Supporting Information

Robust Dirac spin gapless semiconductors in a two-dimensional

oxalate based organic HK lattice

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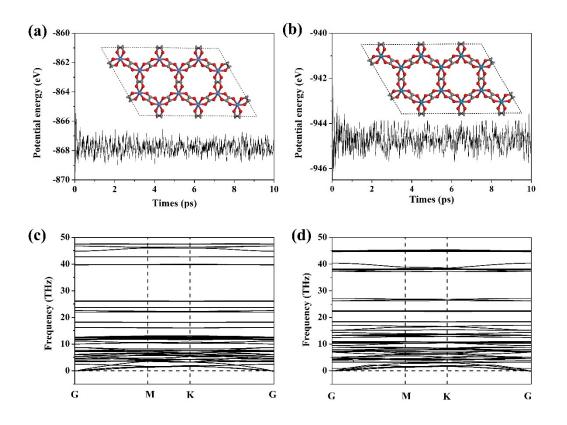


Fig. S1. (a, b) The *ab-initio* molecular dynamic (AIMD) simulation under 300 K with 10 ps timescale, and (c, d) the phonon spectrum of $Ni_2(C_2O_4)_3$ (left) and $Re_2(C_2O_4)_3$ (right), respectively.

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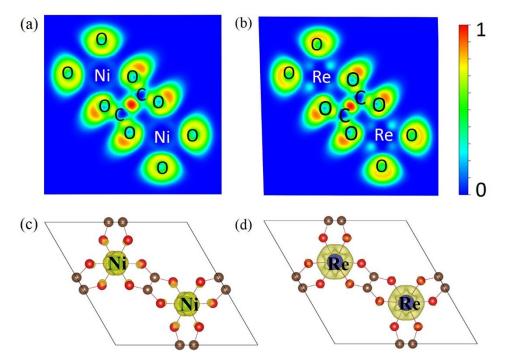


Fig S2. (a, b) The 2D plane ELF of $Ni_2(C_2O_4)_3$ and $Re_2(C_2O_4)_3$ with isosurface 0.4e/Born³. The red and the blue parts denote the electron accumulation and dispersion, respectively. (c, d) The spin charge density of $Ni_2(C_2O_4)_3$ (left) and $Re_2(C_2O_4)_3$ (right). Here the yellow area represents the spin charge localized region.

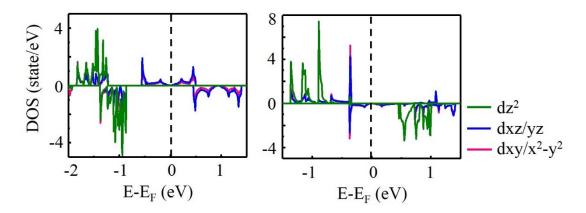


Fig S3. The PDOS of *d* orbitals of Ni and Re in $Ni_2(C_2O_4)_3$ (left) and $Re_2(C_2O_4)_3$ (right), respectively. The Fermi level (E_F) is set to zero.

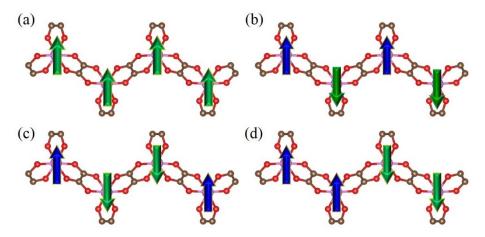


Fig. S4. Four possible magnetic configurations: (a) ferromagnet (FM), (b) Neel antiferromagnet (NAFM), (c) strip antiferromagnet (SAFM) and (d) zigzag antiferromagnet (ZAFM) for $TM_2(C_2O_4)_3$.

Scheme S1. To assess the experimental feasibility of 2D $TM_2(C_2O_4)_3$, the 3×3 supercell of Ag (111) subtract and 4×4 supercell of graphene subtract are screened to grow Ni₂(C₂O₄)₃ and Re₂(C₂O₄)₃. Their lattice mismatch is about 1.25% and 3.69%, respectively. Based on the first-principle calculations, the distance between $TM_2(C_2O_4)_3$ and substrate are about 2.5 Å and 3.3 Å for Ni₂(C₂O₄)₃ and Re₂(C₂O₄)₃. Meanwhile, the structures and magnetic properties are well retained.

