

Controllable Synthesis of Na, K-based Titanium Oxide Nanoribbons as Functional Electrode for Supercapacitor and Separation of Aqueous Ions

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Figure S1. Calibration curve of conductivity to eight saline solution concentration.

Figure S2. Morphological characterization of MTO materials.

Figure S3. Pore width distribution of LTO and KTO materials.

Figure S4. Long-term charge/discharge cycles KTO electrode.

Figure S5. Morphology of KTO after cycling of SEM images.

Figure S6. Schematic illustration of asymmetric CDI device.

Figure S7. Electrosorption/desorption rate performance of CDI device.

Table S1. Values of specific capacitance of different electrodes.

Table S2. Comparison with reported works for the electrochemical performance

Table S3. Comparison with reported works for the desalination performance.

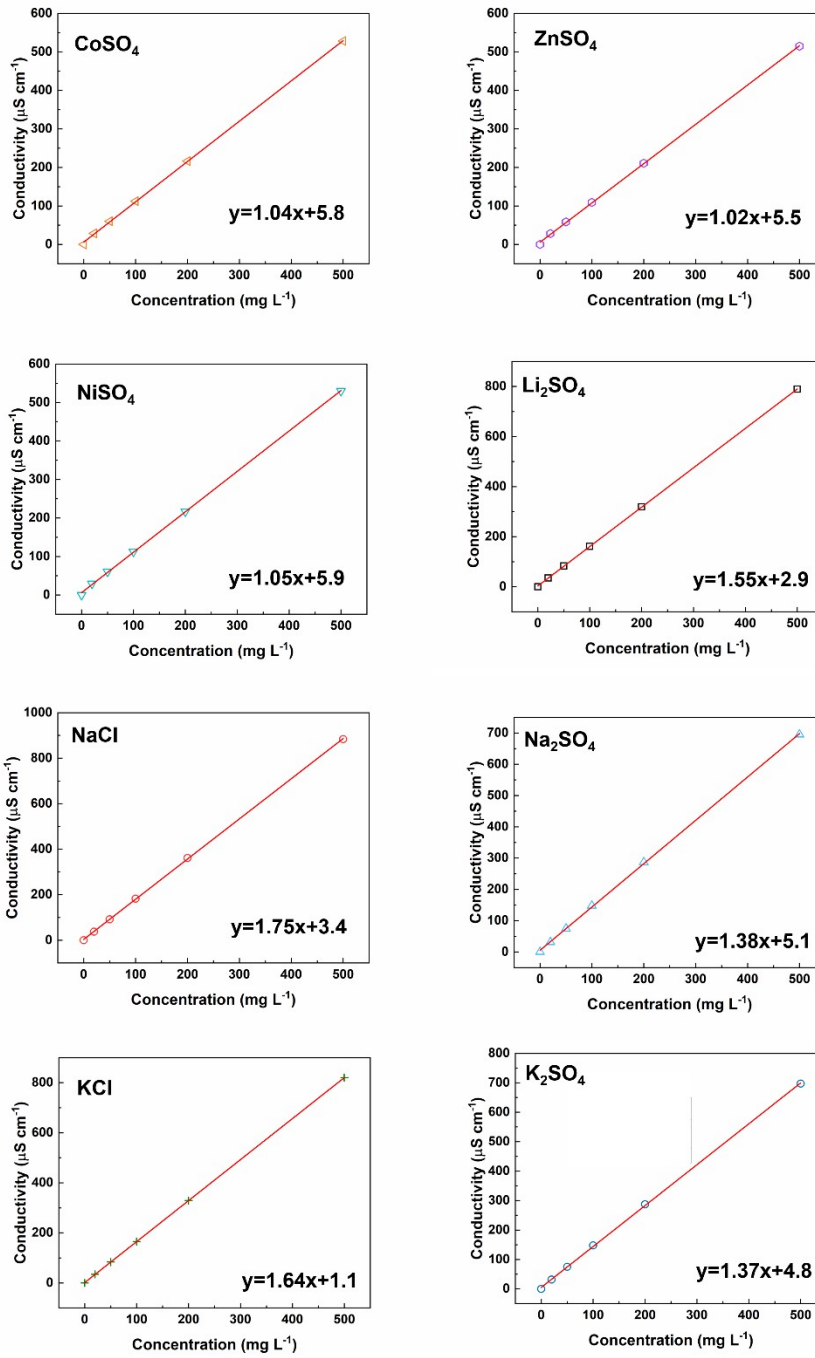


Figure S1. Calibration curve of conductivity to eight saline solution concentration.

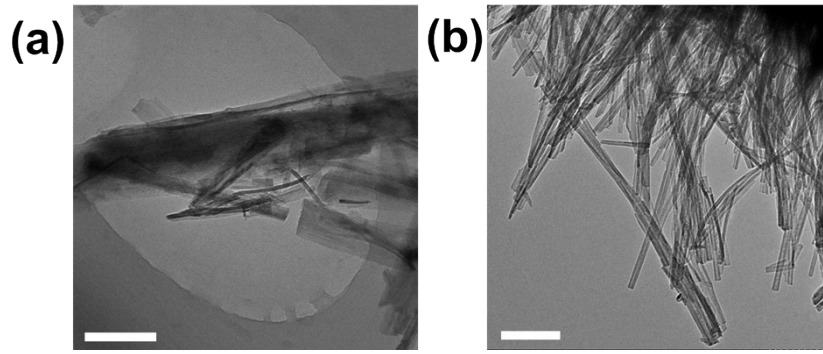


Figure S2. Morphological characterization of MTO materials: (a) KTO; (b)NTO whisker TEM images.

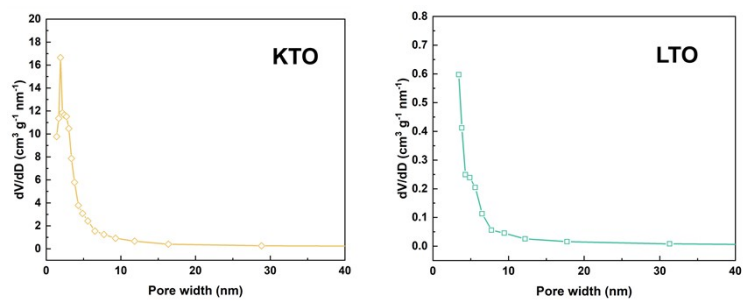


Figure S3. Pore width distribution of LTO and KTO materials.

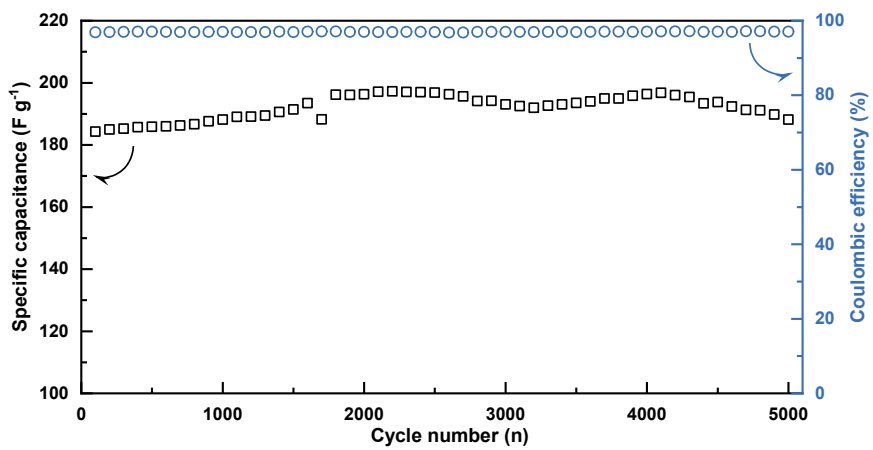


Figure S4. Long-term charge/discharge cycles KTO electrode.

