

## Supporting Information

### Excellent fluoride removal performance by electrospun La-Mn bimetal oxide nanofibers

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#### Abbreviations

$C_0$	initial fluoride concentration ( $\text{mg L}^{-1}$ )
$C_e$	equilibrium fluoride concentration ( $\text{mg L}^{-1}$ )
$m$	mass of La-Mn adsorbent (g)
$V$	volume of fluoride solution (L)
$Q_e$	adsorption capacity at equilibrium ( $\text{mg g}^{-1}$ )
$Q_m$	maximum adsorption capacity ( $\text{mg g}^{-1}$ )
$Q_t$	adsorption capacity at any time $t$ ( $\text{mg g}^{-1}$ )
PFO	pseudo-first order
PSO	pseudo-second order
$k_1$	LPFO rate constant ( $\text{min}^{-1}$ )
$k_2$	LPSO rate constant ( $\text{g mg}^{-1}\text{min}^{-1}$ )
$k_{\text{int}}$	intra-particle diffusion efficient ( $\text{g mg}^{-1}\text{min}^{-1/2}$ )
$K_F$	distribution coefficient ( $\text{L}^{1/n}\text{mg}^{1-1/n}\text{g}^{-1}$ )
$n$	correction factor
$t$	time (min)
$L$	the thickness of boundary layer
$k_{\text{int}}$	constant of intra-particle diffusion ( $\text{g mg}^{-1}\text{min}^{1/2}$ )
$b$	constant of the Langmuir isotherm ( $\text{L mg}^{-1}$ )
$\Delta G$	the Gibbs free energy change ( $\text{kJ mol}^{-1}$ )
$\Delta H$	the enthalpy change ( $\text{kJ mol}^{-1}$ )

- $\Delta S$  the entropy change ( $\text{kJ mol}^{-1} \text{K}^{-1}$ )
- $R$  the ideal gas constant ( $8.314 \text{ J mol}^{-1} \text{K}^{-1}$ )
- $T$  the absolute temperature (K)

Table S1 Comparison of specific surface area and adsorption capacity with other similar studies.

adsorbents	BET ( $\text{m}^2 \text{g}^{-1}$ )	Ref.
CeO <sub>2</sub> -ZrO <sub>2</sub> nanocages	29.61	1
Zr-CTS beads	0.14	2
$\gamma$ -Fe-CTB composite	0.15	3
La <sub>2</sub> O <sub>3</sub> -CeO <sub>2</sub> /laterite (LCL)	14.4	4
ZrO <sub>2</sub> /SiO <sub>2</sub> /Fe <sub>3</sub> O <sub>4</sub>	4.6	5
tin-impregnated chitin (SnC)	2.83	6
Sn(IV) chloride impregnated chitosan (Sn-Ch)	3.02	7

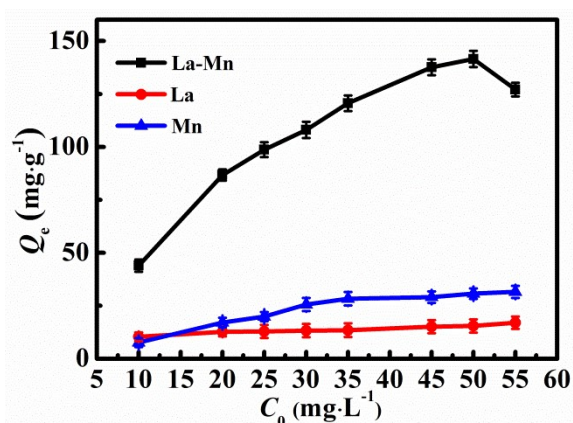


Fig. S1 Effect of fluoride concentration on fluoride ion adsorption on La-Mn bimetal oxide nanofibers, La nanofibers and Mn nanofibers. (dosage = 0.01 g, pH = 3, T = 298 K, V = 50 mL)

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