

Supporting Information

***N*-Nosyl-*O*-Bromoethyl Hydroxylamine Acts as Multifunctional Formaldehyde, Formaldimine, and 1,2-Oxazetidine Surrogate for the C-C and C-O Bond-Forming Reactions**

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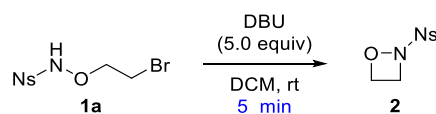
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1. General information

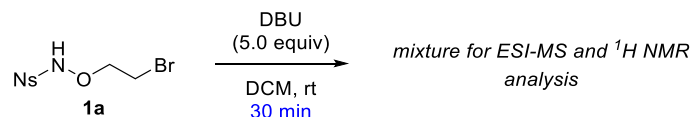
Unless otherwise stated, all reagents were purchased from commercial suppliers and used without further purification. All solvents employed in the reactions were distilled from appropriate drying agents prior to use. Analytical thin layer chromatography (TLC) was performed on precoated silica gel 60 GF254 plates. Visualization on TLC was achieved by use of UV light (254 nm). Flash column chromatography was performed using Tsingdao silica gel. ^1H and ^{13}C NMR spectra were recorded on Agilent 400MR DD2 (400 MHz) spectrometer or Agilent 600MR DD2 (600 MHz) spectrometer. Chemical shifts were reported in parts per million (ppm), and tetramethylsilane (TMS) or the residual solvent peak was used as an internal reference: ^1H NMR (TMS, δ 0.00; CDCl_3 , δ 7.26; acetone- d_6 δ 2.05), ^{13}C NMR (CDCl_3 , δ 77.0; acetone- d_6 δ 29.84, 206.26). Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad), coupling constants (Hz) and integration. High resolution mass spectra (HRMS) were obtained on a Bruker Daltonics Solarix 7.0 Tesla Fourier Transform Ion Cyclotron Resonance (FT-ICR analyzer) Mass Spectrometer using the electrospray ionization (ESI) technique. Reactions carried out at elevated temperature were heated using an oil bath.

2. Detection of the intermediates by ESI-HRMS and ^1H NMR analysis



To a solution of **1a** (33 mg, 0.1 mmol, 1.0 equiv) in DCM (1 mL) was added dropwise of DBU (76 mg, 0.5 mmol, 5.0 equiv) at room temperature. The reaction mixture was stirred at the same temperature for 5 min, then the solution was quenched with saturated NH_4Cl solution (3 mL). The organic layer was separated and the aqueous layer was extracted with DCM (3 mL x 3). The combined organic layer was washed with brine (3 mL), dried over Na_2SO_4 , and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 3/1) to afford the product 1,2-oxazetidinium **2** (15 mg, 60% yield) as a white solid. ^1H NMR

(600 MHz, CDCl₃) δ 8.44 (d, *J* = 8.5 Hz, 2H), 8.22 (d, *J* = 8.5 Hz, 2H), 4.78 (t, *J* = 8.2 Hz, 2H), 4.54 (t, *J* = 8.2 Hz, 2H). The spectral data of **2** was consistent with that reported in the literature.¹



To a solution of **1a** (33 mg, 0.1 mmol, 1.0 equiv) in DCM (1 mL) was added dropwise of DBU (76 mg, 0.5 mmol, 5.0 equiv) at room temperature. After stirring at room temperature for 30 minutes, an aliquot of reaction solution was taken out and diluted with methanol. Then, the mixture was immediately subjected to mass spectra analysis. The ESI-HRMS data was acquired and recorded on a 7 Tesla Solarix FT-ICR mass spectrometer (Bruker Daltonics, Bremen, Germany). By analyzing the data, we assigned the plausible structures of four species **S1-S4** as following (**Figure S1**). They all should derive from a Ns-NHCH₂OCH₃ precursor, indicating that the formaldimine [H₂C=N-Ns] intermediate might be formed in the reaction mixture.

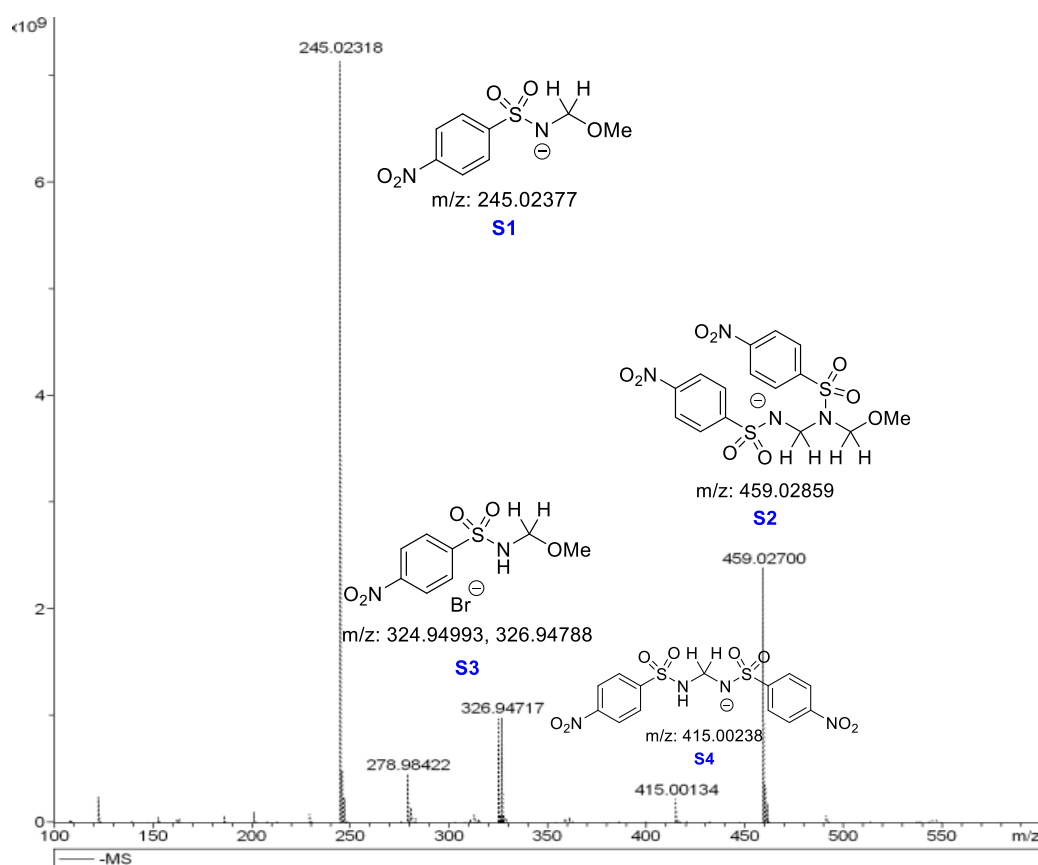


Figure S1. ESI-HRMS data of the reaction mixture (in CH₃OH).

To further confirm the structures of the above **S1-S4**, the reaction mixture was subjected to the mass spectra analysis in CD₃OD solvent (**Figure S2**). Again, the ESI-HRMS data were consistent with the above assigned structures.

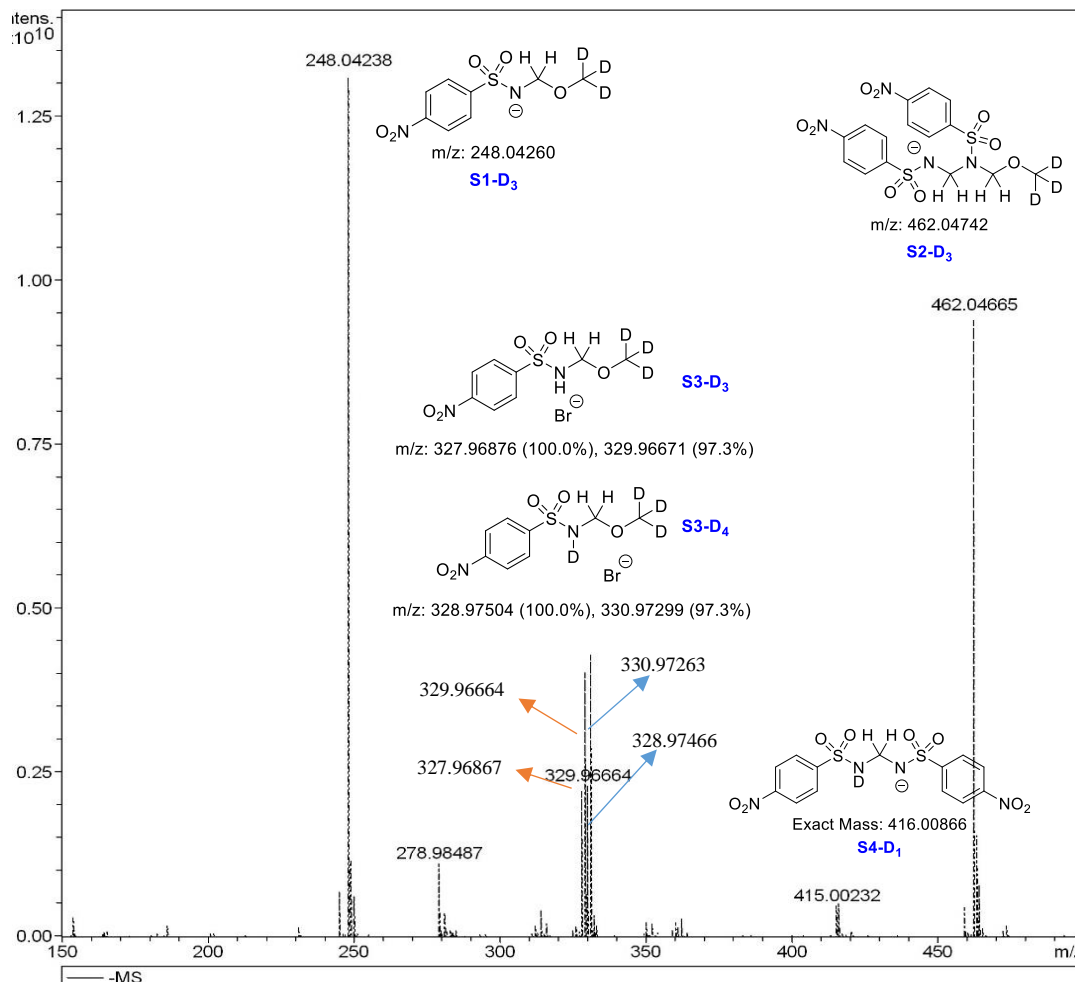


Figure S2. ESI-HRMS data of the reaction mixture (in CD₃OD).

Meanwhile, to gain details of the reaction process, hydroxylamine **1a** (1.0 equiv) and DBU (5.0 equiv) were dissolved in CDCl₃. Then the ¹H NMR spectra of the mixture were recorded at 1 min, 5 min, and 30 min, respectively. As shown in **Figure S3**, a large amount of 1,2-oxazetidine **2** was rapidly generated just after 1 min of the reaction, and the starting material **1a** was almost fully converted to compound **2** after 5 min the reaction. However, when the reaction was extended to 30 min, compound **1a** was finally degraded to nosylamide.

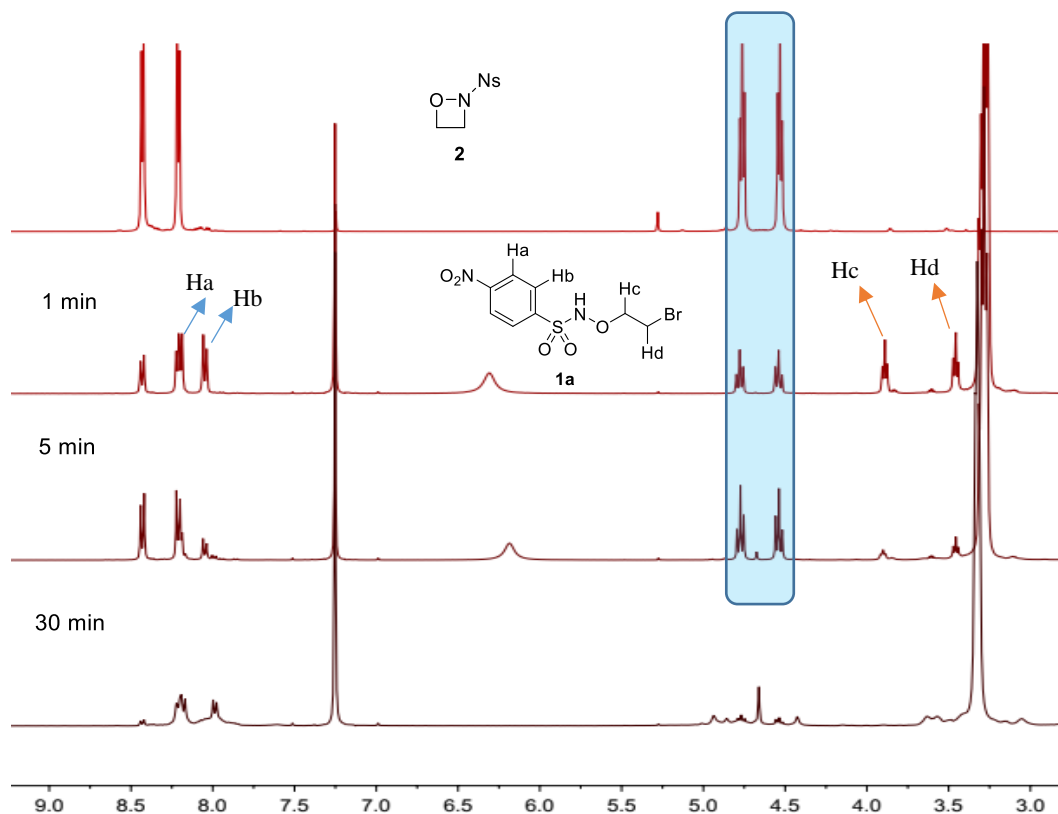


Figure S3. ¹H NMR spectra of the reaction mixture (in CDCl₃).

The final product nosylamide was isolated and identified by ¹H NMR spectra. The whole fragmentation process could also be clearly monitored by TLC method (Figure S4).

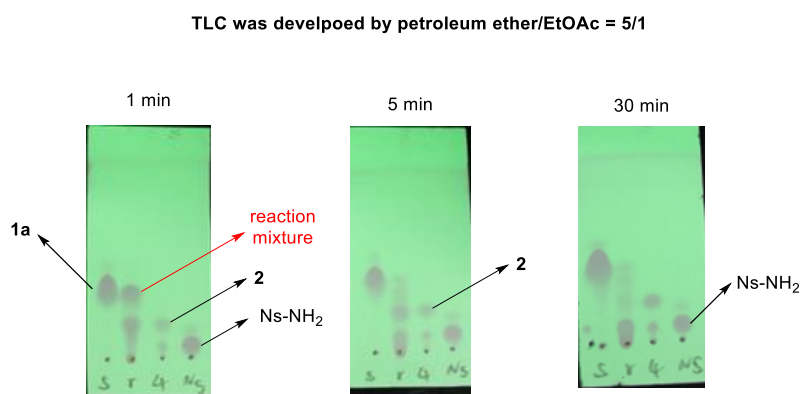
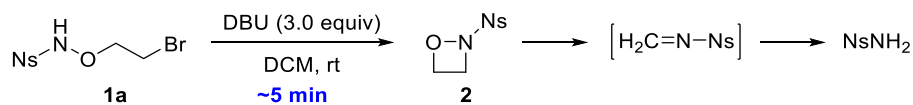


Figure S4. Detection of the intermediates by TLC analysis.

According to the above data and observation, we deduced that *N*-nosyl-*O*-bromoethyl hydroxylamine **1a** rapidly generated the four-membered 1,2-oxazetidine **2** within 5 minutes of the reaction. Then compound **2** probably could in situ generate the formaldimine [H₂C=N-Ns] intermediate in the presence of DBU

base. This intermediate is also highly unstable, which could be finally decomposed to nosylamide.



3. Optimization of the reaction conditions

Table S1. Screening of the bases and solvents^a

entry	Base	solvent	t (h)	yield (%) ^b
1	K ₂ CO ₃ (50%)	DMF	24	47
2	Cs ₂ CO ₃ (50%)	DMF	4	73
3	CsOH (50%)	DMF	4	54
4	K ₂ CO ₃ (s)	DMF	24	35
5	Cs ₂ CO ₃ (s)	DMF	4	71
6	CsOH (s)	DMF	4	60
7	DBU	DMF	1	78
8	TMG	DMF	1	75
9	DBU	DCM	1	85
10	Pyridine	DCM	24	0
11	TEA	DCM	24	0 ^c
12	DBU	PhMe	4	51
13	DBU	Ethyl acetate	4	72
14	DBU	THF	4	63
15	DBU	CH ₃ CN	4	69

^aReaction conditions: **1a** (0.10 mmol), **3a** (0.20 mmol) and base (0.30 mmol) in indicated solvent. ^bIsolated yield. ^c1,2-Oxazetidine was formed.

Table S2. Screening of other reaction parameters^a

entry	3a:1a	T (°C)	t (h)	yield (%) ^b
1	2.5 : 1	rt	1	84
2	2 : 1	rt	1	85
3	1.5 : 1	rt	1	81
4	1 : 1	rt	1	60
5	1 : 1.5	rt	1	34
6	2 : 1	0	24	68
7	2 : 1	40	1	59

^aReaction conditions: **1a**, **3a** and DBU (3.0 equiv) in DCM. ^bIsolated yield.

Table S3. Screening of hydroxylamines^a

entry	substrates	base	solvent	t (h)	yield (%) ^b
1	1a	DBU	DCM	1	85
2	1b	DBU	DCM	24	0 ^c
3	1b	CsOH	DMF	24	0
4	1b	^t BuOK	THF	24	0 ^d
5	1c	DBU	DCM	24	0
6	1c	CsOH	DMF	24	0
7	1c	^t BuOK	THF	24	0 ^c
8	1e	DBU	DCM	1	72
9	1e	CsOH	DMF	1	46
10	1e	^t BuOK	THF	24	0 ^d

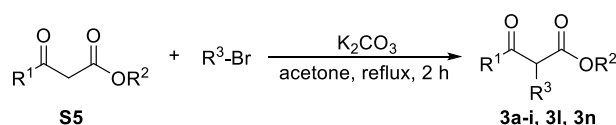
^aReaction conditions: **1** (0.10 mmol), **3a** (0.2 mmol) and base (0.30 mmol) in indicated solvent. ^bIsolated yield. ^c1,2-Oxazetidine was formed. ^dDecomposition of the starting materials.

4. General procedure for the preparation of nucleophile substrates

4.1. General procedure for the preparation of β -keto esters **3**

Compounds **3j**, **3m**, **3p** are commercial available. Other known carbonyl substrates were prepared as following procedures.

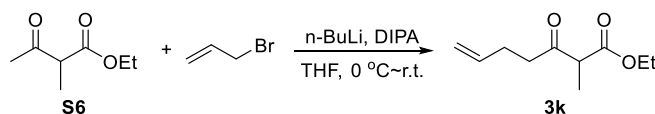
4.1.1 Method A: preparation of β -keto esters **3a-i**, **3l**, and **3n**²



A mixture of **S5** (3 ~ 20 mmol, 1.0 equiv) and anhydrous K_2CO_3 (1.5 equiv) in acetone was stirred at room temperature for five minutes. Then, methyl iodide or corresponding bromide (2.0 equiv) was added carefully. The reaction mixture was heated to refluxed in an oil bath until the completion of the reaction as monitored by TLC. After the mixture was cooled to room temperature, the precipitate was filtered off and the solvent was removed under vacuum. The crude mixture was purified by column chromatography (petroleum ether/EtOAc = 200/1 ~ 20/1) to afford the

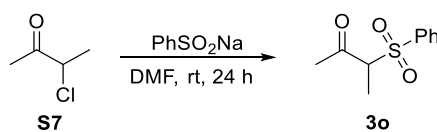
products **3a-i**, **3l** and **3n** in 48-92% yields.

4.1.2 Method B: preparation of β -keto ester **3k**³



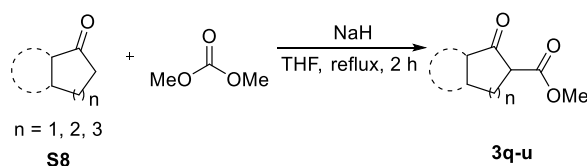
Diisopropylamine (22 mmol, 2.2 equiv) was dissolved in dry tetrahydrofuran (40 mL) and the solution cooled to $0\text{ }^\circ\text{C}$. n -Butyllithium (22 mmol, 2.2 equiv., 2.5 M solution in n -hexane) was slowly added and the reaction mixture was stirred at room temperature for 30 minutes and afterwards cooled to $0\text{ }^\circ\text{C}$. β -ketoester **S6** (10 mmol, 1.0 equiv) was dissolved in dry tetrahydrofuran (5 mL) and slowly added to the solution. The reaction mixture was stirred for 15 minutes after which allyl bromide (12 mmol, 1.2 equiv) was added. The solution was allowed to stir at $0\text{ }^\circ\text{C}$ for 30 minutes and at room temperature for 90 minutes. The reaction mixture was quenched with saturated ammonium chloride (20 mL) and extracted with EtOAc (30 mL x 3). The combined organic layers were washed with brine (20 mL), dried over anhydrous Na_2SO_4 and the solvent was evaporated in vacuo. The crude mixture was purified by column chromatography (petroleum ether/EtOAc = 50/1) to afford the product **3k** as a colorless oil (550 mg, 30% yield).

4.1.3 Method C: preparation of β -keto ester **3o**⁴

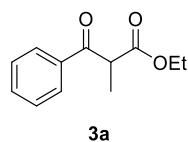


To a solution of α -haloketone **S7** (10 mmol, 1.0 equiv) in DMF (20 mL) was added sodium benzenesulfinate (10 mmol, 1.0 equiv) in one portion and the reaction mixture was stirred at room temperature for 24 h. The reaction was stopped by the addition of water (30 mL) and extracted with EtOAc (30 mL x 3). The combined organic layers were washed with brine (20 mL), dried with NaSO_4 and the solvent was removed under reduced pressure. The crude mixture was purified by column chromatography (petroleum ether/EtOAc = 10/1) to afford the product **3o** as a white solid (1.71 g, 81% yield).

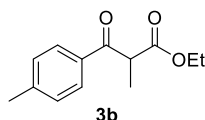
4.1.4 Method D: preparation of β -keto esters **3q-u**⁵



To a suspension of NaH (2.1 equiv) in THF (50 mL) was added dimethyl carbonate (5.0 equiv) at room temperature. The mixture was stirred for 5 minutes and then ketone **S8** (3 ~ 20 mmol, 1.0 equiv) in THF (30 mL) was added dropwise. The reaction mixture was heated to refluxed in an oil bath until the completion of the reaction as monitored by TLC. The resulting mixture was then cooled to 0 °C and 1 M HCl was added until the solution became neutral. The mixture was then poured into water (30 mL) and extracted with EtOAc (30 mL x 3). The combined organic layers were washed with brine (20 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuum. The crude mixture was purified by column chromatography (petroleum ether/EtOAc = 200/1 ~ 100/1) to afford the products **3q-u** in 70-88% yields.

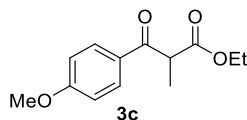


Following the general method A, **3a** was obtained by flash column chromatography (petroleum ether/EtOAc = 200/1) as a colorless oil (1.9 g, 92% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.93 (d, $J = 7.5$ Hz, 2H), 7.52 (t, $J = 7.3$ Hz, 1H), 7.42 (t, $J = 7.6$ Hz, 2H), 4.35 (q, $J = 7.1$ Hz, 1H), 4.09 (q, $J = 7.1$ Hz, 2H), 1.44 (d, $J = 7.2$ Hz, 3H), 1.11 (t, $J = 7.2$ Hz, 3H). The spectral data of **3a** was consistent with that reported in the literature.⁶

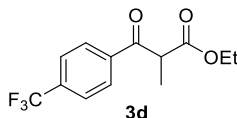


Following the general method A, **3b** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (363 mg, 55% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.88 (d, $J = 8.1$ Hz, 2H), 7.27 (d, $J = 8.1$ Hz,

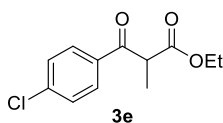
2H), 4.35 (q, $J = 7.1$ Hz, 1H), 4.14 (q, $J = 7.1$ Hz, 2H), 2.41 (s, 3H), 1.47 (d, $J = 7.1$ Hz, 3H), 1.17 (t, $J = 7.1$ Hz, 3H). The spectral data of **3b** was consistent with that reported in the literature.⁶



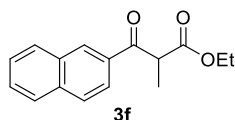
Following the general method A, **3c** was obtained by flash column chromatography (petroleum ether/EtOAc = 50/1) as a colorless oil (1.07 g, 91% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.97 (d, $J = 8.9$ Hz, 2H), 6.94 (d, $J = 8.9$ Hz, 2H), 4.33 (q, $J = 7.1$ Hz, 1H), 4.14 (q, $J = 7.2$ Hz, 2H), 3.87 (s, 3H), 1.47 (d, $J = 7.1$ Hz, 3H), 1.18 (t, $J = 7.2$ Hz, 3H). The spectral data of **3c** was consistent with that reported in the literature.⁷



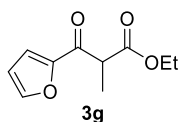
Following the general method A, **3d** was obtained by flash column chromatography (petroleum ether/EtOAc = 30/1) as a colorless oil (1.02 g, 74% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.07 (d, $J = 8.2$ Hz, 2H), 7.74 (d, $J = 8.2$ Hz, 2H), 4.36 (q, $J = 7.1$ Hz, 1H), 4.14 (q, $J = 7.1$ Hz, 2H), 1.50 (d, $J = 7.1$ Hz, 3H), 1.16 (t, $J = 7.1$ Hz, 3H). The spectral data of **3d** was consistent with that reported in the literature.⁸



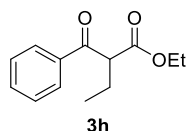
Following the general method A, **3e** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (323 mg, 67% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.92 (d, $J = 8.5$ Hz, 2H), 7.45 (d, $J = 8.5$ Hz, 2H), 4.31 (q, $J = 7.0$ Hz, 1H), 4.14 (q, $J = 7.1$ Hz, 2H), 1.48 (d, $J = 7.1$ Hz, 3H), 1.17 (t, $J = 7.1$ Hz, 3H). The spectral data of **3e** was consistent with that reported in the literature.⁹



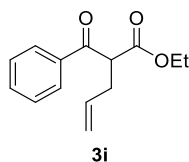
Following the general method A, **3f** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (910 mg, 71% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.51 (s, 1H), 8.04 (dd, $J = 8.7, 1.8$ Hz, 1H), 7.98 (d, $J = 8.1$ Hz, 1H), 7.94 – 7.85 (m, 2H), 7.65 – 7.54 (m, 2H), 4.54 (q, $J = 7.1$ Hz, 1H), 4.21 – 4.11 (m, 2H), 1.56 (d, $J = 7.0$ Hz, 3H), 1.17 (t, $J = 7.1$ Hz, 3H). The spectral data of **3f** was consistent with that reported in the literature.¹⁰



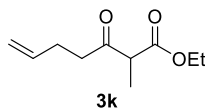
Following the general method A, **3g** was obtained by flash column chromatography (petroleum ether/EtOAc = 20/1) as a colorless oil (456 mg, 86% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.62 – 7.57 (m, 1H), 7.30 – 7.24 (m, 1H), 6.55 (dd, $J = 3.6, 1.7$ Hz, 1H), 4.22 – 4.08 (m, 3H), 1.46 (d, $J = 7.2$ Hz, 3H), 1.18 (t, $J = 7.1$ Hz, 3H). The spectral data of **3g** was consistent with that reported in the literature.⁶



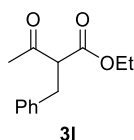
Following the general method A, **3h** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (1.9 g, 86% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.99 (d, $J = 9.3$ Hz, 2H), 7.58 (t, $J = 7.4$ Hz, 1H), 7.47 (t, $J = 7.7$ Hz, 2H), 4.25 – 4.09 (m, 3H), 2.09 – 1.98 (m, 2H), 1.17 (t, $J = 7.1$ Hz, 3H), 0.99 (t, $J = 7.4$ Hz, 3H). The spectral data of **3h** was consistent with that reported in the literature.⁹



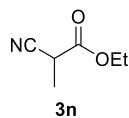
Following the general method A, **3i** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (3.3 g, 72% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.99 (d, $J = 7.5$ Hz, 2H), 7.58 (t, $J = 7.3$ Hz, 1H), 7.47 (t, $J = 7.6$ Hz, 2H), 5.88 – 5.73 (m, 1H), 5.16 – 4.96 (m, 2H), 4.39 (t, $J = 7.2$ Hz, 1H), 4.17 – 4.09 (m, 2H), 2.83 – 2.70 (m, 2H), 1.16 (t, $J = 7.1$ Hz, 3H). The spectral data of **3i** was consistent with that reported in the literature.¹⁰



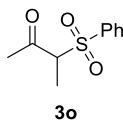
Following the general method B, **3k** was obtained by flash column chromatography (petroleum ether/EtOAc = 50/1) as a colorless oil (550 mg, 30% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 5.91 – 5.69 (m, 1H), 5.08 – 4.94 (m, 2H), 4.19 (q, $J = 7.2$ Hz, 2H), 3.51 (q, $J = 7.1$ Hz, 1H), 2.76 – 2.52 (m, 2H), 2.42 – 2.25 (m, 2H), 1.33 (d, $J = 7.1$ Hz, 3H), 1.27 (t, $J = 7.1$ Hz, 3H). The spectral data of **3k** was consistent with that reported in the literature.³



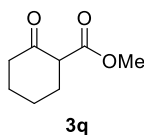
Following the general method A, **3l** was obtained by flash column chromatography (petroleum ether/EtOAc = 50/1) as a colorless oil (1.25 g, 57% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.31 – 7.14 (m, 5H), 4.14 (q, $J = 7.7$ Hz, 2H), 3.78 (t, $J = 7.6$ Hz, 1H), 3.16 (d, $J = 7.7$ Hz, 2H), 2.18 (s, 3H), 1.19 (t, $J = 7.7$ Hz, 3H). The spectral data of **3l** was consistent with that reported in the literature.¹¹



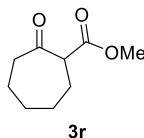
Following the general method A, **3n** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (621 mg, 48% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 4.27 (q, $J = 7.1$ Hz, 2H), 3.54 (q, $J = 7.4$ Hz, 1H), 1.59 (d, $J = 7.4$ Hz, 3H), 1.32 (t, $J = 7.1$ Hz, 3H). The spectral data of **3n** was consistent with that reported in the literature.¹²



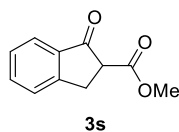
Following the general method C, **3o** was obtained by flash column chromatography (petroleum ether/EtOAc = 10/1) as a white solid (1.71 g, 81% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.78 (d, $J = 8.2$ Hz, 2H), 7.66 (t, $J = 7.4$ Hz, 1H), 7.54 (t, $J = 7.6$ Hz, 2H), 4.16 (q, $J = 7.0$ Hz, 1H), 2.41 (s, 3H), 1.37 (d, $J = 6.9$ Hz, 3H). The spectral data of **3o** was consistent with that reported in the literature.⁴



Following the general method D, **3q** was obtained by flash column chromatography (petroleum ether/EtOAc = 200/1) as a colorless oil (2.3 g, 74% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 12.16 (s, 1H), 3.75 (s, 3H), 2.32 – 2.17 (m, 3H), 1.76 – 1.54 (m, 5H). The spectral data of **3q** was consistent with that reported in the literature.¹³

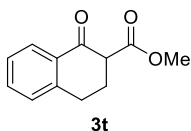


Following the general method D, **3r** was obtained by flash column chromatography (petroleum ether/EtOAc = 200/1) as a colorless oil (357 mg, 70% yield, keto/enol form = 2.8/1). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 12.65 (s, 0.26H), 3.75 – 3.68 (m, 3H), 3.54 (dd, $J = 10.4, 4.0$ Hz, 0.7H), 2.68 – 2.50 (m, 1.2H), 2.46 – 2.31 (m, 1.21H), 2.16 – 2.03 (m, 0.73H), 2.01 – 1.76 (m, 3H), 1.77 – 1.65 (m, 0.75H), 1.65 – 1.51 (m, 1.27H), 1.52 – 1.35 (m, 2H). The spectral data of **3r** was consistent with that reported in the literature.¹³

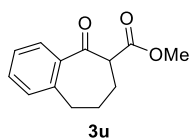


Following the general method D, **3s** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a brown solid (1.56 g, 82%

yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.76 (d, $J = 7.6$ Hz, 1H), 7.61 (t, $J = 7.5$ Hz, 1H), 7.49 (d, $J = 7.7$ Hz, 1H), 7.38 (t, $J = 7.3$ Hz, 1H), 3.80 – 3.69 (m, 4H), 3.60 – 3.47 (m, 1H), 3.37 (dd, $J = 17.3, 8.3$ Hz, 1H). The spectral data of **3s** was consistent with that reported in the literature.¹⁴



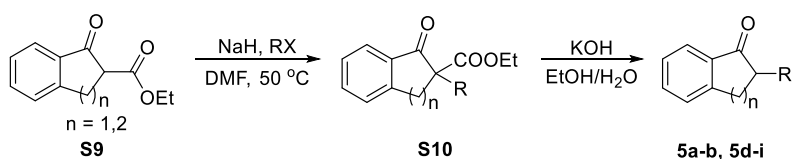
Following the general method D, **3t** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a pale yellow oil (3.6 g, 88% yield, keto/enol form = 1/1). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 12.41 (s, 0.5H), 8.04 (d, $J = 7.7$ Hz, 0.58H), 7.85 – 7.75 (m, 0.49H), 7.49 (t, $J = 7.5$ Hz, 0.61H), 7.37 – 7.21 (m, 2H), 7.17 (d, $J = 7.2$ Hz, 0.42H), 3.82 (s, 1.4H), 3.78 (s, 1.6H), 3.63 (dd, $J = 10.6, 4.6$ Hz, 0.49H), 3.13 – 2.93 (m, 1.26H), 2.81 (t, $J = 7.6$ Hz, 1H), 2.62 – 2.42 (m, 1.43H), 2.43 – 2.30 (m, 0.52H). The spectral data of **3t** was consistent with that reported in the literature.¹⁴



Following the general method D, **3u** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a pale yellow oil (471 mg, 72% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 12.61 (s, 1H), 7.62 (dd, $J = 6.5, 2.5$ Hz, 1H), 7.39 – 7.28 (m, 2H), 7.25 – 7.17 (m, 1H), 3.82 (s, 1.44H), 3.78 (s, 1.56H), 2.64 (t, $J = 6.6$ Hz, 2H), 2.19 – 2.02 (m, 4H). The spectral data of **3u** was consistent with that reported in the literature.¹⁴

4.2 General procedure for the preparation of α -substituted ketones **5**

4.2.1 Method E: preparation of α -substituted ketones **5a-b** and **5d-i**¹⁵

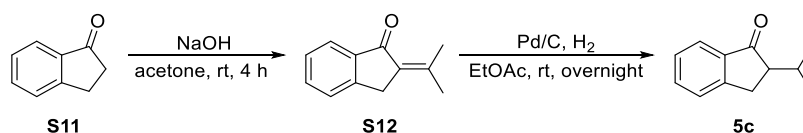


To a solution of **S9** (5 ~ 23 mmol, 1.0 equiv) in dry DMF (12 mL) was added

NaH (60% in oil, 10 mmol, 2.0 equiv), the solution was stirred at room temperature for 30 minutes and methyl iodide (15 mmol, 3.0 equiv) was added. The reaction was stirred at 50 °C in an oil bath for 3h. After completion, the reaction mixture was cooled, quenched with a saturated NH₄Cl aqueous solution (15 mL) and the product was extracted with EtOAc (20 mL × 3). The combined organic layers were washed with brine (20 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuum. The residue was used in the next step without further purification.

The crude **S10** was dissolved in EtOH/H₂O (5/1, 22 mL). After adding KOH (12.5 mmol, 2.5 equiv), the reaction mixture was heated to reflux in an oil bath for 2 hours. The residue was neutralized with a saturated NH₄Cl aqueous solution (10 mL), the volatiles were removed under vacuum, and the resulting aqueous layer was extracted with EtOAc (20 mL × 3). The combined organic layers were washed with brine (20 mL), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash chromatography (petroleum ether/EtOAc = 300/1 ~ 100/1) to afford **5a-b**, **5d-i** in 30-81% overall yields.

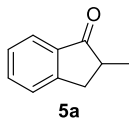
4.2.2 Method F: preparation of α -substituted ketone **5c**¹⁶



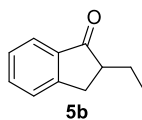
To a solution of **S11** (1.32 g, 10.0 mmol) in anhydrous acetone (20 mL), NaOH (132 mg, 3.3 mmol) was added. The mixture was stirred at room temperature for 4 h and then neutralized with 1 N HCl. Acetone was removed by distillation, after adding water (20 mL) and EtOAc (30 mL), the mixture was partitioned. The organic layer was washed with brine (20 mL), dried over anhydrous Na₂SO₄, and concentrated in vacuum. The residue was purified by flash chromatography (petroleum ether/EtOAc = 80/1) to afford **S12**.

To a solution of **S12** (1.44 g, 10 mmol) in EtOAc (20 mL), 10% Pd/C (20% weight of compound **S12**) was added. The mixture was stirred overnight under a hydrogen atmosphere at room temperature. Insoluble materials were removed by

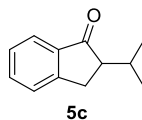
filtration and washed with EtOAc. The filtrate was evaporated to dryness under reduced pressure. The residue was purified by flash chromatography (petroleum ether/EtOAc = 30/1) to afford **5c** in 35% overall yield.



Following the general method E, **5a** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (314 mg, 43% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.73 (d, *J* = 7.6 Hz, 1H), 7.56 (t, *J* = 7.4 Hz, 1H), 7.43 (d, *J* = 7.7 Hz, 1H), 7.34 (t, *J* = 7.4 Hz, 1H), 3.43 – 3.32 (m, 1H), 2.76 – 2.63 (m, 2H), 1.29 (dd, *J* = 7.2, 1.2 Hz, 3H). The spectral data of **5a** was consistent with that reported in the literature.¹⁷

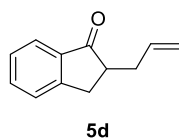


Following the general method E, **5b** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (320 mg, 40% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, *J* = 7.6 Hz, 1H), 7.58 (t, *J* = 7.4 Hz, 1H), 7.46 (d, *J* = 7.6 Hz, 1H), 7.36 (t, *J* = 7.4 Hz, 1H), 3.32 (dd, *J* = 17.2, 7.9 Hz, 1H), 2.82 (dd, *J* = 17.2, 3.9 Hz, 1H), 2.67 – 2.57 (m, 1H), 2.05 – 1.90 (m, 1H), 1.60 – 1.46 (m, 1H), 1.01 (t, *J* = 7.4 Hz, 3H). The spectral data of **5b** was consistent with that reported in the literature.¹⁷

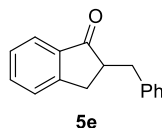


Following the general method F, **5c** was obtained by flash column chromatography (petroleum ether/EtOAc = 30/1) as a colorless oil (610 mg, 35% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.70 (d, *J* = 7.6 Hz, 1H), 7.53 (t, *J* = 7.0 Hz, 1H), 7.43 (d, *J* = 7.7 Hz, 1H), 7.30 (t, *J* = 7.4 Hz, 1H), 3.10 (dd, *J* = 17.4, 8.1 Hz, 1H), 2.89 (dd, *J* = 17.5, 4.0 Hz, 1H), 2.62 (dt, *J* = 8.3, 4.2 Hz, 1H), 2.46 – 2.30 (m, 1H), 1.01 (d, *J* = 6.9 Hz, 3H), 0.75 (d, *J* = 6.8 Hz, 3H). The spectral data of **5c** was

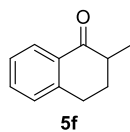
consistent with that reported in the literature.¹⁶



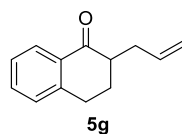
Following the general method E, **5d** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (258 mg, 30% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, *J* = 7.6 Hz, 1H), 7.58 (t, *J* = 8.0 Hz, 1H), 7.45 (d, *J* = 7.6 Hz, 1H), 7.36 (t, *J* = 7.8 Hz, 1H), 5.87 – 5.72 (m, 1H), 5.16 – 5.00 (m, 2H), 3.28 (dd, *J* = 17.3, 7.7 Hz, 1H), 2.86 (dd, *J* = 17.3, 3.8 Hz, 1H), 2.81 – 2.64 (m, 2H), 2.32 – 2.19 (m, 1H). The spectral data of **5d** was consistent with that reported in the literature.¹⁸



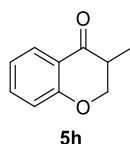
Following the general method E, **5e** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (378 mg, 34% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.79 (d, *J* = 7.5 Hz, 1H), 7.57 (t, *J* = 7.1 Hz, 1H), 7.45 – 7.34 (m, 2H), 7.34 – 7.19 (m, 5H), 3.44 – 3.37 (m, 1H), 3.17 (dd, *J* = 17.3, 7.5 Hz, 1H), 3.07 – 2.95 (m, 1H), 2.92 – 2.82 (m, 1H), 2.75 – 2.59 (m, 1H). The spectral data of **5e** was consistent with that reported in the literature.¹⁵



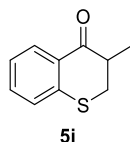
Following the general method E, **5f** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (288 mg, 36% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.01 (d, *J* = 7.8 Hz, 1H), 7.42 (t, *J* = 7.4 Hz, 1H), 7.26 (t, *J* = 7.5 Hz, 1H), 7.20 (d, *J* = 7.6 Hz, 1H), 3.07 – 2.87 (m, 2H), 2.62 – 2.48 (m, 1H), 2.22 – 2.07 (m, 1H), 1.91 – 1.76 (m, 1H), 1.25 (d, *J* = 6.8, 3H). The spectral data of **5f** was consistent with that reported in the literature.⁹



Following the general method E, **5g** was obtained by flash column chromatography (petroleum ether/EtOAc = 300/1) as a colorless oil (3.45 g, 81% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.03 (d, $J = 7.8$ Hz, 1H), 7.44 (t, $J = 7.5$ Hz, 1H), 7.28 (t, $J = 8.2$ Hz, 1H), 7.22 (d, $J = 7.6$ Hz, 1H), 5.91 – 5.76 (m, 1H), 5.15 – 5.03 (m, 2H), 2.98 (dd, $J = 7.8, 4.5$ Hz, 2H), 2.81 – 2.70 (m, 1H), 2.59 – 2.47 (m, 1H), 2.32 – 2.16 (m, 2H), 1.93 – 1.77 (m, 1H). The spectral data of **5g** was consistent with that reported in the literature.¹⁹

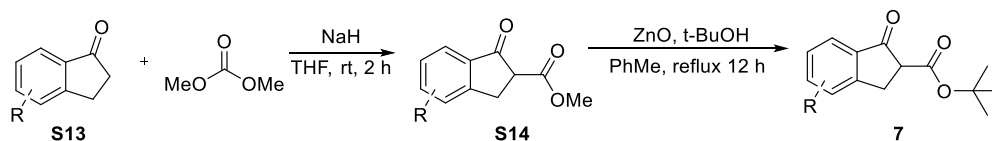


Following the general method E, **5h** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (393 mg, 40% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.90 (d, $J = 7.9$, 1H), 7.51 – 7.42 (m, 1H), 7.01 (t, $J = 7.4$ Hz, 1H), 6.96 (d, $J = 8.4$ Hz, 1H), 4.50 (dd, $J = 11.3, 5.1$ Hz, 1H), 4.20 – 4.10 (m, 1H), 2.93 – 2.80 (m, 1H), 1.22 (d, $J = 7.0$ Hz, 3H). The spectral data of **5h** was consistent with that reported in the literature.²⁰



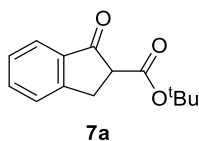
Following the general method E, **5i** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (348 mg, 39% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.09 (d, $J = 7.9$ Hz, 1H), 7.36 (t, $J = 7.5$ Hz, 1H), 7.24 (d, $J = 7.9$ Hz, 1H), 7.16 (t, $J = 7.6$ Hz, 1H), 3.19 – 3.06 (m, 2H), 2.99 – 2.86 (m, 1H), 1.34 (d, $J = 6.9$ Hz, 3H). The spectral data of **5i** was consistent with that reported in the literature.²⁰

4.3. General procedure for the preparation of indanone carboxylates **7**

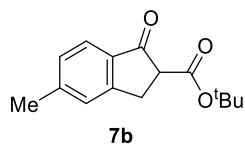


To a suspension of NaH (2.1 equiv) in THF (50 mL) were sequentially added dimethyl carbonate (5.0 equiv) and a catalytic amount of potassium *tert*-butoxide (0.1 equiv) at room temperature. The mixture was stirred for 5 min and then ketone **S13** (1.0 equiv) in THF (30 mL) was added dropwise. The reaction was stirred at room temperature until the completion of the reaction as monitored by ^1H NMR. The resulting thick mixture was then cooled to 0 $^{\circ}\text{C}$ and 1 M HCl was added until the solution became neutral. The mixture was then poured into water (30 mL), extracted with EtOAc (30 mL x 3), washed with brine (20 mL), dried over anhydrous Na_2SO_4 and concentrated in vacuum to afford crude **S14**.

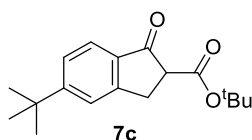
To a flask equipped with a Dean-Stark trap and reflux condenser was added above crude **S14** (1.0 equiv), *tert*-butanol (10 equiv), ZnO (0.2 equiv) and toluene (60 mL). The mixture was heated to reflux in an oil bath until complete conversion. After the mixture was cooled to room temperature, the precipitate was filtered off and the solvent was removed under vacuum. Then the residue was purified by column chromatography (petroleum ether/EtOAc = 100/1 ~ 10/1) to afford the products **7a-l** in 32-67% yields.



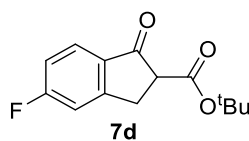
Following the general procedure starting from 1-indanone (10 mmol scale), **7a** was obtained as a pale purple oil (1.4 g, 62% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 100/1). ^1H NMR (400 MHz, CDCl_3) δ 7.76 (d, $J = 7.7$ Hz, 1H), 7.61 (t, $J = 7.4$ Hz, 1H), 7.49 (d, $J = 7.7$ Hz, 1H), 7.38 (t, $J = 7.5$ Hz, 1H), 3.62 (dd, $J = 8.3, 3.9$ Hz, 1H), 3.54 – 3.45 (m, 1H), 3.33 (dd, $J = 17.1, 8.2$ Hz, 1H), 1.49 (s, 9H). The spectral data of **7a** was consistent with that reported in the literature.¹



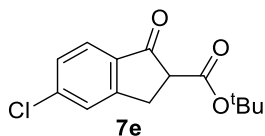
Following the general procedure starting from 5-methyl-1-indanone (6 mmol scale), **7b** was obtained as a white solid (990 mg, 67% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 100/1). ¹H NMR (400 MHz, CDCl₃) δ 7.65 (d, *J* = 7.9 Hz, 1H), 7.28 – 7.22 (m, 1H), 7.19 (d, *J* = 7.8 Hz, 1H), 3.60 (dd, *J* = 8.2, 3.9 Hz, 1H), 3.43 (dd, *J* = 17.1, 3.7 Hz, 1H), 3.27 (dd, *J* = 17.2, 8.2 Hz, 1H), 2.43 (s, 3H), 1.48 (s, 9H). The spectral data of **7b** was consistent with that reported in the literature.¹



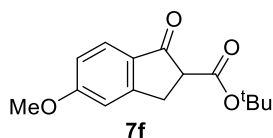
Following the general procedure starting from 5-tert-butyl-indan-1-one (15 mmol scale), **7c** was obtained as a pale purple solid (2.4 g, 56% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 100/1). ¹H NMR (400 MHz, CDCl₃) δ 7.67 (d, *J* = 8.1 Hz, 1H), 7.48 (s, 1H), 7.42 (d, *J* = 8.1 Hz, 1H), 3.59 (dd, *J* = 8.1, 3.9 Hz, 1H), 3.46 (dd, *J* = 17.0, 3.3 Hz, 1H), 3.29 (dd, *J* = 17.1, 8.2 Hz, 1H), 1.48 (s, 9H), 1.34 (s, 9H). The spectral data of **7c** was consistent with that reported in the literature.¹



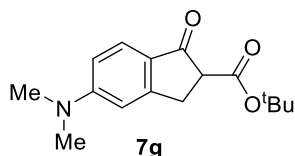
Following the general procedure starting from 5-fluoro-1-indanone (10 mmol scale), **7d** was obtained as a pale yellow solid (1.23 g, 49% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 40/1). ¹H NMR (400 MHz, CDCl₃) δ 7.75 (dd, *J* = 8.1, 5.5 Hz, 1H), 7.14 (d, *J* = 8.1 Hz, 1H), 7.08 (t, *J* = 8.3 Hz, 1H), 3.63 (dd, *J* = 8.1, 3.8 Hz, 1H), 3.48 (dd, *J* = 17.7, 3.6 Hz, 1H), 3.30 (dd, *J* = 17.4, 8.2 Hz, 1H), 1.48 (s, 9H). The spectral data of **7d** was consistent with that reported in the literature.¹



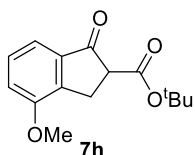
Following the general procedure starting from 5-chloro-1-indanone (10 mmol scale), **7e** was obtained as a pale purple solid (1.65 g, 62% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 40/1). ¹H NMR (400 MHz, CDCl₃) δ 7.67 (d, *J* = 8.1 Hz, 1H), 7.53 – 7.44 (m, 1H), 7.35 (d, *J* = 7.6 Hz, 1H), 3.62 (dd, *J* = 8.1, 3.7 Hz, 1H), 3.52 – 3.40 (m, 1H), 3.30 (dd, *J* = 17.4, 8.2 Hz, 1H), 1.48 (s, 9H). The spectral data of **7e** was consistent with that reported in the literature.¹



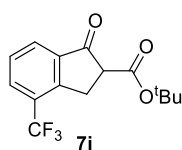
Following the general procedure starting from 5-methoxy-1-indanone (10 mmol scale), **7f** was obtained as a white solid (1.34 g, 51% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 50/1). ¹H NMR (400 MHz, CDCl₃) δ 7.64 (d, *J* = 9.0 Hz, 1H), 6.93 – 6.81 (m, 2H), 3.84 (s, 3H), 3.56 (dd, *J* = 8.0, 3.7 Hz, 1H), 3.40 (dd, *J* = 17.2, 3.1 Hz, 1H), 3.23 (dd, *J* = 17.2, 8.1 Hz, 1H), 1.45 (s, 9H). The spectral data of **7f** was consistent with that reported in the literature.¹



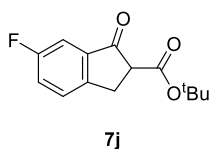
Following the general procedure starting from 5-dimethylamino-1-indanone (10 mmol scale), **7g** was obtained as a yellow solid (1.38 g, 50% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 10/1). ¹H NMR (400 MHz, CDCl₃) δ 7.61 (d, *J* = 8.8 Hz, 1H), 6.66 (dd, *J* = 8.8, 2.0 Hz, 1H), 6.56 (s, 1H), 3.55 (dd, *J* = 8.2, 3.9 Hz, 1H), 3.43 – 3.30 (m, 1H), 3.18 (dd, *J* = 17.0, 8.2 Hz, 1H), 3.08 (s, 6H), 1.48 (s, 9H). The spectral data of **7g** was consistent with that reported in the literature.¹



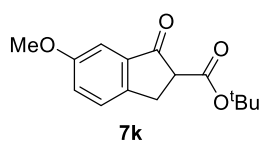
Following the general procedure starting from 4-methoxy-1-indanone (10 mmol scale), **7h** was obtained as a white solid (1.44 g, 55% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 50/1). **¹H NMR** (400 MHz, CDCl₃) δ 7.34 – 7.29 (m, 2H), 7.05 – 6.99 (m, 1H), 3.91 (s, 3H), 3.59 (dd, *J* = 7.7, 3.5 Hz, 1H), 3.47 – 3.15 (m, 2H), 1.48 (s, 9H). The spectral data of **7h** was consistent with that reported in the literature.¹



Following the general procedure starting from 4-methoxy-1-indanone (10 mmol scale), **7i** was obtained as a white solid (1.08 g, 36% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 50/1). **¹H NMR** (400 MHz, CDCl₃) δ 7.90 – 7.78 (m, 1H), 7.73 – 7.52 (m, 1H), 7.50 – 7.37 (m, 1H), 3.67 – 3.41 (m, 3H), 1.58 – 1.42 (m, 9H). The spectral data of **7i** was consistent with that reported in the literature.²¹

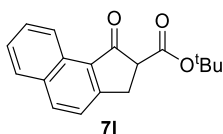


Following the general procedure starting from 6-fluoro-1-indanone (10 mmol scale), **7j** was obtained as a yellow oil (1.65 g, 66% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 100/1). **¹H NMR** (400 MHz, CDCl₃) δ 7.44 (dd, 1H), 7.40 – 7.27 (m, 2H), 3.65 (dd, *J* = 8.2, 3.8 Hz, 1H), 3.48 – 3.38 (m, 1H), 3.28 (dd, *J* = 17.1, 8.1 Hz, 1H), 1.47 (s, 9H). The spectral data of **7j** was consistent with that reported in the literature.¹



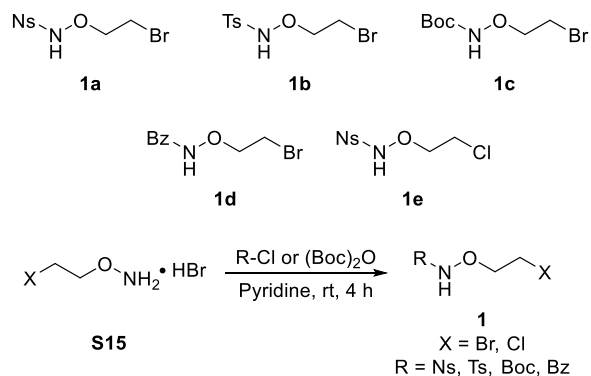
Following the general procedure starting from 6-methoxy-1-indanone (10 mmol

scale), **7k** was obtained as a white solid (1.50 g, 57% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 50/1). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.35 (d, $J = 8.3$ Hz, 1H), 7.20 – 7.13 (m, 2H), 3.80 (s, 3H), 3.61 (dd, $J = 8.0, 3.7$ Hz, 1H), 3.44 – 3.30 (m, 1H), 3.24 (dd, $J = 16.8, 8.0$ Hz, 1H), 1.47 (s, 9H). The spectral data of **7k** was consistent with that reported in the literature.¹



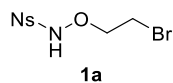
Following the general procedure starting from 2,3-dihydro-1H-cyclopenta[a]naphthalen-1-one (10 mmol scale), **7l** was obtained as a white solid (903 mg, 32% yield) following flash silica gel chromatography (petroleum ether/EtOAc = 100/1). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 9.09 (d, $J = 8.3$ Hz, 1H), 8.04 (t, $J = 9.4$ Hz, 1H), 7.88 (t, $J = 7.7$ Hz, 1H), 7.66 (t, $J = 7.6$ Hz, 1H), 7.54 (dd, $J = 18.4, 8.0$ Hz, 2H), 3.73 (dd, $J = 7.8, 3.5$ Hz, 1H), 3.58 (dd, $J = 17.5, 3.1$ Hz, 1H), 3.39 (dd, $J = 17.6, 7.8$ Hz, 1H), 1.52 (s, 9H). The spectral data of **7l** was consistent with that reported in the literature.¹

5. General procedure for the preparation of hydroxylamines **1**

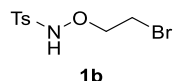


To a solution of **S15**²² (1.0 equiv) in pyridine (25 mL), R-Cl or $(\text{Boc})_2\text{O}$ (1.5 equiv) was added in three portions at 0 °C. The resulting brown suspension was warmed to rt and stirred for 4 h. After completion, the reaction mixture was poured into 1.0 M HCl solution and extracted with EtOAc (25 mL x 3). The organic solvents were combined, washed with saturated NaHCO_3 (20 mL) and H_2O (20 mL), dried

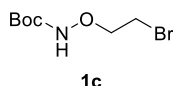
over Na₂SO₄ and concentrated. The residue was purified by column chromatography (petroleum ether/EtOAc = 6/1) to afford **1a-1e** in 37-58% yields.



