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Supporting Information

Identification and in situ removal of inhibitory intermediate to
develop an efficient phytosterol bioconversion process using a
cyclodextrin-resting cells system

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Table S1 Detailed parameters of the macroporous adsorbent resin used.

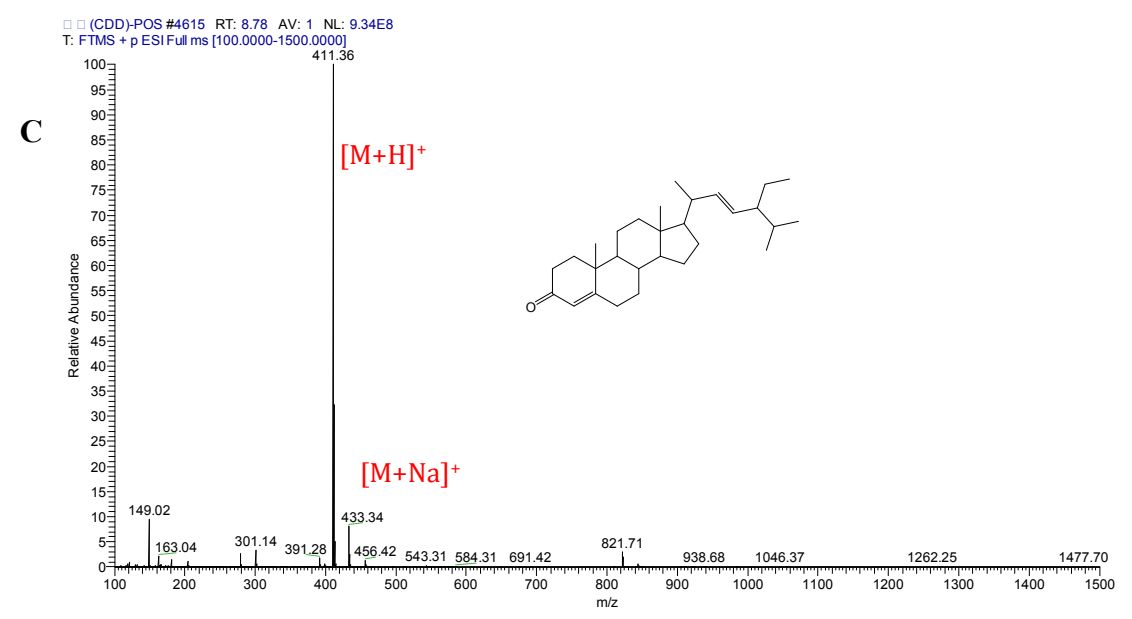
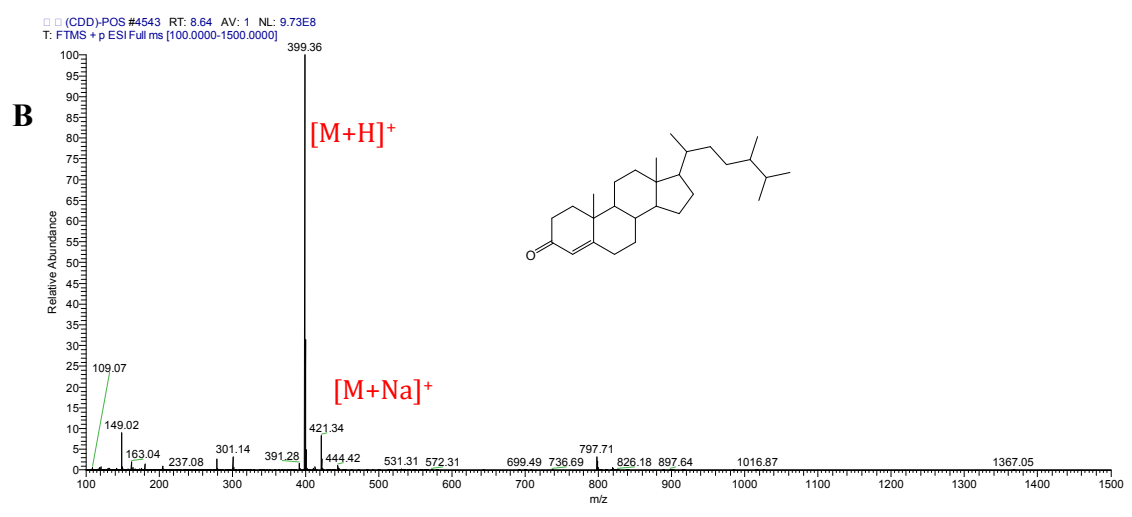
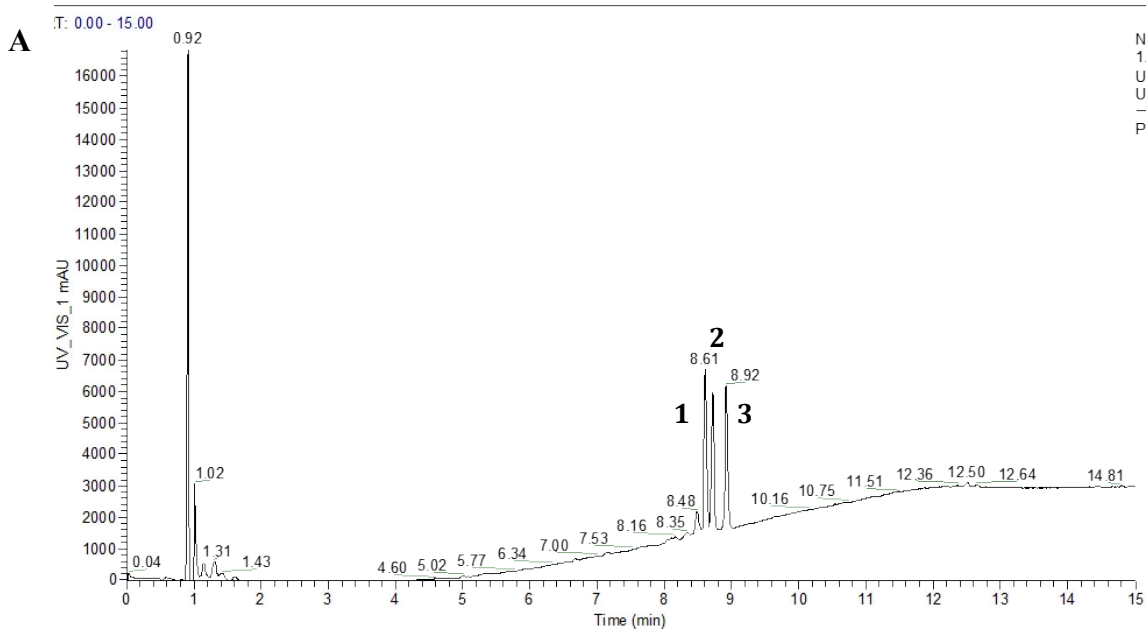
Resin	Manufacturers	Polarity	Specific surface area (m ² /g)	Pore diameter (nm)	Particle size (μm)
D 101	Haiguang Chemical Co., Ltd	Non-polar	480-520	9-10	300-1250
SP 207	Mitsubishi Chemical Industries Limited	Non-polar	900-950	4-5	250-1000
HZ 820	Huazhen Technology Co., Ltd	Non-polar	950-1000	13-15	380-1200
HZ 818	Huazhen Technology Co., Ltd	Non-polar	880-920	7-8	315-1250
HZ 806	Huazhen Technology Co., Ltd	Middle-polar	450-550	6-7	315-1250
HZ 801	Huazhen Technology Co., Ltd	Non-polar	500-550	12-14	315-1250
HPD 826	Baoen Chemical Co., Ltd	Non-polar	485-530	13-14	300-1250
HPD 500	Baoen Chemical Co., Ltd	Polar	500-550	5-6	300-1200
HPD 300	Baoen Chemical Co., Ltd	Non-polar	800-870	5-7	300-1200
HPD 100	Baoen Chemical Co., Ltd	Non-polar	650-700	8-9	300-1250

