

## SUPPLEMENTARY INFORMATION

### A 2D/1D Heterojunction Nanocomposite built from Polymeric Carbon Nitride and MIL-88A(Fe) derived $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> for Enhanced Photocatalytic degradation of Rhodamine-B

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Sl. No.	Fe <sub>2</sub> O <sub>3</sub> Precursor	Synthesis	Photocatalytic condition	Degradation efficiency	Reference No.
1.	MIL-53(Fe)	Hydrothermal	50 mg, 100W LED lamp (420 nm)	92% of TC for 60 m.	52
2	MIL-101	Hydrothermal	300W Xe lamp	Hydrogen evolution	53
3	MIL-100	Hydrothermal	300W Xe lamp	Hydrogen evolution	54
4	Waste iron rust	Calcination	100 mg, Sunlight	99% MO for 120 m	44
5	FeCl <sub>3</sub> ·6H <sub>2</sub> O	Hydrothermal	100 mg, 300W Xe lamp	98% Cr(VI) reduction for 150 m	40
6	Fe(NO <sub>3</sub> ) <sub>3</sub> ·9H <sub>2</sub> O	Calcination	50 mg, 300W Xe lamp	90% Hg reduction for 60 m	37
7	Fe <sub>2</sub> (C <sub>2</sub> O <sub>4</sub> ) <sub>3</sub> ·6H <sub>2</sub> O	Calcination	10 mg,	Removal of Phospate	38
8	FeCl <sub>3</sub> ·6H <sub>2</sub> O	Calcination	100 mg	CO <sub>2</sub> reduction	39
9	FeCl <sub>3</sub> ·6H <sub>2</sub> O	Calcination	20mg, 300W Xe lamp	Hydrogen evolution	34
10	Fe(NO <sub>3</sub> ) <sub>3</sub>	Calcination	65 W CFL lamp, intensity-125 W/m <sup>2</sup> , $\lambda > 400$ nm	94.7% RhB reduction for 140 m	35
11	Fe(NO <sub>3</sub> ) <sub>3</sub> ·9H <sub>2</sub> O	Hydrothermal	70mg, 300W Xe lamp	96.7% RhB reduction for 4 h	42
12	FeCl <sub>3</sub>	Hydrothermal	300W Xe lamp	Completely degraded 4-nitrophenol 100% for 6h	43
13	MIL-88A(Fe)	Calcination	Sunlight	92% of RhB for 60 minutes	<b>This Work</b>

**Table S1. Comparison of the photocatalytic degradation efficiency of MIL-88A derived Fe<sub>2</sub>O<sub>3</sub> /C<sub>3</sub>N<sub>4</sub> with literature**