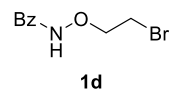
Following the general procedure described above (18 mmol scale), **1a** was obtained by flash column chromatography (petroleum ether/EtOAc = 6/1) as a yellow solid (3.4 g, 58% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.39 (d, *J* = 8.6 Hz, 2H), 8.16 (d, *J* = 8.6 Hz, 2H), 7.44 (s, 1H), 4.32 (t, *J* = 5.7 Hz, 2H), 3.56 (t, *J* = 5.7 Hz, 2H). The spectral data of **1a** was consistent with that reported in the literature.¹



Following the general procedure described above (5 mmol scale), **1b** was obtained by flash column chromatography (petroleum ether/EtOAc = 6/1) as a yellow solid (735 mg, 50% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.83 (d, *J* = 8.2 Hz, 2H), 7.36 (d, *J* = 8.0 Hz, 2H), 6.99 (s, 1H), 4.29 (t, *J* = 5.8 Hz, 2H), 3.56 (t, *J* = 5.8 Hz, 2H), 2.46 (s, 3H). The spectral data of **1b** was consistent with that reported in the literature.²³

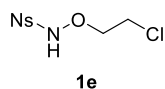


Following the general procedure described above (6.3 mmol scale), **1c** was obtained by flash column chromatography (petroleum ether/EtOAc = 6/1) as a colorless oil (562 mg, 37% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.29 (s, 1H), 4.13 (t, *J* = 6.3 Hz, 2H), 3.53 (t, *J* = 6.3 Hz, 2H), 1.47 (s, 9H). The spectral data of **1c** was consistent with that reported in the literature.²³



Following the general procedure described above (5.6 mmol scale), **1d** was obtained by flash column chromatography (petroleum ether/EtOAc = 100/1) as a colorless oil (700 mg, 51% yield). ¹H NMR (400 MHz, CDCl₃) δ 9.47 (s, 1H), 7.74 (d, *J* = 7.4 Hz, 2H), 7.52 (t, *J* = 6.9 Hz, 1H), 7.41 (t, *J* = 7.4 Hz, 2H), 4.31 (t, *J* = 5.6 Hz,

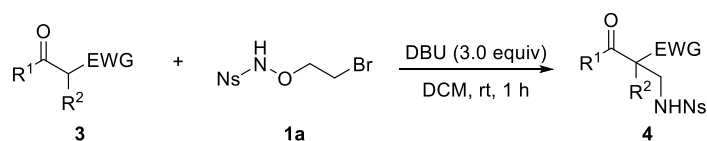
2H), 3.59 (t, $J = 5.6$ Hz, 2H). The spectral data of **1d** was consistent with that reported in the literature.²³



Following the general procedure described above (13.7 mmol scale), **1e** was obtained by flash column chromatography (petroleum ether/EtOAc = 6/1) as a yellow solid (1.62 g, 42% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.41 (d, $J = 8.7$ Hz, 2H), 8.16 (d, $J = 8.7$ Hz, 2H), 7.31 (s, 1H), 4.35 – 4.23 (m, 2H), 3.80 – 3.68 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 150.9, 141.7, 130.1, 124.3, 76.9, 40.9. HRMS (ESI) m/z : [M - H]⁻ Calcd for C₈H₈ClN₂O₅S 278.9848; Found 278.9842. **Melting point:** 100.6-102.5 °C.

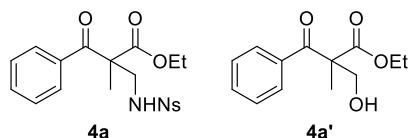
6. General procedure for the C-C bond-forming reactions

6. 1 General procedure for the preparation of aminomethyl products **4**



To a solution of β -keto esters **3** (0.4 mmol, 2.0 equiv) and DBU (91 mg, 0.6 mmol, 3.0 equiv) in DCM (1 mL) was added the solution of **1a** (65 mg, 0.2 mmol, 1.0 equiv) in DCM (1 mL) dropwise at room temperature over 10 minutes. The reaction mixture was stirred at the same temperature for 1 h. After completion, the reaction was quenched with saturated NH₄Cl (2 mL). The organic layer was separated, and the aqueous layer was extracted with DCM (5 mL x 3). The combined organic layers were washed with brine (5 mL), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash column chromatography (petroleum ether/EtOAc = 8/1 ~ 5/1) to afford the products **4a-u** in 65-93% yields.

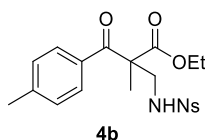
Note: A rough range of 10% to 50% of α -hydroxymethyl products were usually formed in the above Mannich reaction process. Substrates **3a**, **3c**, **3g**, **3k**, and **3t**, which produced the corresponding α -hydroxymethyl products **4a'**, **4c'**, **4g'**, **4k'**, and **4t'**, were exemplified in the following section.



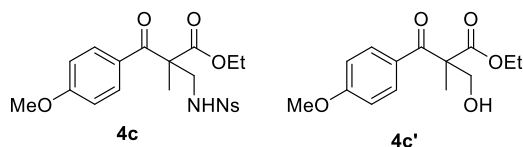
Following the general procedure described above, **4a** (major product, white solid, 71 mg, 85% yield) and **4a'** (minor product, colorless oil, 8 mg, 17% yield) were obtained by flash column chromatography (petroleum ether/EtOAc = 8/1 ~ 6/1).

Data for product **4a**: $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.35 (d, $J = 8.8$ Hz, 2H), 8.04 (d, $J = 8.8$ Hz, 2H), 7.76 (d, 2H), 7.55 (t, 1H), 7.42 (t, 2H), 5.62 (t, $J = 6.8$ Hz, 1H), 4.29 – 4.06 (m, 2H), 3.57 – 3.32 (m, 2H), 1.60 (s, 3H), 1.09 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 197.5, 171.8, 150.0, 146.0, 134.3, 133.5, 128.8, 128.6, 128.2, 124.4, 62.3, 57.7, 49.0, 19.7, 13.7. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{NaO}_7\text{S}$ 443.0883; Found 443.0872. **Melting point**: 150.3-151.4 °C.

Data for product **4a'**: $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.82 (d, $J = 7.6$ Hz, 2H), 7.54 (t, $J = 7.4$ Hz, 1H), 7.43 (t, $J = 7.6$ Hz, 2H), 4.17 (q, $J = 7.1$ Hz, 2H), 3.99 (m, 2H), 2.78 (s, 1H), 1.61 (s, 3H), 1.09 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 199.0, 172.8, 135.3, 133.0, 128.7, 128.5, 67.9, 61.7, 59.0, 18.8, 13.8. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{13}\text{H}_{16}\text{NaO}_4$ 259.0941; Found 259.0933.



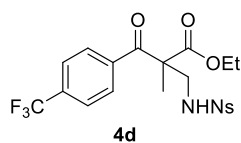
Following the general procedure described above, **4b** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (70 mg, 81% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.33 (d, $J = 8.4$ Hz, 2H), 8.04 (d, $J = 8.5$ Hz, 2H), 7.65 (d, $J = 8.0$ Hz, 2H), 7.20 (d, $J = 8.0$ Hz, 2H), 5.68 (t, $J = 6.7$ Hz, 1H), 4.21 – 4.09 (m, 2H), 3.51 – 3.36 (m, 2H), 2.38 (s, 3H), 1.58 (s, 3H), 1.09 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 196.9, 171.9, 149.9, 146.0, 144.6, 131.6, 129.3, 128.9, 128.2, 124.4, 62.3, 57.6, 49.0, 21.6, 19.8, 13.7. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{20}\text{H}_{22}\text{N}_2\text{NaO}_7\text{S}$ 457.1040; Found 457.1053. **Melting point**: 166.3-168.0 °C.



Following the general procedure described above, **4c** (major product, white solid, 70 mg, 78% yield) and **4c'** (minor product, colorless oil, 14 mg, 27% yield) were obtained by flash column chromatography (petroleum ether/EtOAc = 5/1).

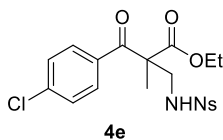
Data for product **4c**: **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.6 Hz, 2H), 8.04 (d, *J* = 8.5 Hz, 2H), 7.76 (d, *J* = 8.8 Hz, 2H), 6.88 (d, *J* = 8.8 Hz, 2H), 5.61 (t, *J* = 6.8 Hz, 1H), 4.24 – 4.09 (m, 2H), 3.86 (s, 3H), 3.50 – 3.32 (m, 2H), 1.59 (s, 3H), 1.11 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 195.8, 172.0, 163.8, 145.0, 146.1, 131.3, 128.2, 126.9, 124.4, 113.8, 62.3, 57.4, 55.5, 49.1, 19.9, 13.8. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₀H₂₂N₂NaO₈S 473.0989; Found 473.1001. **Melting point**: 163.2-164.3 °C.

Data for product **4c'**: **¹H NMR** (400 MHz, CDCl₃) δ 7.82 (d, *J* = 8.8 Hz, 2H), 6.89 (d, *J* = 8.8 Hz, 2H), 4.16 (q, *J* = 7.1 Hz, 2H), 4.00 (d, *J* = 11.2 Hz, 1H), 3.93 (d, *J* = 11.2 Hz, 1H), 3.85 (s, 3H), 2.90 (s, 1H), 1.59 (s, 3H), 1.10 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 197.2, 173.0, 163.4, 131.1, 127.9, 113.7, 67.9, 61.6, 58.6, 55.4, 18.9, 13.8. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₄H₁₈NaO₅ 289.1046; Found 289.1039.

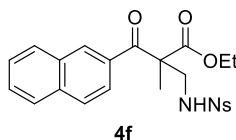


Following the general procedure described above, **4d** was obtained by flash column chromatography (petroleum ether/EtOAc = 5/1) as a white solid (72 mg, 74% yield). **¹H NMR** (400 MHz, acetone-*d*₆) δ 8.45 (d, *J* = 8.8 Hz, 2H), 8.14 (d, *J* = 8.8 Hz, 2H), 7.99 (d, *J* = 8.2 Hz, 2H), 7.87 (d, *J* = 8.3 Hz, 2H), 7.10 (s, 1H), 4.24 – 4.06 (m, 2H), 3.68 – 3.54 (m, 2H), 1.62 (s, 3H), 1.07 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, acetone-*d*₆) δ 196.6, 171.9, 151.0, 147.4, 139.2, 134.3 (q, ²*J*_{C-F} = 32.6 Hz), 130.0, 129.2, 126.6 (q, ³*J*_{C-F} = 7.6 Hz), 125.3, 124.7 (q, ¹*J*_{C-F} = 272.0 Hz), 62.7, 58.8, 49.3, 19.8, 13.9. **HRMS (ESI)** *m/z*: [M - H]⁻ Calcd for C₂₀H₁₈F₃N₂O₇S 487.0792; Found

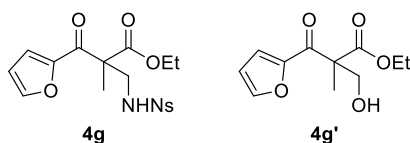
487.0787. **Melting point:** 169.4-170.8 °C.



Following the general procedure described above, **4e** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (76 mg, 83% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.34 (d, *J* = 8.4 Hz, 2H), 8.04 (d, *J* = 8.4 Hz, 2H), 7.71 (d, *J* = 8.2 Hz, 2H), 7.38 (d, *J* = 8.3 Hz, 2H), 5.69 – 5.62 (m, 1H), 4.24 – 4.09 (m, 2H), 3.51 – 3.36 (m, 2H), 1.57 (s, 3H), 1.10 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 196.2, 171.5, 150.0, 146.0, 140.1, 132.5, 130.2, 129.0, 128.2, 124.4, 62.5, 57.8, 48.9, 19.6, 13.7. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₉H₁₉ClN₂NaO₇S 477.0494; Found 477.0507. **Melting point:** 167.2-168.5 °C.



Following the general procedure described above, **4f** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (82 mg, 87% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.37 – 8.21 (m, 3H), 8.03 (d, *J* = 8.5 Hz, 2H), 7.94 – 7.74 (m, 4H), 7.61 (t, *J* = 7.5 Hz, 1H), 7.54 (t, *J* = 7.5 Hz, 1H), 5.76 (t, *J* = 6.8 Hz, 1H), 4.25 – 4.11 (m, 2H), 3.61 – 3.46 (m, 2H), 1.68 (s, 3H), 1.07 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 197.1, 171.9, 149.9, 145.9, 135.4, 132.1, 131.5, 130.6, 129.6, 129.0, 128.4, 128.1, 127.6, 127.1, 124.3, 124.1, 62.3, 57.8, 49.0, 19.9, 13.7. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₃H₂₂N₂NaO₇S 493.1040; Found 493.1039. **Melting point:** 142.0-143.5 °C.

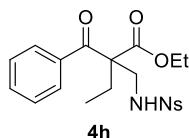


Following the general procedure described above, **4g** (major product, white solid, 53 mg, 65% yield) and **4g'** (minor product, colorless oil, 14 mg, 32% yield) were obtained by flash column chromatography (petroleum ether/EtOAc = 5/1).

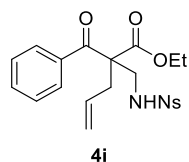
Data for product **4g**: **¹H NMR** (400 MHz, CDCl₃) δ 8.34 (d, *J* = 8.8 Hz, 2H),

8.02 (d, $J = 8.8$ Hz, 2H), 7.53 (s, 1H), 7.23 (d, $J = 3.6$ Hz, 1H), 6.56 – 6.50 (m, 1H), 5.62 (t, $J = 6.8$ Hz, 1H), 4.23 – 4.04 (m, 2H), 3.43 (d, $J = 6.3$ Hz, 2H), 1.56 (s, 3H), 1.10 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 186.0, 171.0, 150.9, 149.9, 146.5, 146.0, 128.2, 124.4, 119.1, 112.7, 62.0, 57.1, 47.8, 18.5, 13.8. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{17}\text{H}_{18}\text{N}_2\text{NaO}_8\text{S}$ 433.0676; Found 433.0686. **Melting point:** 163.1-165.4 °C.

Data for product **4g'**: ^1H NMR (400 MHz, CDCl_3) δ 7.54 (s, 1H), 7.32 – 7.18 (d, $J = 3.5$ Hz, 1H), 6.53 (d, $J = 3.4$ Hz, 1H), 4.24 – 4.08 (m, 2H), 4.03 – 3.91 (m, 2H), 2.80 (t, $J = 6.8$ Hz, 1H), 1.56 (s, 3H), 1.10 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 187.3, 172.1, 151.5, 146.1, 118.4, 112.4, 66.7, 61.5, 58.5, 17.5, 13.9. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{11}\text{H}_{14}\text{NaO}_5$ 249.0733; Found 249.0727.

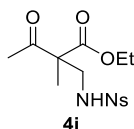


Following the general procedure described above, **4h** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (70 mg, 80% yield). ^1H NMR (400 MHz, CDCl_3) δ 8.33 (d, $J = 8.4$ Hz, 2H), 8.02 (d, $J = 8.4$ Hz, 2H), 7.74 (d, $J = 7.8$ Hz, 2H), 7.54 (t, $J = 7.4$ Hz, 1H), 7.40 (t, $J = 7.7$ Hz, 2H), 5.37 – 5.25 (m, 1H), 4.24 – 4.06 (m, 2H), 3.60 (dd, $J = 13.1, 5.6$ Hz, 1H), 3.45 (dd, $J = 13.1, 7.8$ Hz, 1H), 2.29 – 2.06 (m, 2H), 1.08 (t, $J = 7.1$ Hz, 3H), 0.75 (t, $J = 7.5$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 196.9, 171.6, 150.0, 145.7, 135.1, 133.4, 128.6, 128.3, 128.2, 124.4, 62.1, 61.4, 45.8, 25.8, 13.7, 8.0. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{20}\text{H}_{22}\text{N}_2\text{NaO}_7\text{S}$ 457.1040; Found 457.1051. **Melting point:** 132.0-132.8 °C.

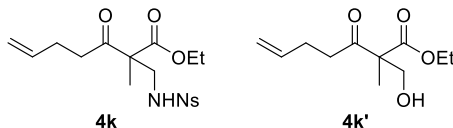


Following the general procedure described above, **4i** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (76 mg, 85% yield). ^1H NMR (400 MHz, CDCl_3) δ 8.31 (d, $J = 8.8$ Hz, 2H), 8.00 (d, $J = 8.8$ Hz, 2H), 7.73 (d, $J = 7.1$ Hz, 2H), 7.55 (t, $J = 7.4$ Hz, 1H), 7.40 (t, $J = 7.8$ Hz, 2H), 5.54 –

5.39 (m, 1H), 5.35 (t, $J = 6.6$ Hz, 1H), 5.06 – 4.93 (m, 2H), 4.25 – 4.09 (m, 2H), 3.59 (dd, $J = 13.2, 5.0$ Hz, 1H), 3.45 (dd, $J = 13.3, 7.5$ Hz, 1H), 2.91 (dd, $J = 14.8, 7.1$ Hz, 1H), 2.82 (dd, $J = 14.8, 7.9$ Hz, 1H), 1.10 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 196.3, 171.1, 150.0, 145.7, 135.0, 133.5, 130.5, 128.6, 128.5, 128.2, 124.4, 120.7, 62.4, 60.7, 46.5, 37.6, 13.7. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{21}\text{H}_{22}\text{N}_2\text{NaO}_7\text{S}$ 469.1040; Found 469.1049. **Melting point:** 145.6-146.4 °C.



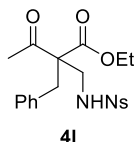
Following the general procedure described above, **4j** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (63 mg, 88% yield). ^1H NMR (400 MHz, CDCl_3) δ 8.36 (d, $J = 8.9$ Hz, 2H), 8.03 (d, $J = 9.0$ Hz, 2H), 5.56 (t, $J = 6.9$ Hz, 1H), 4.19 (q, $J = 7.0$ Hz, 2H), 3.34 – 3.19 (m, 2H), 2.19 (s, 3H), 1.44 (s, 3H), 1.26 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 205.6, 171.0, 150.0, 145.7, 128.2, 124.4, 62.2, 59.8, 47.3, 26.5, 18.2, 13.9. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{14}\text{H}_{18}\text{N}_2\text{NaO}_7\text{S}$ 381.0727; Found 381.0733. **Melting point:** 115.2-116.8 °C.



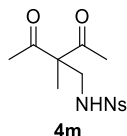
Following the general procedure described above, **4k** (major product, colorless oil, 58 mg, 73% yield) and **4k'** (minor product, colorless oil, 21 mg, 50% yield) were obtained by flash column chromatography (petroleum ether/EtOAc = 8/1).

Data for product **4k**: ^1H NMR (400 MHz, CDCl_3) δ 8.36 (d, $J = 8.5$ Hz, 2H), 8.03 (d, $J = 8.6$ Hz, 2H), 5.82 – 5.66 (m, 1H), 5.51 (t, $J = 6.9$ Hz, 1H), 5.07 – 4.91 (m, 2H), 4.18 (q, $J = 6.3, 5.7$ Hz, 2H), 3.34 – 3.20 (m, 2H), 2.66 – 2.48 (m, 2H), 2.37 – 2.19 (m, 2H), 1.44 (s, 3H), 1.25 (t, $J = 7.1$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 206.9, 171.0, 150.0, 145.7, 136.4, 128.2, 124.4, 115.7, 62.2, 59.6, 47.5, 38.0, 27.5, 18.1, 13.9. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{17}\text{H}_{22}\text{N}_2\text{NaO}_7\text{S}$ 421.1040; Found 421.1047.

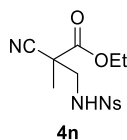
Data for product **4k'**: $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 5.87 – 5.70 (m, 1H), 5.09 – 4.91 (m, 2H), 4.23 (q, $J = 7.1$ Hz, 2H), 3.87 (d, $J = 11.4$ Hz, 1H), 3.80 (d, $J = 11.4$ Hz, 1H), 2.84 – 2.49 (m, 3H), 2.38 – 2.26 (m, 2H), 1.38 (s, 3H), 1.28 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 207.7, 172.4, 136.9, 115.4, 66.6, 61.6, 61.1, 38.4, 27.5, 17.1, 14.0. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{11}\text{H}_{18}\text{NaO}_4$ 237.1097; Found 237.1092.



Following the general procedure described above, **4l** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (75 mg, 86% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.34 (d, $J = 8.9$ Hz, 2H), 7.98 (d, $J = 8.9$ Hz, 2H), 7.25 – 7.18 (m, 3H), 7.09 – 7.00 (m, 2H), 5.24 (t, 1H), 4.32 – 4.17 (m, 2H), 3.38 – 3.14 (m, 4H), 2.25 (s, 3H), 1.30 (t, $J = 7.1$ Hz, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 204.5, 170.4, 150.1, 145.3, 134.3, 129.7, 128.7, 128.2, 127.6, 124.4, 64.4, 62.4, 45.4, 37.9, 27.9, 13.9. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{20}\text{H}_{22}\text{N}_2\text{NaO}_7\text{S}$ 457.1040; Found 457.1050. **Melting point:** 76.2-78.0 °C.

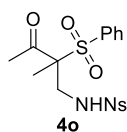


Following the general procedure described above, **4m** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (53 mg, 80% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.36 (d, $J = 8.8$ Hz, 2H), 8.03 (d, $J = 8.7$ Hz, 2H), 5.61 – 5.52 (m, 1H), 3.23 (d, $J = 6.8$ Hz, 2H), 2.17 (s, 6H), 1.44 (s, 3H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 207.2, 150.1, 145.4, 128.2, 124.5, 65.6, 47.1, 26.8, 17.8. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{13}\text{H}_{16}\text{N}_2\text{NaO}_6\text{S}$ 351.0621; Found 351.0626. **Melting point:** 133.3-134.8 °C.

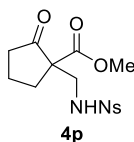


Following the general procedure described above, **4n** was obtained by flash

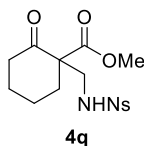
column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (61 mg, 90% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.37 (d, *J* = 8.7 Hz, 2H), 8.07 (d, *J* = 8.6 Hz, 2H), 5.81 (t, *J* = 6.7 Hz, 1H), 4.26 (q, *J* = 7.4, 6.9 Hz, 2H), 3.48 (dd, *J* = 13.6, 6.1 Hz, 1H), 3.41 (dd, *J* = 13.6, 6.3 Hz, 1H), 1.62 (s, 3H), 1.31 (t, *J* = 7.1 Hz, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 167.6, 150.2, 145.5, 128.3, 124.5, 118.1, 63.7, 48.2, 44.6, 20.8, 13.8. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₃H₁₅N₃NaO₆S 364.0574; Found 364.0578. **Melting point:** 128.1-129.5 °C.



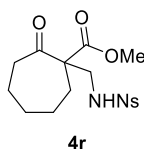
Following the general procedure described above, **4o** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (78 mg, 91% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.4 Hz, 2H), 7.98 (d, *J* = 8.4 Hz, 2H), 7.80 – 7.63 (m, 3H), 7.55 (t, *J* = 7.7 Hz, 2H), 5.34 (t, *J* = 6.7 Hz, 1H), 3.48 (dd, *J* = 13.6, 6.3 Hz, 1H), 3.30 (dd, *J* = 13.5, 6.9 Hz, 1H), 2.43 (s, 3H), 1.75 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 202.7, 150.2, 145.3, 135.1, 134.2, 129.8, 129.3, 128.2, 124.5, 75.7, 45.1, 28.3, 15.4. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₇H₁₈N₂NaO₇S₂ 449.04476; Found 449.04484. **Melting point:** 158.9-160.8 °C.



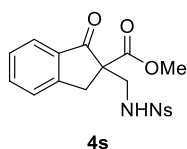
Following the general procedure described above, **4p** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (60 mg, 84% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.36 (d, *J* = 8.7 Hz, 2H), 8.02 (d, *J* = 8.8 Hz, 2H), 5.67 (t, *J* = 6.8 Hz, 1H), 3.68 (s, 3H), 3.30 – 3.14 (m, 2H), 2.58 – 2.19 (m, 4H), 2.17 – 1.93 (m, 2H). **¹³C NMR** (100 MHz, CDCl₃) δ 213.6, 171.4, 150.0, 145.6, 128.2, 124.5, 59.9, 53.0, 45.0, 38.1, 31.7, 19.5. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₄H₁₆N₂NaO₇S 379.0570; Found 379.0580. **Melting point:** 93.5-94.7 °C.



Following the general procedure described above, **4q** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (63 mg, 85% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.36 (d, *J* = 8.4 Hz, 2H), 8.03 (d, *J* = 8.5 Hz, 2H), 5.64 – 5.46 (m, 1H), 3.73 (s, 3H), 3.26 (dd, *J* = 12.7, 8.4 Hz, 1H), 3.11 (dd, *J* = 12.7, 5.2 Hz, 1H), 2.70 – 2.54 (m, 1H), 2.51 – 2.33 (m, 2H), 2.11 – 1.95 (m, 1H), 1.88 – 1.74 (m, 1H), 1.66 – 1.57 (m, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 209.1, 170.7, 150.0, 145.8, 128.2, 124.4, 61.4, 53.0, 47.3, 40.7, 34.1, 26.9, 21.9. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₅H₁₈N₂NaO₇S 393.0727; Found 393.0733. **Melting point:** 152.1-153.6 °C.

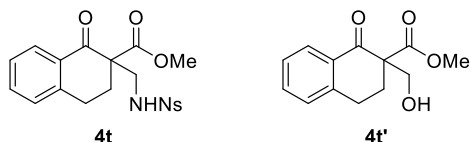


Following the general procedure described above, **4r** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (58 mg, 75% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.8 Hz, 2H), 8.02 (d, *J* = 8.8 Hz, 2H), 5.73 – 5.65 (m, 1H), 3.70 (s, 3H), 3.33 – 3.19 (m, 2H), 2.79 – 2.68 (m, 1H), 2.54 – 2.43 (m, 1H), 2.10 – 1.99 (m, 1H), 1.96 – 1.85 (m, 1H), 1.85 – 1.70 (m, 2H), 1.69 – 1.48 (m, 3H), 1.41 – 1.28 (m, 1H). **¹³C NMR** (100 MHz, CDCl₃) δ 210.2, 171.4, 149.9, 145.8, 128.2, 124.4, 62.7, 52.8, 47.4, 42.5, 31.5, 29.8, 25.7, 24.9. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₆H₂₀N₂NaO₇S 407.0883; Found 407.0892. **Melting point:** 165.4-166.5 °C.



Following the general procedure described above, **4s** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (75 mg, 93% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.5 Hz, 2H), 8.01 (d, *J* = 8.5 Hz, 2H), 7.73 (d, *J* = 7.7 Hz, 1H), 7.68 (t, *J* = 7.5 Hz, 1H), 7.54 (d, *J* = 7.7 Hz, 1H), 7.43

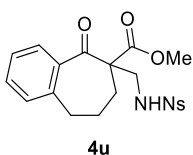
(t, $J = 7.5$ Hz, 1H), 5.77 – 5.70 (m, 1H), 3.65 (s, 3H), 3.58 – 3.48 (m, 3H), 3.30 (dd, $J = 13.0, 8.8$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 200.3, 171.2, 153.2, 150.1, 145.5, 136.2, 134.4, 128.2, 126.8, 125.1, 124.5, 60.1, 53.2, 46.5, 35.6. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{18}\text{H}_{16}\text{N}_2\text{NaO}_7\text{S}$ 427.0570; Found 427.0577. **Melting point:** 144.1-146.0 °C.



Following the general procedure described above, **4t** (major product, white solid, 69 mg, 82% yield) and **4t'** (minor product, colorless oil, 14 mg, 29% yield) were obtained by flash column chromatography (petroleum ether/EtOAc = 8/1 ~ 5/1).

Data for product **4t**: ^1H NMR (400 MHz, CDCl_3) δ 8.36 (d, $J = 8.7$ Hz, 2H), 8.04 (d, $J = 8.7$ Hz, 2H), 7.98 (d, $J = 7.8$ Hz, 1H), 7.52 (t, $J = 7.4$ Hz, 1H), 7.33 (t, $J = 7.6$ Hz, 1H), 7.25 (d, $J = 7.8$ Hz, 1H), 5.88 – 5.79 (m, 1H), 3.65 (s, 3H), 3.51 (dd, $J = 12.6, 8.7$ Hz, 1H), 3.30 (dd, $J = 12.7, 4.6$ Hz, 1H), 3.05 – 2.98 (m, 2H), 2.51 (dt, $J = 13.7, 4.7$ Hz, 1H), 2.30 (dt, $J = 13.6, 7.7$ Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 195.4, 171.1, 150.0, 145.7, 143.1, 134.5, 131.2, 128.9, 128.3, 127.9, 127.1, 124.4, 58.0, 53.0, 46.9, 29.9, 25.6. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{19}\text{H}_{18}\text{N}_2\text{NaO}_7\text{S}$ 441.0727; Found 441.0737. **Melting point:** 177.0-178.4 °C.

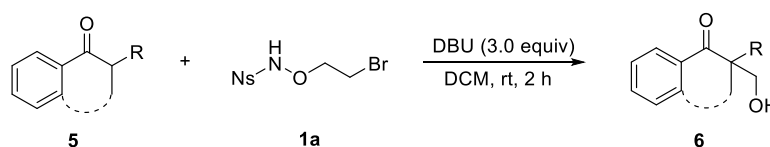
Data for product **4t'**: ^1H NMR (400 MHz, CDCl_3) δ 8.07 (d, $J = 7.9$ Hz, 1H), 7.51 (t, $J = 7.5$ Hz, 1H), 7.34 (t, $J = 7.5$ Hz, 1H), 7.27 – 7.20 (m, 1H), 4.02 (dd, $J = 11.3, 3.4$ Hz, 1H), 3.90 (t, $J = 10.6$ Hz, 1H), 3.72 (s, 3H), 3.28 (dd, $J = 9.9, 4.1$ Hz, 1H), 3.08 – 2.90 (m, 2H), 2.45 (dt, $J = 13.7, 4.6$ Hz, 1H), 2.23 – 2.12 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 197.2, 171.7, 143.2, 134.0, 131.8, 128.8, 127.8, 127.0, 66.3, 59.2, 52.6, 29.2, 25.9. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{13}\text{H}_{14}\text{NaO}_4$ 257.0784; Found 257.0777.



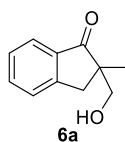
Following the general procedure described above, **4u** was obtained by flash

column chromatography (petroleum ether/EtOAc = 8/1) as a white solid (62 mg, 72% yield). **¹H NMR** (400 MHz, CDCl₃) δ 8.36 (d, *J* = 8.2 Hz, 2H), 8.05 (d, *J* = 8.2 Hz, 2H), 7.45 (d, *J* = 7.5 Hz, 1H), 7.40 (t, *J* = 7.3 Hz, 1H), 7.29 (d, *J* = 7.5 Hz, 1H), 7.14 (d, *J* = 7.3 Hz, 1H), 5.78 – 5.74 (m, 1H), 3.59 (s, 3H), 3.48 – 3.33 (m, 2H), 2.98 – 2.86 (m, 1H), 2.86 – 2.74 (m, 1H), 2.38 – 2.27 (m, 1H), 2.07 – 1.89 (m, 3H). **¹³C NMR** (100 MHz, CDCl₃) δ 204.3, 171.3, 150.0, 145.8, 138.8, 138.7, 132.2, 129.4, 128.2, 126.7, 124.5, 62.0, 52.9, 47.9, 33.0, 30.1, 22.9. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₀H₂₀N₂NaO₇S 455.0883; Found 455.0893. **Melting point**: 189.1-190.7 °C.

6.2 General procedure for the preparation of hydroxymethyl products 6

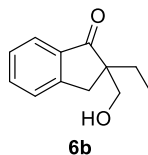


To a solution of α -substituted ketone **5** (0.2 mmol, 1.0 equiv) and DBU (91 mg, 0.6 mmol, 3.0 equiv) in DCM (1 mL) was added the solution of **1a** (98 mg, 0.3 mmol, 1.5 equiv) in DCM (1 mL) dropwise at room temperature over 10 minutes. The reaction mixture was stirred at the same temperature for 2 h. After completion, the solution was quenched with saturated NH₄Cl (2 mL). The organic layer was separated, and the aqueous layer was extracted with DCM (5 mL x 3). The combined organic layers were washed with brine (5 mL), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash column chromatography (petroleum ether/EtOAc = 8/1 ~ 5/1) to afford the products **6a-i** in 61-91% yields.

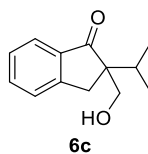


Following the general procedure described above, **6a** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (32 mg, 91% yield). **¹H NMR** (400 MHz, CDCl₃) δ 7.72 (d, *J* = 7.6 Hz, 1H), 7.59 (t, *J* = 7.4 Hz,

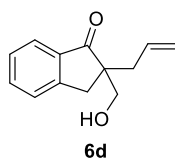
1H), 7.45 (d, $J = 7.6$ Hz, 1H), 7.35 (t, $J = 7.3$ Hz, 1H), 3.82 (d, $J = 10.7$ Hz, 1H), 3.61 (d, $J = 10.7$ Hz, 1H), 3.25 (d, $J = 17.2$ Hz, 1H), 2.88 (d, $J = 17.2$ Hz, 1H), 2.48 (s, 1H), 1.23 (s, 3H). The spectral data of **6a** was consistent with that reported in the literature.²⁴



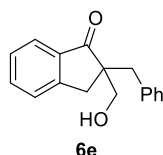
Following the general procedure described above, **6b** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (32 mg, 85% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.71 (d, $J = 7.7$ Hz, 1H), 7.59 (t, $J = 7.4$ Hz, 1H), 7.46 (d, $J = 7.6$ Hz, 1H), 7.35 (t, $J = 7.4$ Hz, 1H), 3.85 (d, $J = 10.7$ Hz, 1H), 3.61 (d, $J = 10.8$ Hz, 1H), 3.12 – 2.94 (m, 2H), 2.46 (s, 1H), 1.81 – 1.61 (m, 2H), 0.80 (t, $J = 7.5$ Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 211.6, 153.9, 136.7, 135.2, 127.4, 126.5, 123.8, 66.5, 54.8, 34.9, 26.5, 8.5. HRMS (ESI) m/z : [M + Na]⁺ Calcd for C₁₂H₁₄NaO₂ 213.0886; Found 213.0878.



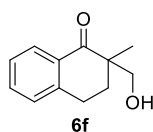
Following the general procedure described above, **6c** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (34 mg, 83% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.72 (d, $J = 7.7$ Hz, 1H), 7.59 (t, $J = 7.4$ Hz, 1H), 7.46 (d, $J = 7.7$ Hz, 1H), 7.35 (t, $J = 7.4$ Hz, 1H), 3.93 – 3.84 (m, 1H), 3.61 (d, $J = 10.8$ Hz, 1H), 3.08 (d, $J = 17.6$ Hz, 1H), 2.73 (d, $J = 17.6$ Hz, 1H), 2.51 (dd, $J = 8.6, 3.5$ Hz, 1H), 2.41 (hept, $J = 6.8$ Hz, 1H), 0.98 (d, $J = 6.8$ Hz, 3H), 0.67 (d, $J = 6.9$ Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 212.7, 154.4, 136.9, 135.1, 127.4, 126.3, 123.7, 66.0, 57.7, 31.6, 29.6, 17.6, 17.5. HRMS (ESI) m/z : [M + Na]⁺ Calcd for C₁₃H₁₆NaO₂ 227.1043; Found 227.1041.



Following the general procedure described above, **6d** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (37 mg, 91% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.70 (d, $J = 7.7$ Hz, 1H), 7.58 (t, $J = 7.4$ Hz, 1H), 7.44 (d, $J = 7.7$ Hz, 1H), 7.33 (t, $J = 7.4$ Hz, 1H), 5.68 – 5.53 (m, 1H), 5.09 (d, $J = 16.9$ Hz, 1H), 5.00 (d, $J = 10.1$ Hz, 1H), 3.83 (d, $J = 10.9$ Hz, 1H), 3.62 (d, $J = 10.9$ Hz, 1H), 3.12 – 2.98 (m, 2H), 2.57 (s, 1H), 2.48 – 2.34 (m, 2H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 210.7, 153.7, 136.3, 135.2, 132.8, 127.4, 126.5, 123.9, 118.8, 66.6, 54.2, 38.1, 34.6. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{13}\text{H}_{14}\text{NaO}_2$ 225.0886; Found 225.0884.

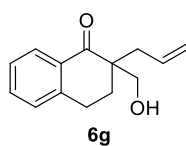


Following the general procedure described above, **6e** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (44 mg, 88% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.71 (d, $J = 7.6$ Hz, 1H), 7.53 (t, $J = 7.4$ Hz, 1H), 7.38 – 7.24 (m, 2H), 7.24 – 7.10 (m, 5H), 3.83 (d, $J = 10.9$ Hz, 1H), 3.62 (d, $J = 10.9$ Hz, 1H), 3.15 (d, $J = 17.4$ Hz, 1H), 3.05 (d, $J = 13.5$ Hz, 1H), 2.97 (d, $J = 13.5$ Hz, 1H), 2.82 (d, $J = 17.4$ Hz, 1H), 2.68 (s, 1H). $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 211.1, 153.5, 136.7, 136.1, 135.2, 130.2, 128.1, 127.4, 126.5, 126.4, 124.0, 66.5, 55.1, 38.6, 34.2. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{17}\text{H}_{16}\text{NaO}_2$ 275.1043; Found 275.1042.

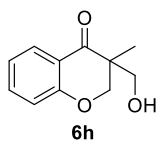


Following the general procedure described above, **6f** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (24 mg, 63% yield). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.01 (d, $J = 7.8$ Hz, 1H), 7.49 (t, $J = 7.4$ Hz, 1H), 7.31 (t, $J = 7.5$ Hz, 1H), 7.25 (d, $J = 7.8$ Hz, 1H), 3.82 – 3.57 (m, 2H), 3.16 (ddd, $J = 17.2, 12.0, 5.0$ Hz, 1H), 2.94 (dt, $J = 17.4, 4.3$ Hz, 1H), 2.85 (t, $J = 6.6$ Hz, 1H), 2.23 (td, $J = 12.8, 5.1$ Hz, 1H), 1.82 – 1.71 (m, 1H), 1.23 (s, 3H). The spectral data of

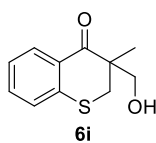
6f was consistent with that reported in the literature.²⁵



Following the general procedure described above, **6g** was obtained by flash column chromatography (petroleum ether/EtOAc = 8/1) as a colorless oil (26 mg, 61% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.01 (d, *J* = 7.8 Hz, 1H), 7.49 (t, *J* = 7.5 Hz, 1H), 7.31 (t, *J* = 7.6 Hz, 1H), 7.25 (d, *J* = 7.8 Hz, 1H), 5.91 – 5.72 (m, 1H), 5.25 – 5.09 (m, 2H), 3.79 – 3.61 (m, 2H), 3.13 (ddd, *J* = 16.9, 11.4, 5.1 Hz, 1H), 2.99 – 2.84 (m, 2H), 2.50 (dd, *J* = 14.0, 6.9 Hz, 1H), 2.34 (dd, *J* = 13.9, 8.0 Hz, 1H), 2.05 (ddd, *J* = 13.7, 11.4, 5.1 Hz, 1H), 1.93 (dt, *J* = 13.8, 4.5 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 203.8, 143.5, 133.7, 132.7, 131.4, 128.8, 127.7, 126.8, 119.0, 66.3, 49.0, 34.7, 27.8, 24.7. HRMS (ESI) *m/z*: [M + Na]⁺ Calcd for C₁₄H₁₆NaO₂ 239.1043; Found 239.1042.



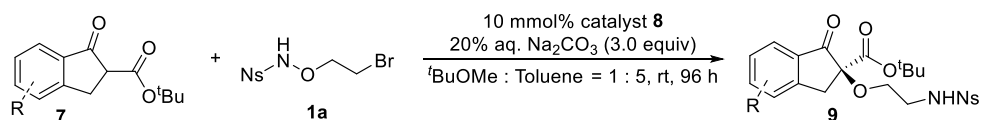
Following the general procedure described above, **6h** was obtained by flash column chromatography (petroleum ether/EtOAc = 5/1) as a colorless oil (31 mg, 81% yield). ¹H NMR (400 MHz, CDCl₃) δ 7.89 (d, *J* = 7.8 Hz, 1H), 7.49 (t, *J* = 7.7 Hz, 1H), 7.08 – 6.85 (m, 2H), 4.52 (d, *J* = 11.5 Hz, 1H), 4.18 (d, *J* = 11.5 Hz, 1H), 3.93 (dd, *J* = 11.1, 6.6 Hz, 1H), 3.59 (dd, *J* = 11.2, 6.3 Hz, 1H), 2.29 (t, *J* = 6.3 Hz, 1H), 1.23 (s, 3H). The spectral data of **6h** was consistent with that reported in the literature.²⁶



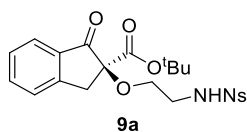
Following the general procedure described above, **6i** was obtained by flash column chromatography (petroleum ether/EtOAc = 5/1) as a colorless oil (26 mg, 62% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.07 (d, *J* = 8.0 Hz, 1H), 7.37 (t, *J* = 7.5 Hz, 1H), 7.23 (d, *J* = 7.9 Hz, 1H), 7.16 (t, *J* = 7.6 Hz, 1H), 3.98 (dd, *J* = 11.3, 6.7 Hz, 1H),

3.66 – 3.57 (m, 2H), 2.78 (d, $J = 13.6$ Hz, 1H), 2.57 (t, $J = 6.7$ Hz, 1H), 1.33 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 199.0, 141.6, 133.3, 129.9, 129.7, 127.2, 124.8, 67.8, 45.8, 34.7, 17.6. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{11}\text{H}_{12}\text{NaO}_2\text{S}$ 231.0450; Found 231.0449.

7. General procedure for the catalytic asymmetric reactions of indanone carboxylates **7** with hydroxylamine **1a**

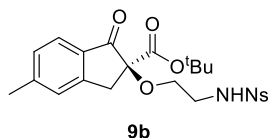


To a solution of *tert*-butyl indanone carboxylate substrates **7** (0.24 mmol, 1.2 equiv), hydroxylamine **1a** (0.2 mmol, 1.0 equiv), and catalyst **8** (16 mg, 0.02mmol, 0.1 equiv) in *t*BuOMe/toluene (2 mL, $v/v = 1/5$) was added 20% aq. Na_2CO_3 (318 μL , 0.6 mmol, 3.0 equiv) at room temperature. The reaction mixture was stirred at the same temperature for 96 h. After completion, the solution was quenched with saturated NH_4Cl (2 mL) and extracted with EtOAc (3 mL x 3), dried over Na_2SO_4 and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 10/1 ~ 5/1) to afford the products **9a-1** in 50-67% yield and 90-97% ee. The absolute configuration of products **9** was determined to be the *S* by comparing their specific rotation value with the reported data in the literature.¹

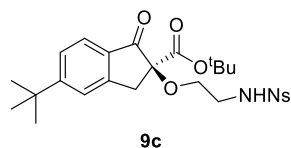


Following the general procedure described above, **9a** was obtained by flash silica chromatography (PE/EtOAc = 10/1) as a white solid (56 mg, 60% yield). The compound was determined to be 91% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 23.9$ min, $t(\text{minor}) = 32.2$. $[\alpha]_{\text{D}}^{25} = -10.9$ ($c = 1.05$, CHCl_3); Reported data from Lit.¹: $[\alpha]_{\text{D}}^{25} = -9.6$ ($c = 0.8$, CHCl_3). ^1H NMR (400 MHz, CDCl_3) δ 8.32 (d, $J = 8.6$ Hz, 2H), 8.08 (d, $J = 8.6$ Hz, 2H), 7.74 (d, $J = 7.6$ Hz, 1H), 7.65 (t, $J = 7.5$ Hz,

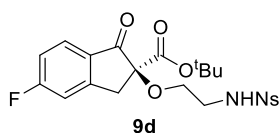
1H), 7.47 – 7.38 (m, 2H), 6.24 (s, 1H), 3.88 – 3.67 (m, 2H), 3.47 (d, $J = 17.2$ Hz, 1H), 3.29 – 3.16 (m, 2H), 3.07 (d, $J = 17.2$ Hz, 1H), 1.37 (s, 9H). The spectral data of **9a** was consistent with that reported in the literature.¹



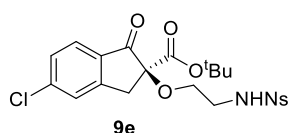
Following the general procedure described above, **9b** was obtained by flash silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (48 mg, 51% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 20.5$ min, $t(\text{minor}) = 25.8$ min. **¹H NMR** (400 MHz, CDCl₃) δ 8.31 (d, $J = 8.8$ Hz, 2H), 8.08 (d, $J = 8.8$ Hz, 2H), 7.64 (d, $J = 8.2$ Hz, 1H), 7.25 – 7.17 (m, 2H), 6.20 (s, 1H), 3.91 – 3.66 (m, 2H), 3.41 (d, $J = 17.2$ Hz, 1H), 3.30 – 3.16 (m, 2H), 2.99 (d, $J = 17.2$ Hz, 1H), 2.46 (s, 3H), 1.38 (s, 9H). The spectral data of **9b** was consistent with that reported in the literature.¹



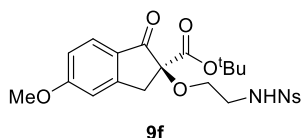
Following the general procedure described above, **9c** was obtained by flash silica chromatography (petroleum ether/EtOAc = 10/1) as a colorless oil (63 mg, 59% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak AD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 19.7$ min, $t(\text{minor}) = 23.5$ min. $[\alpha]_D^{25} = 4.2$ ($c = 0.4$, CHCl₃); Reported data from Lit.¹: $[\alpha]_D^{25} = 3.1$ ($c = 0.5$, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.31 (d, $J = 8.1$ Hz, 2H), 8.08 (d, $J = 8.1$ Hz, 2H), 7.66 (d, $J = 8.1$ Hz, 1H), 7.50 – 7.37 (m, 2H), 6.28 (s, 1H), 3.84 – 3.65 (m, 2H), 3.45 (d, $J = 17.1$ Hz, 1H), 3.27 – 3.14 (m, 2H), 3.03 (d, $J = 17.1$ Hz, 1H), 1.38 (s, 9H), 1.36 (s, 9H). The spectral data of **9c** was consistent with that reported in the literature.¹



Following the general procedure described above, **9d** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (52 mg, 53% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 75/25, 1.0 mL/min, 254 nm, 47 bar, 25 °C; $t(\text{major}) = 14.1$ min, $t(\text{minor}) = 22.8$ min. **¹H NMR** (400 MHz, CDCl₃) δ 8.33 (d, $J = 8.7$ Hz, 2H), 8.09 (d, $J = 8.7$ Hz, 2H), 7.77 (dd, $J = 7.9, 5.4$ Hz, 1H), 7.15 – 7.08 (m, 2H), 6.16 (s, 1H), 3.89 – 3.66 (m, 2H), 3.47 (d, $J = 17.4$ Hz, 1H), 3.31 – 3.15 (m, 2H), 3.08 (d, $J = 17.4$ Hz, 1H), 1.38 (s, 9H). The spectral data of **9d** was consistent with that reported in the literature.¹

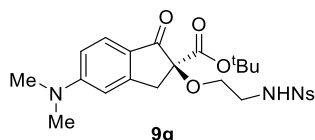


Following the general procedure described above, **9e** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (64 mg, 63% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 27.5$ min, $t(\text{minor}) = 39.0$ min. **¹H NMR** (400 MHz, CDCl₃) δ 8.33 (d, $J = 8.6$ Hz, 2H), 8.08 (d, $J = 8.6$ Hz, 2H), 7.68 (d, $J = 8.1$ Hz, 1H), 7.52 – 7.32 (m, 2H), 6.17 (s, 1H), 3.86 – 3.68 (m, 2H), 3.45 (d, $J = 17.4$ Hz, 1H), 3.29 – 3.15 (m, 2H), 3.07 (d, $J = 17.4$ Hz, 1H), 1.38 (s, 9H). The spectral data of **9e** was consistent with that reported in the literature.¹

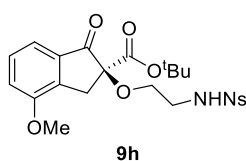


Following the general procedure described above, **9f** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (57 mg, 56% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:**

Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 35.7$ min, $t(\text{minor}) = 44.8$ min. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.31 (d, $J = 8.6$ Hz, 2H), 8.08 (d, $J = 8.6$ Hz, 2H), 7.68 (d, $J = 8.6$ Hz, 1H), 6.94 (d, $J = 8.4$ Hz, 1H), 6.84 (s, 1H), 6.22 (s, 1H), 3.91 (s, 3H), 3.87 – 3.64 (m, 2H), 3.44 (d, $J = 17.2$ Hz, 1H), 3.27 – 3.15 (m, 2H), 2.98 (d, $J = 17.2$ Hz, 1H), 1.39 (s, 9H). The spectral data of **9f** was consistent with that reported in the literature.¹

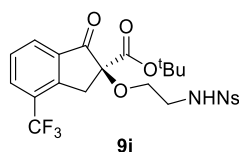


Following the general procedure described above (1.0 equiv of 50% aq. Cs_2CO_3 was used), **9g** was obtained by flash silica chromatography (petroleum ether/EtOAc = 5/1) as a yellow solid (70 mg, 67% yield). The compound was determined to be 96% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 80/20, 1.0 mL/min, 254 nm, 50 bar, 25 °C; $t(\text{major}) = 24.1$ min, $t(\text{minor}) = 31.4$ min. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.27 (d, $J = 8.8$ Hz, 2H), 8.7 (d, $J = 8.7$ Hz, 2H), 7.59 (d, $J = 8.8$ Hz, 1H), 6.75 – 6.62 (m, 1H), 6.46 (s, 1H), 6.31 (s, 1H), 3.82 – 3.67 (m, 2H), 3.34 (d, $J = 17.0$ Hz, 1H), 3.27 – 3.18 (m, 2H), 3.12 (s, 6H), 2.85 (d, $J = 17.0$ Hz, 1H), 1.39 (s, 9H). The spectral data of **9g** was consistent with that reported in the literature.¹

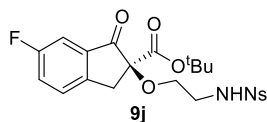


Following the general procedure described above, **9h** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (48mg, 50% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak AD-H, Hexane/*i*-PrOH = 75/25, 1.0 mL/min, 254 nm, 51 bar, 25 °C; $t(\text{major}) = 14.5$ min, $t(\text{minor}) = 20.9$ min. $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.30 (d, $J = 8.6$ Hz, 2H), 8.06 (d, $J = 8.6$ Hz, 2H), 7.42 – 7.29 (m, 2H), 7.07 (d, $J = 7.8$ Hz, 1H), 6.25– 6.17 (m, 1H), 3.90 (s, 3H), 3.85 – 3.65 (m, 2H), 3.35 – 3.17 (m, 3H), 2.81 (d, $J = 17.6$ Hz, 1H), 1.38 (s, 9H). The spectral data of **9h** was consistent with that reported

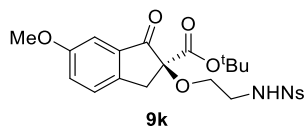
in the literature.¹



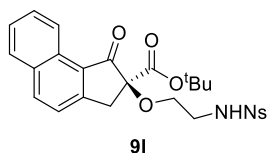
Following the general procedure described above, **9i** was obtained by flash silica chromatography (petroleum ether/EtOAc = 12/1) as a colorless oil (72 mg, 66% yield). The compound was determined to be 95% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 93/7, 1.0 mL/min, 254 nm, 39 bar, 25 °C; $t(\text{major}) = 47.5$ min, $t(\text{minor}) = 53.9$ min. $[\alpha]_D^{25} = -12.2$ ($c = 0.7$, CHCl_3). **¹H NMR** (400 MHz, CDCl_3) δ 8.32 (d, $J = 8.8$ Hz, 2H), 8.08 (d, $J = 8.8$ Hz, 2H), 7.91 (t, $J = 8.3$ Hz, 2H), 7.57 (t, $J = 7.7$ Hz, 1H), 6.14 (s, 1H), 3.89– 3.70 (m, 2H), 3.60 (d, $J = 17.8$ Hz, 1H), 3.29 – 3.14 (m, 3H), 1.36 (s, 9H). **¹³C NMR** (100 MHz, CDCl_3) δ 197.1, 168.7, 149.9, 148.5 (q, $^3J_{\text{C-F}} = 2.5$ Hz), 146.0, 135.5, 132.4 (q, $^2J_{\text{C-F}} = 5.5$ Hz), 128.8, 128.5, 128.4, 128.2, 124.2, 123.4 (q, $^1J_{\text{C-F}} = 272.0$ Hz), 84.6, 84.0, 65.1, 43.4, 38.1, 27.7. **HRMS (ESI)** m/z : $[\text{M} + \text{Na}]^+$ Calcd for $\text{C}_{23}\text{H}_{23}\text{F}_3\text{N}_2\text{NaO}_8\text{S}$ 567.1019; Found 567.1041.



