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Supplementary Information

2 **Sweetsop-like α -Fe₂O₃@CoNi catalyst with superior peroxidase-like activity**
3 **for sensitive and selective detection of hydroquinone**

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14 **Table S1.** Comparison of Kinetic parameters (K_m and V_{max}).

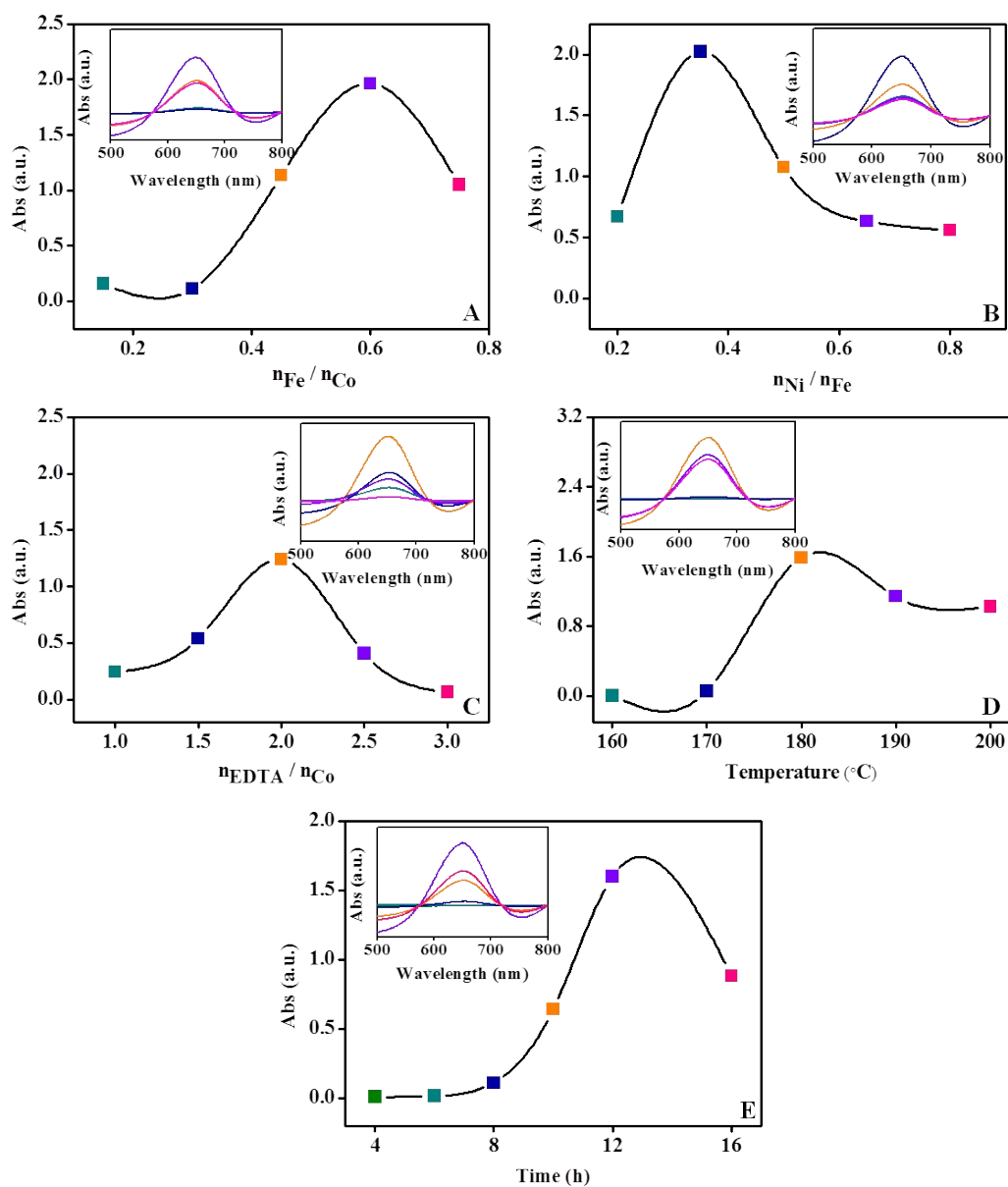
Catalyst	Substrate	$V_{max}(10^{-8}M \cdot s^{-1})$	$K_m(mM)$	Ref.
α -Fe ₂ O ₃ @CoNi	TMB	13.5	0.23	This work
	H ₂ O ₂	9.3	0.42	
Fe SSN	TMB	20.4	0.53	[1]
	H ₂ O ₂	13.2	0.36	
CDs@ZIF-8-a	TMB	1.95	0.232	[2]
	H ₂ O ₂	1.22	0.737	
CB-CQDs	TMB	5.13	0.83	[3]
	H ₂ O ₂	4.09	0.70	
Co ₉ S ₈	TMB	99	1.64	[4]
	H ₂ O ₂	35	7.39	
Por-NiCo ₂ S ₄	TMB	34.86	0.3	[5]
	H ₂ O ₂	4.32	4.5	
HRP	TMB	10	0.43	[6]
	H ₂ O ₂	8.71	3.70	

