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Supporting Information

SBA-15 obtained from rice husk ashes wet-impregnated with metals

(Al, Co, Ni) as efficient catalysts for 1,4-dihydropyridine three-component

reaction

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Figure S1. Batch reaction system utilized in this study. Batch reactors: before reaction (left) and after reaction (right).



Figure S2. Wide-Angle XRD diffractogram of the catalysts prepared.



Figure S3. FT-IR spectra of the catalysts prepared.



Figure S4. $\ensuremath{\mathsf{NH}}_3\ensuremath{\mathsf{-TPD}}$ analysis of the catalysts prepared.



Figure S5. TEM micrographs of the catalysts prepared.



Figure S6. ¹H NMR (CDCl₃, 200 MHz) of dimethyl 1,4-dihydro-2,6-dimethyl-4-phenylpyridine-3,5-dicarboxylate.



Figure S7. FT-IR spectrum of dimethyl 1,4-dihydro-2,6-dimethyl-4-phenylpyridine-3,5-dicarboxylate.



Figure S8. ¹H NMR (CDCl₃, 200 MHz) of dimethyl 1,4-dihydro-2,6-dimethyl-4-(3-nitrophenyl)-pyridine-3,5-dicarboxylate.



Figure S9. FT-IR spectrum of dimethyl 1,4-dihydro-2,6-dimethyl-4-(3-nitrophenyl)-pyridine-3,5-dicarboxylate.



Figure S10. ¹H NMR (CDCl₃, 200 MHz) of dimethyl 1,4-dihydro-2,6-dimethyl-4-(4-methoxyphenyl)-pyridine-3,5-dicarboxylate.



Figure S11. FT-IR spectrum of dimethyl 1,4-dihydro-2,6-dimethyl-4-(4-methoxyphenyl)-pyridine-3,5-dicarboxylate.



Figure S12. ¹H NMR (CDCl₃, 200 MHz) of dimethyl 2,6-dimethyl-4-(4-(methylthio)phenyl)-1,4-dihydropyridine-3,5-dicarboxylate.



Figure S13. FT-IR spectrum of dimethyl 2,6-dimethyl-4-(4-(methylthio)phenyl)-1,4-dihydropyridine-3,5-dicarboxylate.



Figure S14. ¹H NMR (CDCl₃, 200 MHz) of dimethyl 4-(2-fluorophenyl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate.



Figure S15. FT-IR spectrum of dimethyl 4-(2-fluorophenyl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate.



Figure S16. ¹H NMR (dmso-*d6*, 200 MHz) of dimethyl 4-(1*H*-imidazol-2-yl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate.



Figure S17. FT-IR spectrum of dimethyl 4-(1*H*-imidazol-2-yl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylate.