

Selective **Nanomolar** Electrochemical Detection of Serotonin, Dopamine and Tryptophan Using $\text{TiO}_2/\text{RGO}/\text{CPE}$ - Influence of Reducing Agents

Teena Joseph¹, Jasmine Thomas¹, Tony Thomas², Nygil Thomas^{1*}

Supplementary information

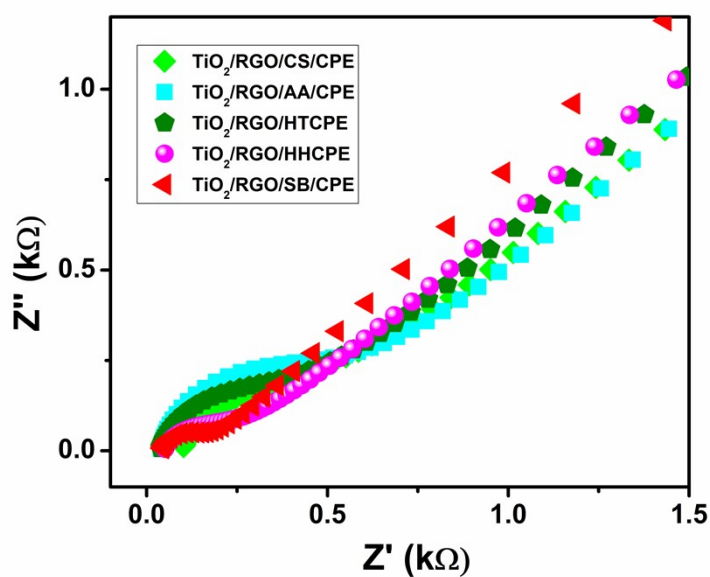


Figure S1: Nyquist plot of bare CPE and different $\text{TiO}_2/\text{RGO}/\text{CPEs}$ (a); EIS parameters: OCP potential = 0.33 V, Perturbation Amplitude = 0.01 V, Linear frequency = $3 \times 10^5 - 1 \times 10^2$ Hz. CVs of 1×10^{-4} M SER at bare CPE and different $\text{TiO}_2/\text{RGO}/\text{CPEs}$ in 0.1 M PBS of pH 7.0 (b); Scan rate: 50 mV/s

