

Electronic Supplementary Information (ESI)

Molecular Conformational Twist Dependent Wide Fluorescence Tuning and White Light Emission in a Single Fluorophore via Halochormism

Parthasarathy Gayathri,^a Probal Nag,^b Neethu Anand,^b Sivarajana Reddy Vennapusa,^b Mehboobali Pannipara,^{c,d} Abdullah G. Al-Sehemi,^{c,d} Dohyun Moon^{*e} Savarimuthu Philip Anthony^{*a}

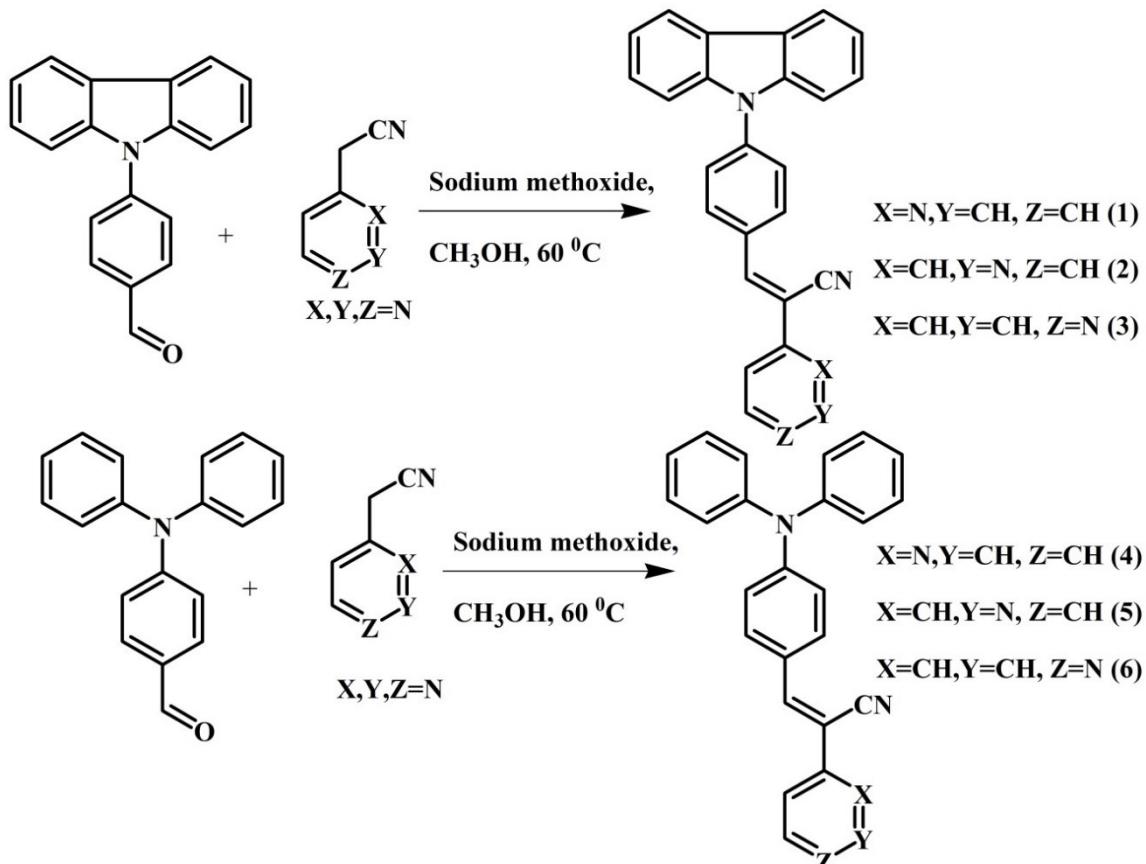
^aSchool of Chemical & Biotechnology, SASTRA Deemed University, Thanjavur-613401, Tamil Nadu, India. Fax: +914362264120; Tel: +914362264101; E-mail: philip@biotech.sastra.edu

^bSchool of chemistry, Indian Institute of Science Education and Research, Thiruvananthapuram, Kerala-695551, India

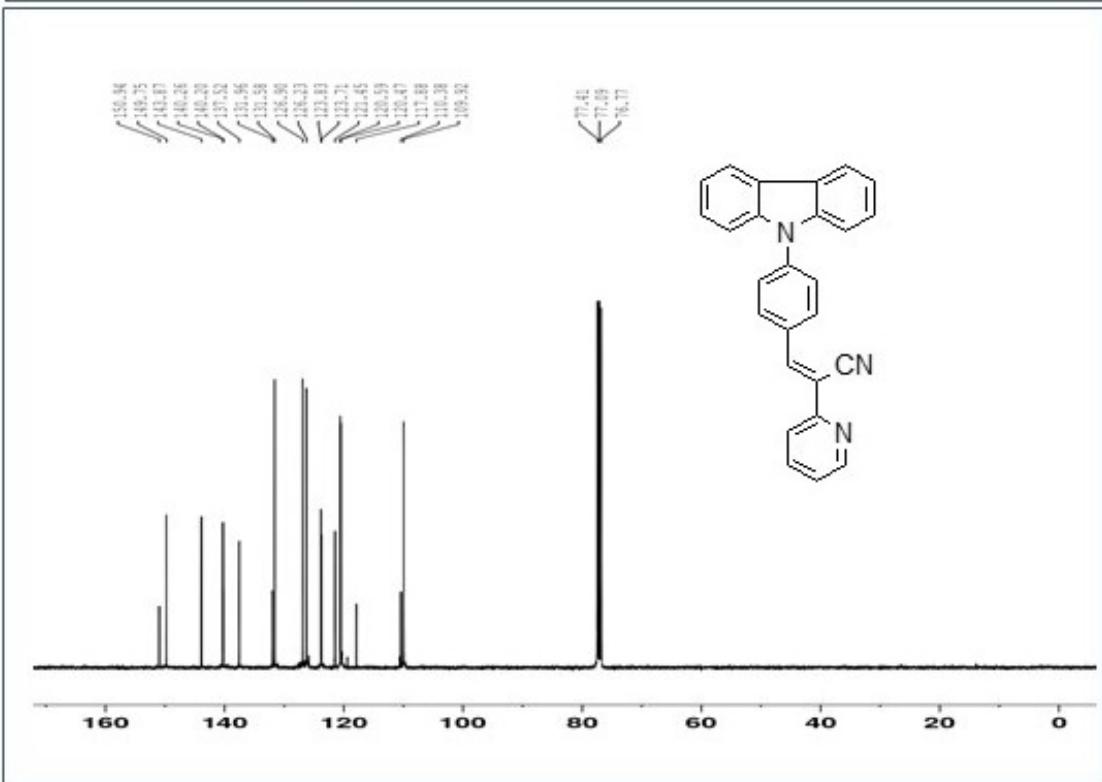
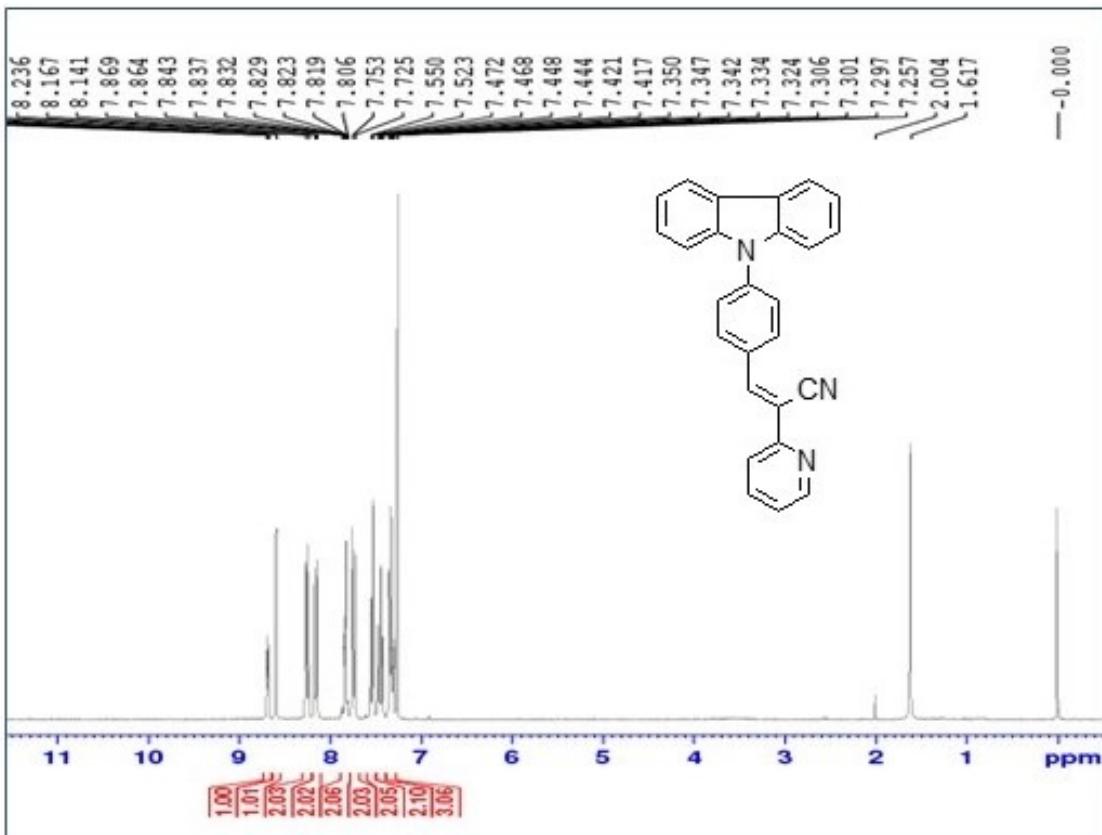
^b)Department of chemistry, King Khalid University, Abha 61413, Saudi Arabia.

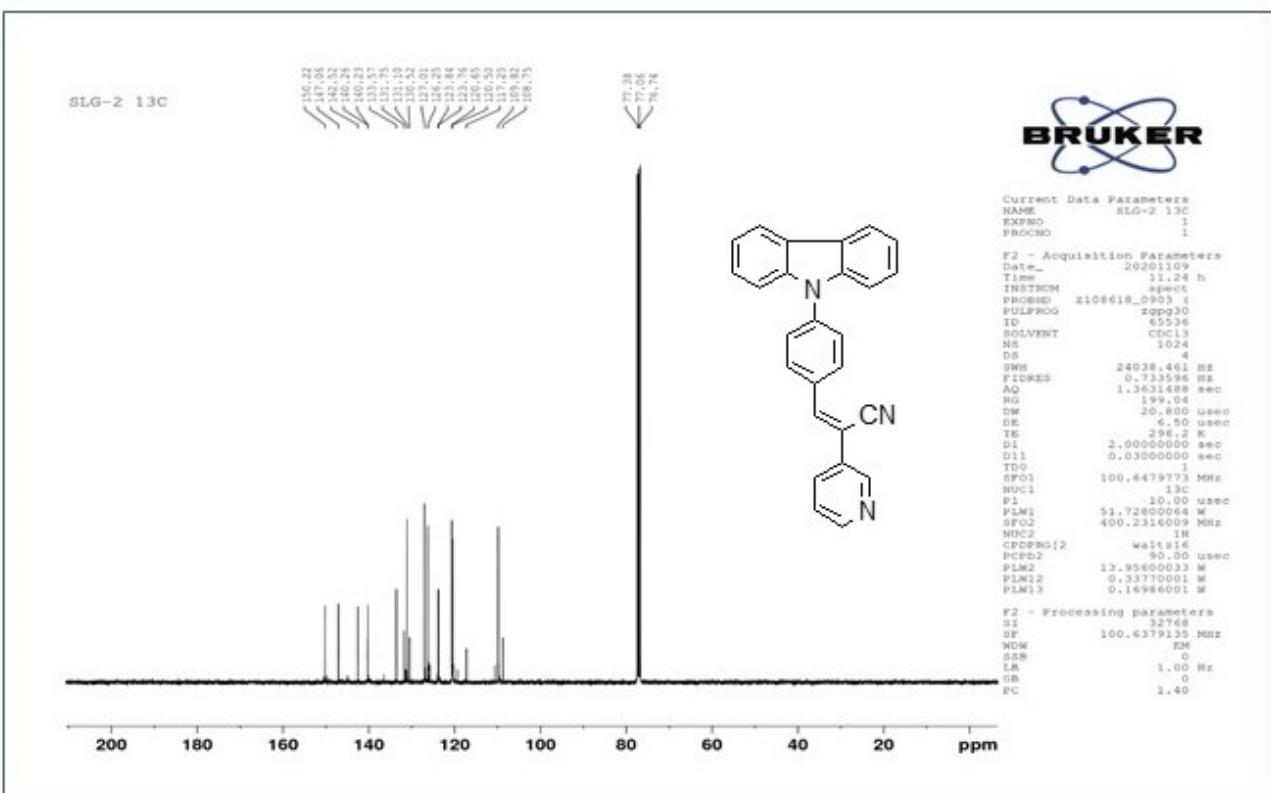
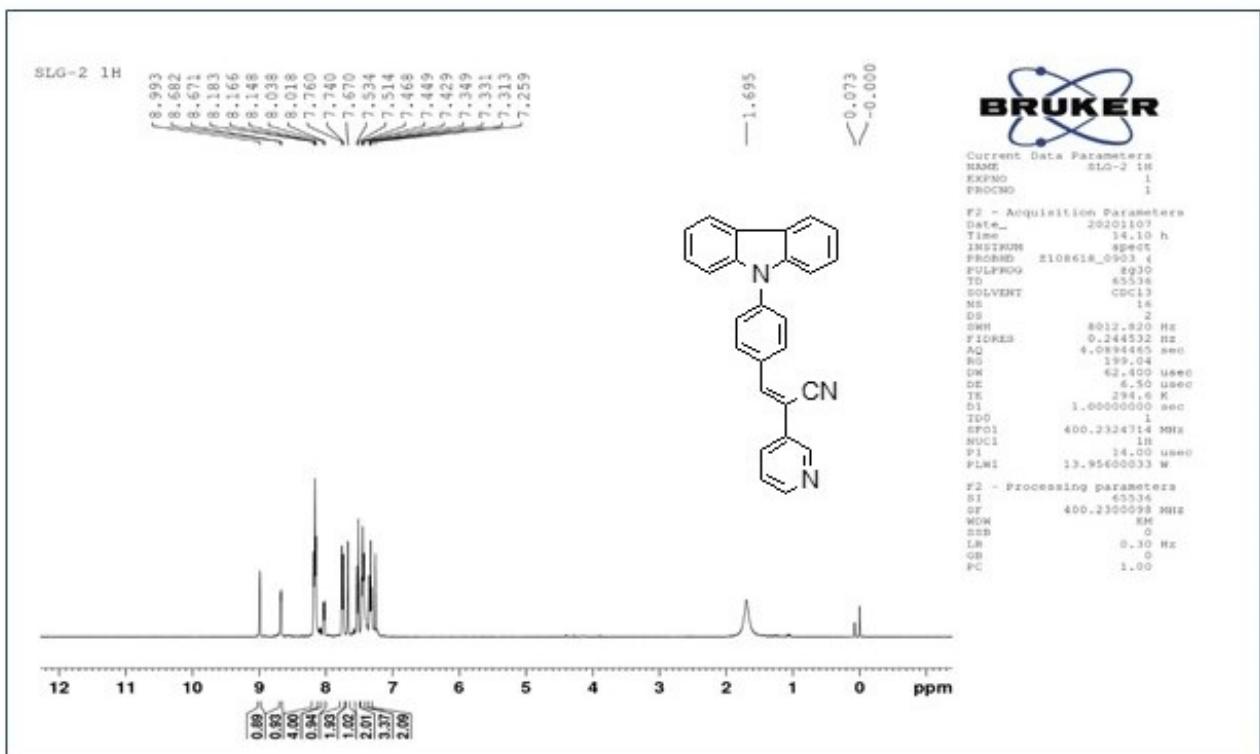
^cResearch center for Advanced Materials Science, King Khalid University, Abha 61413, Saudi Arabia.

^b)Beamline Department, Pohang Accelerator Laboratory, 80 Jigokro-127beongil, Nam-gu, Pohang, Gyeongbuk, Korea, Email: dmoon@postech.ac.kr

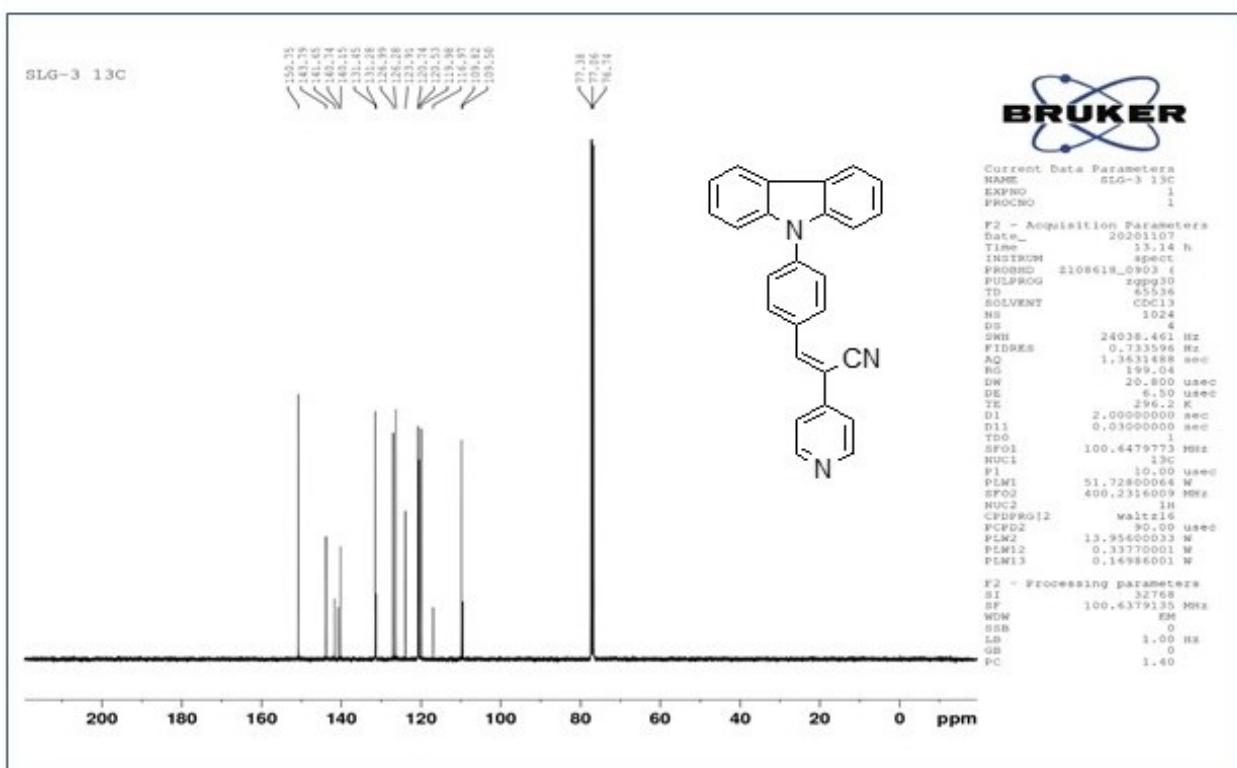
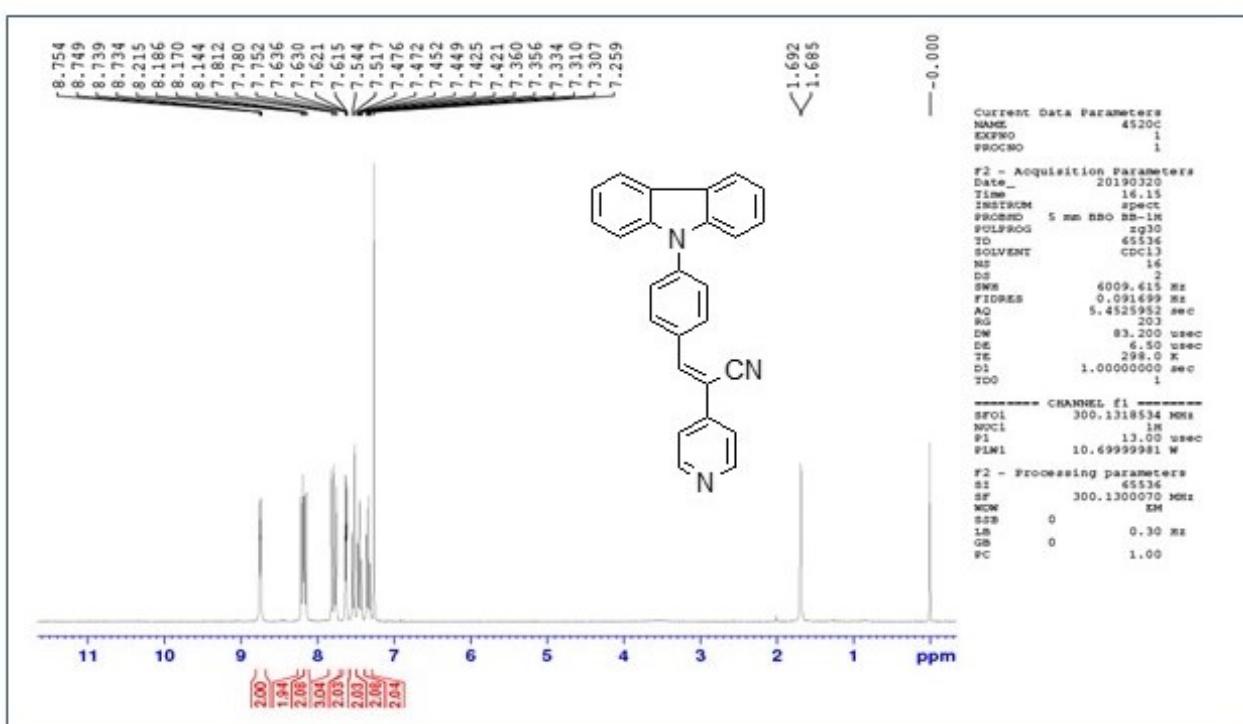


Scheme S1. Synthesis of carbazole and triphenylamine based isomers (**1-6**).

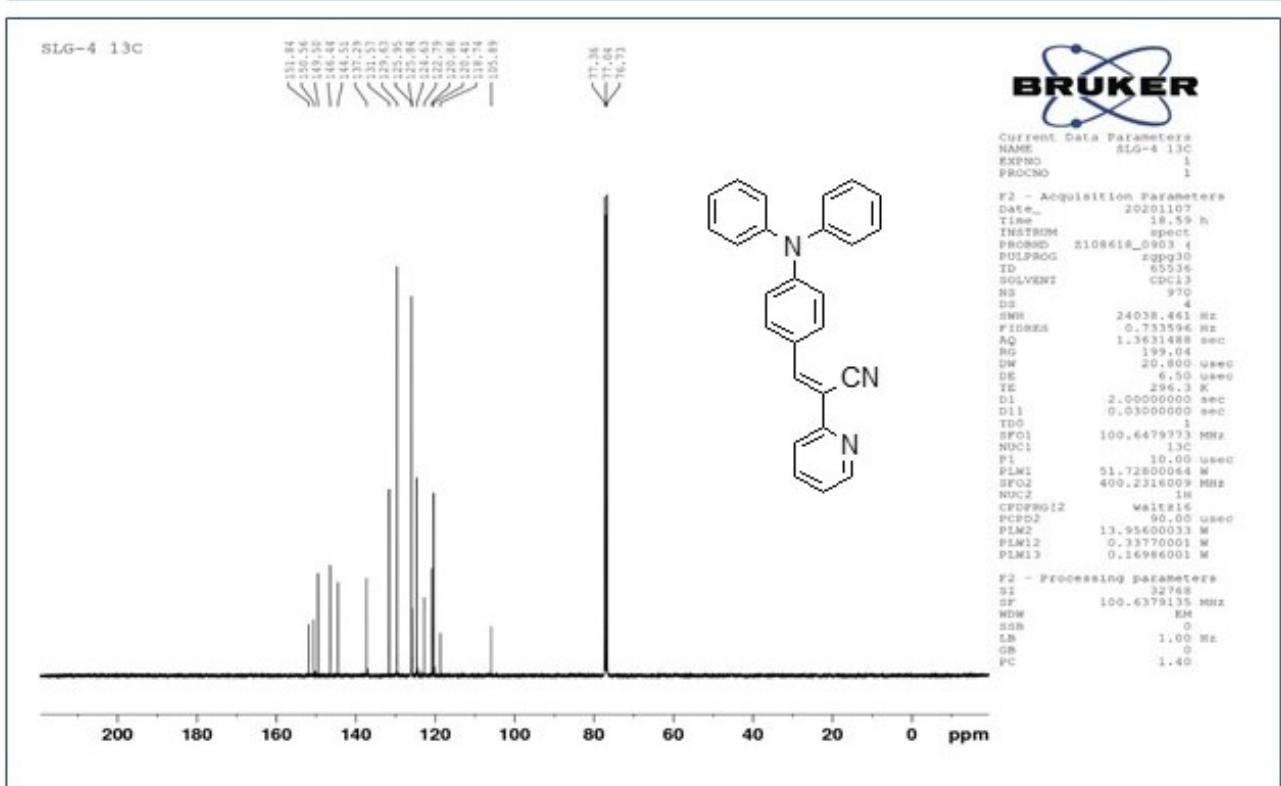
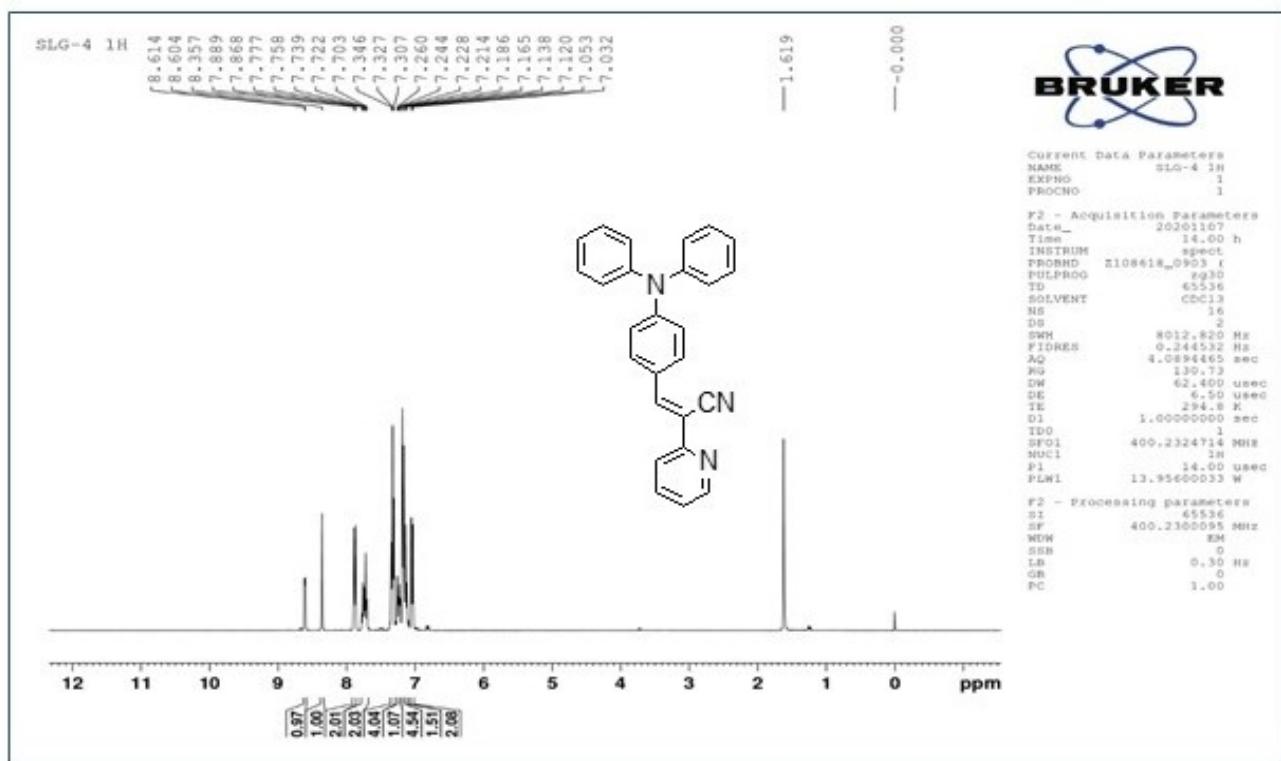




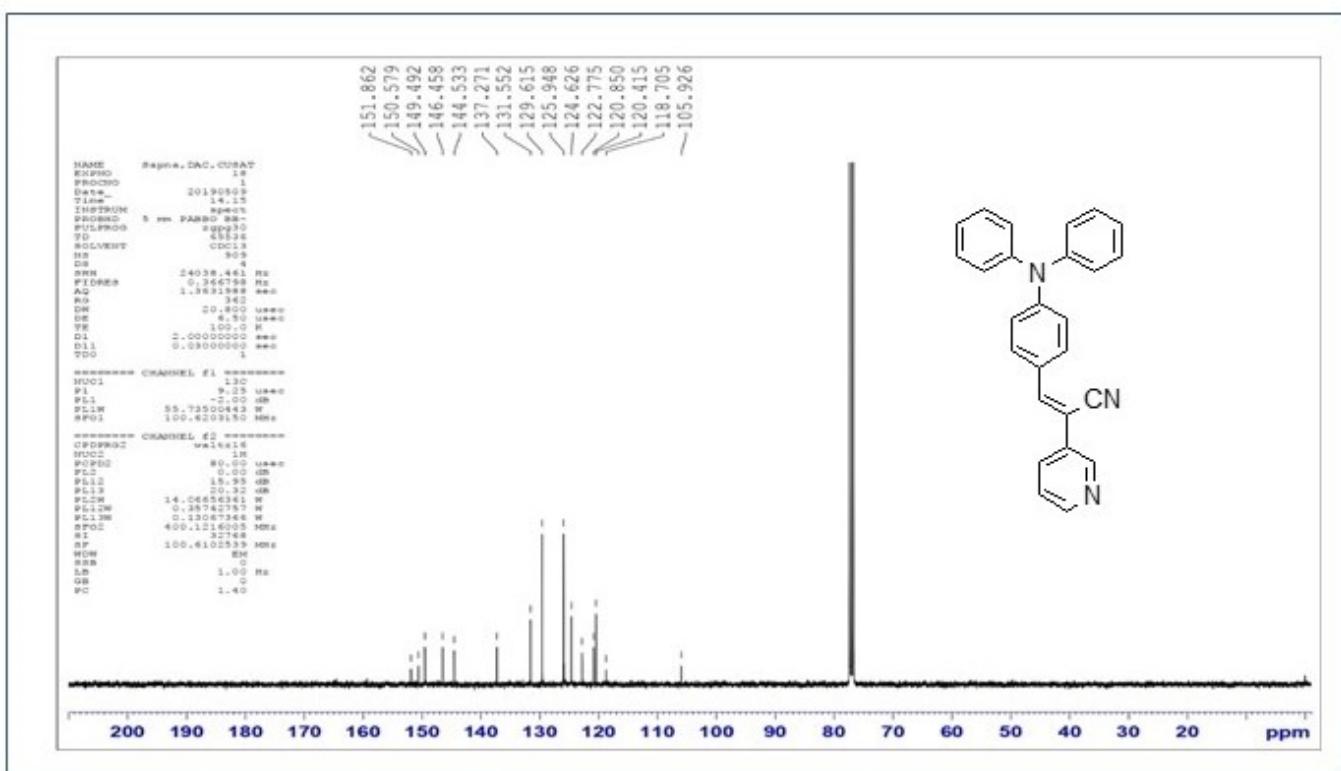
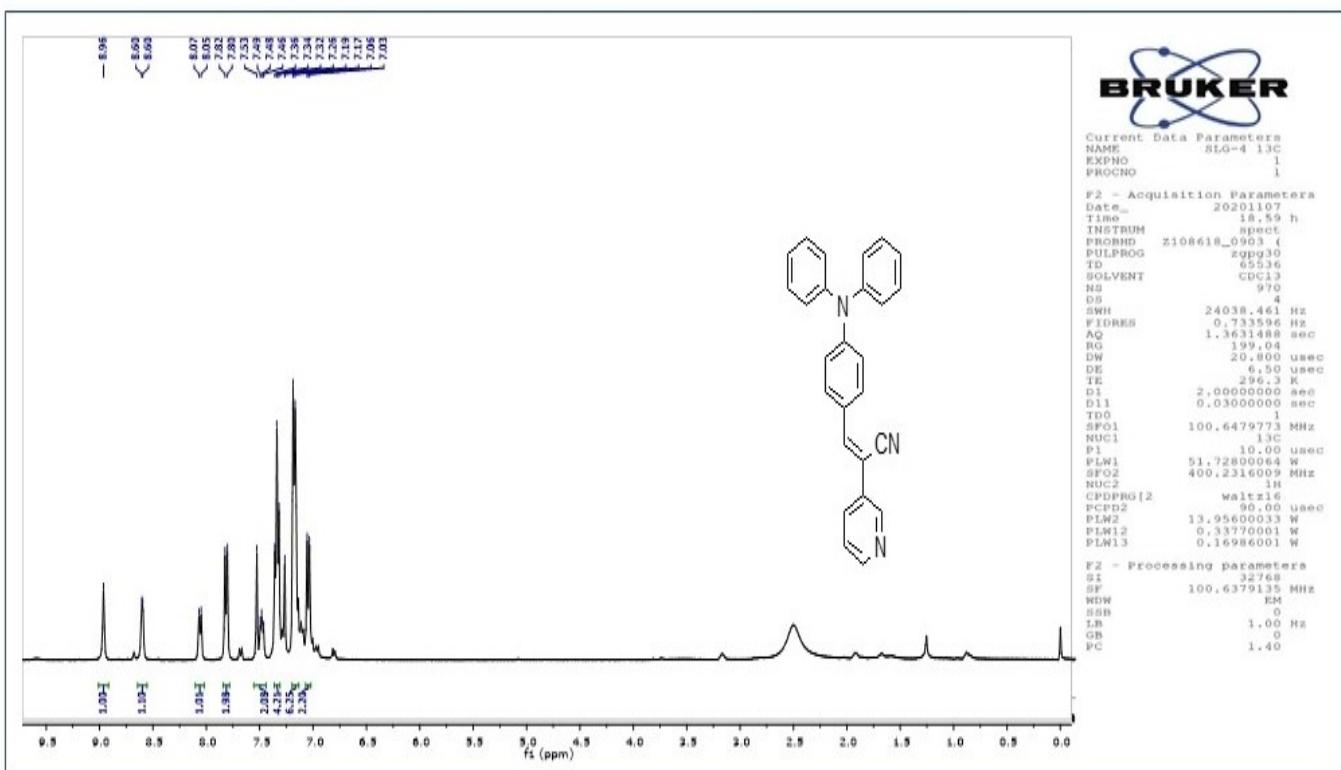
1H and ^{13}C NMR of **2**.



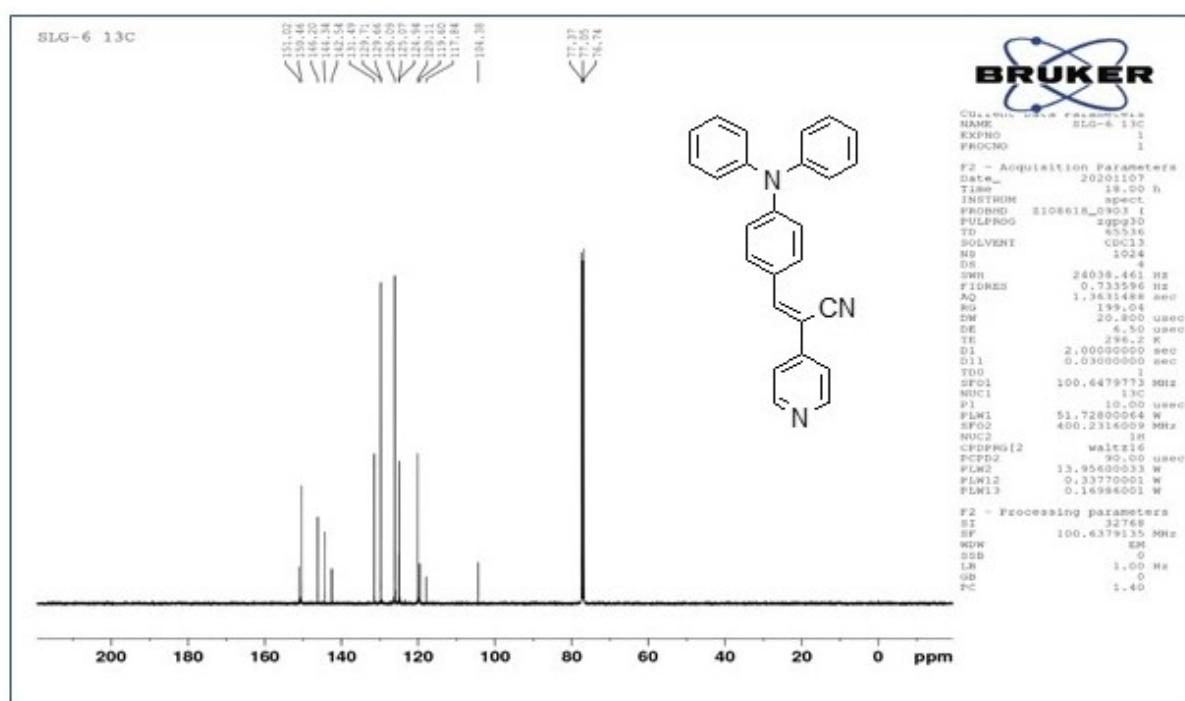
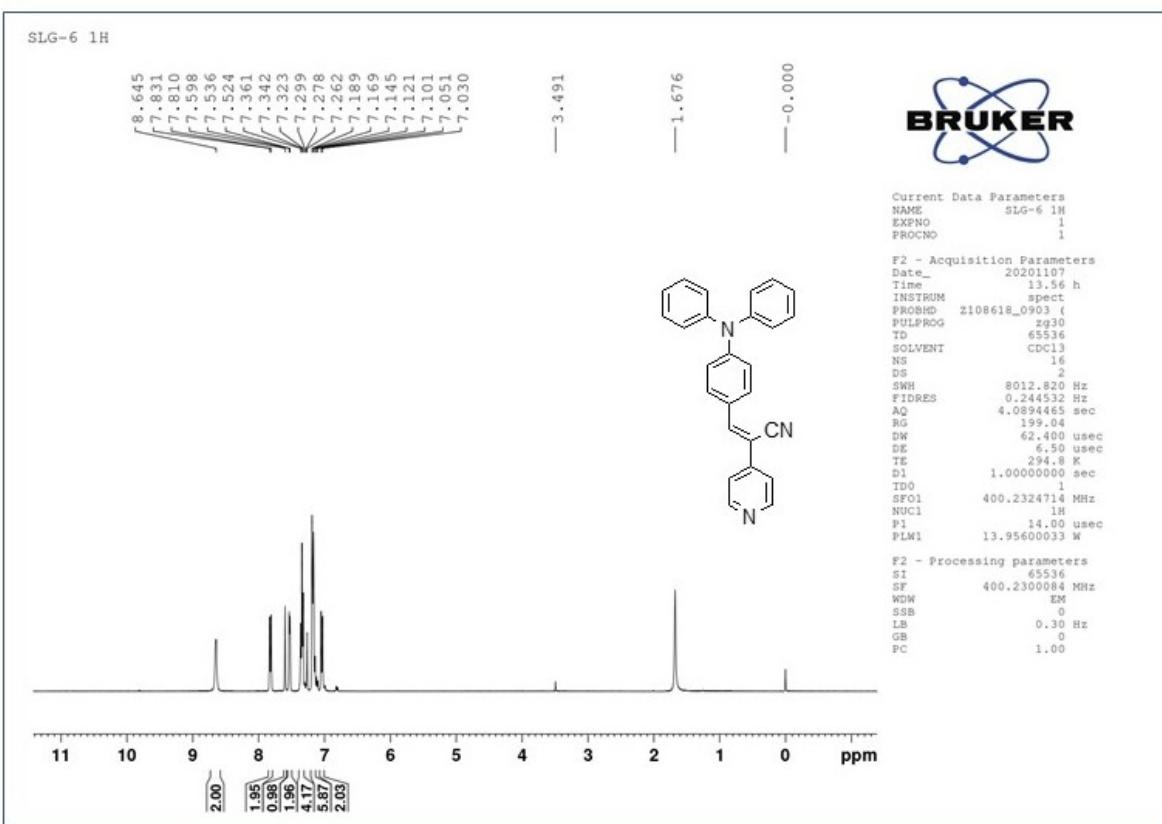
1H and ^{13}C NMR of **3**.



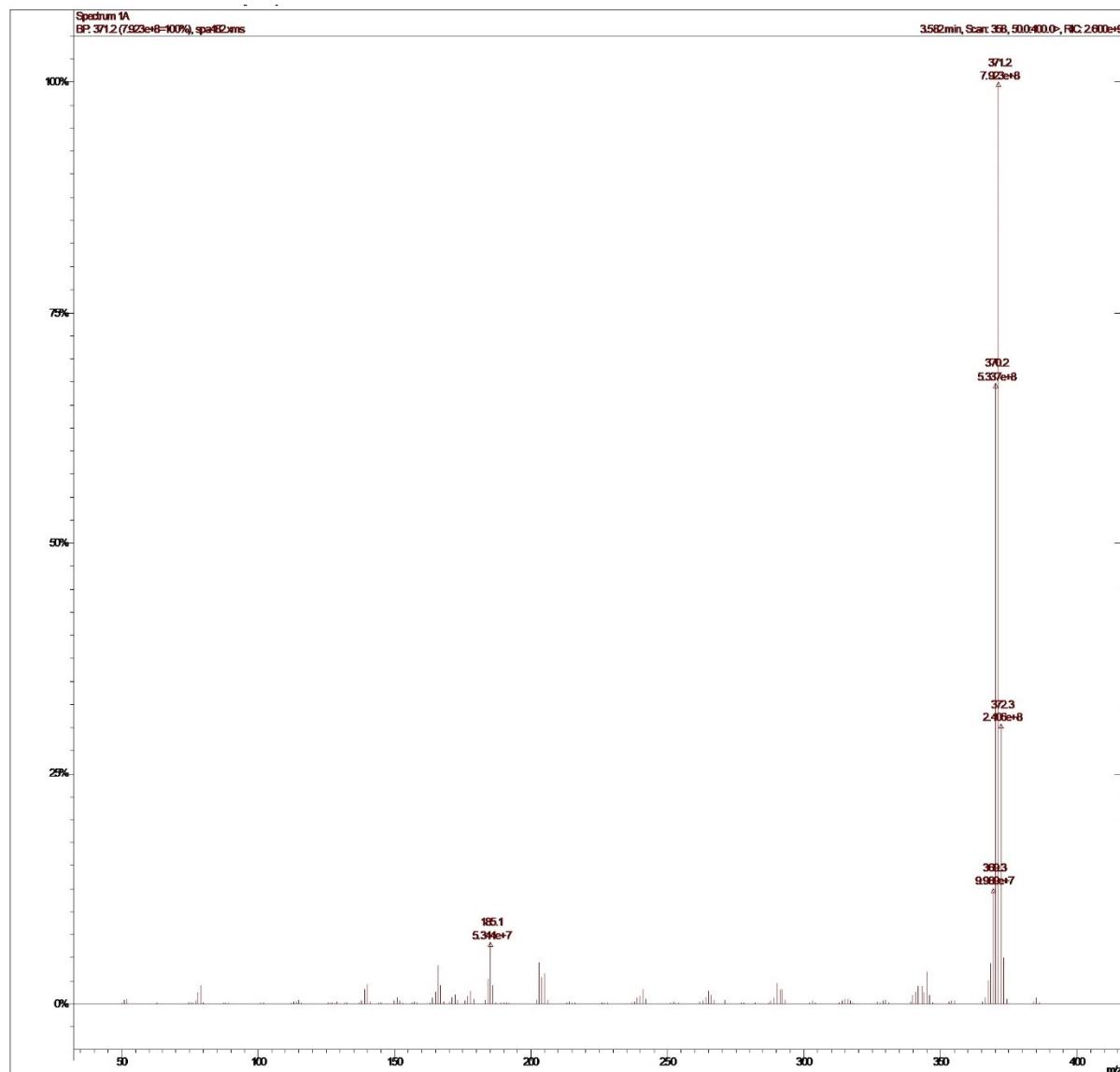
1H and ^{13}C NMR of **4**.



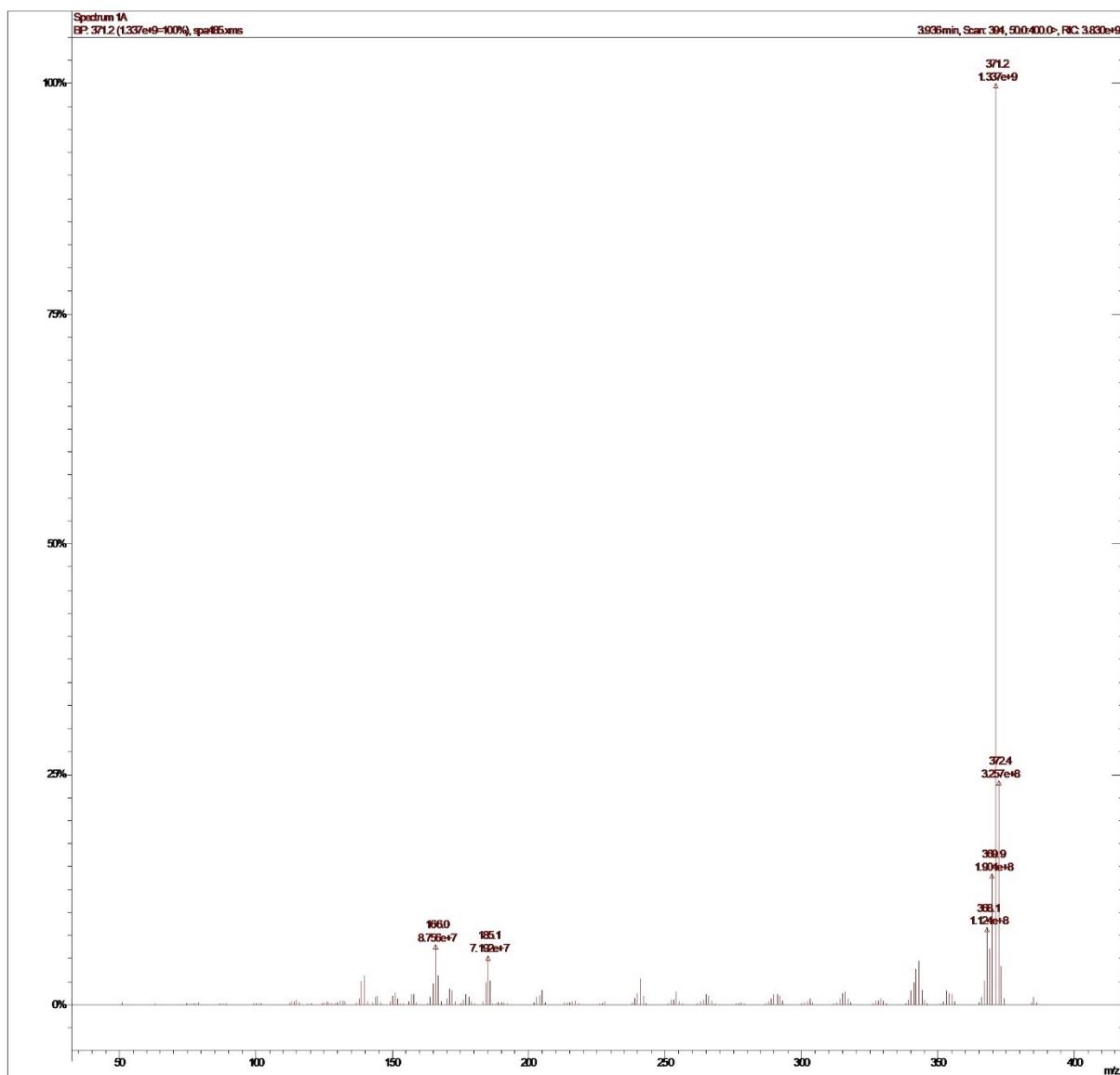
1H and ^{13}C NMR of **5**.



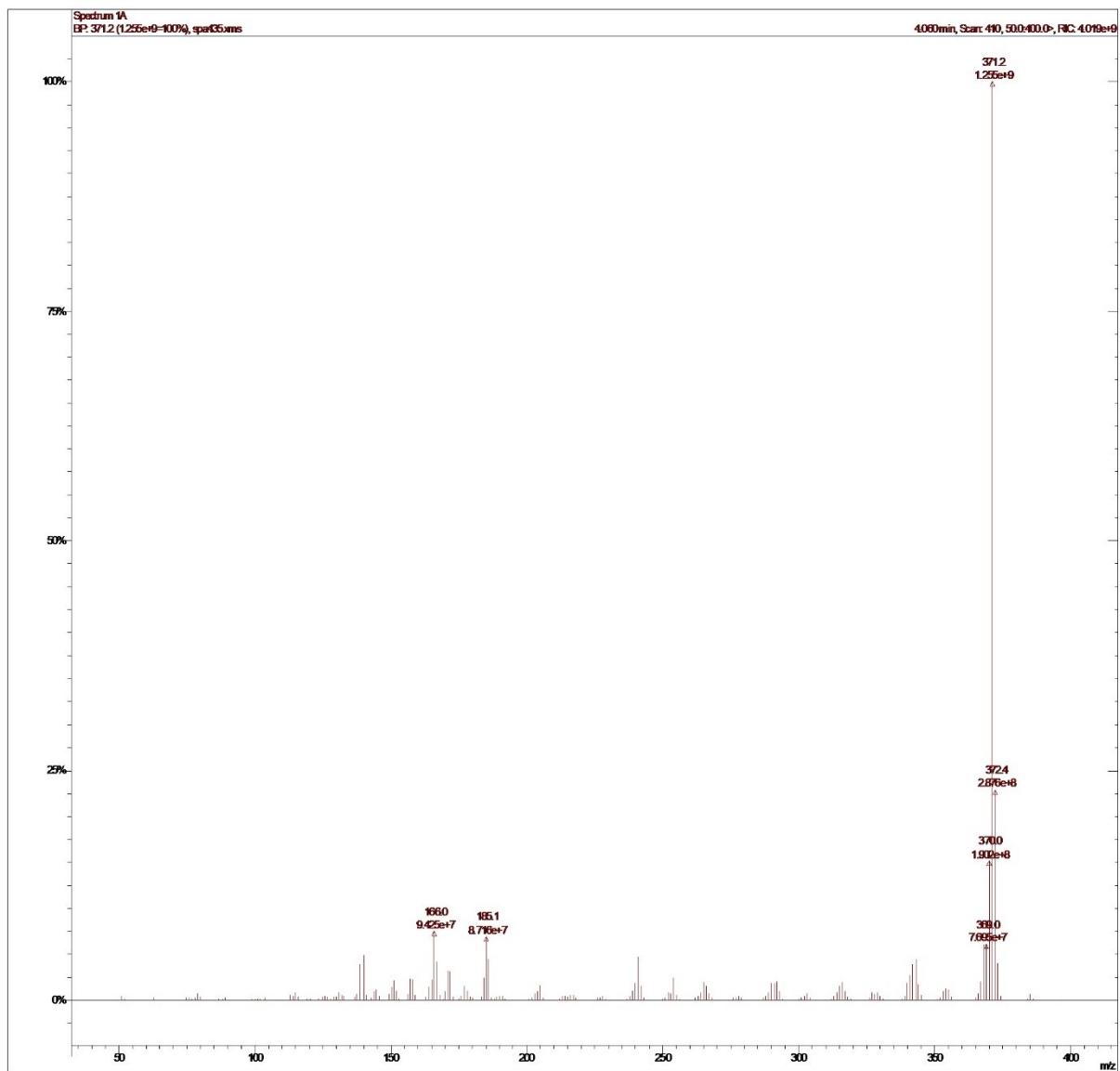
1H and ^{13}C NMR of **6**.



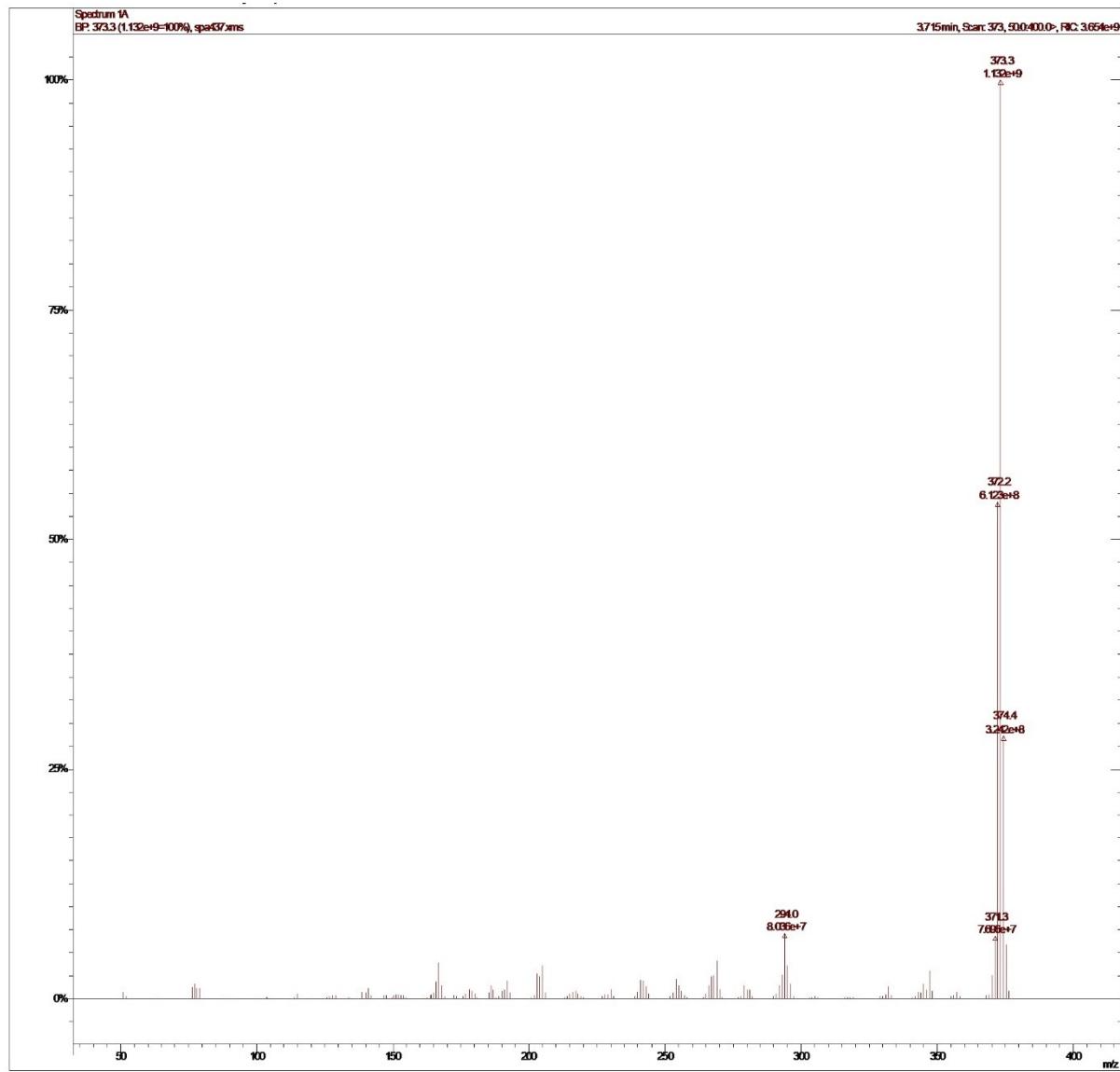
1: m/z calculated for $C_{26}H_{17}N_3$ ($M + H$): 371.14, found: 371.2



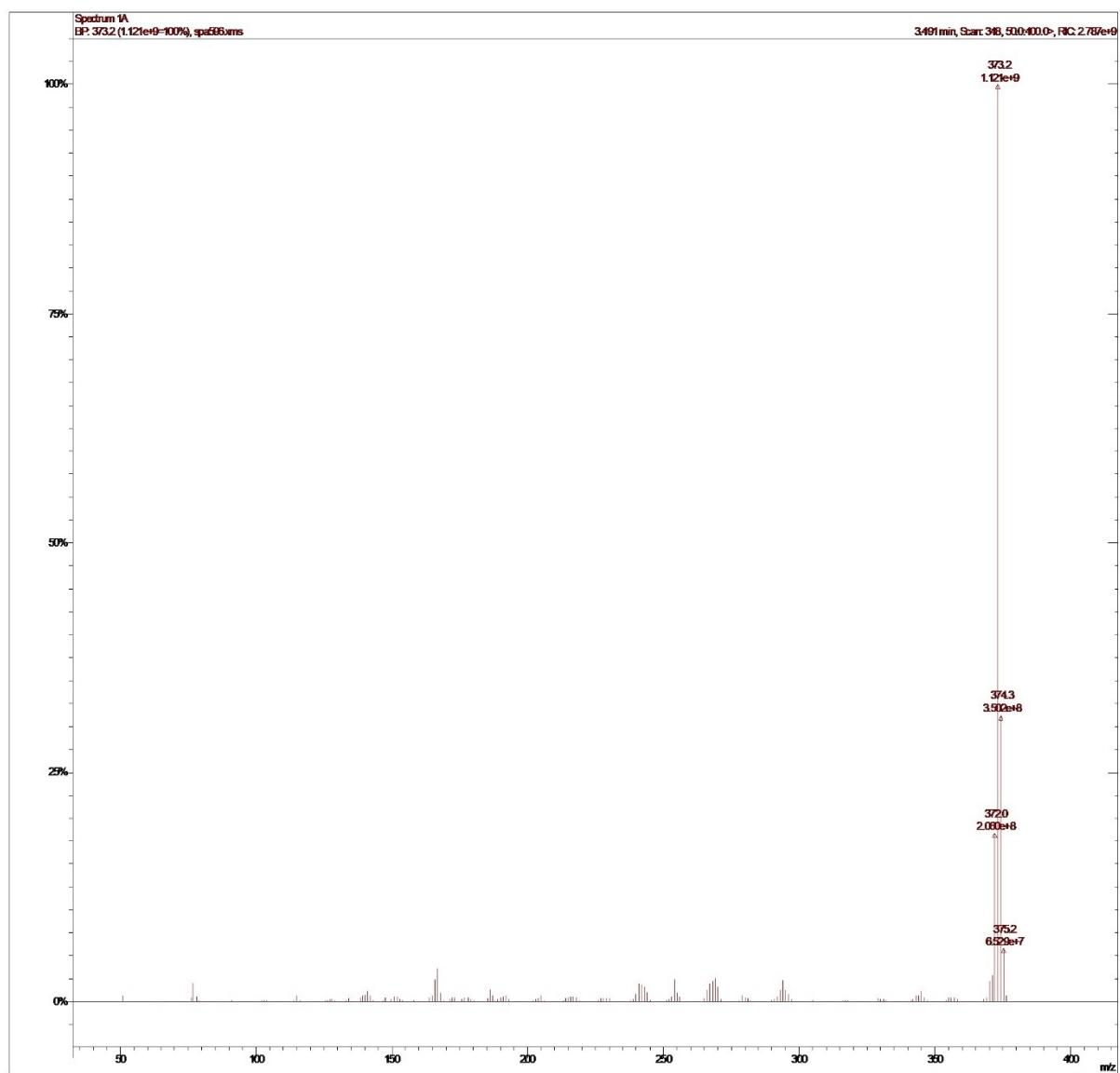
2: m/z calculated for $C_{26}H_{17}N_3$ ($M + H$): 371.14, found: 371.2



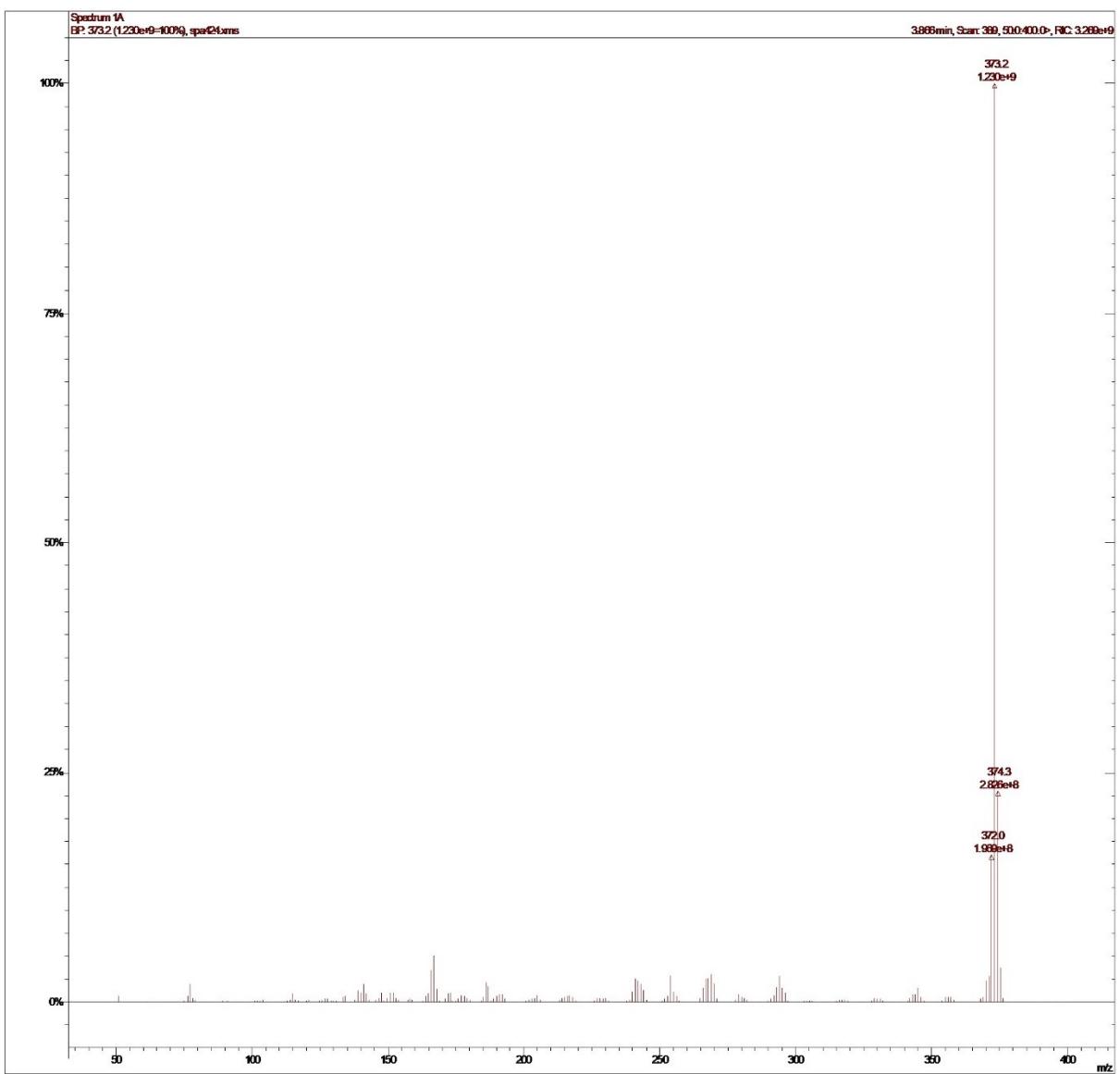
3: m/z calculated for $C_{26}H_{17}N_3$ ($M + H$): 371.14, found: 371.2



4: m/z calculated for $C_{26}H_{19}N_3$ ($M + H$): 373.16, found: 373.3



5: m/z calculated for $C_{26}H_{19}N_3$ ($M + H$): 373.16, found: 373.2



6: m/z calculated for $C_{26}H_{19}N_3$ ($M + H$): 373.16, found: 373.2

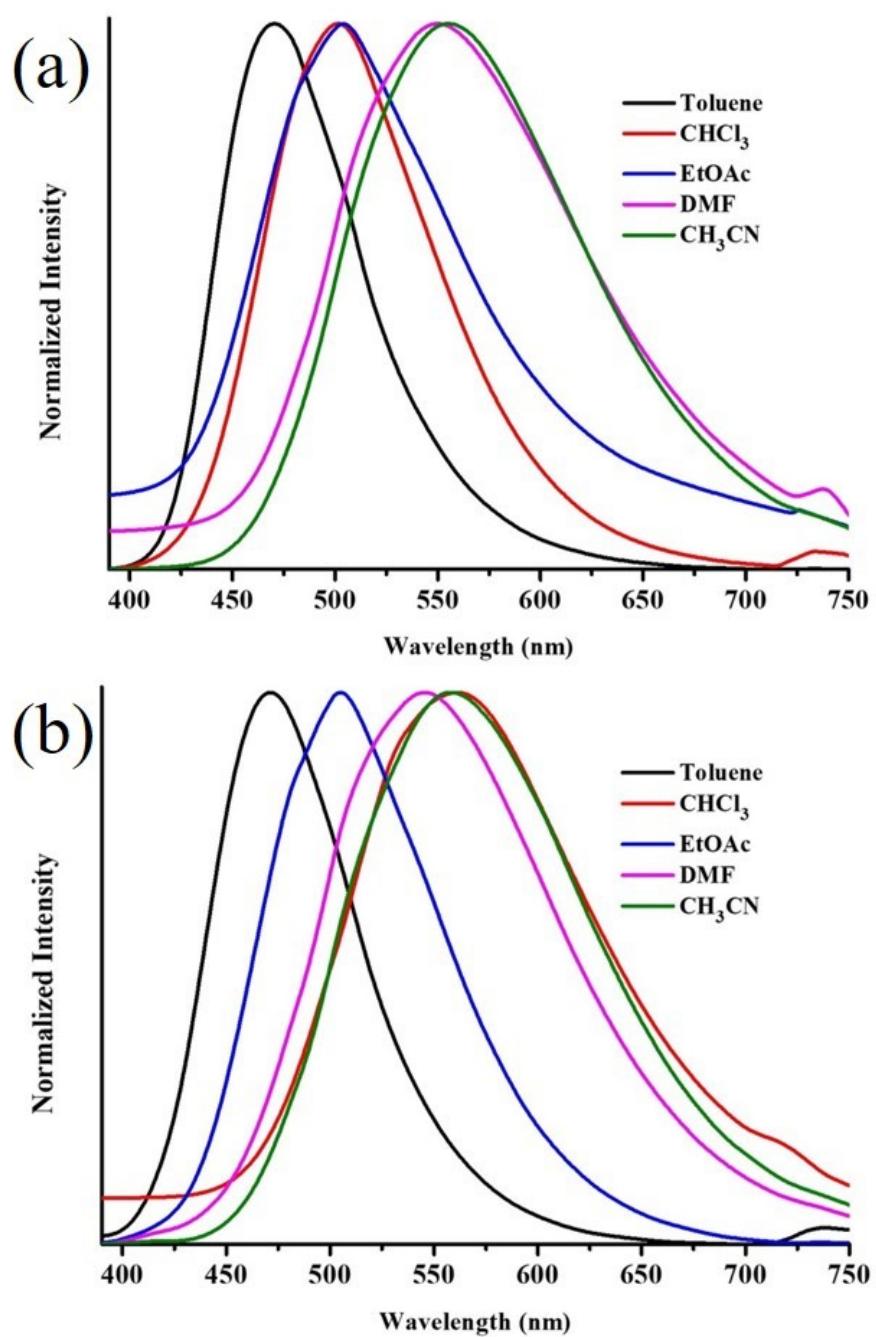


Fig. S1. Fluorescence spectra of (a) **1** and (b) **2** in different solvents (10^{-4} M).

