## Supporting Information:

Hetero-Nanostructures Constructed by 2D Porous Metal Oxide/Hydroxide Nanosheets Supported on 1D Hollow Co<sub>9</sub>S<sub>8</sub>

Nanowire for Hybrid Supercapacitors with High Areal Capacity

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Figure S1 (a,b) Morphology of the pristine carbon cloth used as a current collector



Figure S2 (a) The XRD patterns of CC. (a) The XRD patterns of  $h-Co_9S_8$  NWAs. (c) XPS survey spectrum of CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo hetero-nanostructures. (d) XPS spectra of the O1s.



Figure S3 The FT-IR spectrum of CC/h-Co $_9S_8$ /NiCo-Mo hetero-nanostructures.



Figure S4 (a, b) The CV and GCD curves of  $CC/h-Co_9S_8$  electrode, and (c, d) CC/NiCo-Mo electrode in 2 M KOH.



**Figure S5** The electrochemical performance of the CC/AC electrode in 2 M KOH electrolyte: (a) CV curves, (b) GCD curves.



**Figure S6** (a) CV curves of the as-assembled CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo//AC HSC device measured at different operating voltages at the scan rate of 10 mV·s<sup>-1</sup>. (b) GCD curves of the CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo//AC HSC device at different voltages at a specific current of 10 mA·cm<sup>-2</sup>. (c) Specific discharge capacity of CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo//AC HSC device calculated from GCD curves. (d) EIS curves of the CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo//AC HSC device before and after 10000 cycles and inset showing the LED lights illuminated with CC/h-Co<sub>9</sub>S<sub>8</sub>/NiCo-Mo//AC HSC device connected in series.

Electrode	Specific Capacity	Specific	Reference
		Current	
Co <sub>9</sub> S <sub>8</sub> @Ni(OH) <sub>2</sub>	$0.3 \text{ mA h} \cdot \text{cm}^{-2}$	$2 \text{ mA} \cdot \text{cm}^{-2}$	[1]
Co <sub>9</sub> S <sub>8</sub> @Ni(OH) <sub>2</sub>	0.45 mA h·cm <sup>-2</sup>	0.5 mA·cm <sup>-2</sup>	[2]
Co <sub>9</sub> S <sub>8</sub> NTs@NiCo LDH NSs	1.07 mA h·cm <sup>-2</sup>	$2 \text{ mA} \cdot \text{cm}^{-2}$	[3]
Co <sub>9</sub> S <sub>8</sub> @PPy@NiCo-LDH NTAs	0.74 mA h·cm <sup>-2</sup>	1 mA cm <sup>-2</sup>	[4]
NiCo-LDH/Co <sub>9</sub> S <sub>8</sub>	0.95 mA h·cm <sup>-2</sup>	$7 \text{ mA} \cdot \text{cm}^{-2}$	[5]
NiMoCo-LDH	0.61 mA h·cm <sup>-2</sup>	0.6 mA·cm <sup>-2</sup>	[6]
Co(OH) <sub>2</sub> CO <sub>3</sub> @NiCo-LDH	0.33 mA h·cm <sup>-2</sup>	$5 \text{ mA} \cdot \text{cm}^{-2}$	[7]
NiCo-LDH	1.72 mA h·cm <sup>-2</sup>	$2 \text{ mA} \cdot \text{cm}^{-2}$	[8]
CoS <sub>2</sub> @Ni(OH) <sub>2</sub>	0.58 mA h·cm <sup>-2</sup>	0.7 mA·cm <sup>-2</sup>	[9]
C@MoO <sub>2</sub>	$0.42 \text{ mA h} \cdot \text{cm}^{-2}$	0.5 mA·cm <sup>-2</sup>	[10]
MoO <sub>3</sub>	0.25 mA h cm <sup>-2</sup>	1.5 mA cm <sup>-2</sup>	[11]
MoO <sub>3</sub> @CuO	0.86 mA h cm <sup>-2</sup>	1 mA cm <sup>-2</sup>	[12]
MoO <sub>3</sub> /NiMoO <sub>4</sub>	0.8 mA h cm <sup>-2</sup>	2.2 mA cm <sup>-2</sup>	[13]
MoO <sub>3</sub> -CNF	0.15 mA h cm <sup>-2</sup>	1 mA cm <sup>-2</sup>	[14]
MoO <sub>3</sub> /ZnMoO <sub>4</sub>	110 F/g	$0.5 \mathrm{A} \cdot \mathrm{g}^{-1}$	[15]
h-Co <sub>9</sub> S <sub>8</sub> /NiCo-Mo NSAs	3.07 mA h·cm <sup>-2</sup>	1 mA cm <sup>-2</sup>	This work

**Tabal S1** Comparison of performance between electrode materials prepared in this work and other Ni/Co/Mo-based materials.

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