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## **Supplementary Information**

## Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-Graphene Oxide-Amino Acid Ionic Liquid magnetic solid-phase extraction combined with ICP-OES for speciation of Cr(III) and Cr(VI) in environmental waters

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Fig. S1 SEM images of (a) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and (b) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.

**Fig. S2** FT-IR spectra (a) IL, (b) GO, (c) Fe<sub>3</sub>O<sub>4</sub>, (d) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>, (e) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO, (f) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.

**Fig. S3** Magnetic hysteresis loops of Fe<sub>3</sub>O<sub>4</sub>, Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>, Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.

Fig. S4 TGA spectra of Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO -AAIL.

Fig. S5 Zeta potential plots of Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.

Table S1 Existent forms of Cr(III) and Cr(VI) under different pH.

Table S2 The determination results of Cr(III) and Cr(VI) in certified sample.

**Table S3** The determination results of Cr(III) and Cr(VI) in certified sample and environmental samples by FAAS.



Fig. S1 SEM images of (a) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and (b) Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.



**Fig. S2** FT-IR spectra (a) IL, (b) GO, (c)  $Fe_3O_4$ , (d)  $Fe_3O_4$ -SiO<sub>2</sub>, (e)  $Fe_3O_4$ -SiO<sub>2</sub>-GO, (f)  $Fe_3O_4$ -SiO<sub>2</sub>-GO-AAIL.



**Fig. S3** Magnetic hysteresis loops of Fe<sub>3</sub>O<sub>4</sub>, Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>, Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.



Fig. S4 TGA spectra of Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO and Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO -AAIL.



Fig. S5 Zeta potential plots of Fe<sub>3</sub>O<sub>4</sub>-SiO<sub>2</sub>-GO-AAIL.

Table S1 Existent forms of Cr(III) and Cr(VI) under different pH

pН	3.0	4.0		6.0	6.5	7.0	9.0	>9.0	
Cr(III)	Cr(H <sub>2</sub> O	$_{2}O)_{6}^{3+}$		(OH) <sup>2+</sup>	$Cr(OH)_2^+$	Cr(OH) <sub>3</sub>		Cr(OH) <sub>4</sub> -	
Cr(VI)	HCrO <sub>4</sub> -						CrO <sub>4</sub> <sup>2-</sup>		

Table S2 The determination results of Cr(III) and Cr(VI) in certified sample.

Samples	Certified value (µg· L <sup>-1</sup> )	Added (µg· L <sup>-1</sup> )		Found ( $\mu g \cdot L^{-1}$ )		Recovery (%)	
		Cr(III)	Cr(VI)	Cr(III)	Cr(VI)	Cr(III)	Cr(VI)
GSB 07- 1187-2000	60.30	0.00	0.00	$52.64\pm2.59$	$7.36\pm2.93$	-	-
		50.00	50.00	$103.93\pm3.33$	$59.46 \pm 2.46$	102.6	104.2
		100.00	100.00	$150.55\pm3.98$	$\begin{array}{c} 119.17 \pm \\ 4.50 \end{array}$	97.9	111.8
		150.00	150.00	$201.63\pm1.85$	$166.68 \pm 2.59$	99.3	106.2

Samples	Total Cr ( $\mu g \cdot L^{-1}$ )
GSB 07-1187-2000	60.34
Lake water	14.63
Water Sample Near Pesticide Chemical Plant	34.80
Water Sample Near	20.77
Chemical Plant	29.66

Table S3 The determination results of Cr(III) and Cr(VI) in certified sample and

environmental samples by FAAS.