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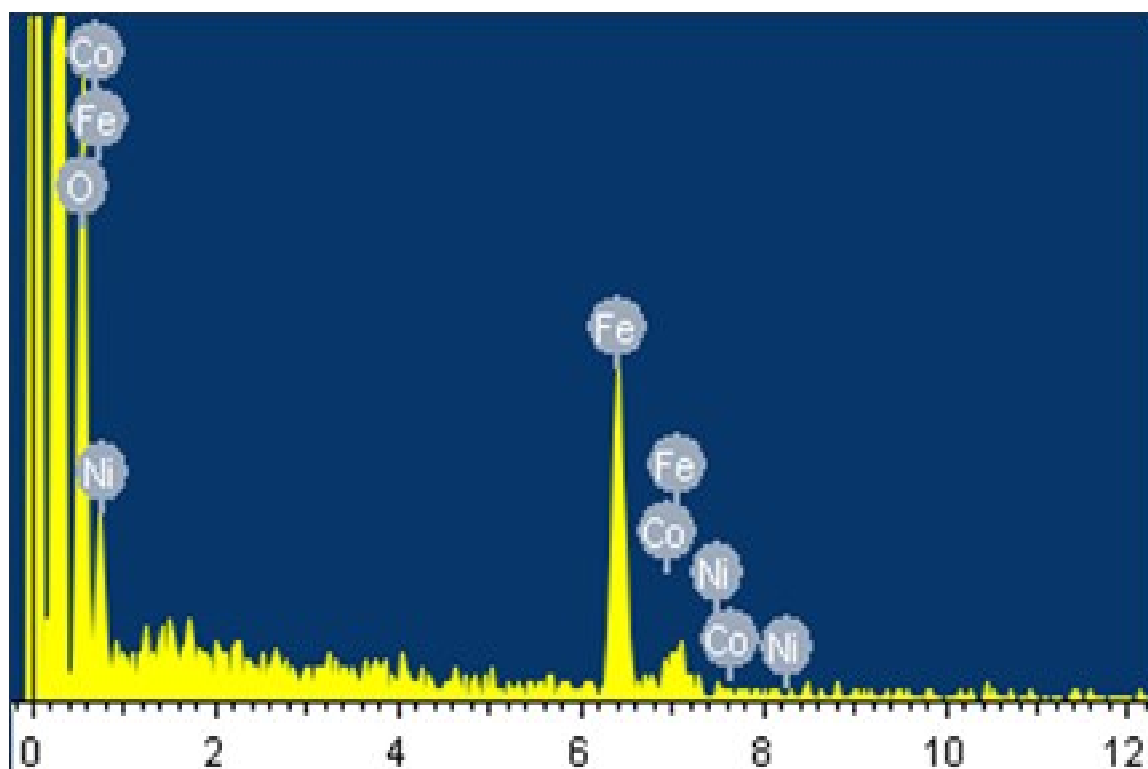
17 **Table S2.** Comparison of the proposed method with other methods for the detection of HQ.