Note: All adsorbents are polystyrene-divinylbenzene type resins. The parameters depicted in the table were obtained from the manufacturer.

Phytosterols (the main components are campesterol, stigmasterol, β -sitosterol) were used as substrates to obtain steroid intermediates catalyzed by *Mycobacterium neoaurum* NwIB-HK86. The separation and purification of the metabolic intermediates adsorbed in resins was performed by silica gel column chromatography. To prepare the silica gel column, 30 g of silica gel (200-300 mesh) was weighed and mixed in hexane and then packed in a glass column (I.D.21 mm \times H200 mm). Subsequently, silica gel with sample were loaded on the column. The intermediates were desorbed by an eluant consisting of hexane and ethyl acetate (hexane: ethyl acetate=5:1, v/v) at a flow rate of 80 mL/h. The eluant was collected per 20 mL. After separation, the samples of the same products identified by TLC were collected together and then dried with a rotary evaporator.

5 μ L of samples were applied to silica gel plate (MERCK TLC Silica gel 60 F₂₅₄, Germany), developed in n-hexane/ethyl acetate/chloroform (5:5:2, v/v/v). The visualization was performed under 254 nm UV light and 20% sulfuric acid solution (115 °C for 15 minutes), respectively.

Multiple types of metabolites produced during bioconversion were adsorbed by the resins. The putative inhibitor was identified using ultra-performance liquid chromatography-tandem mass spectrometry (UPLC-MS/MS, Thermo Scientific *Q-Exactive plus, USA) to determine the molecular weight and fragmentation pattern of these compounds. Masses were recorded in positive mode in the range of 50-6000 m/z. A Hypersil Gold C18 (2.1 mm \times 100 mm) column was used and methanol-water (70% methanol at 0–1 min, linear gradient from 70–100% methanol in 20 min, keep 100% methanol at 20-25 min, and then linear gradient from 100–70% methanol in 3 min, keep for 3 min to equilibrate the column) was used as the mobile phase at a flow rate of 1 mL/min and UV detector at 254 nm.



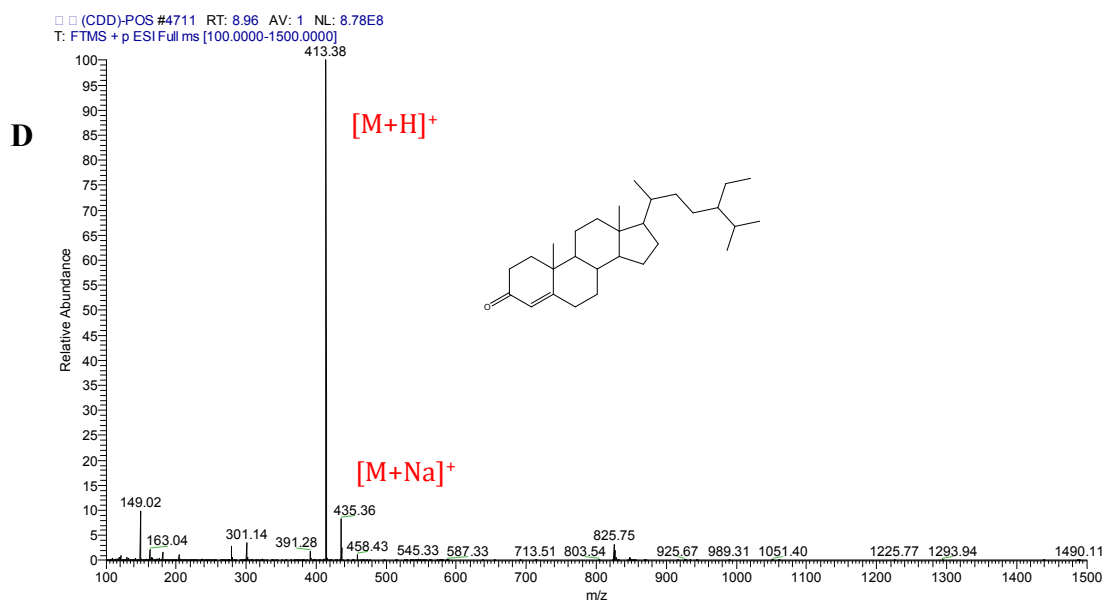


Figure S1. Identification of putative inhibitory intermediates by UPLC-MS/MS. A, UPLC profiles of 4-ene-3-keto steroids; B, MS of 4-ene-3-keto steroids produced from campesterol, corresponding to peak 1 in the UPLC profile; C, MS of 4-ene-3-keto steroids produced from stigmasterol, corresponding to peak 2 in the UPLC profile; D, MS of 4-ene-3-keto steroids produced from β -sitosterol, corresponding to peak 3 in the UPLC profile.