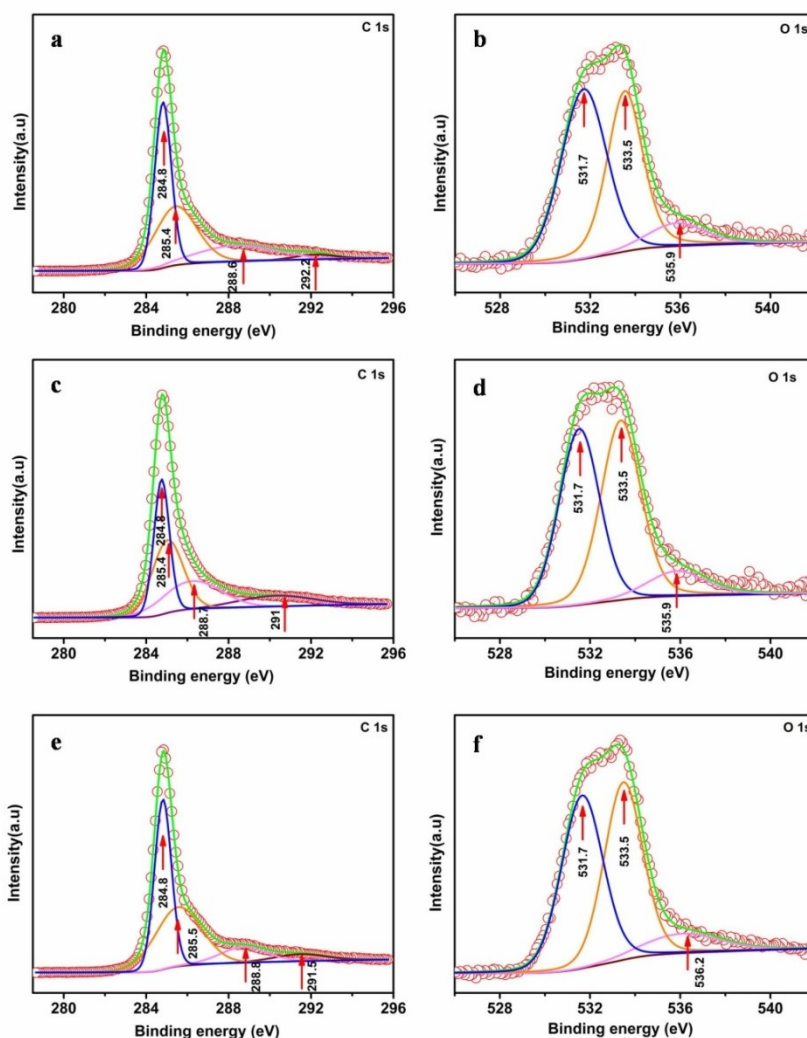
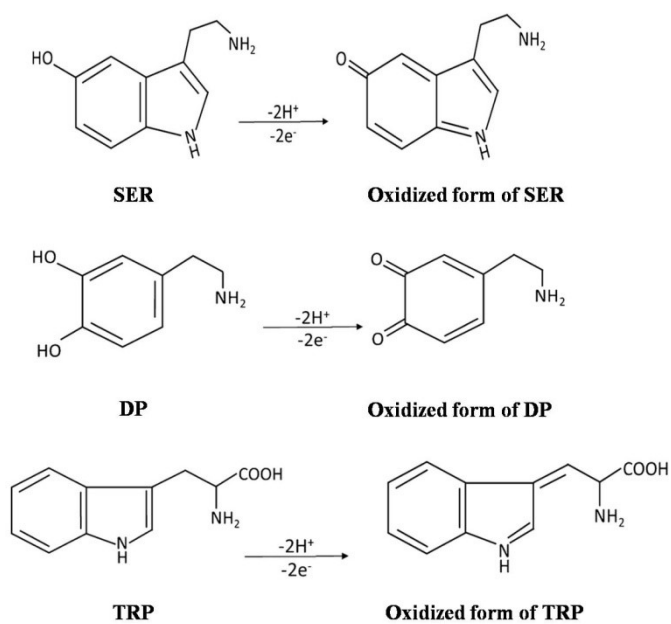


Figure S2: Deconvoluted C1s (a) and O1s (b) XPS spectra of RGSB, Deconvoluted C1s (c) and O1s (d) XPS spectra of RGHH and deconvoluted C1s (e) and O1s (f) XPS spectra of RGHT

Electrode	Anodic Peak Current (μA)	Anodic peak Potential (V)
Bare CPE	4.8	0.313
TiO ₂ /RGO/AA	6.8	0.301
TiO ₂ /RGO/CS	7.227	0.299
TiO ₂ /RGO/HT	8.398	0.304
TiO ₂ /RGO/HH	9.569	0.298
TiO ₂ /RGO/SB	9.904	0.296

Table-T1: The anodic peak current and anodic peak potential of SER using different electrodes



Scheme-1: Proposed electrochemical redox mechanism of SER, DP and TRP.

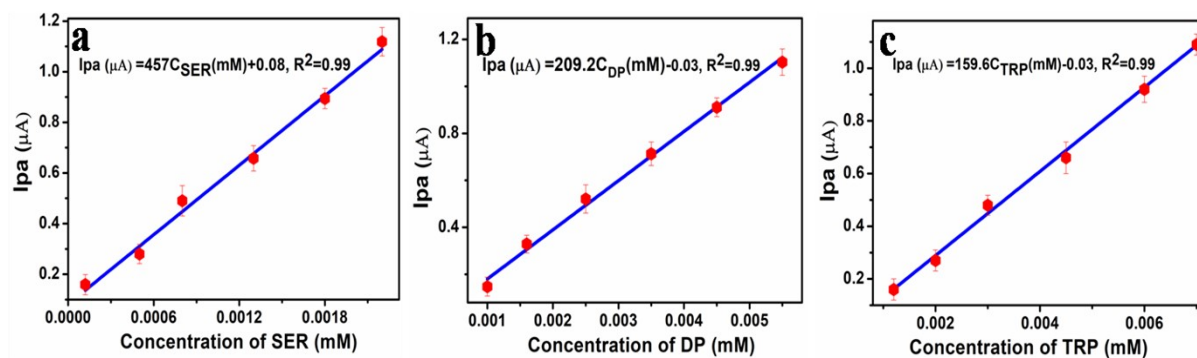


Figure S3: The plot of I_{pa} versus concentration of SER (a), DP (b) and TRP (c) in 0.1 M PBS of pH = 7.0 at $TiO_2/RGO/SB/CPE$ in its mixture.

