## **Supporting Information**

## A significant enhancement of bulk charge separation in photoelectrocatalysis by ferroelectric polarization induced in CdS/BaTiO<sub>3</sub> nanowires

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(Summary of Content: 9 pages, 15 Figures)



Figure S1. XRD patterns of (a) the CdS/BTO, (b) BTO NWs samples around 45°.



Fig. S2 XPS spectra of the CdS/BTO NWs samples: (a) Survey scan, (b) Ti 2p, (c) O 1s, and (d) S 2p.



**Fig. S3** LSV curves of the 10 C, 15 C, 20 C CdS/BTO NWs samples in (a) positive voltage section (-0.1 V to +1.2 V), and (b) negative voltage section (0.2 V to -0.8 V).



Fig. S4 Dark current of the CdS/BTO NWs samples after different polarizations in (a) negative voltage

section (0.2 V to -0.8 V) and (b) positive voltage section (-0.1 V to +1.2 V).



**Fig. S5** LSV curves in the dark with (a) -0.2, -0.4, -0.6, -0.8, -1 V polarization for 1 min, (b) -0.5 V polarization for 2, 4, 6, 8, 10 min, (c) +0.2, +0.4, +0.6, +0.8, +1 V polarization for 1 min, and (d) +0.5 V polarization for 2, 4, 6, 8, 10 min, respectively.



Fig. S6 LSV curves with +3 V polarization of (a) the BTO NWs and (b) CdS/BTO NWs samples.



**Fig. S7** LSV curves of the pure BTO NWs samples after different polarizations in (a) negative voltage section (0.2 V to -0.8 V) and (b) positive voltage section (-0.1 V to +1.2 V).



Fig. S8 Photocurrent densities of the CdS/BTO NWs samples at (a) 0 V, (b) +0.5 V, and (c) -0.5 V bias.





Fig. S10 XRD patterns of the CdS/BTO NWs sample before and after polarization.



Fig. S11 SEM image of the CdS/BTO NWs sample after polarization.



Fig. S12 The equivalent circuit used for modeling the measured electrochemical response.  $R_p$  is related with the kinetics of the interfacial charge transfer reaction,  $R_s$  represents solution resistance. CPE and W represent the pseudocapacitance and Warburg impedance, respectively.

Table. S1 Summary of the impedance fitting data for the CdS/BTO NWs samples.

Sample	$R_p(\Omega)$	$R_{s}(\Omega)$	CPE <sub>p</sub> (mF)
dark current	1333.00	1.78520	0.06534
photocurrent	730.870	2.99120	0.11769
1st +3V 10min	1520.30	1.53840	0.05509
+1V 3min	1229.10	2.78370	0.01561
+0.5V 3min	900.730	1.46280	0.39652
-0.5V 3min	490.030	1.98660	0.02070
-1V 3min	402.040	1.53260	0.02089
1st -3V 10min	30.8510	3.05340	0.14456

2 <sup>nd</sup> -3V 10min	27.3430	2.74870	0.09363
2nd +3V 10min	2019.70	3.68880	0.06978



**Fig. S13** LSV curves of the CdS/BTO NWs samples measured in 1 M  $Na_2SO_3$  electrolyte in (a) negative voltage section (0.1 V to -0.8 V) and (b) positive voltage section (-0.1 V to +0.8 V).

As present in Fig. S13, under the unpolarized condition, the photocurrent at -0.8  $V_{(vs.RHE)}$  can reached 47.71 mA/cm<sup>2</sup>, which is 6.55 mA more than that of the dark current (41.16 mA/cm<sup>2</sup>) and the photocurrent at +0.8 $V_{(vs.RHE)}$  is 71.51 uA/cm<sup>2</sup>, 43.01 uA higher than the dark current (28.49 uA/cm<sup>2</sup>). When we look at the case of the first polarization at -3 V for 10 min polarization, the photocurrent at -0.8  $V_{(vs.RHE)}$  increases

to 66.75 mA/cm<sup>2</sup>, which is 7.9 mA/cm<sup>2</sup> higher than that of the dark current (58.84 mA/cm<sup>2</sup>). The photocurrent at +0.8 V<sub>(vs.RHE)</sub> is 170.97 uA/cm<sup>2</sup>, 162.36 uA/cm<sup>2</sup> larger than the dark current. The experiment adopts the method of applying polarization twice. Then the photocurrent at -0.8 V<sub>(vs.RHE)</sub> increases by 11.07 mA/cm<sup>2</sup> compared with the dark current, which are 74.3 and 85.37 mA/cm<sup>2</sup>, respectively. Meanwhile, the difference of the photocurrent and dark current at +0.8 V is 597.85 uA/cm<sup>2</sup>, which are 267.74 and 865.59 uA/cm<sup>2</sup>, respectively.



Fig. S14 IPCE spectra measured at 0  $V_{(vs.RHE)}$  for CdS/BTO NWs with no poling and after negative polarization with -3 V for 10 min.



**Fig. S15** ESR spectra of the pure BTO and CdS samples under light irradiation for detection of (a)•OH, (b) •-O<sub>2</sub>. ESR spectra of the CdS/BTO NWs samples with no poling and polarization at +3 V/-3 V for 10 min under light irradiation for detection of (c) •OH and (d)•-O<sub>2</sub>.