

Supporting Information

Ultrathin mesoporous graphitic carbon nitride nanosheets with functional cyano groups decoration and nitrogen vacancy defects for efficient selective CO₂ photoreduction

Junyi Li ^{a, b, ‡}, Xiaohan Wang ^{a, ‡}, Liang Huang ^a, Liang Tian ^a, Menny Shalom ^b,
Chunyan Xiong ^c, Haijun Zhang ^{a, *}, Quanli Jia^d, Shaowei Zhang ^e, Feng Liang ^{a, *}

a The State Key Laboratory of Refractories and Metallurgy, Wuhan University of
Science and Technology, Wuhan 430081, China

b Department of Chemistry and Ilse Katz Institute for Nanoscale Science and
Technology, Ben-Gurion University of the Negev, Beer-Sheva 8410501, Israel

c School of Chemical Engineering and Pharmacy, Wuhan Institute of Technology,
Wuhan 430074, China

d High Temperature Ceramic Institute, Zhengzhou University, Zhengzhou 450052,
PR China

e College of Engineering, Mathematics and Physical Sciences, University of Exeter,
Exeter Ex4 4QF, U.K.

* Corresponding author. Assoc. Prof. Feng Liang, E-mail: liangfengref@wust.edu.cn;
Prof. Haijun Zhang, E-mail: zhanghaijun@wust.edu.cn, Tel: +86-27-68862829

‡ These authors contributed equally to this work.

Table S1 Textural parameters of nitrogen sorption analysis and elemental analysis for BCN and CNNS.

sample	S_{BET} (m^2/g)	Pore volume (cm^3/g)	Pore size (nm)	Elemental analysis (atom%)				
				C	N	H	O	C/N
BCN	7.0	0.63	5.25	34.1	49.9	13.7	1.1	0.68
CNNS	512.6	1.72	8.95	23.3	32.0	31.5	13.1	0.73

Table S2 Elemental compositions of BCN and CNNS based on XPS results.

	BCN	CNNS
C (at.%)	51.5% C-C (16.3%) C-NH _x (x=1, 2) (14.0%) C≡N (5.4 %) N=C-(N) ₂ (64.3%)	67.7% C-C (64.5%) C-NH _x (5.6%) C≡N (9.1%) N=C-(N) ₂ (20.8%)
N (at.%)	41.7% C=N-C (57.7%) N-(C) ₃ (24.1%) N-H (18.2%)	13.7% C=N-C (55.1%) C ₃ -N (26.1%) C≡N (11.6%) N-H (7.2%)
O (at.%)	6.8% H ₂ O (88.9%) O ₂ (11.1%)	18.6% H ₂ O (46.6%) O ₂ (53.4%)

Table S3 Comparison of photocatalytic CO₂ reduction performance of various materials.

Photocatalysts	Conditions	Rate ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	References	Year
CNNS	H ₂ O, Xe lamp, 5 h ($\lambda \geq 200 \text{ nm}/\lambda \geq 420 \text{ nm}$)	CH ₄ : 50.8/23.0, CO: 5.1/1.9	This work	
BCN	H ₂ O, Xe lamp, 5 h ($\lambda \geq 200 \text{ nm}/\lambda \geq 420 \text{ nm}$)	CO: 2.6/1.1	This work	
N-CQDs-TiO₂	H ₂ O, Xe lamp, 6 h	CH ₄ : 3.98, CO: 6.13	[S1]	2018
g-C₃N₄	H ₂ O, Xe lamp, 5 h ($\lambda > 200 \text{ nm}$)	CH ₄ : 0.24, CO: 2.1	[S2]	2016
P-doped g-C₃N₄	H ₂ O, Vis ($\lambda \geq 420 \text{ nm}$), 4 h	CO: 2.37, CH ₄ : 1.81	[S3]	2018
Co₃O₄/CNS	H ₂ O, Xe lamp, 4 h	CO: 13.31, CH ₄ : 3.17	[S4]	2020
WO₃/g-C₃N₄	H ₂ O, UV ($254 \leq \lambda < 420 \text{ nm}$)/Vis ($\lambda \geq 420 \text{ nm}$), 4 h	CO: 14.60/1.37, CH ₄ : 10.37/0.75	[S5]	2020
Bi₄O₅Br₂	H ₂ O, Xe lamp, 2 h	CO: 3.16, CH ₄ : 0.5	[S6]	2019
Ru/g-C₃N₄	H ₂ O, Xe lamp, 4 h ($420 \leq \lambda < 780 \text{ nm}$)	CO: (4.78), CH ₄ : (0.78)	[S7]	2018