No.	Method	Materials	Linear Rang(μM)	LOD(μM)	Ref.
1	Colorimetry	Pt/CdS	1.0-10	0.165	[7]
2	Colorimetry	ZnO/ZnFe ₂ O ₄	0-150	3.75	[6]
3	Colorimetry	NiCo ₂ O ₄	5-110	2.70	[8]
4	Fluorescence	SiQDs	6-100	2.63	[9]
5	Fluorescence	g-CNQDs	0.5-11.6	0.04	[10]
6	Fluorescence	N/S/P-codoped CDs	0.56-375	0.16	[11]
7	ECL	MOF-rGO	10-200	0.66	[12]
8	ECL	CoTFPP/GO/GCE	1-200	0.21	[13]
9	ECL	CoFe ₂ Se ₄ /PCF	0.5-200	0.13	[14]
10	Colorimetry	α -Fe ₂ O ₃ @CoNi	0.5-30	0.16	This work

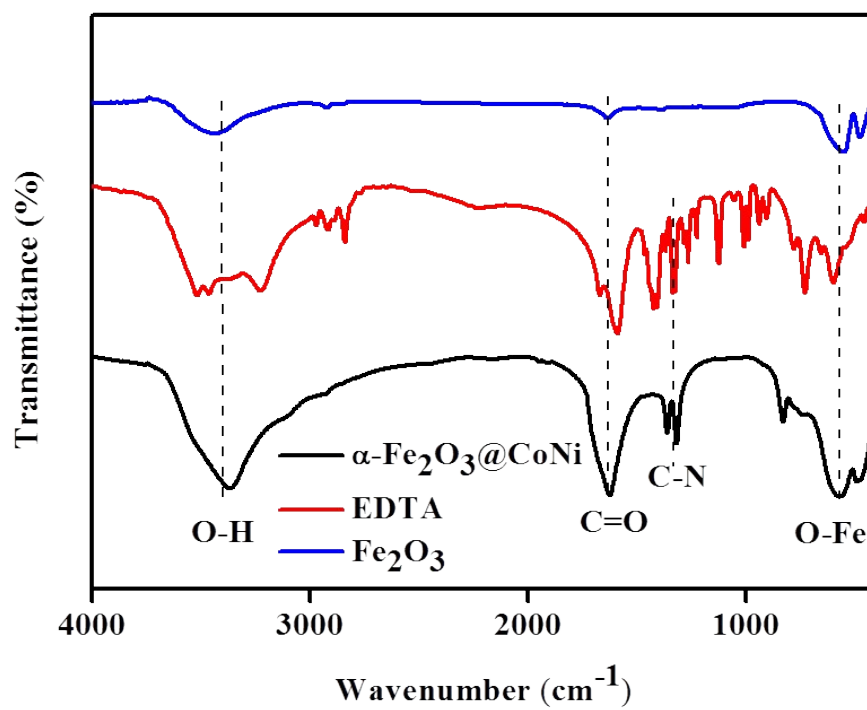
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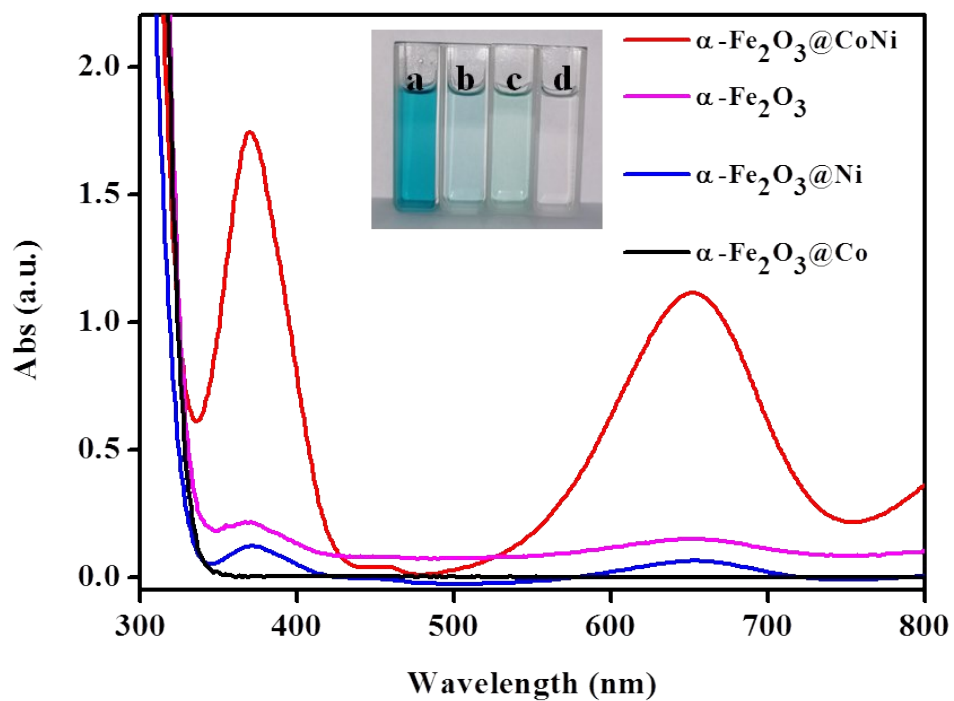
19 **Fig. S1** Optimization of Co, Ni and EDTA dosage (A, B, C), synthesis temperature and
 20 time (D and E), (inset) the oxTMB of corresponding absorption spectra.



