

Supporting Information

Aqueous media ultra-sensitive detection of antibiotics via highly stable luminescent 3D Cadmium-based MOF

Muhammad Asad^a, Shan Wang^{a*}, Qian-You Wang^a, Lin-Ke Li^{a*}, Muhammad Imran Anwar^b, Ayesha Younas^c, Shuang-Quan Zang^a

^aHenan Key Laboratory of Crystalline Molecular Functional Materials, Henan International Joint Laboratory of Tumor Theranostical Cluster Materials, Green Catalysis Center, and College of Chemistry, Zhengzhou University, Zhengzhou 450001, P. R. China

^bGreen Catalysis Center and College of Chemistry, Zhengzhou University, 450001 Zhengzhou, Henan, PR China

^cDepartment of Pharmaceutics, School of Pharmaceutical Sciences, Zhengzhou University, Zhengzhou 450001, Henan, P.R. China

Table of contents

Section	Detail	Page.no
i	Experimental: Materials and methods	S3
ii	Crystallographic data collection and structural refinement	S3
iii	Crystal data for Compound 1 (Table. S1)	S4
iv	Selected Bond Lengths and Bond Angles (Table. S2) (Table.S3)	S5
v	Thermal stability test for 1 (TGA) (Fig. S1)	S6
vi	SEM and Mapping (Fig. S2)	S6
vii	Quenching and LoD calculations for antibiotics (Figs. S3-S10)	S7-S15
viii	Calculated K_{sv} & K_q and LoD values for common selected antibiotics (Table. S4)	S16
ix	Recyclability study (Fig. S11)	S16
x	Uv-visible spectroscopy (Fig. S12)	S17
xi	TCSPC calculation for 1 and different antibiotics (Fig. S13) (Table. S5)	S18
xii	DFT calculations HOMO-LUMO (Fig. S14) (Table. S6)	S19
xiii	References	S20

i. Materials and methods

All the chemical reagents and solvents were purchased of analytical grade from certified sellers and used without any further refinement. X-ray diffraction (PXRD) patterns were obtained with monochromatized Cu K α ($\lambda = 1.540598 \text{ \AA}$) on a Rigaku D/Max-2500PC diffractometer at ambient temperature. FT-IR spectra were recorded with wavelength range 4000-400 cm^{-1} using BRUKER ALPHA II spectrophotometer. TGA analysis was conducted via TA Q50 thermal analyzer with range of 30–800 $^{\circ}\text{C}$ at the rate of 10 $^{\circ}\text{C min}^{-1}$ under ambient environment. Scanning electron microscopic (SEM) study was investigated with Zeiss Sigma 500 microscope. UV-visible spectroscopy was performed via Hitachi UH4150 spectrophotometer. DFT calculations were carried out using Gaussian 09w package with B3LYP function and 6-31+G* basis set. Elemental analyses (EA) were carried out with a Perkin-Elmer 240 elemental analyzer. Photoluminescence emission and excitation spectra were recorded at ambient temperature with a HORIBA FluoroLog-3 fluorescence spectrometer, while time-resolved decay measurements were carried on HORIBA FluoroLog-3 fluorescence spectrometer equipped with a laser of 370 nm working in time-correlated single-photon counting mode (TCSPC) with a time resolution of 100 ns.

ii. Crystallographic data collection and structural refinement

Single crystal X-ray diffraction (SC-XRD) measurement for Cd-based MOF (**1**) was performed at 200 K on a Rigaku XtaLAB Pro diffractometer with Cu-K α radiation ($\lambda = 1.54184 \text{ \AA}$). All the structures were solved with direct methods (SHELXS)¹ and refined by full-matrix least-squares on F^2 using OLEX2,² which utilizes the SHELXL-2015 module³. The imposed restraints and constraints (ISOR, DFIX, DELU, TRIA, etc.) in the least-squares refinement of each structure were commented in the corresponding crystallographic CIF files. The crystal structures are visualized by DIAMOND3.2. Detailed information about the X-ray crystallography data, intensity collection procedure, and refinement results for the compound are summarized in Table S1.

iii. Table S1 Crystal data for Compound 1

Empirical formula	C ₉₃ H ₇₇ Cd ₂ N ₇ O ₁₁
Formula weight	846.71
Temperature/K	200.00(10)
Crystal system	Monoclinic
Space group	P2/c
a/Å	14.4126(2)
b/Å	20.6834(4)
c/Å	16.4770(3)
α/°	90
β/°	104.140(2)
γ/°	90
Volume/Å ³	4763.00(15)
Z	2
ρ _{calc} /g/cm ³	1.181
μ/mm ⁻¹	4.030
F(000)	1736.0
Crystal size/mm ³	0.03 × 0.03 × 0.02
Radiation	CuKα (λ = 1.54184)
2θ range for data collection/°	6.324 to 145.55
Index ranges	-17 ≤ h ≤ 17, -24 ≤ k ≤ 23, -20 ≤ l ≤ 18
Reflections collected	31299
Independent reflections	9196 [R _{int} = 0.0516, R _{sigma} = 0.0572]
Data/restraints/parameters	9196/95/536
Goodness-of-fit on F ²	1.050
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0574, wR ₂ = 0.1523
Final R indexes [all data]	R ₁ = 0.0798, wR ₂ = 0.1626
Largest diff. peak/hole / e Å ⁻³	1.13/-1.07

iv. Table S2. Selected Bond lengths for 1

Atom	Atom	Length/Å
Cd1	N1 ³	2.300(5)
Cd1	N2	2.312(5)
Cd1	O1	2.451(3)
Cd1	O1 ⁴	2.399(3)
Cd1	O2	2.373(4)
Cd1	O3	2.493(3)
Cd1	O4	2.302(4)
N1	Cd1 ⁵	2.300(5)
O1	Cd1 ⁴	2.399(3)

¹2-X,+Y,3/2-Z; ²-X,+Y,1/2-Z; ³+X,1+Y,+Z; ⁴1-X,+Y,1/2-Z; ⁵+X,-1+Y,+Z

Table S3. Selected Bond angles for 1

Atom	Atom	Atom	Angle/°
N1 ³	Cd1	O1	93.03(14)
N1 ³	Cd1	O3	86.78(14)
N2	Cd1	O1	86.38(15)
N2	Cd1	O1 ⁴	90.19(15)
N2	Cd1	O2	87.57(17)
N2	Cd1	O3	95.73(15)
O1 ⁴	Cd1	O1	74.25(12)
O1 ⁴	Cd1	O3	145.01(11)
O1	Cd1	O3	140.43(11)
O2	Cd1	O1	54.22(11)
O2	Cd1	O1 ⁴	128.46(11)
O2	Cd1	O3	86.32(11)
O4	Cd1	N2	87.40(17)
O4	Cd1	O1 ⁴	91.33(12)
O4	Cd1	O1	164.25(12)
O4	Cd1	O2	139.89(12)
O4	Cd1	O3	54.70(12)
Cd1 ⁴	O1	Cd1	105.46(12)