Following the general procedure described above, **9j** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (57 mg, 58% yield). The compound was determined to be 93% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 31.2$ min, $t(\text{minor}) = 40.9$ min. **¹H NMR** (400 MHz, CDCl_3) δ 8.34 (d, $J = 8.8$ Hz, 2H), 8.09 (d, $J = 8.8$ Hz, 2H), 7.49 – 7.31 (m, 3H), 6.20 (t, $J = 4.4$ Hz, 1H), 3.90 – 3.69 (m, 2H), 3.44 (d, $J = 17.0$ Hz, 1H), 3.25 – 3.19 (m, 2H), 3.04 (d, $J = 17.0$ Hz, 1H), 1.38 (s, 9H). The spectral data of **9j** was consistent with that reported in the literature.¹



Following the general procedure described above, **9k** was obtained by flash silica chromatography (petroleum ether/EtOAc = 8/1) as a white solid (60 mg, 59% yield). The compound was determined to be 90% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; $t(\text{major}) = 30.4$ min, $t(\text{minor}) = 41.2$ min. **¹H NMR** (400 MHz, CDCl₃) δ 8.31 (d, $J = 8.6$ Hz, 2H), 8.08 (d, $J = 8.5$ Hz, 2H), 7.40 – 7.18 (m, 2H), 7.14 (s, 1H), 6.25 (s, 1H), 3.89 – 3.68 (m, 5H), 3.39 (d, $J = 16.8$ Hz, 1H), 3.30 – 3.16 (m, 2H), 2.98 (d, $J = 16.9$ Hz, 1H), 1.37 (s, 9H). The spectral data of **9k** was consistent with that reported in the literature.¹



Following the general procedure described above, **9l** was obtained by flash silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (64 mg, 61% yield). The compound was determined to be 97% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 88/12, 1.0 mL/min, 254 nm, 39 bar, 25 °C; $t(\text{major}) = 47.6$ min, $t(\text{minor}) = 54.5$ min. $[\alpha]_{\text{D}}^{25} = 21.4$ ($c = 2.21$, CHCl₃); Reported data from Lit.¹: $[\alpha]_{\text{D}}^{25} = 19.1$ ($c = 2.25$, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.97 (d, $J = 8.3$ Hz, 1H), 8.29 (d, $J = 8.8$ Hz, 2H), 8.15 – 8.05 (m, 3H), 7.93 (d, $J = 8.1$ Hz, 1H), 7.72 (t, $J = 7.6$ Hz, 1H), 7.61 (t, $J = 7.5$ Hz, 1H), 7.48 (d, $J = 8.4$ Hz, 1H), 6.27 (s, 1H), 3.95 – 3.73 (m, 2H), 3.59 (d, $J = 17.4$ Hz, 1H), 3.31 – 3.11 (m, 3H), 1.39 (s, 9H). The spectral data of **9l** was consistent with that reported in the literature.¹

8 General procedure for the catalytic asymmetric Mannich reactions

8.1 Screening of the reaction conditions

The following catalysts **S16-S19** and **8** were known compounds and were prepared from literature.¹

Table S4. Catalyst screening for reaction of 7a and 1a^a

S16: R² = Ph, Ar = Ph
S17: R² = pivaloyl, Ar = Ph
S18: R² = pivaloyl, Ar = 2-naphthyl
S19: R² = pivaloyl, Ar = anthryl
8: Ar = 4-^tBu-Ph

entry	T (°C)	t (h)	Cat.	base	yield of 10a (%) ^b	ee of 10a (%) ^c
1	rt	96	S16	20% aq Na ₂ CO ₃	25	8
2	rt	96	S17	20% aq Na ₂ CO ₃	20	15
3	rt	96	S18	20% aq Na ₂ CO ₃	22	26
4	rt	96	S19	20% aq Na ₂ CO ₃	29	33
5	rt	96	8	20% aq Na ₂ CO ₃	45	42

^aReaction conditions: **1a** (0.1 mmol), **7a** (0.12 mmol), catalyst (10 mol%) and base (0.3 mmol) indicated temperature.

^bIsolated yield. ^cDetermined by HPLC analysis.

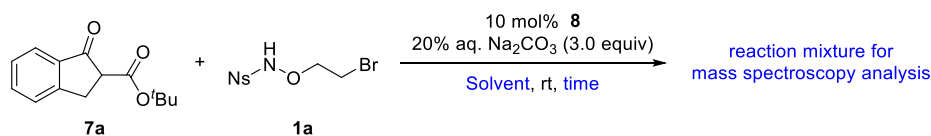
Table S5. Initial results for reaction of benzoyl acetate and 1a^a

entry	T (°C)	base	yield (%) ^b	ee (%) ^c
1	rt	Cs ₂ CO ₃ (50%)	0	0
2	rt	KOH (50%)	21	5

8, Ar = 4-^tBu-Ph

^aReaction conditions: **1a** (0.1 mmol), **S20** (0.12 mmol), catalyst (10 mol%) and base (0.3 mmol) indicated temperature. ^bIsolated yield. ^cDetermined by HPLC analysis.

8.2 Investigation of the solvent effect via mass and NMR spectroscopy



To a solution of *tert*-butyl indanone carboxylate **7a** (56 mg, 0.24 mmol, 1.2 equiv), hydroxylamine **1a** (65 mg, 0.2 mmol, 1.0 equiv), and catalyst **8** (16 mg,

0.02mmol, 0.1 equiv) in toluene or toluene/^tBuOMe (v/v = 5/1) (2 mL) was added 20% aq. Na₂CO₃ (318 uL, 0.6 mmol, 3.0 equiv) at room temperature. After stirring at room temperature for 12 h, 48 h, and 64 h, an aliquot of reaction solution was taken out from toluene or toluene/^tBuOMe solvents. Then, the mixture was subjected to mass spectroscopy analysis. The recorded ESI-MS data and assigned structures of possible compounds were demonstrated in following **Figures S5-6**.

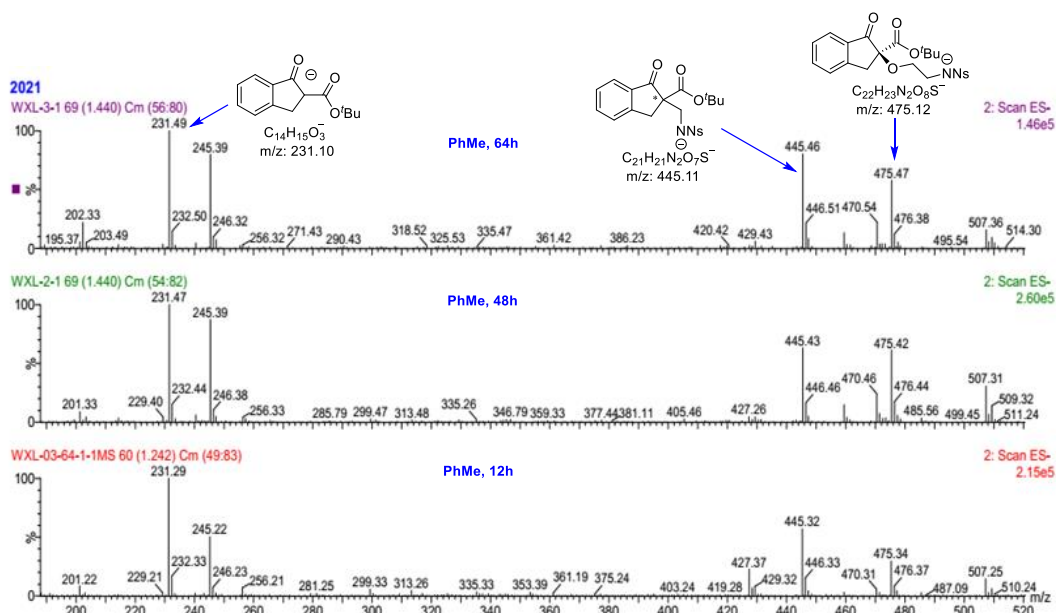


Figure S5. ESI-MS data of the reaction mixture (toluene as solvent)

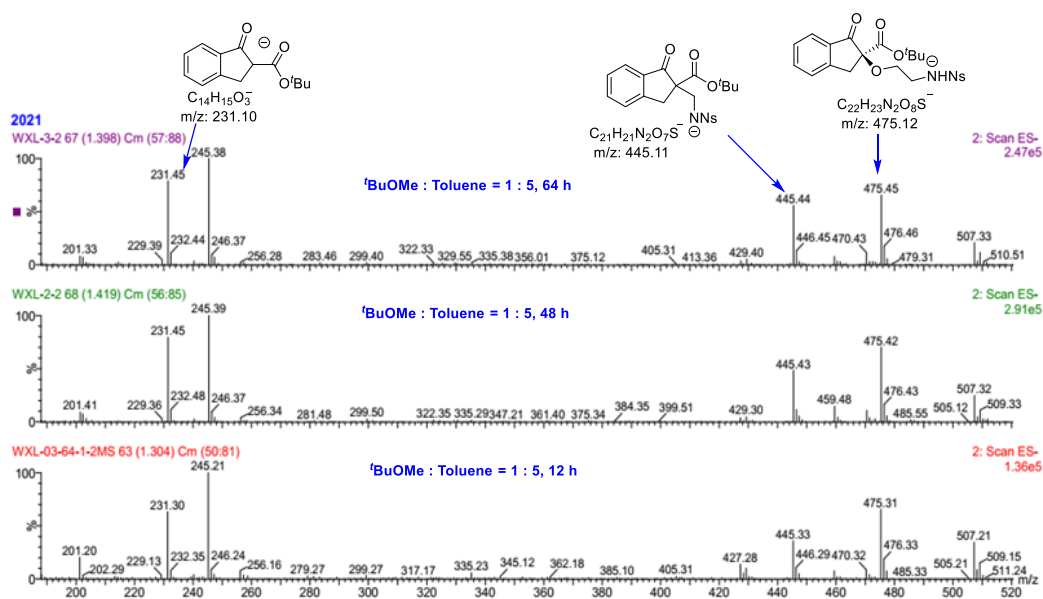
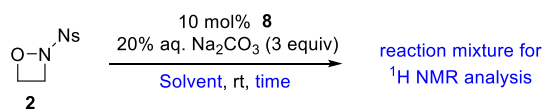


Figure S6. ESI-MS data of the reaction mixture (toluene/^tBuOMe as solvents)



To a solution of 1,2-oxazetidine **2** (12 mg, 0.05 mmol, 1.0 equiv) and catalyst **8** (4 mg, 0.005 mmol, 0.1 equiv) in toluene or toluene/ t BuOMe (v/v = 5/1) (0.5 mL) was added 20% aq. Na_2CO_3 (80 μL , 0.15 mmol, 3.0 equiv), the reaction mixture was stirred at room temperature. After stirring at room temperature for 12 h, 48 h, and 72 h, the crude reaction mixture was subjected to $^1\text{H NMR}$ analysis (mesitylene was used as internal standard). The $^1\text{H NMR}$ spectra were shown in following **Figures S7-14**.

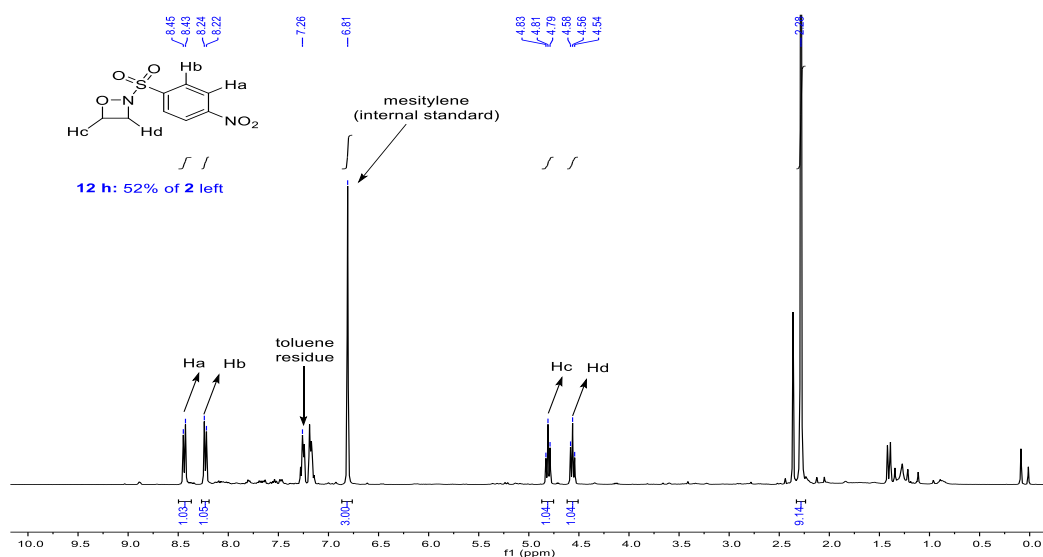


Figure S7. Crude $^1\text{H NMR}$ of compound 2 (toluene as solvent, 12 h)

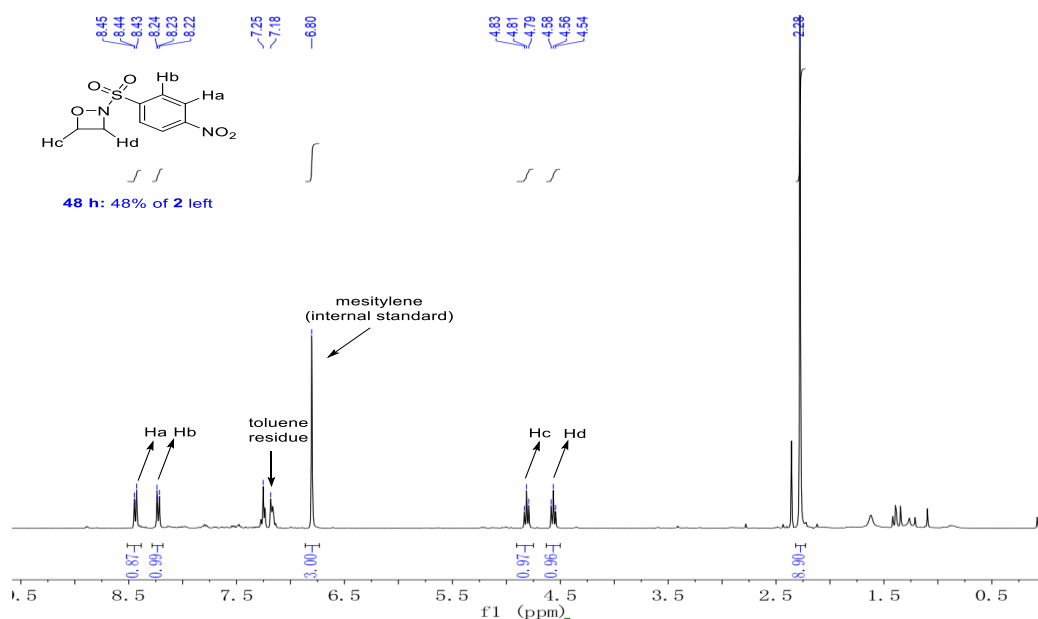


Figure S8. Crude $^1\text{H NMR}$ of compound 2 (toluene as solvent, 48 h)

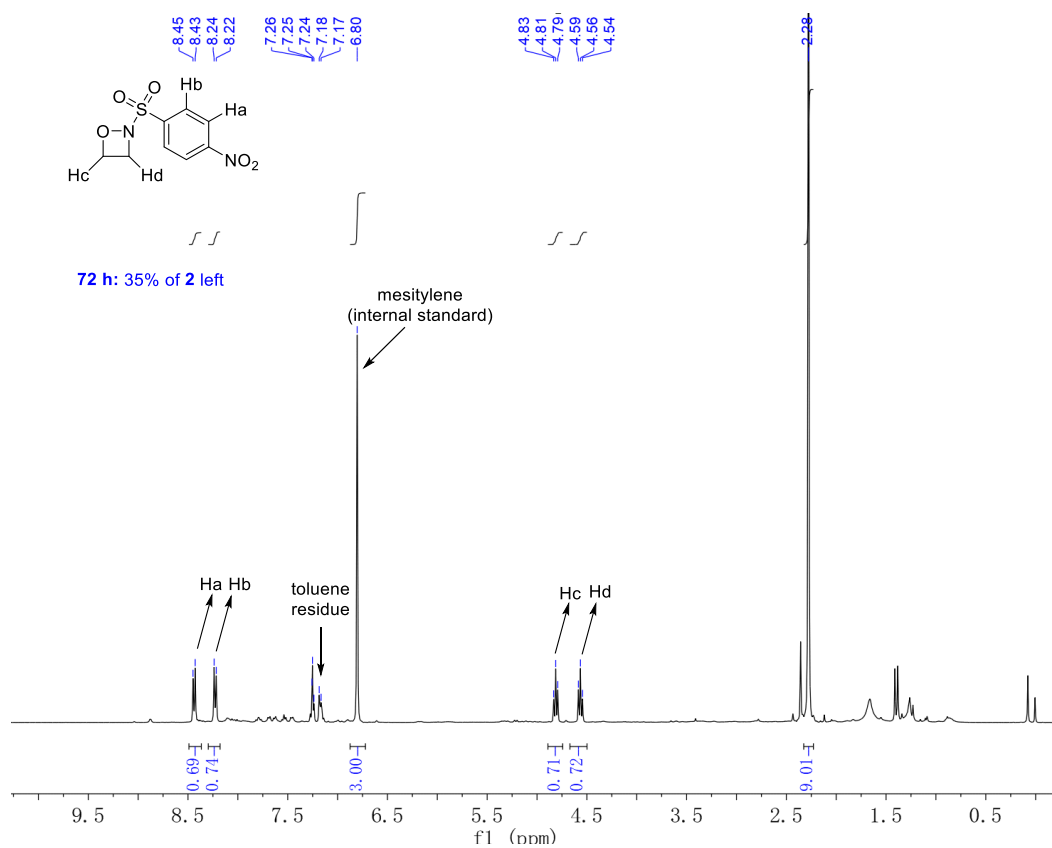


Figure S9. Crude ¹H NMR of compound 2 (toluene as solvent, 72 h)

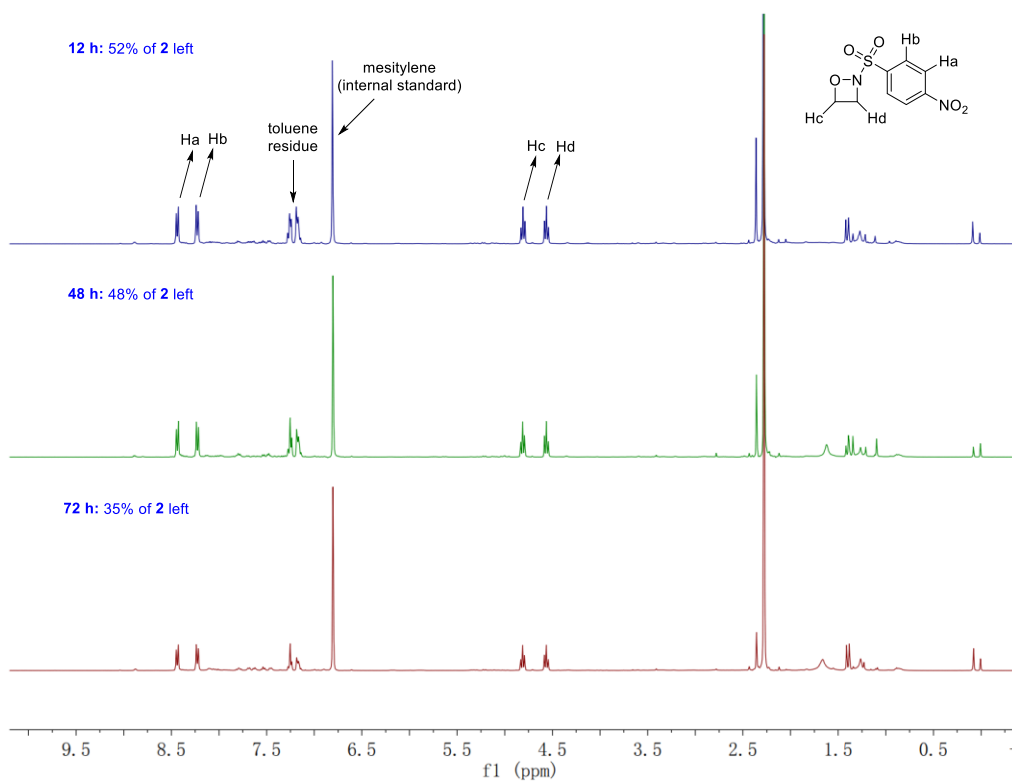


Figure S10. Comparison of the crude ¹H NMR (toluene as solvent)

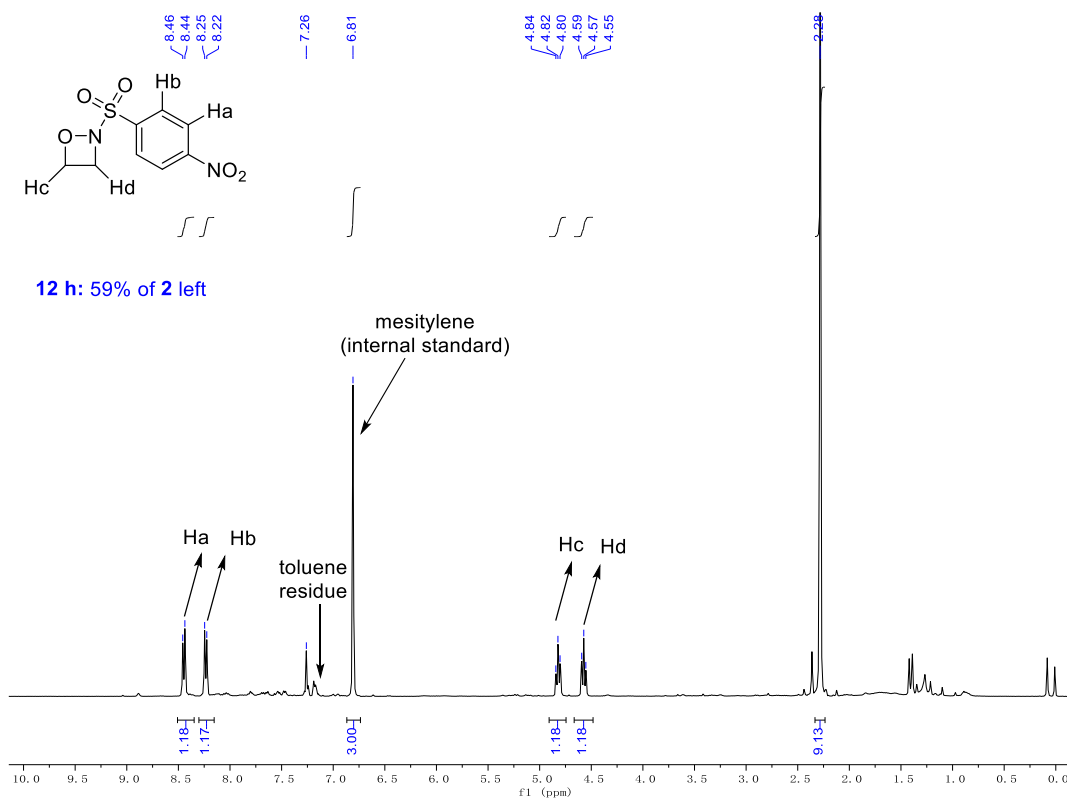


Figure S11. Crude ^1H NMR of compound **2** (t BuOMe/Toluene as solvents, 12 h)

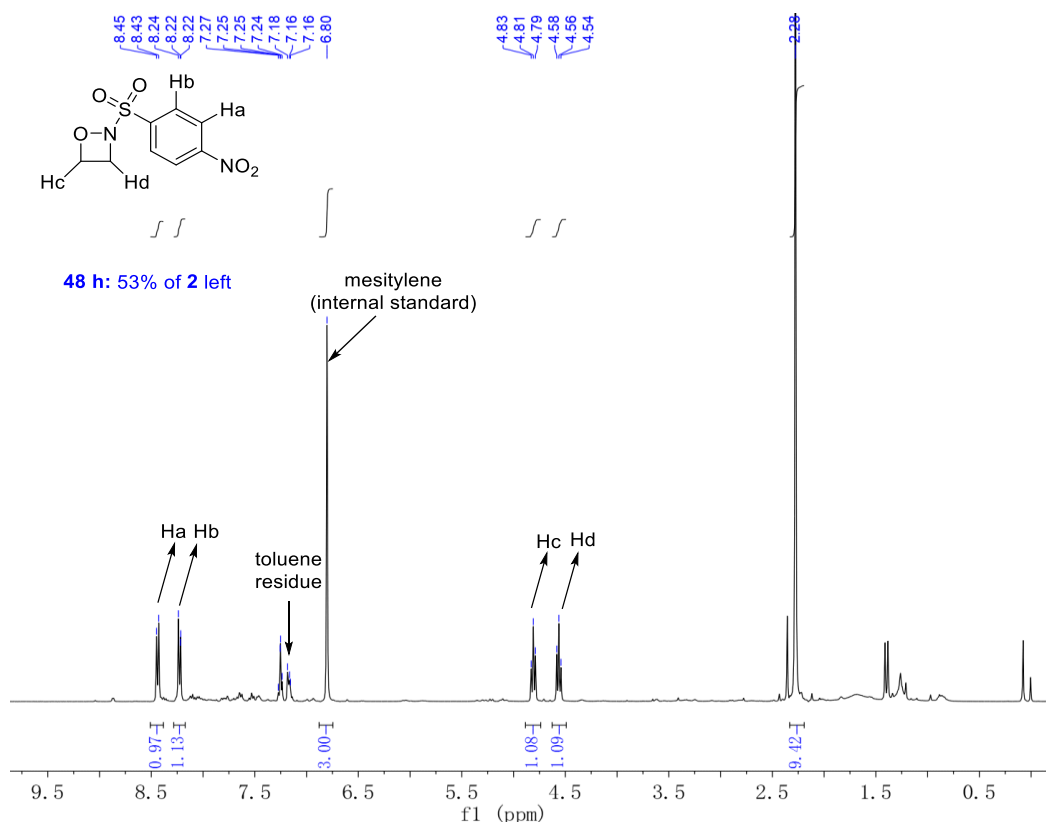


Figure S12. Crude ^1H NMR of compound **2** (t BuOMe/Toluene as solvents, 48 h)

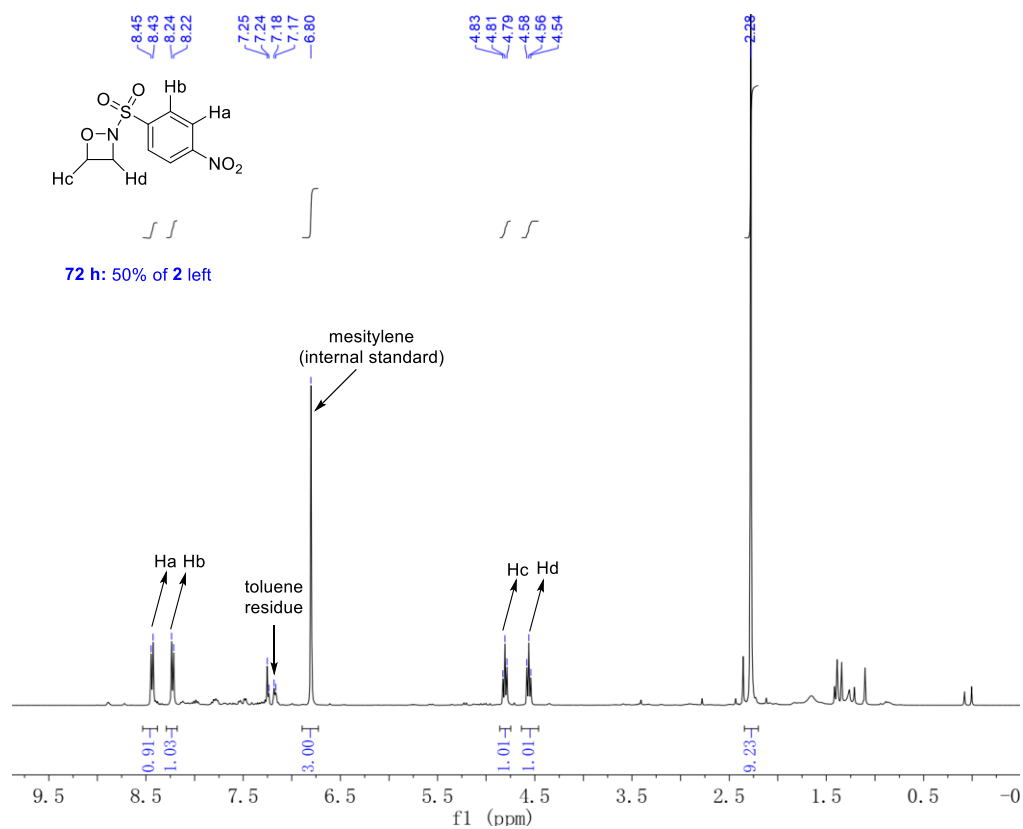


Figure S13. Crude ¹H NMR of compound 2 (^tBuOMe/Toluene as solvents, 72 h)

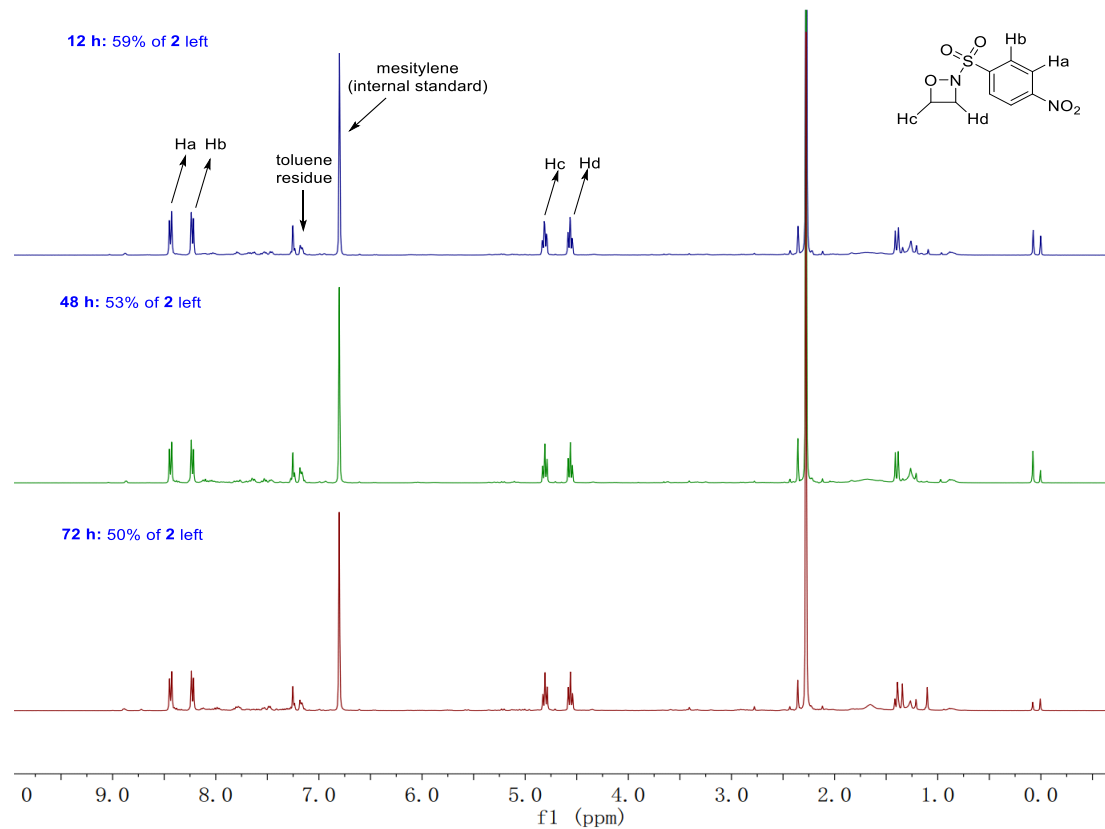
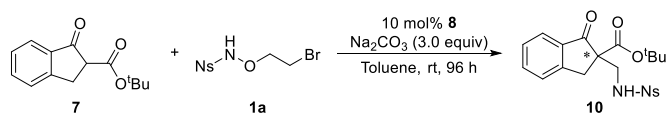
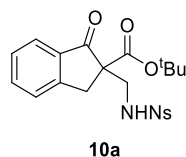


Figure S14. Comparison of the crude ¹H NMR (^tBuOMe/Toluene as solvents)

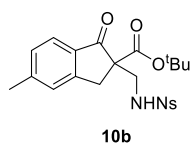
8.3 General procedure for catalytic asymmetric Mannich reactions



To a solution of *tert*-butyl indanone carboxylate substrates **7** (0.24 mmol, 1.2 equiv), hydroxylamine **1a** (0.2 mmol, 1.0 equiv), and catalyst **8** (16 mg, 0.02mmol, 0.1 equiv) in toluene (2 mL) was added 20% aq. Na₂CO₃ (318 uL, 0.6 mmol, 3.0 equiv) at room temperature. The reaction mixture was stirred at the same temperature for 96 h. After completion, the solution was quenched with saturated NH₄Cl (2 mL) and extracted with EtOAc (3 mL x 3), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 10/1 ~ 5/1) to afford the products **10a-d** in 40-45% yield and 42-56% ee.

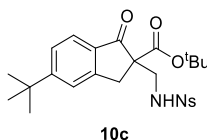


Following the general procedure described above, **10a** was obtained by flash silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (40 mg, 45% yield). The compound was determined to be 42% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak AD-H, Hexane/*i*-PrOH = 70/30, 1.0 mL/min, 254 nm, 55 bar, 25 °C; t(major) = 13.9 min, t(minor) = 26.8 min. $[\alpha]_D^{25} = -12.6$ (c = 0.3, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.36 (d, *J* = 8.5 Hz, 2H), 8.00 (d, *J* = 8.5 Hz, 2H), 7.77 – 7.63 (m, 2H), 7.53 (d, *J* = 7.5 Hz, 1H), 7.42 (t, *J* = 7.3 Hz, 1H), 5.67 (dd, *J* = 9.3, 4.4 Hz, 1H), 3.45 (d, *J* = 19.2 Hz, 3H), 3.22 (dd, *J* = 12.8, 9.1 Hz, 1H), 1.33 (s, 9H). **¹³C NMR** (100 MHz, CDCl₃) δ 200.8, 170.2, 153.1, 150.1, 145.5, 135.9, 134.6, 128.2, 128.1, 126.7, 124.9, 124.5, 83.4, 60.6, 46.5, 35.9, 27.7. **HRMS (ESI)** m/z: [M + Na]⁺ Calcd for C₂₁H₂₂N₂NaO₇S 469.1040; Found 469.1045. **Melting point:** 153.2-154.5 °C.

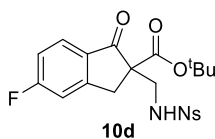


Following the general procedure described above, **10b** was obtained by flash

silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (40 mg, 43% yield). The compound was determined to be 56% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; *t*(major) = 19.8 min, *t*(minor) = 23.5 min. $[\alpha]_D^{25} = -27.4$ (*c* = 1.25, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.8 Hz, 2H), 8.00 (d, *J* = 8.8 Hz, 2H), 7.62 (d, *J* = 7.9 Hz, 1H), 7.31 (s, 1H), 7.23 (d, *J* = 7.8 Hz, 1H), 5.64 (dd, *J* = 8.9, 4.1 Hz, 1H), 3.41 (s, 3H), 3.21 (dd, *J* = 12.7, 9.2 Hz, 1H), 2.47 (s, 3H), 1.33 (s, 9H). **¹³C NMR** (100 MHz, CDCl₃) δ 200.2, 170.3, 153.6, 150.1, 147.5, 145.6, 132.4, 129.4, 128.2, 127.0, 124.7, 124.5, 83.3, 60.7, 46.6, 35.8, 27.7, 22.2. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₂H₂₄N₂NaO₇S 483.1209; Found 483.1196. **Melting point:** 161.7-162.6 °C.



Following the general procedure described above, **10c** was obtained by flash silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (45 mg, 45% yield). The compound was determined to be 48% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; *t*(major) = 11.7 min, *t*(minor) = 15.8 min. $[\alpha]_D^{25} = -21.6$ (*c* = 1.15, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.35 (d, *J* = 8.6 Hz, 2H), 8.00 (d, *J* = 8.6 Hz, 2H), 7.65 (d, *J* = 8.1 Hz, 1H), 7.54 – 7.40 (m, 2H), 5.68 (dd, *J* = 8.6, 3.9 Hz, 1H), 3.50 – 3.32 (m, 3H), 3.22 (dd, *J* = 12.5, 9.3 Hz, 1H), 1.37 (s, 9H), 1.34 (s, 9H). **¹³C NMR** (100 MHz, CDCl₃) δ 200.3, 170.3, 160.5, 153.5, 150.1, 145.5, 132.2, 128.2, 126.0, 124.5, 123.2, 83.3, 60.8, 46.6, 36.0, 35.7, 31.1, 27.7. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₅H₃₀N₂NaO₇S 525.1666; Found 525.1676. **Melting point:** 158.4-159.3 °C.

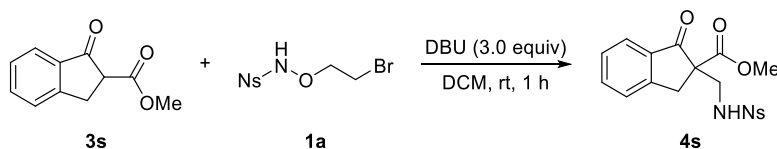


Following the general procedure described above, **10d** was obtained by flash

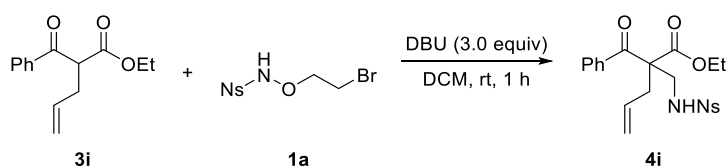
silica chromatography (petroleum ether/EtOAc = 10/1) as a white solid (37mg, 40% yield). The compound was determined to be 52% ee by HPLC analysis. **HPLC conditions:** Daicel Chiralpak OD-H, Hexane/*i*-PrOH = 85/15, 1.0 mL/min, 254 nm, 41 bar, 25 °C; *t*(major) = 21.3 min, *t*(minor) = 25.8 min. $[\alpha]_D^{25} = -19.1$ (*c* = 2.12, CHCl₃). **¹H NMR** (400 MHz, CDCl₃) δ 8.36 (d, *J* = 8.7 Hz, 2H), 8.01 (d, *J* = 8.8 Hz, 2H), 7.75 (dd, *J* = 8.5, 5.2 Hz, 1H), 7.23 – 7.05 (m, 2H), 5.62 (dd, *J* = 8.9, 4.4 Hz, 1H), 3.55 – 3.38 (m, 3H), 3.22 (dd, *J* = 12.8, 9.1 Hz, 1H), 1.33 (s, 9H). **¹³C NMR** (100 MHz, CDCl₃) δ 198.9, 169.7, 167.8 (d, ¹*J*_{C-F} = 258.2 Hz), 156.2 (d, ³*J*_{C-F} = 10.4 Hz), 150.1, 145.4, 131.1, 128.2, 127.3 (d, ³*J*_{C-F} = 11.5 Hz), 124.5, 116.6 (d, ²*J*_{C-F} = 24.2 Hz), 113.5 (d, ²*J*_{C-F} = 23.0 Hz), 83.6, 61.0, 46.3, 35.7, 27.7. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₁H₂₁FN₂NaO₇S 487.0956; Found 487.0951. **Melting point:** 155.6-156.6 °C.

9. Gram-scale synthesis and synthetic applications

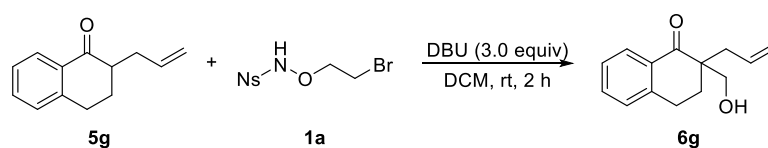
9.1 Gram-scale synthesis of compounds 4s, 4i, and 6b



To a solution of **3s** (1.9 g, 10 mmol, 2.0 equiv) and DBU (2.3 g, 15 mmol, 3.0 equiv) in DCM (25 mL) was added the solution of **1a** (1.6 g, 5 mmol, 1.0 equiv) in DCM (25 mL) at room temperature over 10 minutes. The reaction mixture was stirred at the same temperature for 1 h. After completion, the solution was quenched with saturated NH₄Cl (20 mL) and extracted with DCM (40 mL x 3). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 8/1) to afford the product **4s** as a white solid (1.83 g, 91% yield).

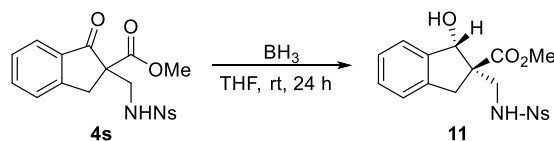


To a solution of **3i** (1.86 g, 8 mmol, 2.0 equiv) and DBU (1.81 g, 12 mmol, 3.0 equiv) in DCM (20 mL) was added the solution of **1a** (1.3 g, 4 mmol, 1.0 equiv) in DCM (20 mL) at room temperature over 10 minutes. The reaction mixture was stirred at the same temperature for 1 h. After completion, the solution was quenched with saturated NH₄Cl (20 mL) and extracted with DCM (30 mL x 3). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 8/1) to afford the product **4i** as a white solid (1.4 g, 78% yield).



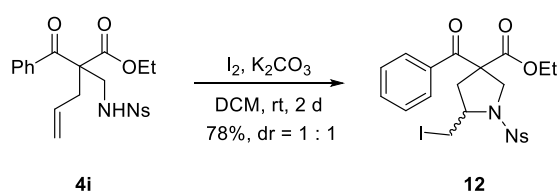
To a solution of α -substituted ketone **5g** (1.86 g, 10 mmol, 1.0 equiv) and DBU (4.54 g, 30 mmol, 3.0 equiv) in DCM (30 mL) was added the solution of **1a** (4.88 g, 15 mmol, 1.5 equiv) in DCM (30 mL) at room temperature over 10 min. The reaction mixture was stirred at the same temperature for 2 h. After completion, the solution was quenched with saturated NH₄Cl (20 mL) and extracted with DCM (30 mL x 3). The combined organic layers were dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 8/1) to afford the product **6g** as a colorless oil (1.13 g, 52% yield).

9.2 Synthetic applications



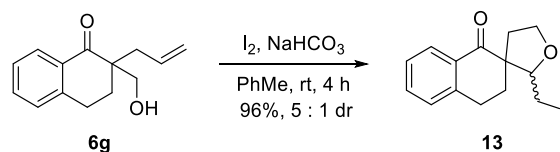
To a solution of **4s** (81 mg, 0.2 mmol, 1.0 equiv) in anhydrous THF (2 mL) was added dropwise of borane (1.0 M solution in THF, 0.4 mL, 0.4 mmol, 2.0 equiv) at room temperature under an N₂ atmosphere. After stirring for 24 h at the same temperature, the mixture was quenched with EtOH (2 mL) and concentrated. The residue was diluted with H₂O (3 mL) and extracted with EtOAc (5 mL x 3). The combined organic layers were dried over Na₂SO₄ and concentrated in vacuum. The

residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 5/1) to afford the product **11** as a white solid (50 mg, 61% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.31 (d, *J* = 8.6 Hz, 2H), 7.97 (d, *J* = 8.6 Hz, 2H), 7.39 – 7.30 (m, 1H), 7.30 – 7.21 (m, 2H), 7.20 – 7.08 (m, 1H), 5.82 – 5.73 (m, 1H), 5.72 – 5.65 (m, 1H), 3.73 (s, 3H), 3.42 – 3.28 (m, 2H), 3.23 (d, *J* = 16.4 Hz, 1H), 3.13 – 2.97 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 174.8, 149.9, 145.6, 141.3, 138.2, 128.9, 128.2, 127.5, 124.7, 124.4, 124.2, 79.2, 58.3, 52.8, 45.7, 38.1. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₁₈H₁₈N₂NaO₇S 429.0727; Found 429.0730. **Melting point:** 110.3-112.1 °C.



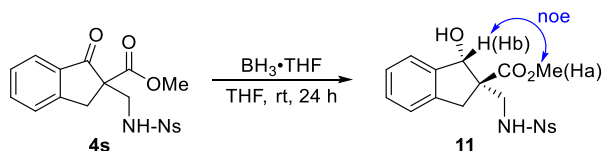
K₂CO₃ (83 mg, 0.6 mmol, 3.0 equiv) and I₂ (152 mg, 0.6 mmol, 3.0 equiv) were added to a solution of the **4i** (89 mg, 0.2 mmol, 1.0 equiv) in DCM (4 mL) at room temperature. The reaction mixture was stirred at the same temperature for 2 days. Then, a saturated aqueous solution of Na₂S₂O₃ (5 mL) was added, the phases were separated and the aqueous phase was extracted with DCM (5 mL x 3). The organic phases were combined, washed with brine (3mL), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 8/1) to afford the product **12** as a white solid (89 mg, 78% yield, dr = 1:1). ¹H NMR (400 MHz, CDCl₃) δ 8.44 – 8.34 (m, 1H), 8.11 – 8.01 (m, 2H), 7.81 – 7.71 (m, 2H), 7.63 – 7.52 (m, 1H), 7.52 – 7.39 (m, 2H), 7.40 – 7.30 (m, 1H), 4.40 (dd, *J* = 12.2, 2.5 Hz, 0.56H), 4.20 (dd, *J* = 12.0, 2.4 Hz, 0.36H), 4.17 – 4.00 (m, 2.56H), 3.97 – 3.79 (m, 1.39H), 3.73 – 3.63 (m, 0.57H), 3.61 – 3.47 (m, 1H), 3.46 – 3.36 (m, 0.38H), 3.06 – 2.83 (m, 1H), 2.37 (ddd, *J* = 13.6, 7.8, 2.7 Hz, 0.4H), 2.22 (ddd, *J* = 13.5, 8.0, 2.6 Hz, 0.56H), 1.03 – 0.88 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 192.2, 191.8, 170.8, 170.0, 150.2, 150.1, 143.9, 142.4, 134.3, 134.1, 133.9, 132.4, 128.9, 128.7, 128.7, 128.6, 128.5, 128.4, 124.4, 124.1, 62.5, 62.5, 62.1, 61.3, 60.5, 60.2, 54.8, 54.7, 40.6, 40.0, 13.6, 13.5, 11.6, 10.9. **HRMS (ESI)** *m/z*: [M + Na]⁺ Calcd for C₂₁H₂₁IN₂NaO₇S 595.0006; Found 595.0019. **Melting point:** 111.7-113.5

°C.



NaHCO₃ (50 mg, 0.6 mmol, 3.0 equiv) and I₂ (127 mg, 0.5 mmol, 2.5 equiv) were added to a solution of **6g** (43 mg, 0.2 mmol, 1.0 equiv) in PhMe (2 mL) at room temperature. The reaction mixture was stirred at the same temperature for 4 h. Then, a saturated aqueous solution of Na₂S₂O₃ (5 mL) was added, the phases were separated and the aqueous phase was extracted with DCM (5 mL x 3). The organic phases were combined, washed with brine (3mL), dried over Na₂SO₄ and concentrated in vacuum. The residue was purified by flash silica gel chromatography (petroleum ether/EtOAc = 15/1) to afford the product **13** as a brown oil (66 mg, 96% yield, dr = 5:1). ¹H NMR (400 MHz, CDCl₃) δ 8.04 (d, *J* = 7.7 Hz, 1H), 7.47 (t, *J* = 6.9 Hz, 1H), 7.31 (t, *J* = 7.1 Hz, 1H), 7.22 (d, *J* = 7.6 Hz, 1H), 4.37 – 4.26 (m, 0.18H), 4.25 – 4.10 (m, 1.94H), 3.93 (d, *J* = 8.8 Hz, 0.89H), 3.85 (d, *J* = 8.9 Hz, 0.18H), 3.42 – 3.25 (m, 2H), 3.07 – 2.91 (m, 2H), 2.57 (dd, *J* = 12.6, 6.4 Hz, 0.95H), 2.33 (dd, *J* = 12.9, 7.5 Hz, 0.15H), 2.26 – 2.19 (m, 2H), 2.09 (dd, *J* = 12.9, 7.1 Hz, 0.17H), 1.59 (dd, *J* = 12.6, 8.9 Hz, 0.9H). ¹³C NMR (100 MHz, CDCl₃) δ 199.1, 198.3, 143.0, 143.0, 133.6, 131.4, 131.0, 128.6, 128.0, 126.8, 79.5, 78.5, 75.2, 75.1, 54.6, 42.1, 40.7, 33.4, 32.4, 26.7, 26.1, 10.2, 8.5. HRMS (ESI) *m/z*: [M + Na]⁺ Calcd for C₁₄H₁₅INaO₂ 365.0009; Found: 365.0010.

10. Determination of the relative configuration of compound 11



NOE analysis was used to determine the relative configuration of compound **11**. As shown in **Figure S15**, when selecting to irradiate Ha (OMe) proton signal in compound **11**, a clear NOE effect between Hb and Ha (OMe) was observed. This result indicated the protons of Hb and Ha (CO₂Me) are on the same side of the ring,

namely, they are mutually *cis* to each other.

