

## SUPPORTING INFORMATION

### **A plant root-derived Ni-ecocatalyst for Suzuki cross coupling of aryl iodides**

Lucie Cases, Pauline Adler, Sébastien Diliberto, Clotilde Boulanger and Claude Grison\*

Laboratory of Bio-inspired Chemistry and Ecological Innovations UMR 5021 CNRS –  
University of Montpellier, Cap Delta, 1682 Rue de la Valsière, 34790 Grabels, France

\*Corresponding author: [claude.grison@cnrs.fr](mailto:claude.grison@cnrs.fr).

<b>I. Infrared spectra .....</b>	<b>2</b>
<b>1. IR spectra of the biomaterials – Before and after functionalization .....</b>	<b>2</b>
<b>2. Infrared spectrum of the Eco-Ni(HCOO)<sub>2</sub> ecocatalyst used for Suzuki coupling reaction .....</b>	<b>3</b>
<b>II. Description of the products .....</b>	<b>3</b>
<b>1. NMR spectroscopic description.....</b>	<b>3</b>
<b>2. NMR spectra .....</b>	<b>5</b>
<b>3. GCMS and GCFID analyses of standards and synthesized products .....</b>	<b>9</b>
<b>III. XPS analyses of the Eco-Ni(HCOO)<sub>2</sub> ecocatalyst .....</b>	<b>13</b>

# I. Infrared spectra

## 1. IR spectra of the biomaterials – Before and after functionalization

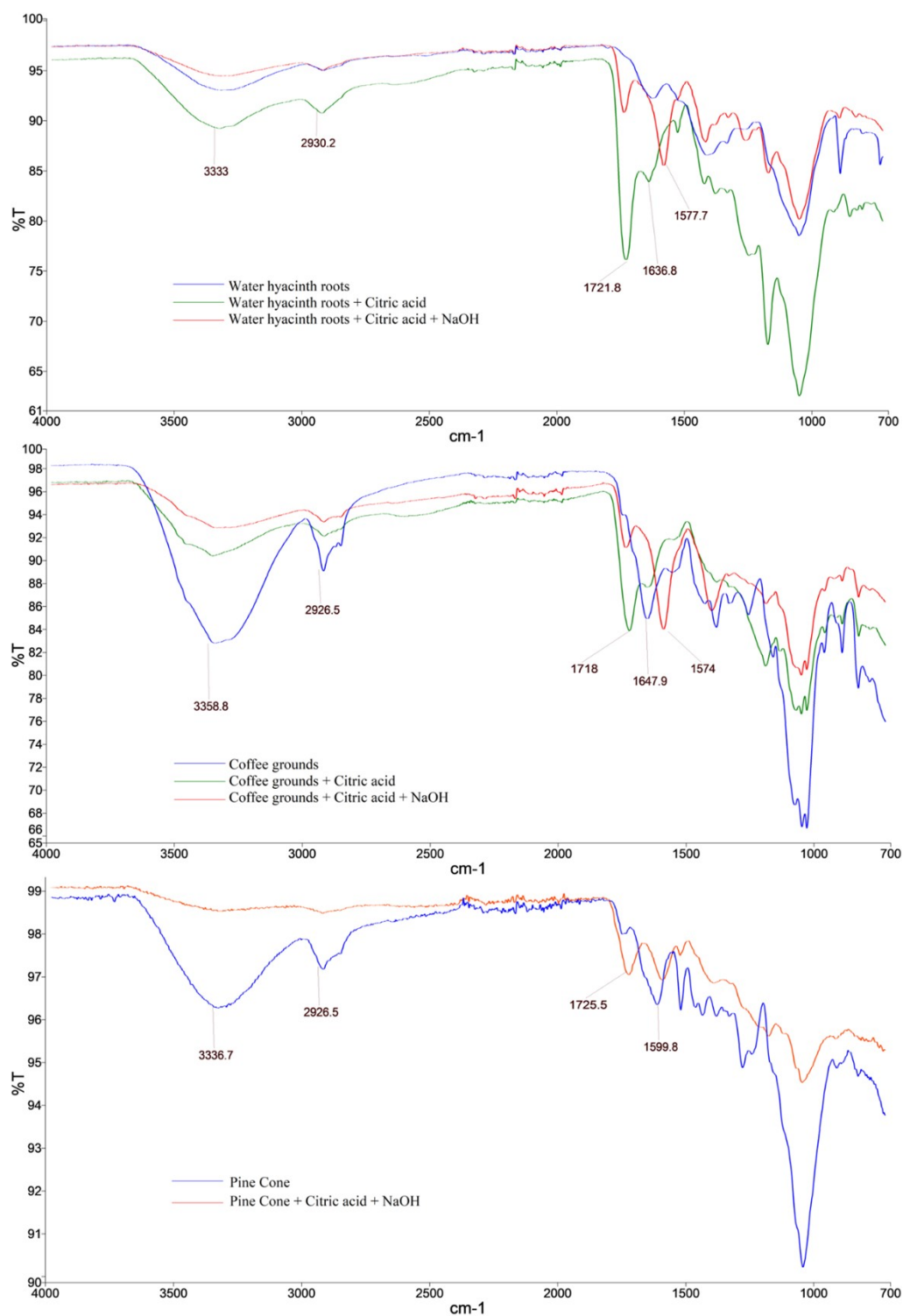


Figure 1. Infrared spectra of water hyacinth, coffee grounds and pinecone before (blue), after functionalization (green) and after titration with NaOH (red).

2. Infrared spectrum of the Eco-Ni(HCOO)<sub>2</sub> cocatalyst used for Suzuki coupling reaction

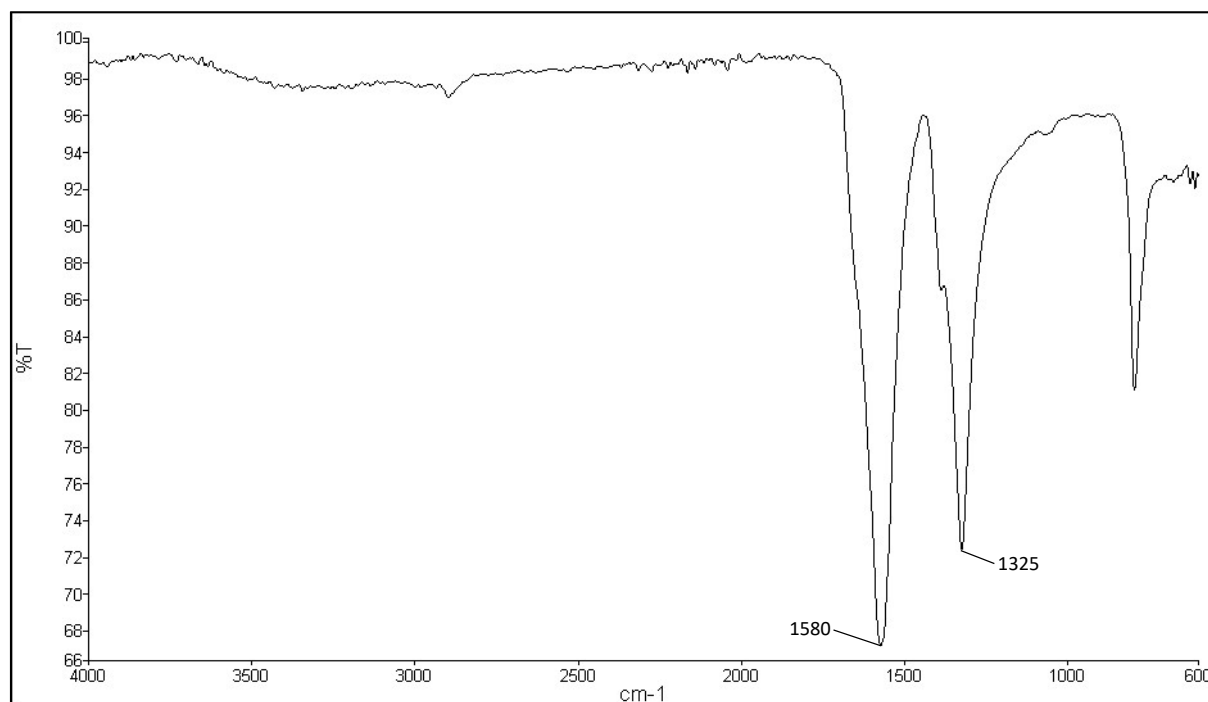
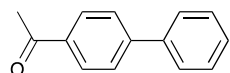