¹2-X,+Y,3/2-Z; ²-X,+Y,1/2-Z; ³+X,1+Y,+Z; ⁴1-X,+Y,1/2-Z; ⁵+X,-1+Y,+Z

v. The thermal stability test of 1

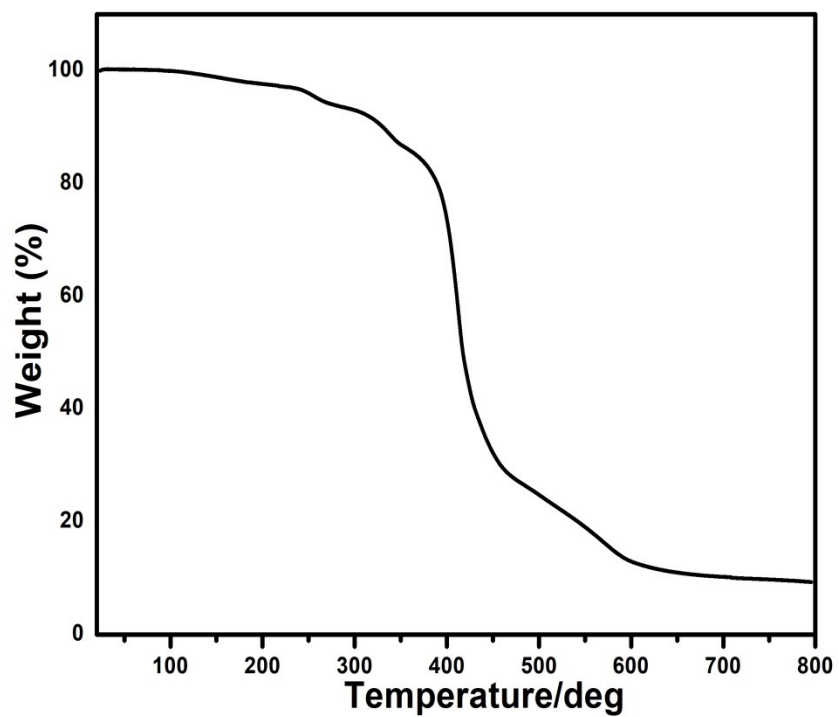


Fig. S1 Thermal stability test (TGA) of 1

vi. SEM and Mapping

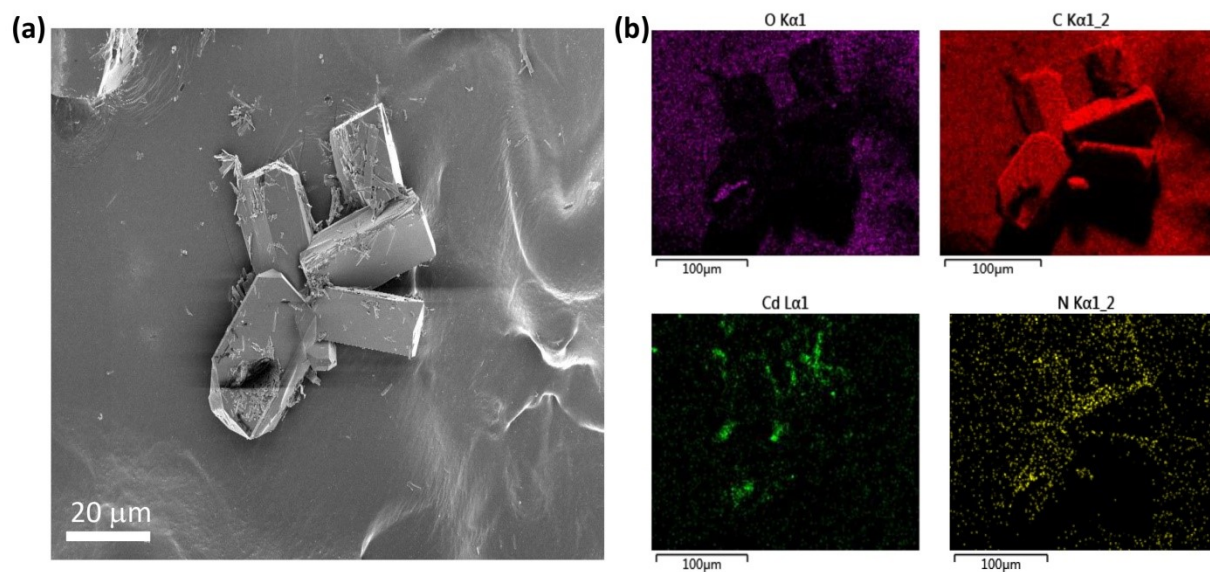


Fig. S2 (a) SEM image of 1 (b) Elemental mapping images of 1.

vii. Quenching and LoD calculations for antibiotics

Standard Deviation (σ) calculation:

Blank Readings	FL Intensity
Reading 1	2.53358
Reading 2	2.53358
Reading 3	2.47324
Reading 4	2.35596
Reading 5	2.29437
Standard Deviation (σ)	0.108251

(i)

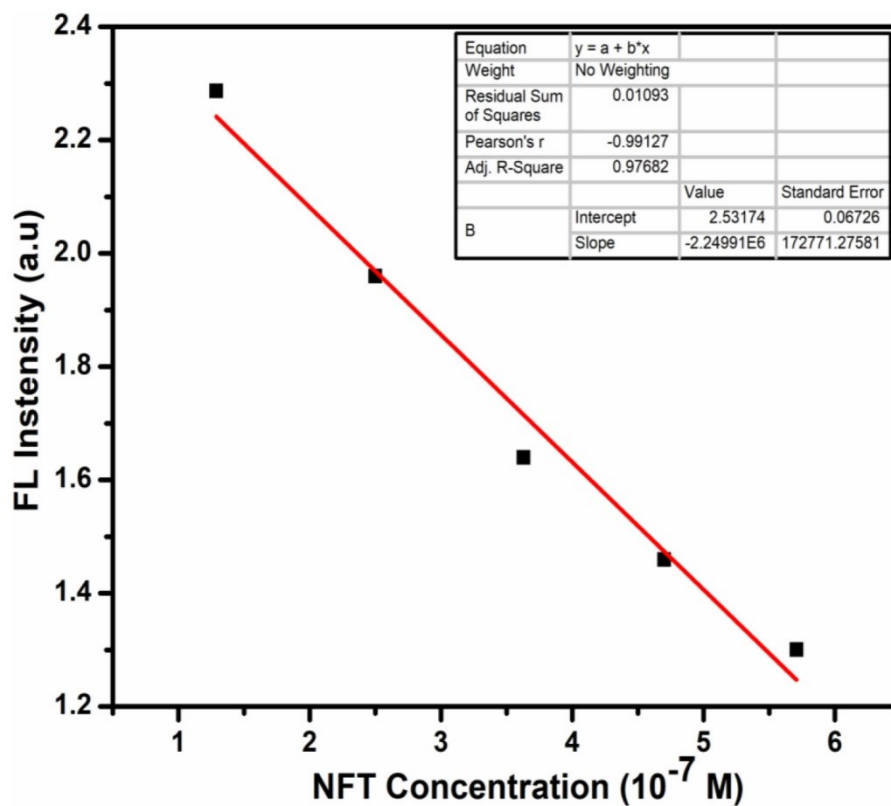


Fig. S3 LoD calculation for NFT antibiotic

Calculation of Detection limit (LOD) for NFT:

Slope (m)	2.24991E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.14 μ M

(ii)

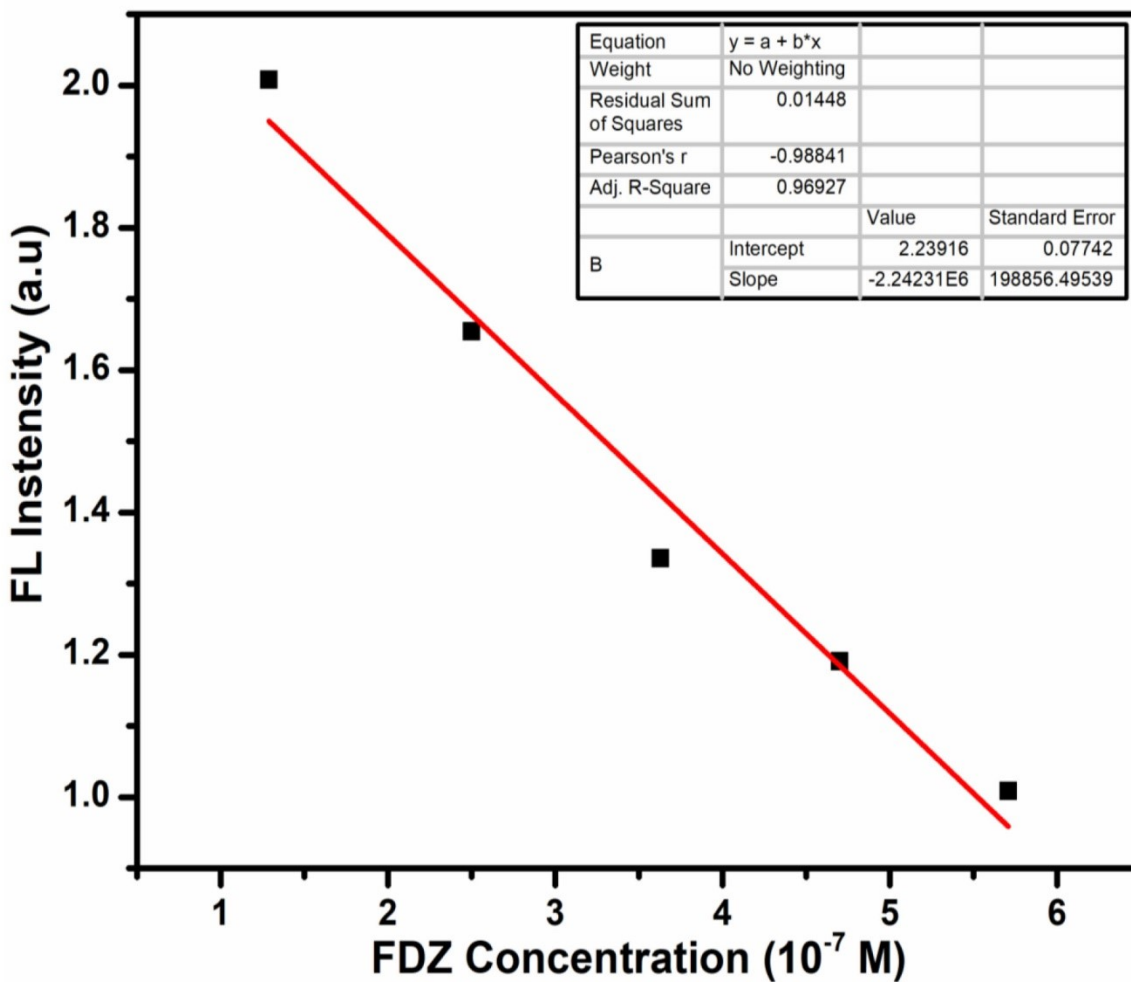


Fig. S4 LoD calculation for FDZ antibiotic

Calculation of Detection limit (LOD) for FDZ:

Slope (m)	2.24231E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.14 μ M

(iii)

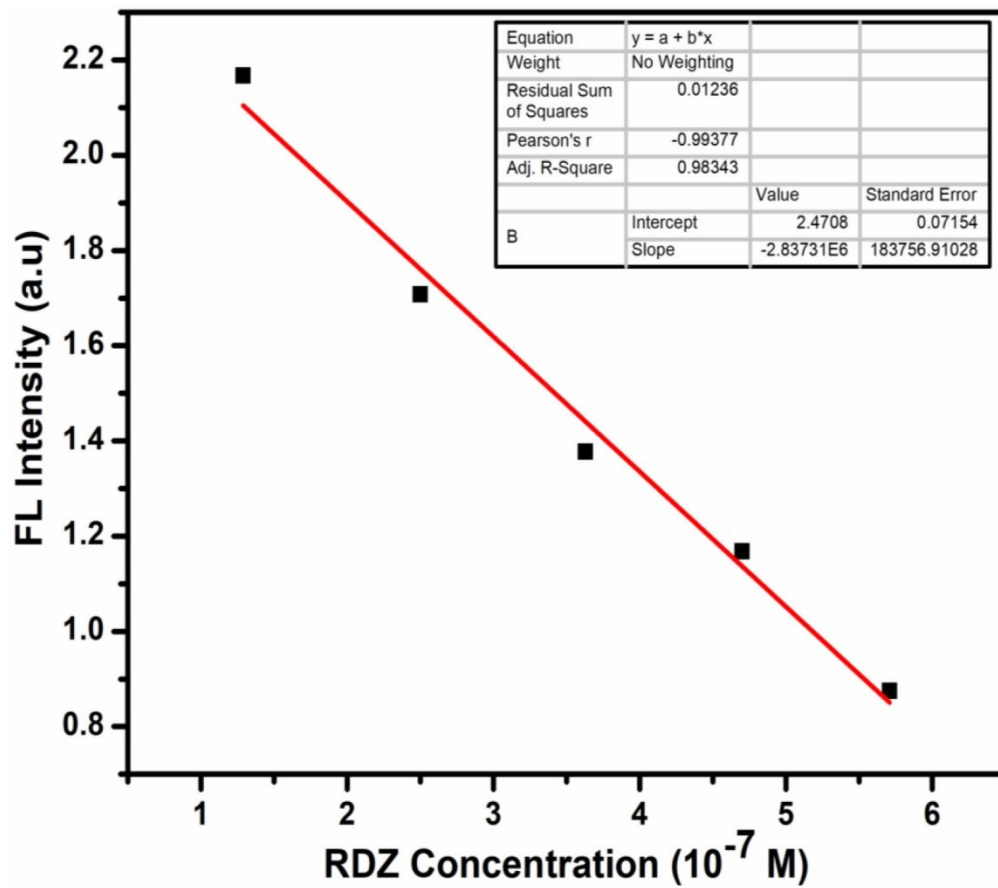


Fig. S5 LoD calculation for RDZ antibiotic

Calculation of Detection limit (LOD) for RDZ:

Slope (m)	2.83731E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.11 μ M

(iv)

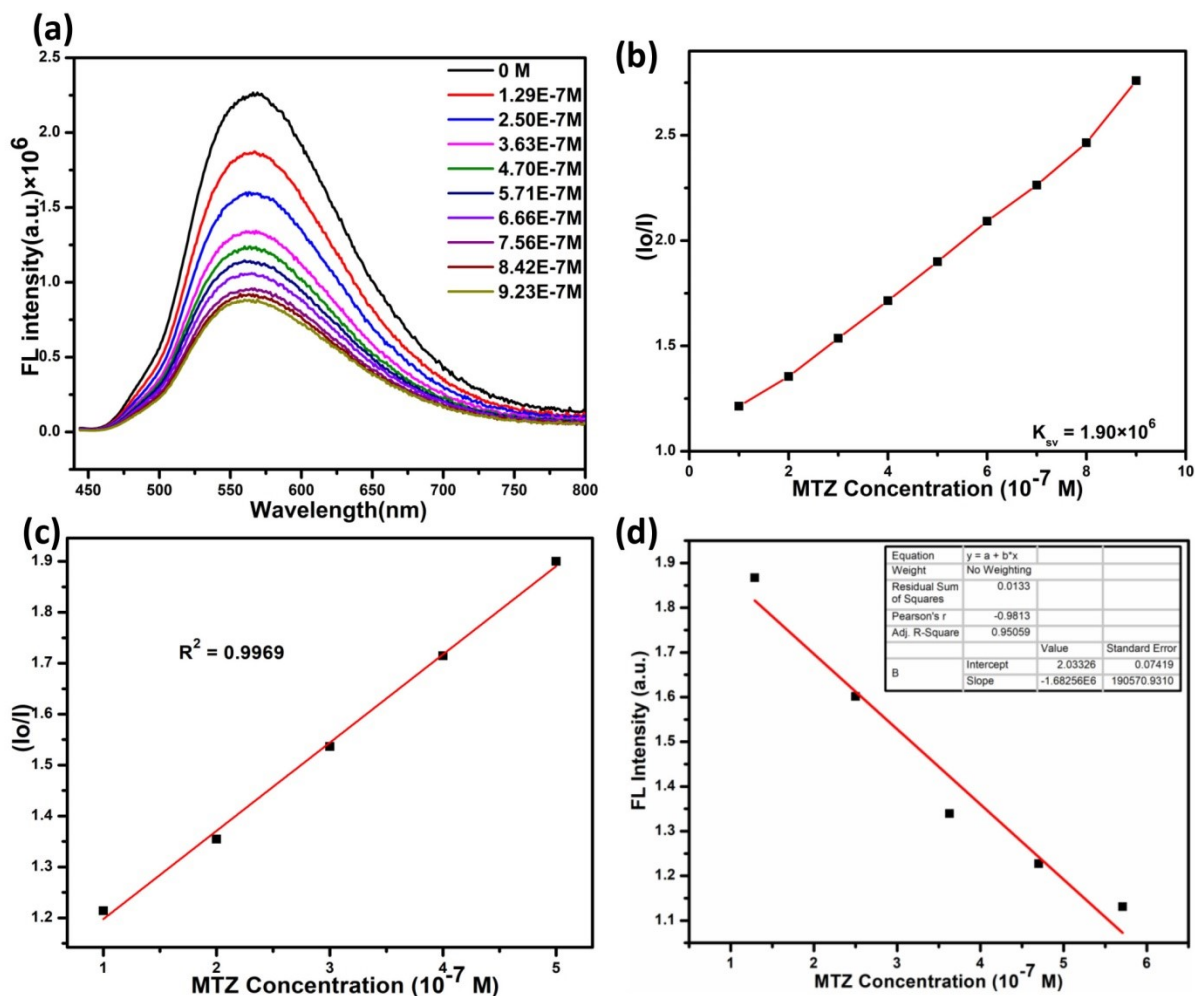


Fig. S6 Quenching titration graph and LoD calculation for MTZ antibiotic

Calculation of Detection limit (LOD) for MTZ:

Slope (m)	1.68256E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.19 μ M

(V)

