Electronic Supplementary Material (ESI) for Inorganic Chemistry Frontiers. This journal is © the Partner Organisations 2021

Hollow Bimetallic Selenide Derived from Hierarchical MOF-based Prussian

Blue Analogue for Urea Electrolysis

Huizhu Xu, Ke Ye*, Kai Zhu, Yinyi Gao, Jinling Yin, Jun Yan, Guiling Wang,

 $Dianxue\ Cao^*$

Key Laboratory of Superlight Materials and Surface Technology of Ministry of

Education, College of Materials Science and Chemical Engineering, Harbin

Engineering University, Harbin 150001, China



Figure S1. HRTEM image of PBA@MOF-Ni/Se.

^{*}Corresponding authors.

E-mail addresses: yekehrbeu@163.com (K. Ye); caodianxue@hrbeu.edu.cn (D. Cao)



Figure S2. XPS spectra of (a) Co 2p; (b) N 1s; (c) C 1s of PBA@MOF-Ni and

PBA@MOF-Ni/Se.



Figure S3. (a) Polarization curves of PBA@MOF-Ni/Se under different reaction conditions towards HER in 1.0 M KOH with 0.5 M urea; (b) Polarization curves of PBA@MOF-Ni/Se under different reaction conditions towards UOR in 1.0 M KOH with 0.5 M urea.



Figure S4. LSV curves towards HER of PBA@MOF-Ni/Se and commercial Pt/C.



Figure S5. CV curves of (a) PBA@MOF-Ni/Se; (b) PBA@Ni(OH)₂/Se; (c) MOF-Ni/Se; (d) Ni foam; (e) ECSA evaluation towards HER in 1.0 M KOH with 0.5 M urea.



Figure S6. CV curves of (a) PBA@MOF-Ni/Se; (b) PBA@Ni(OH)₂/Se; (c) MOF-Ni/Se; (d) Ni foam; (e) ECSA evaluation towards UOR in 1.0 M KOH with 0.5 M urea.



Figure S7. Chronopotentiometry curve of PBA@MOF-Ni/Se towards UOR at 10 mA cm⁻² for 20h.



Figure S8. XPS spectra of (a) Ni 2p (b) Se 3d and (c) Co 2p of PBA@MOF-Ni/Se before and after UOR.



Figure S9. XRD pattern of PBA@MOF-Ni/Se after CP test towards UOR.



Figure S10. (a) SEM image of PBA@MOF-Ni/Se after CP test towards UOR; (b)

TEM image of PBA@MOF-Ni/Se after CP test towards UOR.

Electrocatalysts	Electrolyte	Potential (V vs. RHE) at 10 mA cm ⁻²	Ref
PBA@MOF-Ni/Se	1.0 M KOH 0.5 M Urea	1.319 V	This work
CoS ₂	1.0 M KOH 0.3 M Urea	1.4 V	1
Ni ₂ P/Fe ₂ P	1.0 M KOH 0.5 M Urea	1.36 V	2
Ni/NiO	1.0 M KOH 0.33 M Urea	1.33 V	3
MnO ₂ /MnCo ₂ O ₄	1.0 M KOH 0.5 M Urea	1.33 V	4
NiMoO ₄ ·H ₂ O	1.0 M KOH 0.5 M Urea	1.35 V	5
Ni MOF	1.0 M KOH 1.0 M Urea	1.36 V	6

 Table S1. Comparison of the electrocatalytic activity of PBA@MOF-Ni/Se towards

 UOR in alkaline media with catalysts reported previously.

HC-NiMoS/Ti	1.0 M KOH 0.5 M Urea	1.38 V	7
NF-G-Mn	1.0 M KOH 0.5 M Urea	1.33 V	8
CoN NF/NF	1.0 M KOH 0.5 M Urea	1.342 V	9
Ni _(10%) Pd _(10%) /OMC	1.0 M KOH 0.33 M Urea	1.346 V	10

 Table S2. Comparison of the electrocatalytic performance of PBA@MOF-Ni/Se ||

 PBA@MOF-Ni/Se towards overall urea electrolysis in alkaline media with catalysts

 reported previously.