SnS₂/g-C₃N₄	H ₂ O, Xe lamp, 4 h (λ > 420 nm)	CH ₄ : 0.64, CH ₃ OH: 2.3	[S8]	2017
PdO/TiO₂	H ₂ O, UV light, 2 h	CO: 0.12, CH ₄ : 13.99	[S9]	2019
O-doped g-C₃N₄	H ₂ O, Xe lamp (λ > 420 nm)	CH ₃ OH: 0.88	[S10]	2017
MnO₂/g-C₃N₄	H ₂ O, Xe lamp, 6 h	CO: 3.4	[S11]	2017
NiO/g-C₃N₄	H ₂ O, Xe lamp, 8 h	CO: 4.17	[S12]	2018
Pt@Bi-TiO₂	H ₂ O, Hg lamp, 10 h	CH ₄ : 2.06	[S13]	2020
Au/C₃N₄	H ₂ O, Xe lamp, 2 h	CO : 6.59, CH ₄ : 1.55	[S14]	2018
Bi₂MoO₆	H ₂ O, Xe lamp, 6 h	CO : 3.62	[S15]	2019
Ni-Bi co-doped TiO₂	H ₂ O, 250W Hg lamp, 10 h	CH ₄ : 2.11	[S16]	2020

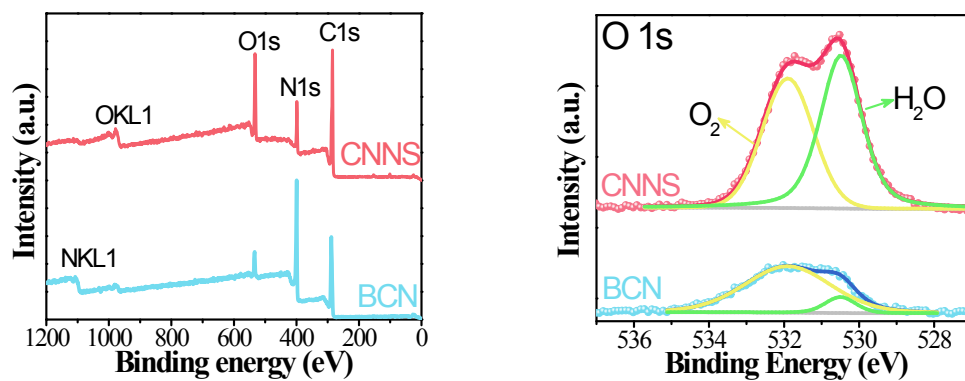


Figure S1 XPS wide spectra and high resolution O 1s XPS spectra of BCN and CNNS.

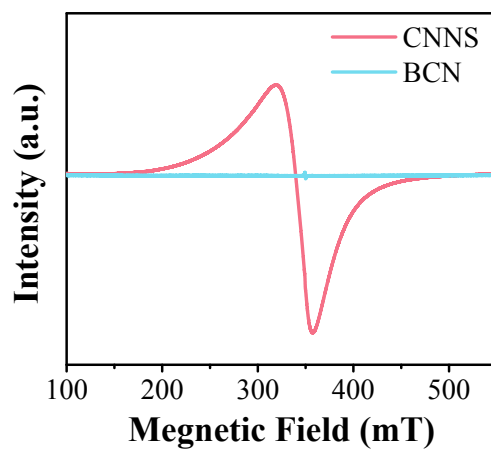


Figure S2 EPR spectra of BCN and CNNS.

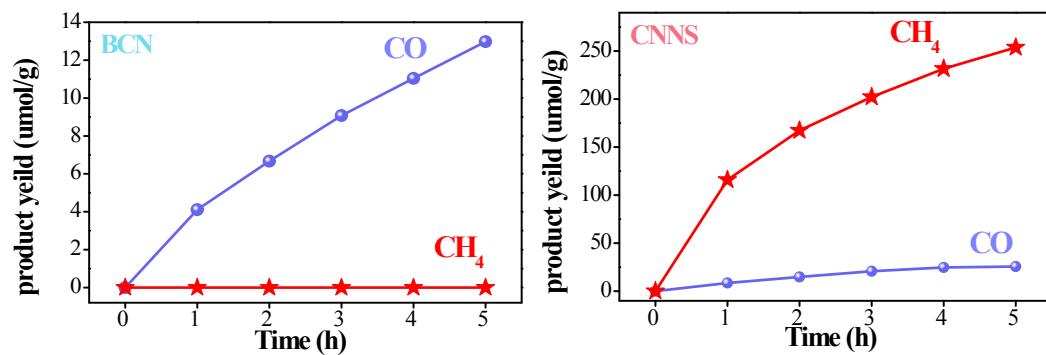


Figure S3. Time-production plots of the CO and CH₄ generated from the photocatalytic CO₂ reduction over the BCN (left) and CNNS (right). (photocatalysis condition: 0.05 g photocatalysts, 5 mL H₂O, Light $\lambda \geq 200$ nm, 20°C)