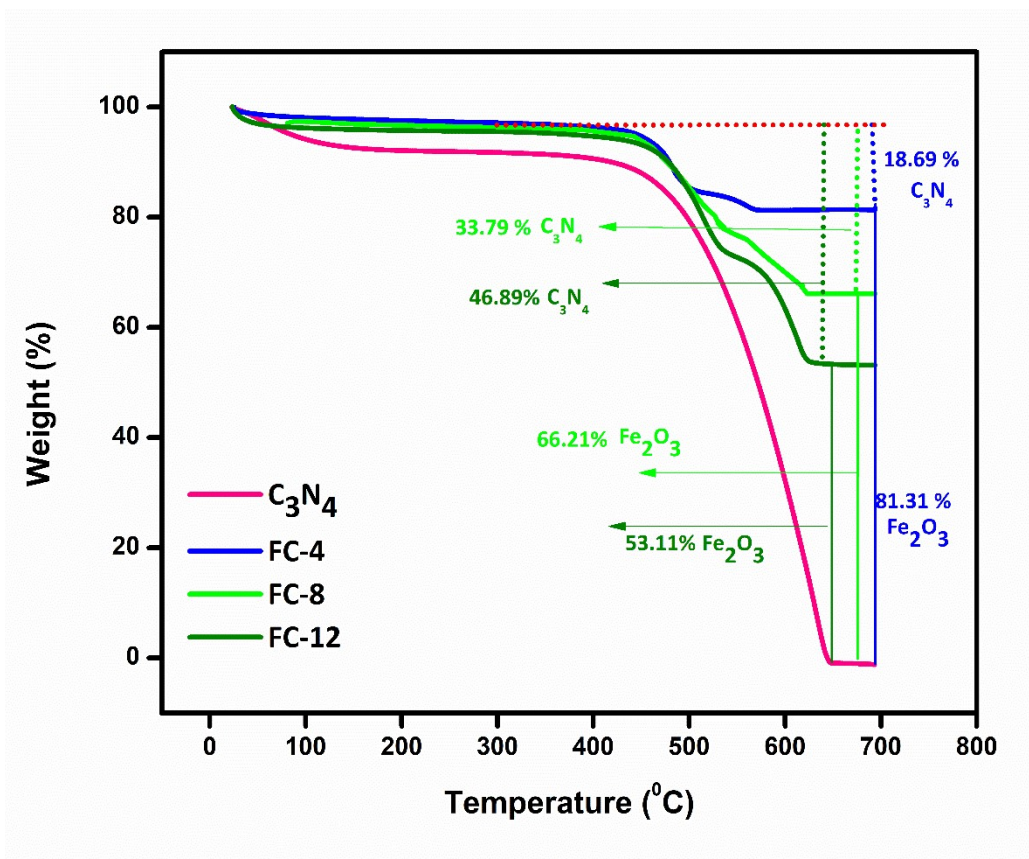


Figure S1. Thermogravimetry curve of  $C_3N_4$ , FC-4, FC-8 and FC-12

Sample Name	% of $C_3N_4$ evolved	% of $C_3N_4$ left in the residue	% of Residue ( $Fe_2O_3/C_3N_4$ )	Reference
MIL-88A	0%	-	45% Corresponding to only $Fe_2O_3$	RSC Adv., 2015, 5, 32520–32530
$C_3N_4$	100%	-	0 %	This work
$Fe_2O_3/C_3N_4$ (FC-4)	18.69%	36.31%	81.31%	This work
$Fe_2O_3/C_3N_4$ (FC-8)	33.79%	21.21%	66.21%	This work
$Fe_2O_3/C_3N_4$ (FC-12)	46.89%	8.11%	53.11%	This work

Table S2. Tabulated data of weight % of  $C_3N_4$  evolved and % of  $C_3N_4$  left in the residue by the decomposition of FC-4, FC-8 and FC-12

