

**Dual-role Magnesium Aluminate Ceramic Film as an Advanced Separator and
Polysulfide Trapper in Li-S battery: Experimental and DFT investigations**

Divyamahalakshmi Muthuraj ^a, Raja Murugan ^a, Pavul Raj Rayappan ^a, Ganapathi Rao
Kandregula ^a, Kothandaraman Ramanujam ^{a,b,*}

^a Clean Energy Lab, Department of Chemistry, Indian Institute of Technology Madras,
Chennai-600036, India

^b Potential Centre of Excellence: Advanced Centre for Energy Storage and Conversion,
Indian Institute of Technology Madras, Chennai-600036, India

* Corresponding author E-mail address: rkraman@iitm.ac.in

Table S1. Porosity calculation for Celgard[®], MAO, and MAO/MWCNT separator

Separator	Mass of dry separator (mg)	Mass of DME/DOL absorbed separator (mg)	Mass of DME/DOL absorbed (mg)	Porosity (%) estimated using eq. 1
Celgard [®]	4.2	7.1	2.9	38.5
MAO	46.3	66.8	20.5	62.6
MAO/MWCNT	48.8	76.4	27.5	68.1

When the thickness of the MAO separator is less than 100 microns, the mechanical strength of the separator is poor and tend to break (brittle). Therefore, we have limited the thickness of the separator to be 100 μm and it is highly flexible. Further studies are required to fabricate a thin MAO separator for realistic future application in battery, and we will work towards it in future.

Table S2. Electrical parameters of Li|Celgard®|Li symmetric cell over time (derived from Nyquist plot Fig. 2d).

R_{el} - electrolyte resistance,

R_{pass} - passivation layer resistance

R_i - interfacial charge transfer resistance

Y_{pass}, Y_i - admittance

n_{pass}, n_i - CPE index

Time (h)	χ^2	R_{el}(Ω)/ % Error	R_{pass}(Ω)/ % Error	Y_{pass} (mho\timesSⁿ)/ % Error	n_{pass}/ % Error	R_i (Ω)/ % Error	Y_i (mho\timesSⁿ)/ % Error	n_i/ % Error
0	7.5 \times 10 ⁻⁵	3.4/0.8	17.3/15.6	1.9 \times 10 ⁻¹ / 12.2	0.67/ 11.4	140.3/0.2	4.9 \times 10 ⁻⁶ / 1.4	0.82/ 0.1
24	1.0 \times 10 ⁻³	4.1/1.3	31.9/16.6	7.8 \times 10 ⁻² / 44	0.59/ 49	281.1/0.9	3.7 \times 10 ⁻⁶ / 4.1	0.83/ 0.4
48	4.2 \times 10 ⁻³	4.4/2.4	33.1/11.1	5.5 \times 10 ⁻² / 18.7	0.56/ 10.1	319.5/3	2.9 \times 10 ⁻⁶ / 15.3	0.88/ 2.1
72	1.3 \times 10 ⁻³	3.3/9.3	81.2/33.2	1.7 \times 10 ⁻² / 47.2	0.33/ 20.7	342.2/3.5	5.9 \times 10 ⁻⁶ / 13.3	0.77/ 2.4
144	6.2 \times 10 ⁻³	3.9/3.3	55.1/17	2.1 \times 10 ⁻⁴ / 11.7	0.15/ 19.1	360.6/17. 2	2.6 \times 10 ⁻⁶ / 29.1	0.85/ 3.7
216	3.1 \times 10 ⁻³	3.7/3	56.4/8.1	7.9 \times 10 ⁻⁴ / 12.1	0.64/ 10.2	470/2.8	1.7 \times 10 ⁻⁶ / 7.2	0.84/ 0.8
408	7.5 \times 10 ⁻³	2.7/7.6	52.2/13.1	9.5 \times 10 ⁻⁴ / 18.3	0.61/ 16.3	469.2/4.7	2.0 \times 10 ⁻¹ / 16.1	0.83/ 2.0

Table S3. Electrical parameters of Li|MAO/MWCNT|Li symmetric cell over time (derived from Nyquist plot Fig. 2e)

