

Supplementary Information

Production of Nitrogen Doped Carbon Quantum Dots with Controllable Emission Wavelength, Excellent Sensing of Fe³⁺ in Aqueous Solution and Potential Application for Stealth QR Coding at Visible Regime

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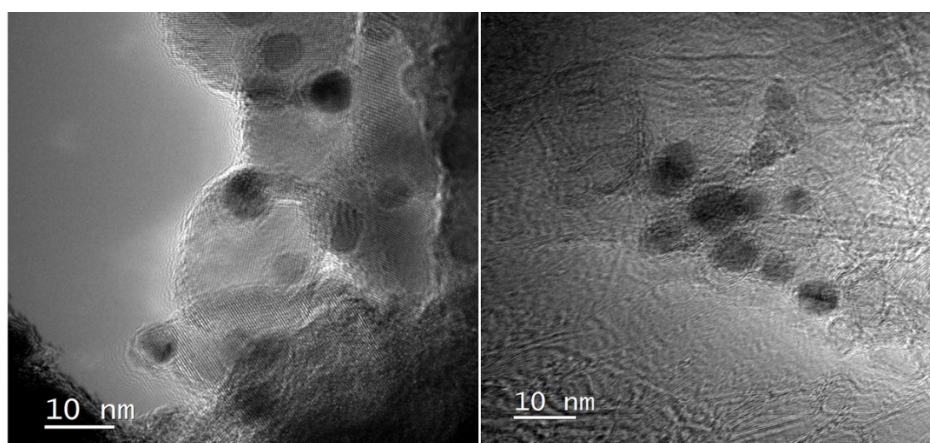


Figure S1. TEM images of ST3 made CQDs.

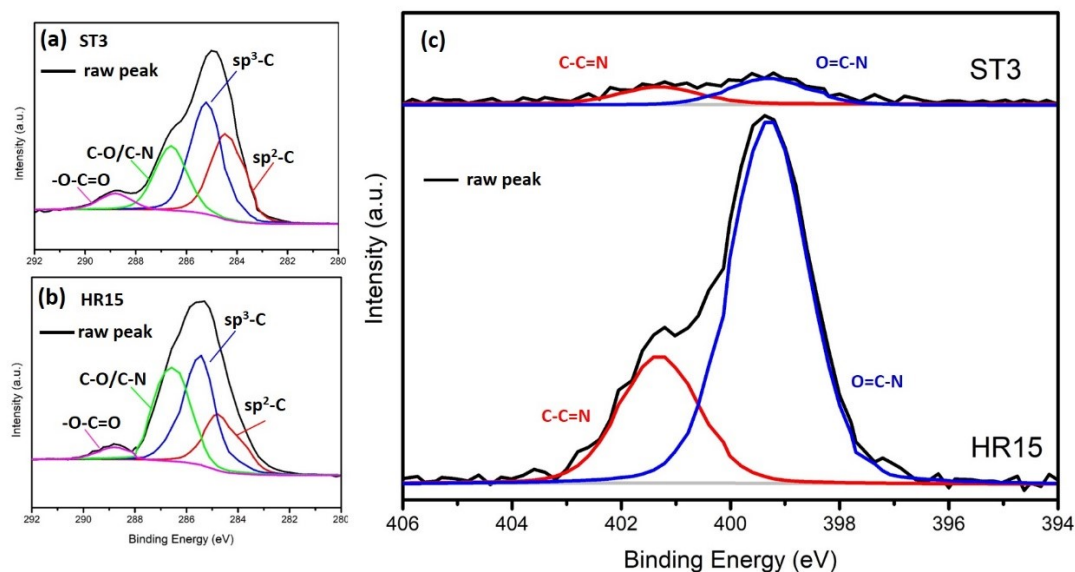


Figure S2. The surface functional groups of the N-CQDs were investigated using XPS spectra. The C1s high-resolution XPS spectrum of (a) ST3 and (b) HR15 shows three peaks assigned to carbon atoms in C=C (284.40 eV), C-C(285.2 eV), C-O (286.2 eV), and -O-C=O (288.8 eV). (c)The high-resolution N1s XPS spectrum (Figure 4C) exhibits two peaks located at 399.29 and 401.32 eV, which can be attributed to C=C-N and O=C-N, respectively.