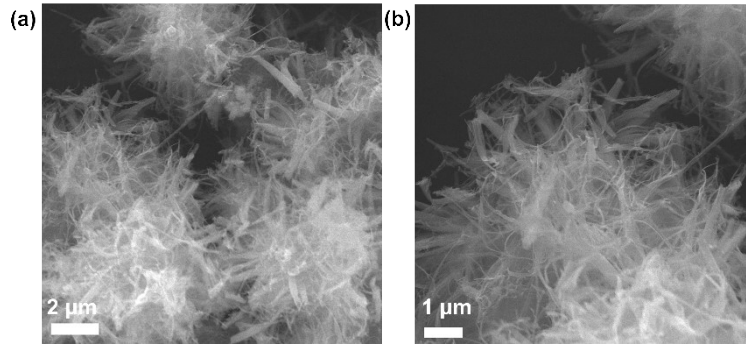


Figure S5. Morphology of KTO after cycling of SEM images.

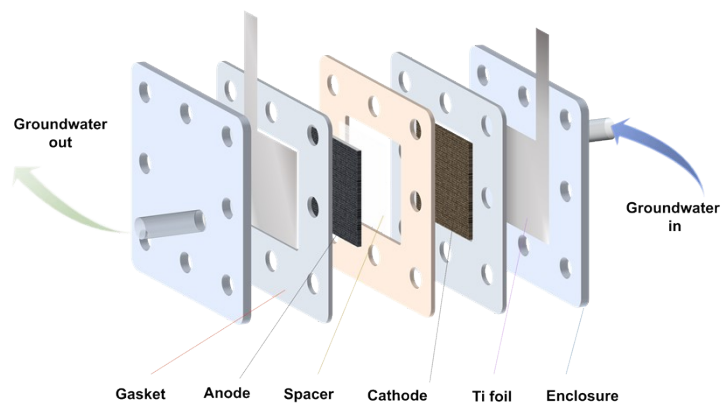


Figure S6. Schematic illustration of asymmetric CDI device.

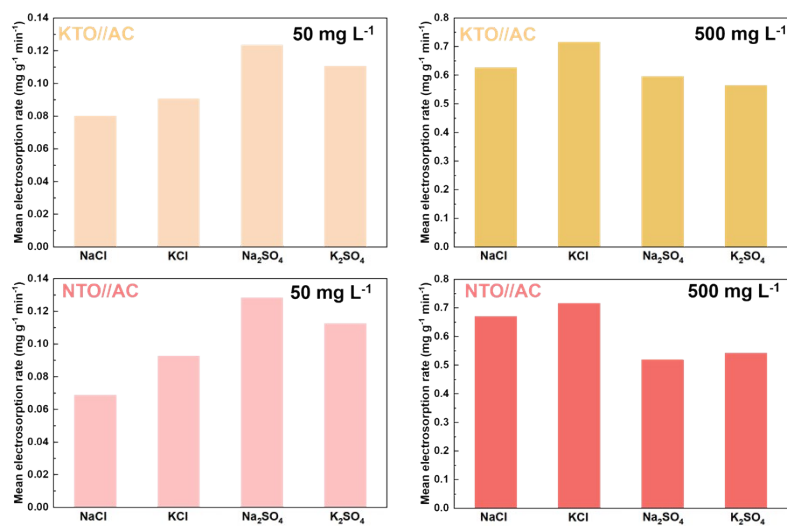


Figure S7. Electrodesorption/desorption rate performance of NTO//AC and KTO//AC CDI device in various saline solutions with different concentration.

Table S1 Values of specific capacitance depending on galvanostatic discharge curves of different electrodes.