Figure 2. Infrared spectrum of Eco-Ni(HCOO)<sub>2</sub>

## II. Description of the products

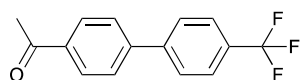
### 1. NMR spectroscopic description

#### 1-(1,1'-biphenyl)-4-yl)ethan-1-one – [1]



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.03 (d, *J* = 12 Hz, 2H), 7.69 (d, *J* = 8 Hz, 2H), 7.63 (d, *J* = 8 Hz, 2H), 7.49-7.40 (m, 3H), 2.64 (s, 3H).

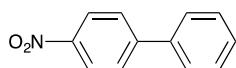
#### 1-(4'-(trifluoromethyl)-[1,1'-biphenyl]-4-yl)ethan-1-one – [2]



**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 8.05 (d, *J* = 8.3 Hz, 1H), 7.72 (s, 2H), 7.68 (d, *J* = 8.3 Hz, 1H), 2.64 (s, 2H).

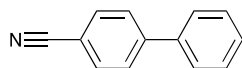
**<sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)** δ 197.63 (Ccarbonyl), 144.15 (Cq), 143.38 (Cq), 136.60 (Cq), 130.22 (q, *J* = 32.7 Hz, Cq), 129.05 (2C), 127.60 (2C), 127.45 (2C), 125.89 (q, *J* = 3.4 Hz, 2C), 124.14 (q, *J* = 272.2 Hz, CF<sub>3</sub>), 26.65 (s).

#### 4-nitrobiphenyl – [4]



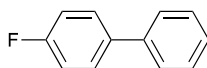
**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 8.34 – 8.21 (m, 2H), 7.76 – 7.68 (m, 2H), 7.65 – 7.57 (m, 2H), 7.55 – 7.27 (m, 4H).

#### 4-cyanobiphenyl – [5]



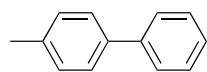
**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 7.73 – 7.64 (m, 4H), 7.61 – 7.55 (m, 2H), 7.51 – 7.44 (m, 2H), 7.43 – 7.39 (m, 1H).

#### 4-fluoro-1,1'-biphenyl – [6]



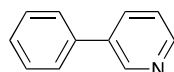
**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 7.73 – 7.61 (m, 4H), 7.47 (t, *J* = 8 Hz, 2H), 7.37 (t, *J* = 8 Hz, 1H), 7.24 (d, *J* = 9 Hz, 2H).

#### 4-methyl-1,1'-biphenyl – [7]



**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 7.66-7.63 (m, 2H), 7.56 (d, *J* = 8 Hz, 2H), 7.45 (t, *J* = 8 Hz, 2H), 7.34 (t, *J* = 8 Hz, 1H), 7.24 (d, *J* = 8 Hz, 2H), 2.38 (s, 3H).

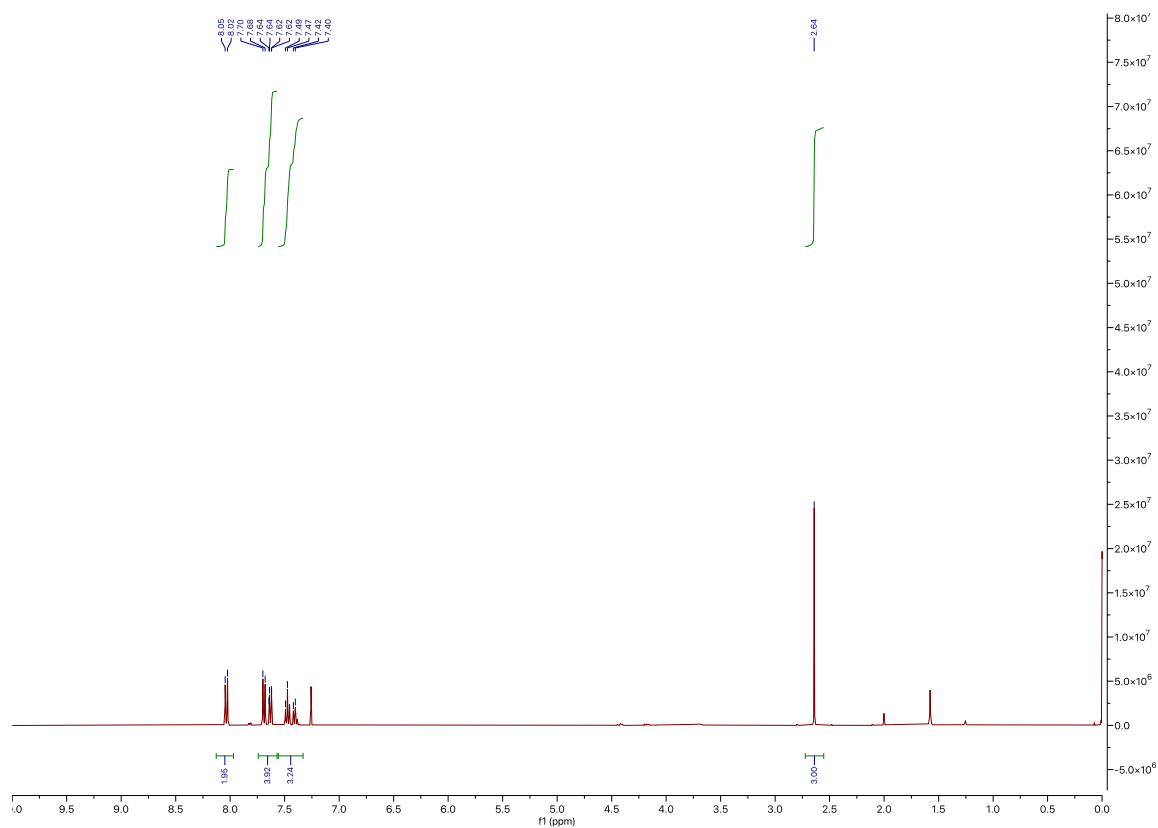
#### 3-phenylpyridine – [8]



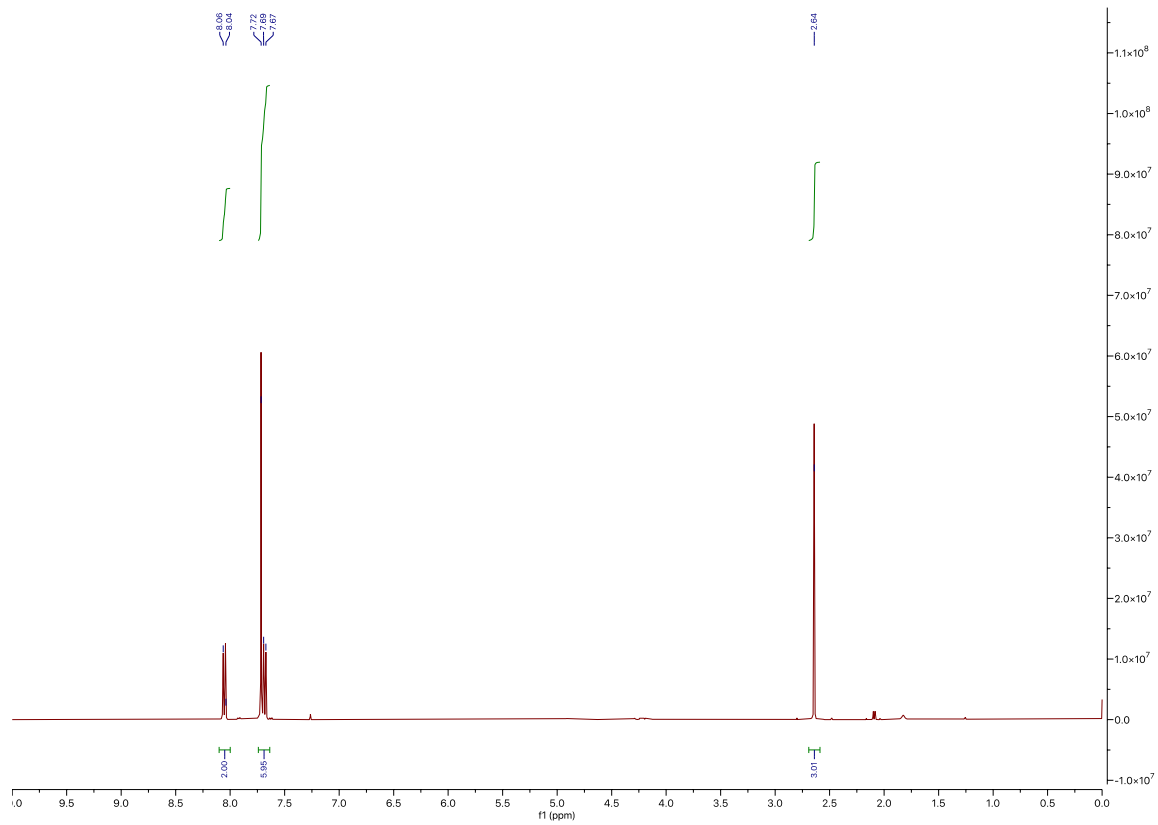
**<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)** δ 8.83 (d, *J* = 4 Hz, 2H), 8.57 (d, *J* = 4 Hz, 2H), 7.86 (d, *J* = 8 Hz, 1H), 7.89-7.83 (m, 1H), 7.57 (d, *J* = 8 Hz, 2H), 7.63 (d, *J* = 8 Hz, 2H), 7.47 (t, *J* = 8 Hz, 4 Hz, 2H), 7.45-7.33 (m, 3H).

## 2. NMR spectra

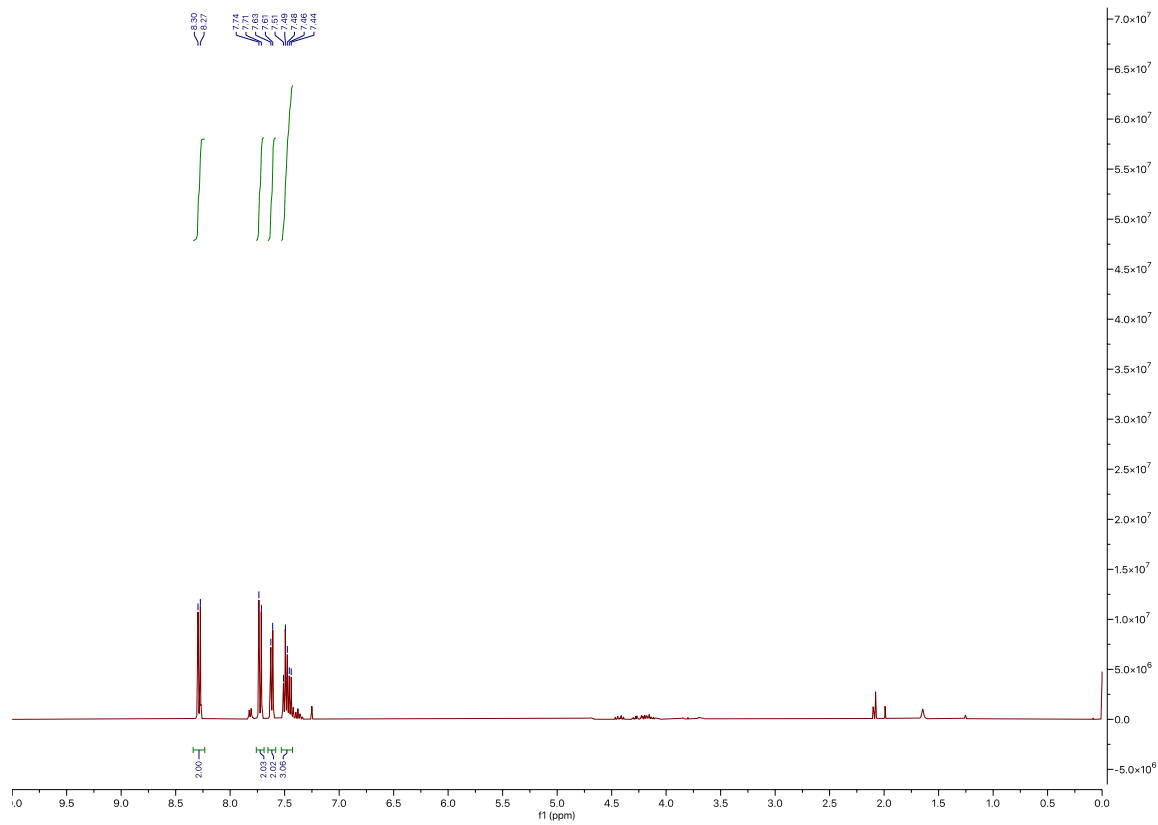
### 1-(1,1'-biphenyl)-4-yl)ethan-1-one – [1]



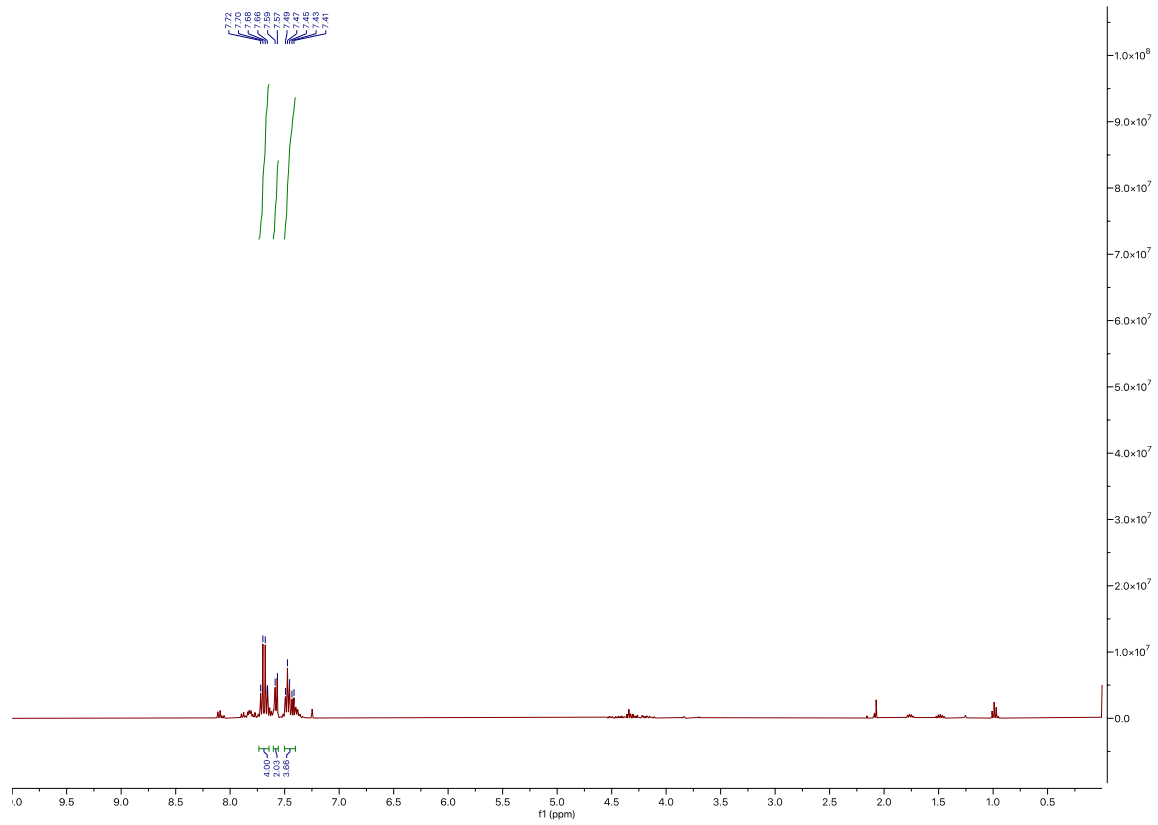
### 1-(4'-(trifluoromethyl)-[1,1'-biphenyl]-4-yl)ethan-1-one – [2]



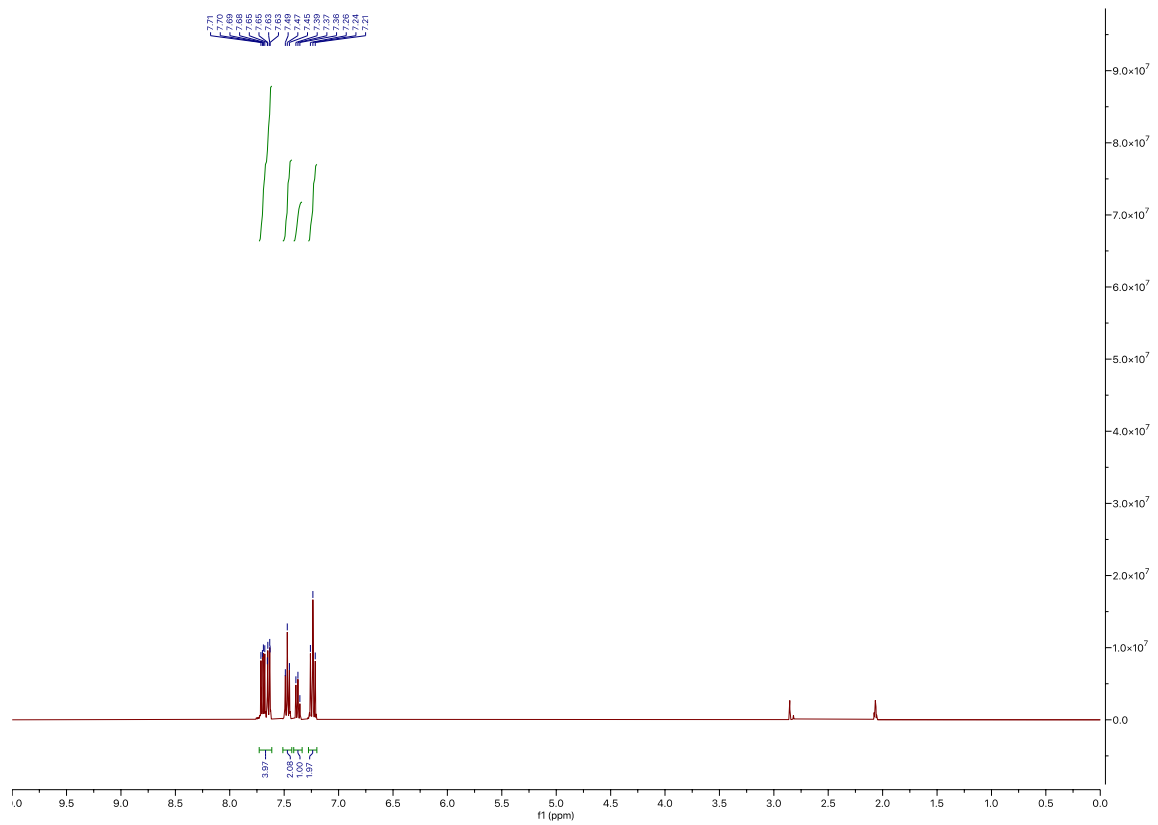
### 4-nitrophenyl – [4]



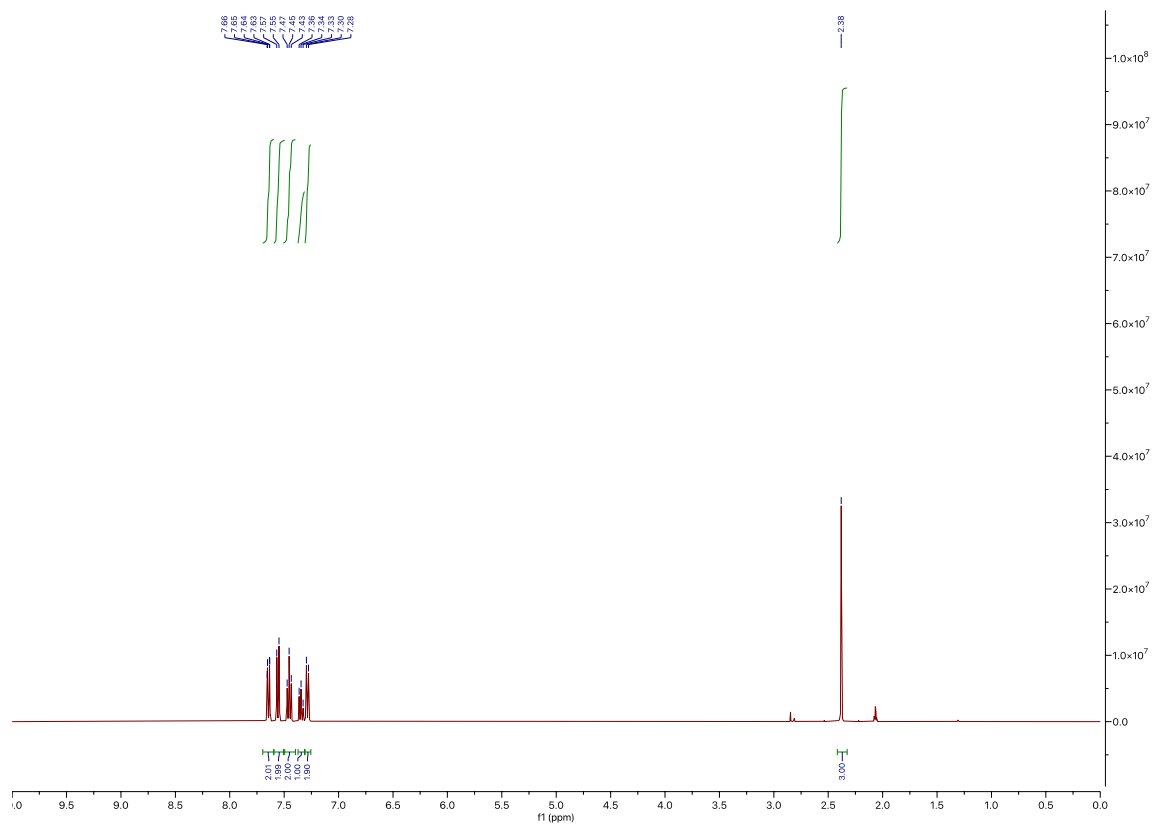
### 4-cyanobiphenyl – [5]



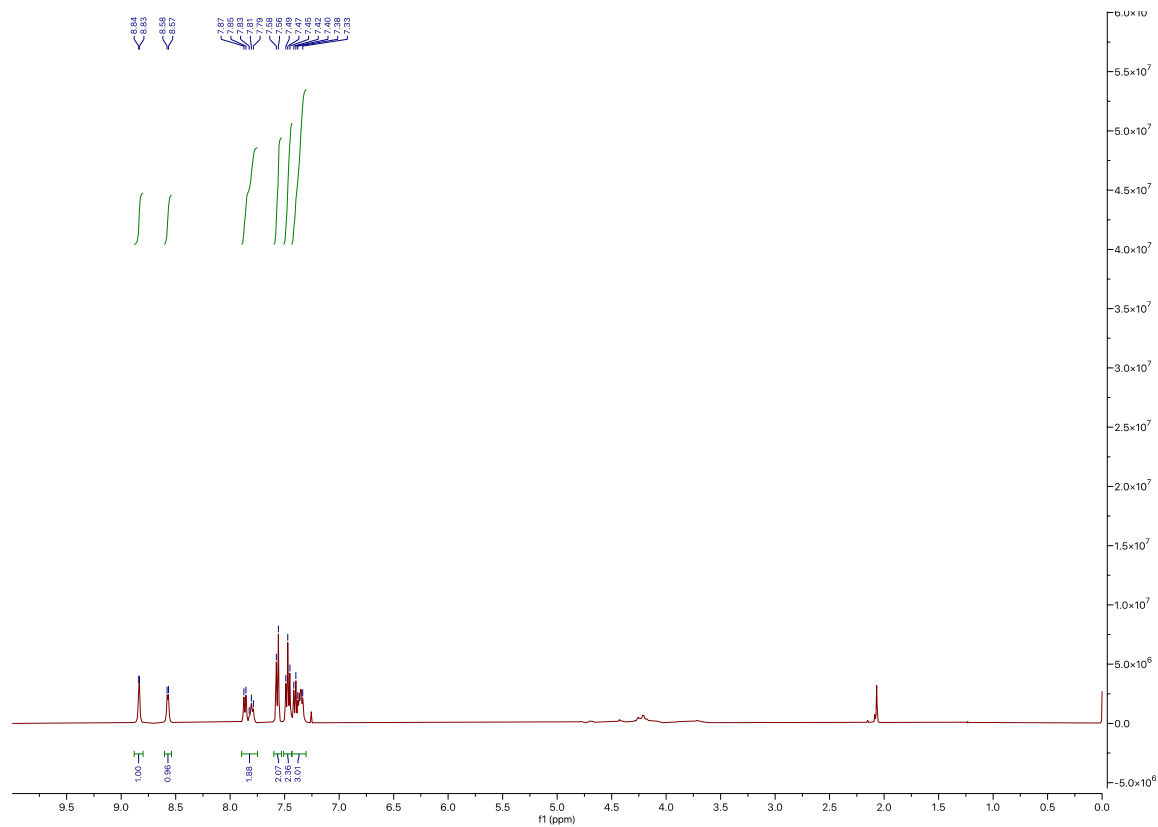
### 4-fluoro-1,1'-biphenyl – [6]



### 4-methyl-1,1'-biphenyl – [7]



### 3-phenylpyridine – [8]





### 3. GCMS and GCFID analyses of standards and synthesized products

#### 1-(1,1'-biphenyl)-4-yl)ethan-1-one – [1] – expected mass: 196.09

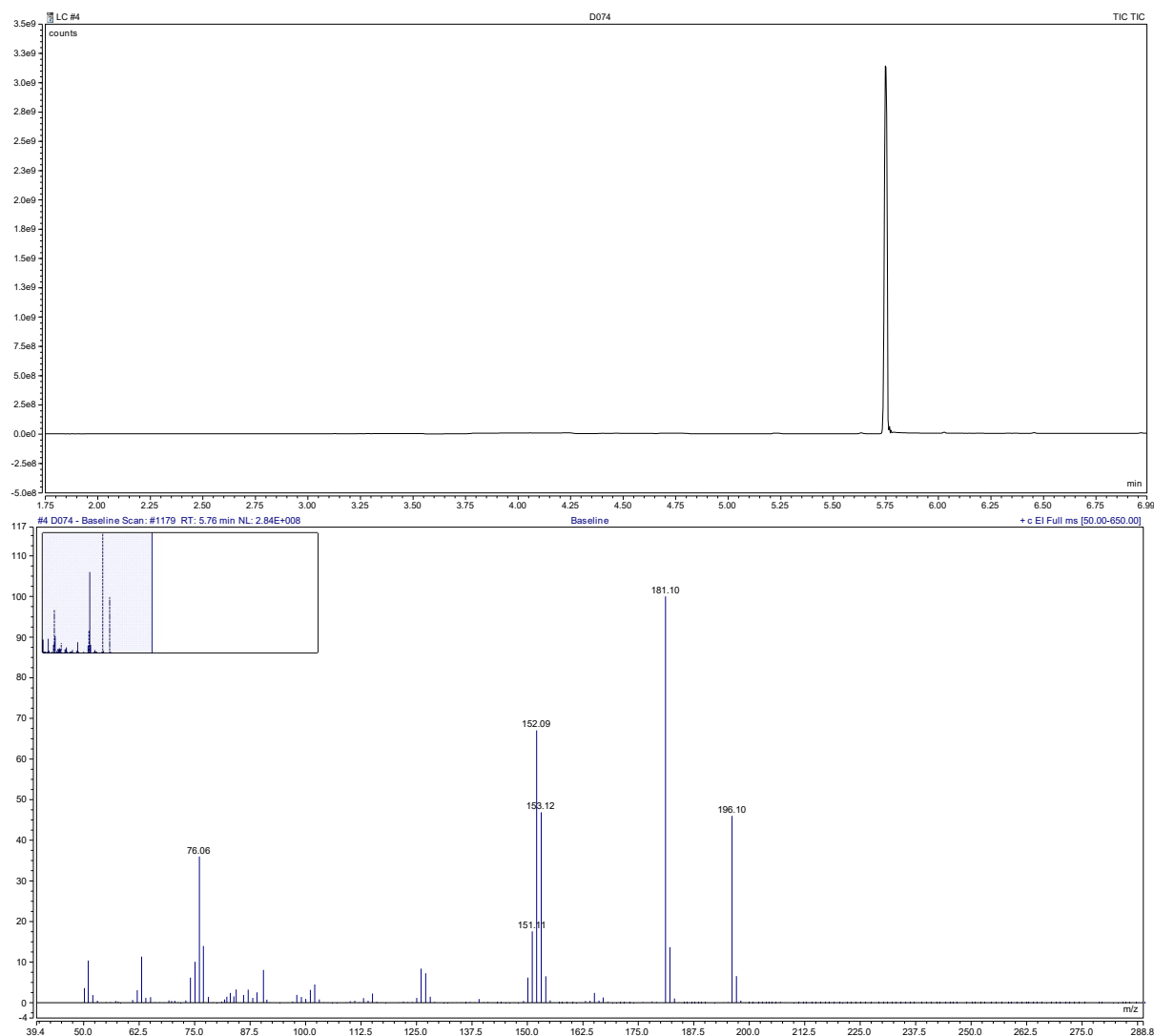


Figure 3. GCMS spectrum of synthesized product [1]

#### 1-(4'-(trifluoromethyl)-[1,1'-biphenyl]-4-yl)ethan-1-one – [2] – expected mass: 264.04

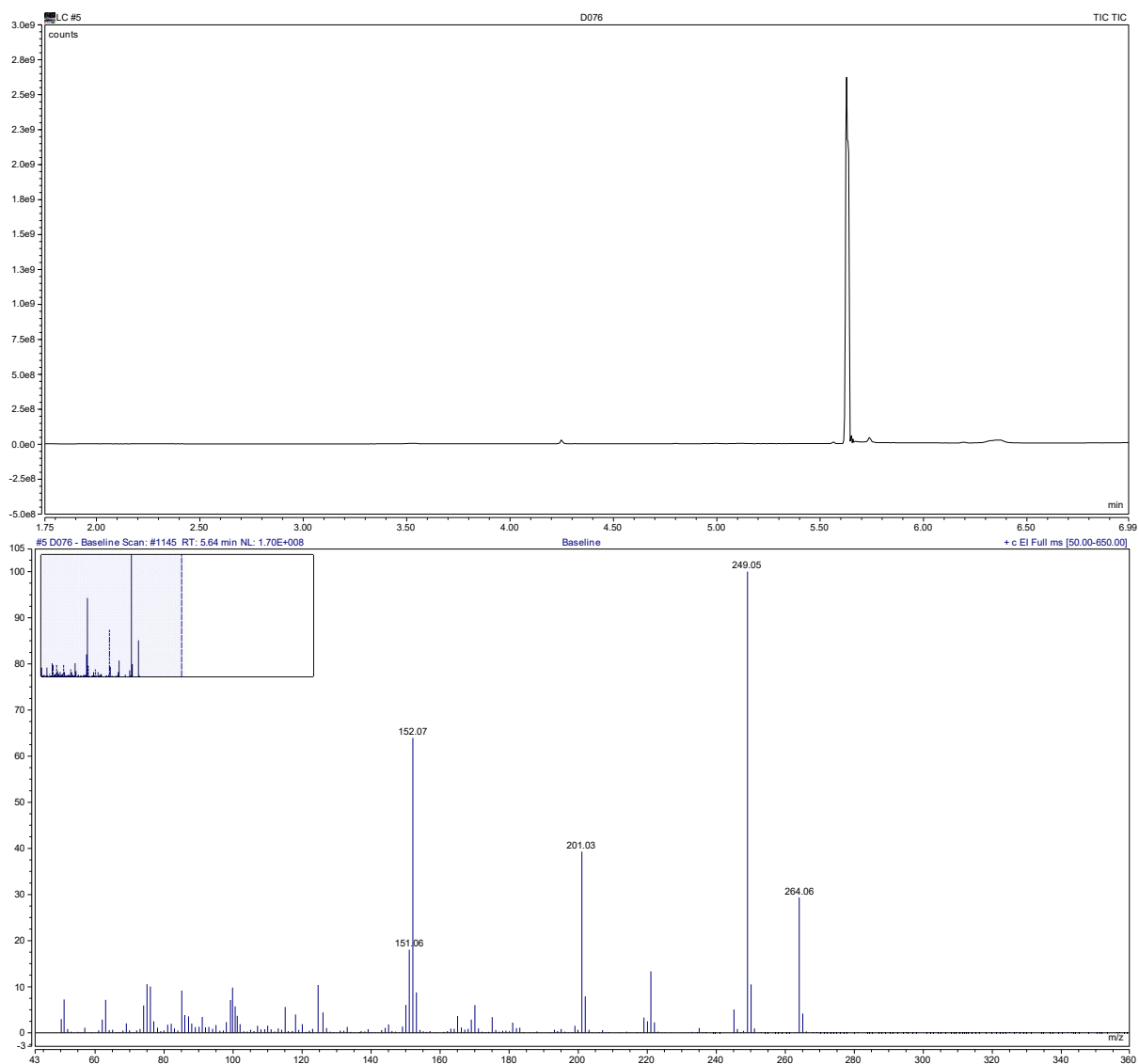


Figure 4. GCMS spectrum of synthesized product [2]

**4-nitrobiphenyl – [4] - expected mass: 199.06 Da**

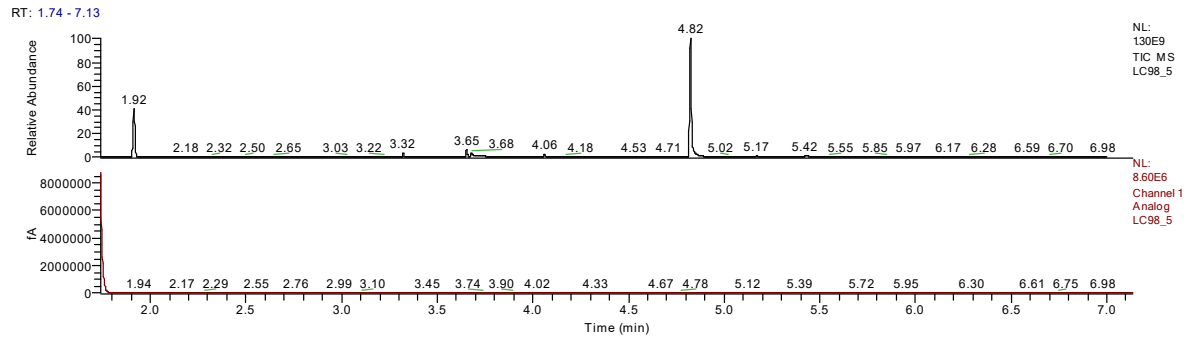


Figure 5. GCMS (black) & GCFID (red) spectra of synthesized product [4]

#### 4-cyanobiphenyl – [5] - expected mass: 179.07 Da

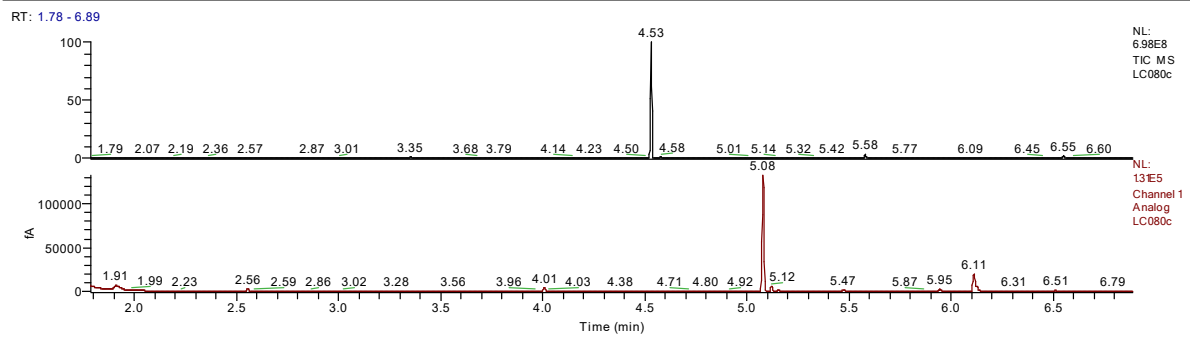


Figure 6. GCMS (black) & GCFID (red) spectra of synthesized product [5]

#### 4-fluoro-1,1'-biphenyl – [6] - expected mass: 172.07 Da

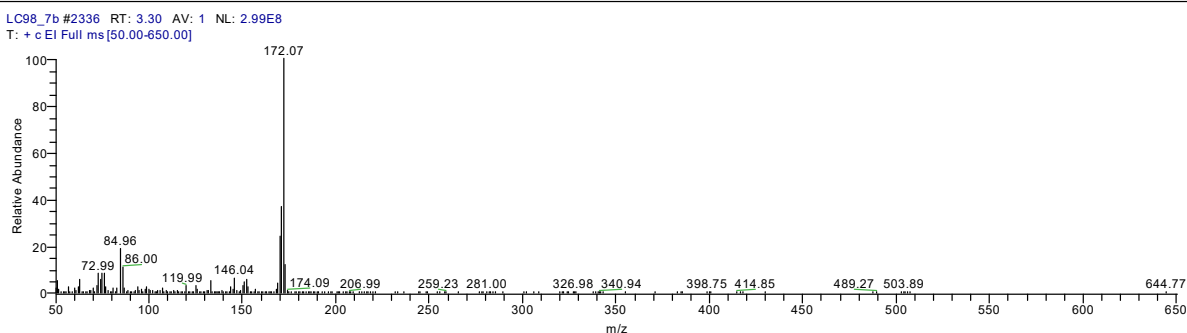
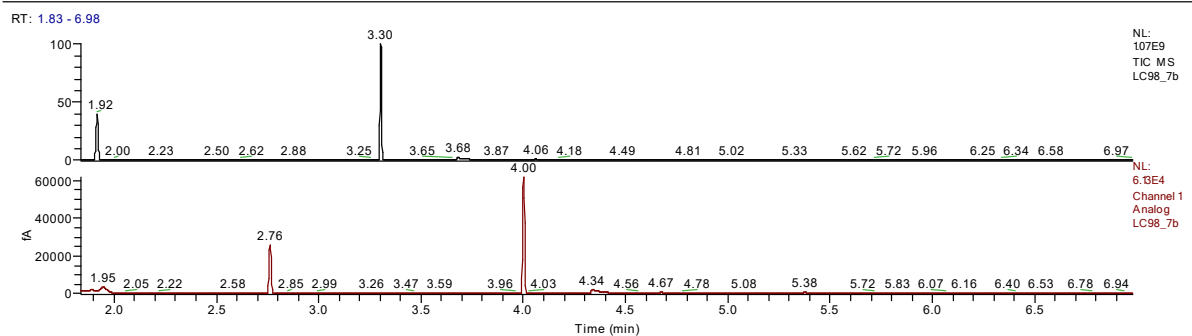


Figure 7. GCMS (black) & GCFID (red) spectra of synthesized product [6]

### 4-methyl-1,1'-biphenyl – [7] - expected mass: 168.09 Da

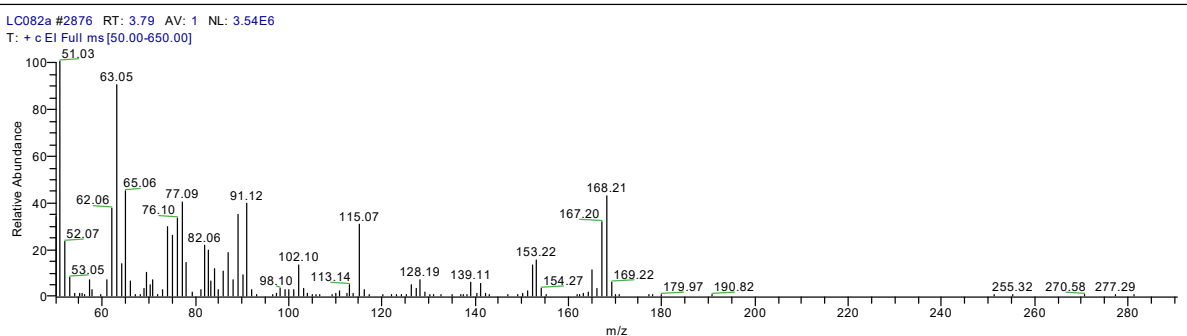
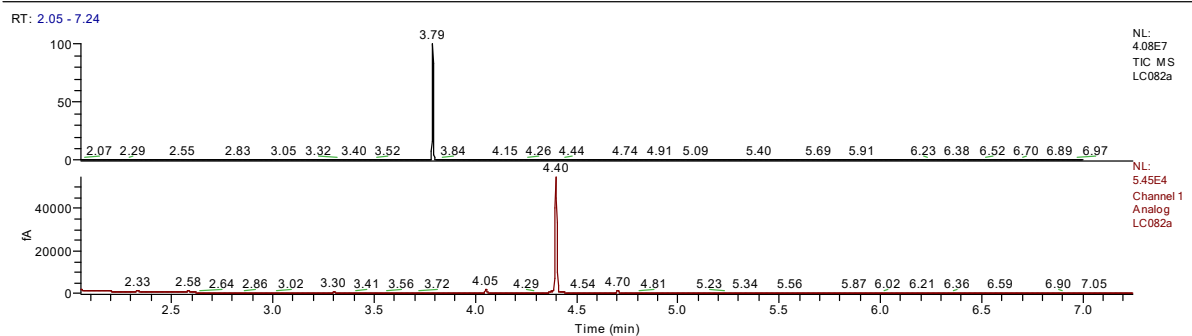


Figure 8. GCMS (black) & GCFID (red) spectra of synthesized product [7]

### 3-phenylpyridine – [8] – expected mass: 155.07 Da

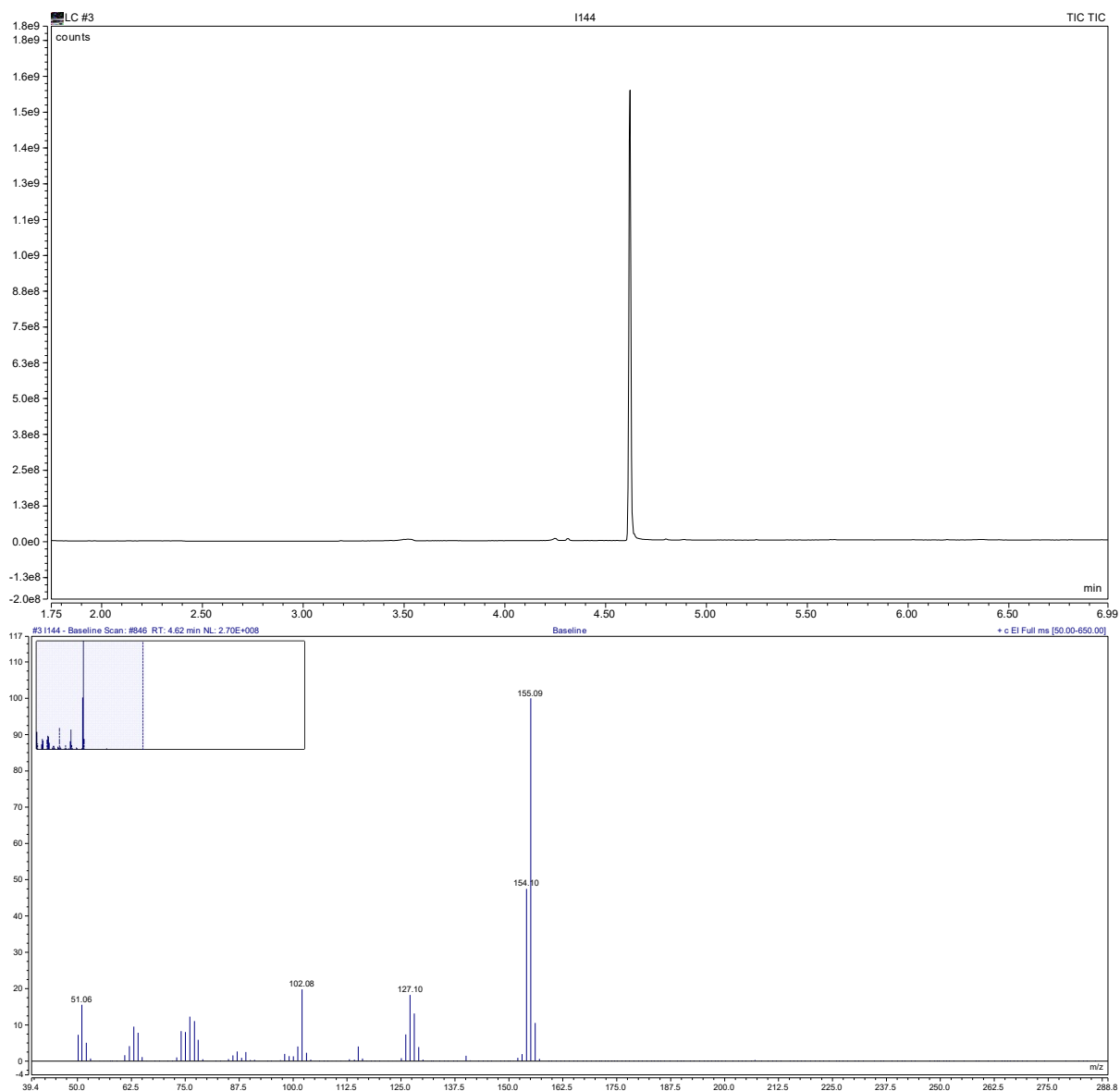
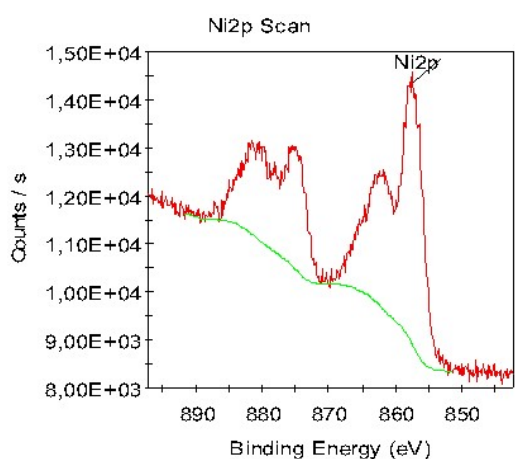


Figure 9. GCMS spectra of synthesized product [8]

### III. XPS analyses of the Eco-Ni(HCOO)<sub>2</sub> ecocatalyst

Eco-Ni(HCOO)<sub>2</sub> before reaction



Eco-Ni(HCOO)<sub>2</sub> after reaction

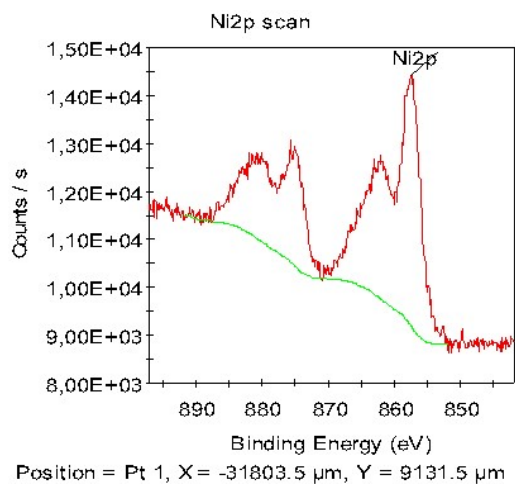


Figure 17. XPS analyses of the ecocatalyst before (left) and after the reaction (right)