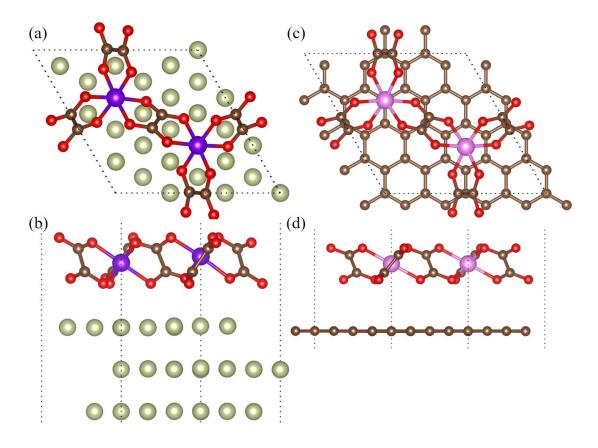


Fig. S5. (a) Top view and (b) side view of $Ni_2(C_2O_4)_3$ on Ag(111) substrate, and (c) top view and (d) side view of $Re_2(C_2O_4)_3$ on graphene monolayer.

Scheme S2. Considering the first and second nearest-neighbor magnetic exchange interactions, the Hamiltonian of a classical Heisenberg model can be written as Eq. (S1)

$$H = -\sum_{mn} J_1 M_m M_n - \sum_{kl} J_2 M_k M_l - \sum_{ij} J_3 M_i M_j$$
(S1)

where J_1 , J_2 and J_3 are the first, second and third nearest-neighbor exchange parameters, respectively; the magnetic moments for per TM in 2D Ni₂(C₂O₄)₃ and Re₂(C₂O₄)₃ are 0.83 μ_B and 1.82 μ_B . Based on the calculated total energies of different magnetic configurations, three different magnetic configurations are chosen.

The exchange parameters J_1 , J_2 and J_3 for Ni₂(C₂O₄)₃ can be derived by following equations

$$E(FM) = -(6J_1 + 12J_2 + 6J_3) M^2$$

$$E(AFM_1) = -(2J_1 - 4J_2 - 6J_3) M^2$$

$$E(AFM_2) = -(-6J_1 + 12J_2 - 6J_3) M^2$$

$$E(AFM_3) = -(-2J_1 + 12J_2 - 6J_3) M^2$$

$$2E_4 + 224E_2 + 192E_2$$

$$J_1 = \frac{-192E_1 + 224E_2 + 192E_3}{3456M^2}$$

$$J_2 = \frac{96E_1 - 112E_{2\mathbb{Z}} + 120E_3}{3456M^2}$$

$$J_3 = \frac{192E_1 + 64E_{2\mathbb{Z}} - 192E_3}{3456M^2}$$

Here, $E_1 = E(AFM_1) - E(FM)$, $E_2 = E(AFM_2) - E(FM)$ and $E_3 = E(AFM_3) - E(FM)$, M= 0.83 μ_B . For $\text{Re}_2(\text{C}_2\text{O}_4)_3$, it only has one kind of AFM configuration (Néel AFM as shown in Fig. S4(b)) and other two configurations are ferrimagnetic. Only considering one AFM configuration, the expression of E(FM) and E(AFM) can be written as:

$$E(FM) = -6J_1 M^2$$

$$E(AFM) = -2J_1 M^2$$

and

$$J = \frac{E}{4M^2}$$

Here, E = E(AFM) - E(FM), $M = 1.82 \mu_B$.

The calculated J_1 for Re₂(C₂O₄)₃ is 3.73 meV. The J_1 , J_2 and J_3 for Ni₂(C₂O₄)₃ are 69.75 meV, -1.38 meV, and -6.28 meV, respectively, which show that the second and third neighbor coupling are very weak and can be neglected, and first neighbor magnetic exchange couplings are FM coupling.