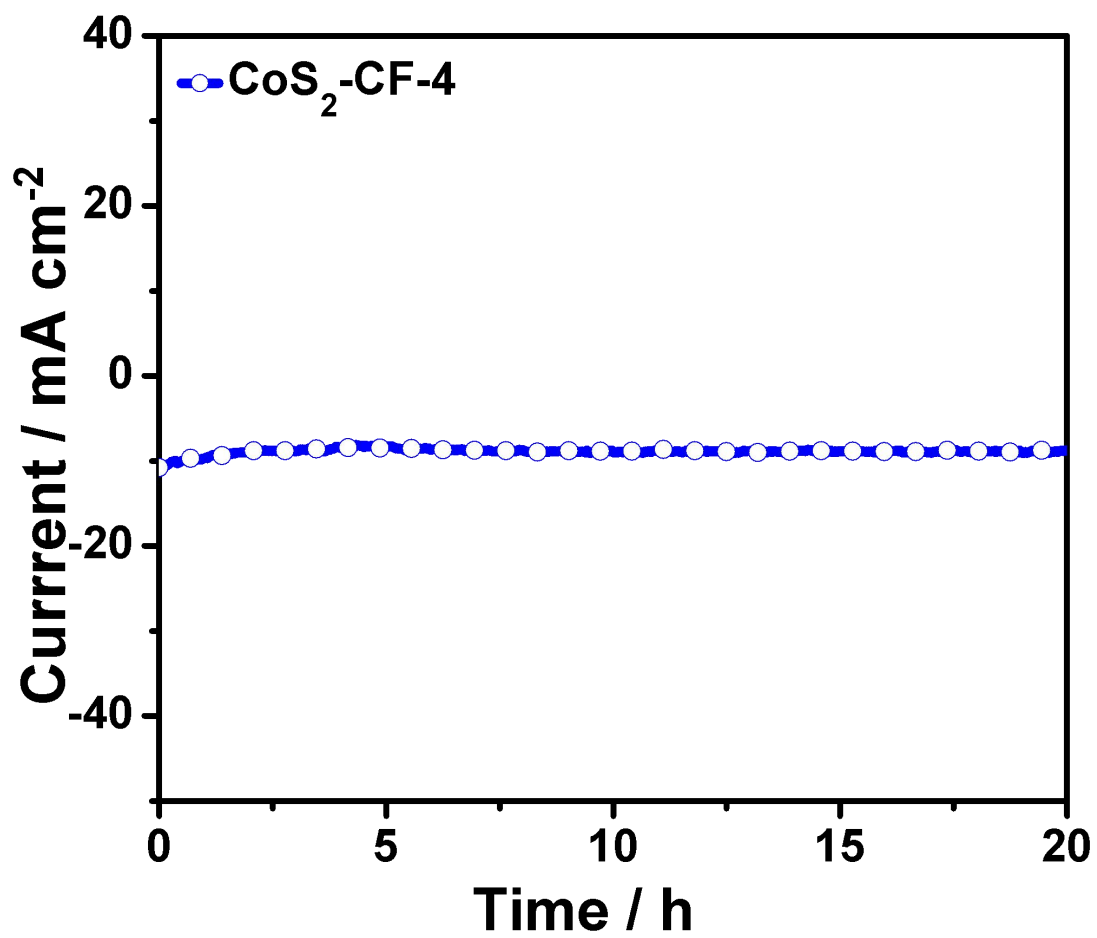
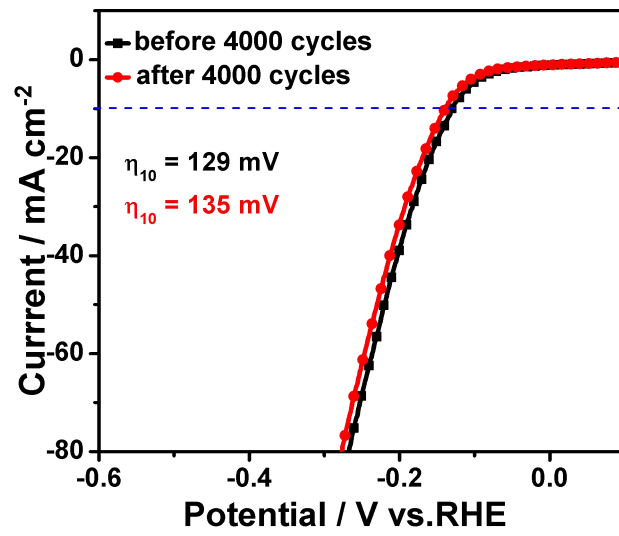


Figure S11. Chronoamperometry test of CoS<sub>2</sub>-CF-4.



**Figure S12.** The stability of carbon rod electrode instead of platinum plate.



**Table S1.** Elemental analysis of CoS<sub>2</sub>-CF-4 composite.

<b>Element</b>	<b>NetCounts</b>	<b>Weight %</b>	<b>Atom %</b>
C K	2639	12.79	22.38
N K	256	2.30	3.45
O K	9284	38.81	50.98
P K	85	0.17	0.12
S K	10248	22.41	14.69
Co K	2684	23.52	8.39
		100	100



**Table S2.** XPS spectra fitting result of CoS<sub>2</sub>-CF-4 composite.

<b>C 1s</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
C-C/C=C	284.58	0.64	25261
C-S	284.94	0.64	14695
C-O/C-N	285.43	1.2	22424
<b>N 1s</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
Oxidized N	402.22	1.59	5140
Graphitic N	400.64	1.16	606
<b>S 2p</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
S-O	169.4	1.46	16268
S-2p <sub>1/2</sub>	163.88	1.14	1112
S-2p <sub>3/2</sub>	163.06	1.18	675
<b>Co 2p</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
satelite	804.35	4.1	8148
Co <sup>2+</sup>	782.43	2.5	19669
Co <sup>2+</sup>	798.68	2.43	9317
Co <sup>2+</sup>	785.36	3.44	11178
<b>O 1s</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
C-OH	533.38	1.17	36230
C-O/ O-C-N	532.44	0.81	45410
Co-O	531.81	0.52	14121
<b>P 2p</b>			
Peak Label	Position(eV)	FWHM(eV)	Area
Oxidized-P	135.13	1.59	609
Graphitic N	134.09	2.03	403

FWHM: full width half maximum.

Atomic percent of CoS<sub>2</sub>-CF-4 composite with XPS measurements.

Material	C1s	N1s	S2p	Co2p	O1s	P2p
Atomic %	76.69	2.51	3.83	0.38	16.36	0.58

Through XPS measurements, the atomic percent of C could be obtained of 76.69 %, which was calculated to the mass content of 66.92 wt %. Besides, we calculated the S

of 8.91 wt % and Co of 1.63 wt % through the XPS analysis. The mass percentage of N was 2.19 wt%.

**Table S3.** The mechanism process in the electrochemical hydrogen evolution reaction (HER) .

Step	Reaction
Volmer step	$\text{H}^+ + \text{e}^- \rightarrow \text{H}_{\text{ad}}$
Heyrovsky step	$\text{H}_{\text{ad}} + \text{H}^+ + \text{e}^- \rightarrow \text{H}_2$
Tafel step	$2\text{H}_{\text{ad}} \rightarrow \text{H}_2$

**Table S4.** HER activity of CoS<sub>2</sub>- CF- X and CF-X composite

Catalyst	Current Density (j, mA·cm <sup>-2</sup> )	Overpotential at the corresponding j(mV)	Tafel slope (mV·dec <sup>-1</sup> )
CoS <sub>2</sub> -CF-1	10	273	93
CoS <sub>2</sub> -CF-2	10	250	92
CoS <sub>2</sub> -CF-3	10	157	82
CoS <sub>2</sub> -CF-4	10	121	68
CoS <sub>2</sub> -CF-5	10	135	72
CF-1	10	556	196
CF-2	10	531	188
CF-3	10	454	173
CF-4	10	401	166
CF-5	10	425	167