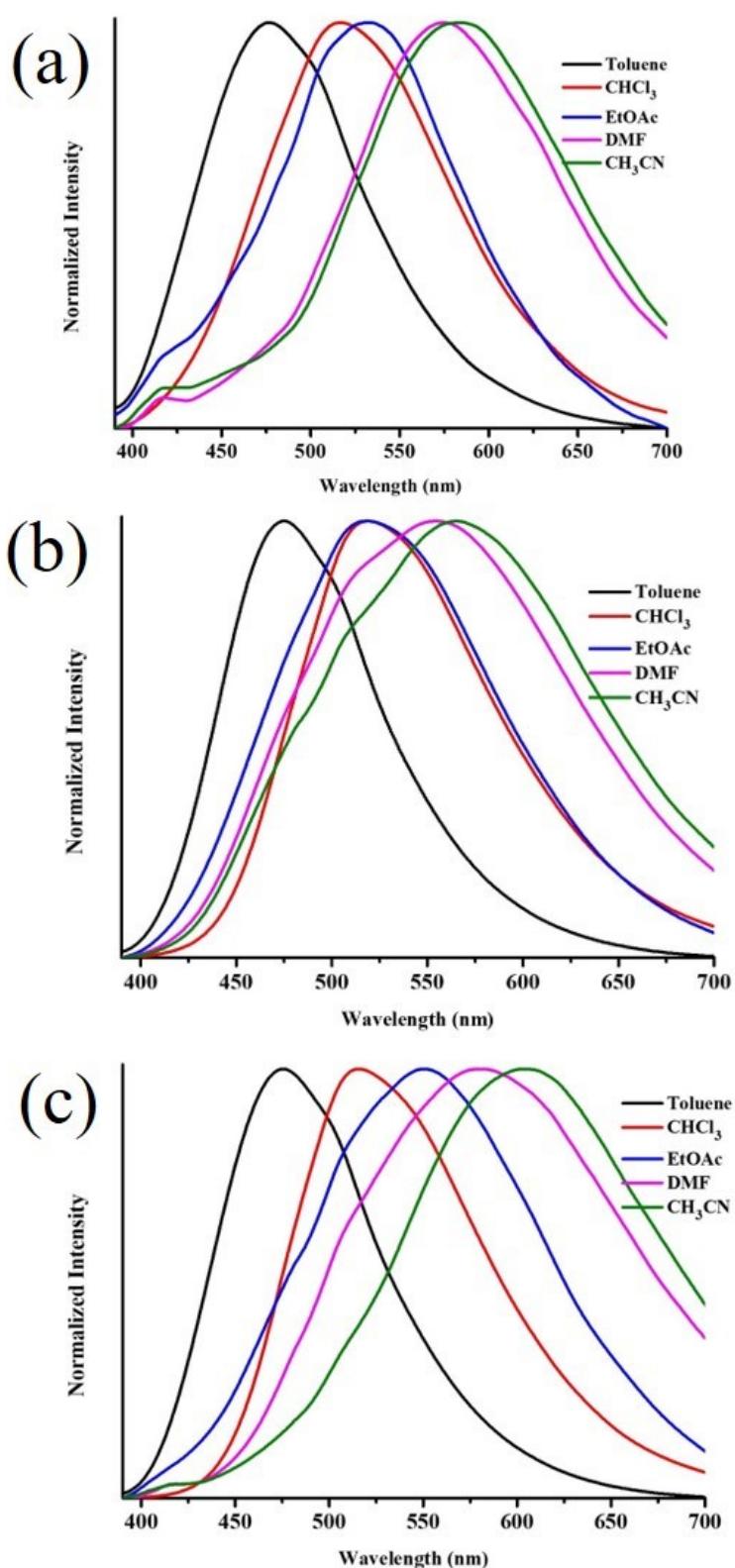


Fig. S2. Fluorescence spectra of (a) **4**, (b) **5** and (c) **6** in different solvents (10^{-4} M).

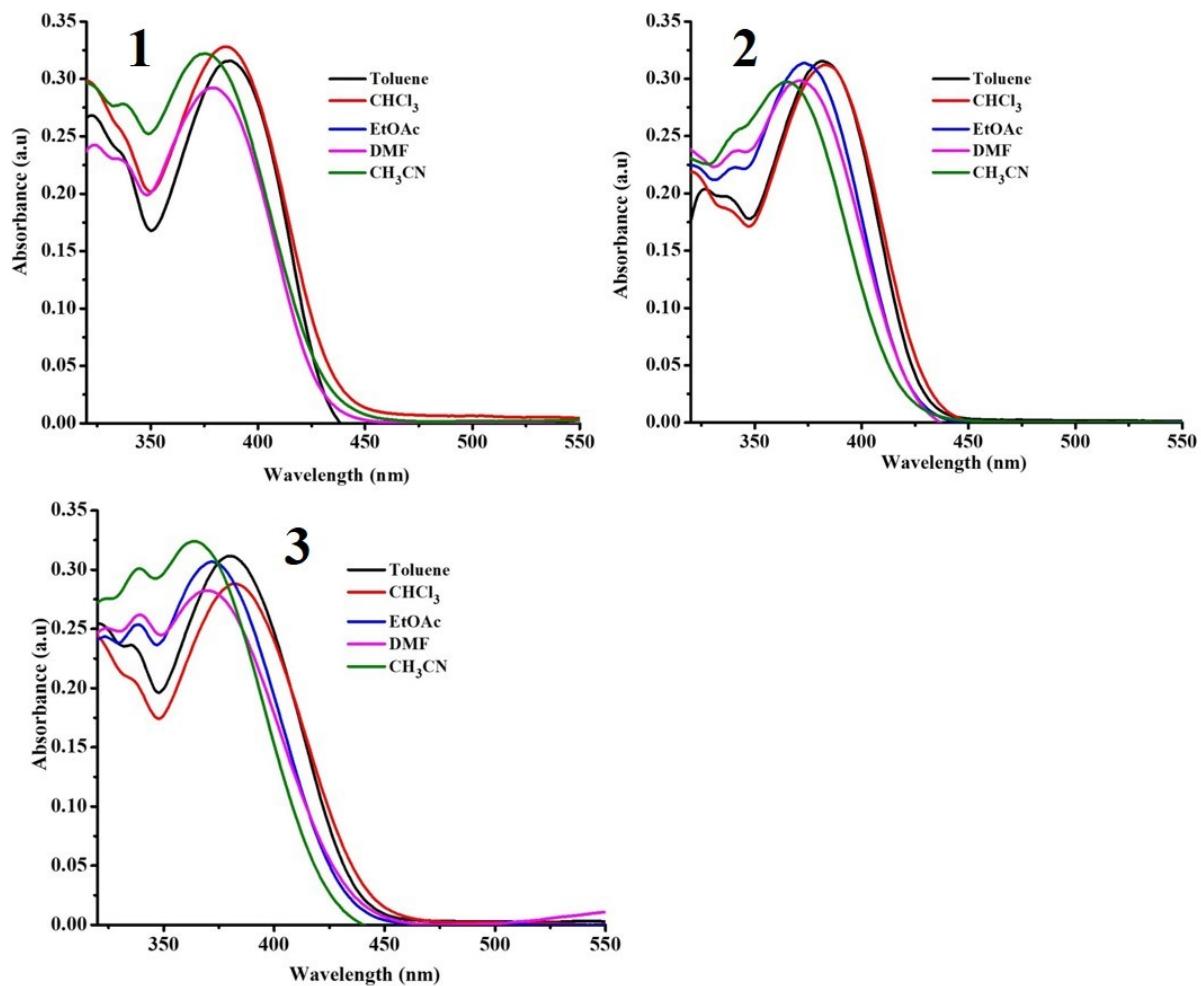


Fig. S3. Absorption spectra of **1-3** in different solvents (10^{-4} M).

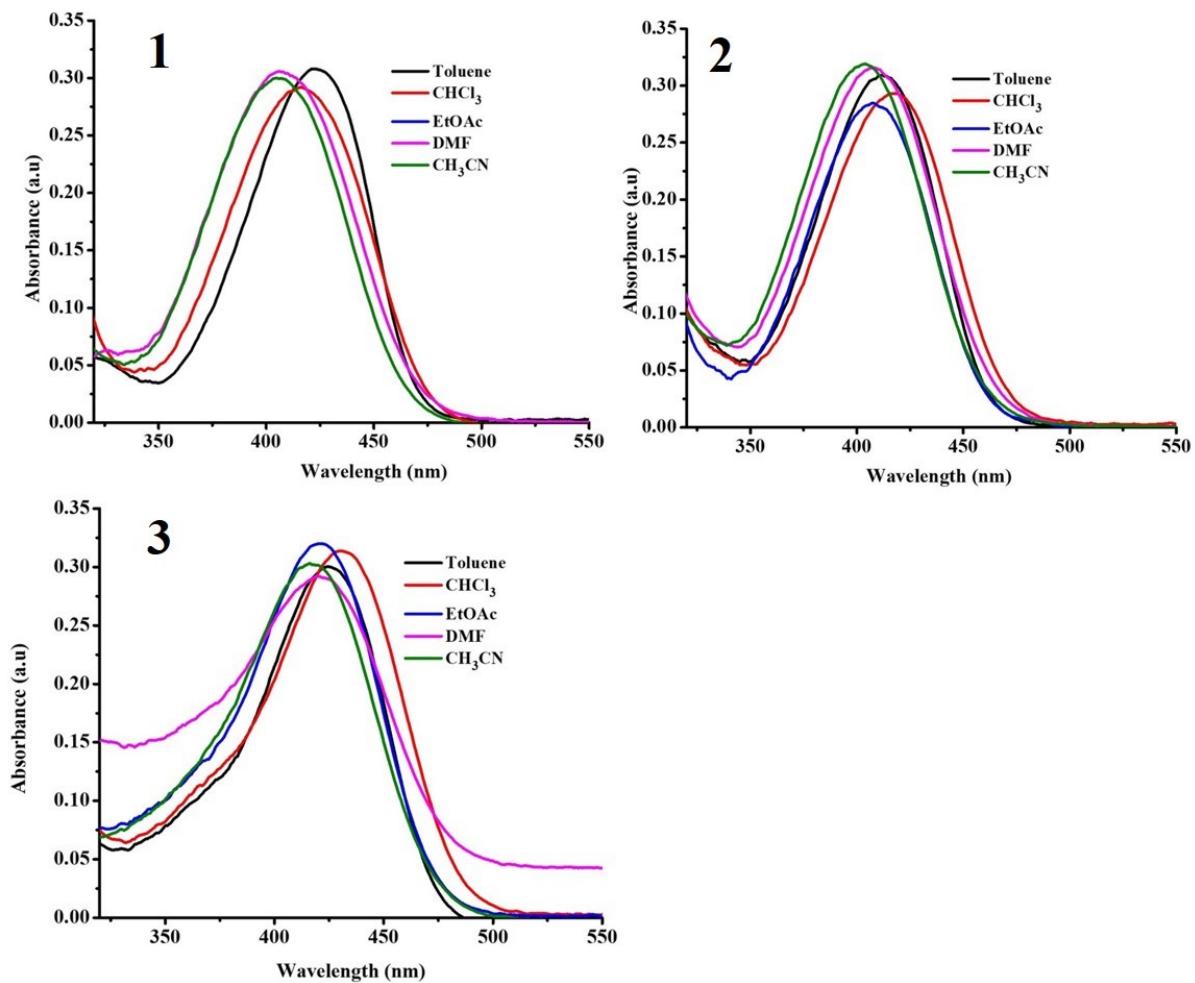


Fig. S4. Absorption spectra of **4-6** in different solvents (10^{-4} M).

Table S1. Fluorescence efficiency of **1-6** compared to quinine sulfate in different solvent polarity.

Solvents	Quantum Yield (Φ_f)					
	1	2	3	4	5	6
Toluene	0.142	0.28	0.095	0.036	0.21	0.085
CHCl ₃	0.116	0.09	0.068	0.035	0.096	0.051
Ethyl acetate	0.132	0.25	0.088	0.033	0.023	0.031
DMF	0.160	0.163	0.13	0.030	0.059	0.036
CH ₃ CN	0.051	0.16	0.098	0.025	0.085	0.015

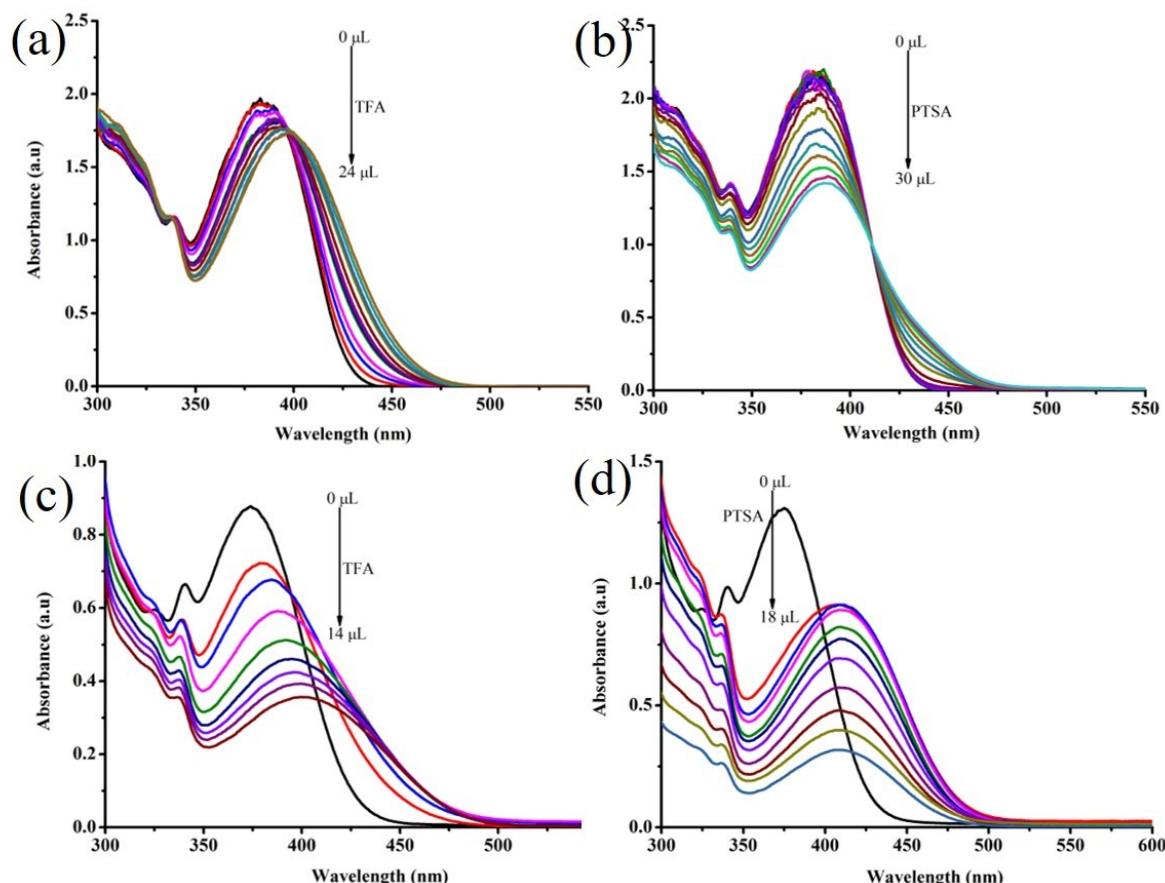


Fig. S5. Absorption changes of **2** (10⁻⁴ M) upon addition of (a, c) TFA (10⁻² M) and (b, d) PTSA (10⁻² M) in (a, b) toluene and (c, d) CHCl₃.

