

## **Preparation of magnetic silica supported Brønsted acidic ionic liquids for the depolymerization of lignin to aromatic monomers**

Mingyu Cui<sup>‡a</sup>, Dingkai Wang<sup>‡a</sup>, Yanjun Li<sup>a,b</sup>, Wei Zhao<sup>a\*</sup>, Chong Liang<sup>a</sup>, Xutang Liu<sup>a</sup>, Shuiyuan Fu<sup>a</sup>, Luyao

Wang<sup>a</sup>, Xianyong Wei<sup>a</sup>

<sup>a</sup> Jiangsu Province Engineering Research Center of Fine Utilization of Carbon Resources, China University of Mining & Technology, Xuzhou 221116, Jiangsu, China

<sup>b</sup> Shaanxi Key Laboratory of Low Metamorphic Coal Clean Utilization, School of Chemistry and Chemical Engineering, Yulin University, Yulin 719000, Shaanxi, China

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\* Corresponding author

E-mail address: [zhaow1965@163.com](mailto:zhaow1965@163.com).

‡ These authors contributed equally to this work.

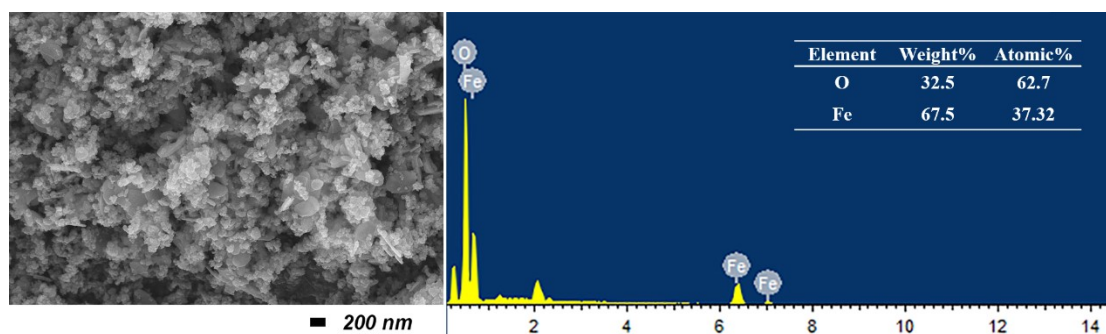


Fig. S1 SEM-EDS image of  $\text{Fe}_3\text{O}_4$ .

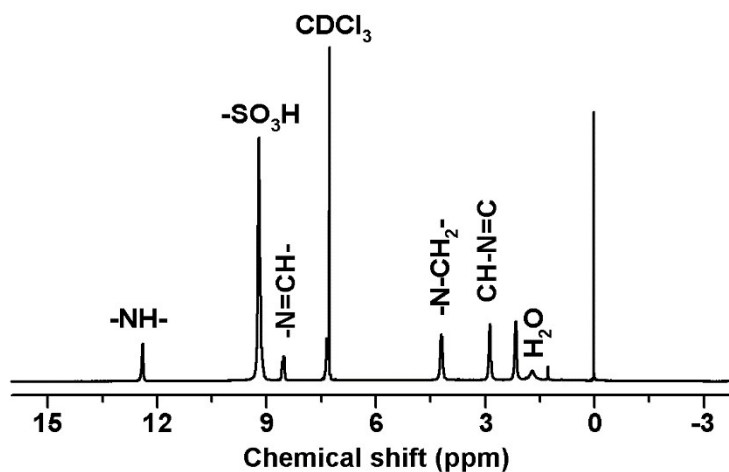


Fig. S2  $^1\text{H}$ -NMR spectra of BAIL. <sup>a</sup>

<sup>a</sup>  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ),  $\delta$  12.40 (s, 1H), 9.22 (s, 9H), 8.95–8.23 (m, 1H), 7.32 (d,  $J = 41.5$  Hz, 3H), 4.21 (s, 2H), 4.00 – 3.64 (m, 1H), 2.87 (s, 2H), 2.12 (t,  $J = 28.9$  Hz, 2H), 1.79 (d,  $J = 94.2$  Hz, 1H), 1.44 – 1.11 (m, 1H), 0.02 (s, 1H).

