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Supporting Information

Alkaline Metal and Alkaline Earth Metal Salts of Di(1H-tetrazol-5-

yl)Methanone (DTO): Energetic Catalysts for Ammonium

Perchlorate Decomposition

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	1	2	3	4		
Formula	$C_3H_6N_8Na_2O_4$	$C_6H_{20}N_{16}Na_4O_{12}$	$C_3H_2K_2N_8O_2$	$C_{3}K_{0.5}N_{8}ORb_{1.5}$		
M (g mol ⁻¹)	264.14	600.34	260.33	311.86		
Cryst. Syst.	triclinic	orthorhombic	monoclinic	triclinic		
Space group	P-1	Pnma	$P2_1$	P-1		
a (Å)	6.5949(13)	6.8000(14)	7.1376(14)	4.4887(9)		
b (Å)	8.7129(17)	7.5000(15)	17.029(3)	7.5710(15)		
c (Å)	9.1609(18)	24.200(5)	7.6232(15)	12.714(3)		
α (°)	102.33(3)	90	90	97.31(3)		
β (°)	90.26(3)	90.	111.38(3)	92.74(3)		
γ (°)	105.15(3)	90	90	106.32(3)		
V (Å ³)	495.34(19)	1234.2(4)	862.8(3)	409.70(16)		
Ζ	2	2	4	2		
ρ (g cm ⁻³)	1.771	1.615	2.004	2.528		
T (K)	153.15	153.15	153.15	153.15		
F (000)	268.0	616.0	520.0	294.0		
μ (mm ⁻¹)	0.225	0.202	1.093	9.226		
Reflns. collected	3918	2553	5224	3076		
Indep. Reflns.	2206	1138	2768	1429		
Params.	154	109	271	129		
R _{int.}	0.0348	0.0285	0.0270	0.0329		
S	1.105	1.264	1.053	1.099		
R1 (I>2o(I)) ^a	0.0407	0.0940	0.0255	0.0329		
R1 (all data)	0.0437	0.0844	0.0260	0.0359		
wR2 (I> $2\sigma(I)$) ^a	0.1046	0.1869	0.0620	0.0726		
wR2 (all data)	0.1110	0.1878	0.0623	0.0742		
Largest diff. peak /hole (e Å ⁻ ³)	0.33/-0.35	0.39/-0.66	0.22/-0.28	0.56/-0.57		
CCDC	2078900	2078909	2078919	2078913		
${}^{a}R1 = \Sigma F_{o} - F_{c} / \Sigma F_{o} . \ wR2 = [\Sigma w (F_{o}{}^{2} - F_{c}{}^{2})^{2} / \Sigma w (F_{o}{}^{2})^{2}]^{1/2}; \ w = 1 / [\sigma^{2}(F_{o}{}^{2}) + (xP)^{2} + yP].$						

Table S1 X-ray crystallographic data for compounds 1, 2, 3, 4

	5	6	7			
Formula	$C_3H_{10}MgN_8O_6$	$C_3H_{10}CaN_8O_6$	$C_3H_{10}N_8O_6Sr$			
M (g mol ⁻¹)	278.50	294.27	341.81			
Cryst. Syst.	orthorhombic	orthorhombic	orthorhombic			
Space group	Pbcn	$P2_{1}2_{1}2_{1}$	$P2_{1}2_{1}2_{1}$			
a (Å)	11.890(2)	6.5466(13)	6.6230(13)			
b (Å)	16.700(3)	12.561(3)	12.839(3)			
c (Å)	11.228(2)	13.047(3)	13.282(3)			
α (°)	90	90	90			
β (°)	90	90	90			
γ (°)	90	90	90			
V (Å ³)	2229.4(8)	1072.9(4)	1129.3(4)			
Ζ	8	4	4			
ρ (g cm ⁻³)	1.659	1.822	2.010			
T (K)	153.15	153.15	153.15			
F (000)	1152.0	332.0	680.0			
μ (mm ⁻¹)	0.200	0.627	4.814			
Reflns. collected	5924	4309	4560			
Indep. Reflns.	2515	2250	2529			
Params.	164	163	163			
R _{int.}	0.0439	0.0289	0.0364			
S	1.116	1.049	1.073			
R1 (I>2σ(I)) ^a	0.0589	0.0354	0.0312			
R1 (all data)	0.0687	0.0362	0.0320			
wR2 (I> $2\sigma(I)$) ^a	0.1288	0.0823	0.0726			
wR2 (all data)	0.1363	0.0831	0.0732			
Largest diff. peak /hole (e Å ⁻³)	0.36/-0.35	0.26/-0.26	0.34/-0.96			
CCDC	2078915	2078916	2078917			
${}^{a}R1 = \Sigma F_{o} - F_{c} / \Sigma F_{o} . wR2 = [\Sigma w(F_{o}{}^{2} - F_{c}{}^{2})^{2} / \Sigma w(F_{o}{}^{2})^{2}]^{1/2}; w = 1 / [\sigma^{2}(F_{o}{}^{2}) + (xP)^{2} + yP].$						

 Table S2 X-ray crystallographic data for compounds 5, 6, 7

compound	The first exothermic decomposition peak temperatures (°C)					
	5°C·min ⁻¹	10°C·min ⁻¹	15°C·min ⁻¹	20°C·min ⁻¹		
1	252.3	254.2	256.1	257.5		
2	258.8	260.2	261.4	262.8		
3	223.5	225.2	227.3	229.0		
4	242.3	244.1	246.2	248.0		
5	220.1	221.9	223.5	225.0		
6	200.6	202.3	203.8	205.2		
7	201.7	203.8	208.4	210.6		

Table S3. The first exothermic decomposition peak temperatures tested at different heating rates of compounds 1 to 7.

Figure S1. DSC curve of 1.



Figure S2. DSC curve of 2.



Figure S3. DSC curve of 3.



Figure S4. DSC curve of 4.



Figure S5. DSC curve of 5.



Figure S6. DSC curve of 6.



Figure S7. DSC curve of 7.



Figure S8. TG-DTG curve of 1.



Figure S9. TG-DTG curve of 2.



Figure S10. TG-DTG curve of 5.



Figure S11. TG-DTG curve of 6.



Figure S12. TG-DTG curve of 7.



Figure S13. DSC curves of mixed 1+AP, 3+AP, 4+AP and 7+AP.

