

## **Electronic Supporting Information**

### **Organic-inorganic nanohybrids based on AIE luminogens- functional polymer and CdTe/ZnS QDs: morphologies, optical properties, and applications**

Bingfeng Shi,<sup>a</sup> Jianhua Lü,<sup>b</sup> Ying Liu,<sup>a</sup> Yang Xiao<sup>a</sup> and Changli Lü<sup>\*a</sup>

<sup>a</sup>*Institute of Chemistry, Northeast Normal University, Changchun 130024, P. R. China.*

<sup>b</sup>*Narcotics Control School, Yunnan Police College, Kunming 650223, P. R. China.*

\* Corresponding author.

*E-mail addresses:* [lucl055@nenu.edu.cn](mailto:lucl055@nenu.edu.cn) (C. Lü).

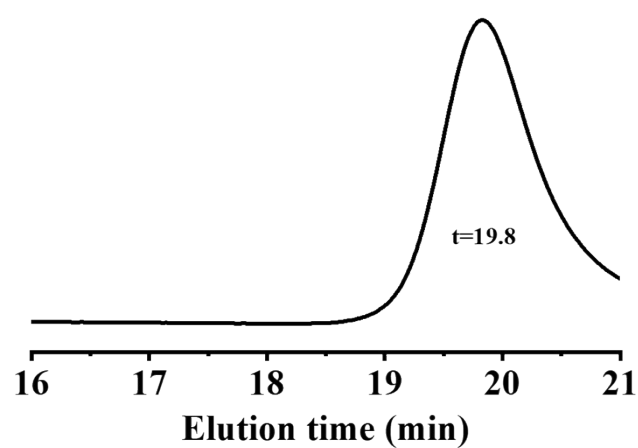


Fig. S1 GPC trace of T-PNI polymer.

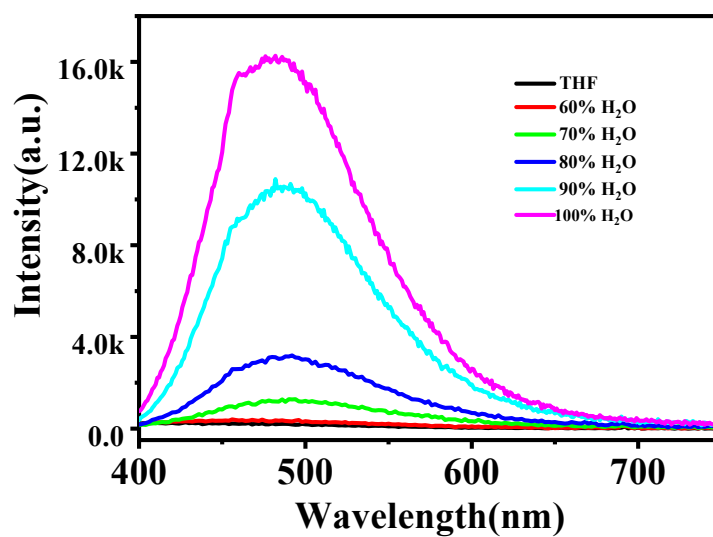


Fig. S2 PL spectra of T-PNI in THF/water mixtures with different water contents ( $1 \text{ mg} \cdot \text{mL}^{-1}$ ),  $\lambda_{\text{ex}} = 346 \text{ nm}$ .

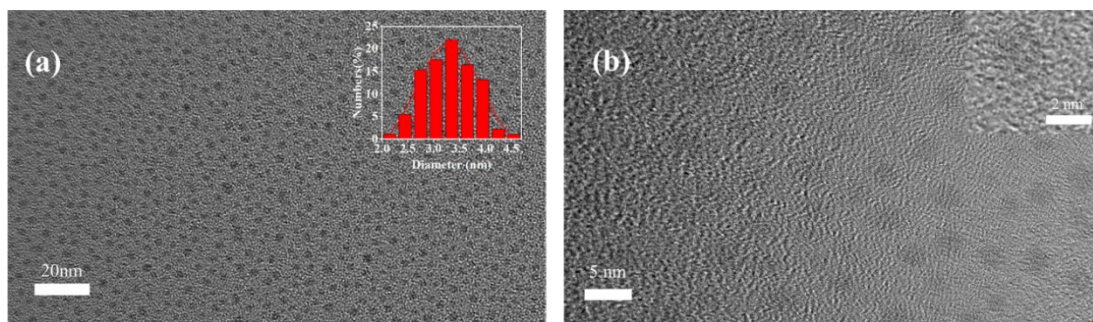


Fig. S3 (a) TEM image of CdTe/ZnS QDs in water and their size-distribution histogram (inset); (b) HRTEM image of CdTe/ZnS QDs.

