

## Supporting Information

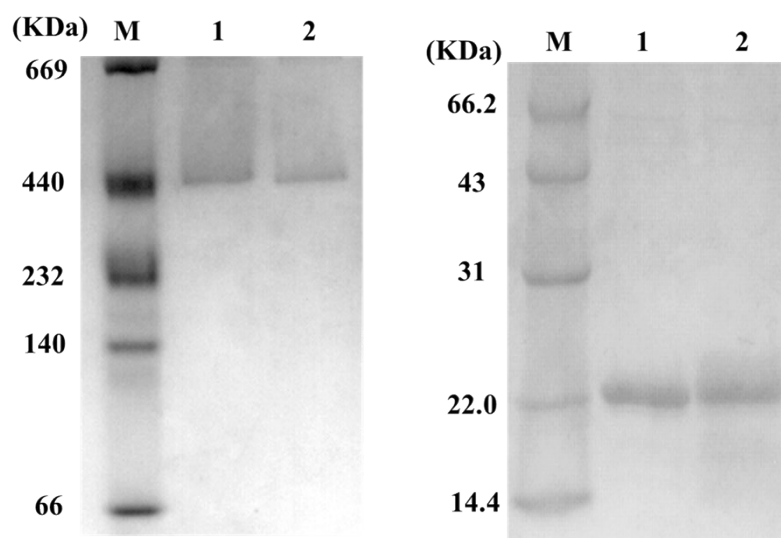
for

### **Ferritin-catalyzed synthesis of ferrihydrite nanoparticles with high mimetic peroxidase activity for biomolecules detection**

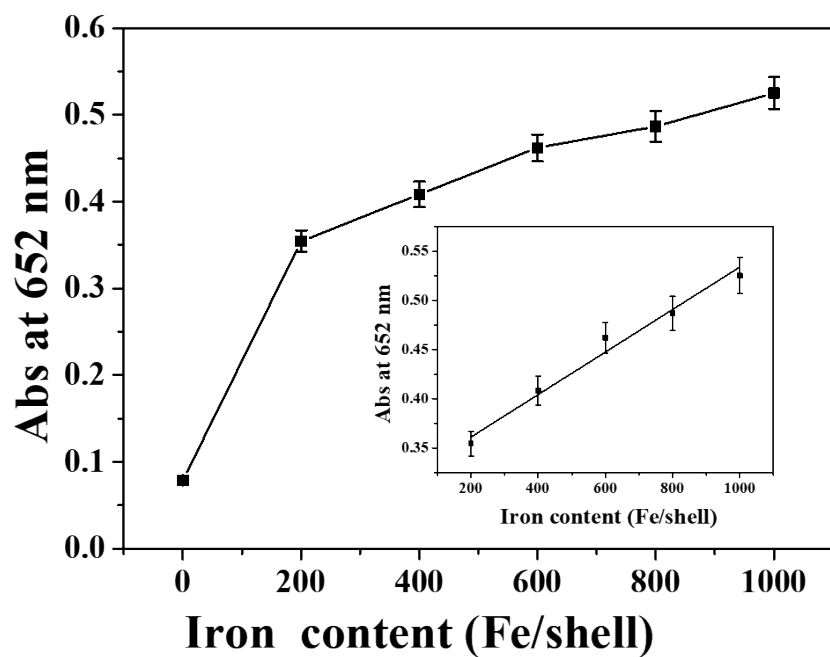
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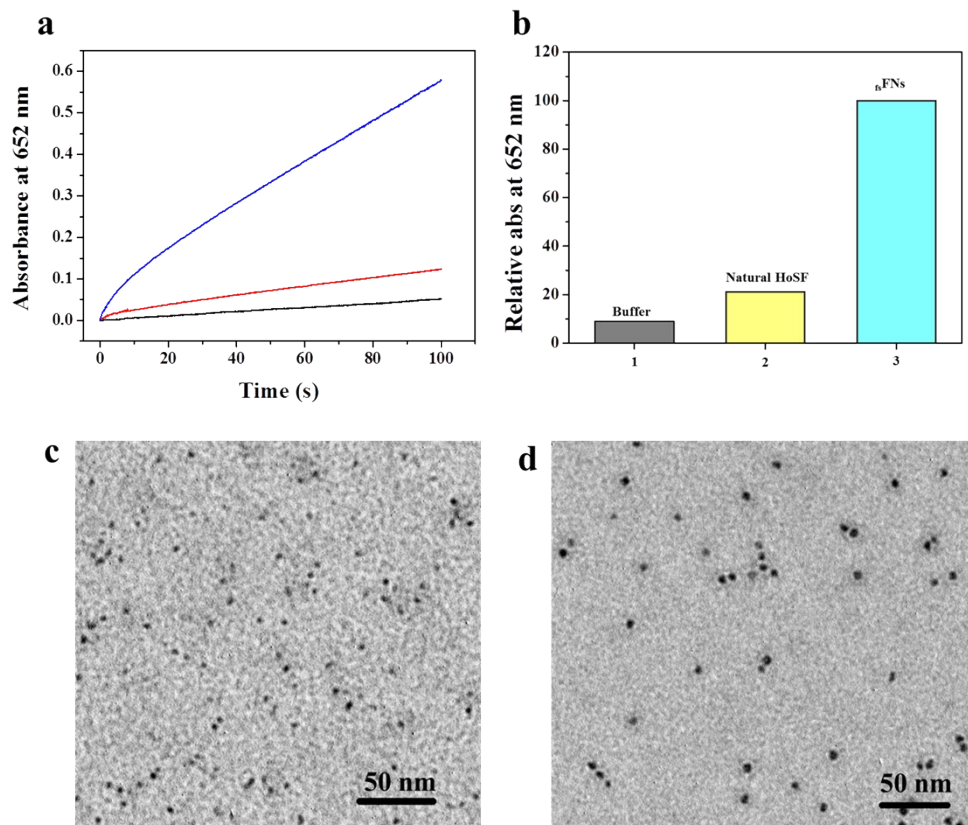
## Supplementary Figures (Figs. S1-S5)