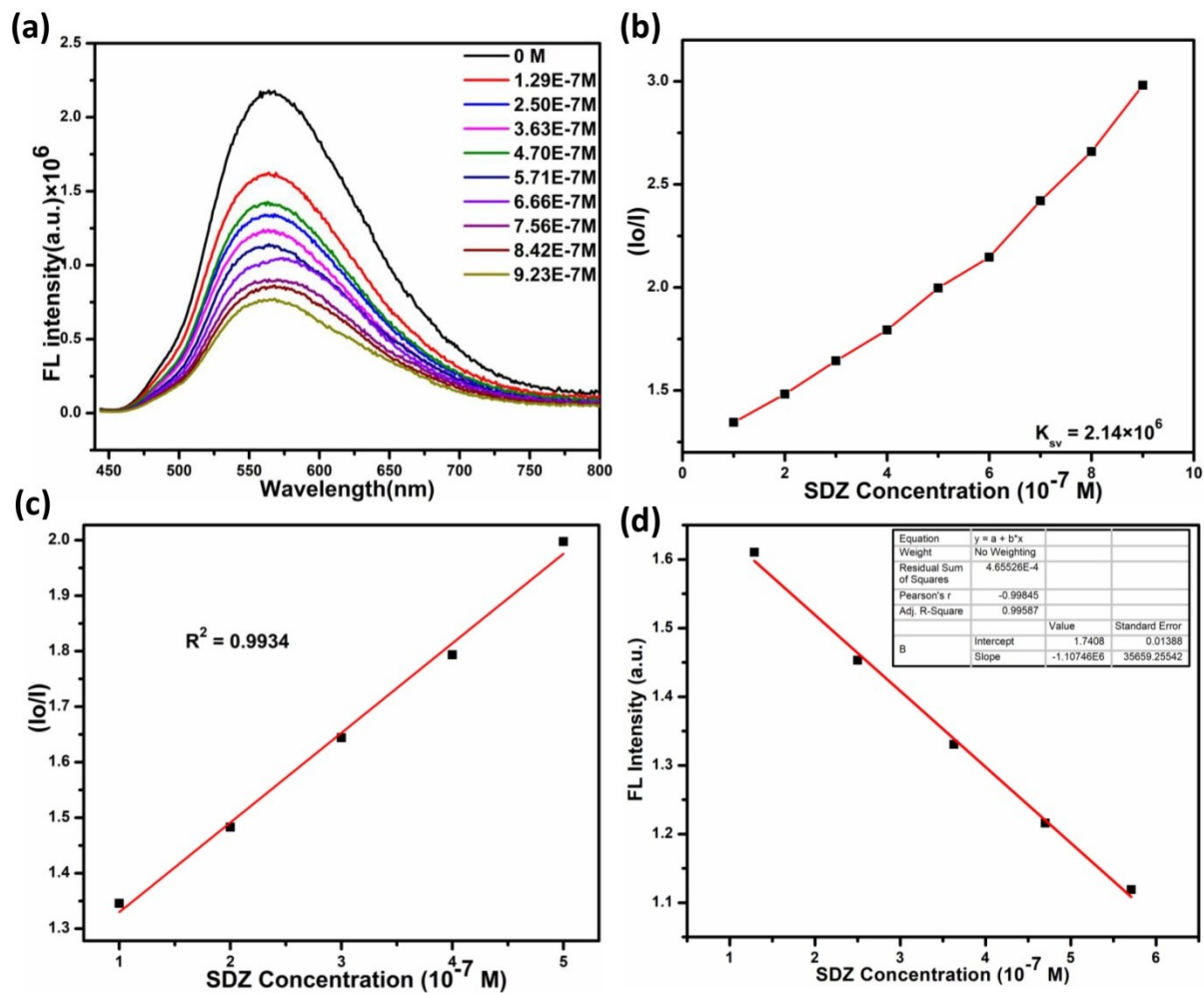


Fig. S7 Quenching titration graph and LoD calculation for SDZ antibiotic

Calculation of Detection limit (LOD) for SDZ:

Slope (m)	1.10746E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.29 μ M

(vi)

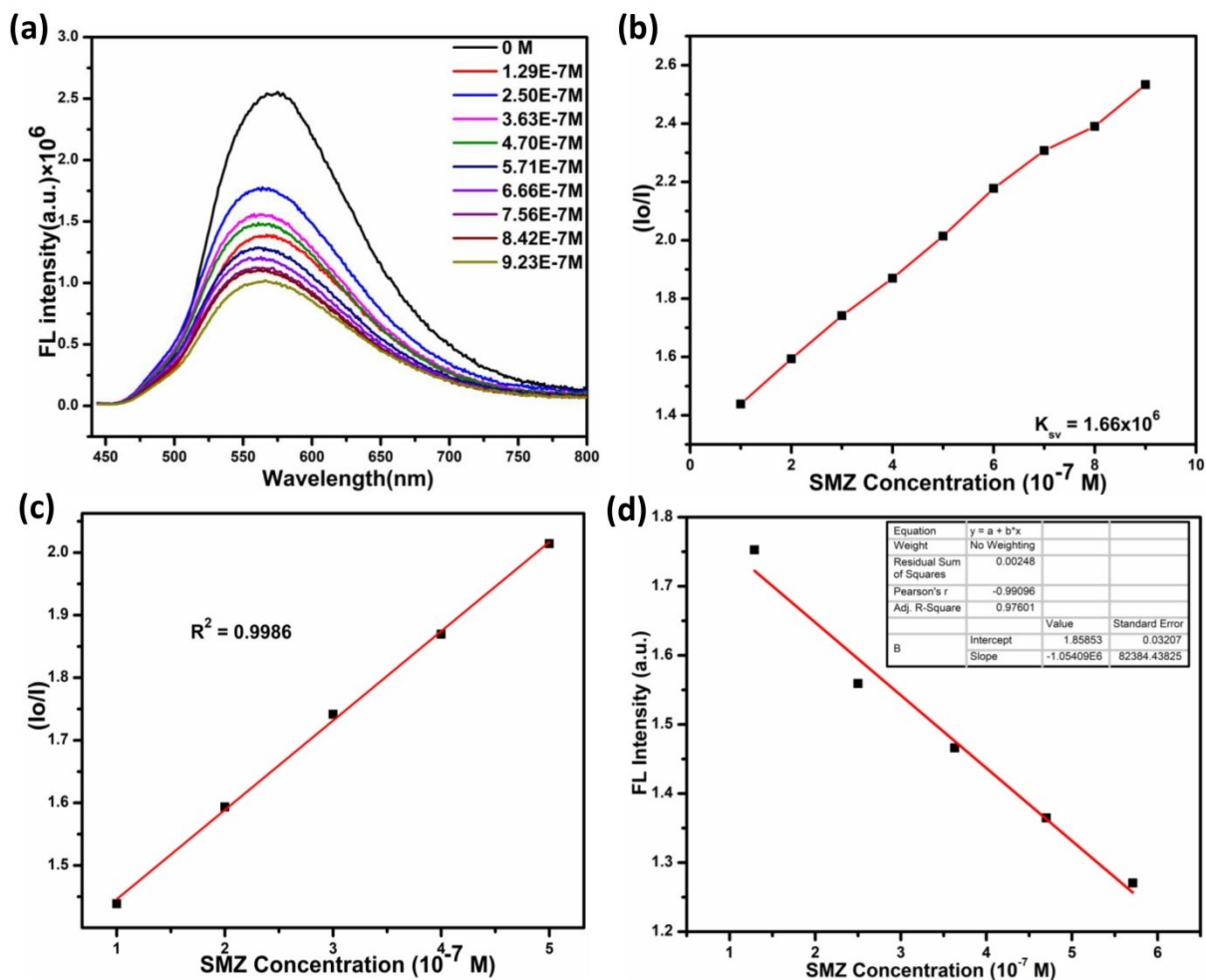


Fig. S8 Quenching titration graph and LoD calculation for SMZ antibiotic

Calculation of Detection limit (LOD) for SMZ:

Slope (m)	1.05409E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.30 μ M

(vii)