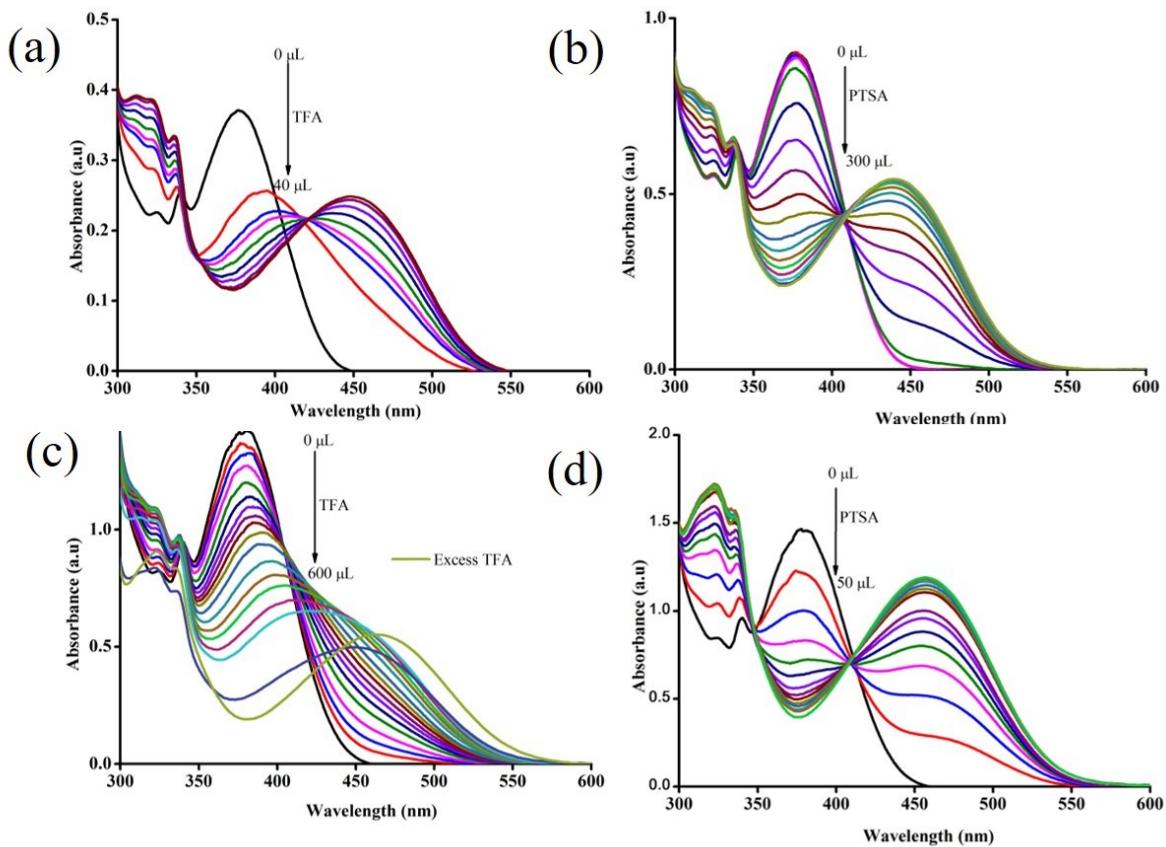


Fig. S6. Absorption changes of **3** (10⁻⁴ M) upon addition of (a, c) TFA (10⁻² M) and (b, d) PTSA (10⁻² M) in (a, b) toluene and (c, d) CHCl₃..

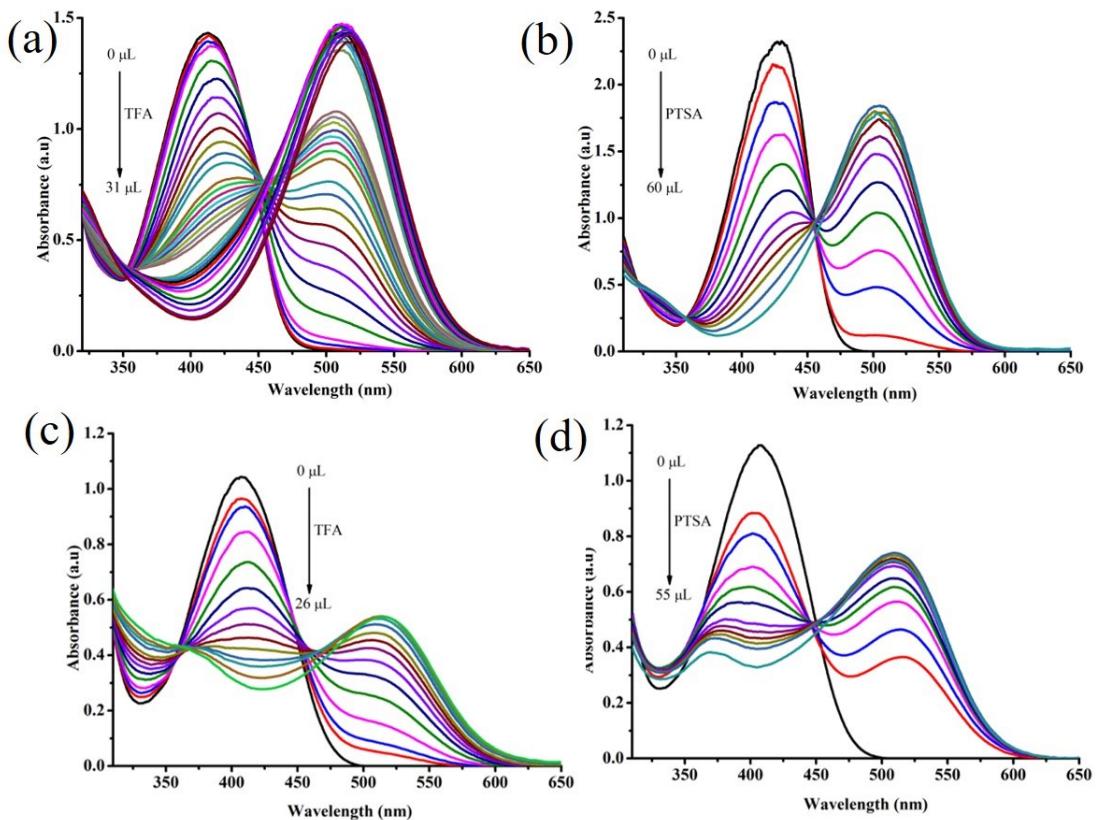


Fig. S7. Absorption changes of **4** (10^{-4} M) upon addition of (a, c) TFA (10^{-2} M) and (b, d) PTSA (10^{-2} M) in (a, b) toluene and (c, d) CHCl_3 .

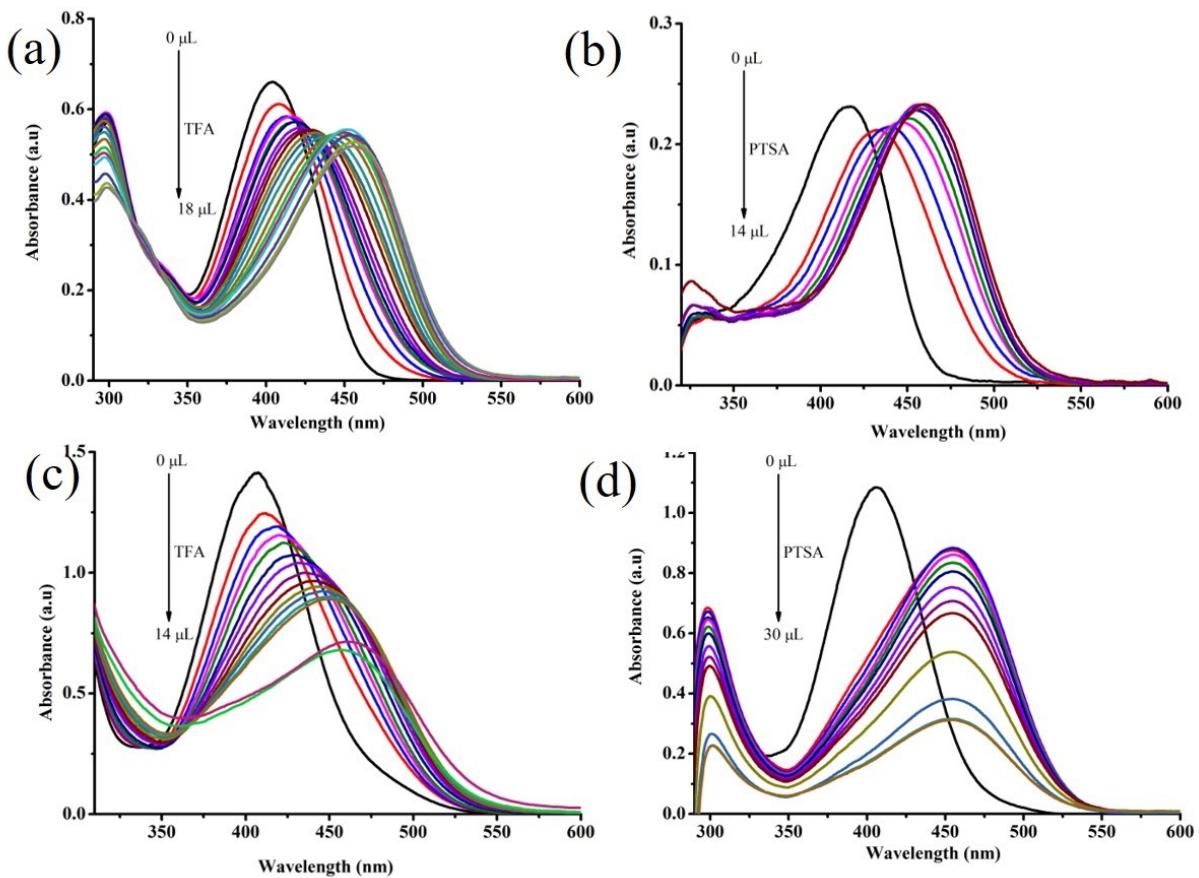


Fig. S8. Absorption changes of **5** (10^{-4} M) upon addition of (a, c) TFA (10^{-2} M) and (b, d) PTSA (10^{-2} M) in (a, b) toluene and (c, d) CHCl_3 .

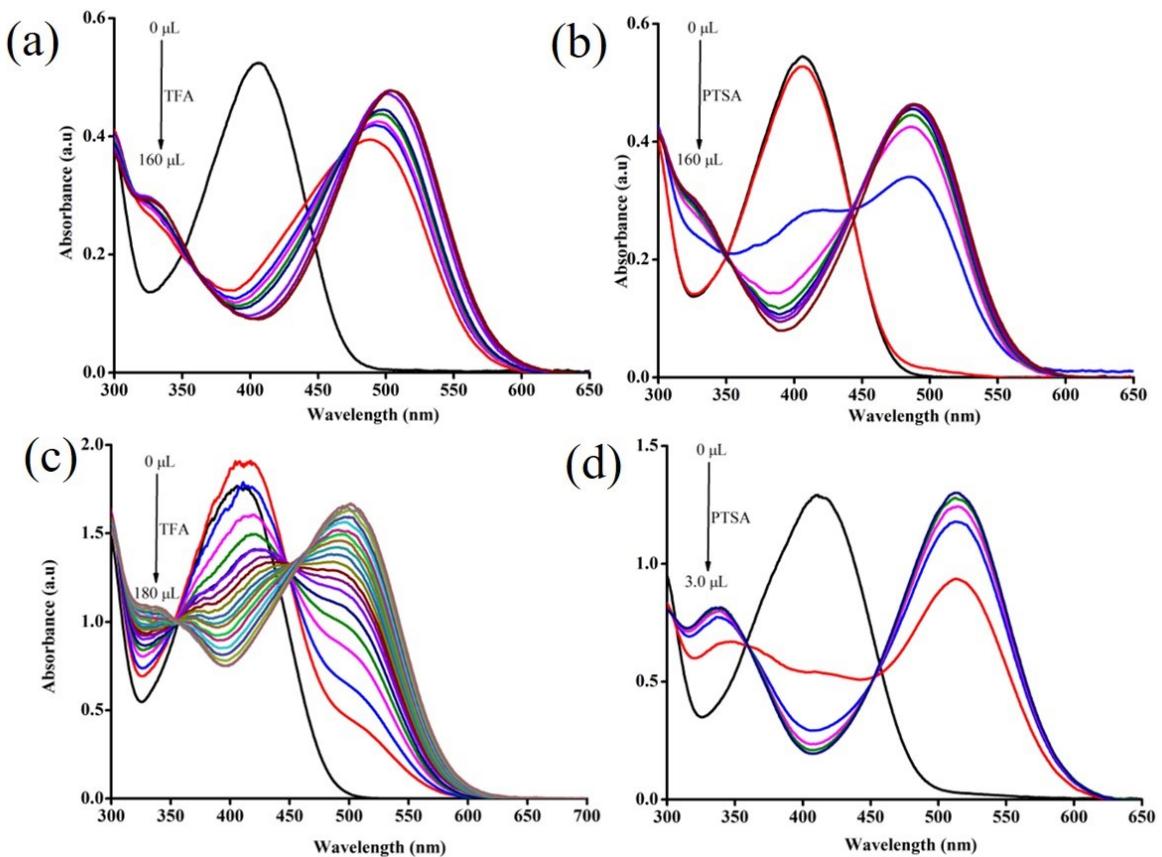


Fig. S9. Absorption changes of **6** (10^{-4} M) upon addition of (a, c) TFA (10^{-2} M) and (b, d) PTSA (10^{-2} M) in (a, b) toluene and (c, d) CHCl_3 .

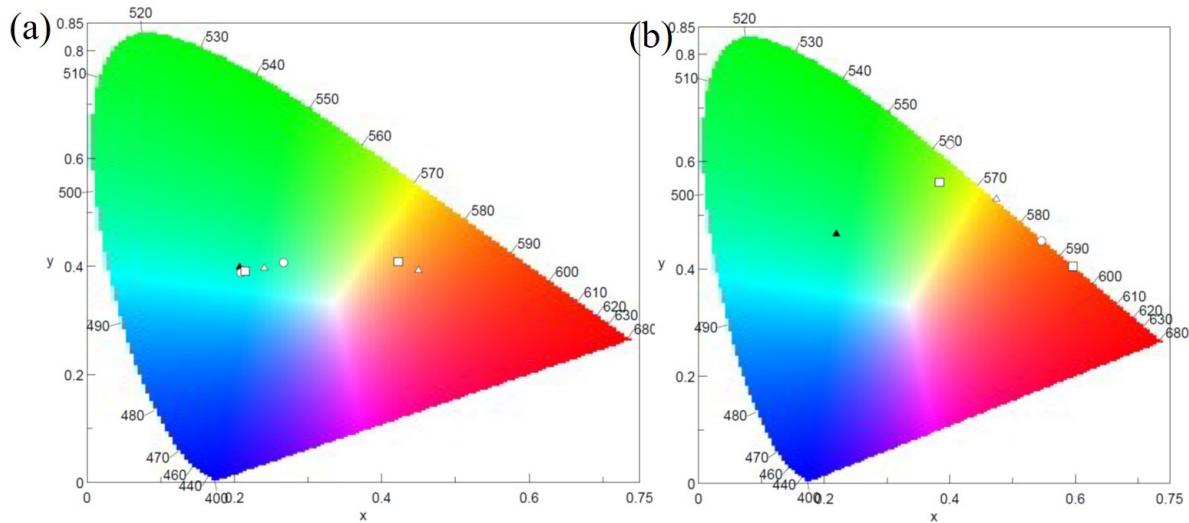


Fig. S10. CIE 1931 chromaticity plot with emission color coordinates of **1** in CHCl_3 by adding (a) TFA and (b) PTSA (10^{-2} M).

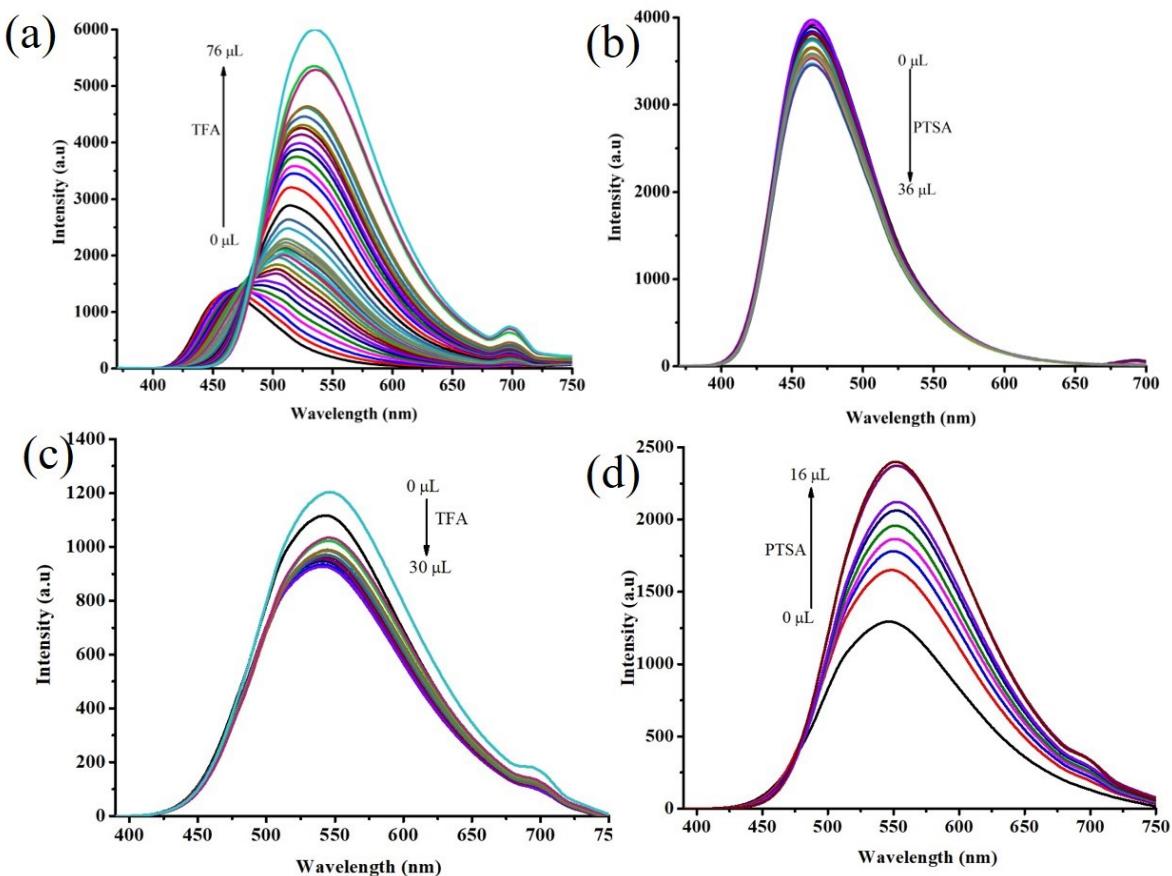


Fig. S11. Fluorescence spectra of **2** (10^{-4} M) upon addition of (a, c) TFA (10^{-2} M) and (b, d) PTSA (10^{-2} M) in (a, b) toluene and (c, d) CHCl_3 . $\lambda_{\text{exc}} = 370$ nm.