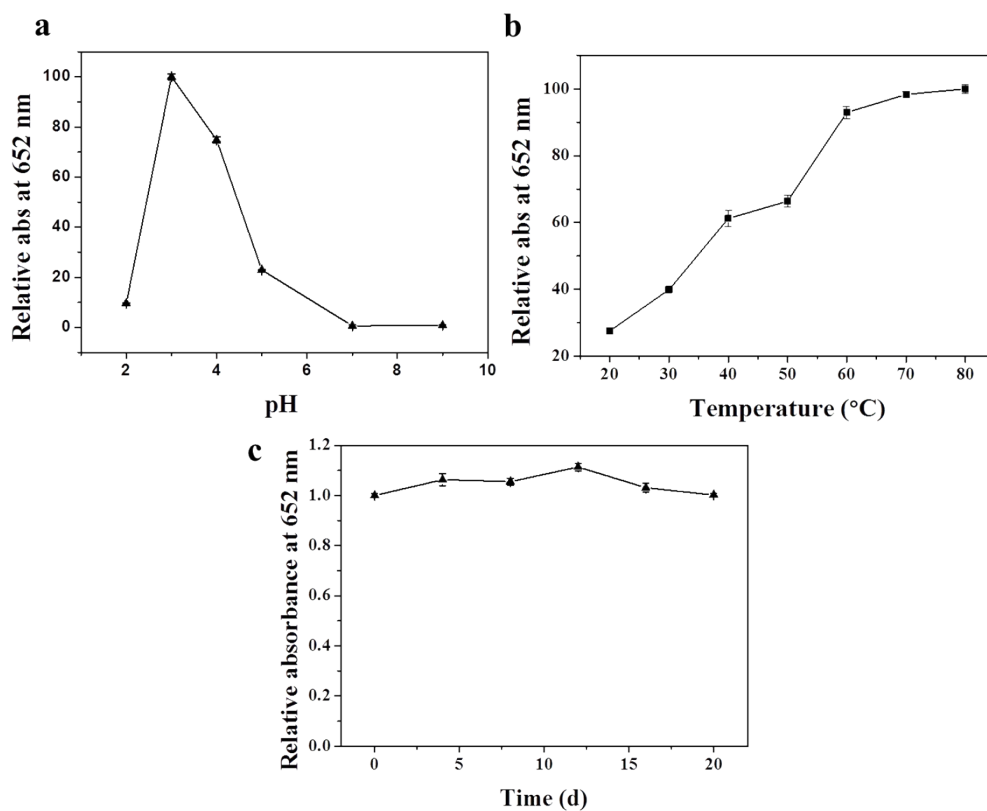
**Figure S1.** (a) SDS-PAGE and (b) Native-PAGE analyses. Land M, protein markers and their corresponding molecular masses; Land 1, purified rHuHF; Land 2, holo rHuHF with loading 800 iron/protein within the inner cavity.



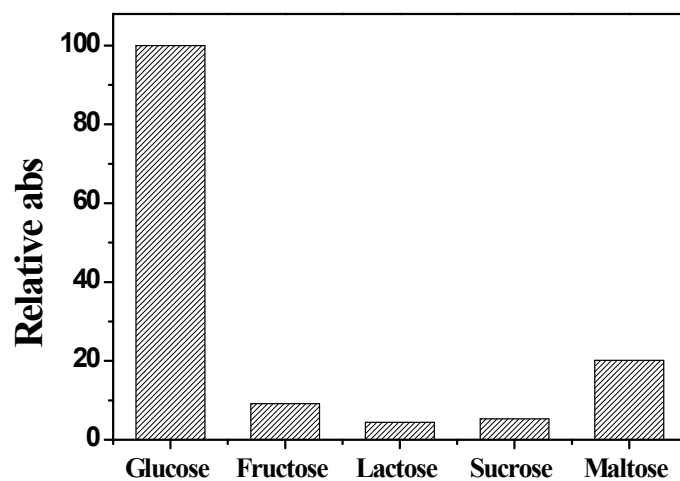
**Figure S2.** Relationship between the absorbance at 652 nm and the iron content. Insert: A near linear relationship between the absorbance at 652 nm and the iron content. Conditions: 1.0 mM TMB, 10 mM H<sub>2</sub>O<sub>2</sub>, and [apo ferritin or f<sub>s</sub>FNs] = 0.25 μM.



**Figure S3.** Comparison of the catalytic activity between natural holo horse spleen ferritin (HoSF) and  $f_s$ FNs. (a) Time-dependence absorbance changes at 652 nm of buffer (black),  $f_s$ FNs (blue) and natural holo HoSF (red). (b) Relative catalytic activity of buffer,  $f_s$ FNs and natural holo HoSF with respect to 100% absorbance assumed for  $f_s$ FNs at 652 nm. TEM images of  $f_s$ FNs containing 800 iron/shell (c) and natural holo HoSF (d) without uranyl acetate staining.



**Figure S4.** Effect of different experimental conditions on the peroxidase-like activity of  $f_s$ FNs, which include (a) pH value, (b) temperature, and (c) storage time. Conditions: 1.0 mM TMB, 10 mM  $H_2O_2$ , and  $[f_sFNs] = 0.25 \mu M$ .



**Figure S5.** Determination of the selectivity for glucose detection. The concentration of glucose and fructose, lactose, sucrose, and maltose are 80  $\mu$ M.

## Supplementary table

**Table 1.** Comparison of kinetic parameters of TMB oxidation by  $f_s$ FNs and HRP.

Catalyst	Substrate	$V_{max}$ (M s <sup>-1</sup> )	$K_m$ (M)	Ref.
$f_s$ FNs	TMB	$4.4 \times 10^{-7}$	$4.5 \times 10^{-4}$	This work
$f_s$ FNs	H <sub>2</sub> O <sub>2</sub>	$4.5 \times 10^{-8}$	$3.1 \times 10^{-5}$	This work
HRP	TMB	$3.3 \times 10^{-8}$	$3.2 \times 10^{-4}$	[1]
HRP	H <sub>2</sub> O <sub>2</sub>	$2.4 \times 10^{-8}$	$5.5 \times 10^{-5}$	[1]
Au-Ft	TMB	$7.5 \times 10^{-8}$	$9.7 \times 10^{-5}$	[2]
Au-Ft	H <sub>2</sub> O <sub>2</sub>	$9.3 \times 10^{-8}$	$199.4 \times 10^{-3}$	[2]
Pt-Ft	TMB	$5.6 \times 10^{-7}$	$2.2 \times 10^{-4}$	[3]
Pt-Ft	H <sub>2</sub> O <sub>2</sub>	$3.2 \times 10^{-4}$	$187.3 \times 10^{-3}$	[3]
M-HFn-Co <sub>x</sub> Fe <sub>3-x</sub> O <sub>4</sub>	TMB	$1.3 \times 10^{-8}$	$1.3 \times 10^{-4}$	[4]
M-HFn-Co <sub>x</sub> Fe <sub>3-x</sub> O <sub>4</sub>	H <sub>2</sub> O <sub>2</sub>	$0.7 \times 10^{-8}$	$17.0 \times 10^{-3}$	[4]
Fe <sub>3</sub> O <sub>4</sub> MNPs	TMB	$3.4 \times 10^{-8}$	$9.8 \times 10^{-5}$	[5]
Fe <sub>3</sub> O <sub>4</sub> MNPs	H <sub>2</sub> O <sub>2</sub>	$9.8 \times 10^{-8}$	$154 \times 10^{-3}$	[5]

## References

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