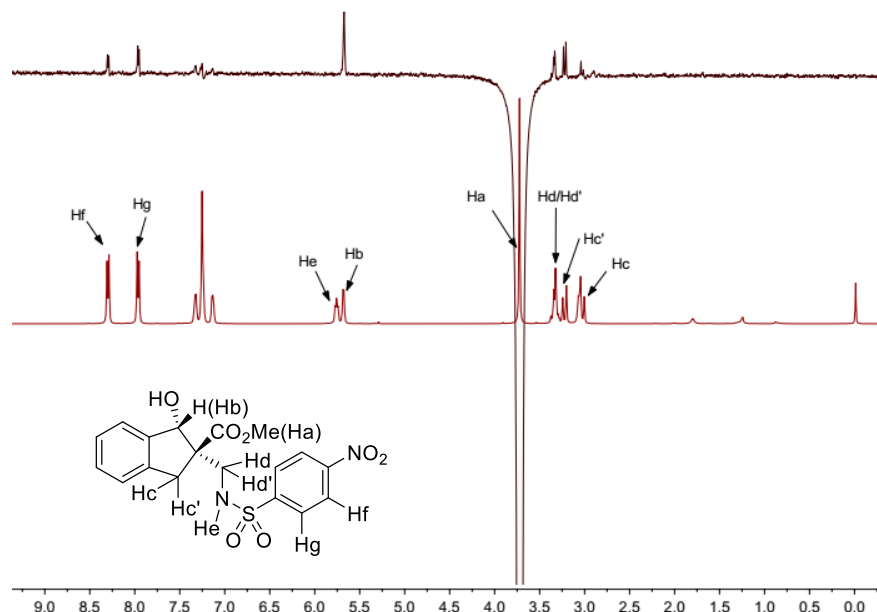


Figure S15. Comparison of the ¹H NMR and NOE spectra of compound 11

11. References

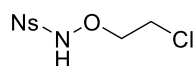
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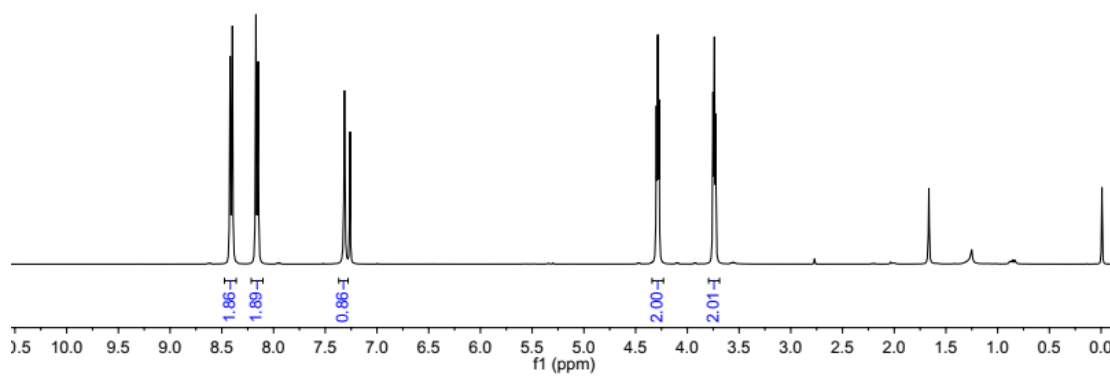
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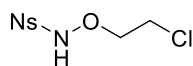
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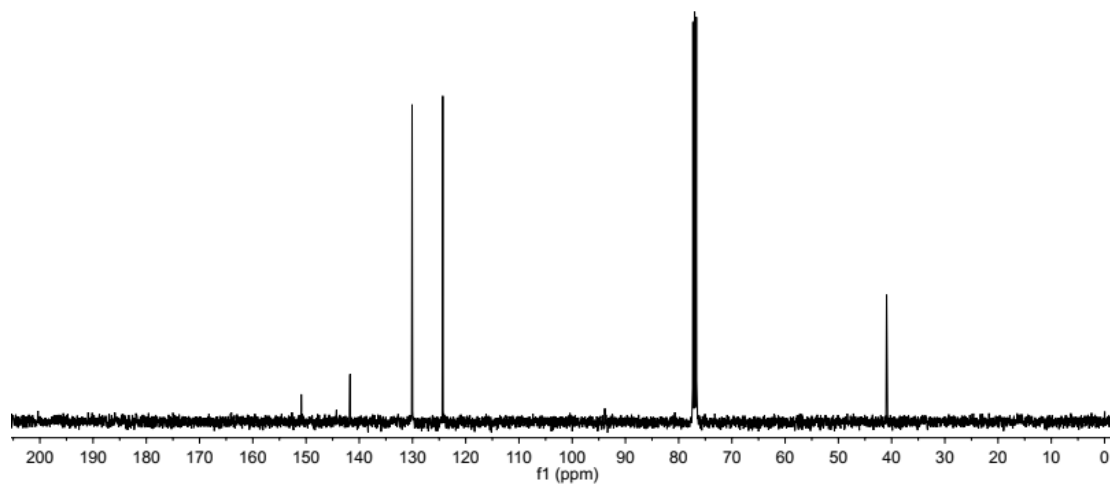
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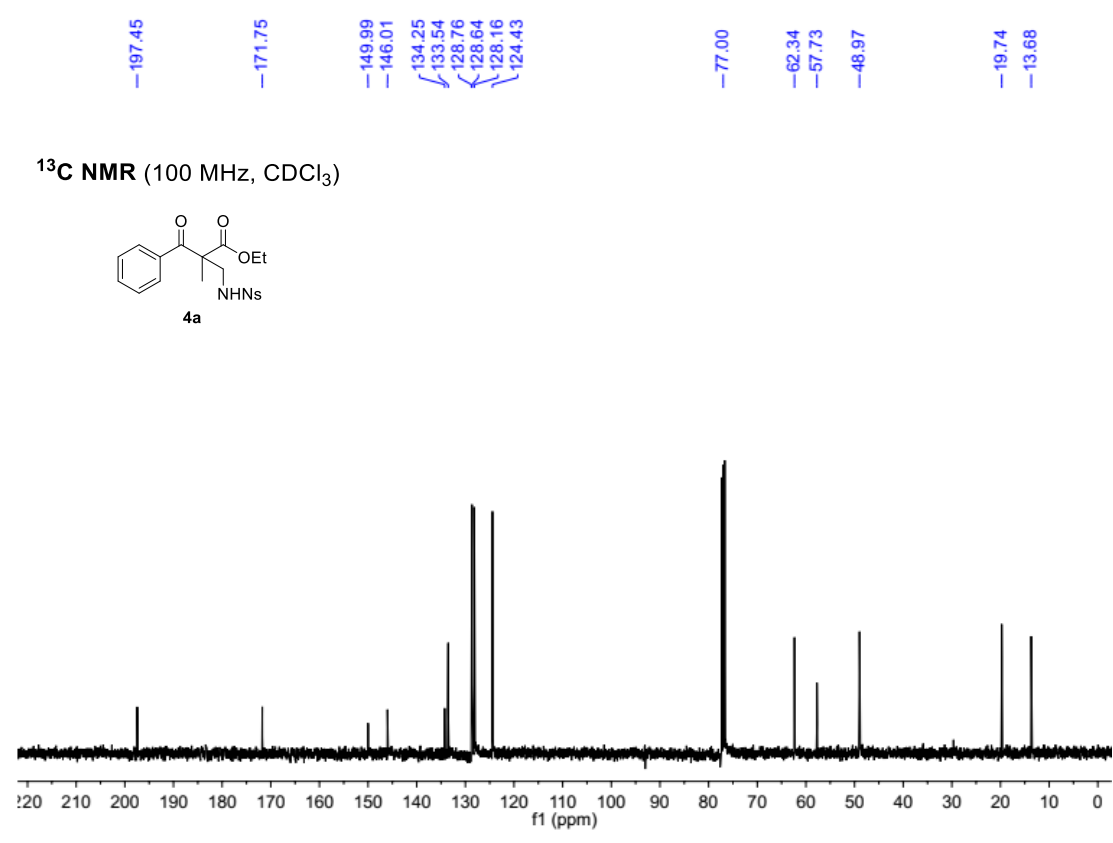
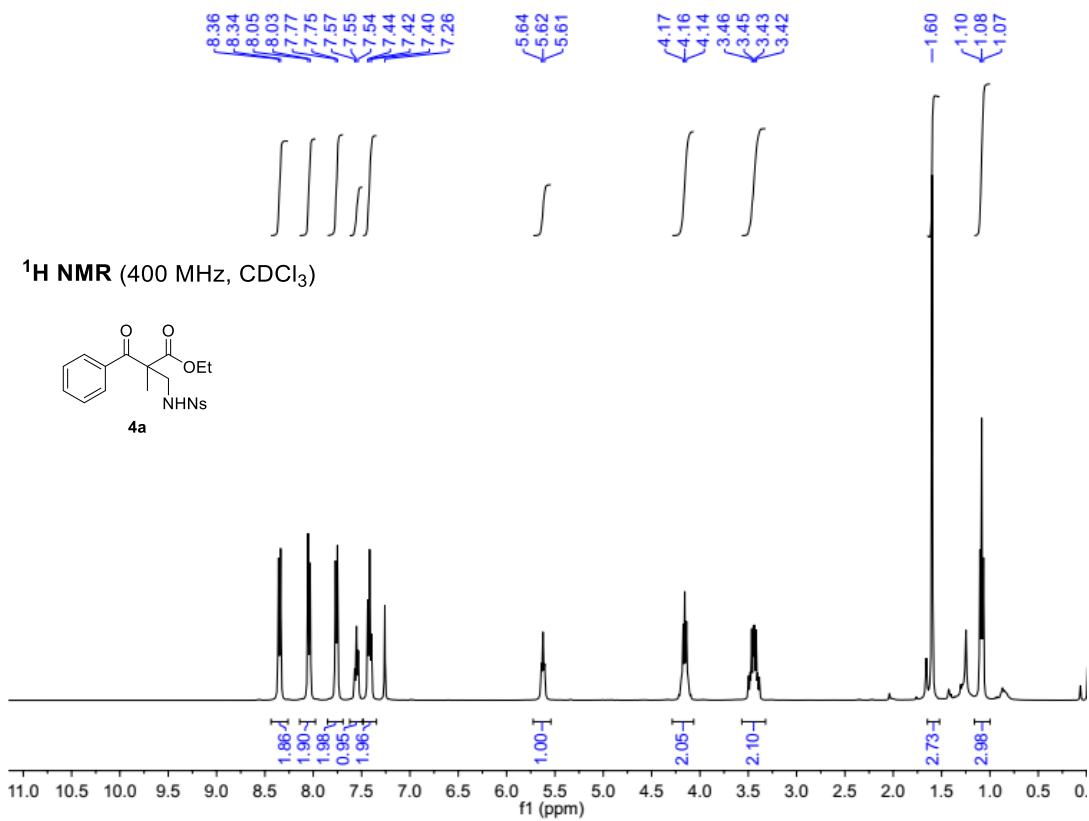


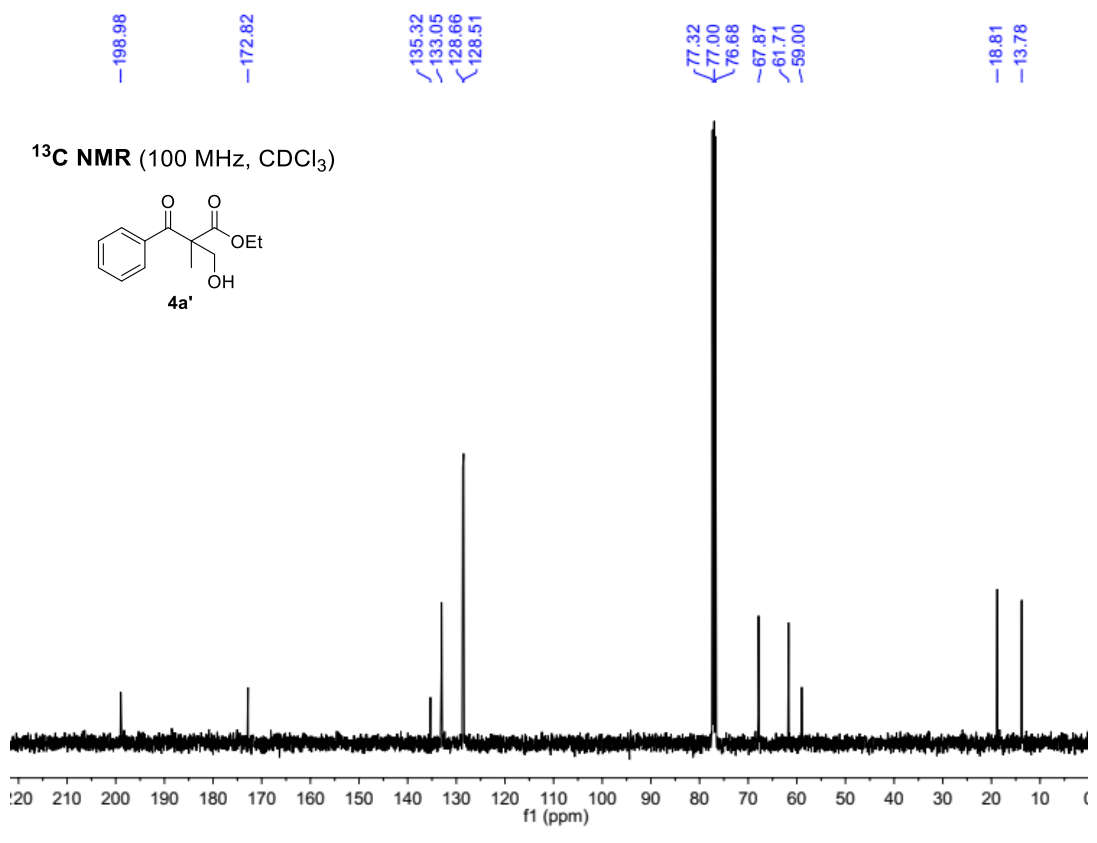
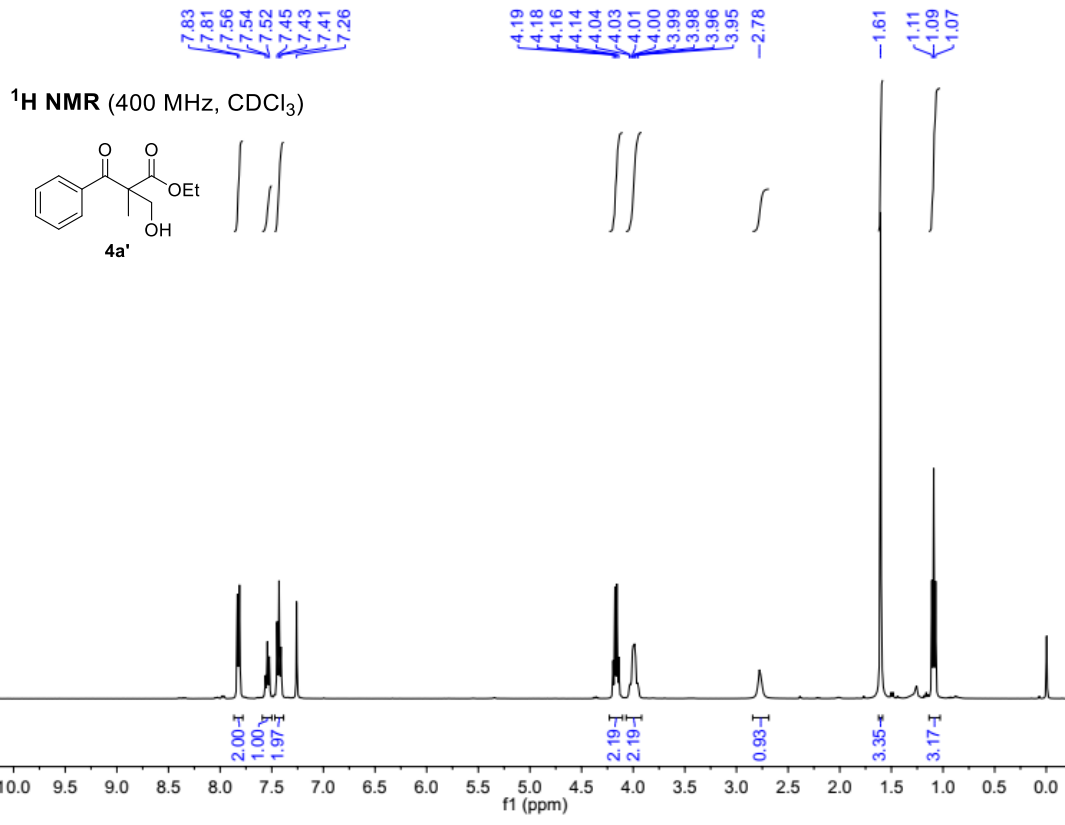
^{13}C NMR (100 MHz, CDCl_3)

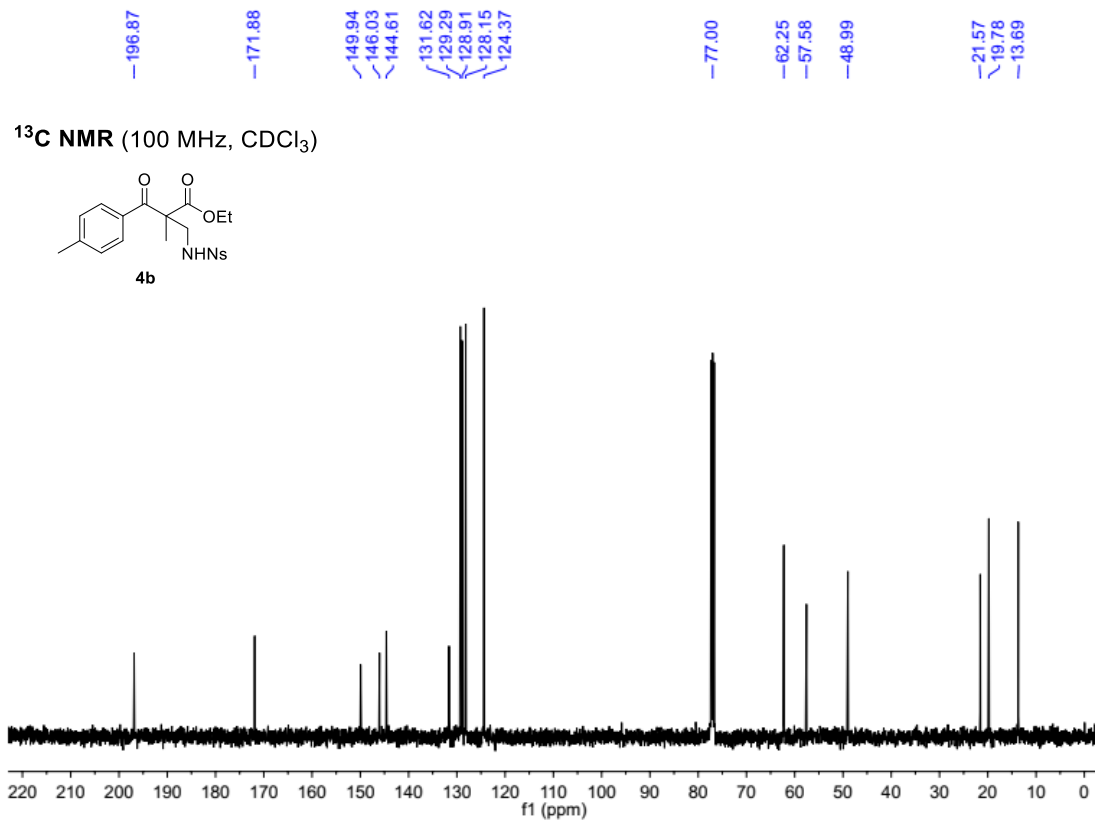
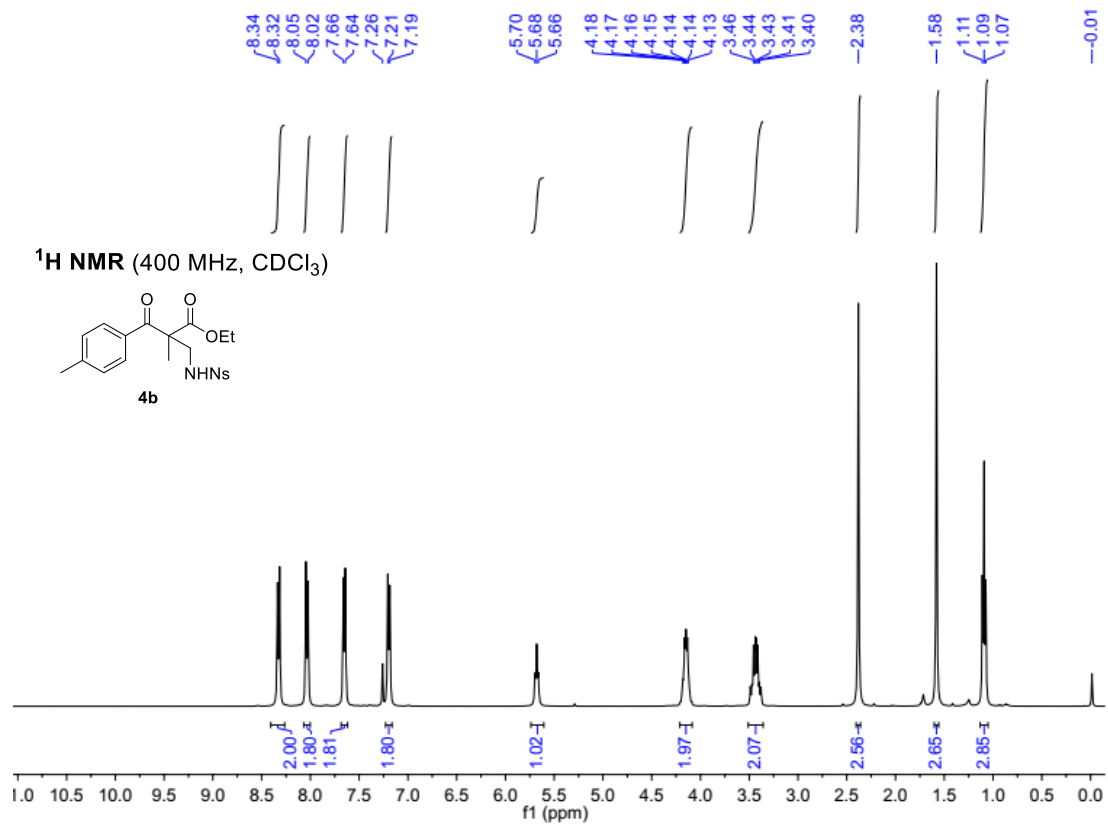


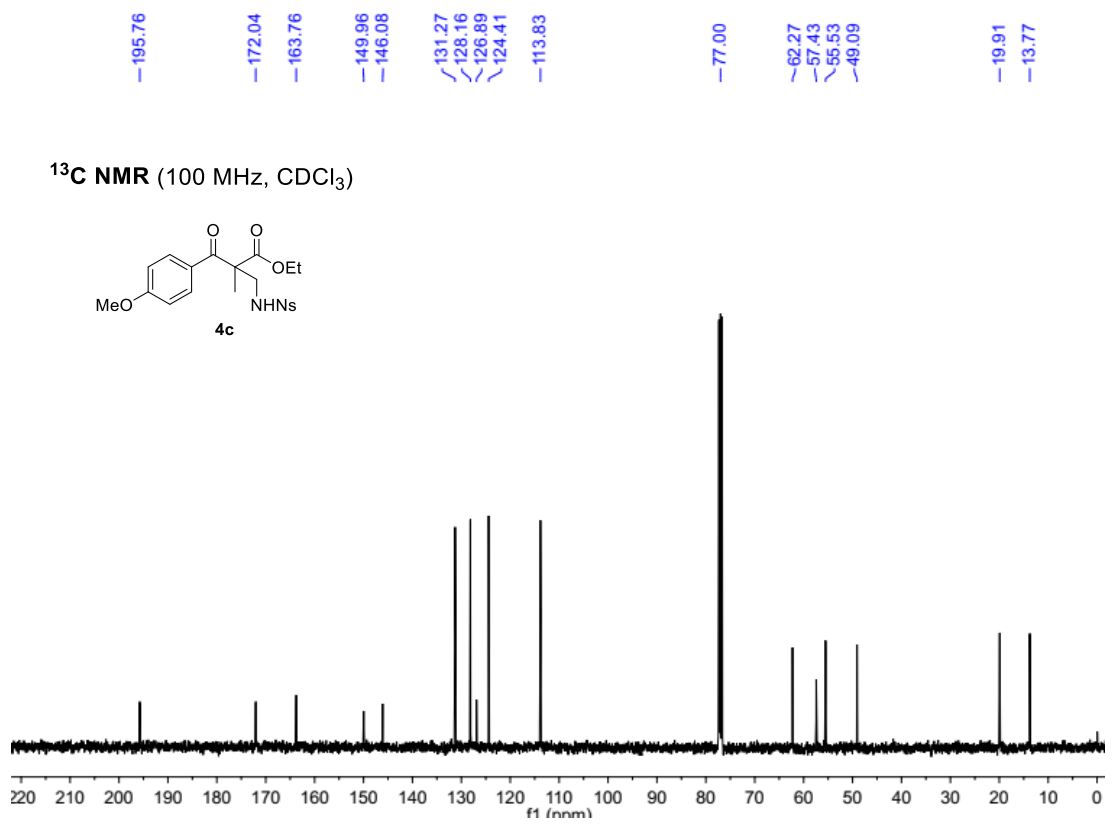
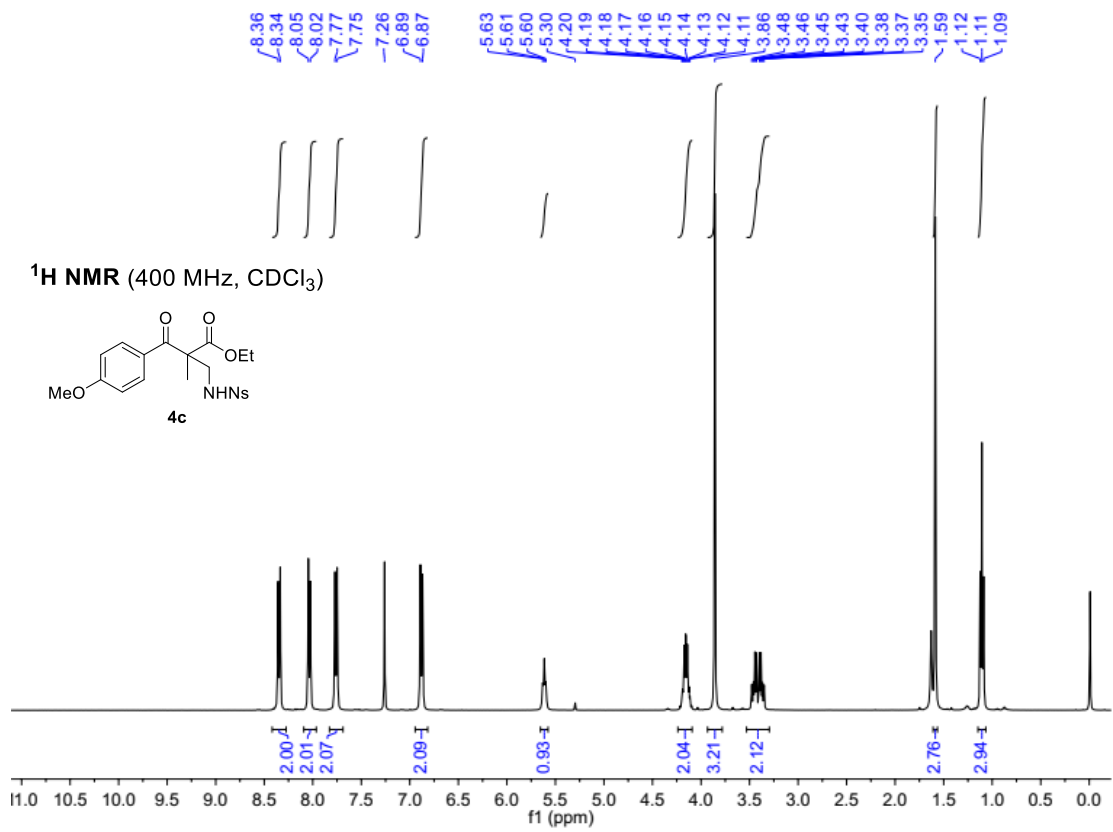
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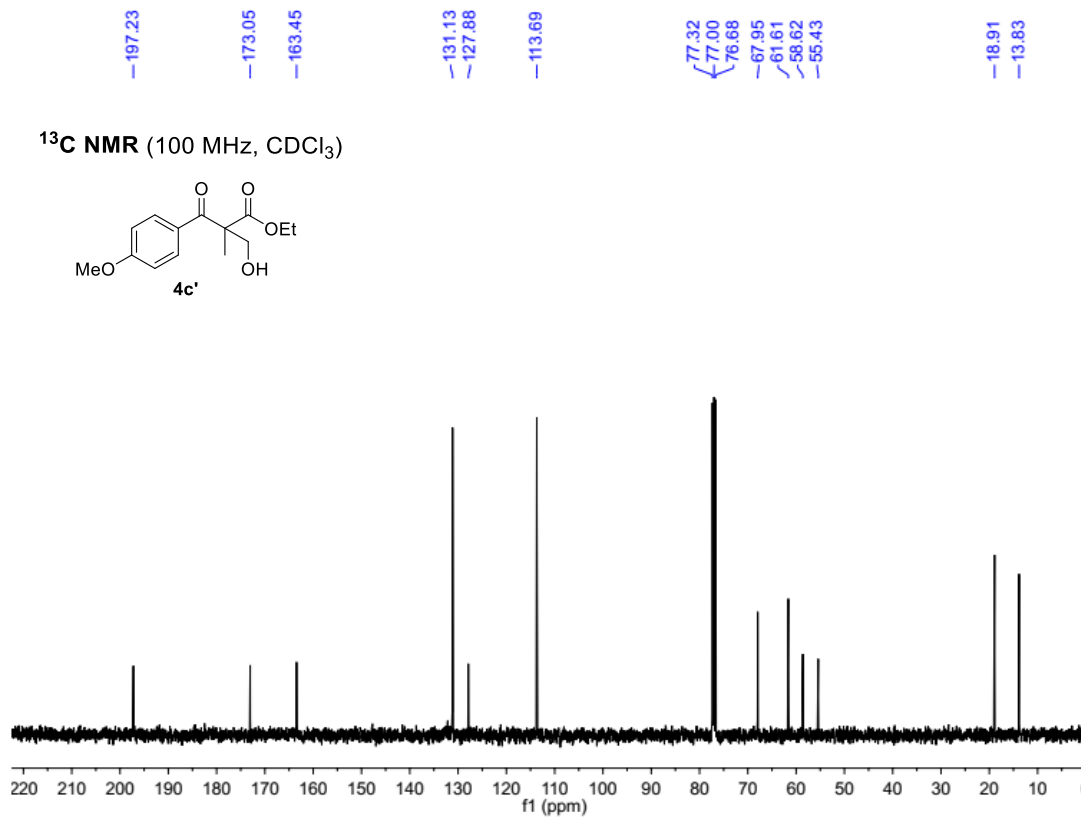
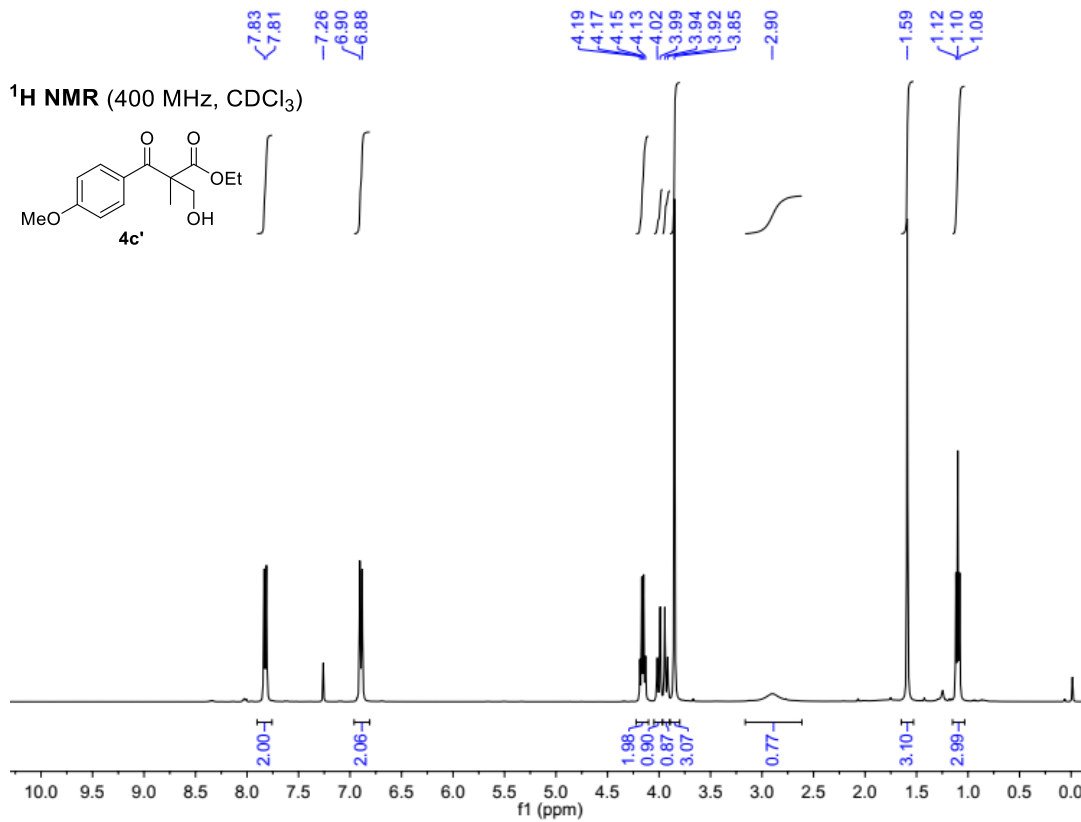


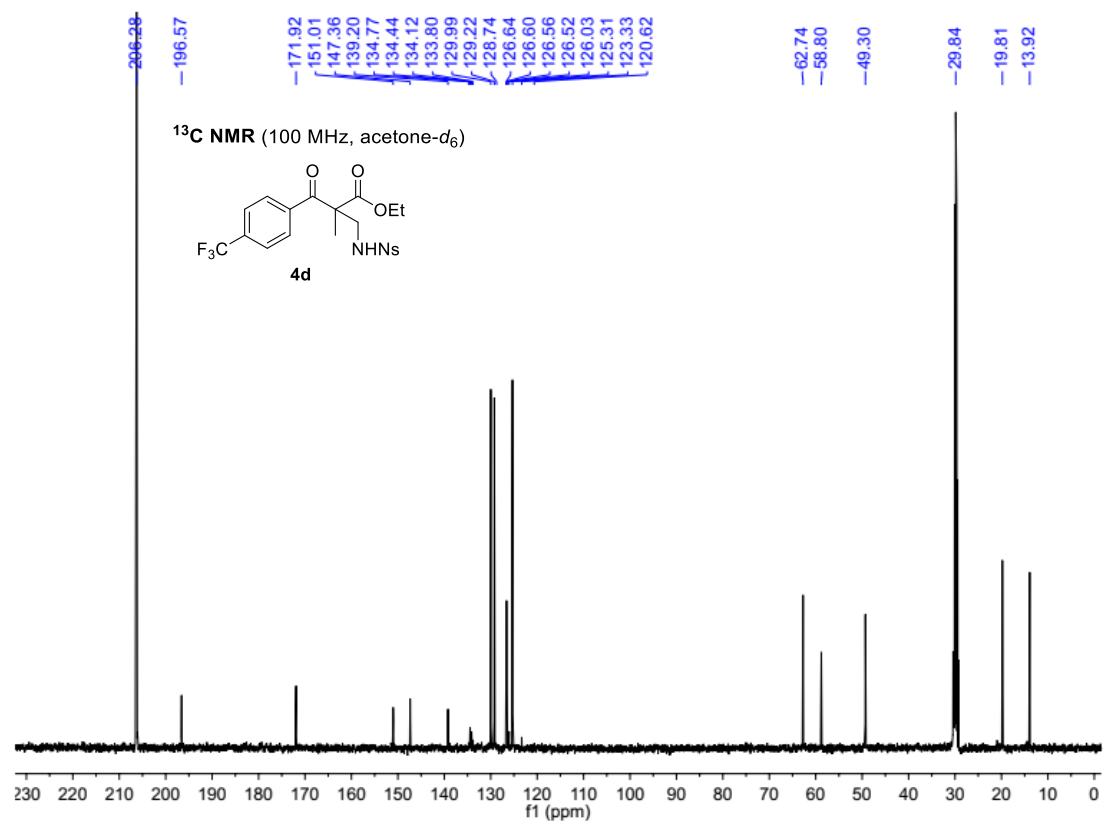
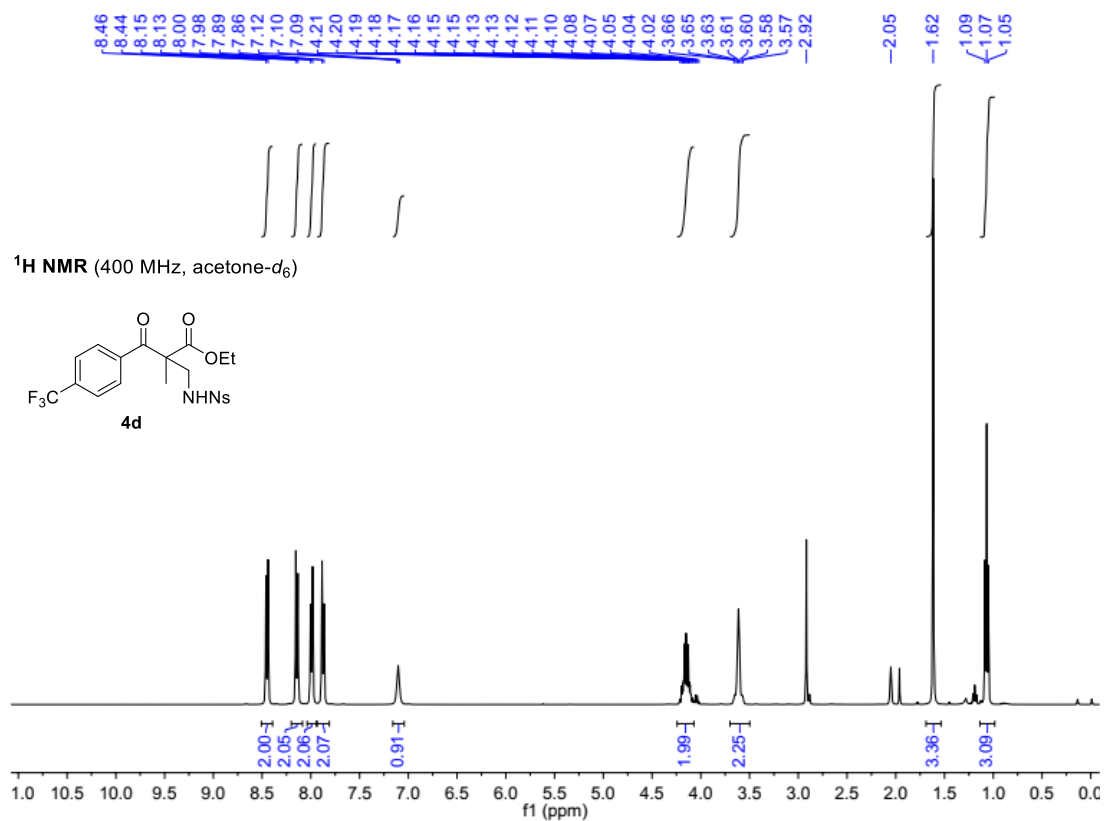


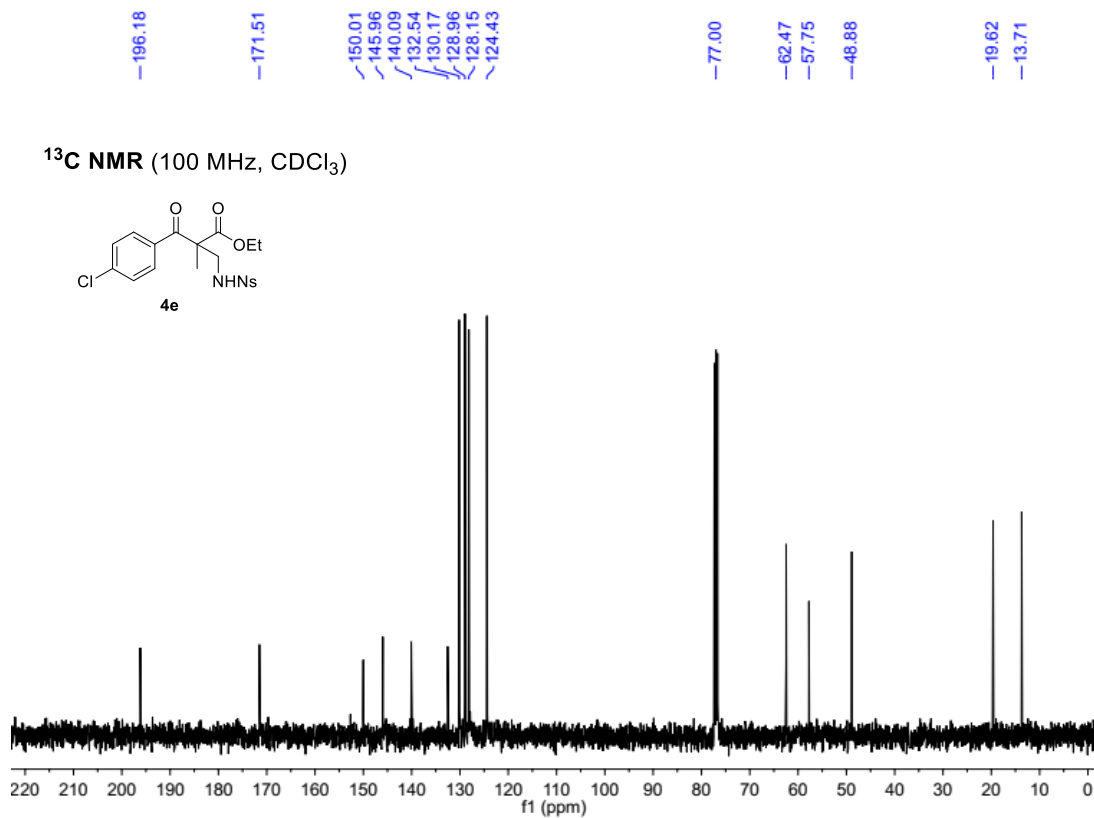
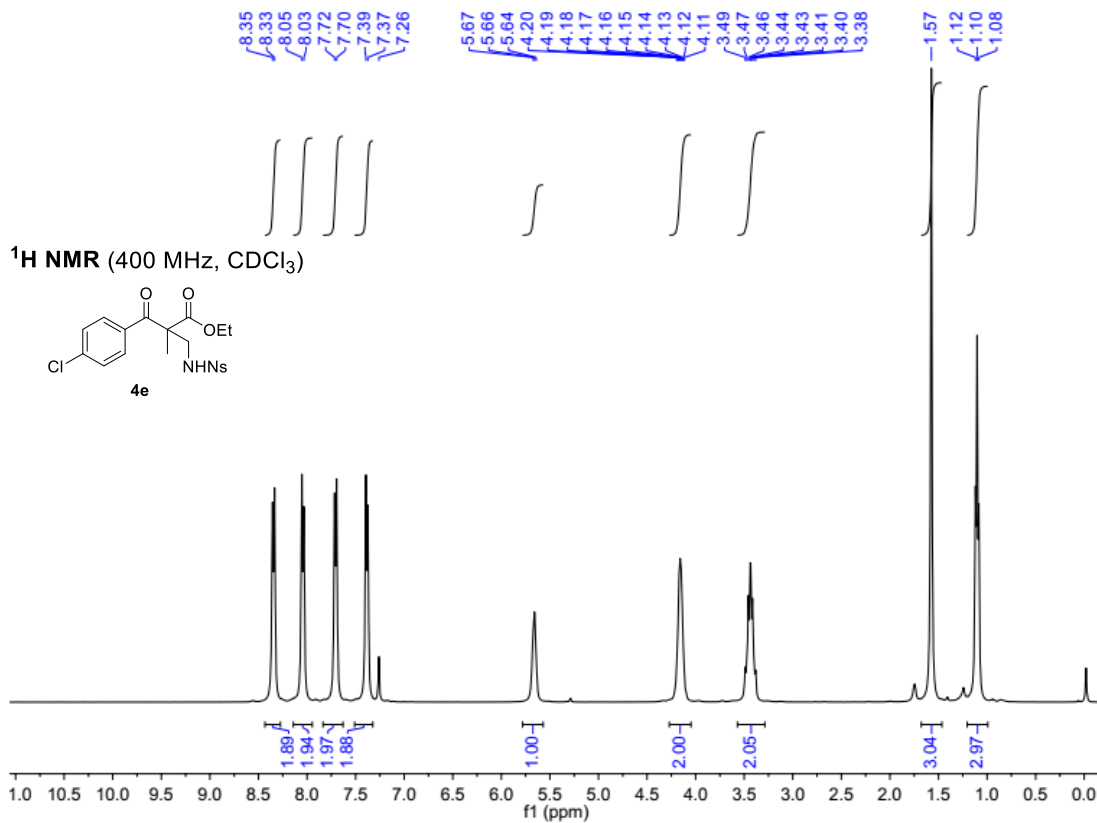


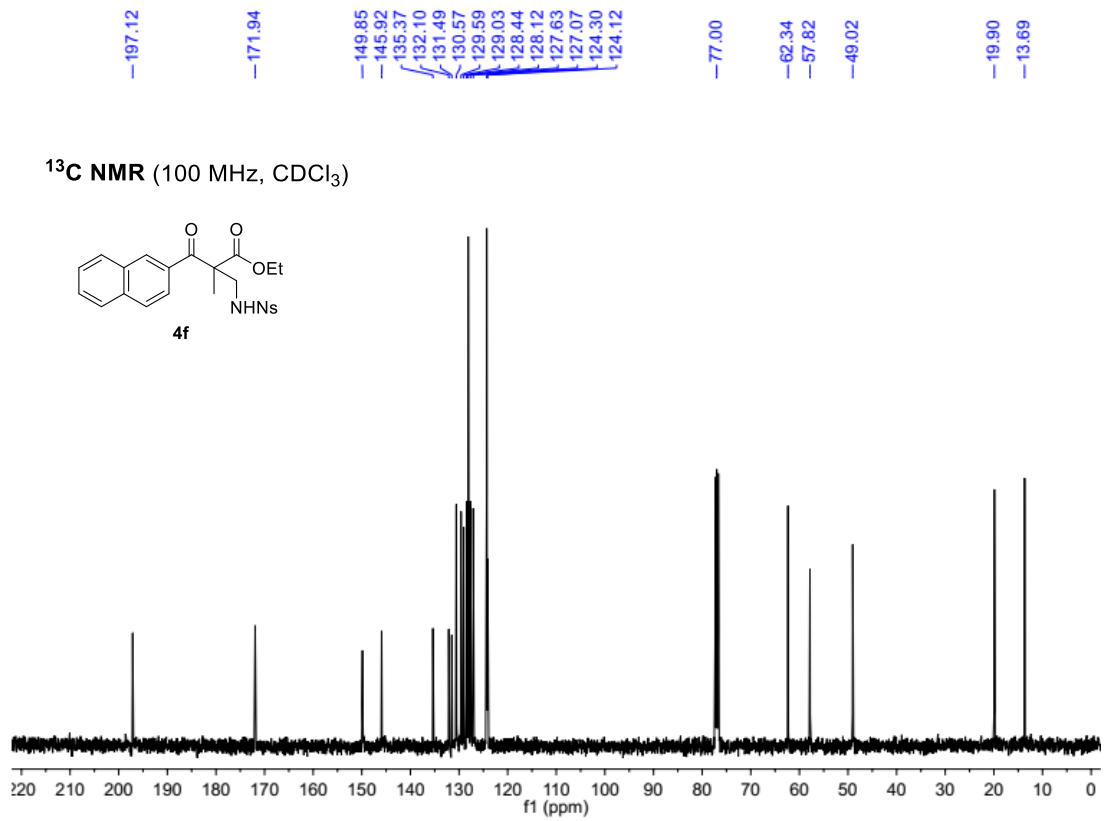
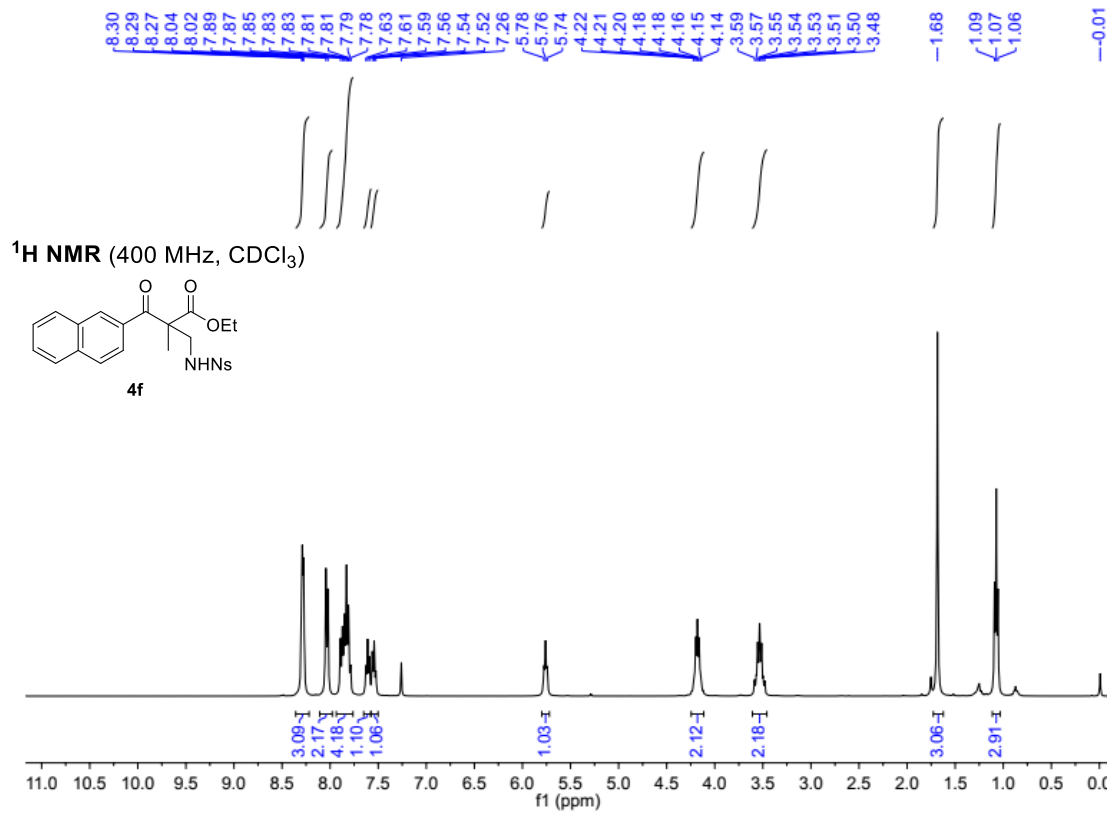


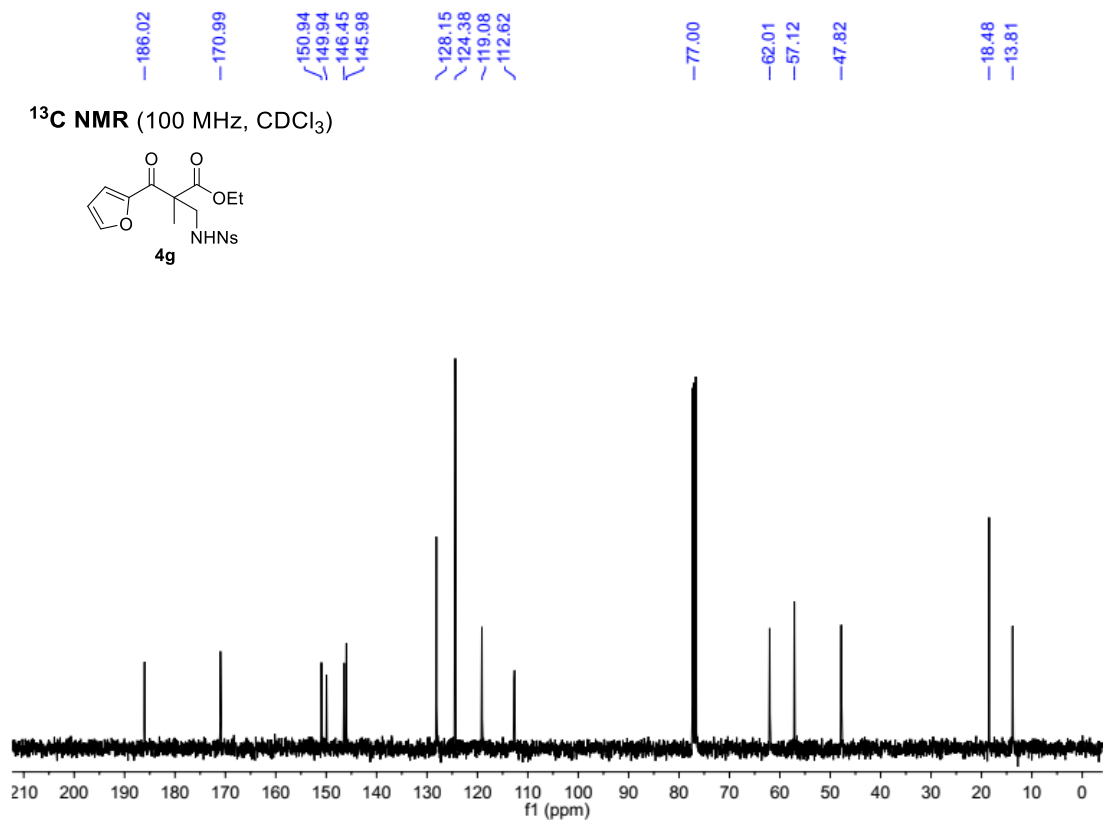
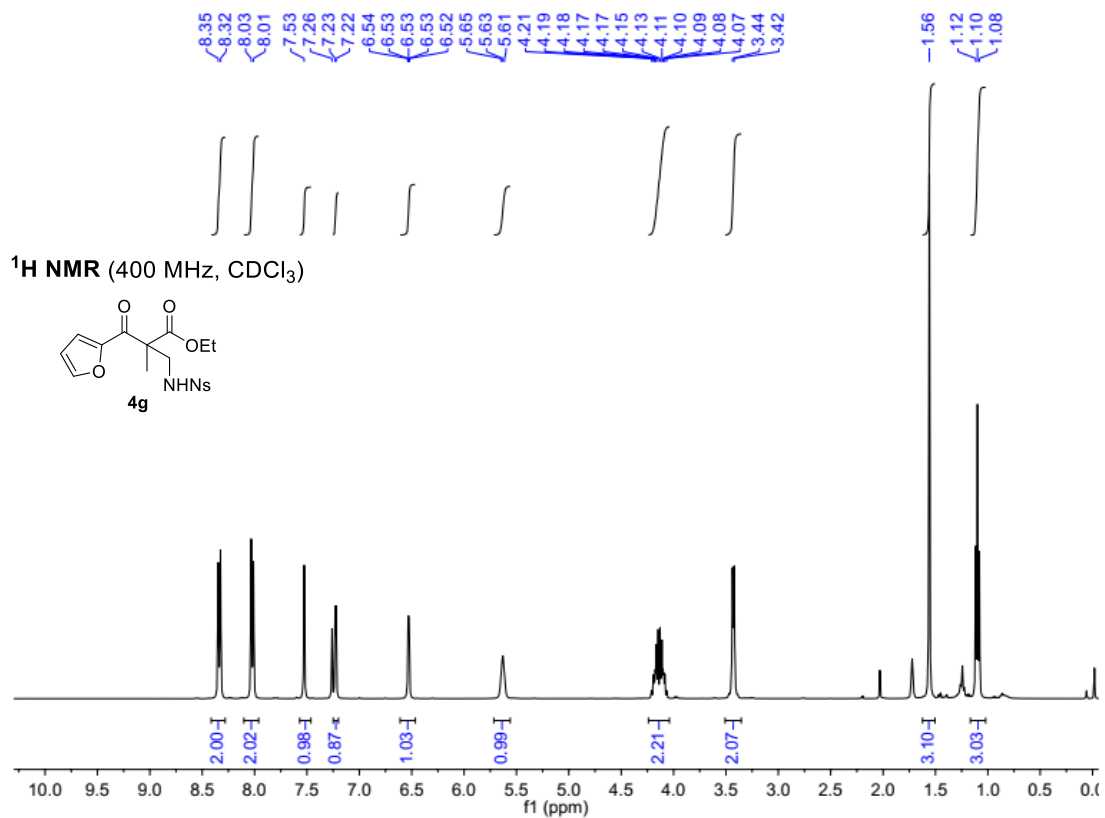


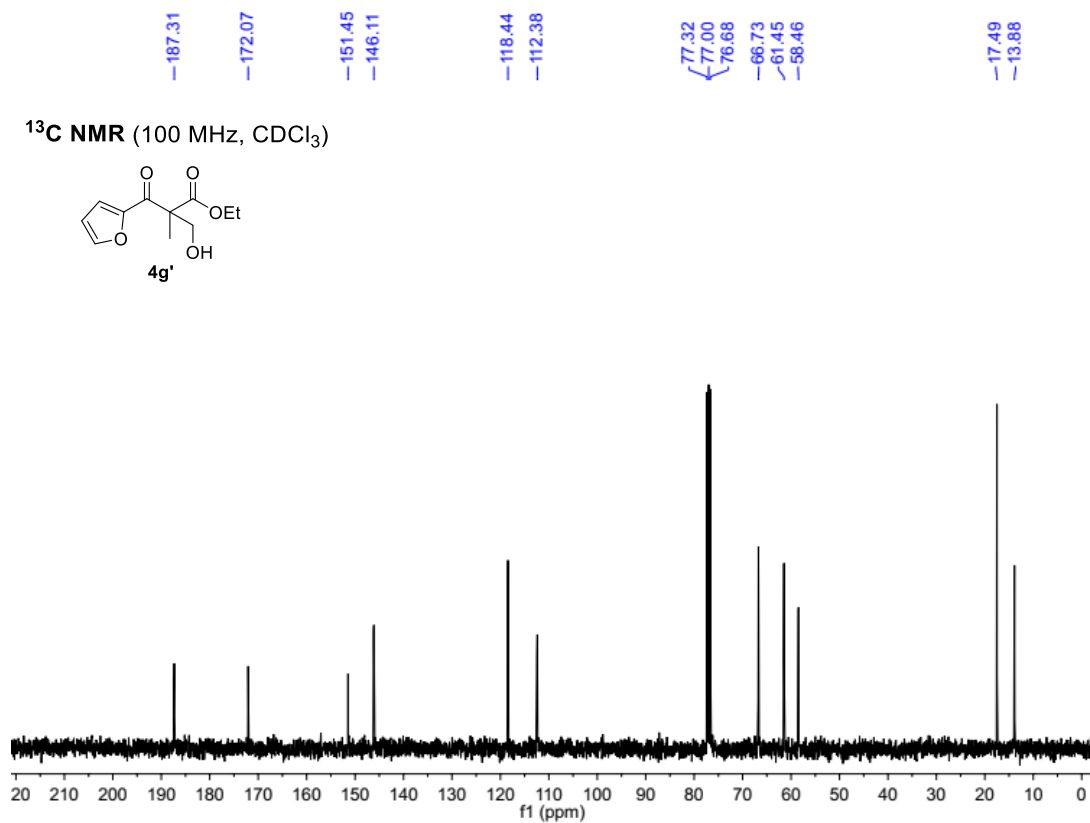
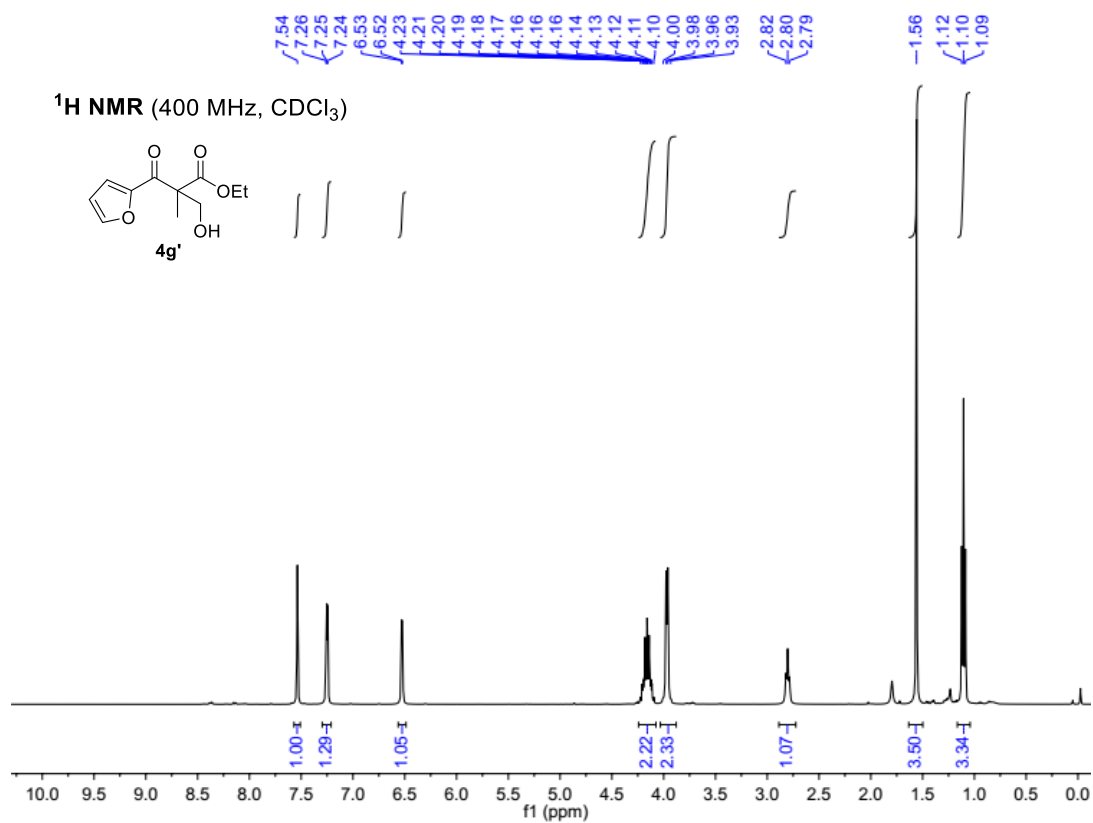