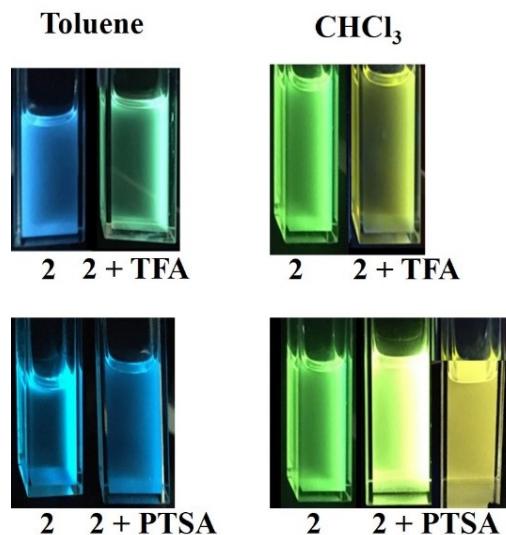


Fig. S12. Digital fluorescence images of **2** (10^{-4} M) upon addition of TFA (10^{-2} M) and PTSA (10^{-2} M) in toluene and CHCl_3 . $\lambda_{\text{exc}} = 365$ nm.

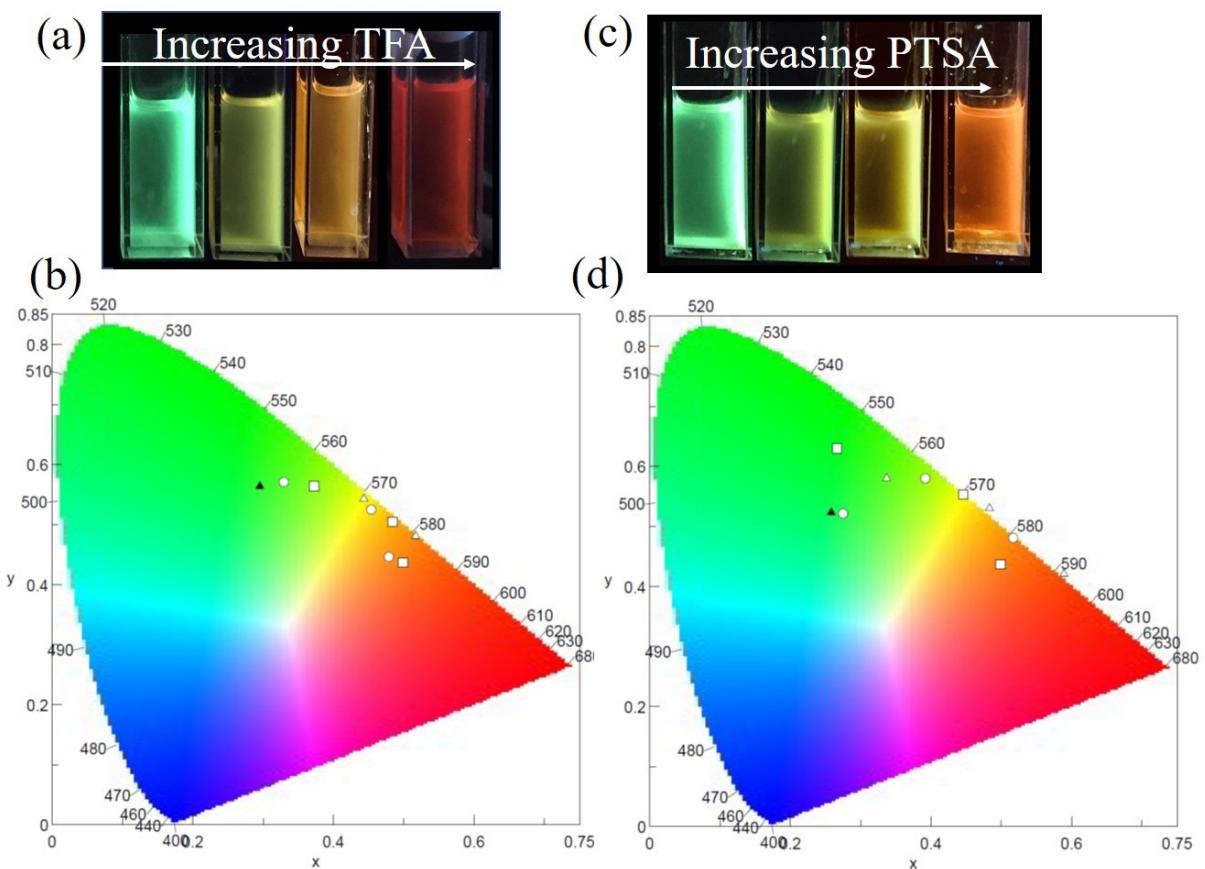


Fig. S13. (a, c) Digital fluorescence images and (b, d) CIE 1931 chromaticity plot with emission colour coordinates of **3** (10^{-4} M) upon adding (a, b) TFA (10^{-2} M) and (c, d) PTSA (10^{-2} M) in CHCl_3 . $\lambda_{\text{exc}} = 365$ nm (for digital images) and 370 nm (for spectra)..

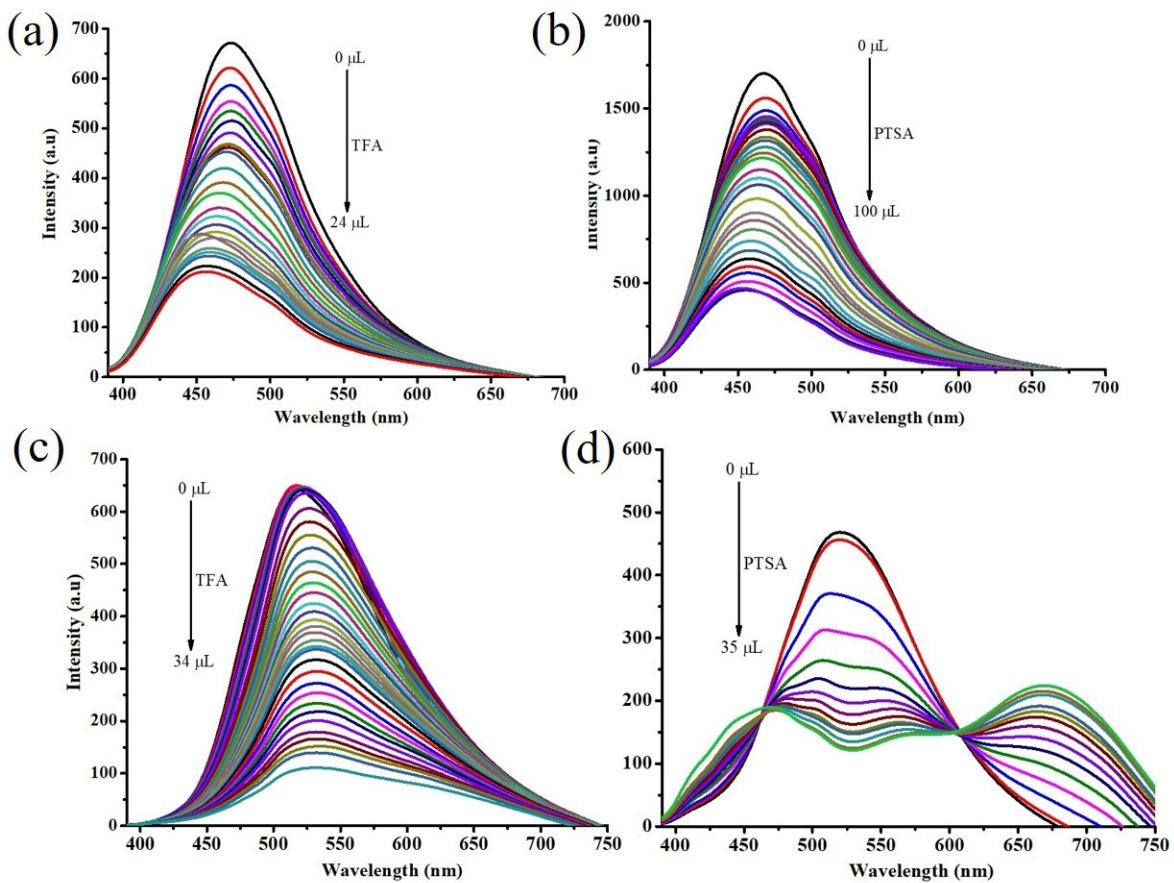


Fig. S14. Fluorescence spectra of **5** (10^{-4} M) upon addition of (a, c) TFA (10^{-2} M) and (b, d) PTSA (10^{-2} M) in (a, b) toluene and (c, d) CHCl₃. $\lambda_{\text{exc}} = 370$ nm.

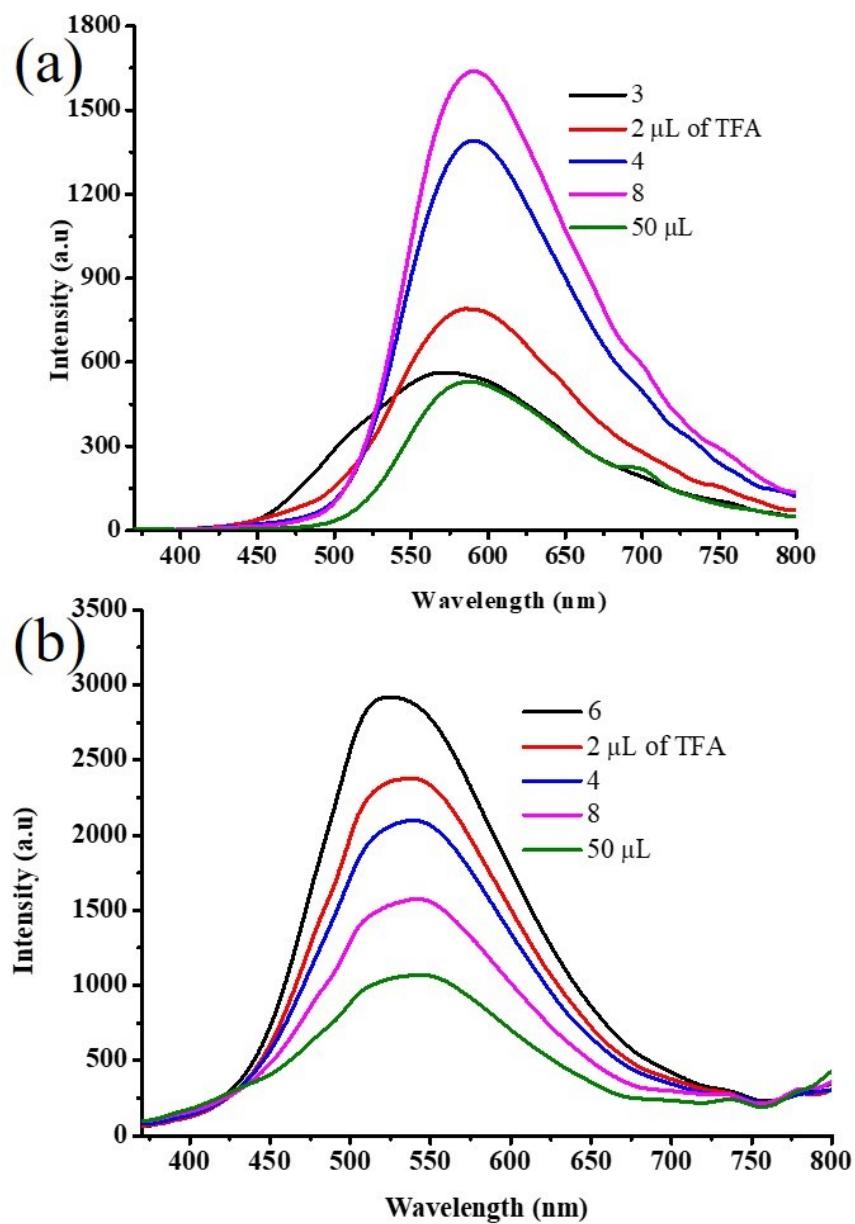


Fig. S15. Fluorescence spectra of (a) **3** (10^{-4} M) and (b) **6** (10^{-4} M) upon addition of TFA (10^{-2} M) in DMF. $\lambda_{\text{exc}} = 370$ nm..

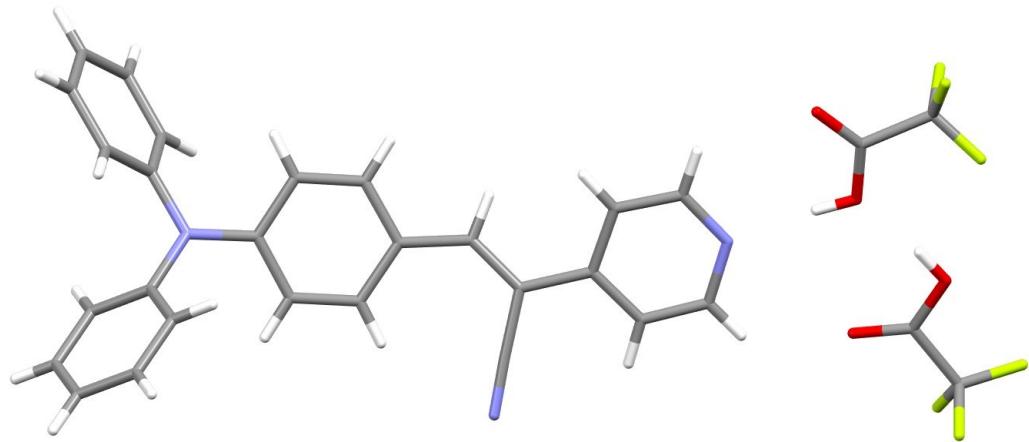


Fig. S16. Molecular structure of 6-TFA in the crystal lattice.

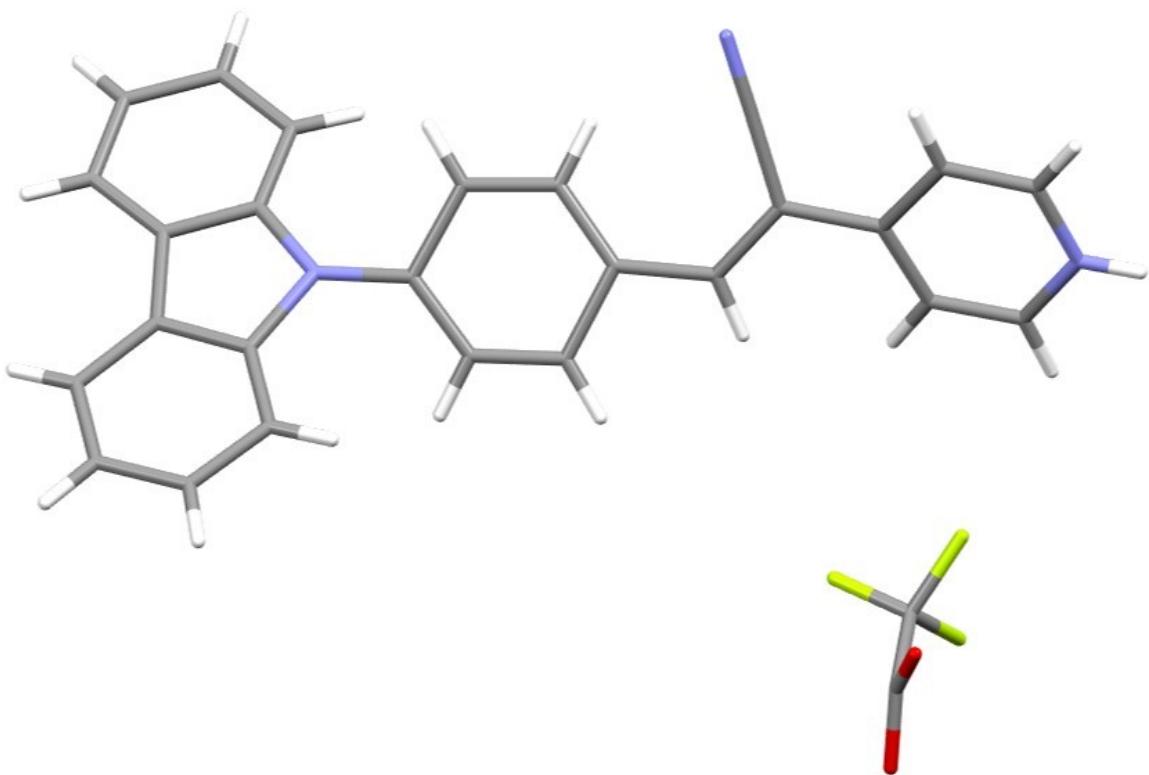


Fig. S17. Molecular structure of 3-TFA in the crystal lattice.