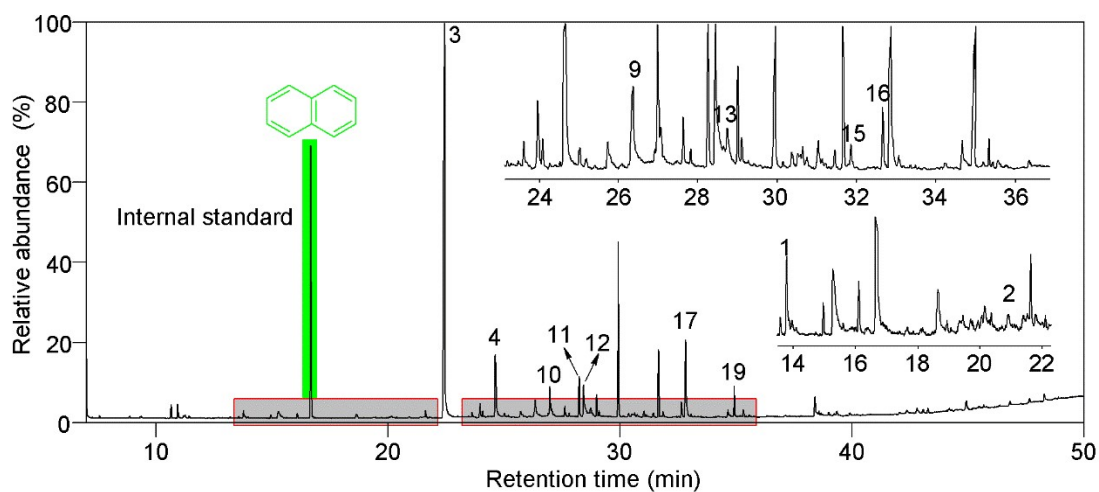
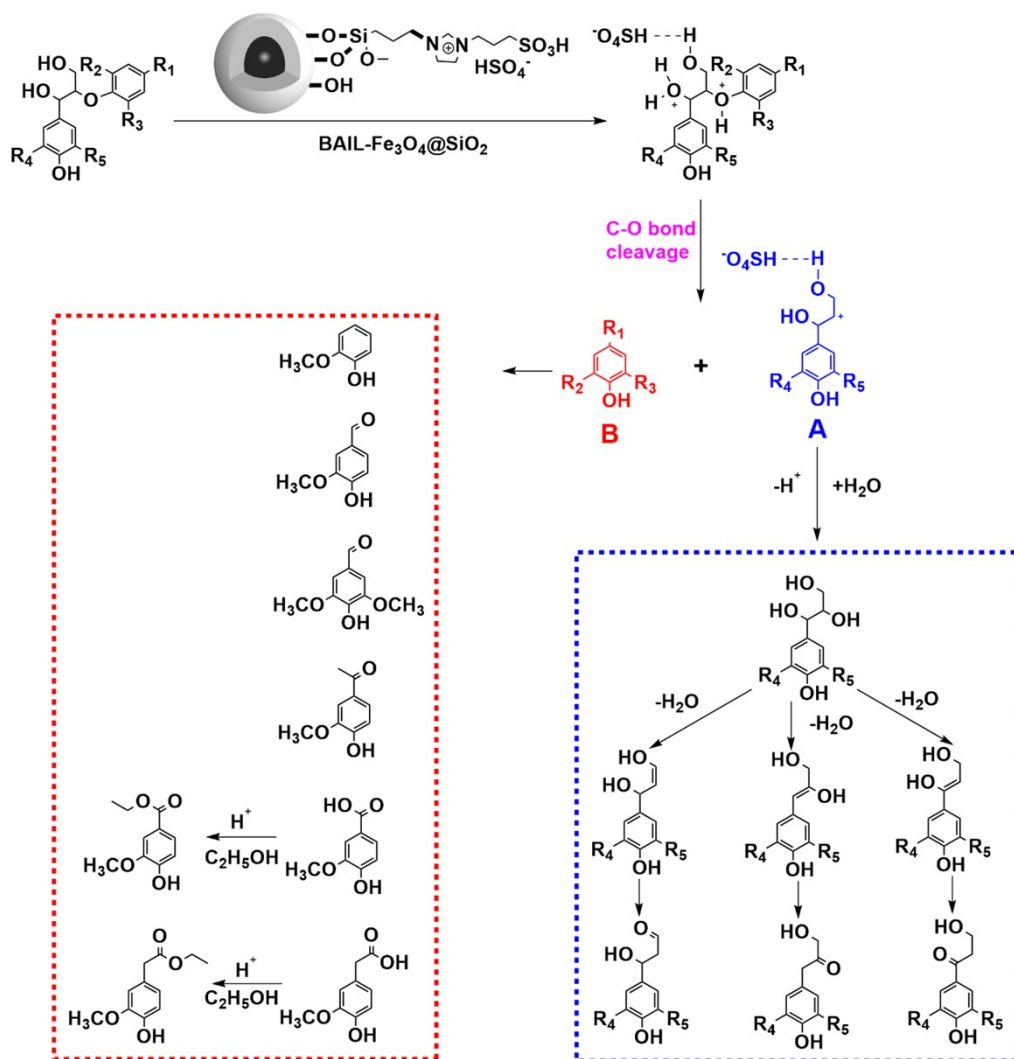


Fig. S3 Total ions chromatogram of liquid products from lignin depolymerization over 40 wt% BAIL (30wt%)- $\text{Fe}_3\text{O}_4$ @ $\text{SiO}_2$  at 160  $^\circ\text{C}$  for 3 h. (Identified by GC/MS, naphthalene as the internal standard)



**Fig. S4** Possible mechanism proposed for the catalytic depolymerization of lignin over 40 wt% BAIL (30wt%)-Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub>.

**Table S1** Lignin depolymerization products. <sup>a</sup> (Identified by GC-MS)

Classification	Entry <i>b</i>	Compounds	Formula	Yield of main products (%)			
				None	40 wt% Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub>	20 wt% BAIL	40 wt% BAIL (30 wt%)- Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub>
Phenol	1	2-Methoxyphenol	C <sub>7</sub> H <sub>8</sub> O <sub>2</sub>	0.27	0.17	0.7	0.33
	6	2,6-Di-tert-butylphenol	C <sub>14</sub> H <sub>22</sub> O		0.14		
	2	4-Hydroxybenzaldehyde	C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>		0.15		
Aryl aldehyde	3	Vanillin	C <sub>8</sub> H <sub>8</sub> O <sub>3</sub>	11.9 3	12.96	9.38	18.42
	13	4-Hydroxy-3,5-dimethoxybenzaldehyde	C <sub>9</sub> H <sub>10</sub> O <sub>4</sub>	0.13	0.22		0.75
	7	Methyl 2-(4-hydroxy-3-methoxyphenyl) acetate	C <sub>10</sub> H <sub>12</sub> O <sub>4</sub>			0.8	
Aryl ester	10	Ethyl 4-hydroxy-3-methoxybenzoate	C <sub>10</sub> H <sub>12</sub> O <sub>4</sub>	0.21	0.40	7.97	1.11
	11	Ethyl homovanillate	C <sub>11</sub> H <sub>14</sub> O <sub>4</sub>	0.10	0.18	0.93	1.33
	16	2-Methoxy-4-(2-oxopropyl) phenyl acetate	C <sub>12</sub> H <sub>14</sub> O <sub>4</sub>				0.54
	18	Butyl (4-methylpentyl) phthalate	C <sub>18</sub> H <sub>26</sub> O <sub>4</sub>			0.50	
Aryl ketone	19	Dibutyl phthalate	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>		0.09		0.95
	4	Apocynin	C <sub>9</sub> H <sub>10</sub> O <sub>3</sub>	0.95	1.48	2.15	3.06
	17	2-Hydroxy-1-(4-hydroxy-3-methoxyphenyl) ethan-1-one	C <sub>9</sub> H <sub>10</sub> O <sub>4</sub>				3.18
Aryl acid	9	4-Hydroxy-3-methoxybenzoic acid	C <sub>8</sub> H <sub>8</sub> O <sub>4</sub>	0.46	1.01		1.20
	12	Homovanillic acid	C <sub>9</sub> H <sub>10</sub> O <sub>4</sub>	0.29	0.73		1.91
Aryl alcohol	8	Homovanillyl alcohol	C <sub>9</sub> H <sub>12</sub> O <sub>3</sub>		0.12		
Heteroatoms-containing aromatic monomers	5	S-methyl 4-(methylthio) benzothioate	C <sub>9</sub> H <sub>10</sub> OS <sub>2</sub>		0.13		
	14	4-(Methylphenylthio) acetaldehyde diethyl acetal	C <sub>13</sub> H <sub>20</sub> O <sub>2</sub> S	0.13			
	15	Ethamivan	C <sub>12</sub> H <sub>17</sub> NO <sub>3</sub>				0.25

<sup>a</sup> Reaction conditions: 160 °C, 3 h, 0.5 g of lignin, 30 mL of ethanol, and 1 MPa of N<sub>2</sub>.<sup>b</sup> Relative to Fig. S3.

**Table. S2** Mass distribution and DBE distribution of the species identified with Q-Exactive Orbitrap MS in the ethanol-soluble products.

Catalyst	Mass distribution of the species identified <sup>a</sup>				DBE distribution of the species identified <sup>b</sup>			Total
	<150	150~250	250~350	350~450	1-3	4-7	≥8	
None	19	41	39	8	20	56	31	107
BAIL	9	30	17	5	21	30	10	61
Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub>	13	31	21	7	27	33	12	72
BAIL (30 wt%)- Fe <sub>3</sub> O <sub>4</sub> @SiO <sub>2</sub>	21	43	42	14	14	63	43	120

<sup>a</sup> Number of the species identified with Q-Exactive Orbitrap MS in different molecular mass ranges.

<sup>b</sup> Number of the species identified with Q-Exactive Orbitrap MS in different DBE ranges.