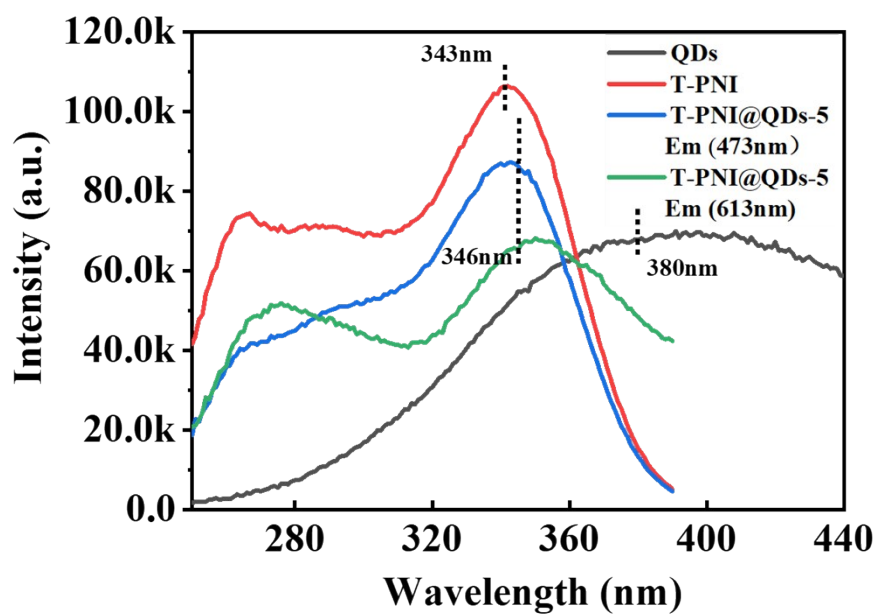


Fig. S4 PL excitation spectra of T-PNI, red-emitting CdTe@ZnS QDs, and T-PNI@QDs systems in water at 25 °C.

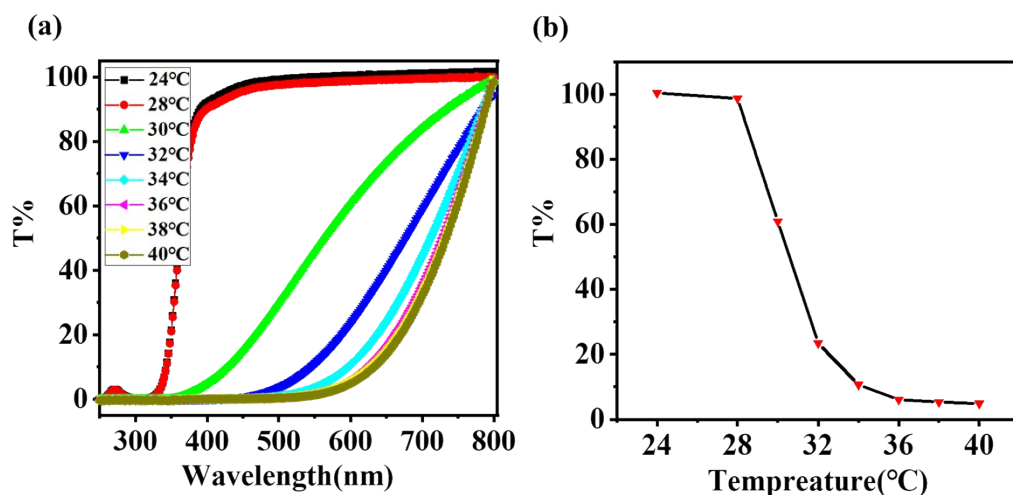


Fig. S5 Optical transmittance (a) and the relative transmittance at 600 nm (b) of aqueous solutions of T-PNI at different temperatures