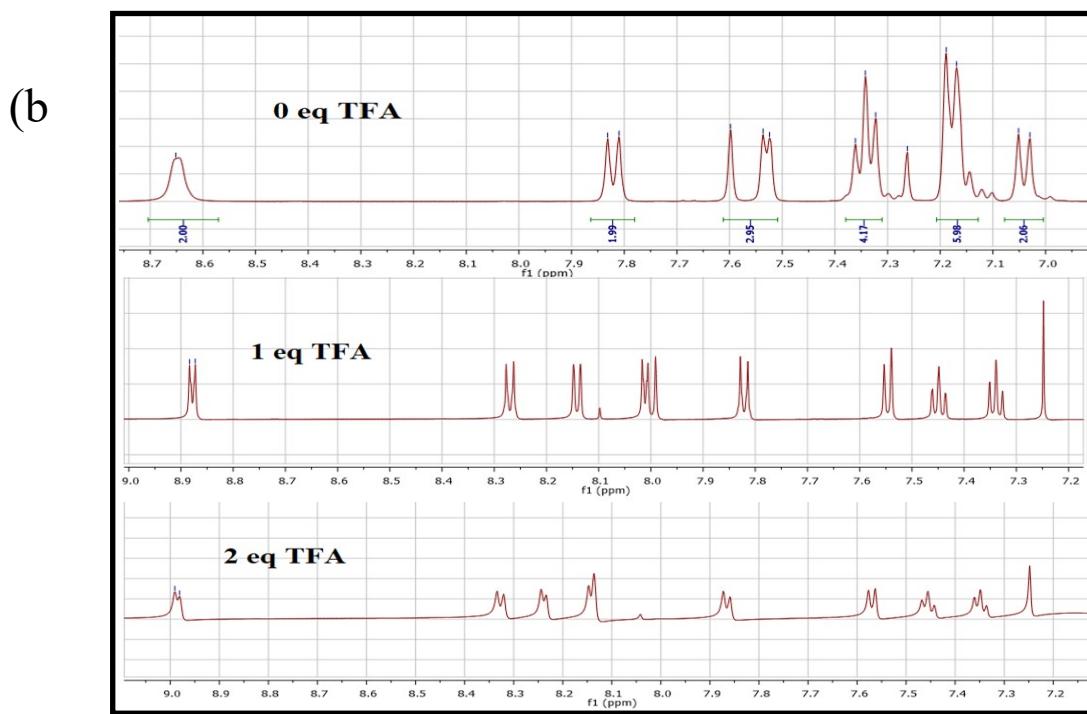
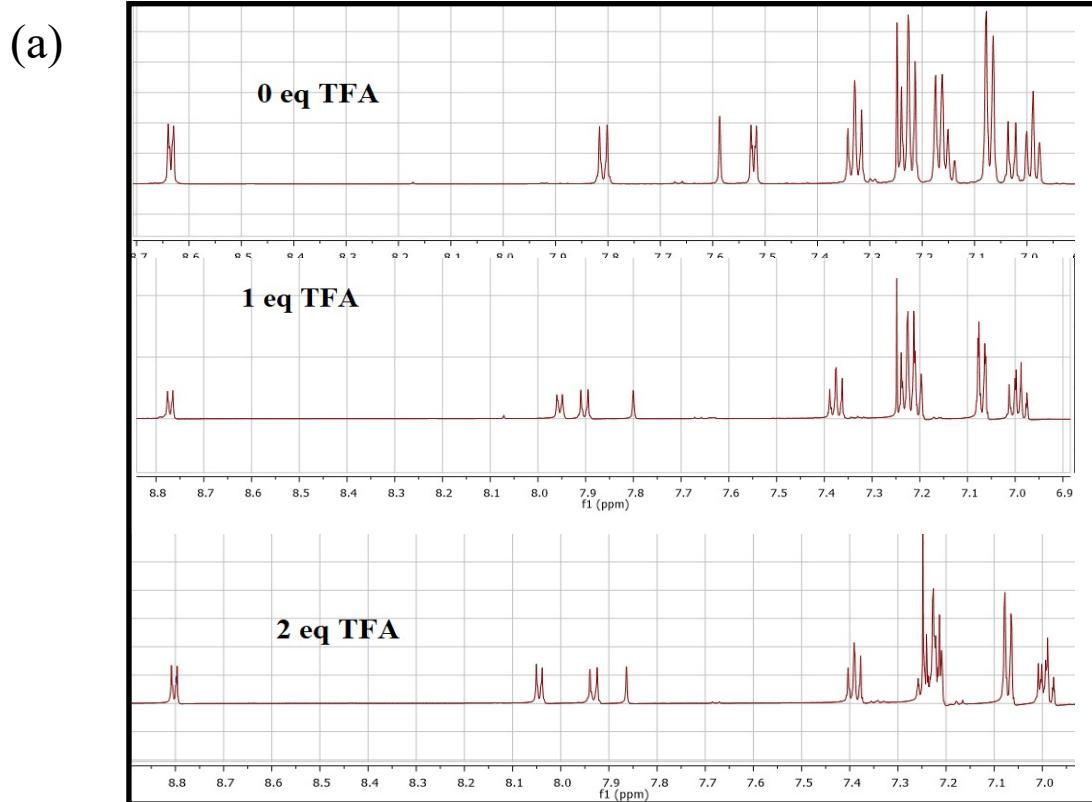


Fig. S18. NMR spectra of (a) 3 and 3-TFA and (b) 6 and 6-TFA.

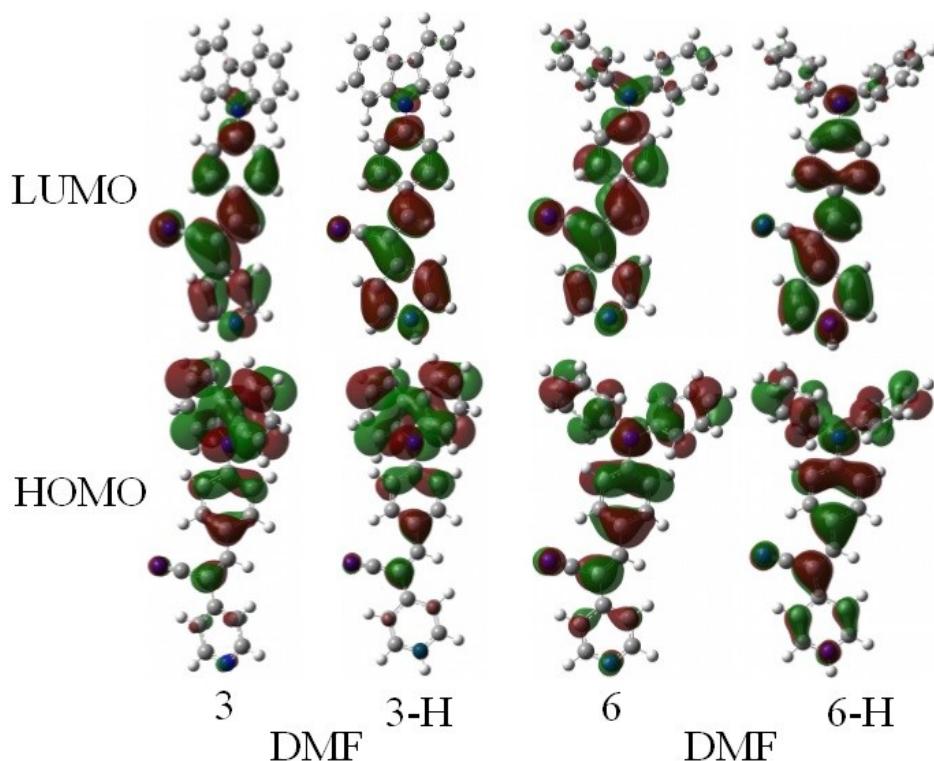


Fig. S19. Calculated frontier orbital for **3**, **3-H**, **6** and **6-H** in DMF.

Table S2. Calculated energy level of HOMO, LUMO and optical band gap.

		HOMO (eV)	LUMO (eV)	Band gap (eV)
3	Toluene	-5.57	-2.49	-3.08
	CHCl ₃	-5.57	-2.48	-3.09
	DMF	-5.58	-2.47	-3.11
3-H	Toluene	-6.49	-4.50	-1.99
	CHCl ₃	-6.07	-3.89	-2.18
	DMF	-5.74	-3.32	-2.42
6	Toluene	-5.29	-2.21	-3.08
	CHCl ₃	-5.29	-2.23	-3.06
	DMF	-5.29	-2.25	-3.04
6-H	Toluene	-6.50	-4.15	-2.35
	CHCl ₃	-6.02	-3.59	-2.43
	DMF	-5.62	-3.08	-2.54

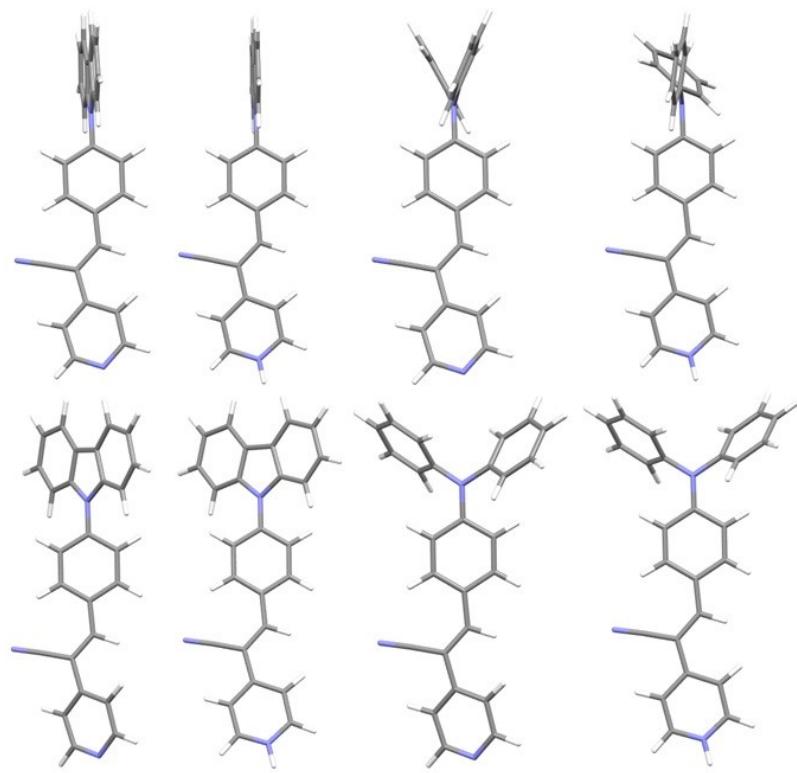


Fig. S20. Calculated ground and excited state (S1) molecular conformation of (a) **3** and **3-H** and (b) **6** and **6-H** in toluene.

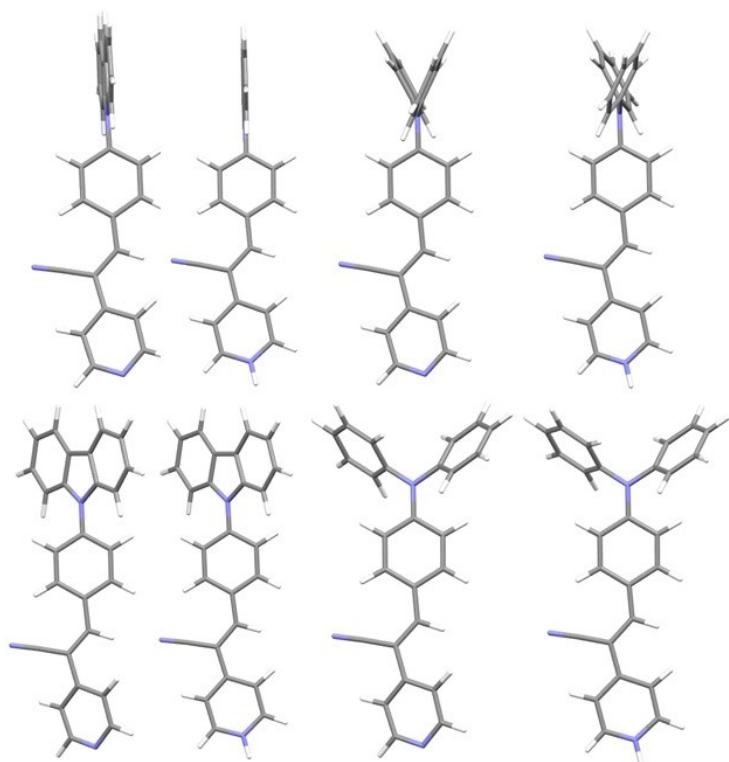


Fig. S21. Calculated ground and excited state (S1) molecular conformation of (a) **3** and **3-H** and (b) **6** and **6-H** in CHCl₃.

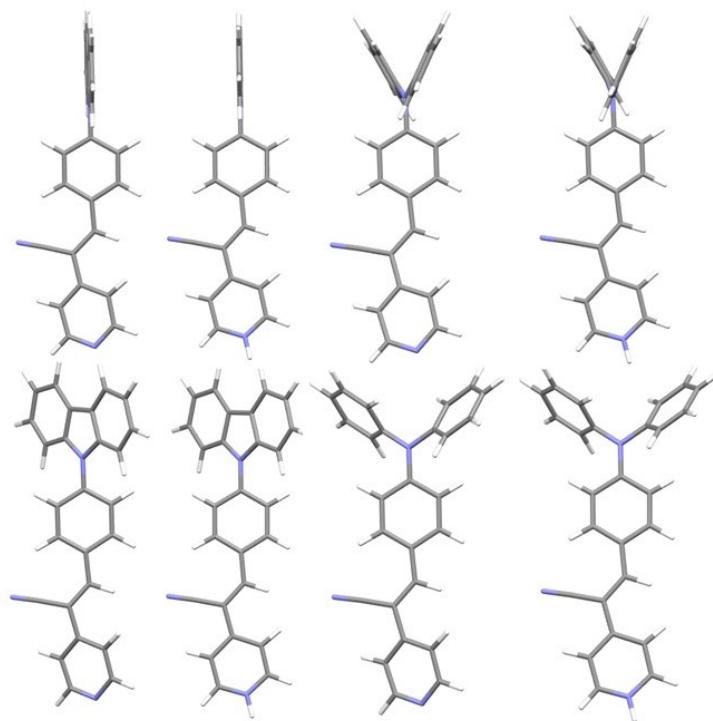
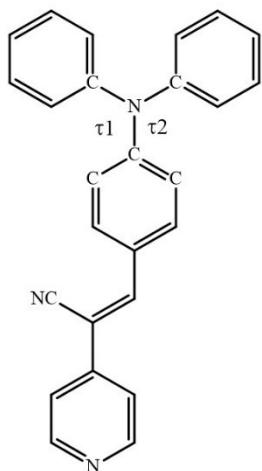


Fig. S22. Calculated ground and excited state (S1) molecular conformation of (a) **3** and **3-H** and (b) **6** and **6-H** in DMF.

Table S3. Comparison of torsion angle of ground/excited state **3** and **6** (protonated/deprotonated) in different solvents.



		Ground state		Excited state (S1)	
		Torsion angle (τ_1)	Torsion angle (τ_2)	Torsion angle (τ_1)	Torsion angle (τ_2)
3	Toluene	49.69	49.36	89.48	89.49
	CHCl ₃	49.85	49.55	89.28	89.14
	DMF	50.10	49.84	85.40	85.34
3-H	Toluene	42.20	42.13	89.69	90.38
	CHCl ₃	43.94	43.58	89.99	90.51
	DMF	46.19	46.08	90.24	90.13
6	Toluene	27.96	27.69	84.71	85.15
	CHCl ₃	27.16	26.56	81.21	81.44
	DMF	26.18	25.52	72.87	72.94
6-H	Toluene	18.86	18.26	81.22	96.60
	CHCl ₃	19.45	18.97	91.87	91.40
	DMF	20.67	20.06	85.44	85.30