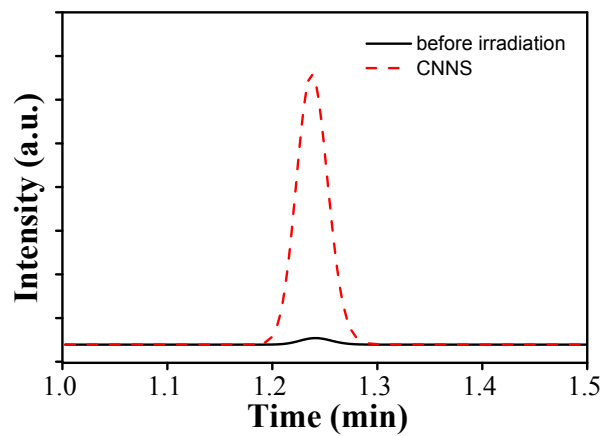


Figure S4. The GC spectra of resulting O₂ after photocatalytic CO₂-reduction reactions (visible-light-driven, $\lambda \geq 420$ nm) over CNNS.

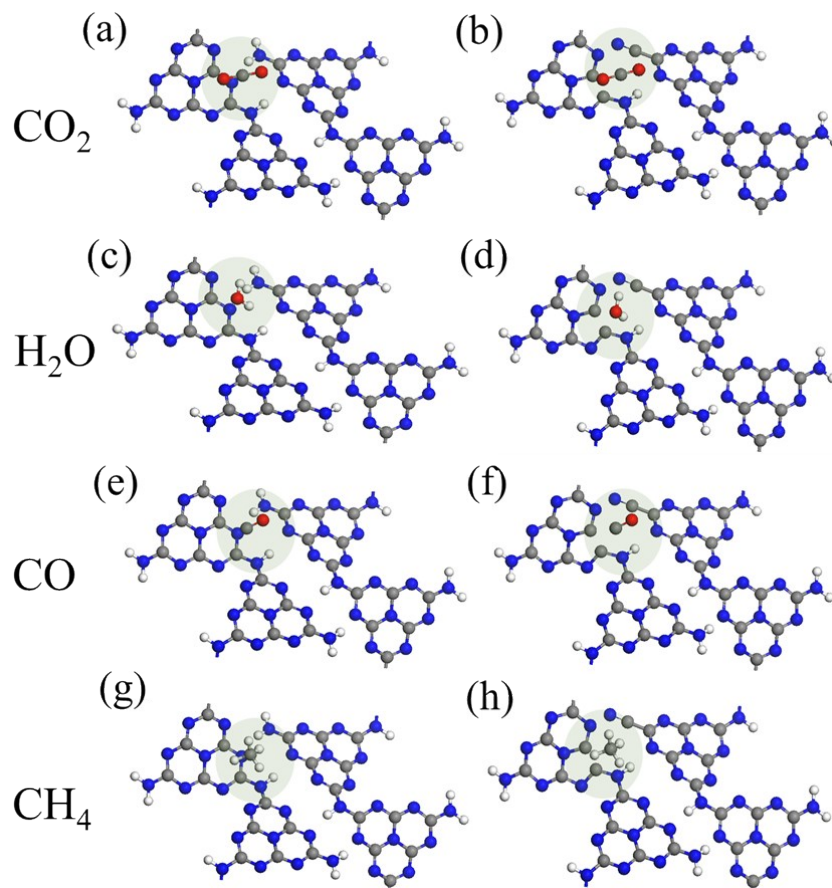


Figure S5 CO₂, H₂O, CO, CH₄ adsorption geometry on BCN (a, c, e, and g) and CNNS with nitrogen vacancy and cyano group (b, d, f, and h).

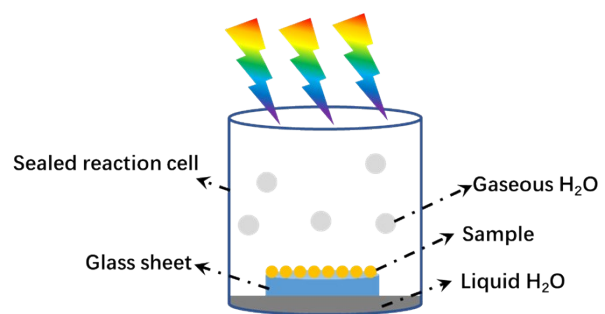


Figure S6 Illustration of the experimental setup for photocatalytic CO₂ reduction.