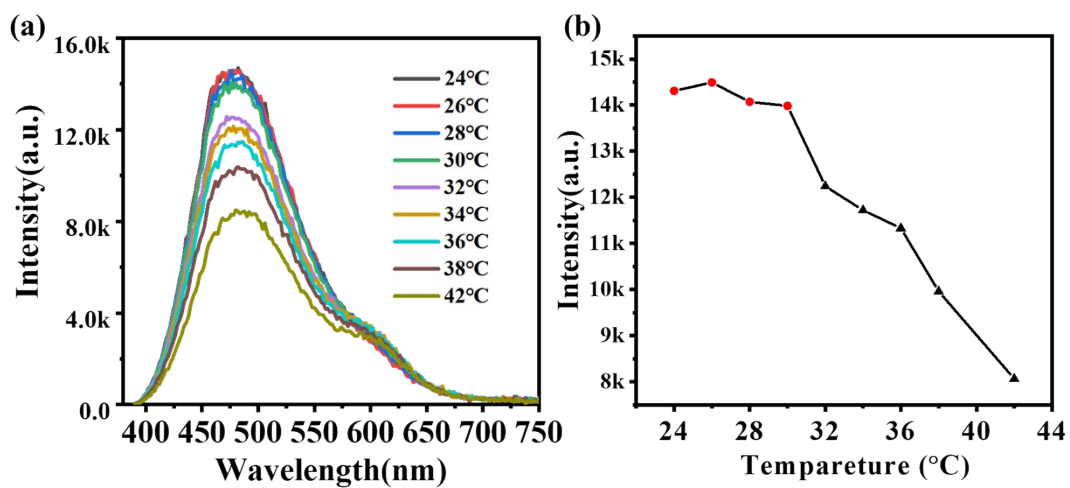


Fig. S6 (a) PL emission spectra of T-PNI at different temperatures. (b) Temperature dependence PL intensity of T-PNI.

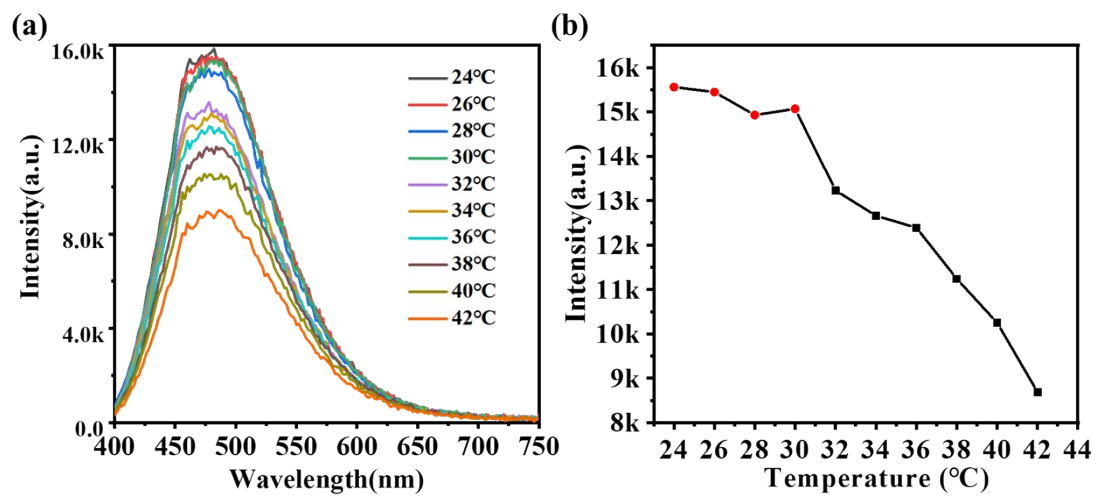


Fig. S7 (a) PL emission spectra of T-PNI-SH at different temperatures. (b) Temperature dependence PL intensity of T-PNI-SH.