Quantitative analysis of 4-ene-3-keto steroids was performed by HPLC equipped with an Agilent ZORBAX 300SB-C8 (4.6 mm× 250 mm) column. Methanol-water (80:20, v/v) was used as the mobile phase at a flow rate of 1 mL/min and UV detector at 254 nm. 4-ene-3-keto steroids content in 12 h samples of the control group, which total peak areas was 298.48, was set to a relative concentration of 1 in the manuscript (the relative content of 4-ene-3-keto steroids can be calculated according to the following formula:

$$\text{Relative content of 4-ene-3-keto steroids} = \frac{\text{Area}(4\text{-ene-3-keto steroids})}{298.48}$$

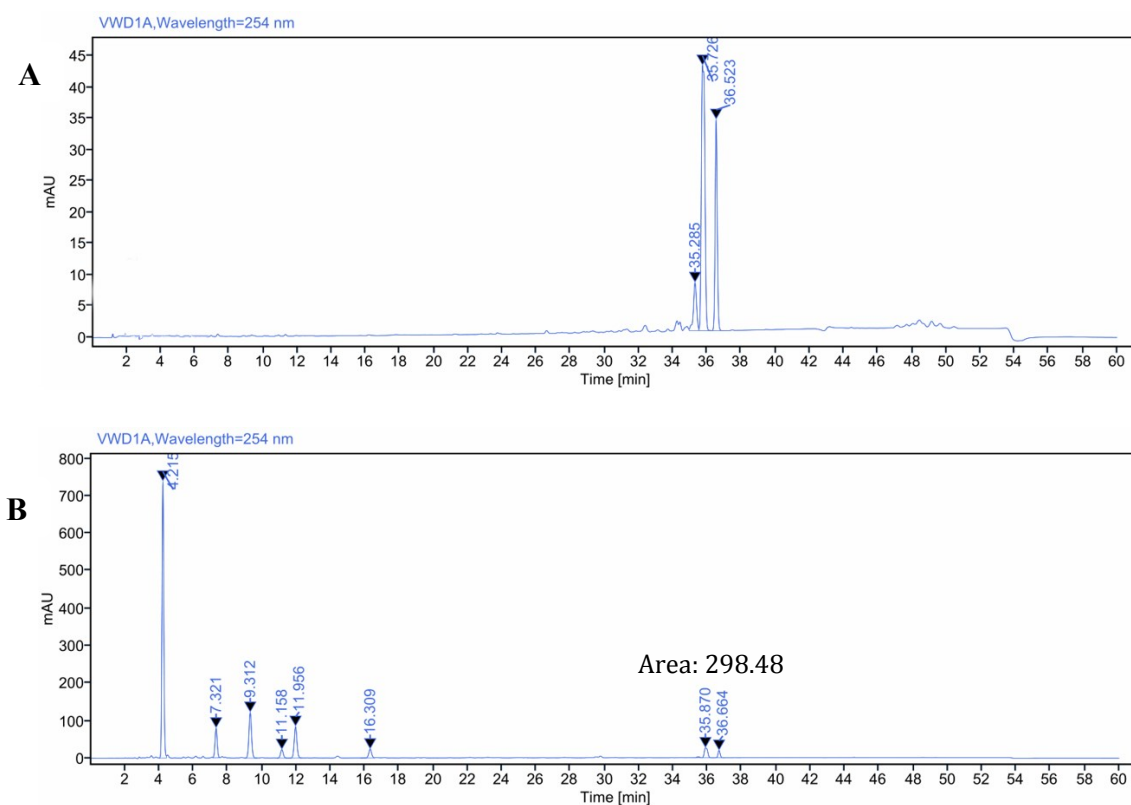


Figure S2. HPLC profiles of 4-ene-3-keto steroids. A, 4-ene-3-keto steroids obtained by silica gel column chromatography; B, 4-ene-3-keto steroids content in 12 h samples of the control group.