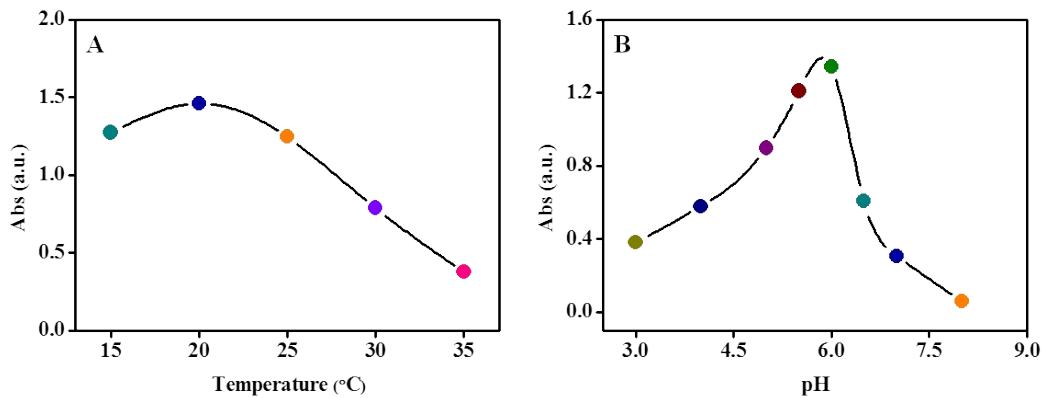
21 Fig. S2 Energy dispersive spectrometer (EDS) image of α -Fe₂O₃@CoNi.



