One-Step Green Synthesis of 2D Ag Dendrites-Embedded Biopolymer Hydrogel Beads for Catalytic Reactor

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Supporting Tables

Table S1. Activity parameter, κ , values for various heterogeneous catalysts, which is obtained by dividing the reaction rate constant by the total weight of used catalyst.

Heterogeneous catalysts	Total weight of used catalyst [mg]	Reaction rate constant (k) $[10^{-3} s^{-1}]$	к [s ⁻¹ g ⁻¹]	Ref.
Hollow PS/Au-Hollow PS/Au-Ag NPs	0.40	2.14	5.35	1
Hollow PS/Pt-Ag NPs	0.40	4.57	33.43	1
PS with an open hole/Ag NPs	0.12	1.74	14.5	2
Porous PS/Ag NPs	0.15	5.79	38.6	3
Cross-linked PS/Au NPs	100	25	0.25	4
Hybrid cryogel/Ag NPs	20	5.03	0.25	5
Sodium polyacrylate water ball/Ag NPs	20	7.67	0.38	6
Mg-Al layer double hydroxide/Au-Ag NPs	1	0.48	0.48	7
Spherical covalent organic framework/Ag NPs	25	17.67	0.71	8
Covalent organic framework@Fe3O4/Au NPs	3	3.7	1.23	9
Spindle-shaped PANI/Au NPs	31.5	40.4	1.28	10
SiO2@PPy/Au NPs	31.5	44.2	1.4	11
Heterocycle-modified PS/Au NPs	15	30.7	2.05	12
Ni@SiO2 magnetic hollow microspheres/Au	4	10	2.5	13
ZnO@Fe3O4/Ag NPs	3	8.18	2.73	14
Self-assembly of surface-modified Au NPs	0.9	2.58	2.87	15
Fe3O4@PS/Ag	2	8.9	4.45	16
Ca-Al layer double hydroxide hybrid@Fe3O4/Ag NPs	1	5.3	5.3	17
CNT@PZS/Au NPs	0.3	1.78	5.93	18
MWCNT@S.lavandulifolia/Ag NPs	3	19.2	6.4	19
Graphitic carbon nitride/Au NPs	2	15	7.5	20
Silica nanosheet/Ag NPs	10	80.19	8.02	21
PAM@PPy@GO/Ag NPs	4	33.8	8.45	22
Metal-organic framework/Au NPs	1.5	24.2	16.13	23
CaCO3/Au NPs	1	18	18	24
Thiol-functionalized nGO@PEG/Au NPs	0.07	1.73	24.71	25
Fe3C@NG/Au	0.5	17	34	26
Alginate bead/Ag dendrites	0.4	16.09	40.23	This work

Supporting Figures



Fig. S1. Photographs of droplets of aqueous alginate solution containing Au nanocrystals dropped into (a) 0.125 M AgNO₃ aqueous solution and (b) pure water. (c) UV-vis absorbance spectrum and TEM image of Au nanocrystals used as pigments in this experiment.



Fig. S2. Photographs of hydrogel beads which was aged in the aqueous solution containing AgNO₃ (a) in the darkroom for 3 h and (b) under the sun light for 3 h. (c)-(d) Photographs of alginate beads obtained after immersing in 0.125 M AgNO₃ and aging at room temperature for 3 hours under light irradiation. In this experiment, Xenon lamp was used instead of UV lamp and the beads were exposed to light without stirring. (e)-(f) TEM images of Ag nanocrystals formed in the alginate beads shown in (c)-(d).



Fig. S3. FTIR spectra of pure Na-alginate (red) and alginate beads embedded with 2D Ag dendrites (black).



Fig. S4. UV–Vis absorption spectra for 10 min after the alginate hydrogel beads cross-linked by Ba^{2+} ions (i.e., not covered with Ag nanocrystals) were added to the solution that contained NaBH₄ and 4-NP.

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