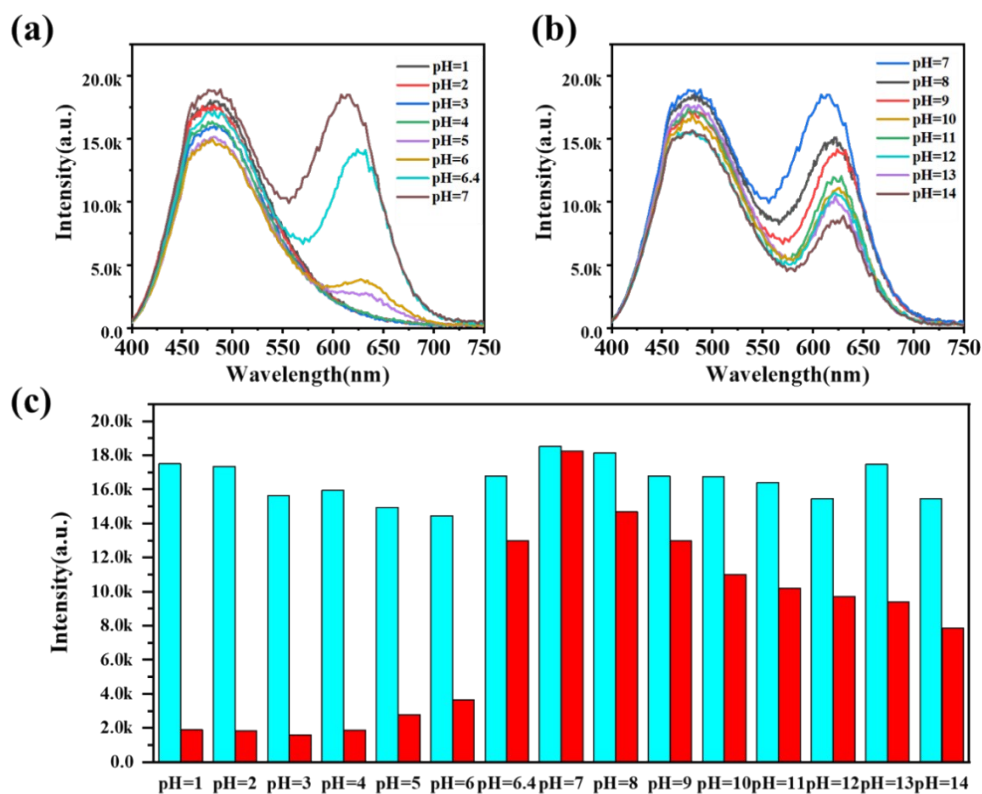


Fig. S8 PL intensity of T-PNI@QDs-5 at different pH values: (a) pH=1-7; (b) pH=7-14. (c) PL intensities of T-PNI@QDs-5 in different pH values. Blue bars: intensity at 473 nm; red bars: intensity at 613 nm.