Time (h)	χ^2	$R_{el}(\Omega)$ / % Error	$R_{pass}(\Omega)$ / % Error	$Y_{pass}(\text{mho}\times\text{S}^n)$ / % Error	n_{pass} / % Error	$R_i(\Omega)$ / % Error	$Y_i(\text{mho}\times\text{S}^n)$ / % Error	n_i / % Error
0	4.3×10^{-4}	8.5/1.3	39.6/15.7	3.2×10^{-2} / 14.8	0.53/ 15.8	121.2/6.5	6.9×10^{-5} / 13.2	0.59/ 1.9
24	5.7×10^{-4}	10.2/1.8	53.4/70	3.9×10^{-2} / 42	0.36/ 56.1	136.5/12. 6	1.3×10^{-4} / 21.2	0.53/ 4.2
48	9.8×10^{-4}	4.9/2.1	19.5/14	2.6×10^{-3} / 12.1	0.32/ 56.5	111.4/14. 0	9.6×10^{-6} / 21	0.78/ 2.29
72	2.9×10^{-3}	3.8/6.0	16.1/9.4	4.5×10^{-2} / 49.7	0.39/ 46.3	90.8/21.7 2	2.5×10^{-4} / 34.5	0.5/ 6.5
144	1.8×10^{-3}	4.1/8.9	8.3/56.1	6.3×10^{-3} / 3.6	0.46/ 1.1	82.4/5.6	9.2×10^{-6} / 8.5	0.81/ 5.3
216	1.0×10^{-3}	4.4/2.1	10.5/10.9	1.2×10^{-5} / 17.9	0.78/ 2.1	89.2/9.8	2.6×10^{-7} / 53.1	0.11/ 5.7
408	9.8×10^{-4}	4.1/2.5	16.2/15.4	1.2×10^{-5} /27.7	0.76/ 3.2	68.4/7.6	6.3×10^{-7} / 51.2	0.94/ 4

Table S4. CV results of S electrode at various scan rate using MAO/MWCNT separator and Li-ion diffusivity values

Scan rate/ (mV s ⁻¹)	V _o (I') / V	I _o (I')/ mA	V _o (II') / V	I _o (II')/ mA	V _r (I)/ V	I _r (I)/ mA	V _r (II))/ V	I _r (II)/ mA	Voltage difference V(I-I')	Voltage difference V(II-II')
0.01	2.4	0.28	2.35	0.59	2.35	0.23	2.05	0.62	0.05	0.29
0.02	2.41	0.39	2.33	0.83	2.34	0.33	2.04	0.82	0.07	0.28
0.04	2.42	0.59	2.35	1.1	2.33	0.48	2.01	1.04	0.09	0.33
0.08	2.43	1.02	2.37	1.63	2.32	0.75	2.0	1.42	0.12	0.37
0.1	2.44	1.23	2.38	1.85	2.31	0.83	1.99	1.59	0.13	0.38
0.2	2.49	1.94	2.43	2.64	2.28	1.21	1.96	2.1	0.21	0.47
Diffusion coefficient t/ cm² s⁻¹	3.7× 10⁻⁷		2.04× 10⁻⁸		1.3× 10⁻⁷		1.1× 10⁻⁸			

Table S5. CV results of S electrode at various scan rates using Celgard® separator and Li-ion diffusivity values

Scan rate/ mV s ⁻¹	V _o (I')/ V	I _o (I')/ mA	V _o (II')/ V	I _o (II')/ mA	V _r (I)/ V	I _r (I)/ mA	V _r (II)/ V	I _r (II)/ mA	Voltage difference V(I-I')	Voltage difference V(II-II')
0.01	2.39	0.27	2.29	0.44	2.32	0.15	2.06	0.64	0.07	0.23
0.02	2.42	0.41	2.34	0.63	2.31	0.32	2.03	0.72	0.11	0.31
0.04	2.47	0.74	2.39	0.90	2.28	0.46	1.98	0.82	0.19	0.42
0.08	2.53	1.19	2.47	1.24	2.25	0.66	1.92	1.06	0.28	0.55
0.1	2.56	1.32	2.51	1.33	2.23	0.70	1.89	1.11	0.33	0.62
Diffusion coefficient/ cm² s⁻¹	4×10⁻⁷		9.9×10⁻⁹			9.8×10⁻⁸		2.9×10⁻⁹		