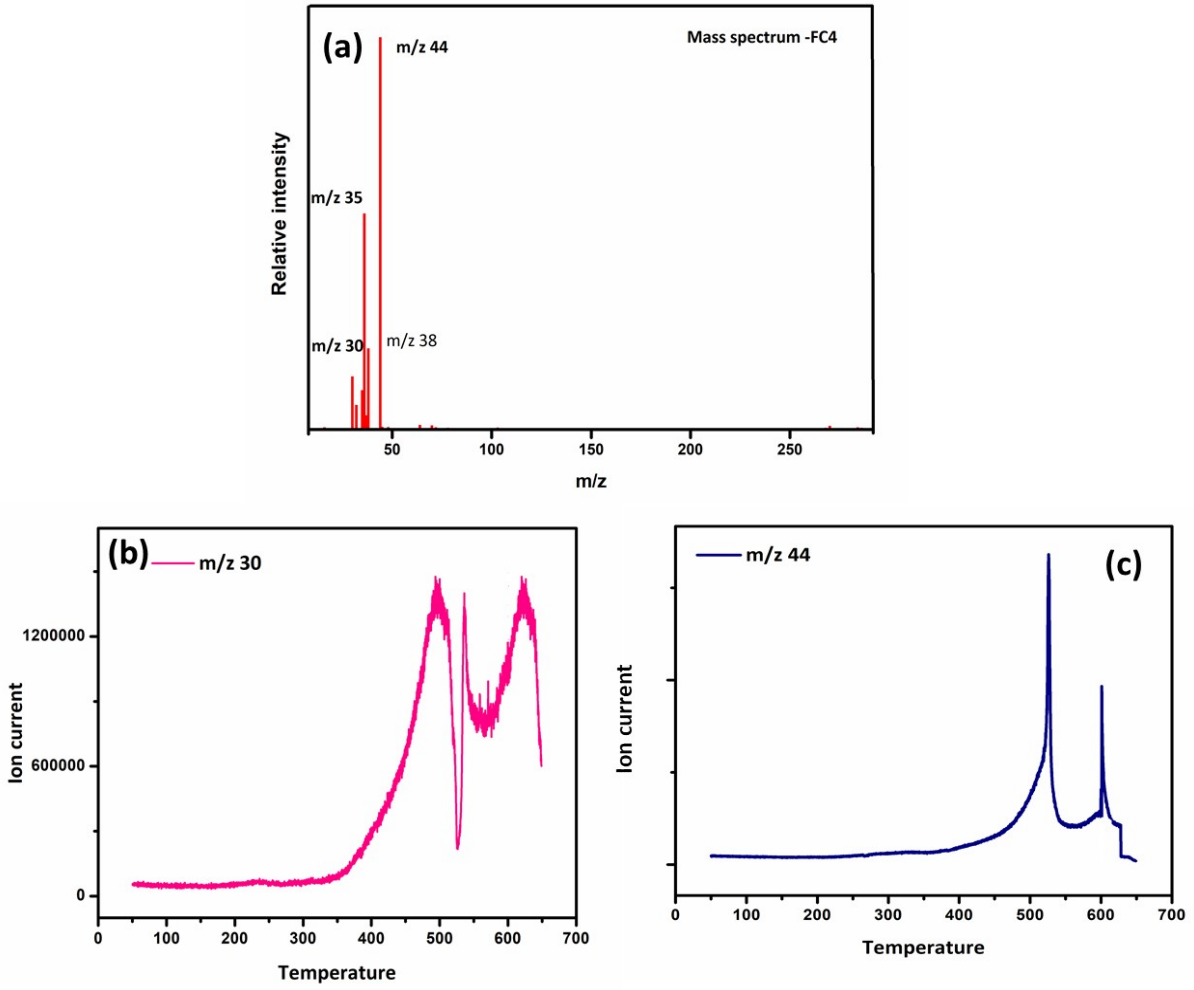


Figure S2. a) Mass spectrum of FC-4 b) Gas chromatogram (m/z=30) c) Gas chromatogram (m/z=44)

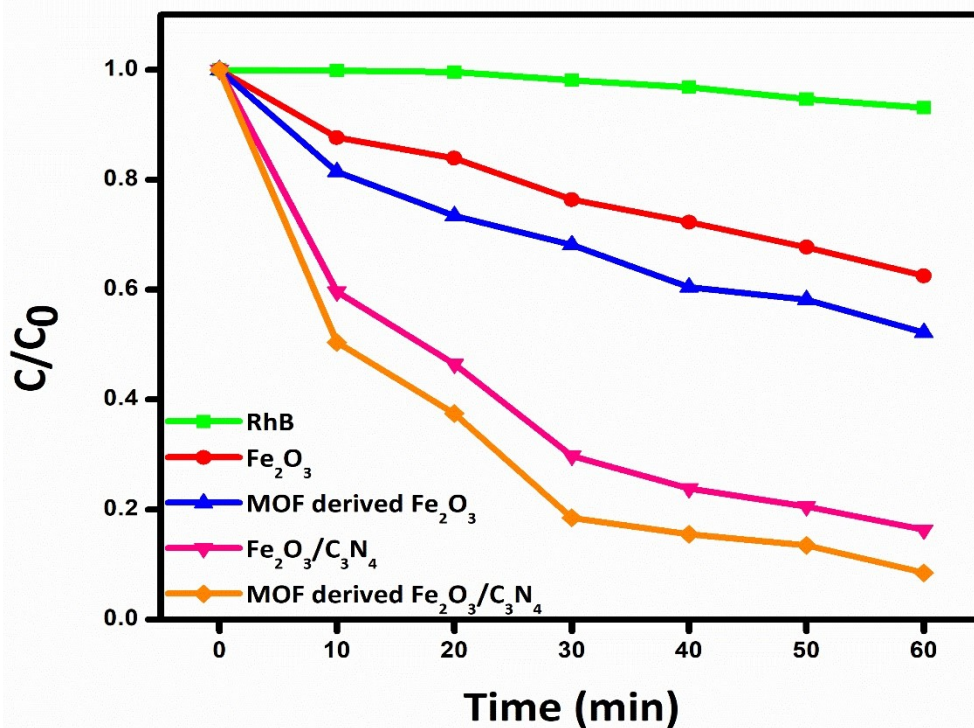


Figure S3. Photocatalytic degradation curves of RhB using  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/C<sub>3</sub>N<sub>4</sub> and MOF-derived  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/C<sub>3</sub>N<sub>4</sub>

Parameters	Fe <sub>2</sub> O <sub>3</sub>	g-C <sub>3</sub> N <sub>4</sub>
Bandgap E <sub>g</sub> (eV)	2.0	2.97
Absolute electronegativity $\chi$ (eV)	5.82	4.72
Free electrons energy E <sup>c</sup> (eV)	4.50	4.50
Valence band position (E <sub>VB</sub> ) (eV)	+2.32	+1.705
Conduction band position E <sub>CB</sub> (eV)	+0.32	-1.265

Table S3. Bandgap, Conduction and Valence band values of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and C<sub>3</sub>N<sub>4</sub>