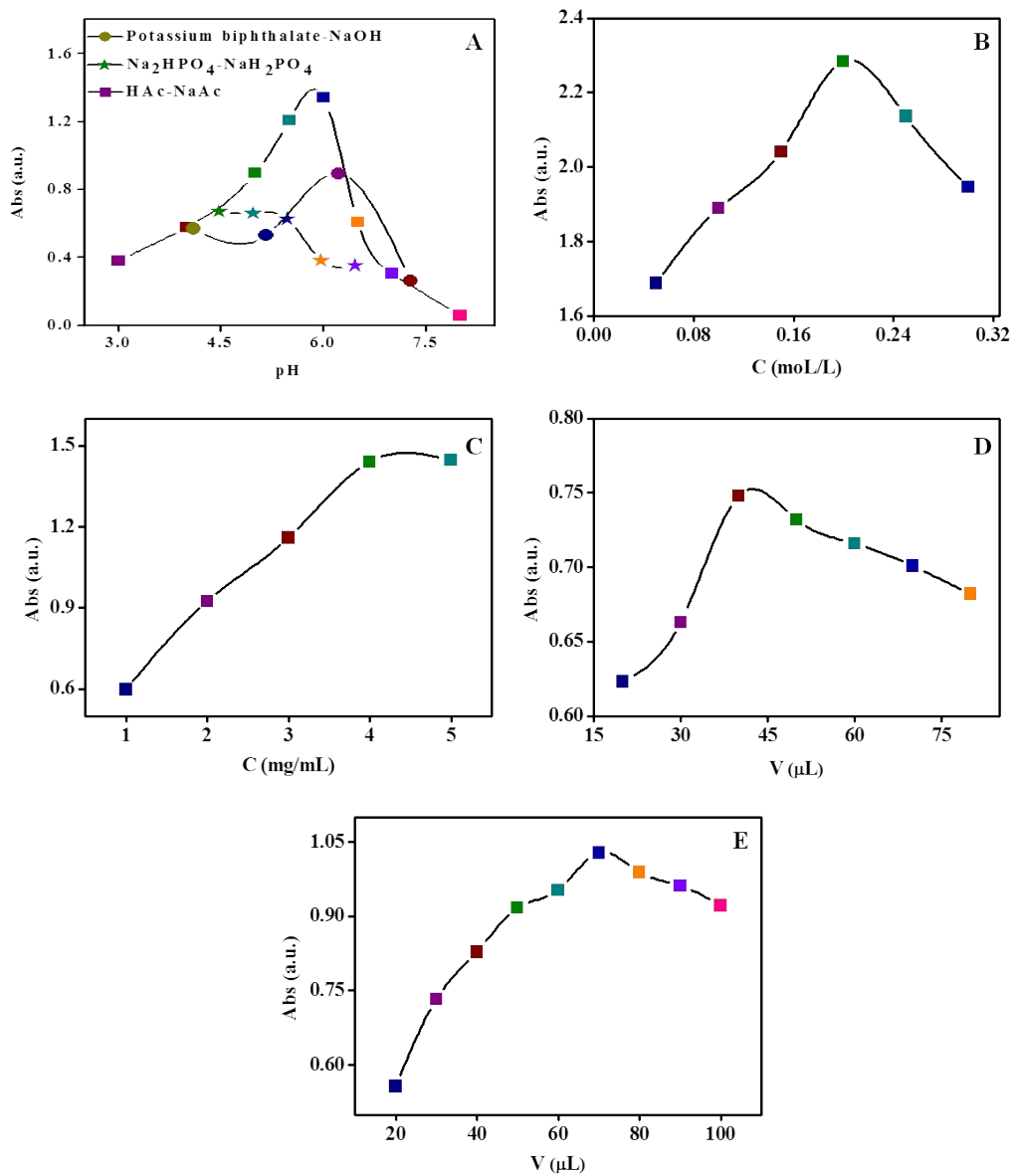
22 **Fig. S3** The FT-IR spectra of $\alpha\text{-Fe}_2\text{O}_3\text{@CoNi}$, commercial Fe_2O_3 and EDTA.