Catalyst	Electrolyte	Cell voltage at 10 mA cm ⁻²	Ref
PBA@MOF-Ni/Se	1.0 M KOH 0.5 M Urea	1.495 V	This work
CoS ₂	1.0 M KOH 0.3 M Urea	1.59 V	1
MnO ₂ /MnCo ₂ O ₄	1.0 M KOH 0.5 M Urea	1.55 V	4
HC-NiMoS/Ti	1.0 M KOH 0.5 M Urea	1.59 V	7
NF-Pt/C	1.0 M KOH 0.5 M Urea	1.63 V	11
NF-Pt/C NF-IrO ₂	1.0 M KOH 0.5 M Urea	1.72 V	11
NiMoS	1.0 M KOH 0.5 M Urea	1.59 V	12

References

S. Wei, X. Wang, J. Wang, X. Sun, L. Cui, W. Yang, Y. Zheng, J. Liu, CoS₂ nanoneedle array on Ti mesh: A stable and efficient bifunctional electrocatalyst for urea-assisted electrolytic hydrogen production, Electrochim. Acta 246 (2017) 776-782.
 L. Yan, Y. Sun, E. Hu, J. Ning, Y. Zhong, Z. Zhang, Y. Hu, Facile in-situ growth of Ni₂P/Fe₂P nanohybrids on Ni foam for highly efficient urea electrolysis, J. Colloid Interf. Sci. 541 (2019) 279-286

 [3] B. Zhang, S. Wang, Z. Ma, Y. Qiu, Ni0-rich Ni/NiO nanocrystals for efficient water to hydrogen conversion via urea electro-oxidation, App. Surf. Sci. 496 (2019) 143710.

[4] C. Xiao, S. Li, X. Zhang, D. R. MacFarlane, MnO₂/MnCo₂O₄/Ni heterostructure with quadruple hierarchy: a bifunctional electrode architecture for overall urea oxidation, J. Mater. Chem. A 5(2017) 7825-7832.

[5] Z. Yu, C. Lang, M. Gao, Y. Chen, Q. Fu, Y. Duan, S. Yu, Ni–Mo–O nanorodderived composite catalysts for efficient alkaline water-to-hydrogen conversion via urea electrolysis, Energy Environ. Sci. 11(2018) 1890.

[6] V. Maruthapandian, S. Kumaraguru, S. Mohan, V. Saraswathy, S. Muralidharan. An Insight on the Electrocatalytic Mechanistic Study of Pristine Ni MOF(BTC) in Alkaline Medium for Enhanced OER and UOR, ChemElectroChem 5 (2018).

[7] X. Wang, J. Wang, X. Sun, S. Wei, L. Cui, W. Yang, J. Liu, Hierarchical corallike NiMoS nanohybrids as highly efficient bifunctional electrocatalysts for overall urea electrolysis, Nano Res. 11 (2018) 988-996. [8] S. Chen, J. Duan, A. Vasileff, S. Z. Qiao, Size Fractionation of Two-Dimensional Sub-Nanometer Thin Manganese Dioxide Crystals towards Superior Urea Electrocatalytic Conversion, Angew. Chem. Int. Ed. 55 (2016) 3804-3808.

[9] Y. Chen, P. Sun, W. Xing, Cobalt nitride nanoflakes supported on Ni foam as a high-performance bifunctional catalyst for hydrogen production via urea electrolysis, J. Chem. Sci. 131 (2019) 101.

[10]N. Muthuchamy, S. Jang, J. C. Park, S. Park, K. H. Park, Bimetallic NiPd Nanoparticle-Incorporated Ordered Mesoporous Carbon as Highly Efficient Electrocatalysts for Hydrogen Production via Overall Urea Electrolysis, ACS Sustainable Chem. Eng. 7 (2019) 15526-15536.

[11]S. Chen, J. Duan, A. Vasileff, S. Z. Qiao, Size fractionation of two-dimensional sub-nanometer thin manganese dioxide crystals towards superior urea electrocatalytic conversion, Angew. Chem. Int. Ed. 55 (2016) 3804-3808.

[12]X. Wang, J. Wang, X. Sun, S. Wei, L. Cui, W. Yang, J. Liu, Hierarchical corallike NiMoS nanohybrids as highly efficient bifunctional electrocatalysts, Nano Res. 11(2018) 988.