

Electronic Supplementary Information

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Water-soluble Al(III), Fe(III) and Cu(II) formazanates: synthesis, structure, applications in alkane and alcohol oxidations

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1. Catalysis

Table S1. MW-assisted oxidation of cyclohexane to cyclohexanol and cyclohexanone catalyzed by formazanate metal complexes **1-3**^a.

Entry	Cat.	Cat. (μmol)	CyH (mmol)	Yield (%) ^b			TON ^c	
				K	A	Total		
1	2	0.50	1.0	3.5	19.5	23.0	230	
2		0.50	2.5	1.4	8.4	9.8	490	
5		1.00	0.5	8.4	7.5	15.9	79	
6		1.00	1.0	2.0	26.6	28.6	286	
7		1.00	1.5	1.7	27.2	28.9	433	
8		1.00	2.0	1.9	22.6	24.5	490	
9		1.00	2.5	2.9	8.3	11.2	279	
10		1.50	1.0	9.6	18.9	28.4	189	
11		1.50	2.5	5.1	10.7	15.8	263	
12		1.75	1.0	6.1	17.9	24.0	120	
13		1.75	2.5	4.9	10.3	15.2	224	
14		2.00	1.0	6.2	14.3	20.5	117	
15		2.00	2.5	4.0	5.8	9.8	123	
16		3	0.50	1.0	2.8	15.8	18.6	372
17			0.50	2.5	2.2	5.8	8.0	400
18	1.00		0.5	7.9	7.9	15.8	79	
19	1.00		1.0	3.8	16.6	20.4	204	
20	1.00		1.5	1.6	15.1	16.7	251	
21	1.00		1.75	1.3	12.1	13.4	235	
22	1.00		2.0	1.0	11.0	12.0	240	
23	1.00		2.5	0.5	9.2	9.7	242	
24	1.50		1.0	5.0	12.1	17.1	116	
25	1.50		2.5	1.8	9.4	11.2	164	
26	2.00		1.0	1.4	13.2	14.6	73	
27	2.00		2.5	1.8	8.1	9.9	123	

^a Reaction conditions unless stated otherwise: 3 mL MeCN, farmazan catalyst (0.5-2.0 μmol), cyclohexane (1.0-2.5 mmol), oxidant, H₂O₂ (5 mmol), 80 °C, 20 W, 1 h. ^b Moles of product/100 moles of cyclohexane (A = cyclohexanol and K = cyclohexanone). ^c Overall TON (moles of products/mole of catalyst). GC analysis performed upon addition of PPh₃ (conversion of CyOOH into cyclohexanol prior to GC analysis).

2. X-ray analysis

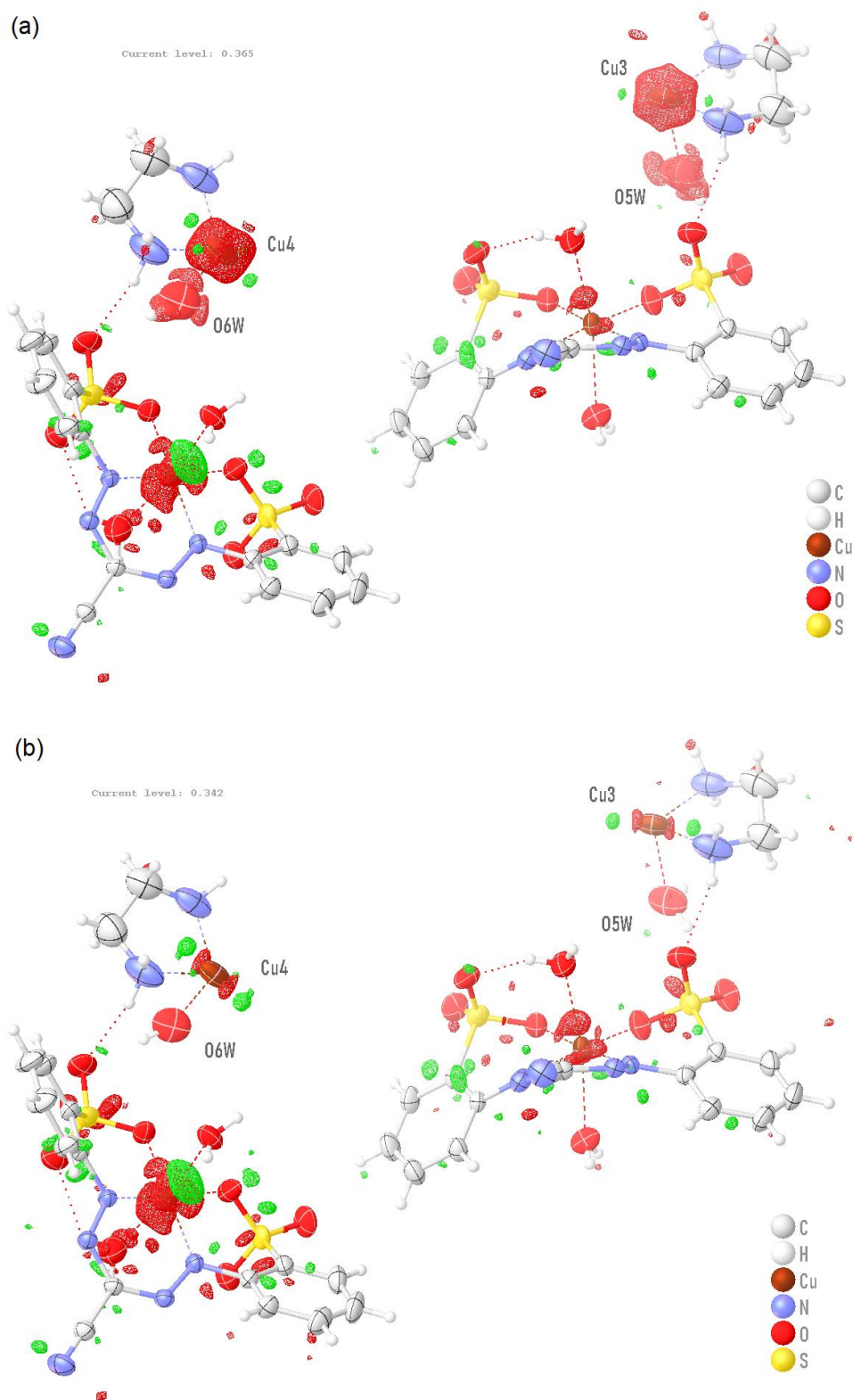


Figure S1. Residual density maps before (a) and after (b) freely refinement of occupancies of Cu3, Cu4 and of the attached water molecules O5w, O6w.