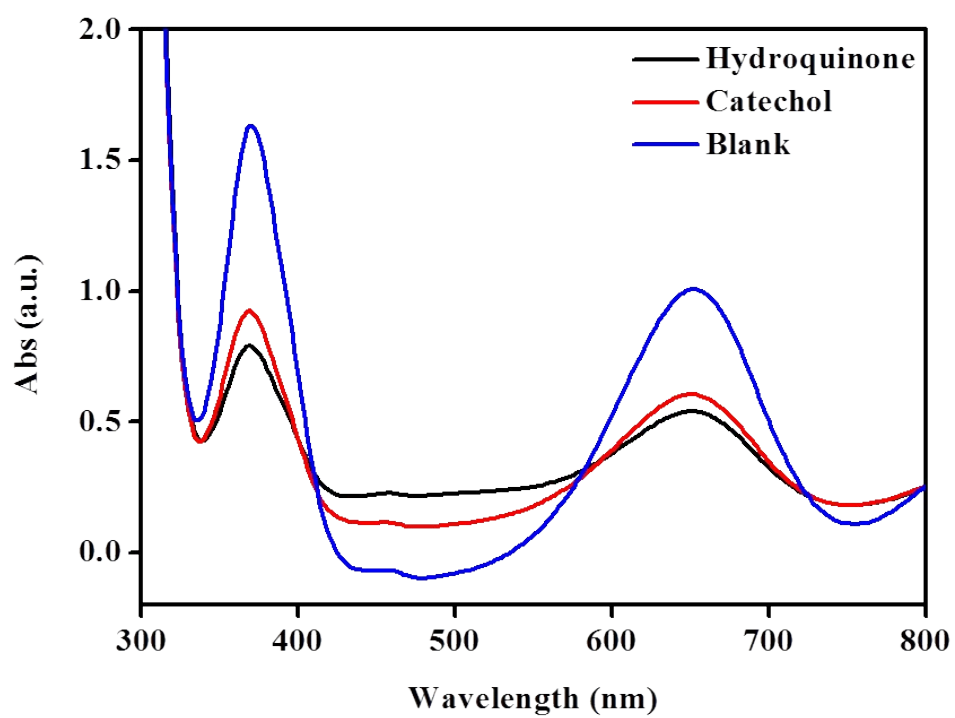
23 **Fig. S4** The UV-vis absorption spectrum of commercial Fe₂O₃, α -Fe₂O₃@Co, α -
 24 Fe₂O₃@Ni and α -Fe₂O₃@CoNi in TMB/H₂O₂ system. (inset) The color of
 25 corresponding solutions. (a) TMB+H₂O₂+ α -Fe₂O₃@CoNi, (b) TMB+H₂O₂+Fe₂O₃ (c)
 26 TMB+H₂O₂+ α -Fe₂O₃@Ni, (c) TMB+H₂O₂+ α -Fe₂O₃@Co.



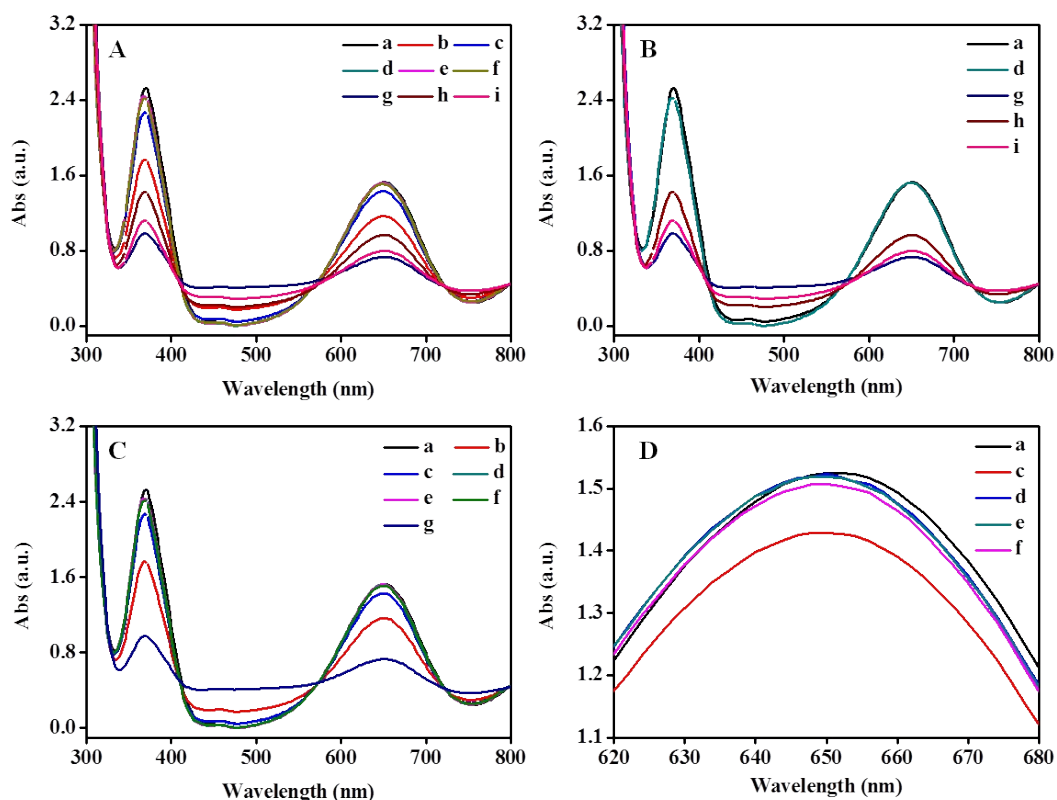
28 **Fig. S5** Effects of temperature (A) and pH (B) on the catalytic activity of α -
29 Fe₂O₃@CoNi.



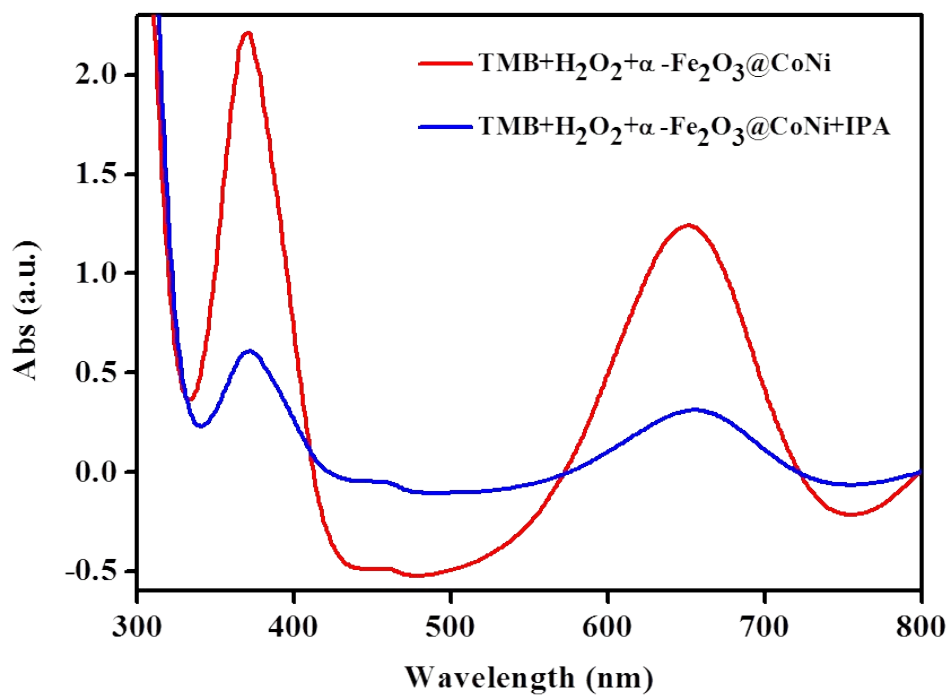
31 **Fig. S6** Effects of type and concentration of buffer solution (A, B), concentration of α -
 32 $\text{Fe}_2\text{O}_3@\text{CoNi}$ (C), dosage of TMB (D) and H_2O_2 (E) on the catalytic activity of α -
 33 $\text{Fe}_2\text{O}_3@\text{CoNi}$.



35 **Fig. S7** The UV-vis absorption spectrum of hydroquinone (30 μM) and Catechol (30
36 μM) in $\alpha\text{-Fe}_2\text{O}_3\text{@CoNi/TMB/H}_2\text{O}_2$ system and blank control.



38 **Fig. S8** The UV-vis absorption spectrum of TMB+H₂O₂+α-Fe₂O₃@CoNi system with
 39 different alumina additions (a, blank; b-f, 0.1-0.5g alumina added into 2 mL 0.006 M
 40 HQ solution; g, catechol solution before alumina treatment; h-i, HQ solution before
 41 and after alumina treatment, respectively).



43 **Fig. S9** The effect of isopropyl alcohol (IPA) on the catalytic activity of α -Fe₂O₃@CoNi.

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