*V_o,V_r – oxidation and reduction potential

I_o, I_r – Peak current for oxidation and reduction process

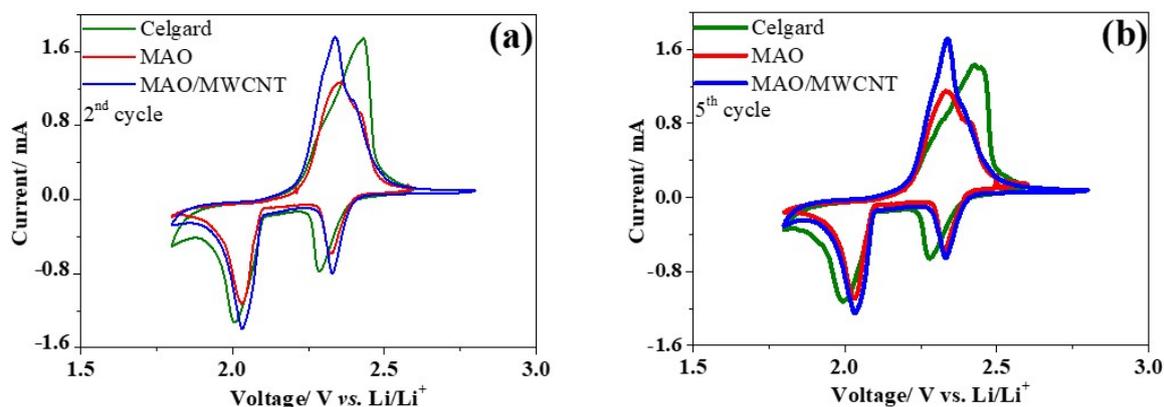


Fig. S1: CV of Li-S cell using Celgard®, MAO and MAO/MWCNT separator at 0.05 mV s^{-1} scan rate for 2nd cycle and 5th cycle

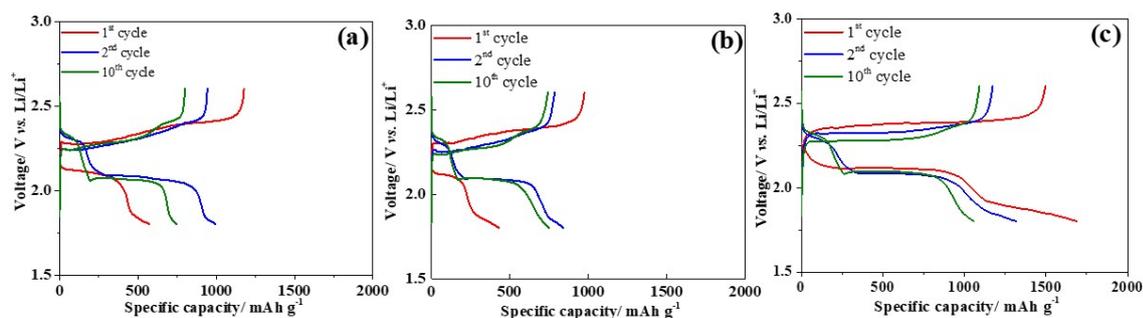


Fig. S2: Galvanostatic charge-discharge profile of Li-S cell for 1st, 2nd, and 10th cycle at 0.2 C rate while using (a) Celgard®, (b) MAO, and (c) MAO/MWCNT separators

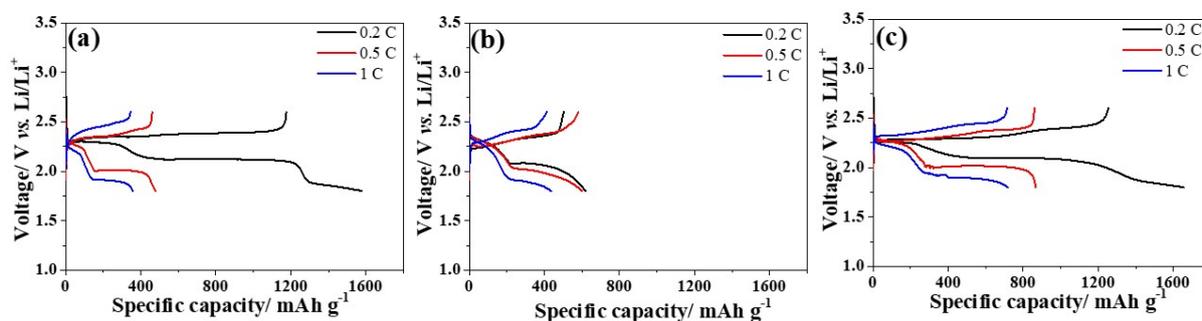


Fig. S3: Galvanostatic charge-discharge profile of Li-S cell at 0.2 C, 0.5 C, 1 C rate while using (a) Celgard®, (b) MAO, and (c) MAO/MWCNT separators