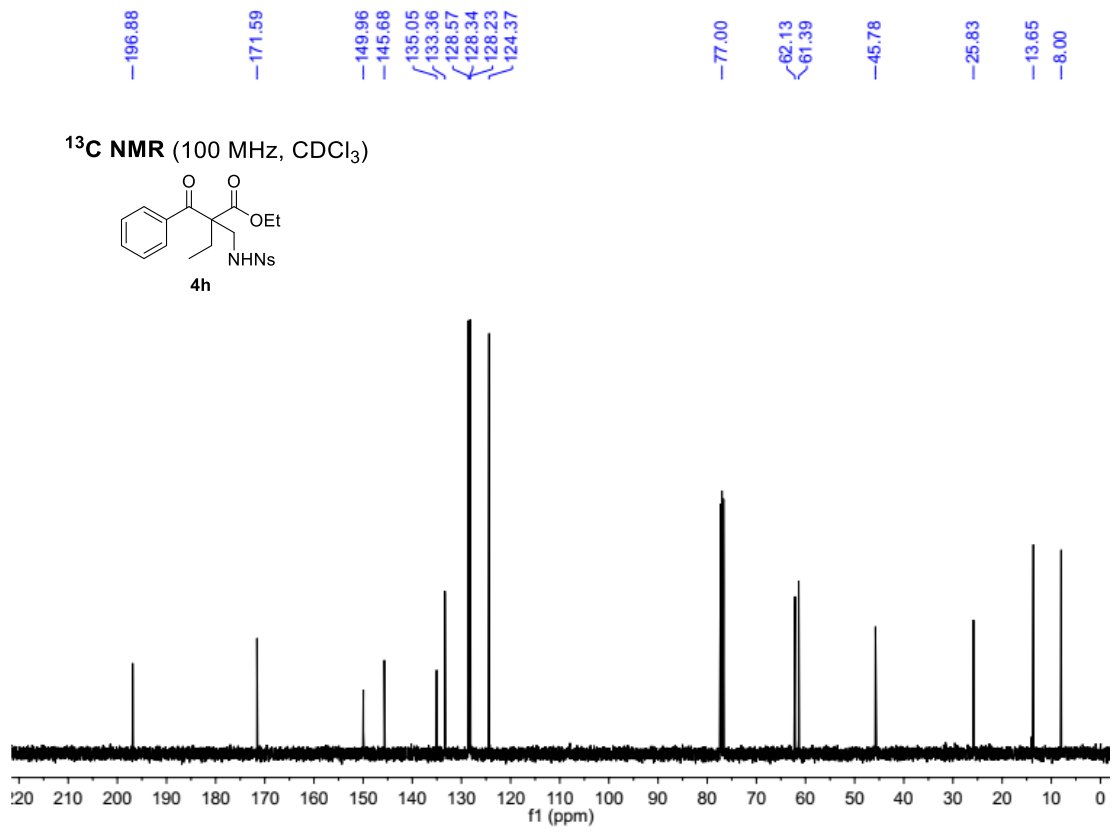
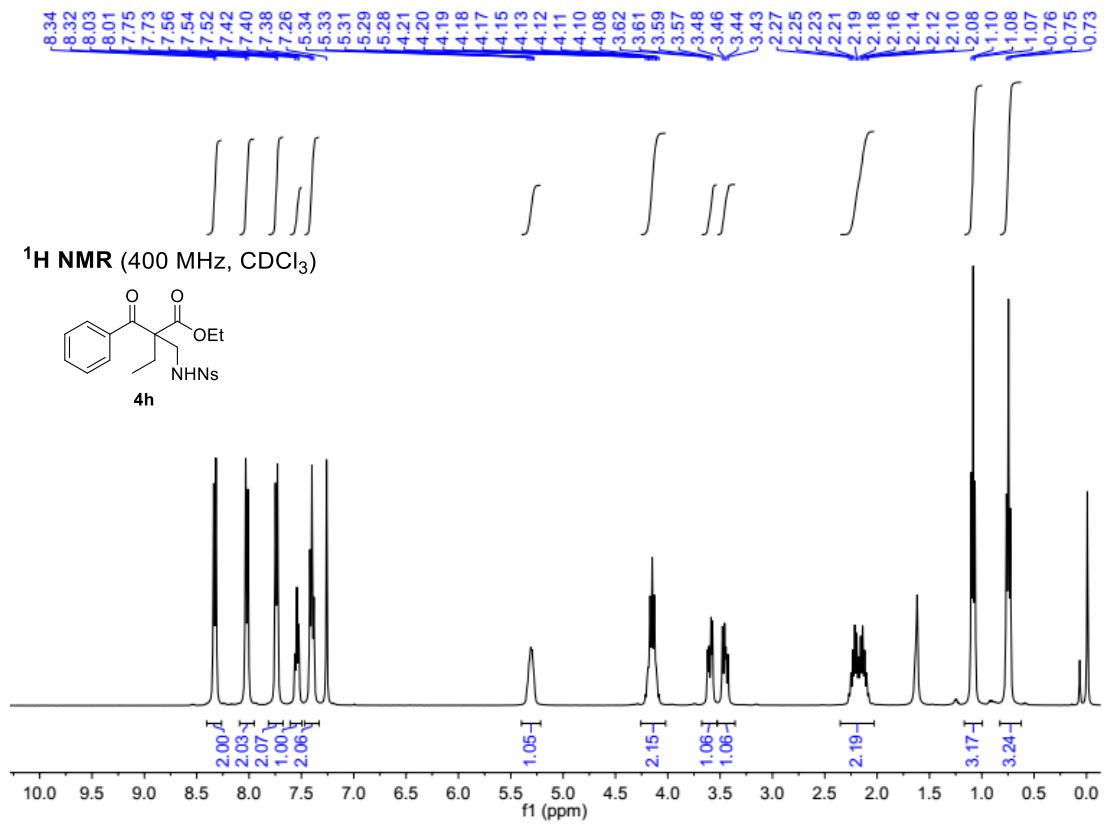


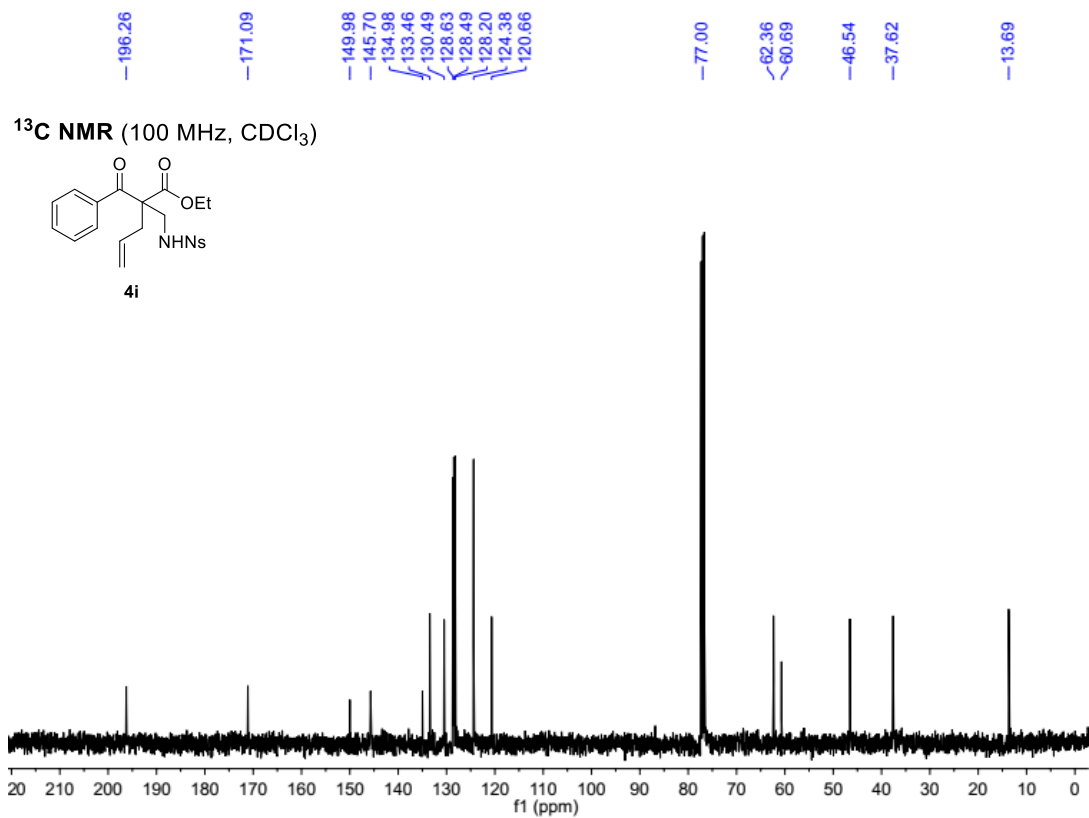
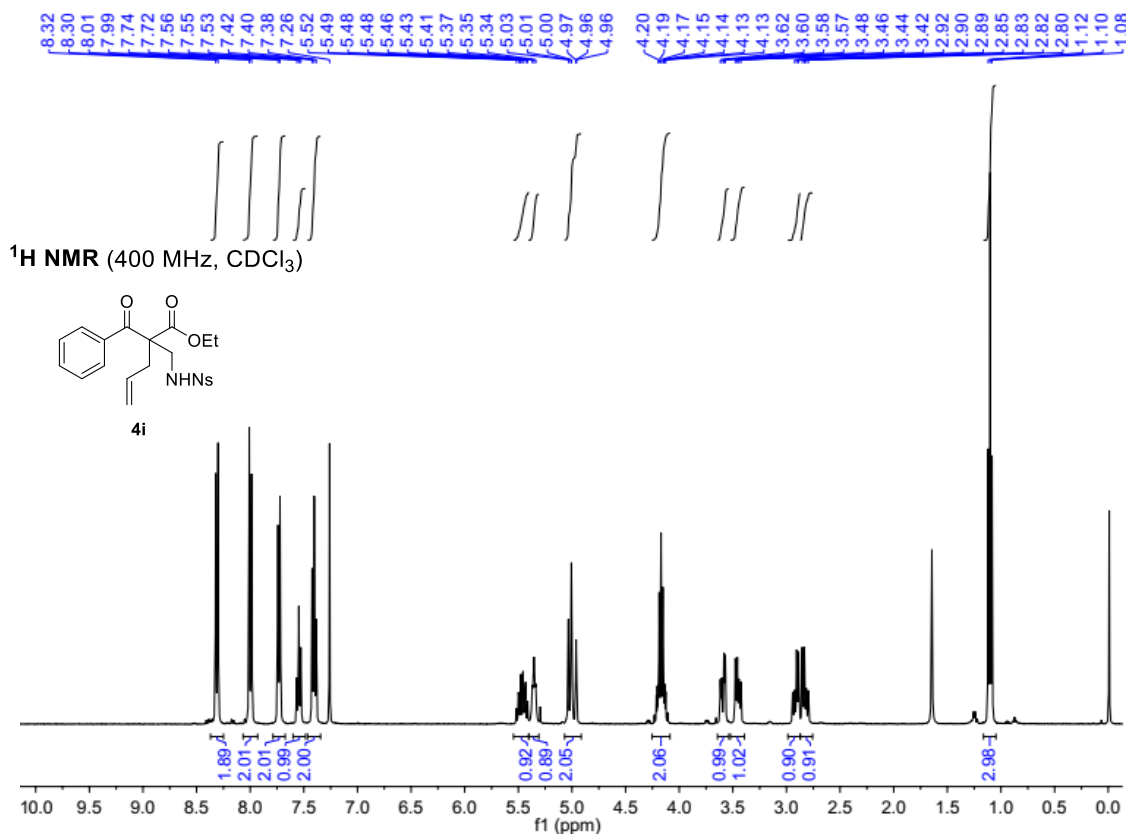


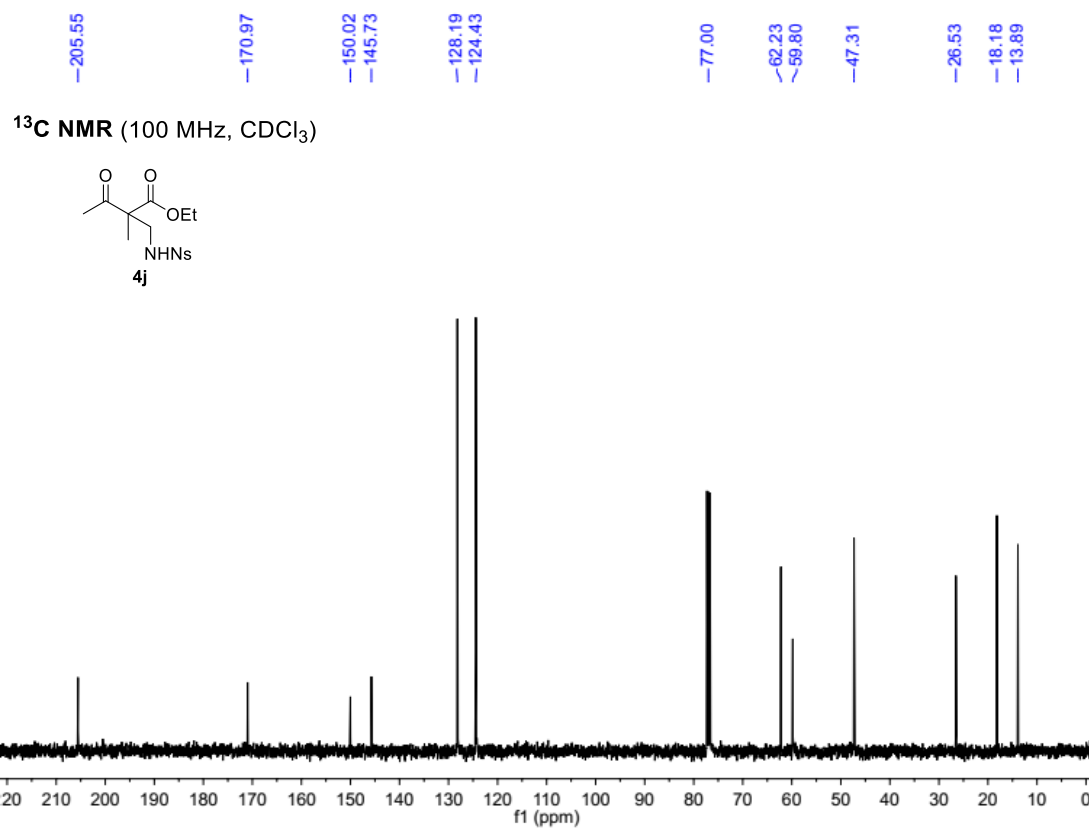
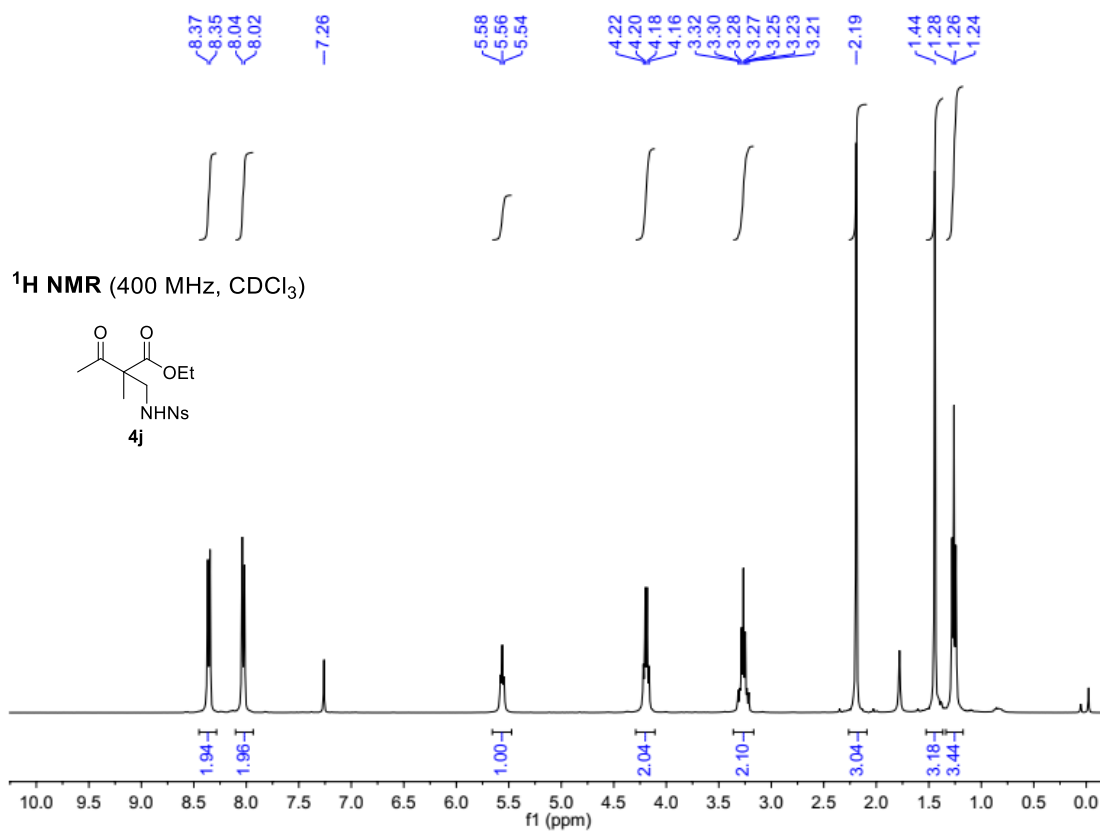


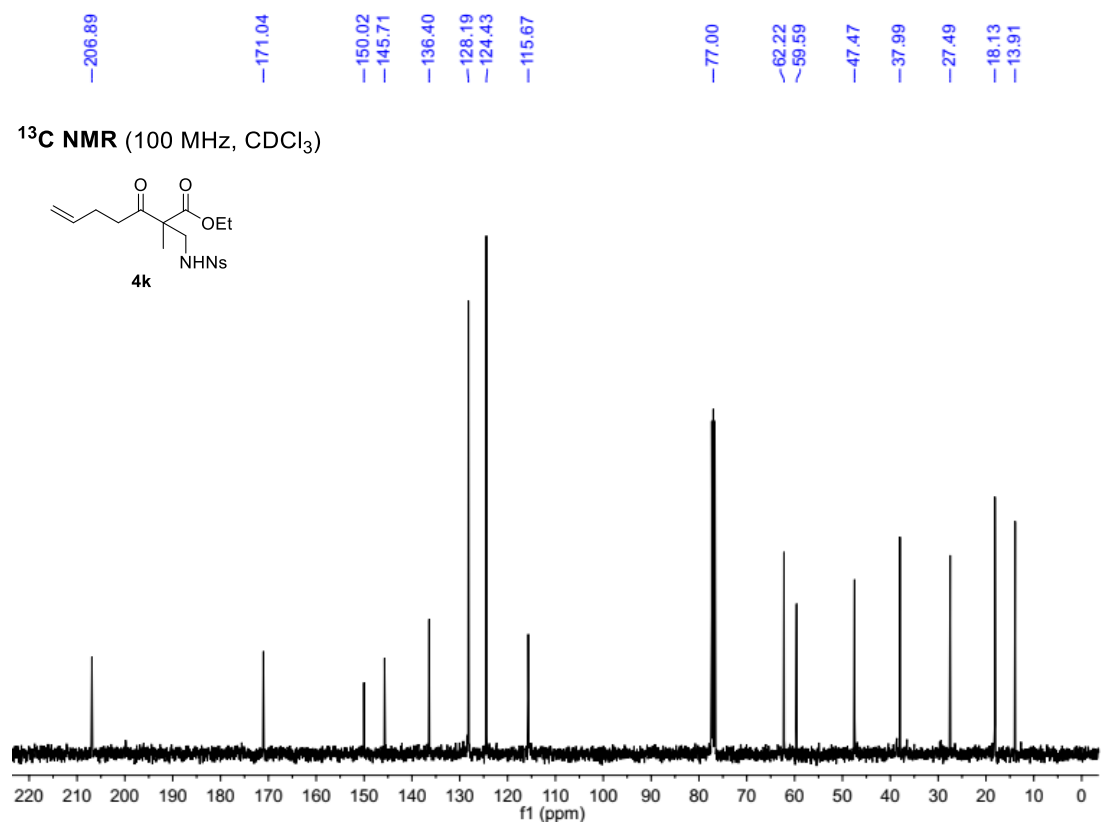
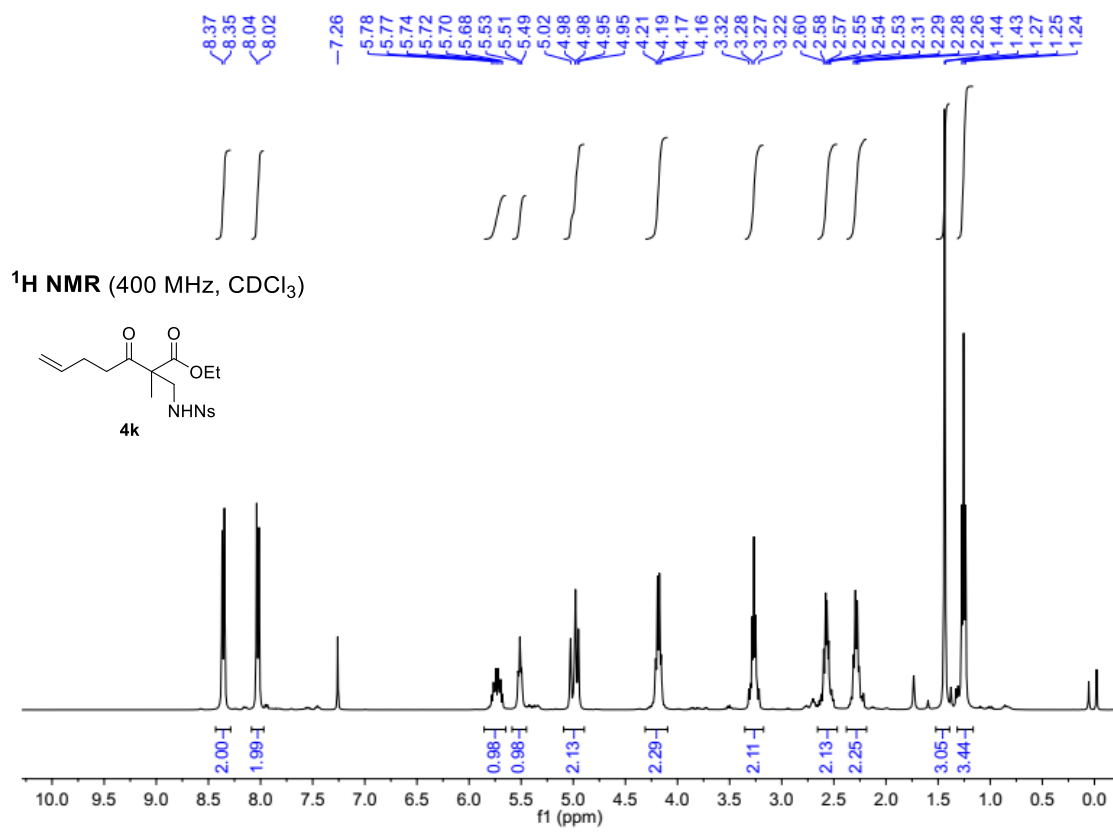


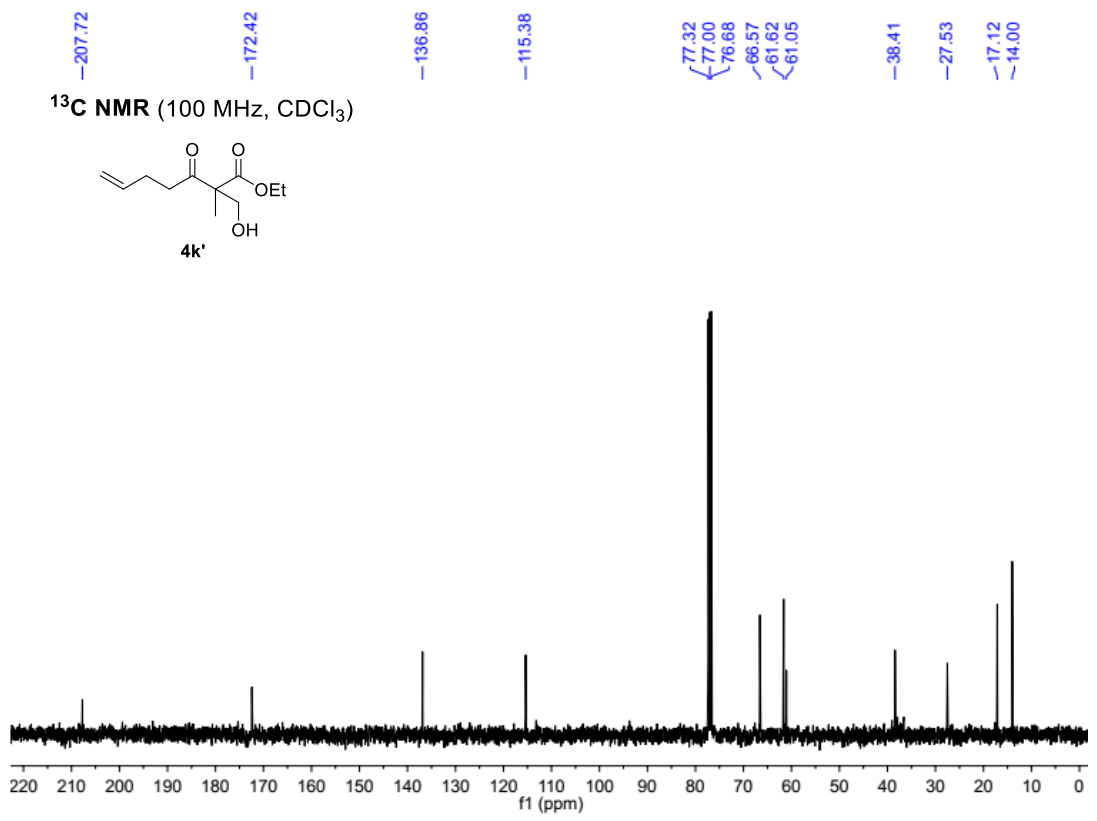
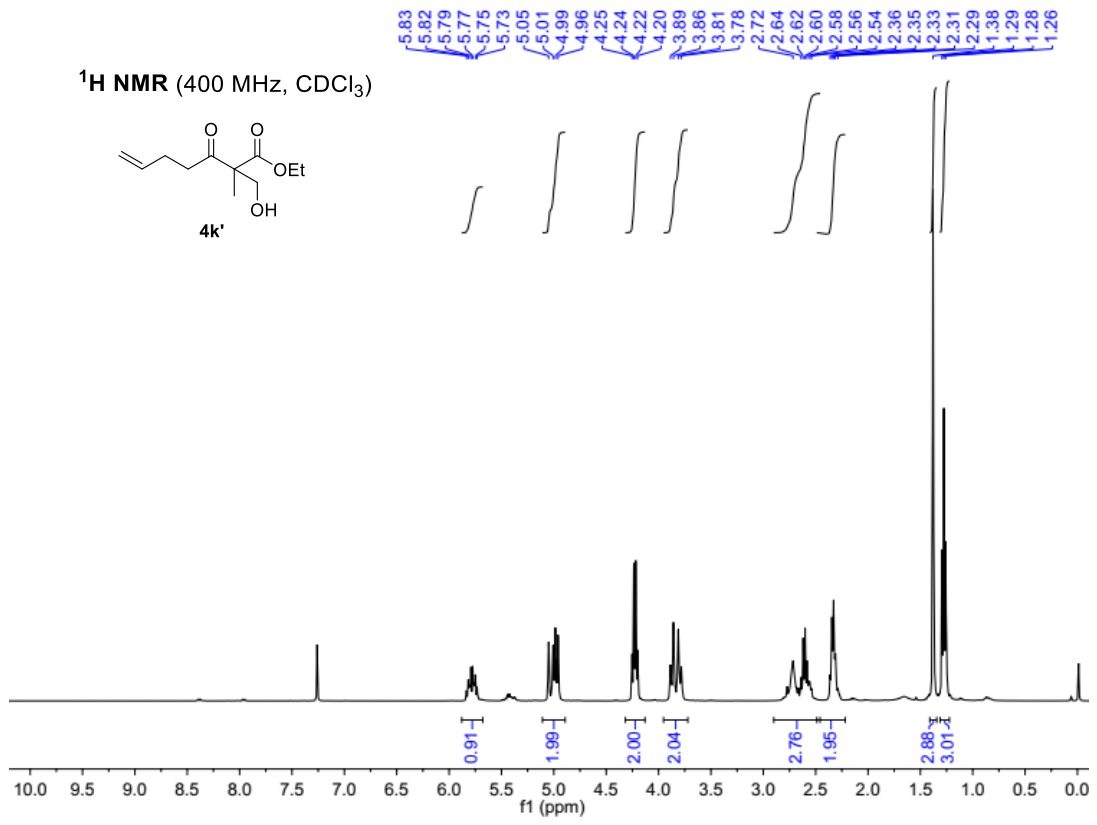


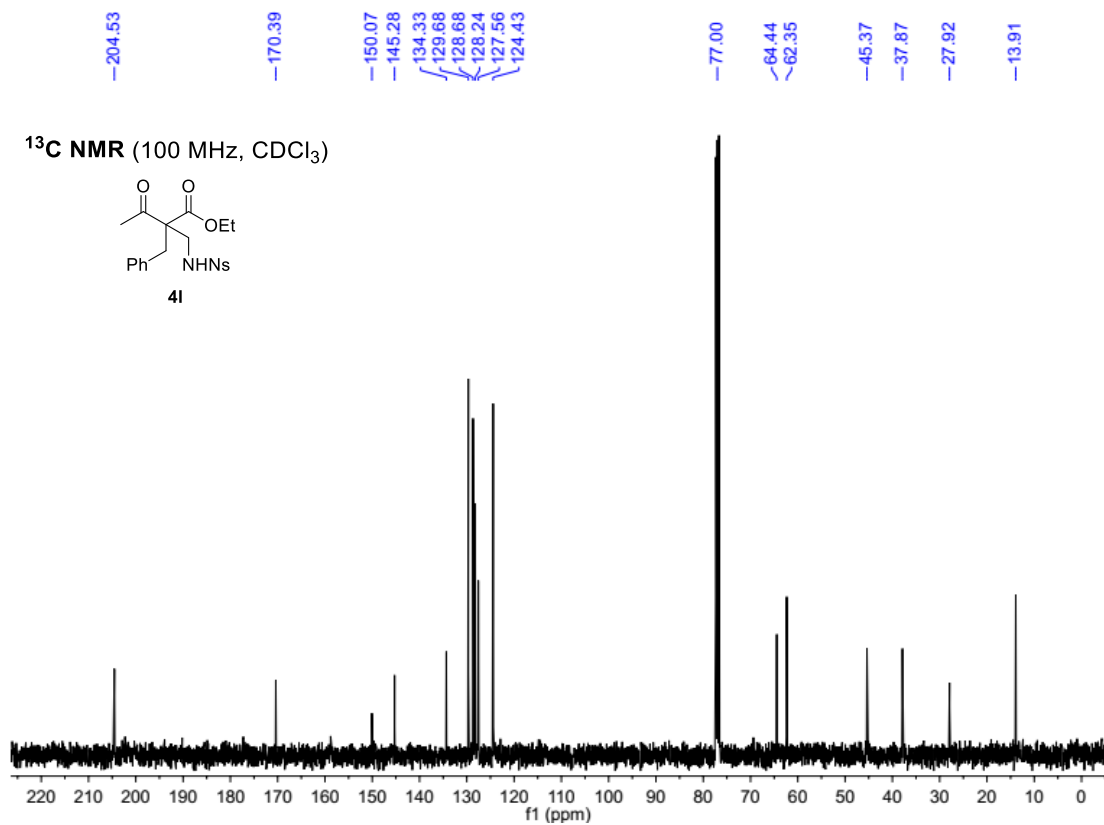
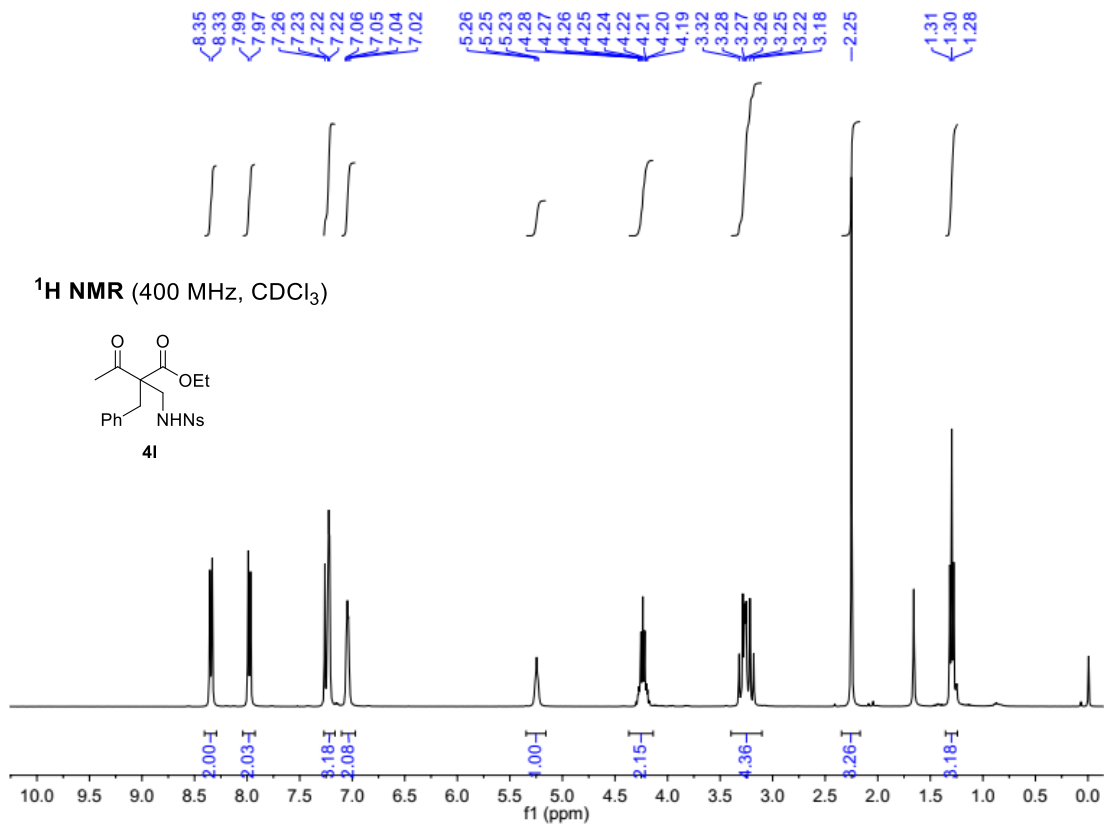


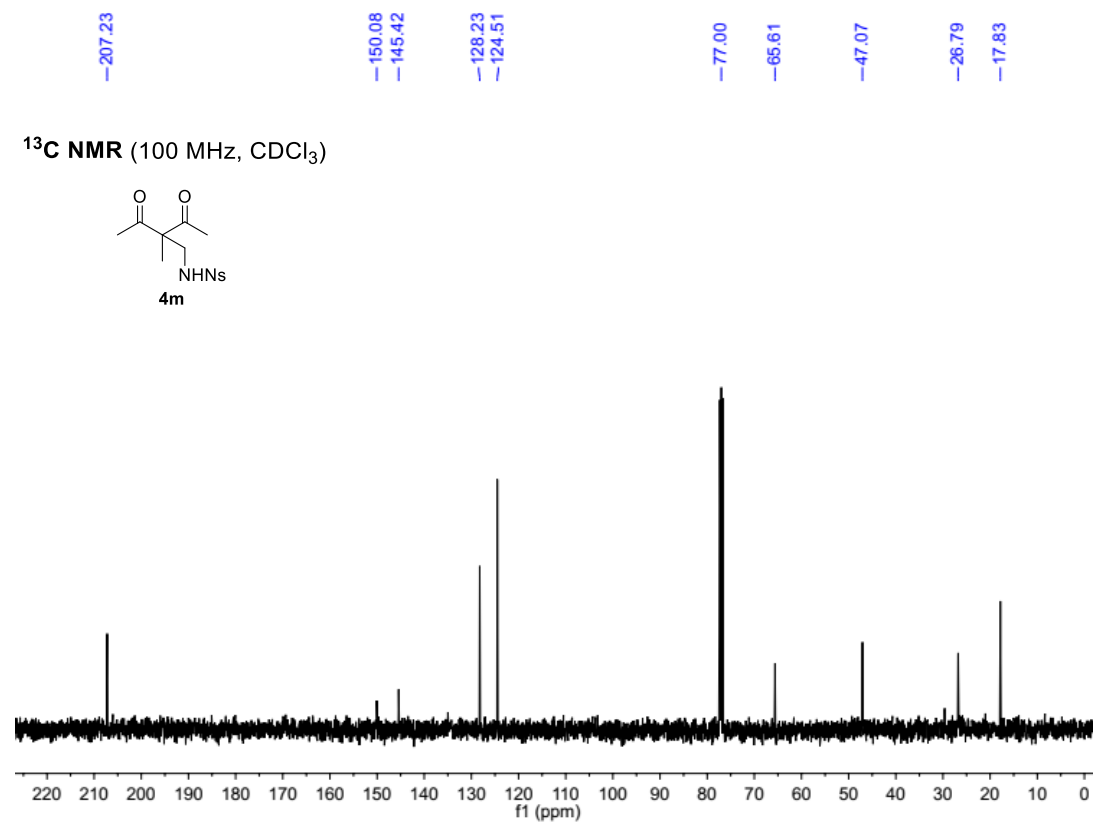
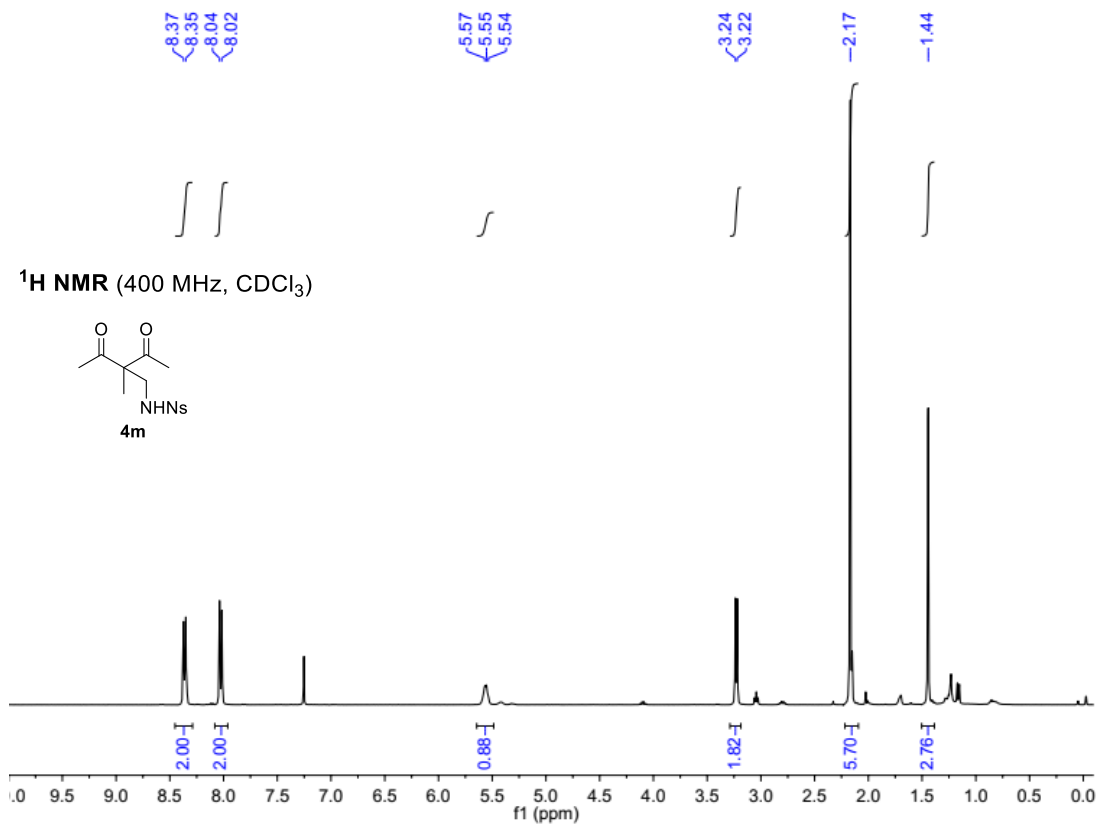


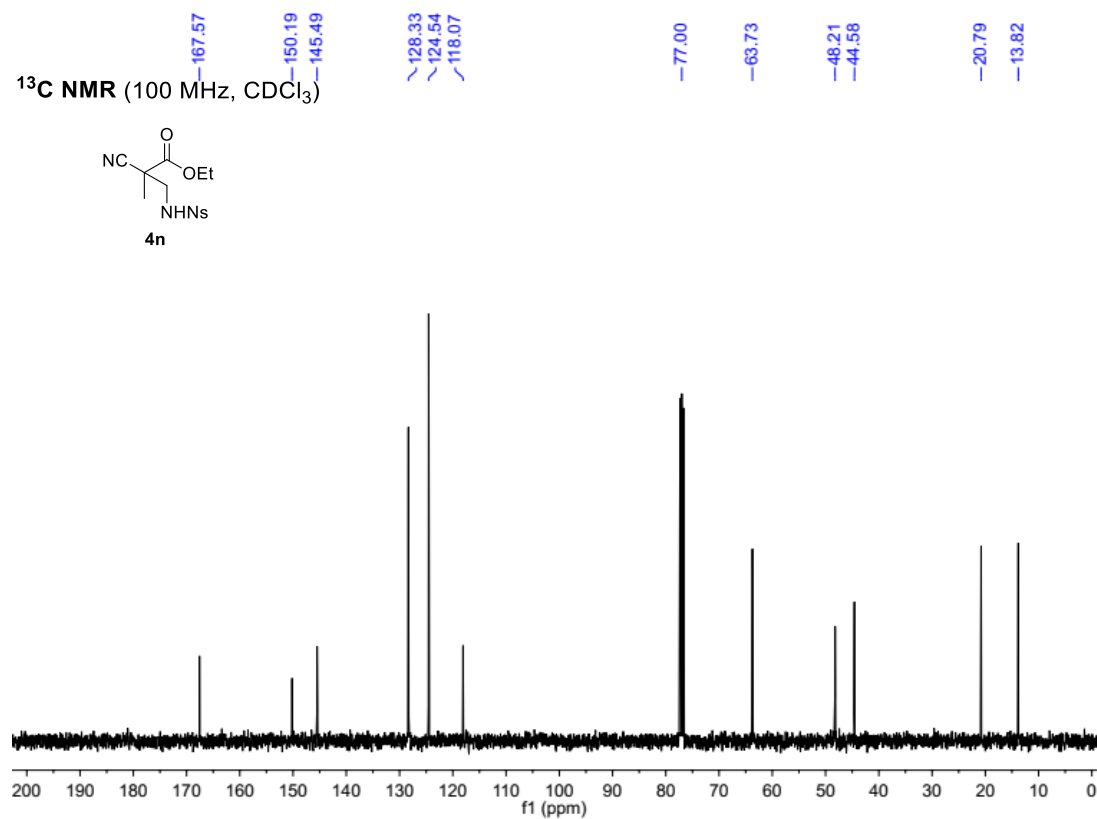
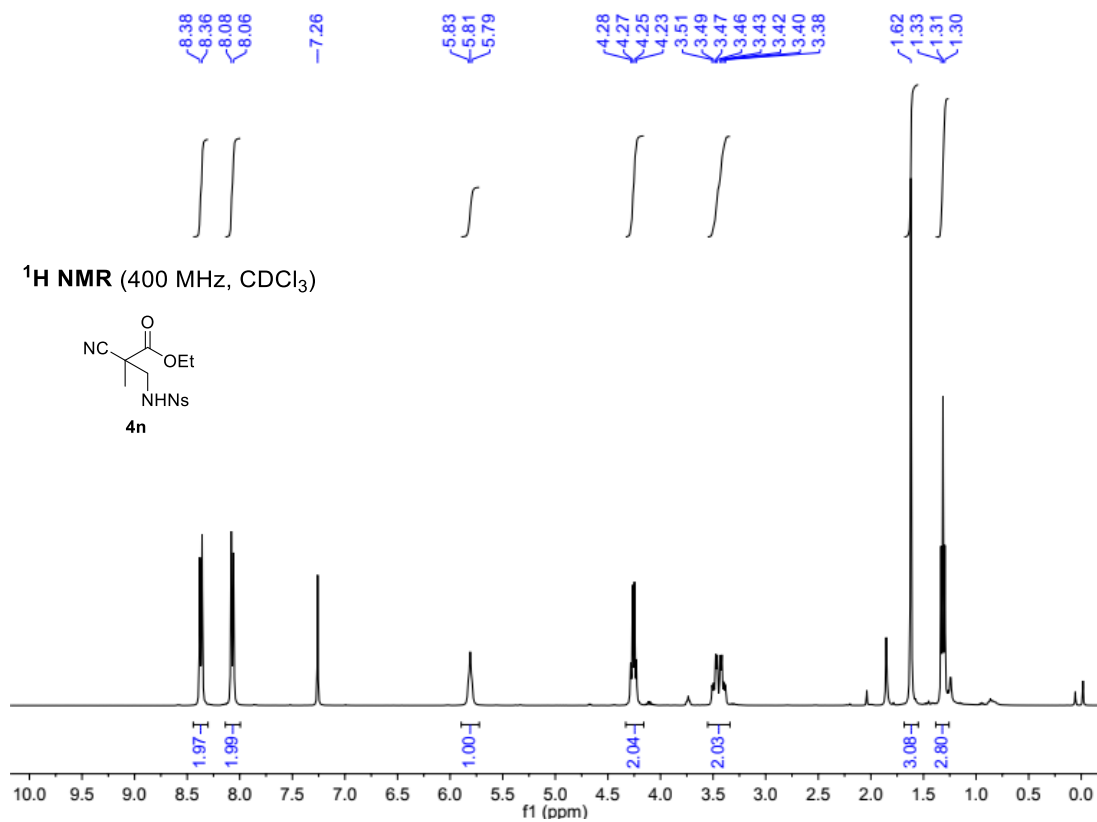


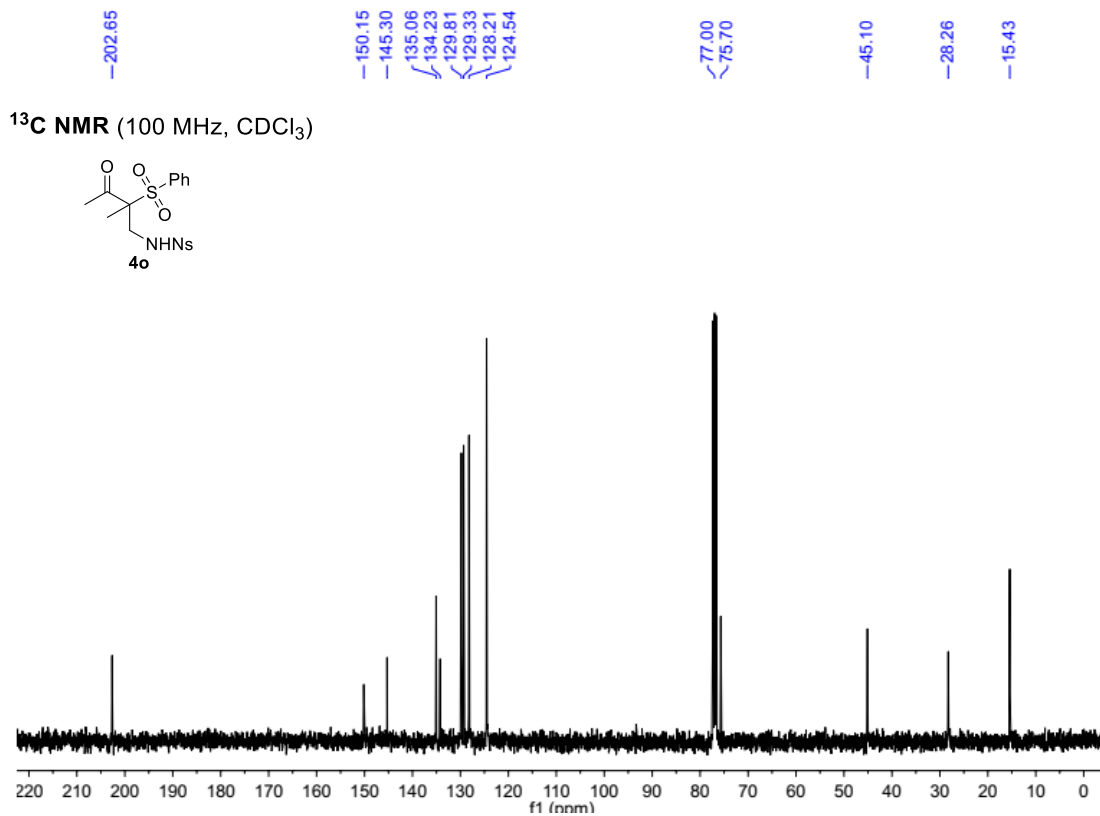
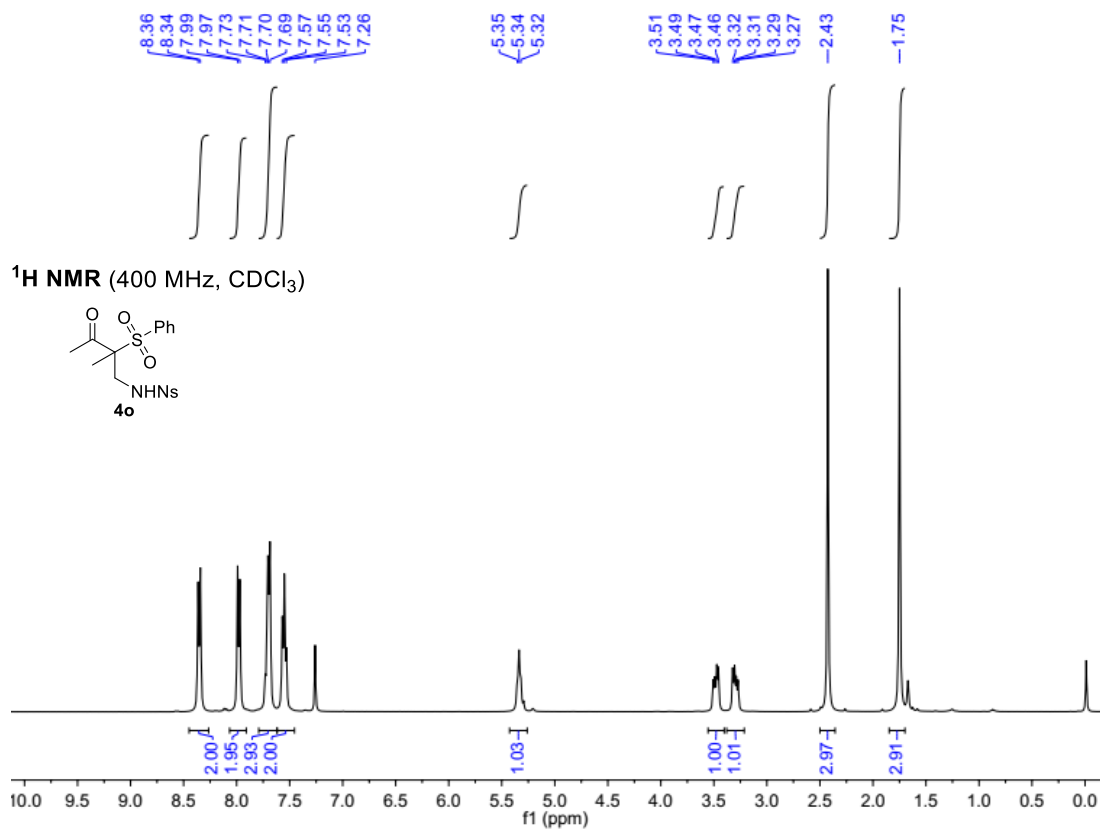


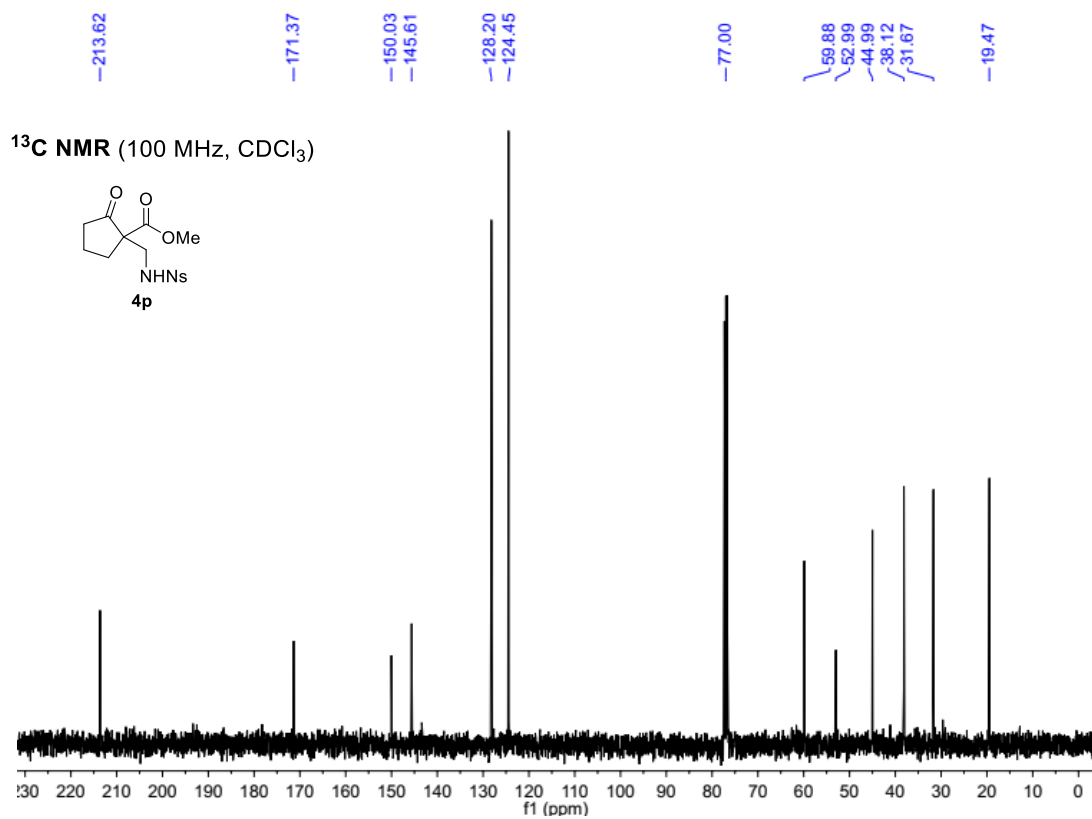
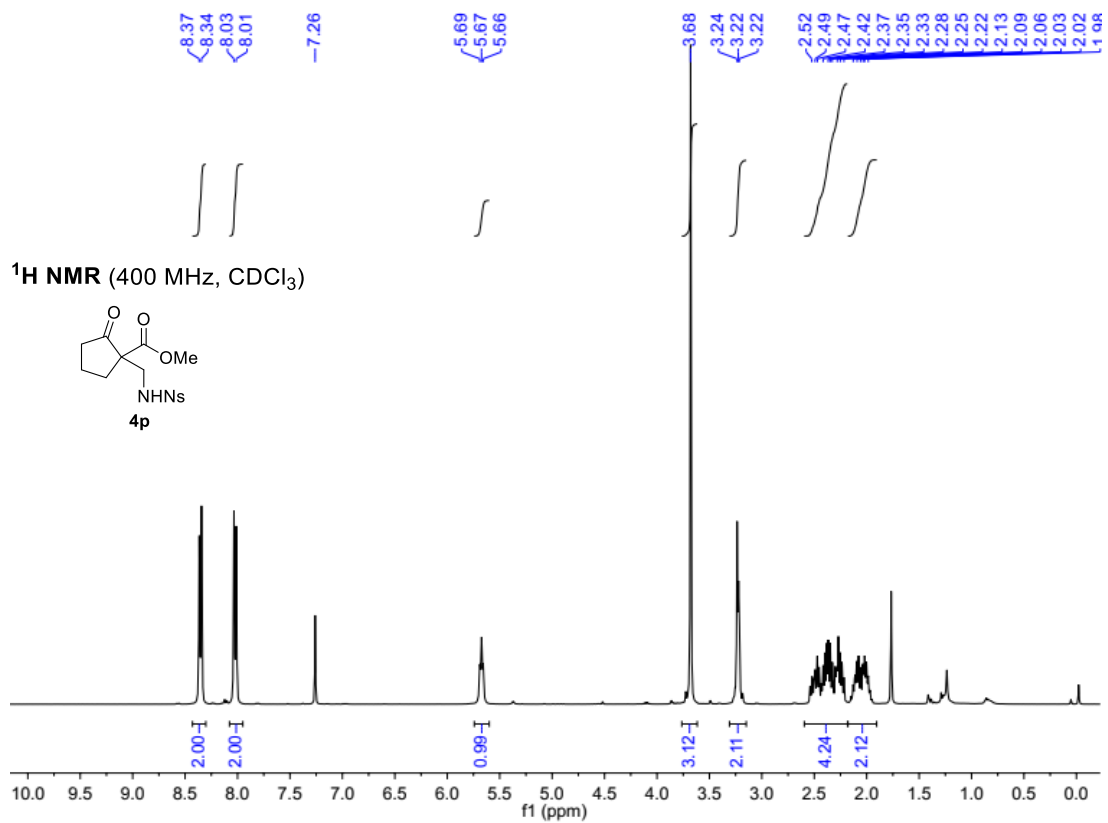