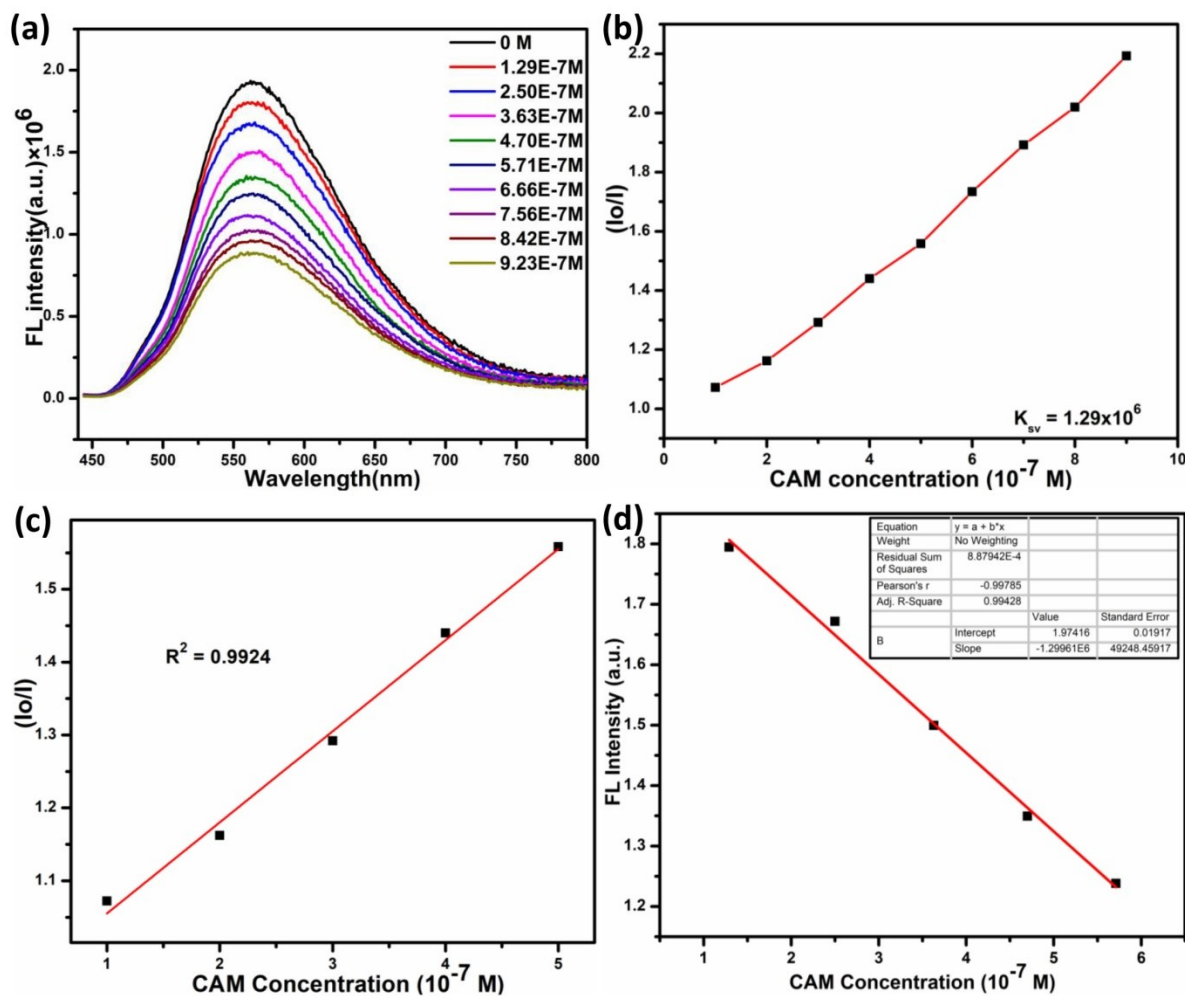


Fig. S9 Quenching titration graph and LOD calculation for CAM antibiotic

Calculation of Detection limit (LOD) for CAM:

Slope (m)	1.29961E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.24 μ M

(viii)

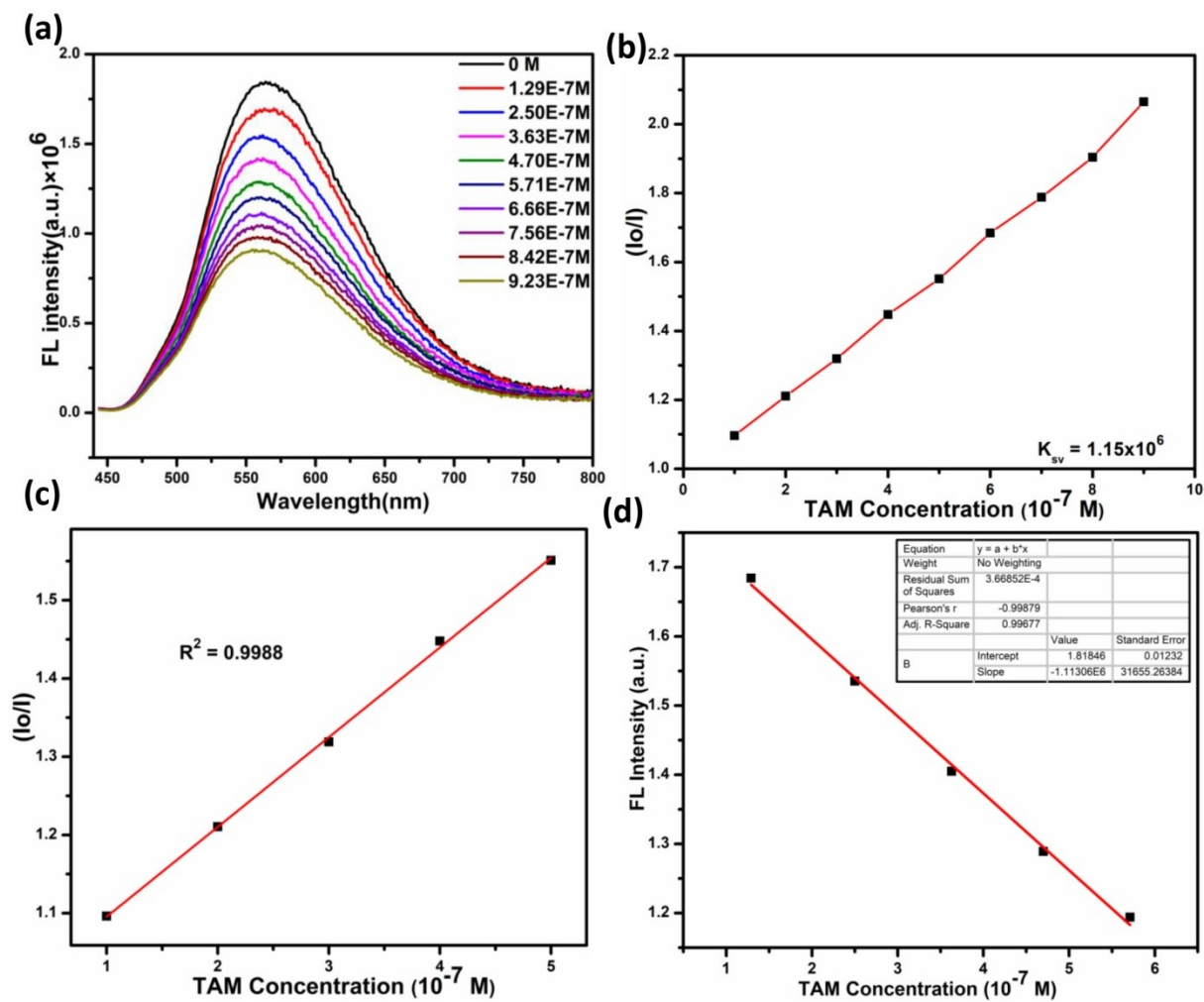


Fig. S10 Quenching titration graph and LoD calculation for TAM antibiotic

Calculation of Detection limit (LOD) for TAM:

Slope (m)	1.11306E+6
Standard deviation (σ)	0.108251
Limit of detection ($3\sigma/m$)	0.29 μ M

viii. Table. S4 Calculated K_{sv} , K_q and LoD values for common antibiotics

Selected analytes	K_{sv} (M^{-1})	K_q ($M^{-1}S^{-1}$)	Detection limits (μM)
Nitrofuratoin(NFT)	2.92×10^6	0.19×10^{15}	0.14
Furazolidone (FDZ)	3.57×10^6	0.23×10^{15}	0.14
Ronidazole (RDZ)	5.16×10^6	0.33×10^{15}	0.11
Metronidazole (MTZ)	1.91×10^6	0.12×10^{15}	0.19
Sulfadiazine (SDZ)	2.14×10^6	0.14×10^{15}	0.29
Sulfamethazine (SMZ)	1.66×10^6	0.10×10^{15}	0.30
Chloramphenicol(CAM)	1.29×10^6	0.84×10^{14}	0.24
Thiamphenicol (TAM)	1.15×10^6	0.75×10^{14}	0.29

ix. Recyclability study

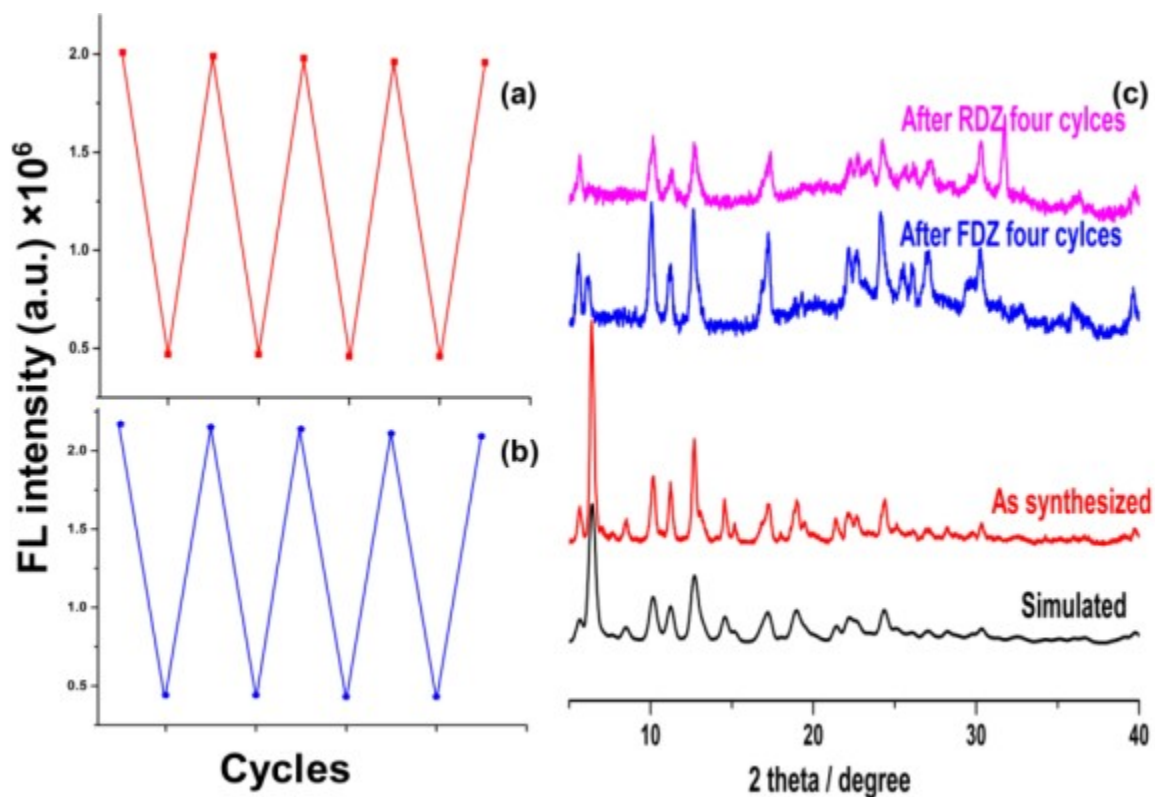


Fig. S11 Fluorescence intensity of **1** after four cycles detection of (a) FDZ, (b) RDZ and (c) P-XRD patterns of **1** after 4 cycles sensing of FDZ and RDZ compared with the simulated and as-synthesized.

x. Uv-visible spectroscopy

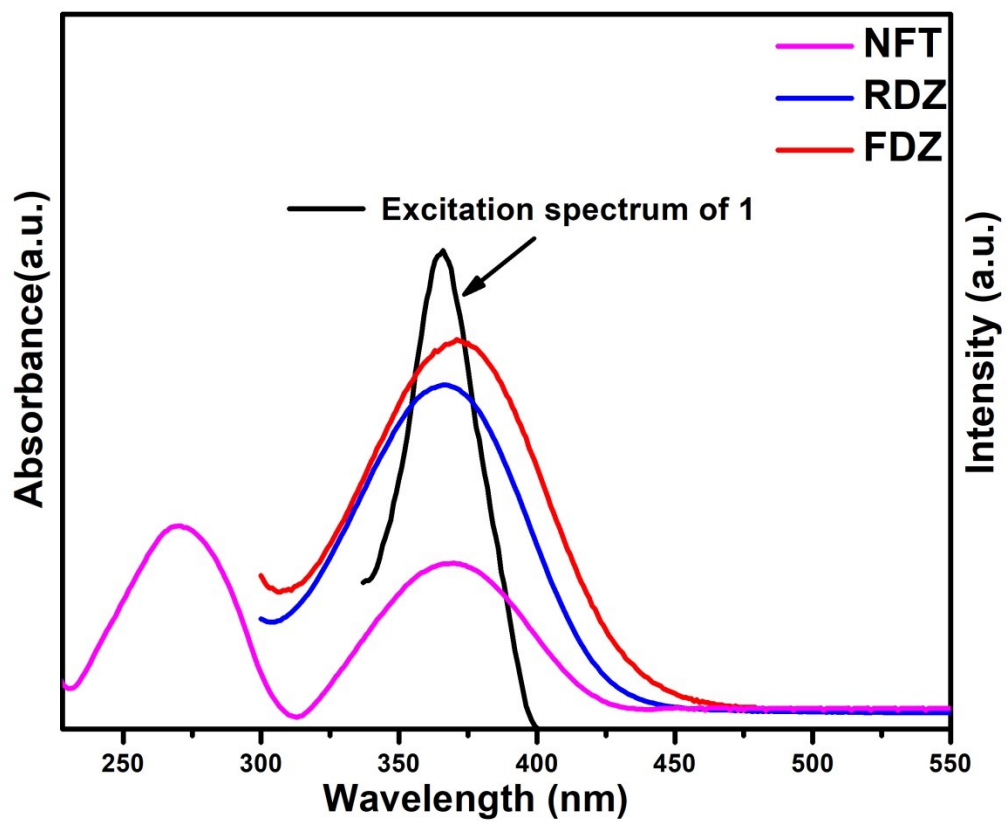


Fig. S12 UV-vis spectra of NFT, RDZ, and FDZ, and excitation spectra of 1

xi. TCSPC calculation for **1** and different antibiotics

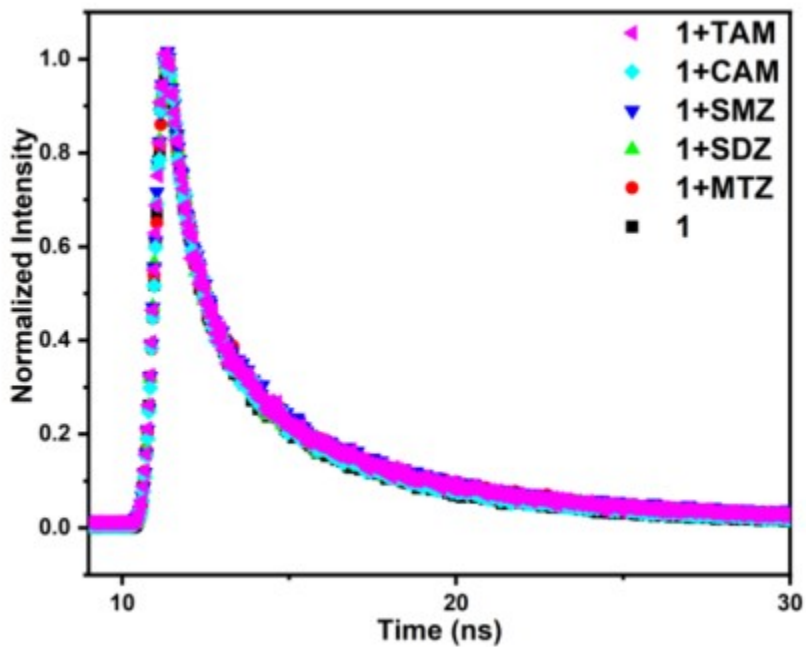


Fig. S13 Time-resolved fluorescence decay profile for **1** and different antibiotics.

Table. S5 Decay life-time calculations for **1** and antibiotics

Samples	T1 (ns)	T2 (ns)	B1	B2	(T1×B1+T2×B2) (ns)
1	19.5985	5.586098	0.6877	0.3123	15.22
1+NFT	7.10502	20.48647	0.3125	0.6875	16.30
1+FDZ	6.43309	19.7718	0.2546	0.7454	16.37
1+RDZ	21.53541	8.148406	0.6784	32.16	17.23
1+ MTZ	7.18421	20.3890	30.45	69.55	16.36
1+SDZ	6.30467	19.6943	0.2793	0.7207	15.95
1+SMZ	19.82028	6.30224	0.7100	0.2900	15.90
1+CAM	6.25478	20.07246	0.3073	0.6927	15.82
1+TAM	5.81566	19.47863	0.3038	0.6962	15.32

xii. DFT calculations

All the DFT calculations were performed to find HOMO-LUMO bandgap for **1** and different antibiotics. All theoretical calculations were carried out using Gaussian 09w package with B3LYP function and 6-31+G* basis set ⁴⁻⁶.

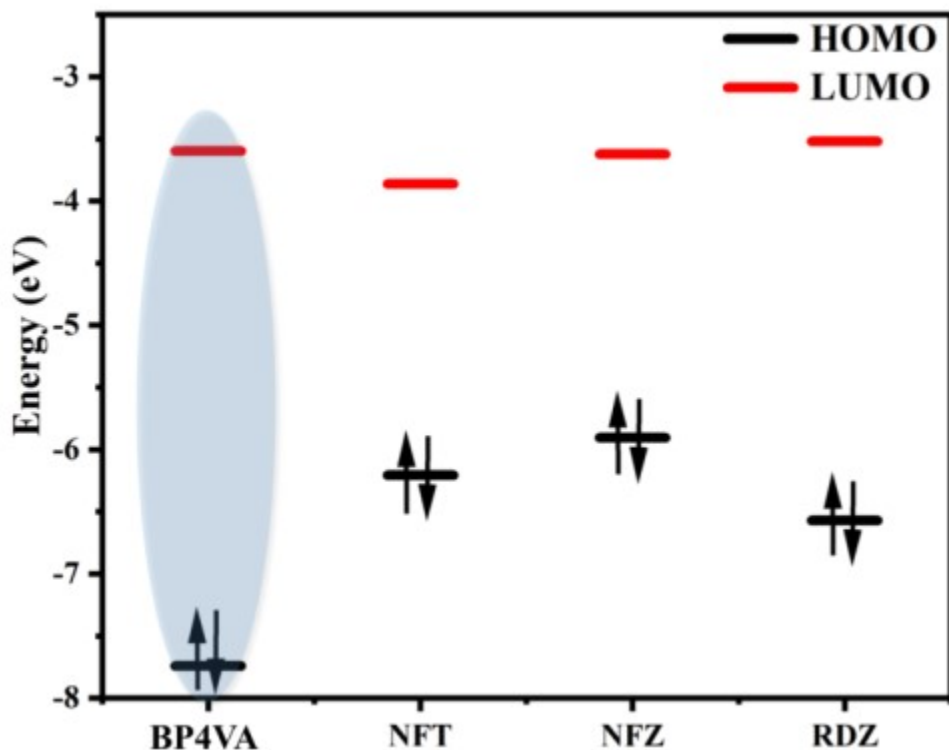


Fig. S14 Theoretically calculated HOMO and LUMO energies for ligand and different antibiotics.

Table. S6 Theoretically calculated HOMO-LUMO energy levels for different antibiotics

Analytes	HOMO (eV)	LUMO (eV)	Band gap
BP ₄ V ₄	-7.7419	-3.599	-4.1429
NFT	-6.205	-3.864	-2.341
NFZ	-5.905	-3.624	-2.281
RDZ	-6.57	-3.52	-3.05
SDZ	-7.564	0.0163	-7.547
MTZ	-6.42	-3.43	-2.99
SMZ	-5.457	-1.898	-3.559
CAM	-6.58	-3.69	-2.89
TAM	-7.09	-1.52	-5.57

xiii. References

1. G. M. Sheldrick, A short history of SHELX. *Acta Crystallogr., Sect. A: Found. Crystallogr.* **2008**, *64* (1), 112-122.
2. O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. Howard, H. Puschmann, OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Crystallogr.* **2009**, *42* (2), 339-341.
3. G. M. Sheldrick, Crystal structure refinement with SHELXL. *Acta Crystallogr. C Struct. Chem.* **2015**, *71* (1), 3-8.
4. C. Lee, W. Yang, R. G. Parr, Development of the Colle-Salvetti correlation-energy formula into a functional of the electron density. *J. Physical rev.* **1988**, *37* (2), 785.
5. B. Miehlich, A. Savin, H. Stoll, H. Preuss, Results obtained with the correlation energy density functionals of Becke and Lee, Yang and Parr. *Chem. Phys. Lett* **1989**, *157* (3), 200-206.
6. M. Frisch, G. Trucks, H. Schlegel, G. Scuseria, M. Robb, J. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. Petersson, Gaussian 09; Gaussian, Inc. Inc." Wallingford, CT. **2009**, *32*, 5648-5652.