References

[S1] Li M, Wang M, Zhu L, et al. Facile microwave assisted synthesis of N-rich carbon quantum dots/dual-phase TiO₂ heterostructured nanocomposites with high activity in CO₂ photoreduction[J]. Applied Catalysis B: Environmental, 2018, 231: 269-276.

[S2] Wang H, Sun Z, Li Q, et al. Surprisingly advanced CO₂ photocatalytic conversion over thiourea derived g-C₃N₄ with water vapor while introducing 200–420 nm UV light[J]. Journal of CO₂ Utilization, 2016, 14: 143-151.

[S3] Li M, Zhang L, Fan X, et al. Core-shell LaPO₄/g-C₃N₄ nanowires for highly active and selective CO₂ reduction[J]. Applied Catalysis B: Environmental, 2017, 201: 629-635.

[S4] Hangfan Ma X L, Shiyang Fan , Zhifan Yin , Guoqiang Gan , Meichun Qin , Penglei Wang , Yaxuan Li , Lianzhou Wang In Situ Formation of Interfacial Defects between Co-Based Spinel Carbon Nitride Hybrids for Efficient CO₂ Photoreduction[J]. ACS Applied Energy Materials, 2020, 3(5): 5083–5094.

[S5] Li X, Song X, Ma C, et al. Direct Z-Scheme WO₃/Graphitic Carbon Nitride Nanocomposites for the Photoreduction of CO₂[J]. ACS Applied Nano Materials, 2020, 3(2): 1298-1306.

[S6] Jin X, Lv C, Zhou X, et al. A bismuth rich hollow Bi₄O₅Br₂ photocatalyst enables dramatic CO₂ reduction activity[J]. Nano Energy, 2019, 64: 103955.

[S7] Ye F, Wang F, Meng C, et al. Crystalline phase engineering on cocatalysts: A promising approach to enhancement on photocatalytic conversion of carbon dioxide

- to fuels[J]. Applied Catalysis B: Environmental, 2018, 230: 145-153.
- [S8] Di T, Zhu B, Cheng B, et al. A direct Z-scheme g-C₃N₄/SnS₂ photocatalyst with superior visible-light CO₂ reduction performance[J]. Journal of Catalysis, 2017, 352: 532-541.
- [S9] Mekasuwandumrong O, Jantarasorn N, Panpranot J, et al. Synthesis of Cu/TiO₂ catalysts by reactive magnetron sputtering deposition and its application for photocatalytic reduction of CO₂ and H₂O to CH₄[J]. Ceramics International, 2019, 45(17): 22961-22971.
- [S10] Fu J, Zhu B, Jiang C, et al. Hierarchical Porous O-Doped g-C₃N₄ with Enhanced Photocatalytic CO₂ Reduction Activity[J]. Small, 2017, 13(15).
- [S11] Wang M, Shen M, Zhang L, et al. 2D-2D MnO₂/g-C₃N₄ heterojunction photocatalyst: In-situ synthesis and enhanced CO₂ reduction activity[J]. Carbon, 2017, 120: 23-31.
- [S12] Tang J-y, Guo R-t, Zhou W-g, et al. Ball-flower like NiO/g-C₃N₄ heterojunction for efficient visible light photocatalytic CO₂ reduction[J]. Applied Catalysis B: Environmental, 2018, 237: 802-810.
- [S13] Mohsen Moradi F K, Afsanehsadat Larimi Pt nanoparticles decorated Bi-doped TiO₂ as an efficient photocatalyst for CO₂ photo-reduction into CH₄[J]. Solar Energy, 2020, 211: 100-110.
- [S14] Li H, Gao Y, Xiong Z, et al. Enhanced selective photocatalytic reduction of CO₂ to CH₄ over plasmonic Au modified g-C₃N₄ photocatalyst under UV-vis light irradiation[J]. Applied Surface Science, 2018, 439: 552-559.

[S15] Jun Di X Z, Cheng Lian, Mengxia Ji, Jiexiang Xia, Jun Xiong, Wu Zhou, Xingzhong Cao, Yuanbin She, Honglai Liu, Kian Ping Loh, Stephen J. Pennycook, Huaming Li, Zheng Liu. Atomically-thin Bi_2MoO_6 nanosheets with vacancy pairs for improved photocatalytic CO_2 reduction[J]. *Nano Energy*, 2019, 61: 54-59.

[S16] Nematollahi R, Ghotbi C, Khorasheh F, et al. Ni-Bi co-doped TiO_2 as highly visible light response nano-photocatalyst for CO_2 photo-reduction in a batch photo-reactor[J]. *Journal of CO_2 Utilization*, 2020, 41.