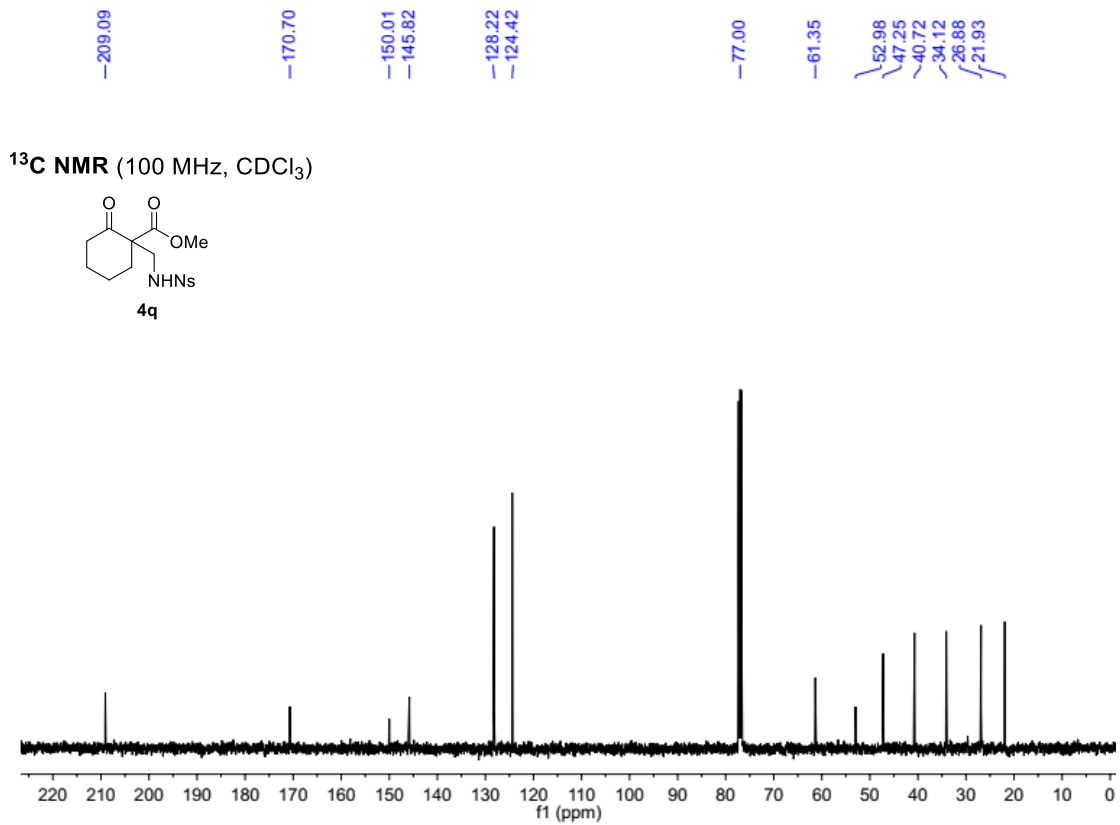
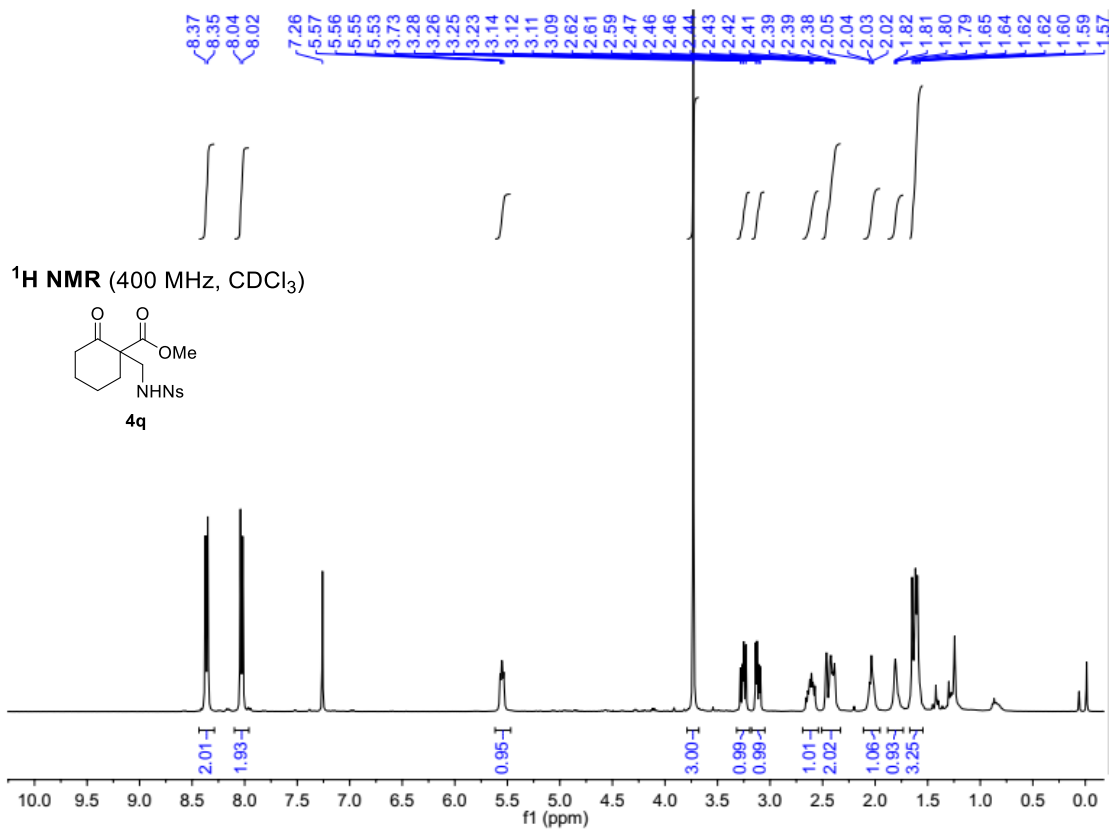


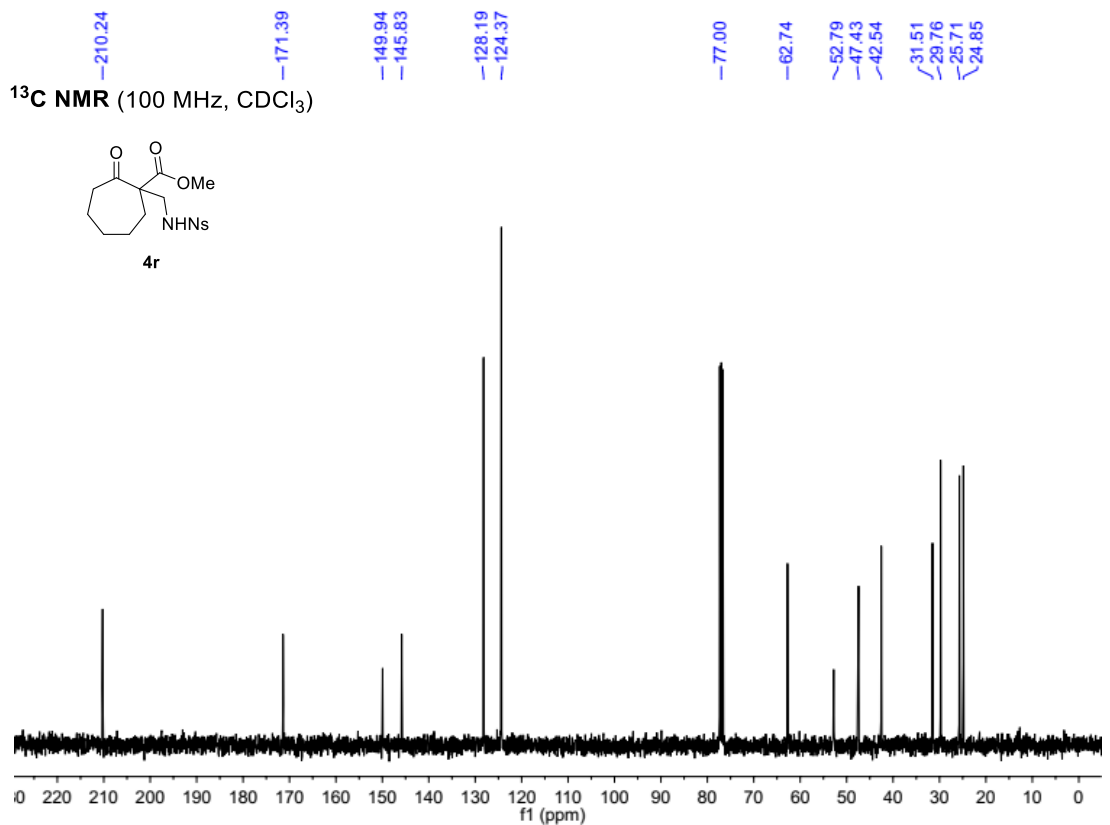
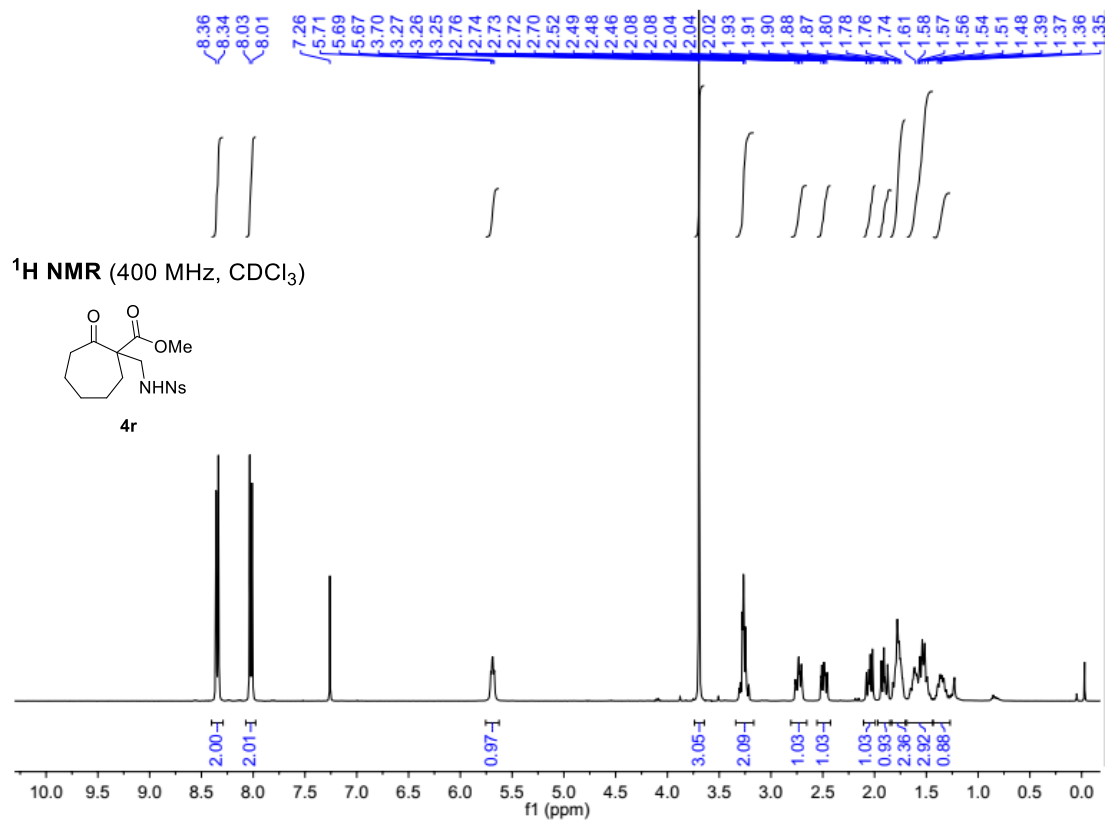


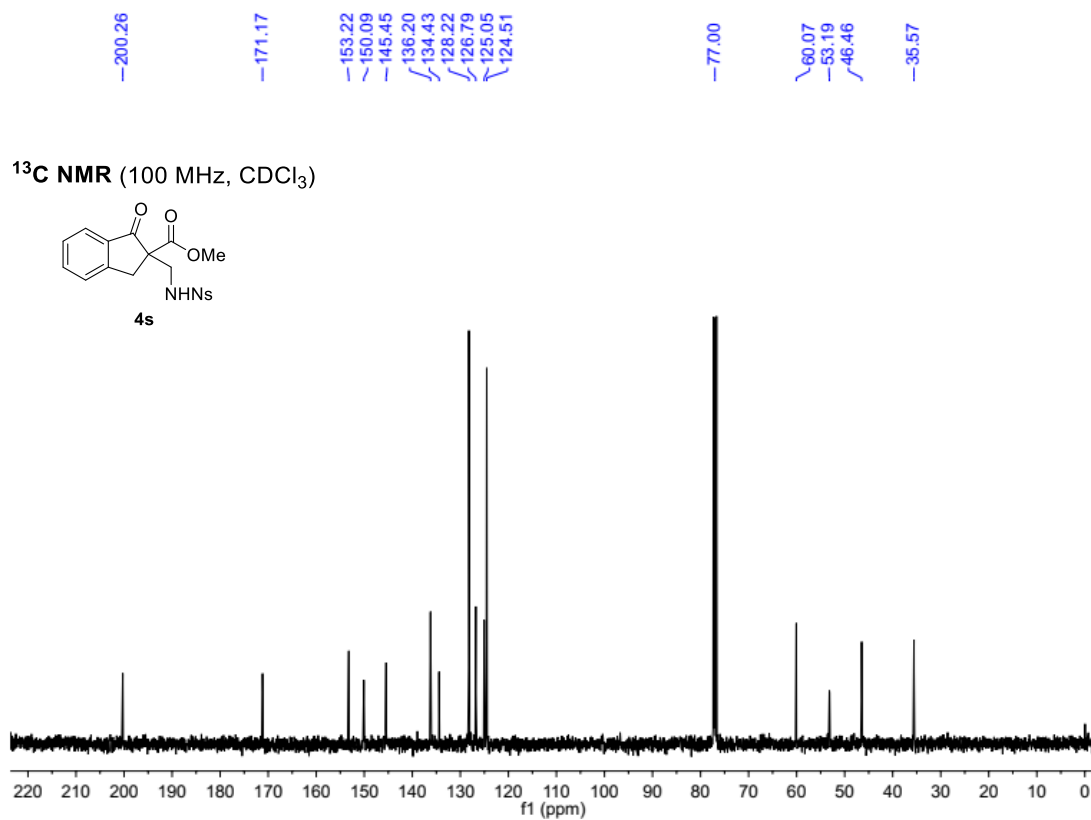
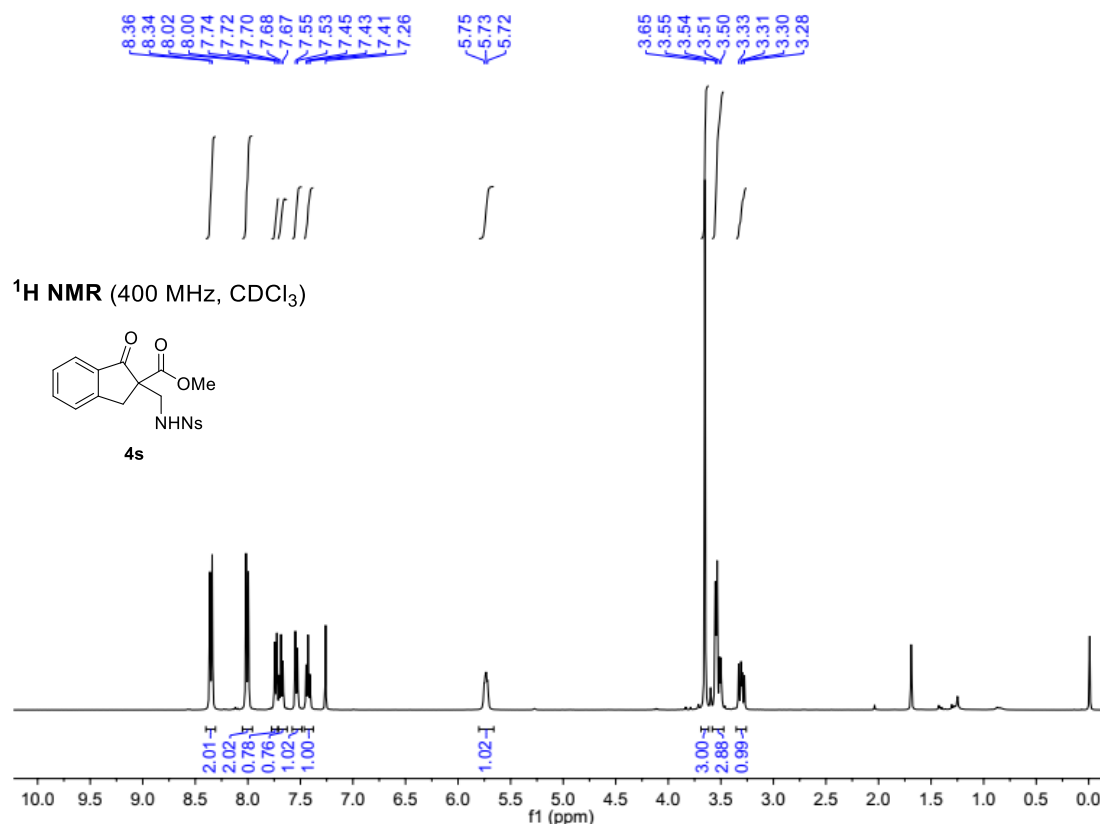


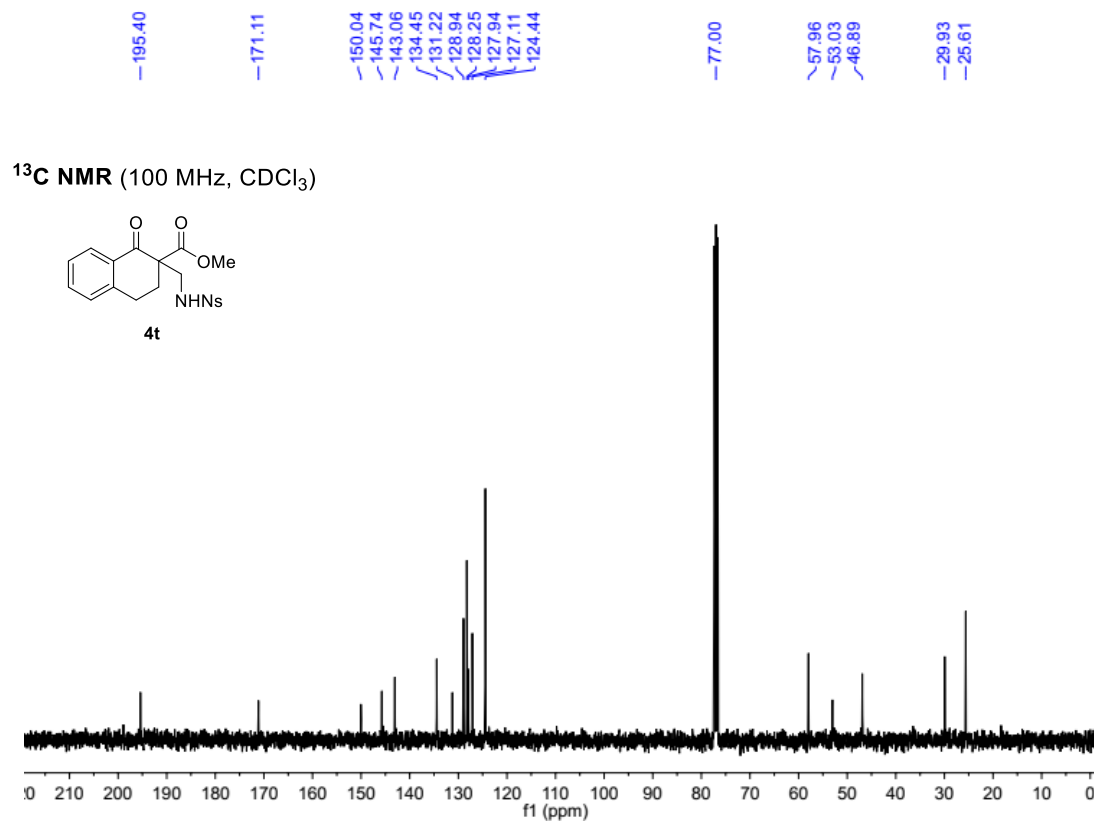
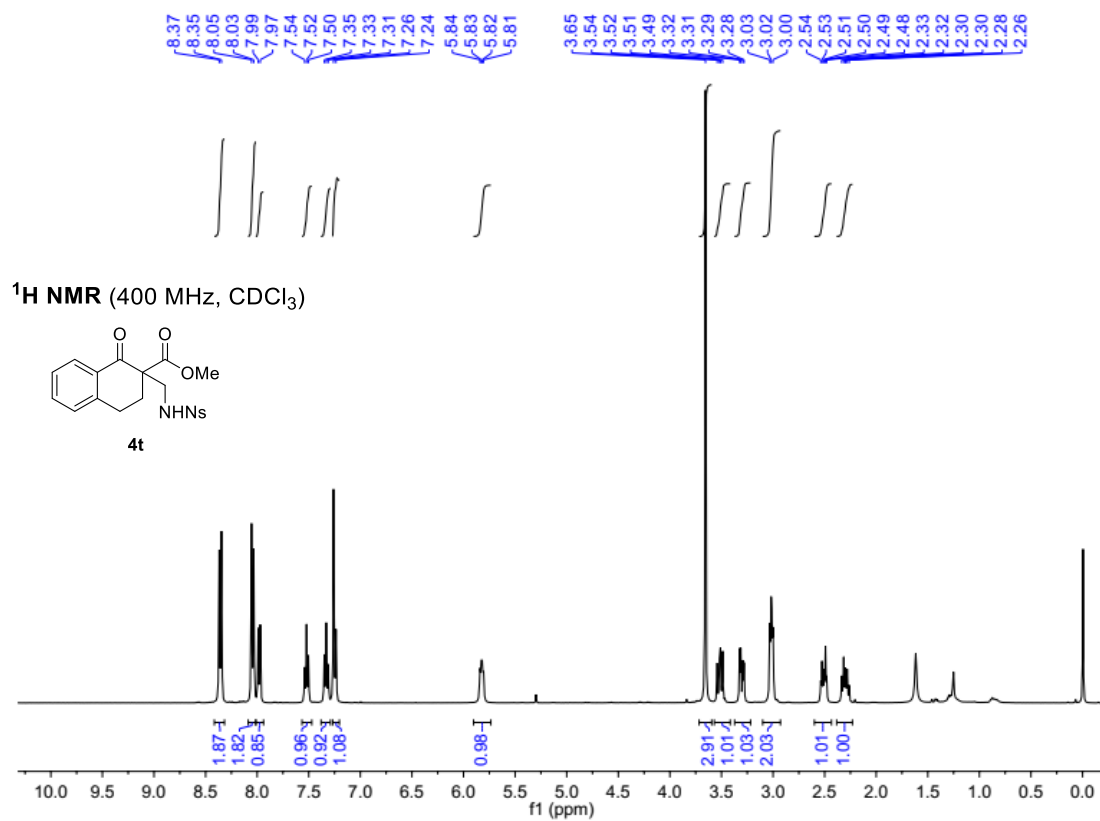


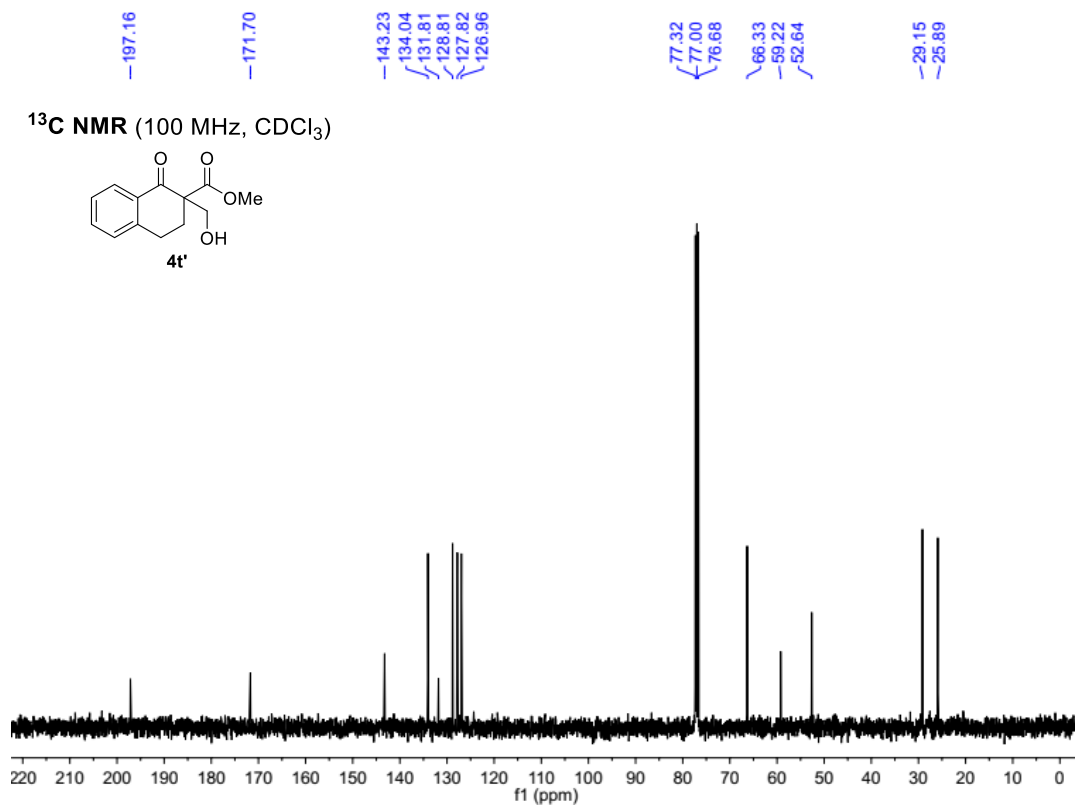
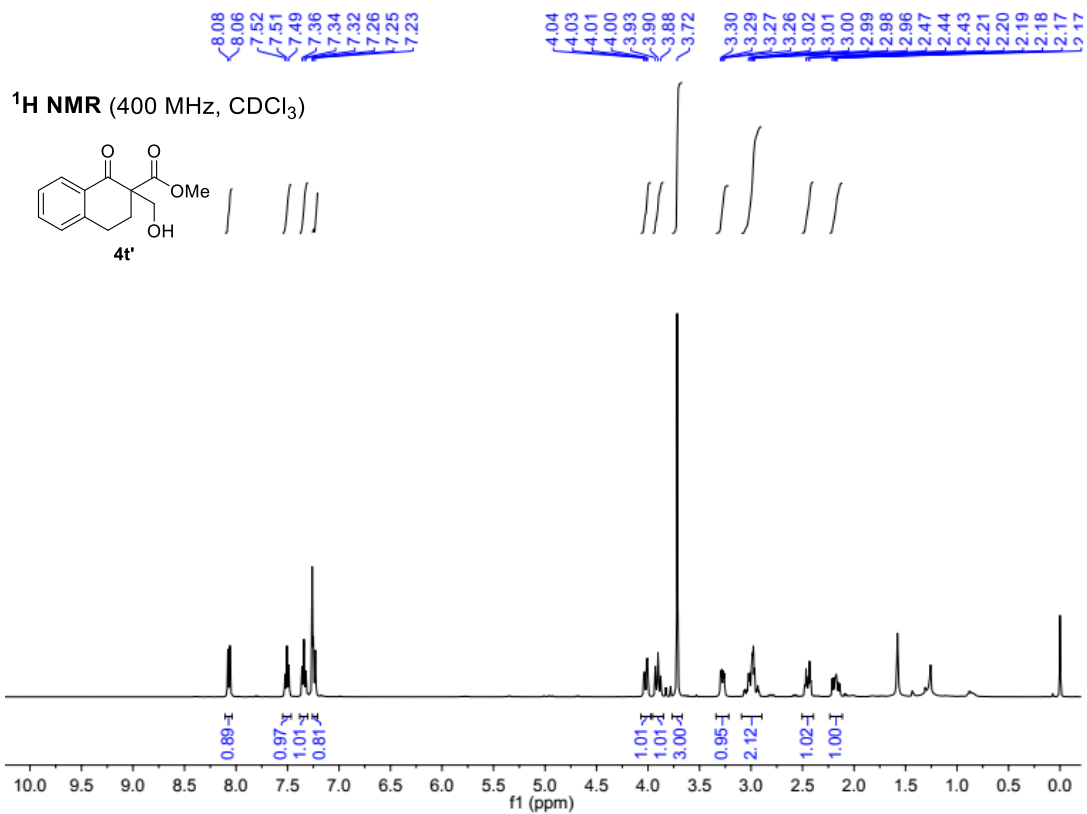


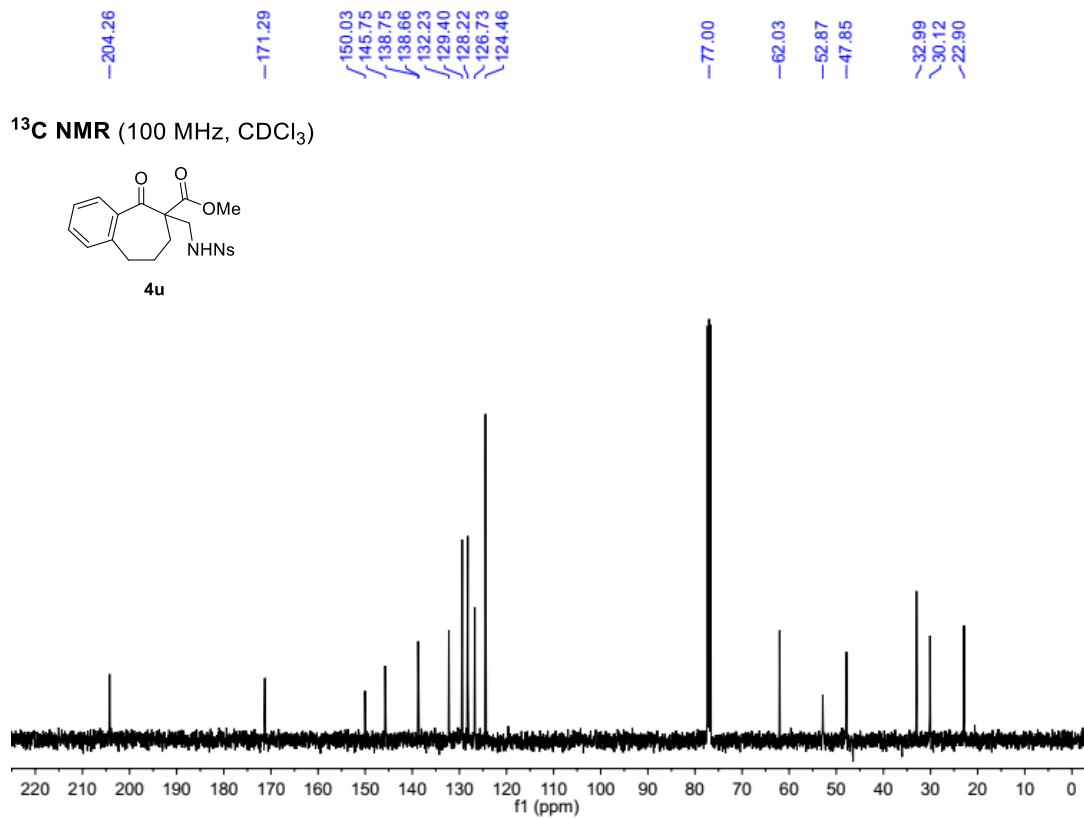
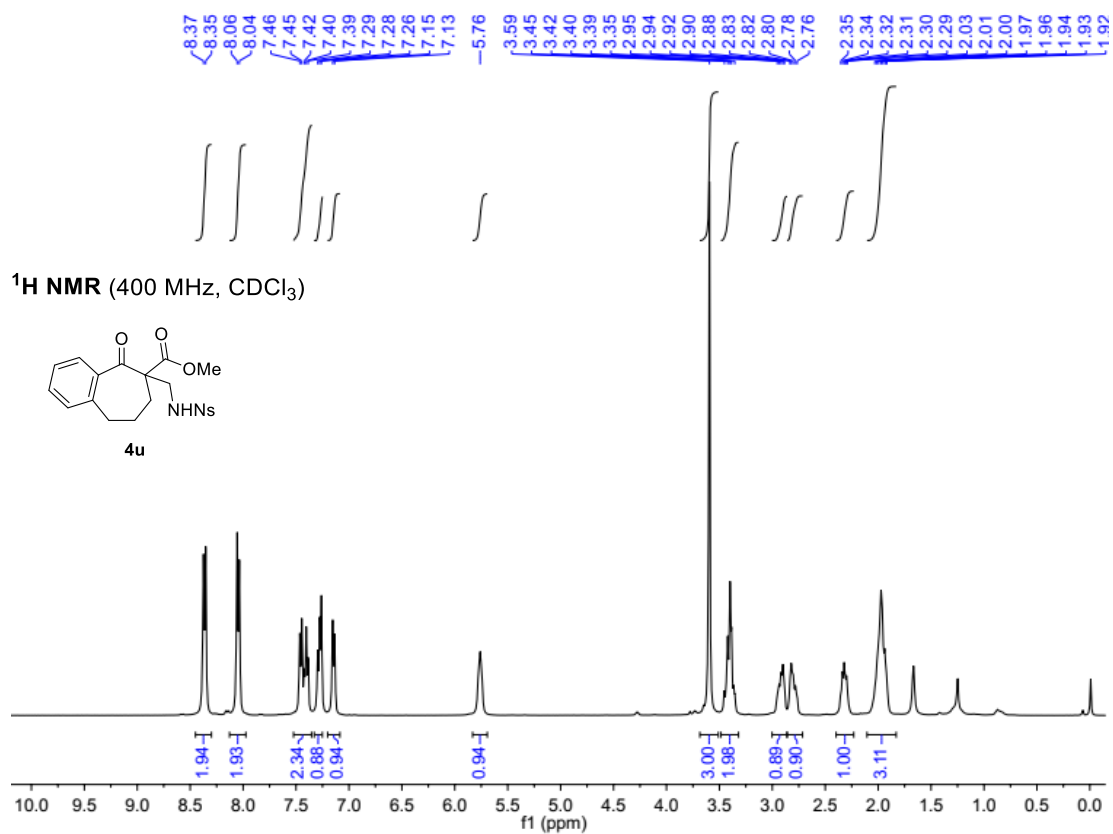


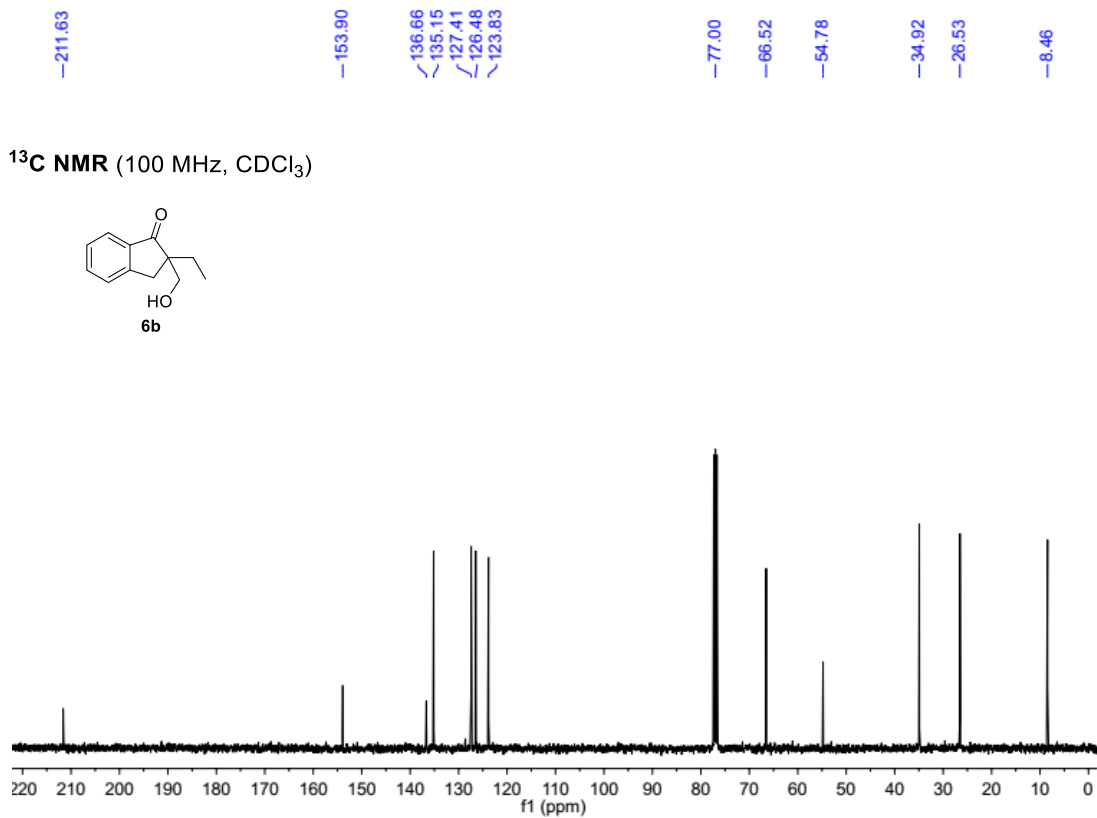
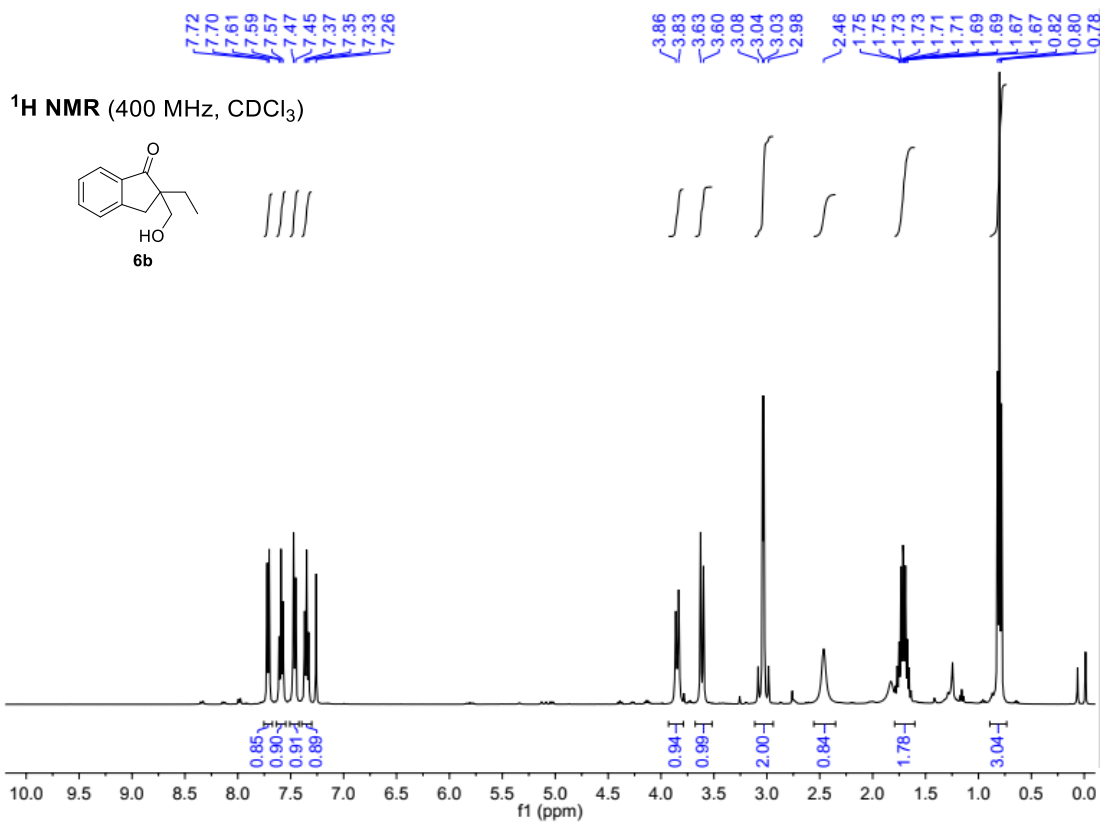


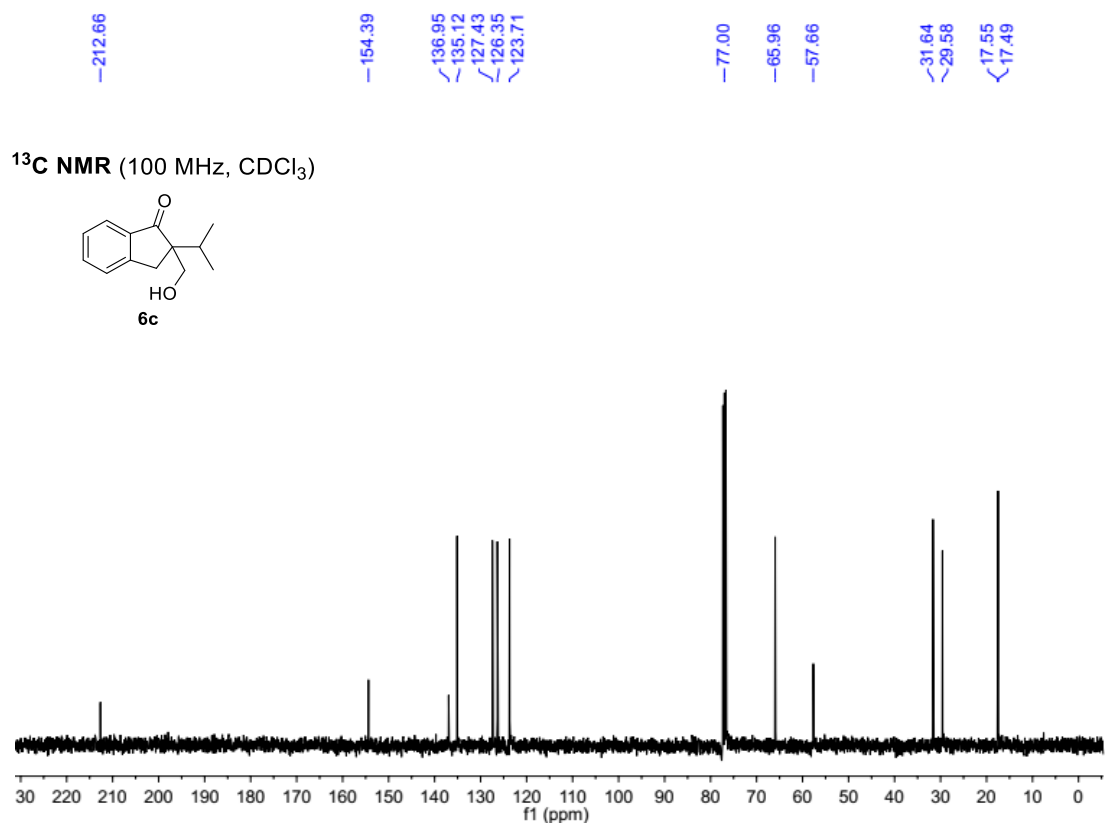
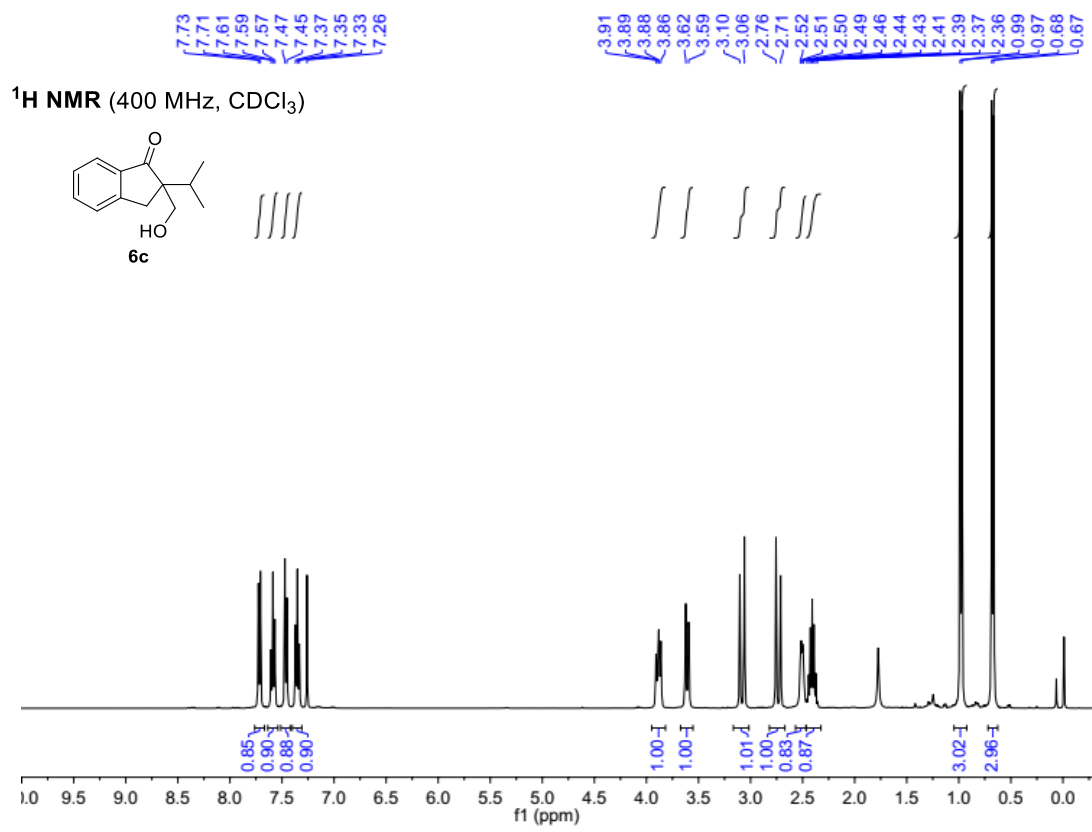


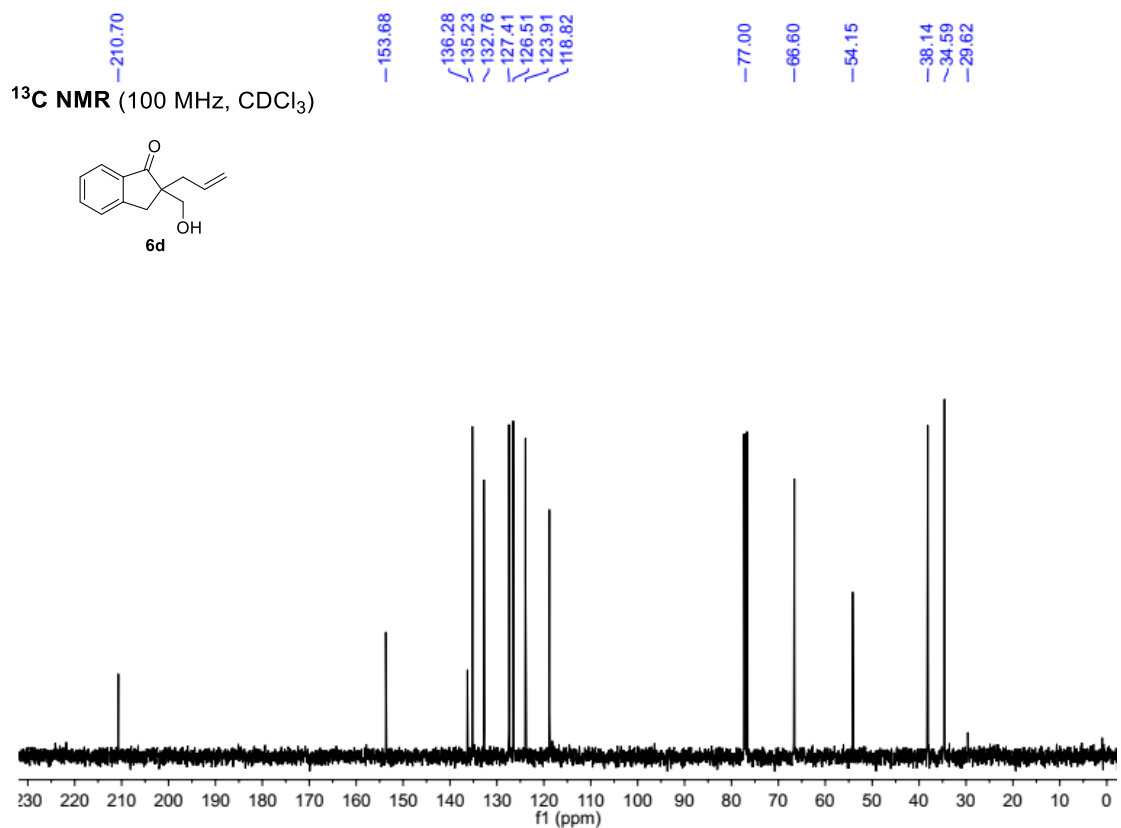
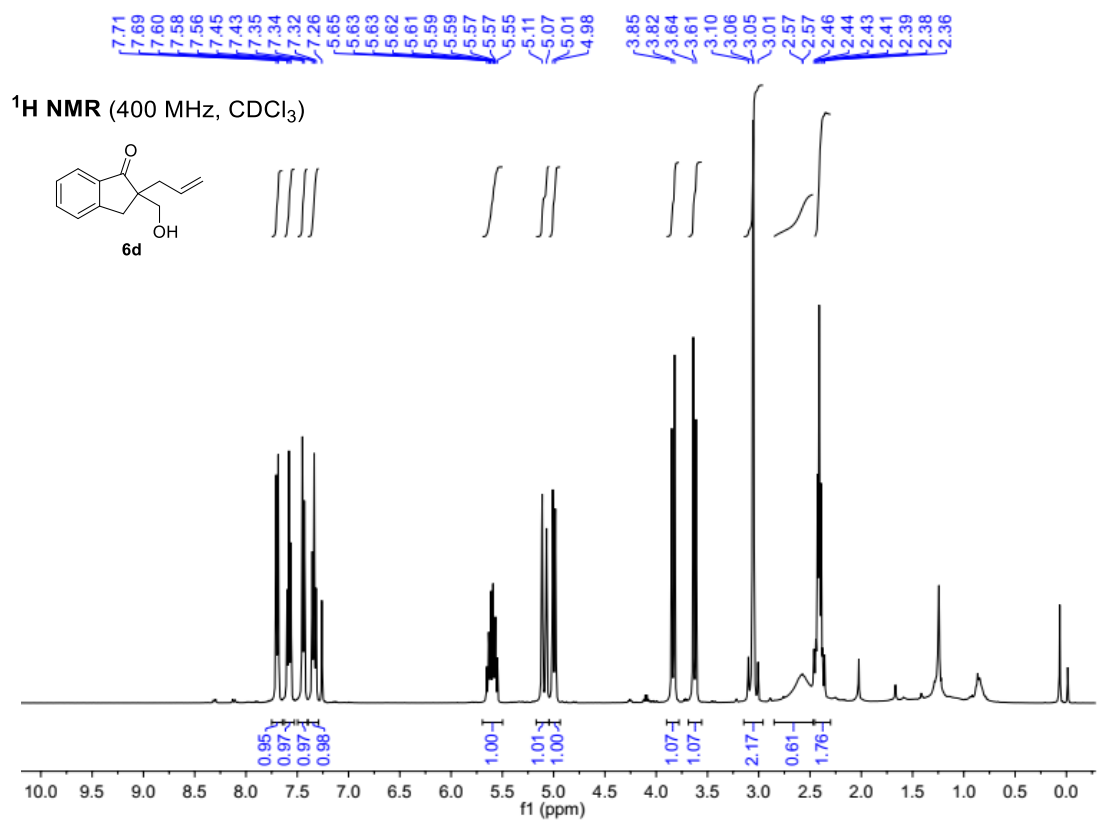