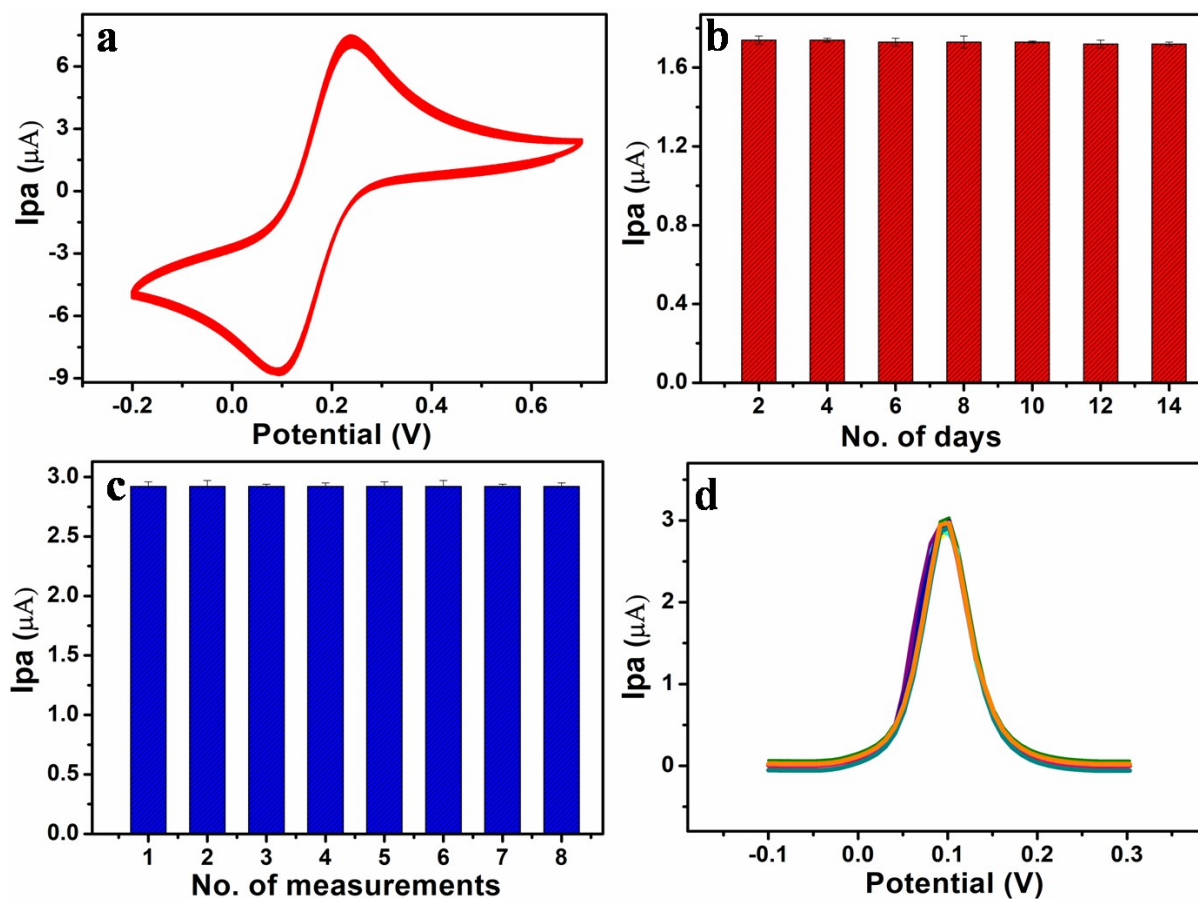


Figure S4: Stability and reproducibility results (a - d) at $\text{TiO}_2/\text{RGO}/\text{SB}/\text{CPE}$.