Current density A g ⁻¹		1	2	5	10	20
Specific capacitance F g ⁻¹	LTO	172.5	155.3	137.1	119.2	93.6
	NTO	182.3	166.1	145.9	128.0	99.8
	KTO	196.6	184.2	159.1	137.0	108.3

Table S2 Comparison of NTO and KTO with reported works for the electrochemical performance

Materials	Specific capacitance (F g ⁻¹)	Capability rate (%)	Electrolyte	Ref.
SWCNTs/TiO ₂ nanocomposite	144	91	PVA/H ₂ SO ₄	1
Mn ₃ O ₄ nanorod	136.5	36.6	Na ₂ SO ₄	2
C-TiO ₂	214	67.3	Na ₂ SO ₄	3
TiO ₂ @BAC	402.5	21.1	KOH	4
TiO ₂ -CNT	345.7	43.1	H ₂ SO ₄	5
Sn-doped TiO ₂ NTAs	329	-	H ₂ SO ₄	6
Na ₂ Ti ₃ O ₇ /SWCNTs	576.01		KOH	7
TiO ₂ /CDC	173.2	60	Na ₂ SO ₄	8
NTO	182.3	54.7	KOH	This work
KTO	196.6	55.1	KOH	This work

Table S3 Comparison of different desalination materials with reported works for the desalination performance

Electrode	Saline solution	Capacity	Work voltage	Ref.
ACC:ZnO	NaCl , 100 mg L ⁻¹	8.5 mg g ⁻¹	1.2 V	11
ZnO/AC	NaCl , 500 mg L ⁻¹	9.4 mg g ⁻¹	1.2 V	12
Gr/SnO ₂	NaCl , 30 mg L ⁻¹	1.49 mg g ⁻¹	1.4 V	13
Fe ₃ O ₄ /rGO	CaCO ₃ , 35 mg L ⁻¹	4.3 mg g ⁻¹	1.5 V	14
Zr/ACC	NaCl , 5.8 mg L ⁻¹	2.05 mg g ⁻¹	1.0 V	15
MoS ₂	NaCl , 400 mg L ⁻¹	8.81 mg g ⁻¹	1.2 V	16
TiO ₂ /CNTs	NaCl , 500 mg L ⁻¹	4.3 mg g ⁻¹	1.2 V	17
TiO ₂ NTs/HCF/CF	Na ₂ SO ₄ , 55 mg L ⁻¹	1.69 mg g ⁻¹	1.6 V	18
Ti-C	NaCl , 50 mg L ⁻¹	7.11 mg g ⁻¹	1.4 V	19
ZnO/ACC	NaCl , 1660 mg L ⁻¹	5.72 mg g ⁻¹	1.5 V	20
rGO@MnO ₂	NaCl , 50 mg L ⁻¹	3.50 mg g ⁻¹	1.2 V	21
Carbon aerogel/MnO ₂	NaCl , 1461 mg L ⁻¹	6.4 mg g ⁻¹	1.5 V	22
RuO ₂ (20)-AC	NaCl , 292.2 mg L ⁻¹	11.3 mg g ⁻¹	1.2 V	23
MO/ACF	Cd(NO ₃) ₂ , 18 mg L ⁻¹	10.5 mg g ⁻¹	1.5 V	24
ACC-TiO ₂	NaCl , 1000 mg L ⁻¹	5.3 mg g ⁻¹	1.5 V	12
TiO ₂ /rGO	NaCl , 500 mg L ⁻¹	25.0 mg g ⁻¹	1.2 V	9
Ti-C	NaCl , 50 mg L ⁻¹	9.61 mg g ⁻¹	1.4 V	19
MnO ₂ /CNT-CS	NaCl , 877 mg L ⁻¹	10.07 mg g ⁻¹	1.6 V	25
Mn@C-400	NaCl , 25 mg L ⁻¹	5.4 mg g ⁻¹	1.2 V	26
CNTs/rGO	NaCl , ~500 mg L ⁻¹	1.4 mg g ⁻¹	1.2 V	10
Ti-ACC	NaCl , 250 mg L ⁻¹	2.05 mg g ⁻¹	1.0 V	27
KTO//AC	KCl , 500 mg L ⁻¹	7.35 mg g ⁻¹	1.6 V	This work

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