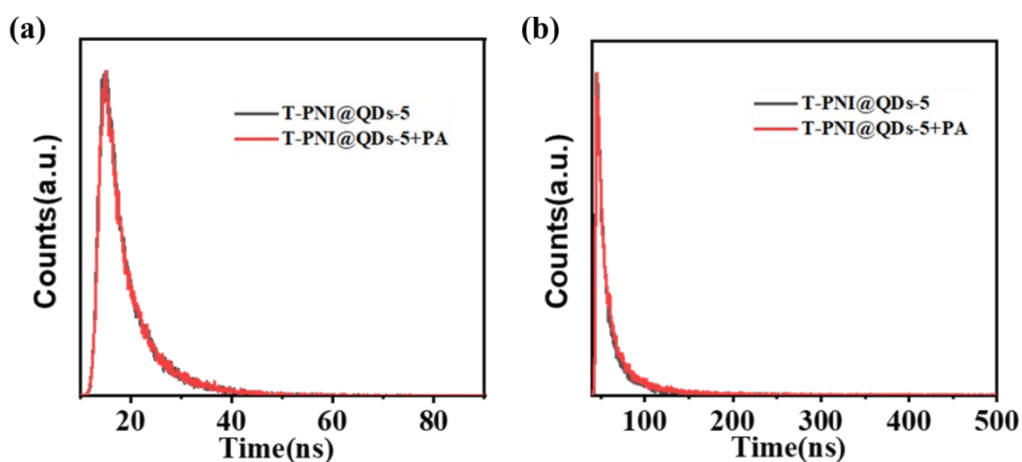
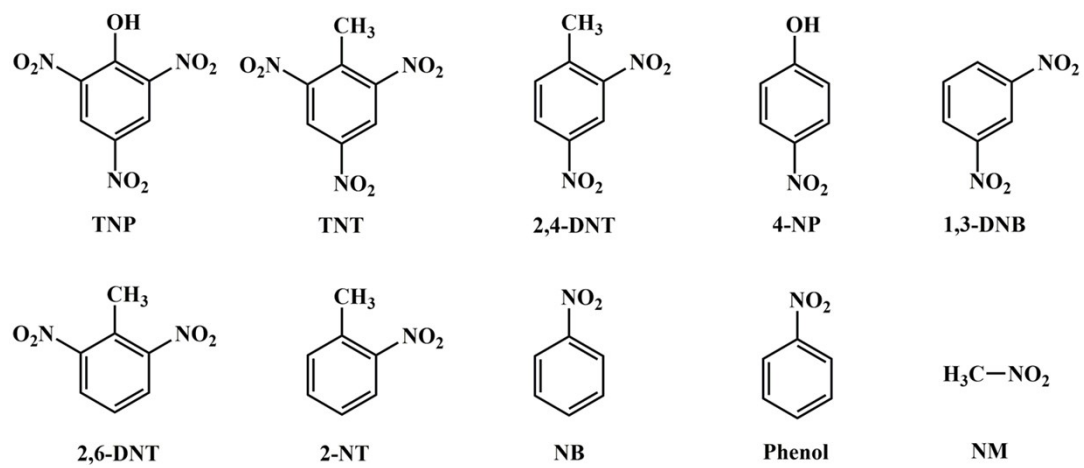


Fig. S9 (a) Fluorescence decay curves of blue-light of T-PNI@QDs-5 and T-PNI@QDs-5+PA; (b) Fluorescence decay curves of red-light of T-PNI@QDs-5 and T-PNI@QDs-5+PA.



Scheme S1 Structures of different nitro-compounds used in the experiments.

Table S1 A comparative study of the K<sub>sv</sub>, detection limit and medium used for PA detection of some recent representative reports.

Publication	Material used	K <sub>sv</sub> (M <sup>-1</sup> )	Detection limit	Medium Used
Present work	T-PNI@QDs hybrid nanostructure	$2.67 \times 10^4$	4.09 $\mu$ M (blue light)	Water
		$2.89 \times 10^4$	3.79 $\mu$ M (red light)	
<i>Sensor Actuat. B Chem.</i> , <b>2017</b> , 248, 223	Antipyrine Schiff base AIE sensor	$1.91 \times 10^5$	19.1 $\mu$ M	Water
<i>J. Mol. Liq.</i> , <b>2018</b> , 262, 446	9-Anthraldehyde-based AIE sensor	$1.89 \times 10^5$	8.07 $\mu$ M	Water
<i>Chem. Commun.</i> , <b>2016</b> , 52, 11284	TPE functionalized metal–organic framework	$2.8 \times 10^4$	—	Methanol
<i>RSC Adv.</i> , <b>2015</b> , 5, 76670	TPE-based oxacalixarenes	$1.7 \times 10^4$	0.1 mM	Water/THF (95/5)
<i>Polym. Chem.</i> , <b>2018</b> , 9, 3158	CdTe QDs/block copolymer hybrid assemblies	—	1.27 $\mu$ M	Water
<i>Spectrochim. Acta A Mol. Biomol. Spectrosc.</i> , <b>2019</b> , 222, 117168	Barbituric acid derivatives	$4.1 \times 10^4$	2.4 $\mu$ M,	Water/THF (90/10)
<i>Spectrochim. Acta A Mol. Biomol. Spectrosc.</i> , <b>2020</b> , 233, 118221	Amine functionalized CdSe@SiO <sub>2</sub> NPs	$4.0 \times 10^4$	0.05 $\mu$ M,	Water
<i>Mater. Chem. Phys.</i> , <b>2021</b> , 260, 124130	Fe <sub>2</sub> O <sub>3</sub> -CdSe nanocomposite	$4.3 \times 10^4$	2.2 $\mu$ M	DMSO