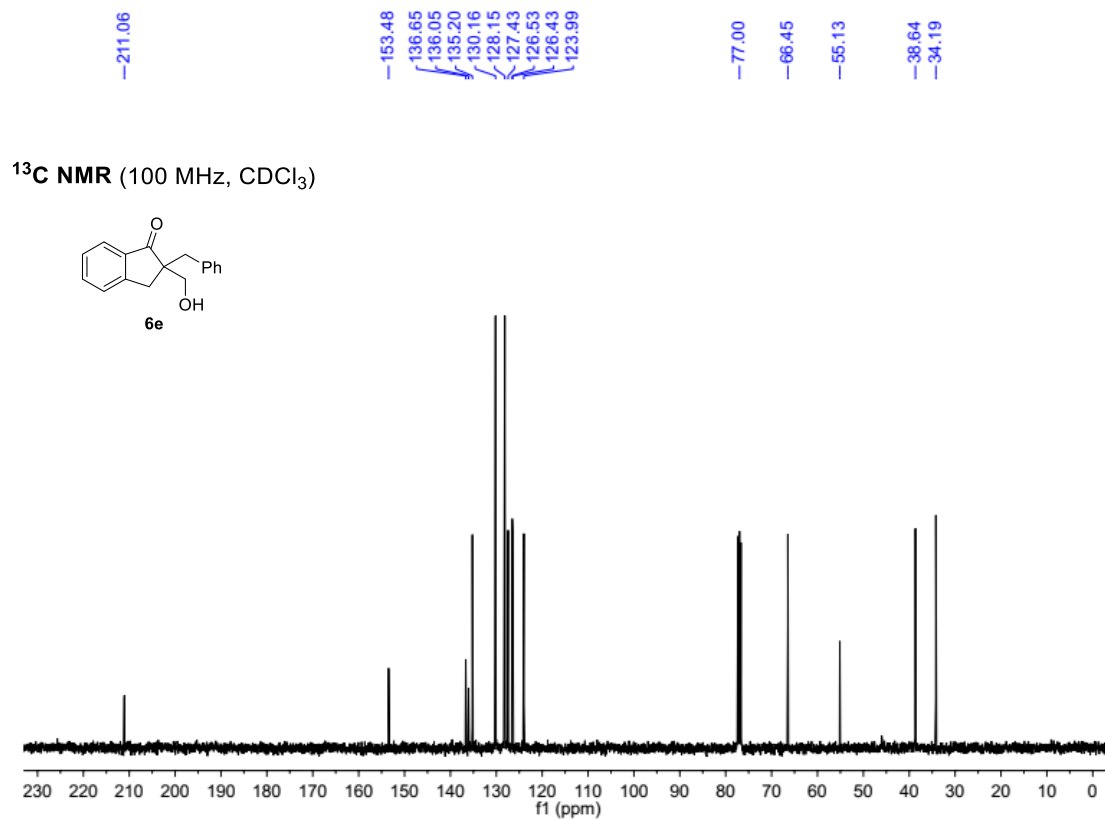
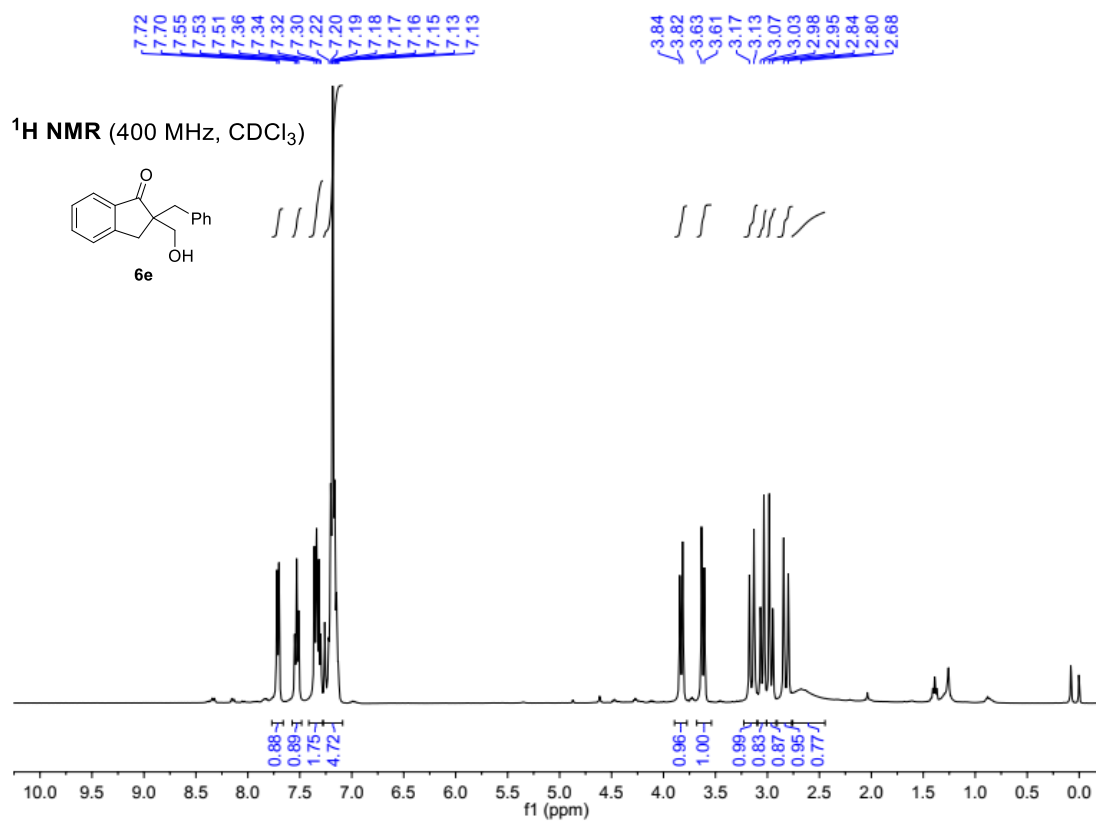


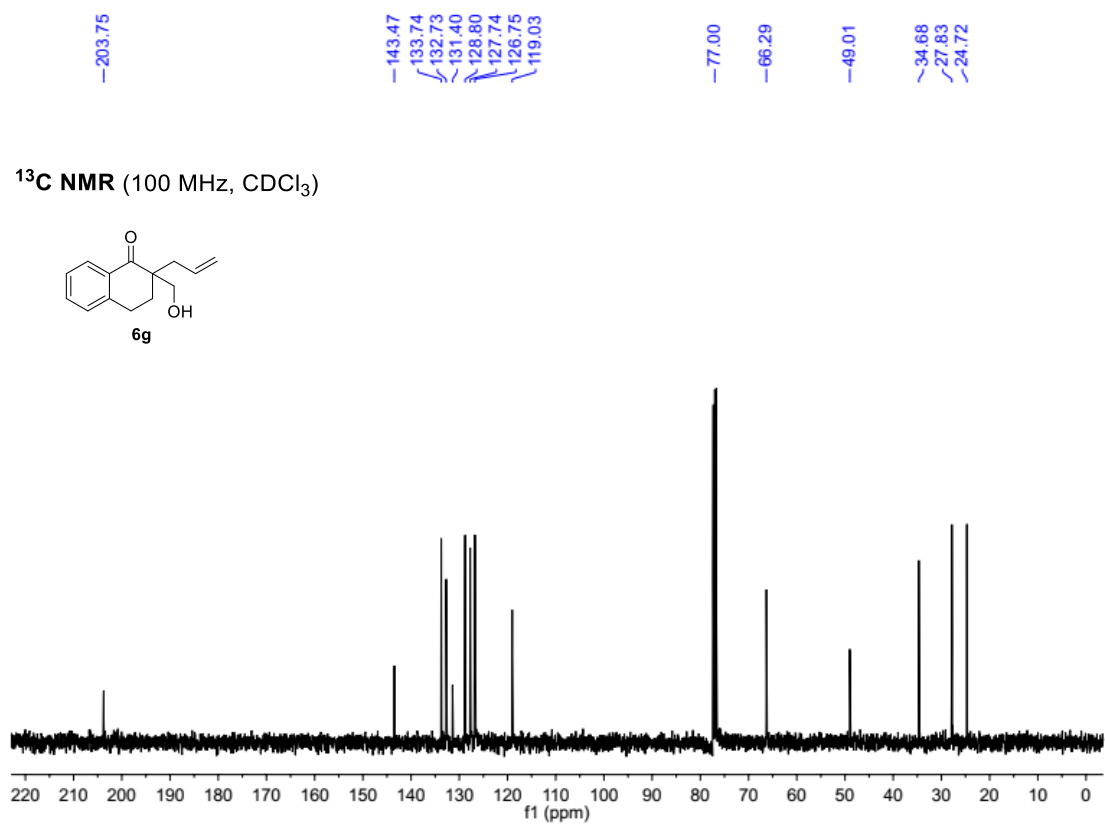
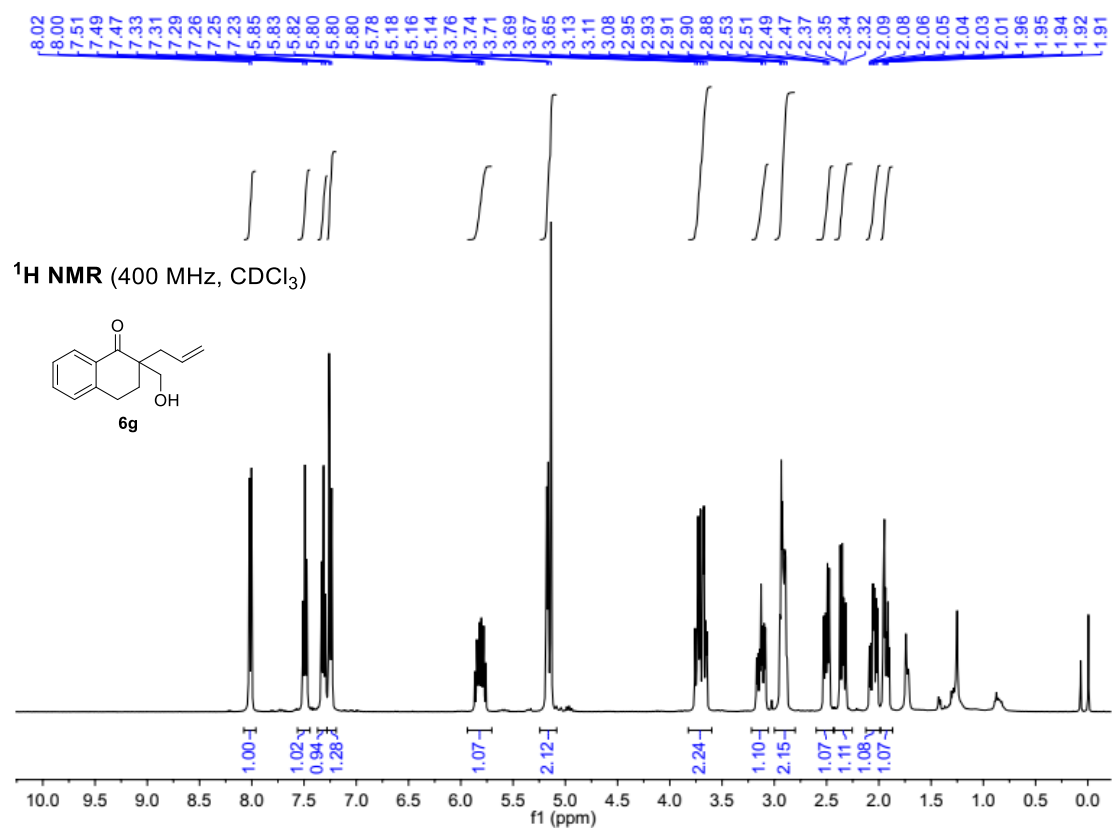


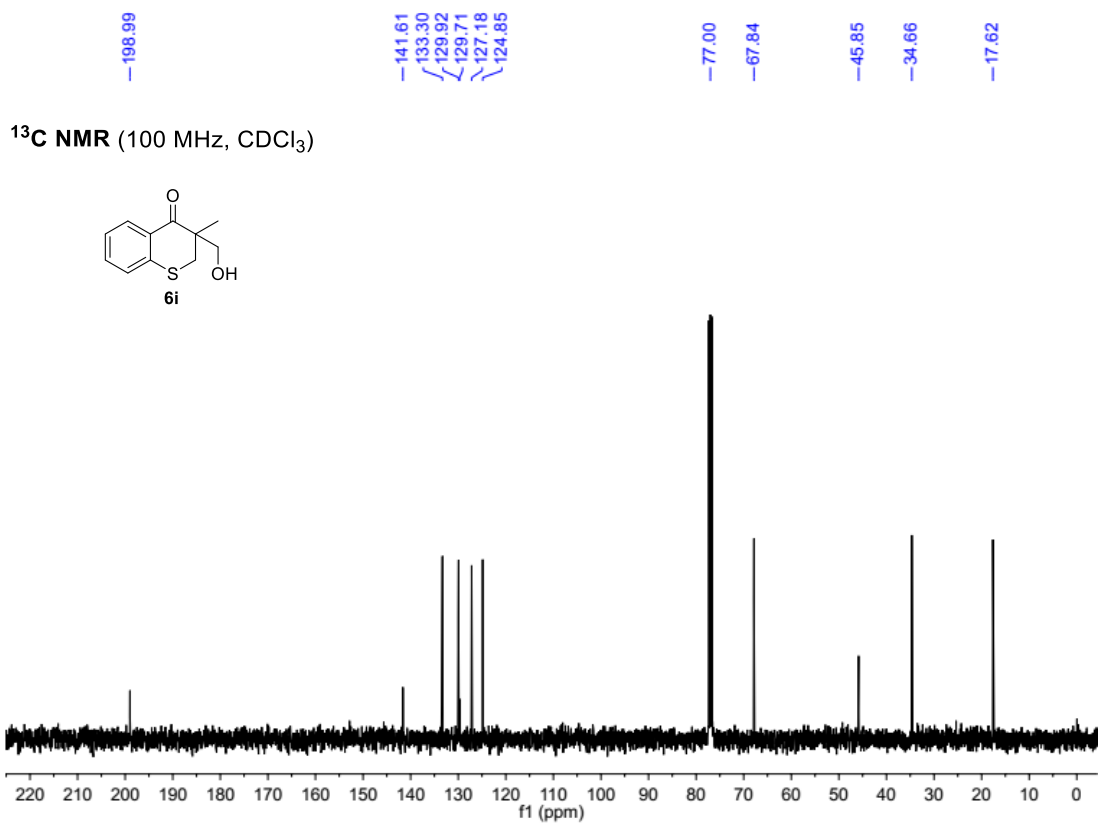
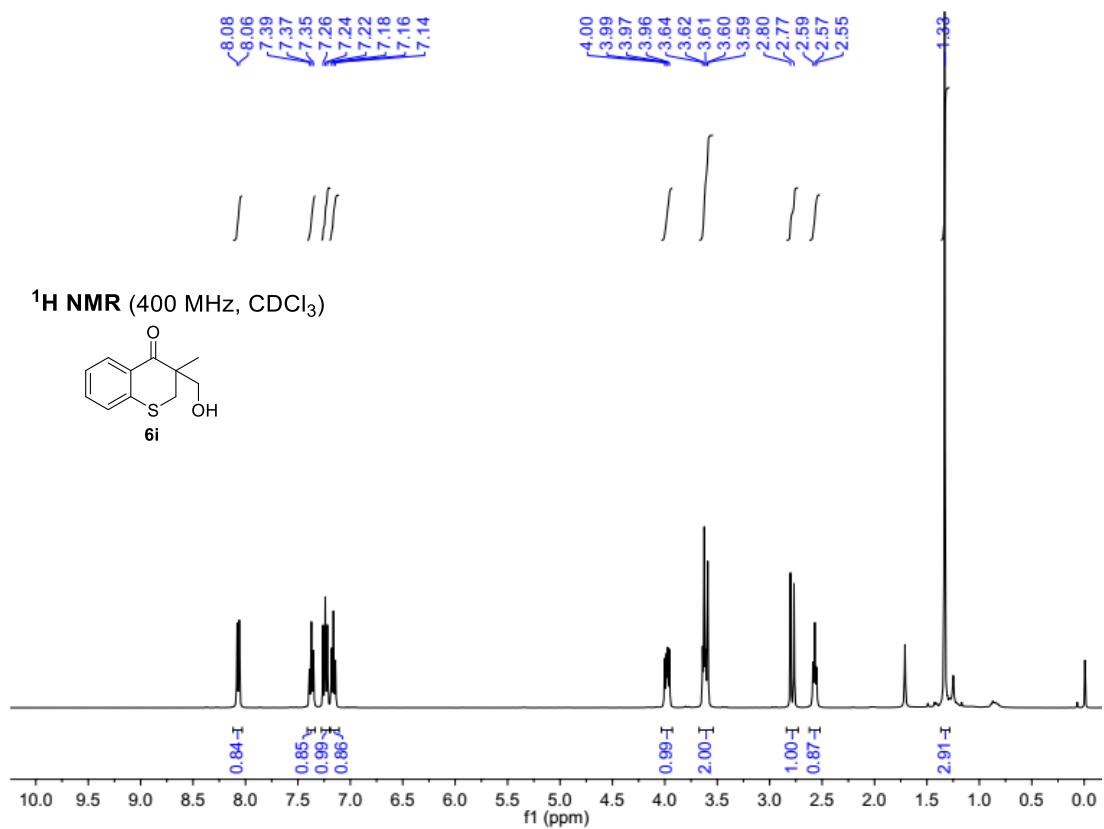




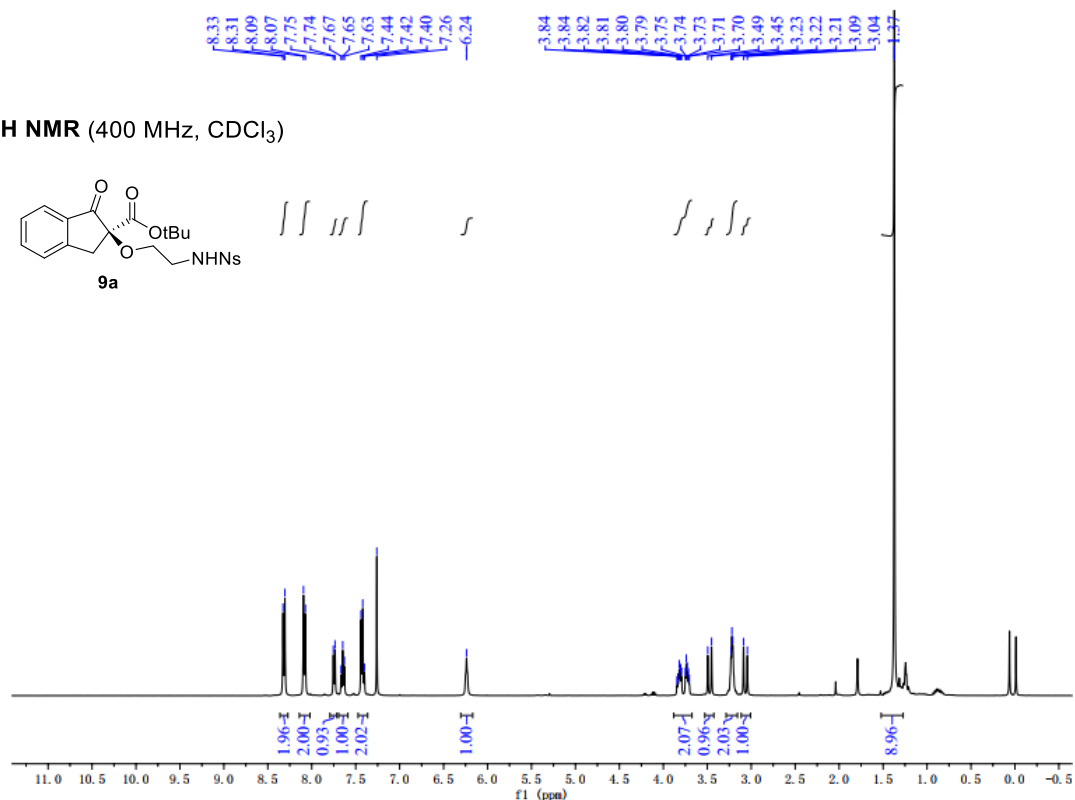
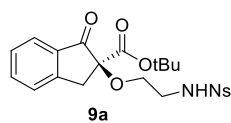






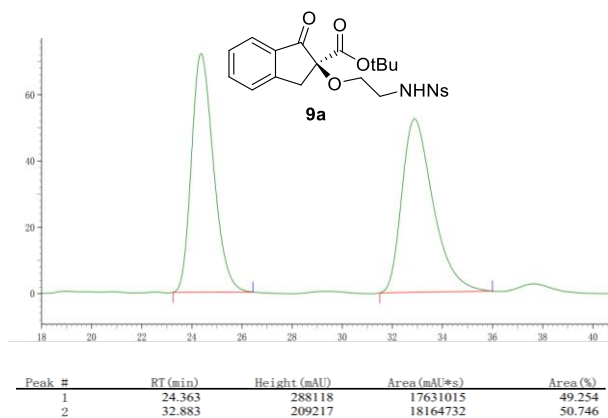


¹H NMR (400 MHz, CDCl₃)

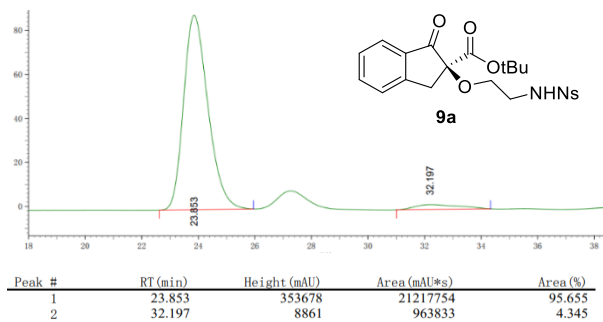


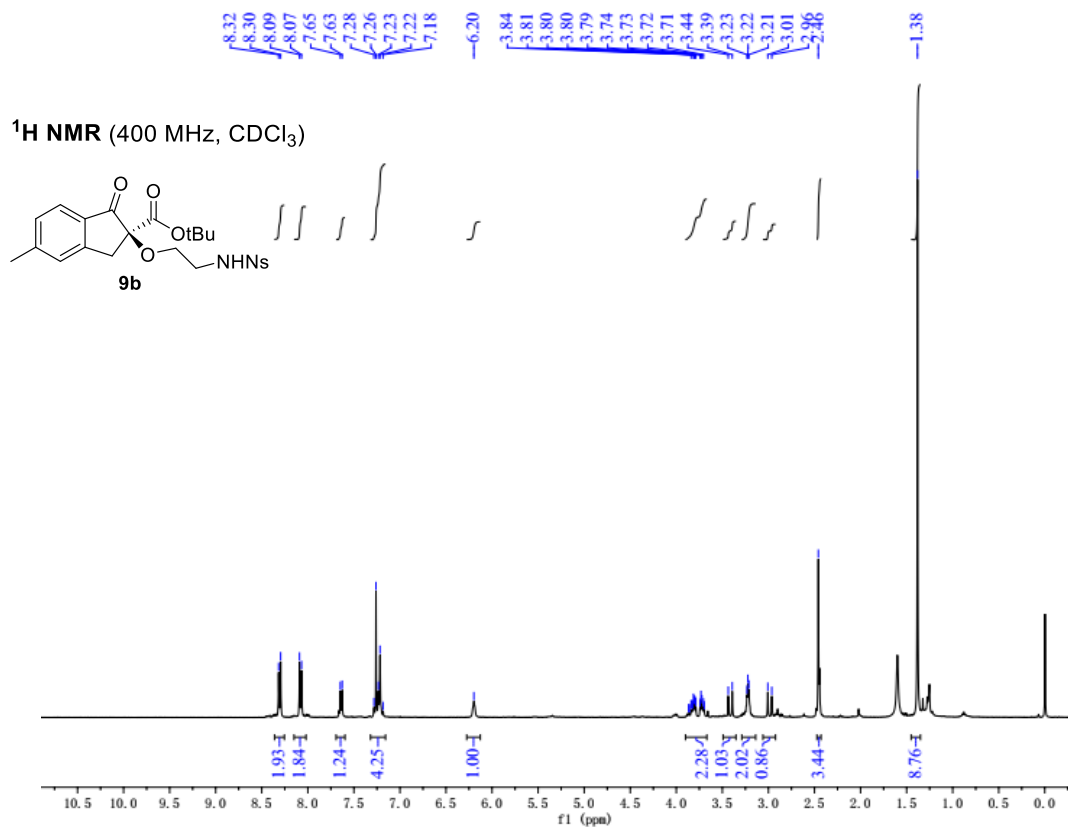
HPLC spectrum of compound **9a**:

Racemate:



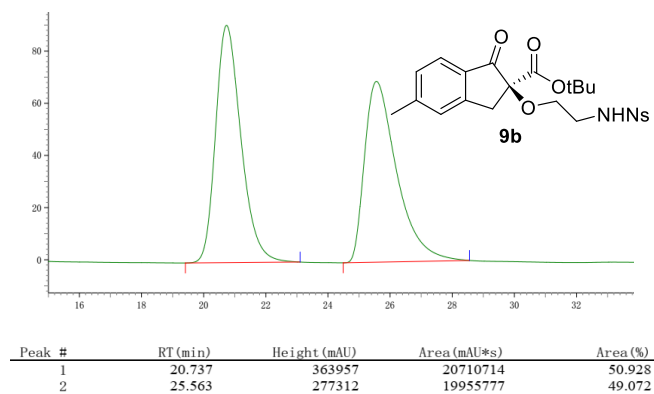
Enantioselective:



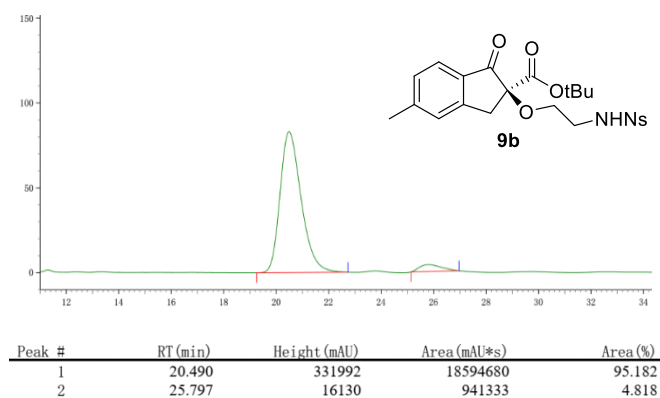


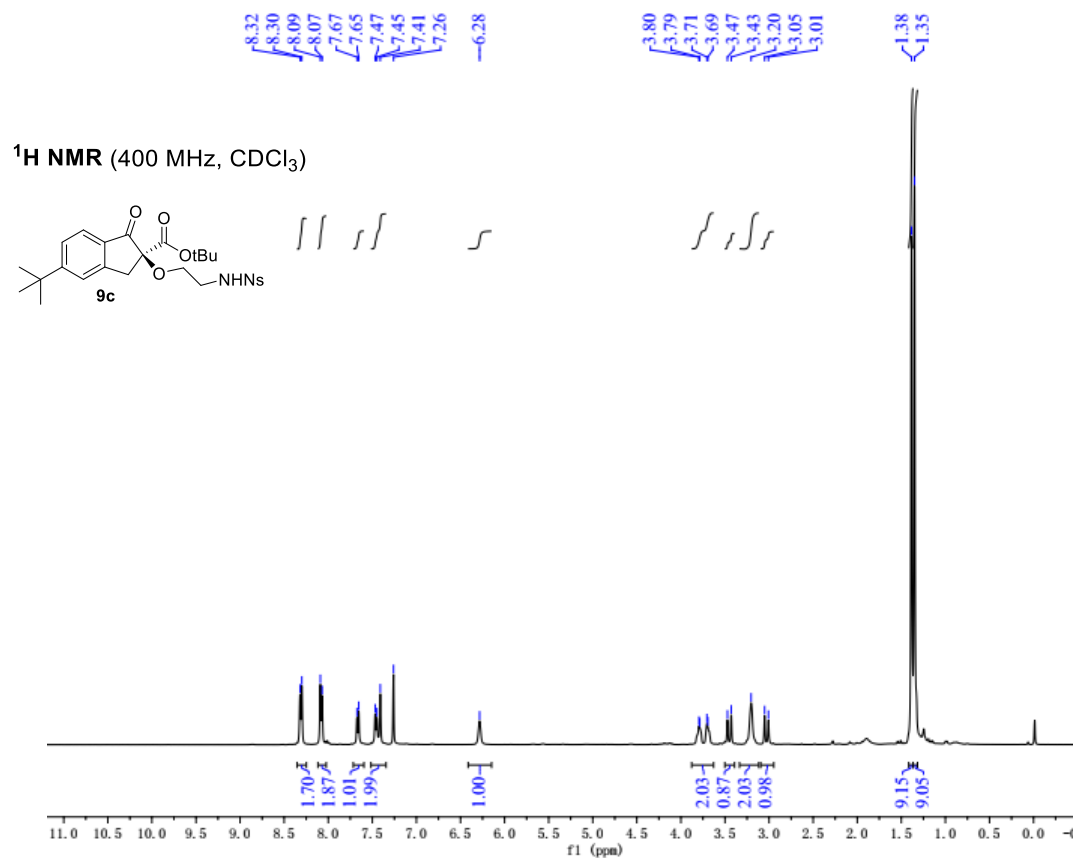
HPLC spectrum of compound **9b**:

Racemate:



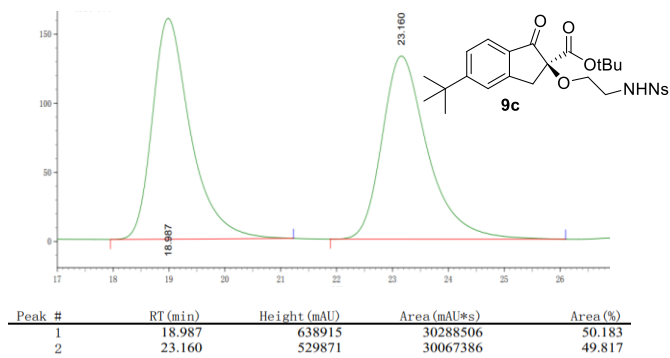
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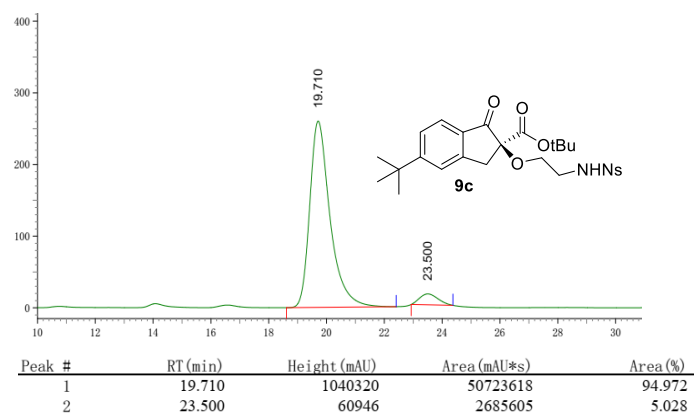


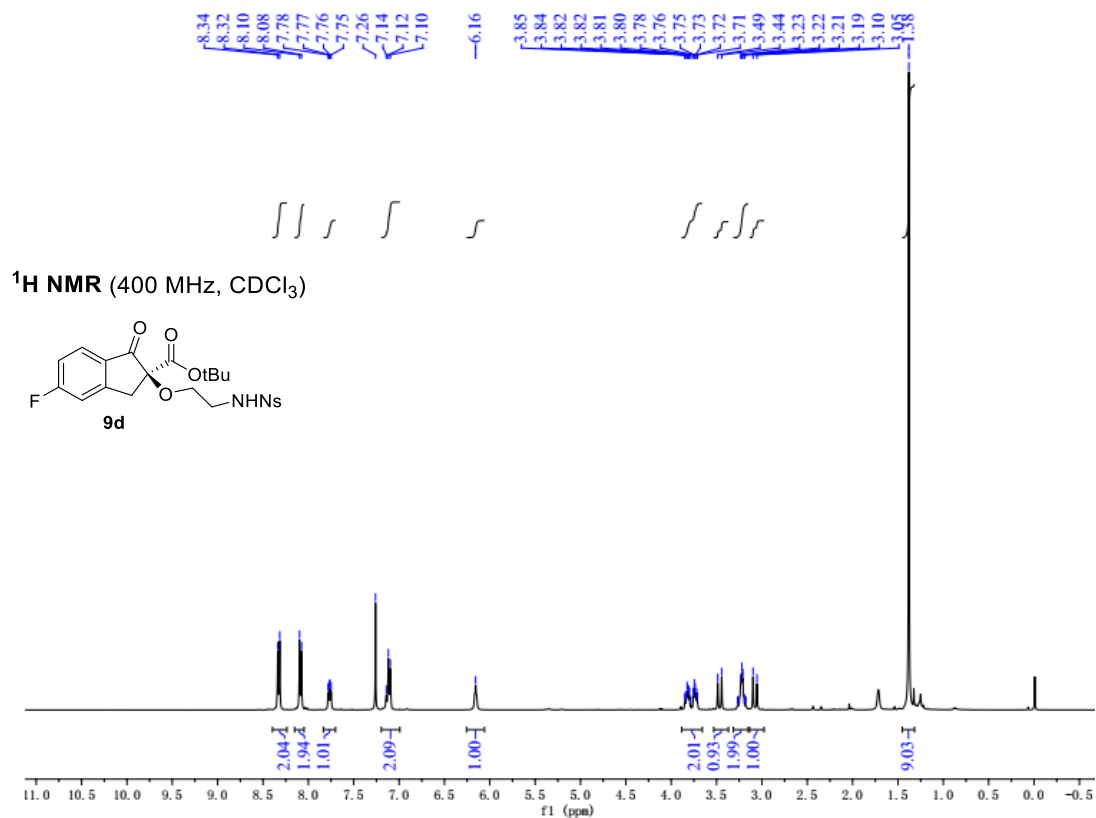
HPLC spectrum of compound **9c**:

Racemate:



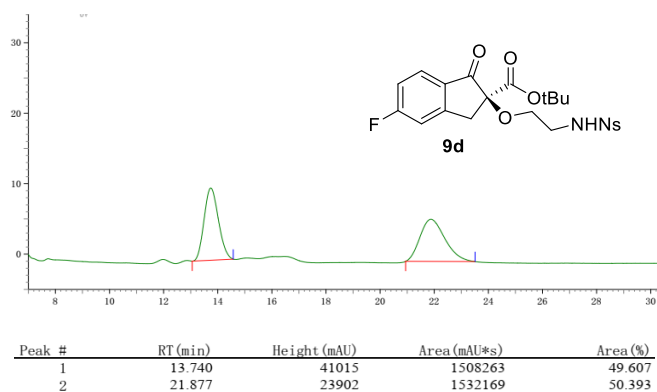
Enantioselective:



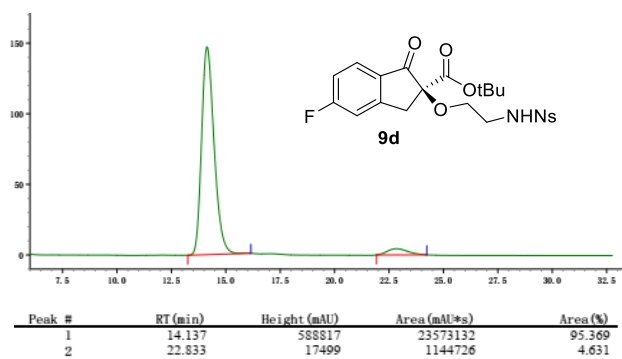


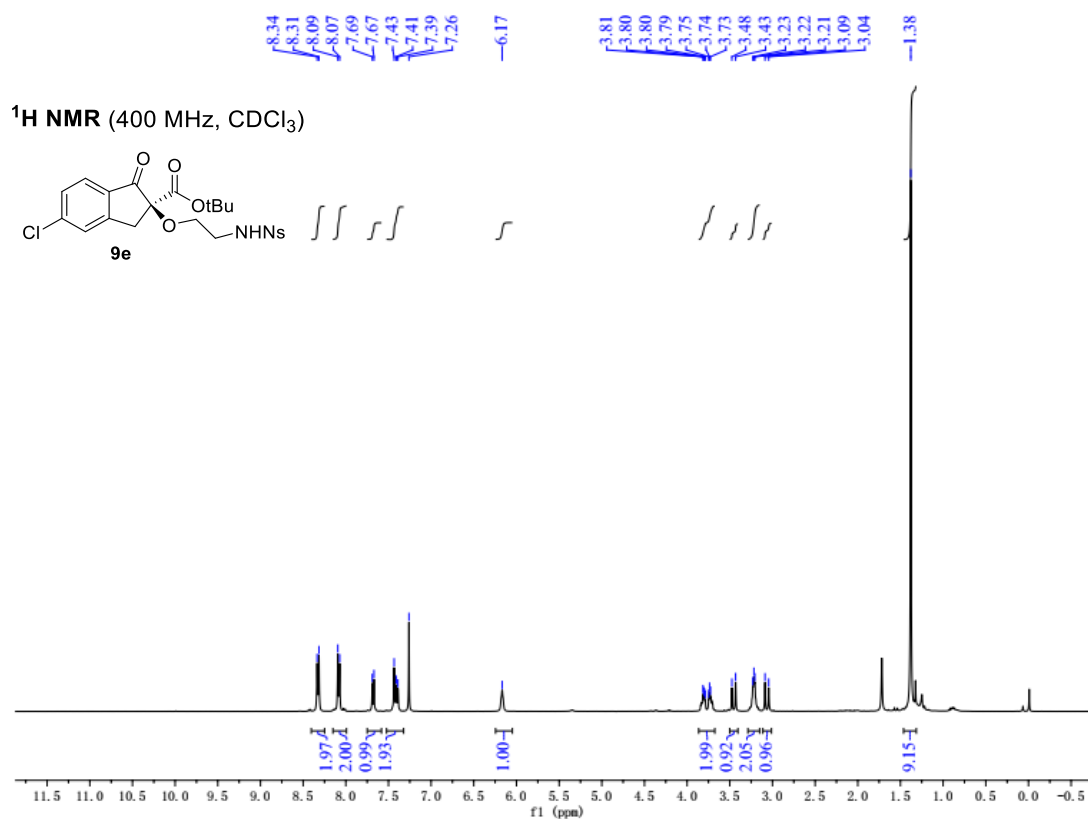
HPLC spectrum of compound **9d**:

Racemate:



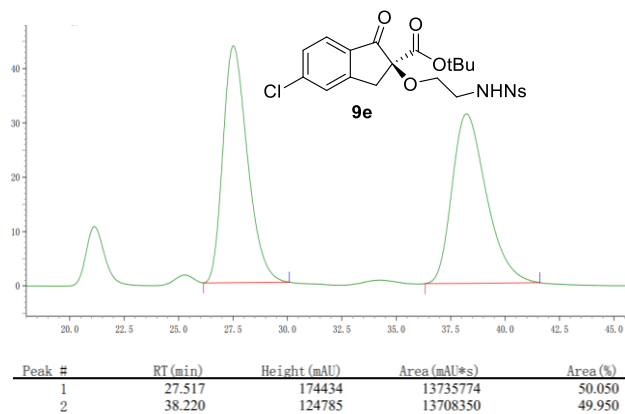
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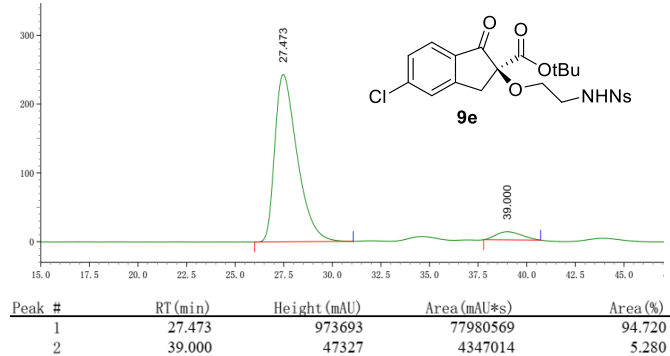


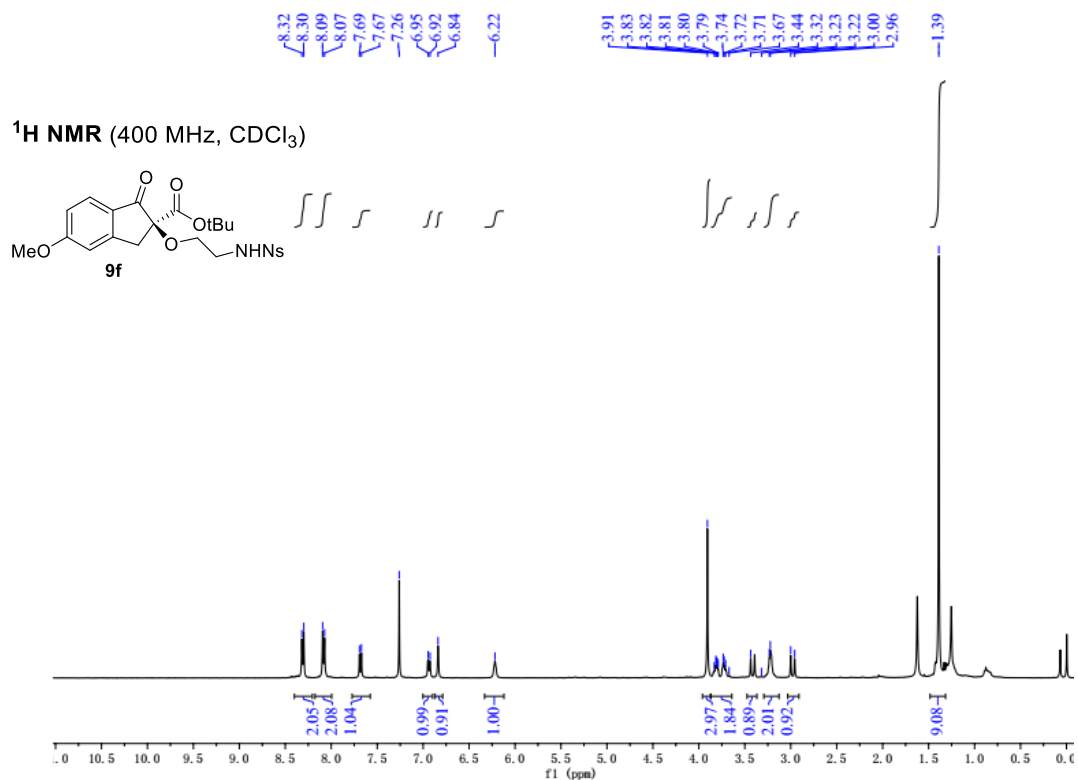
HPLC spectrum of compound **9e**:

Racemate:



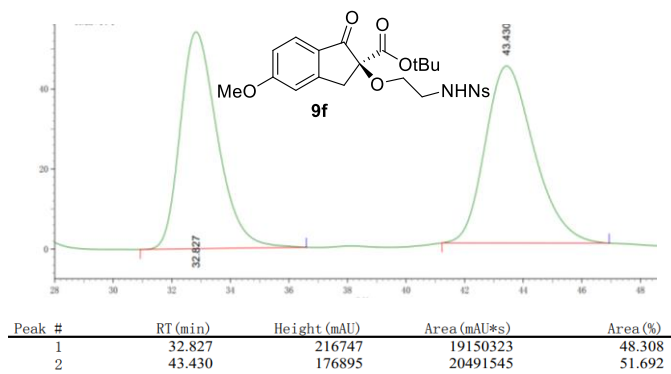
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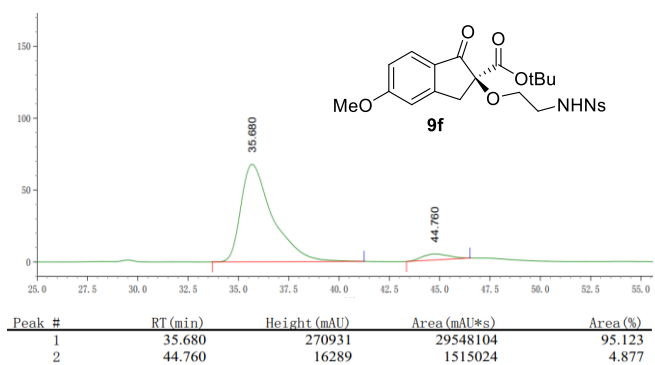


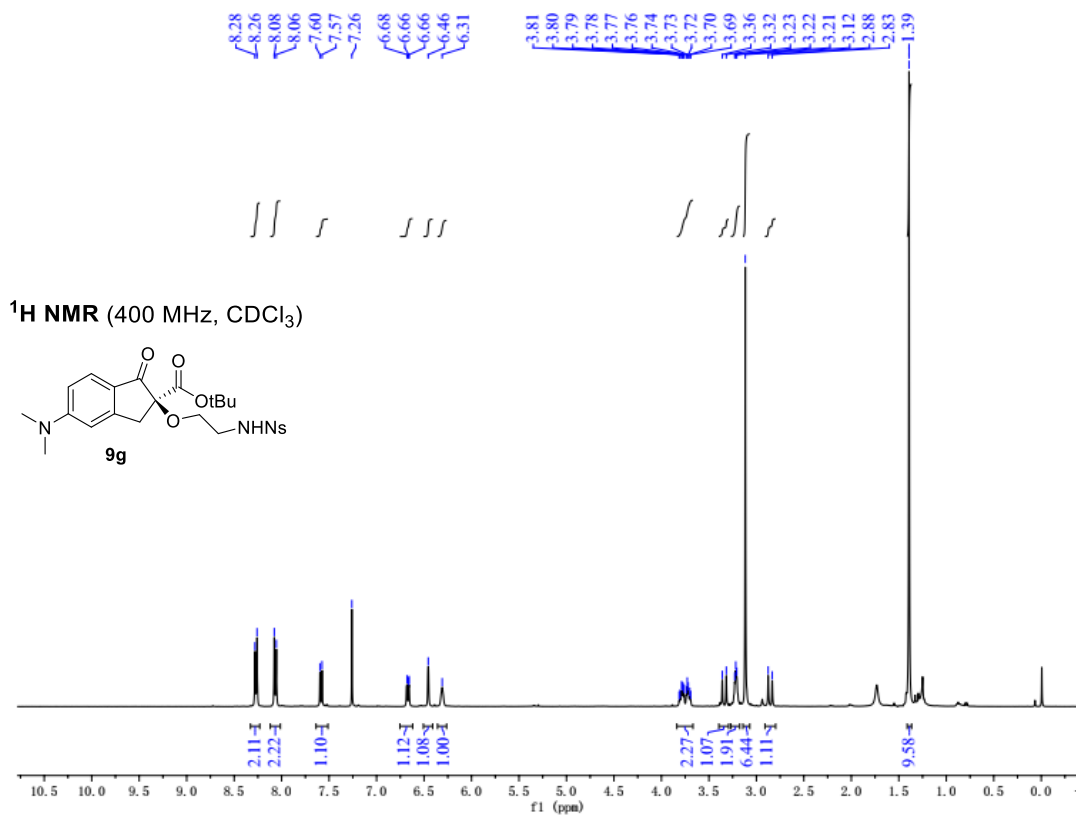
HPLC spectrum of compound **9f**:

Racemate:



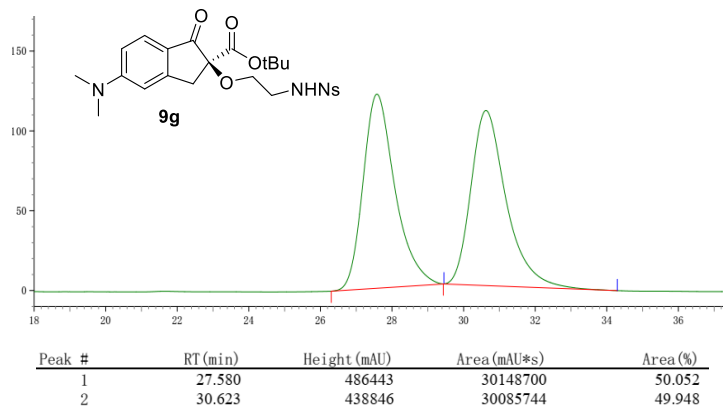
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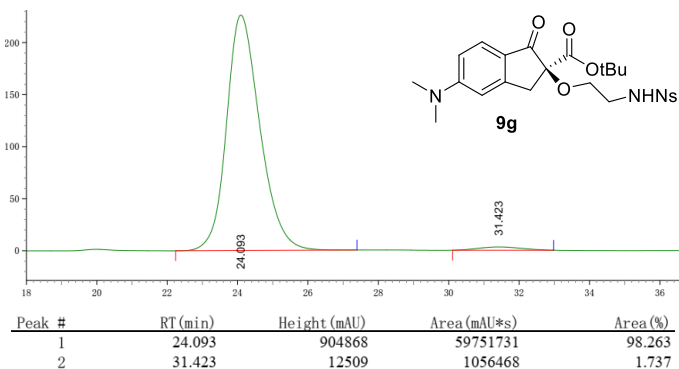


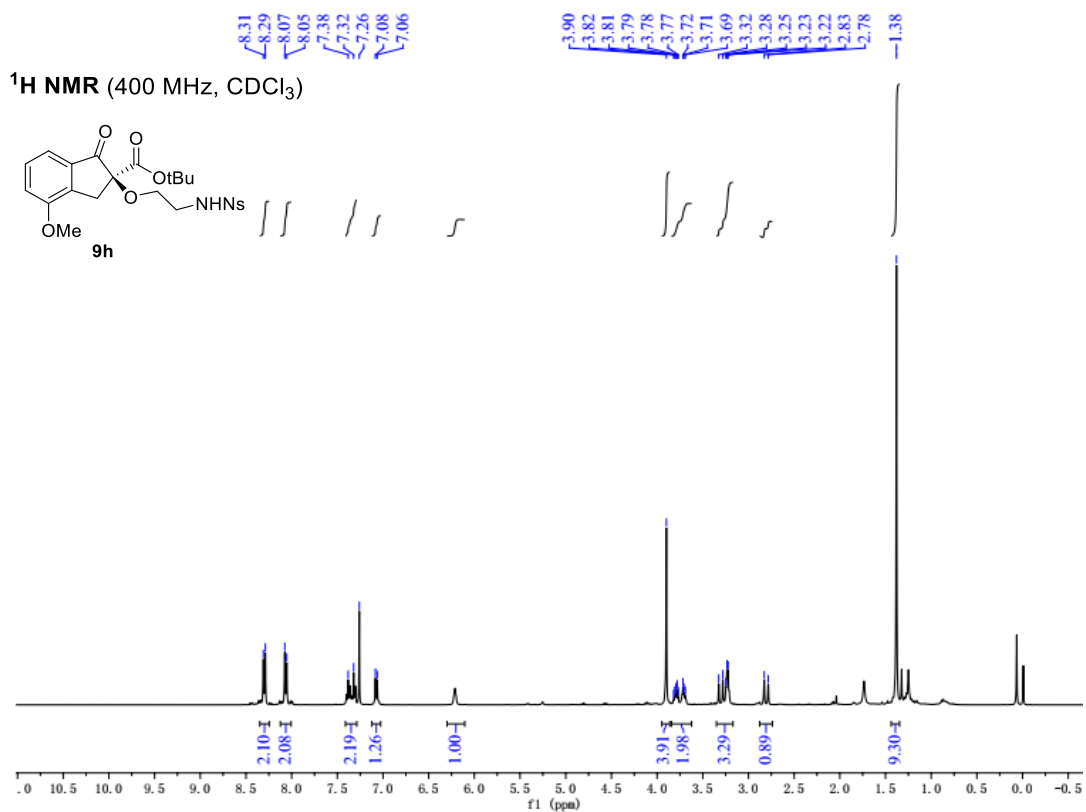
HPLC spectrum of compound **9g**:

Racemate:



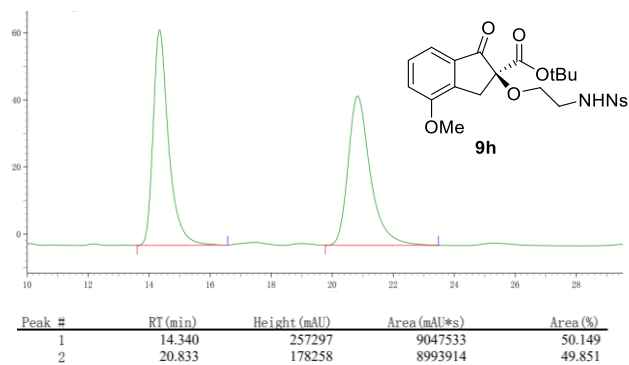
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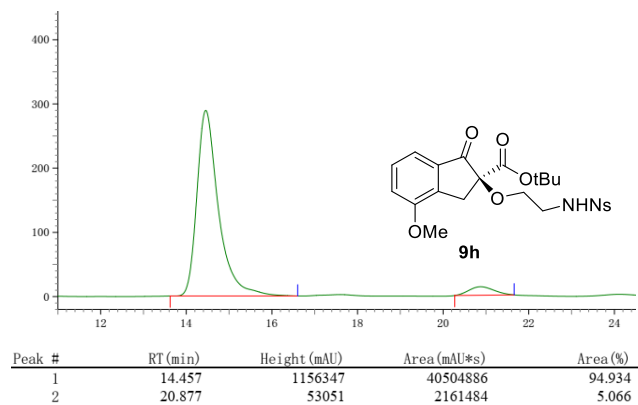


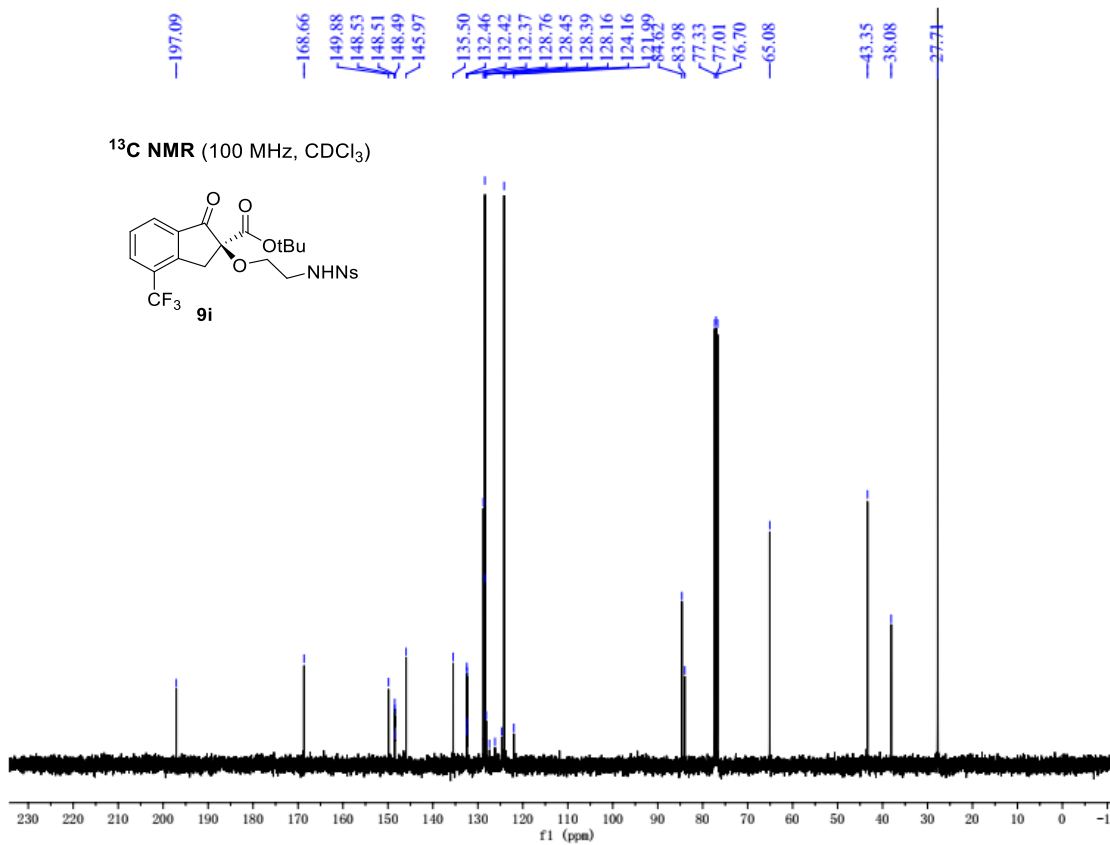
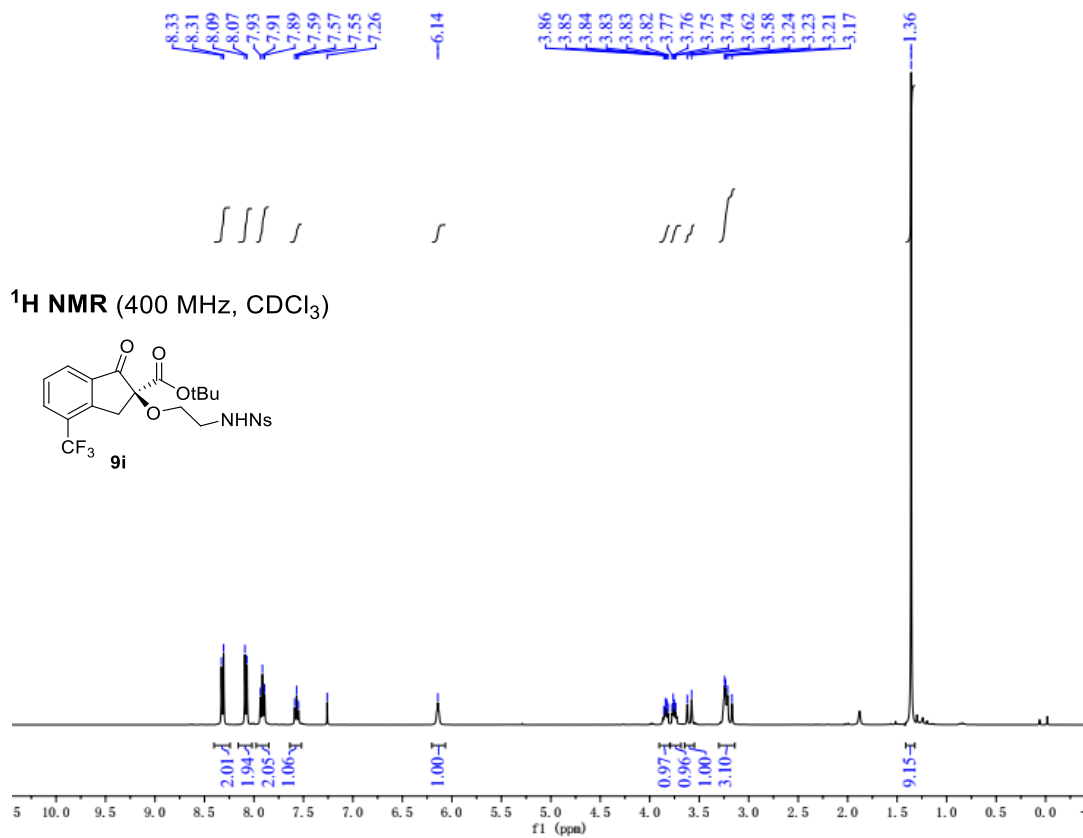
HPLC spectrum of compound **9h**:

Racemate:



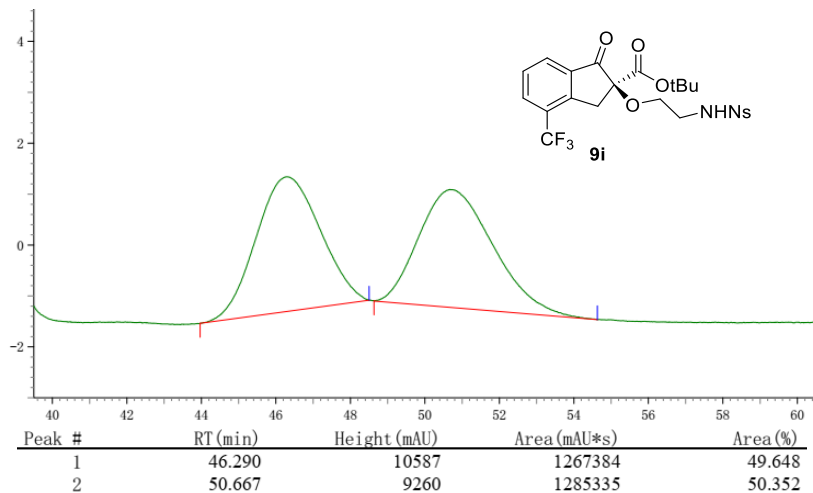
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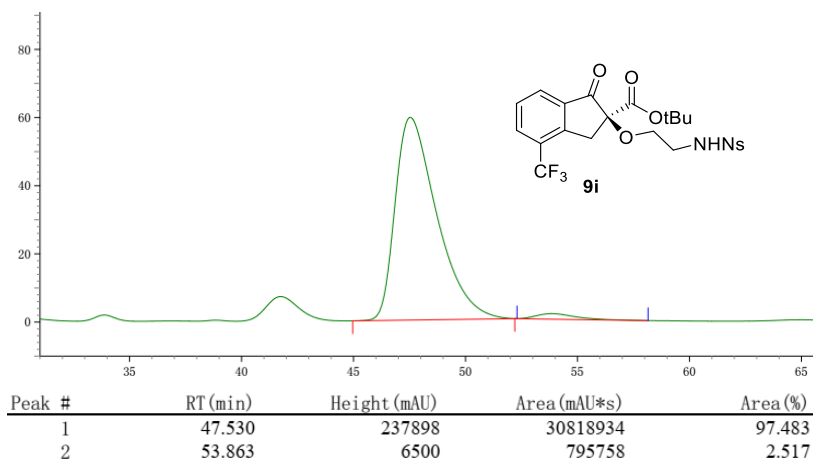


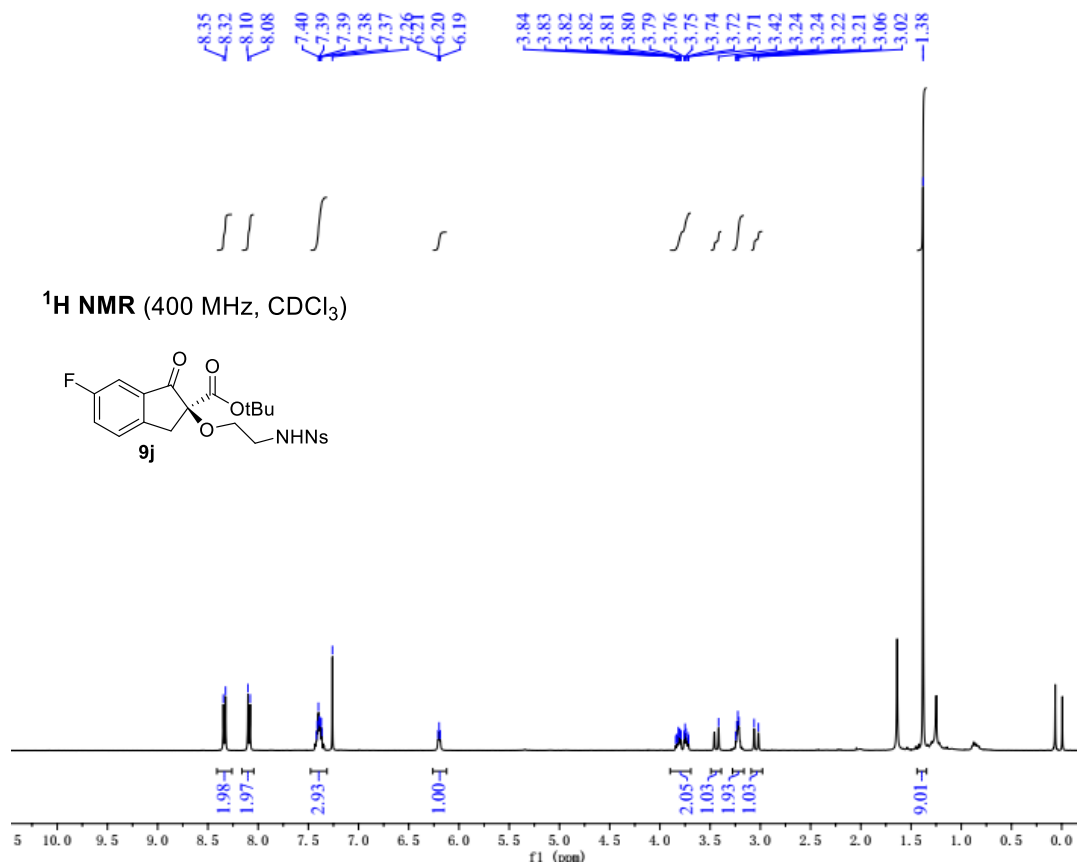
HPLC spectrum of compound **9i**:

Racemate:



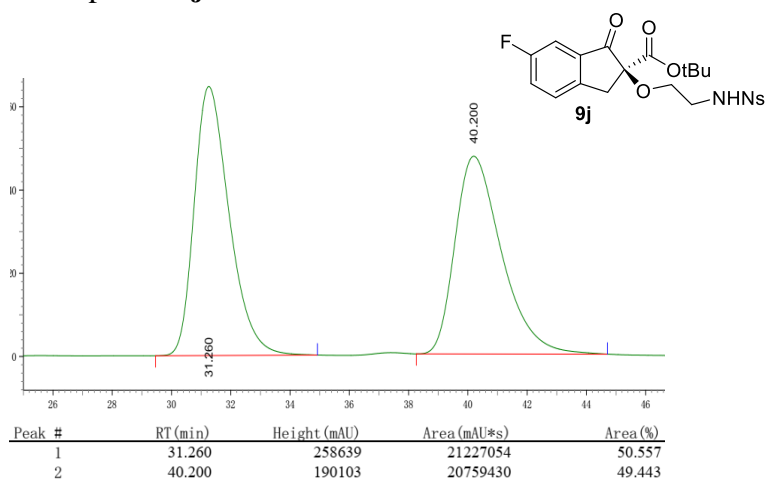
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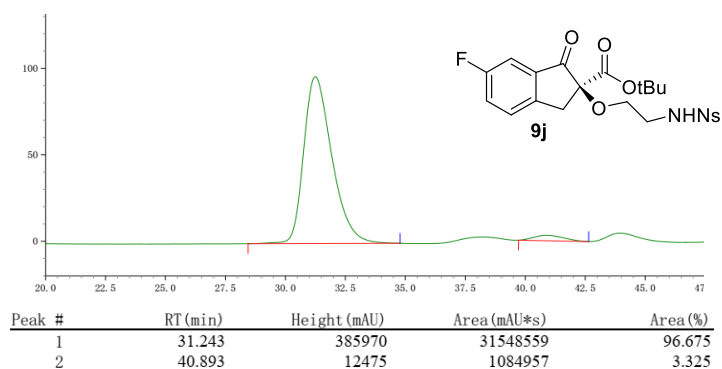


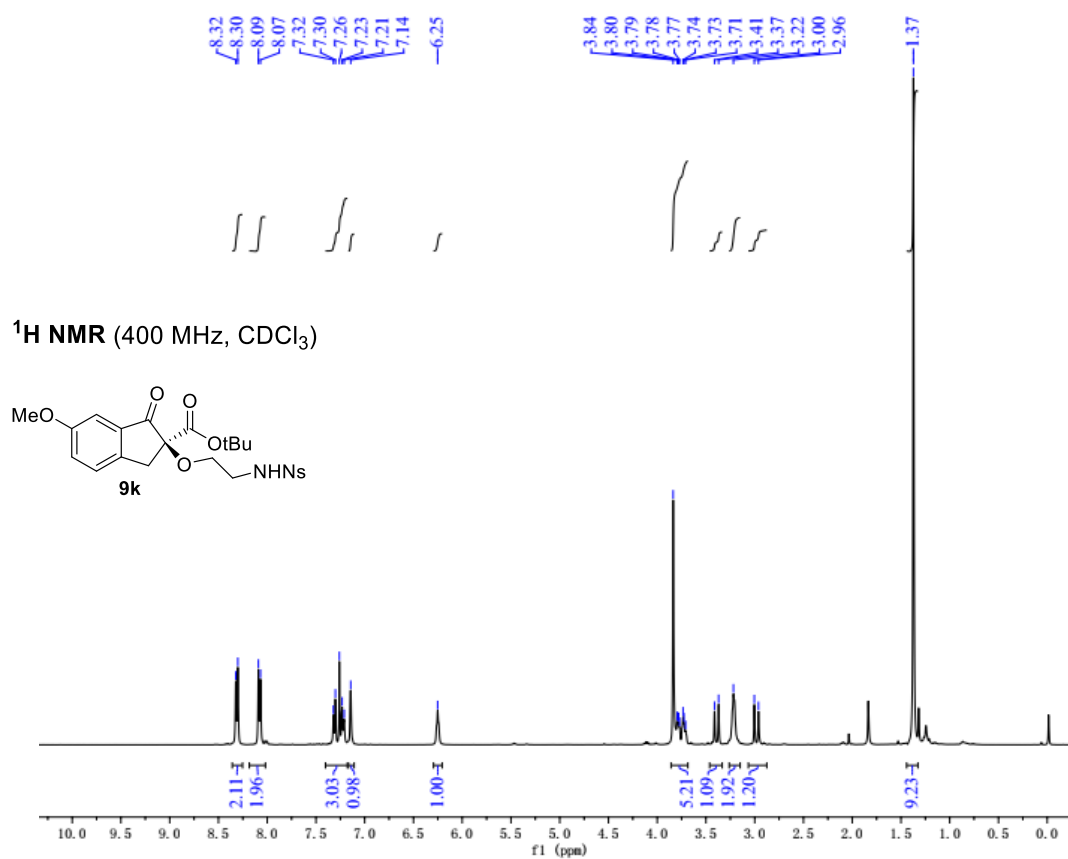
HPLC spectrum of compound **9j**:

Racemate:



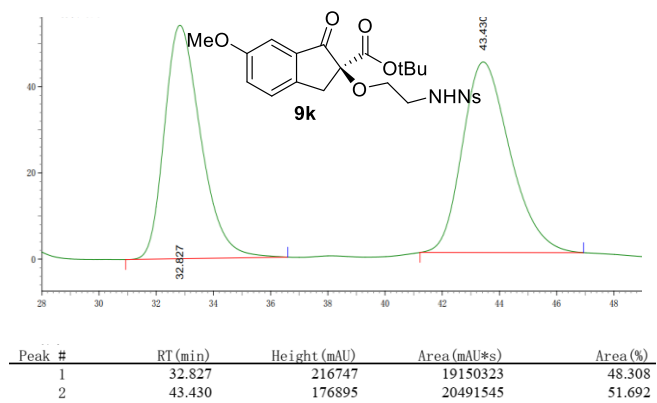
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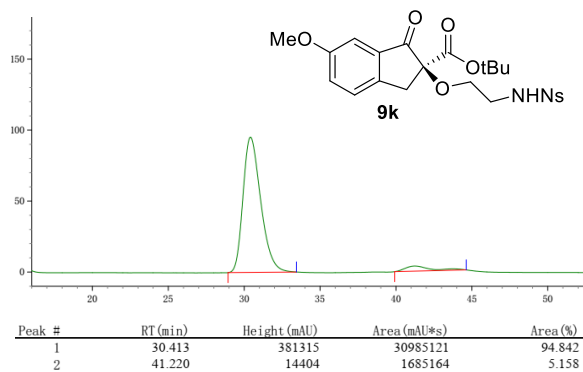


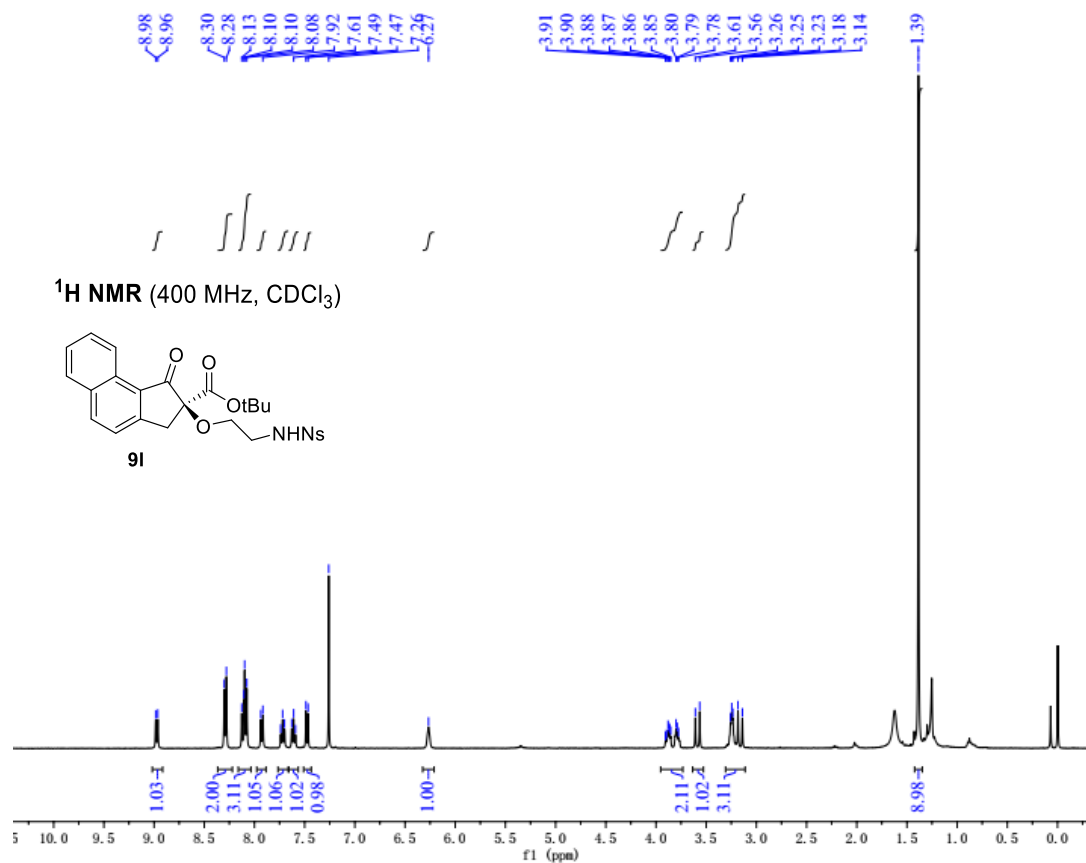
HPLC spectrum of compound **9k**:

Racemate:



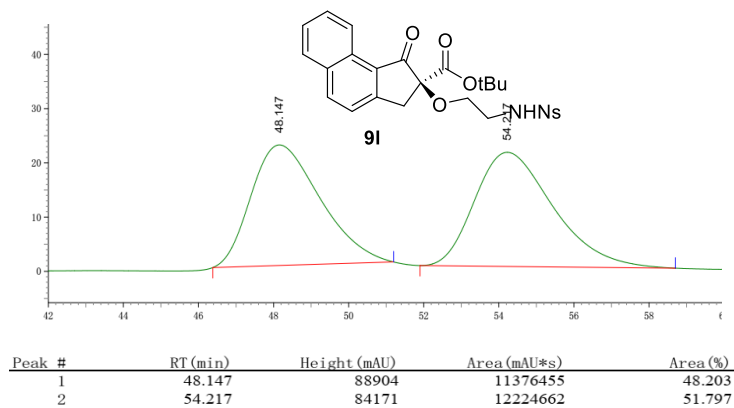
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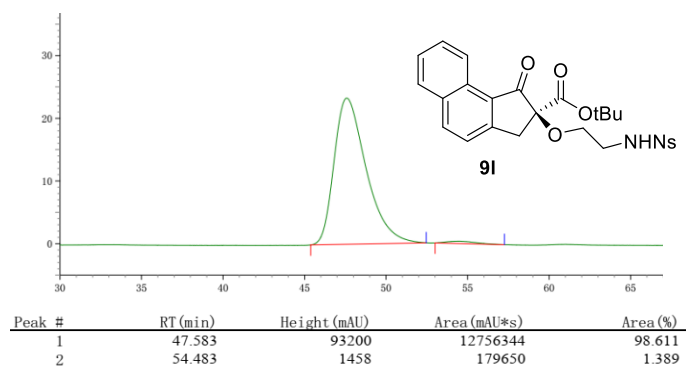


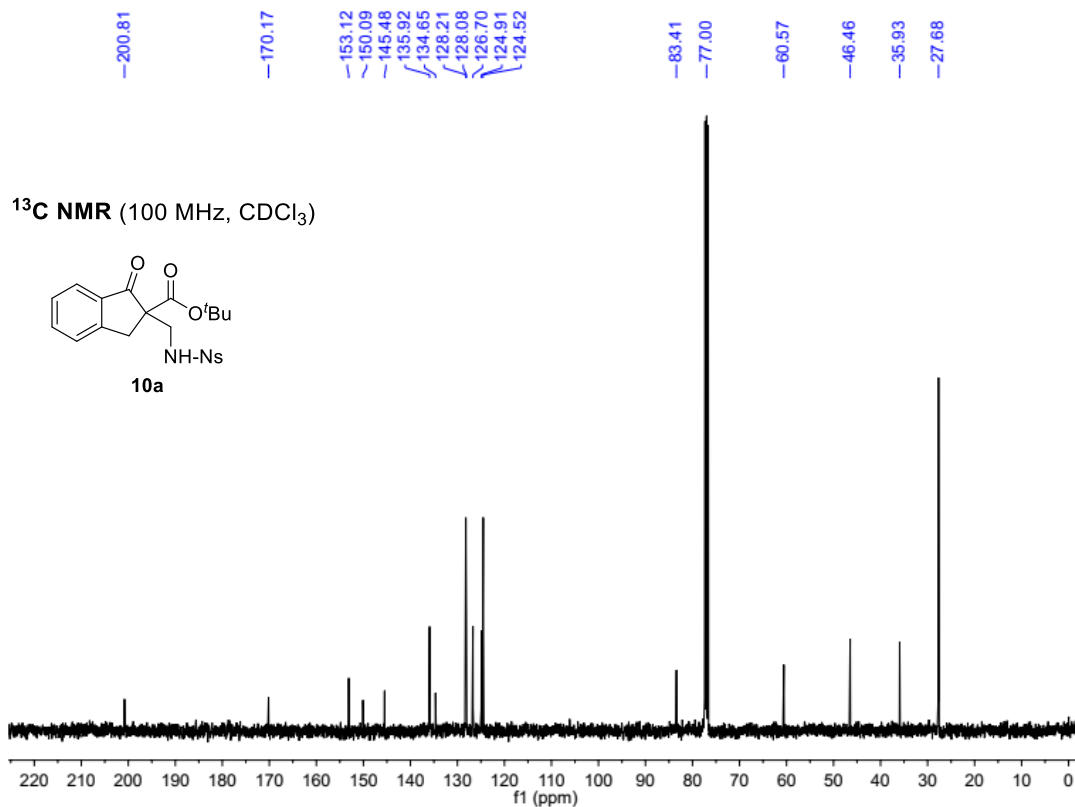
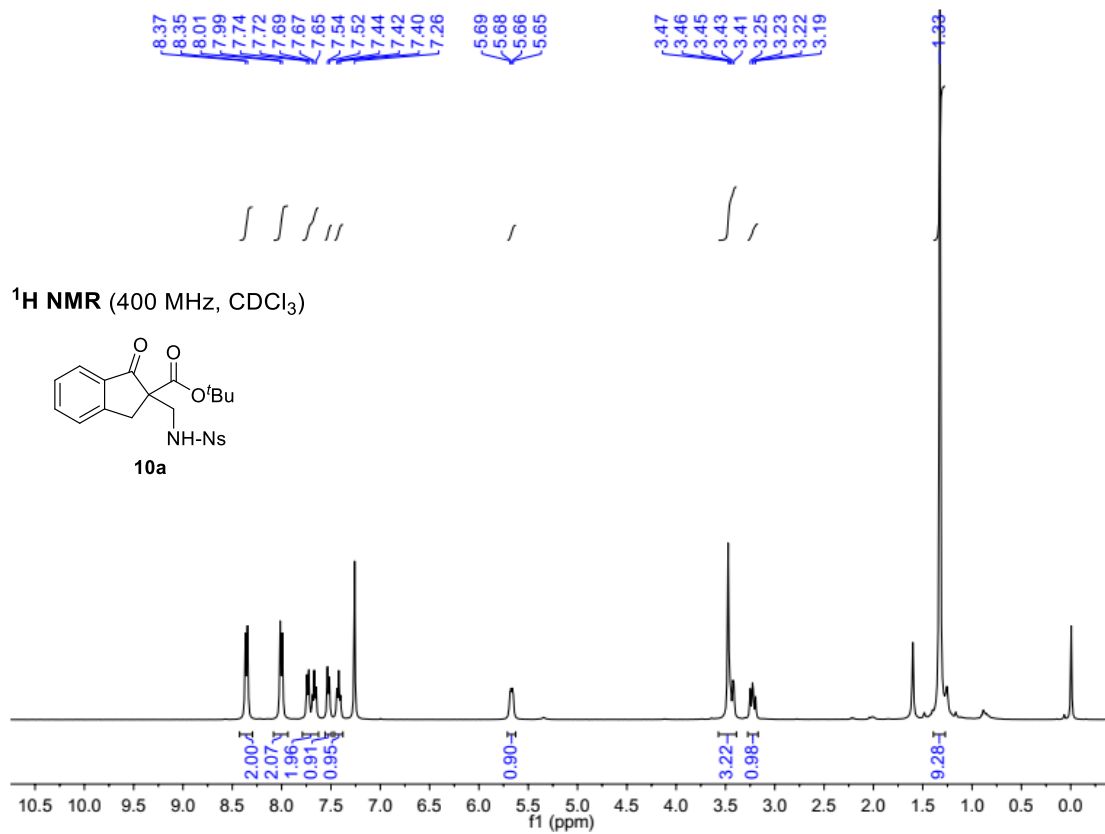
HPLC spectrum of compound **9l** :

Racemate:



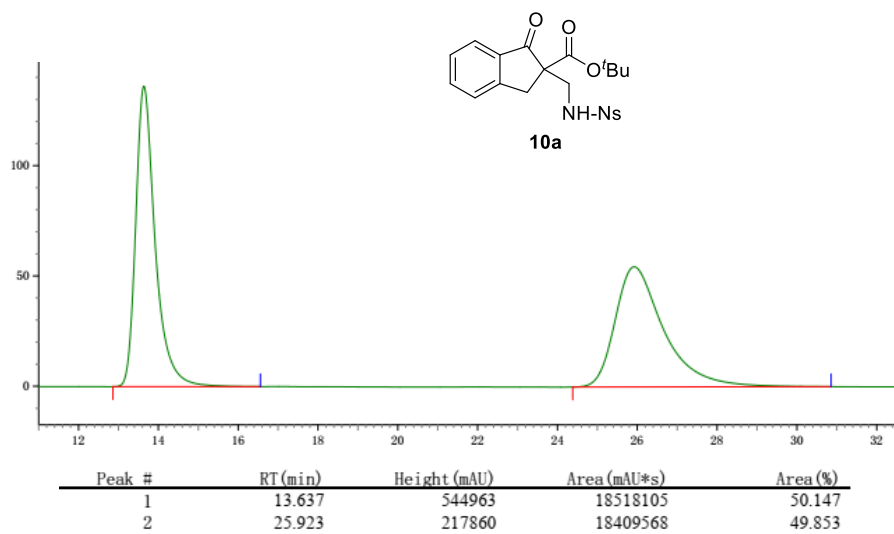
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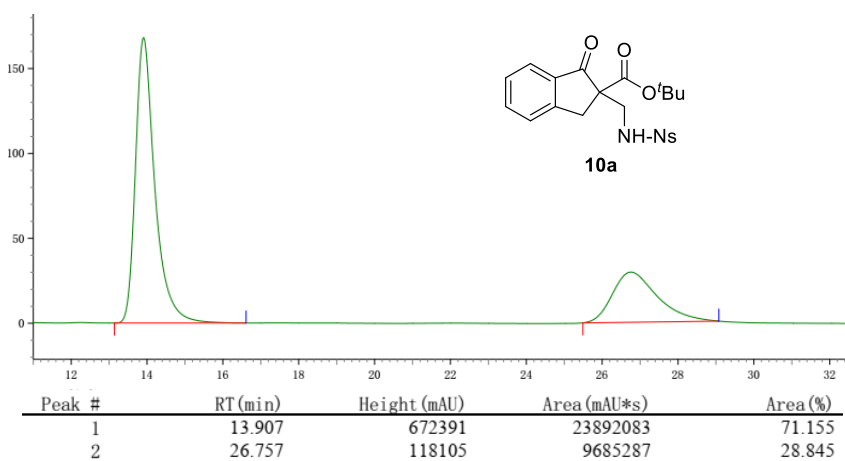


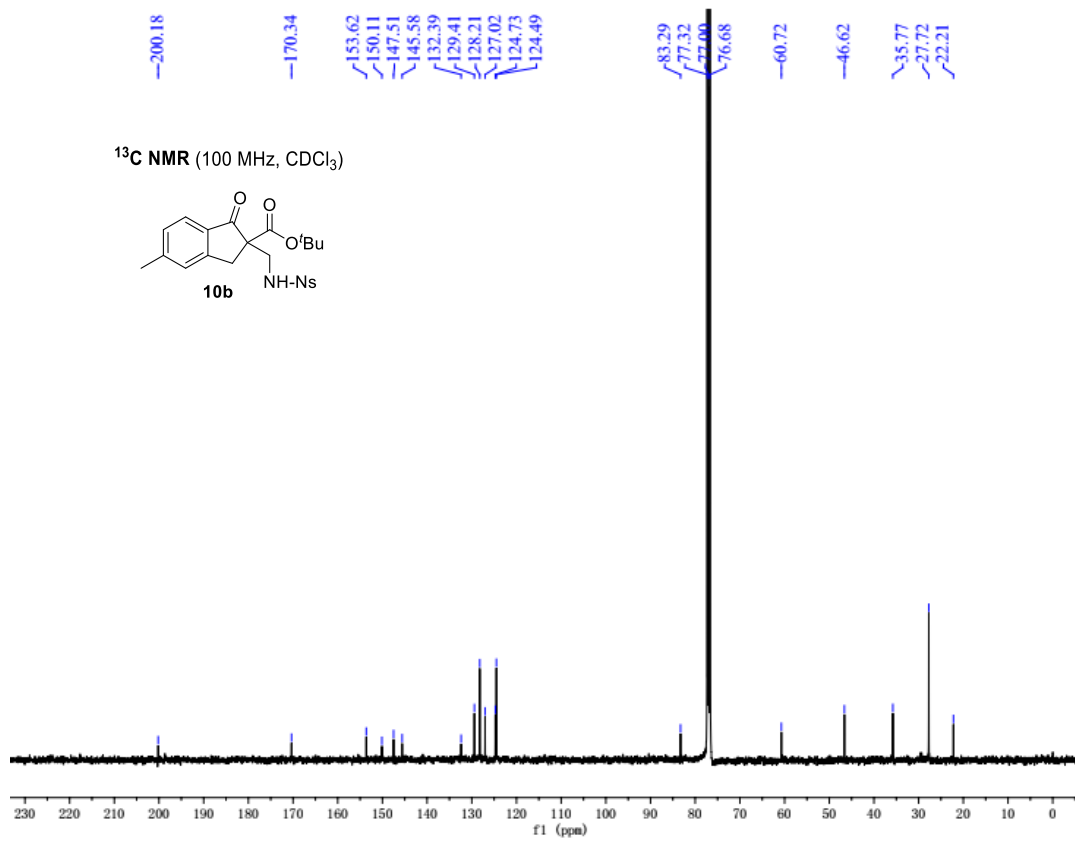
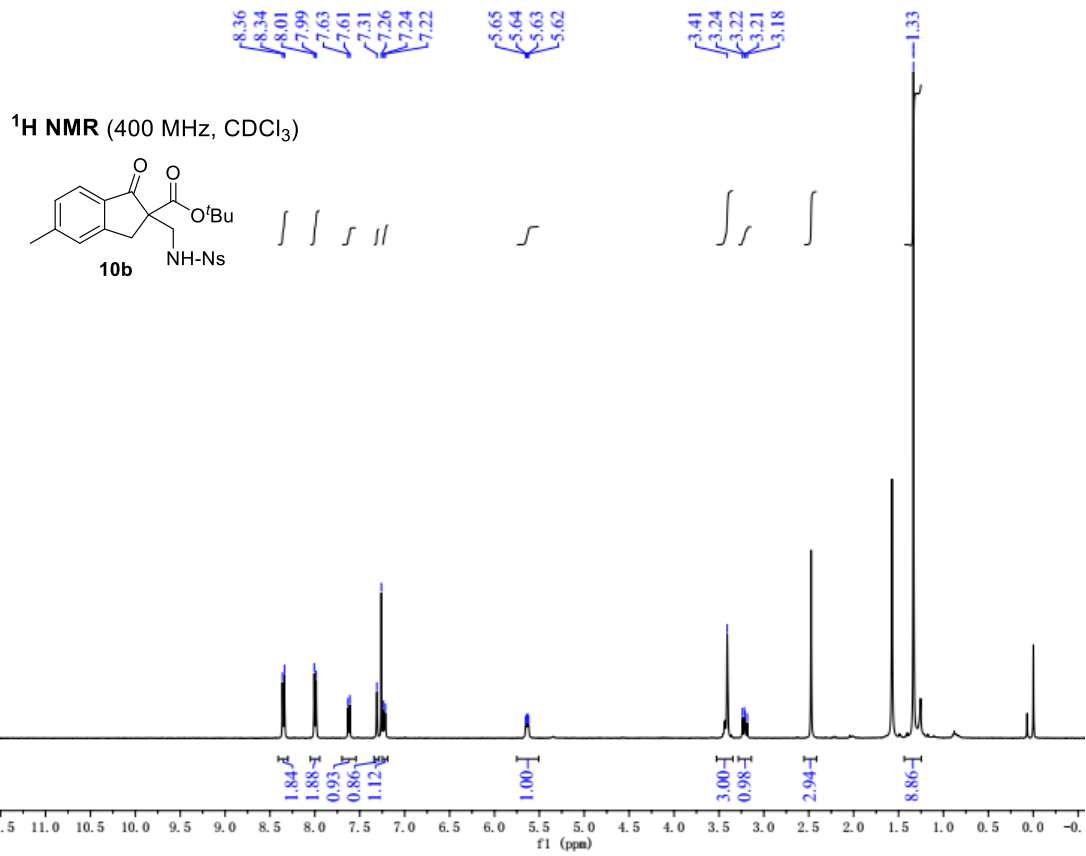
HPLC spectrum of compound **10a**:

Racemate:



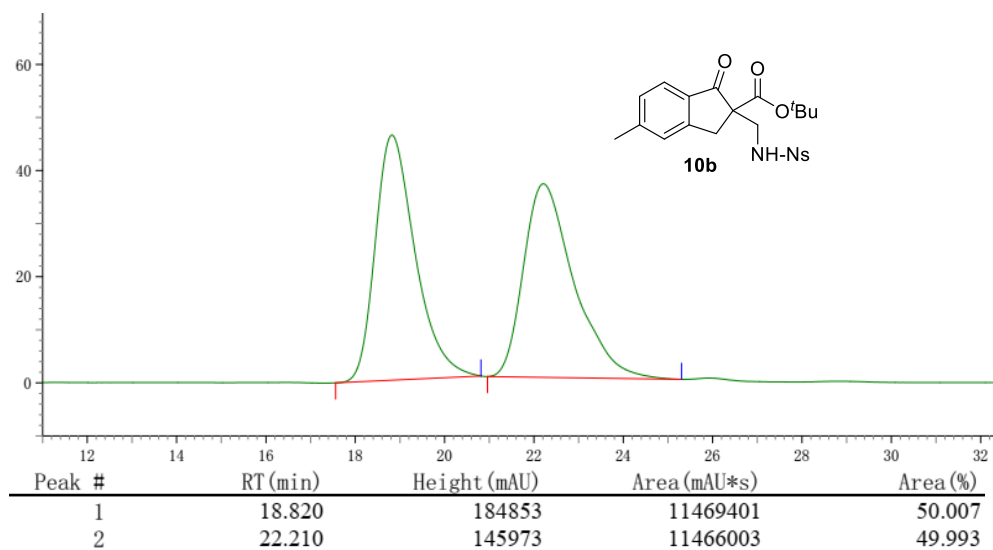
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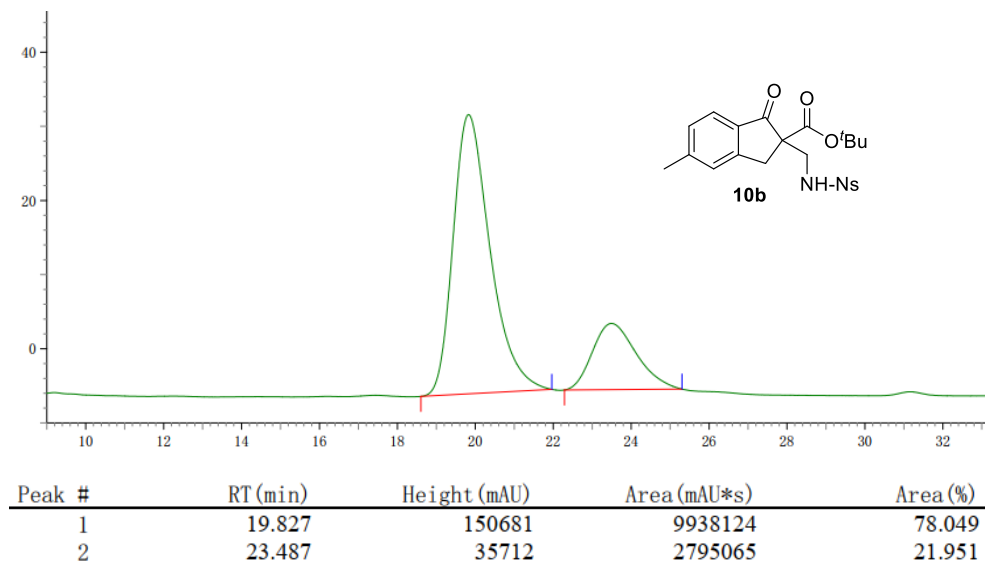


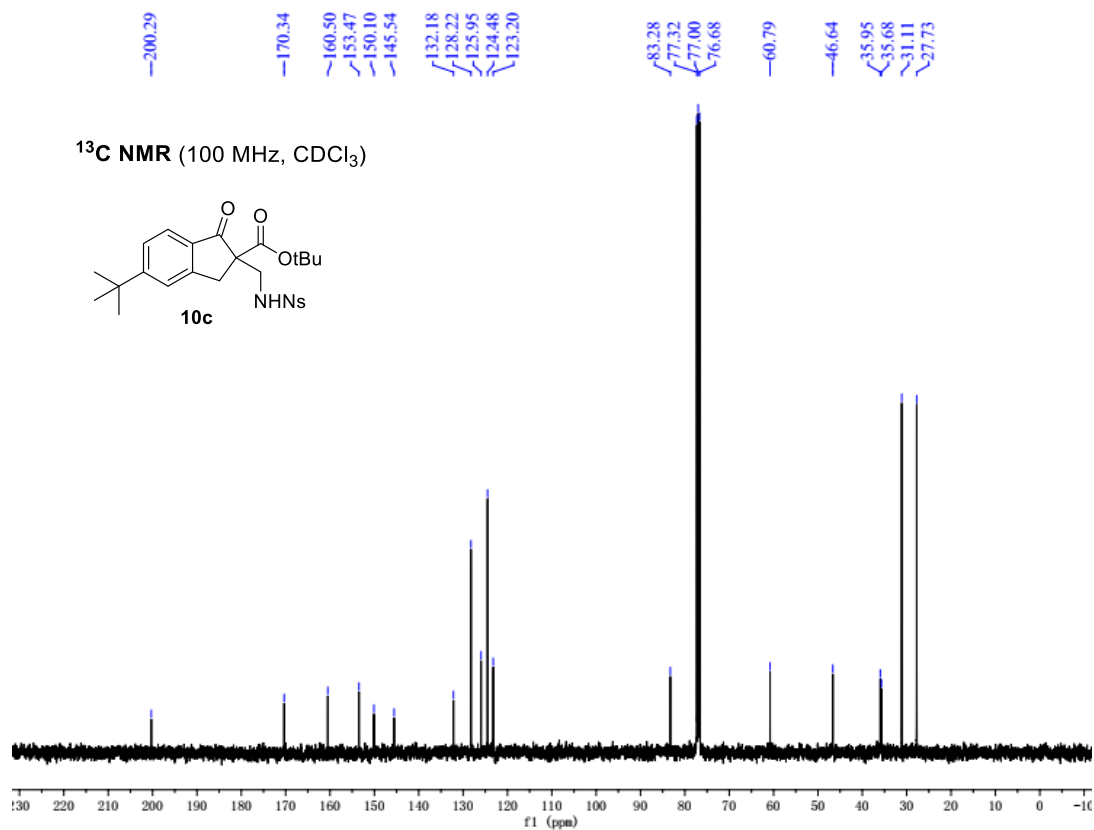
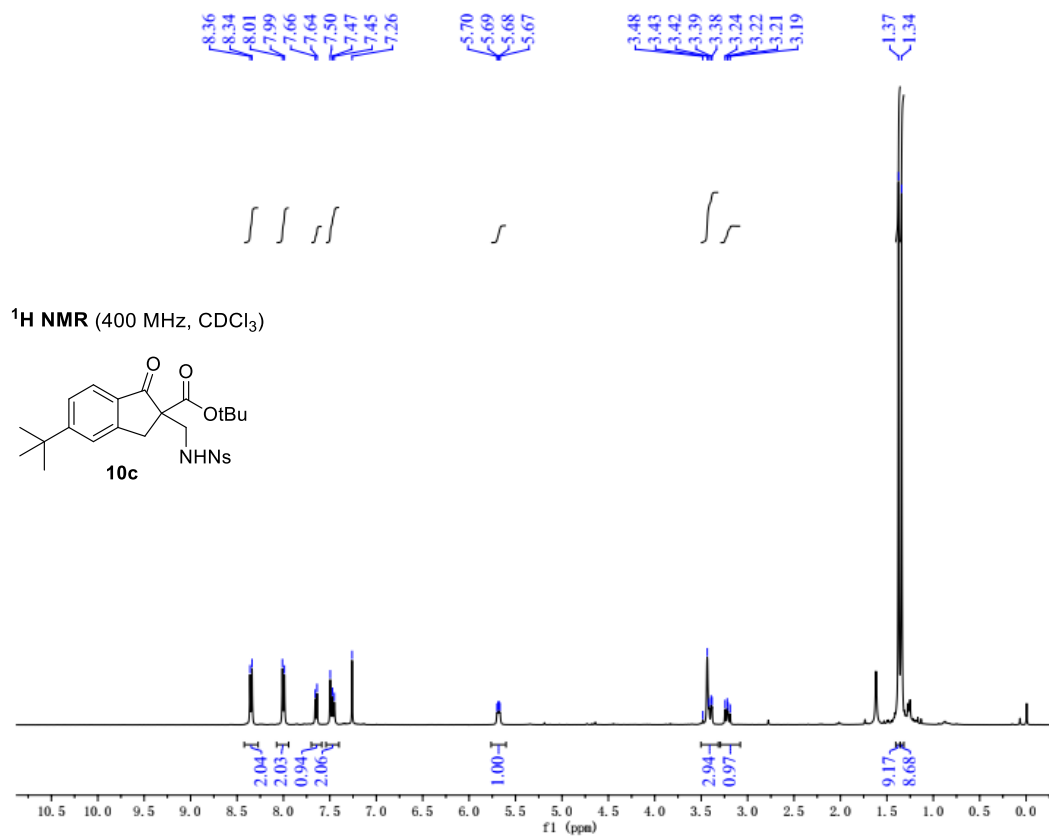
HPLC spectrum of compound **10b**:

Racemate:



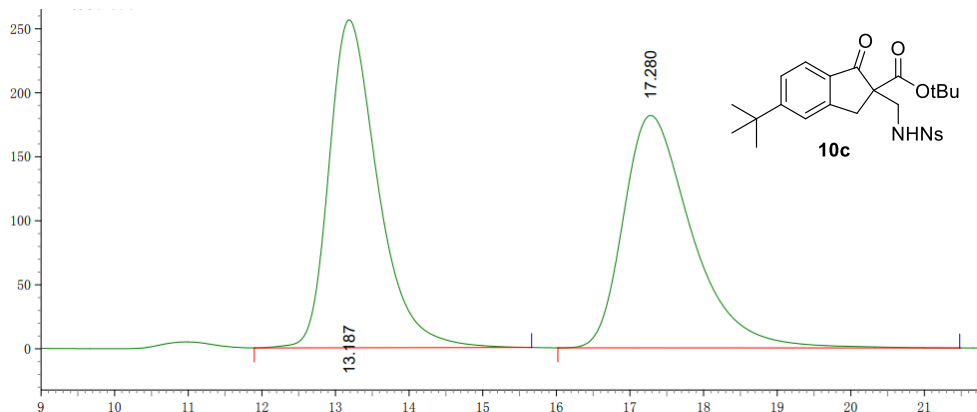
Enantioselective:





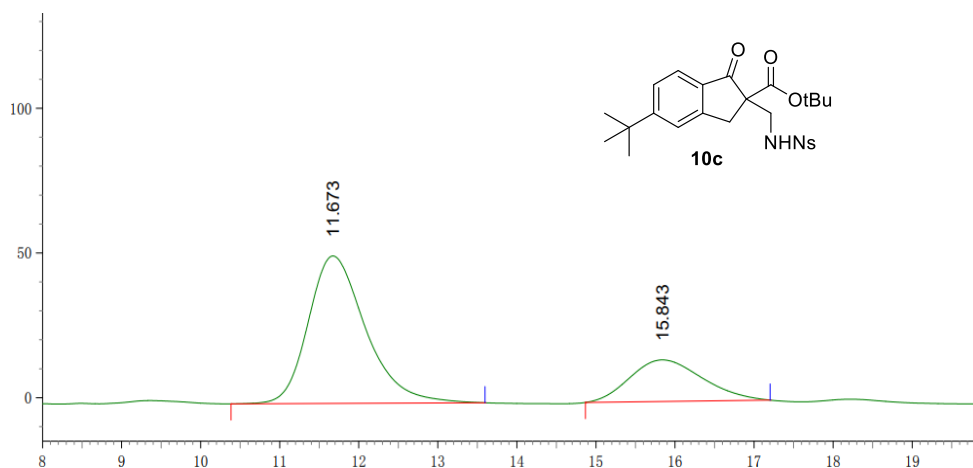
HPLC spectrum of compound **10c**:

Racemate:

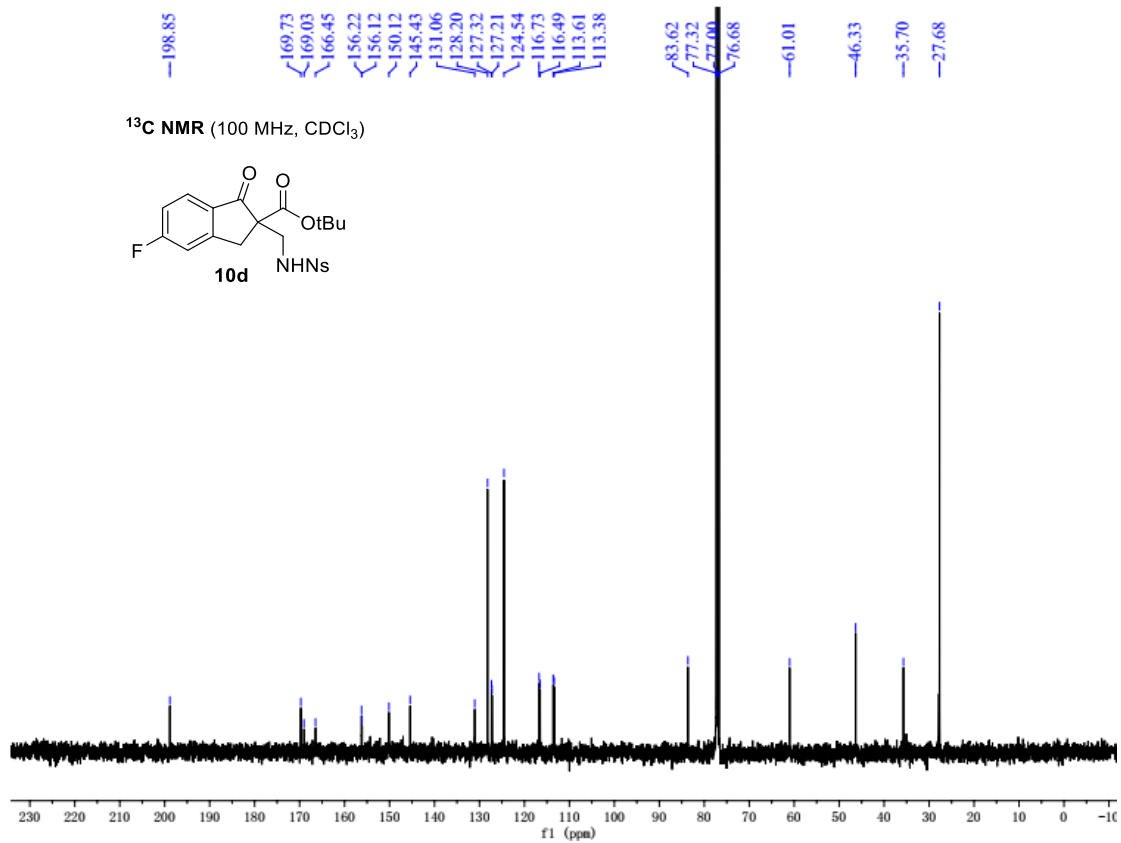
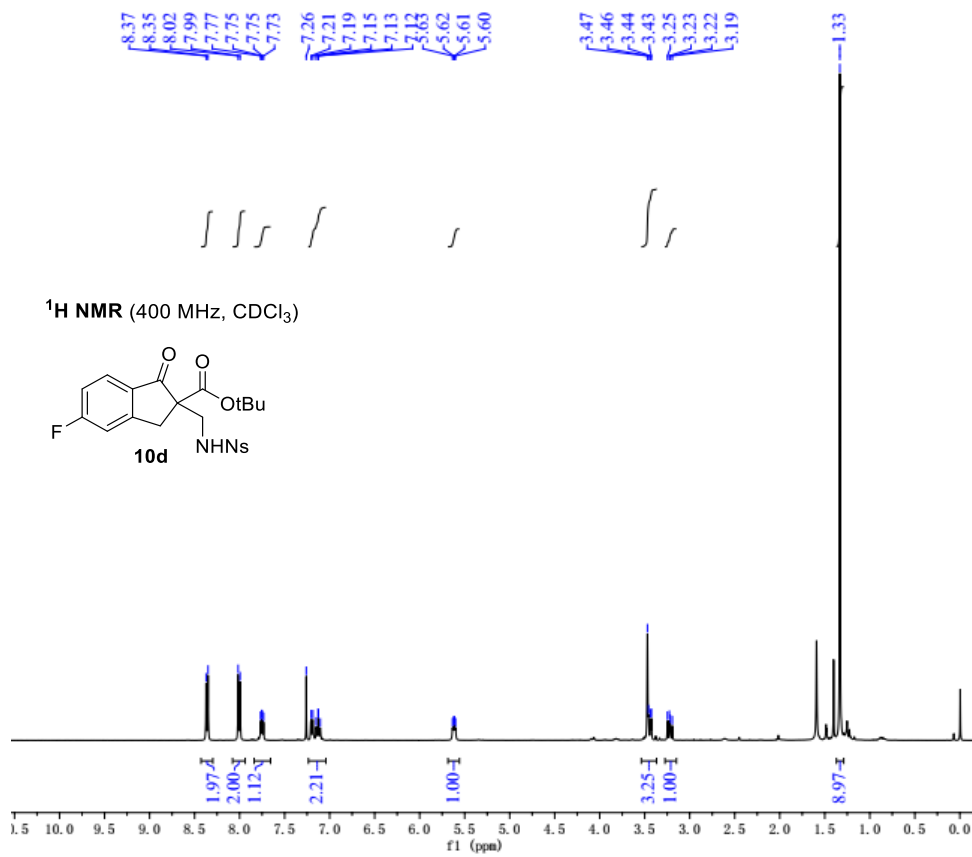


Peak #	RT (min)	Height (mAU)	Area (mAU*s)	Area (%)
1	13.187	1024541	47243497	50.224
2	17.280	726191	46821786	49.776

Enantioselective:

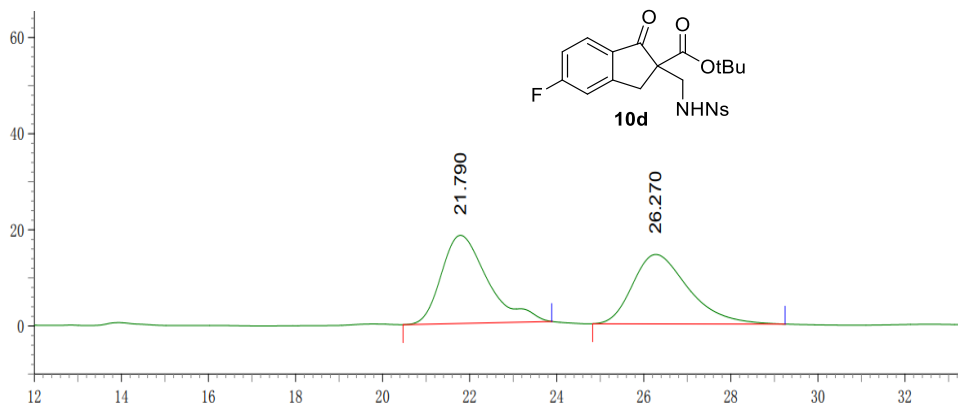


Peak #	RT (min)	Height (mAU)	Area (mAU*s)	Area (%)
1	11.673	203746	10282570	73.747
2	15.843	57456	3660435	26.253



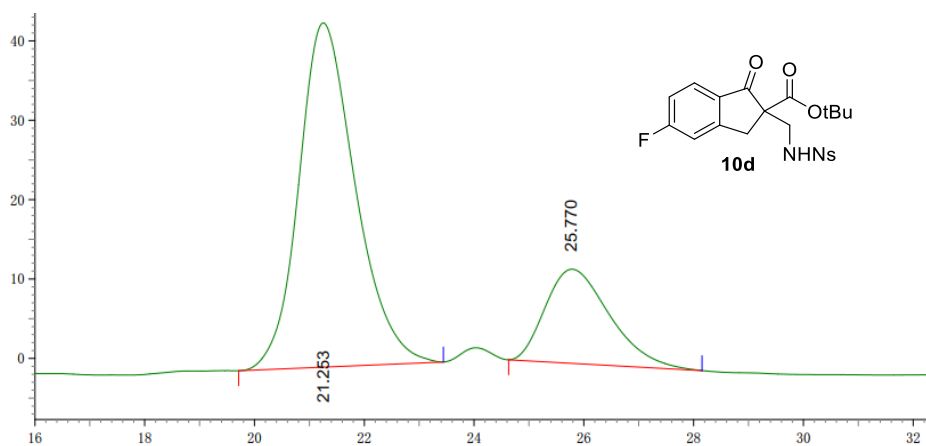
HPLC spectrum of compound **10d**:

Racemate:



Peak #	RT (min)	Height (mAU)	Area (mAU*s)	Area (%)
1	21.790	73359	5416753	51.302
2	26.270	57765	5141784	48.698

Enantioselective:



Peak #	RT (min)	Height (mAU)	Area (mAU*s)	Area (%)
1	21.253	173453	12407201	76.275
2	25.770	47379	3859122	23.725