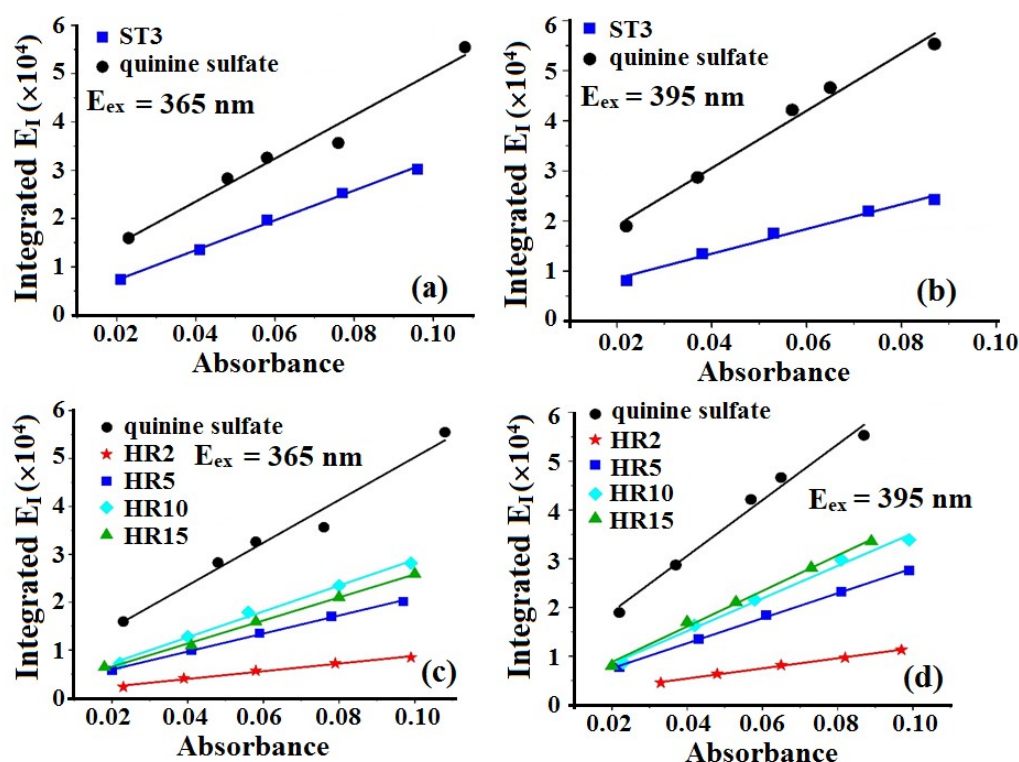


Figure S3. Integrated E_I - absorbance profiles of ST3 and quinine sulfate at $E_{ex} = 365$ nm (a) and 395 nm (b). Integrated E_I - absorbance profiles of HR and quinine sulfate at $E_{ex} = 365$ nm (c) and 395 nm (d).

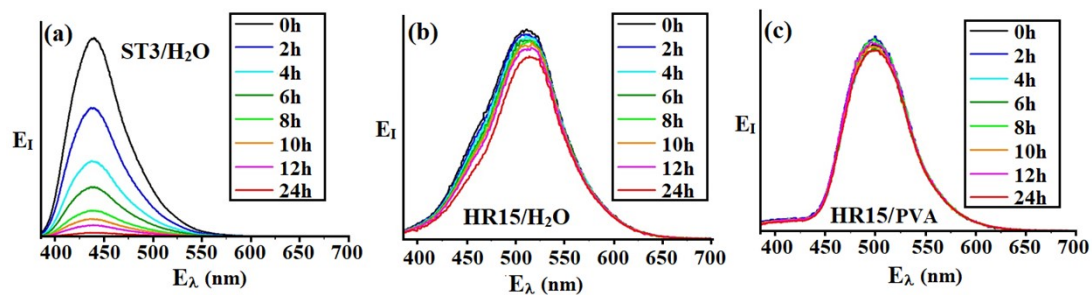


Figure S4. E_I - E_λ plots of ST3/H₂O (a), HR15/H₂O (b) and HR15/PVA at different irradiation times (c). $E_{ex} = 395$ nm.

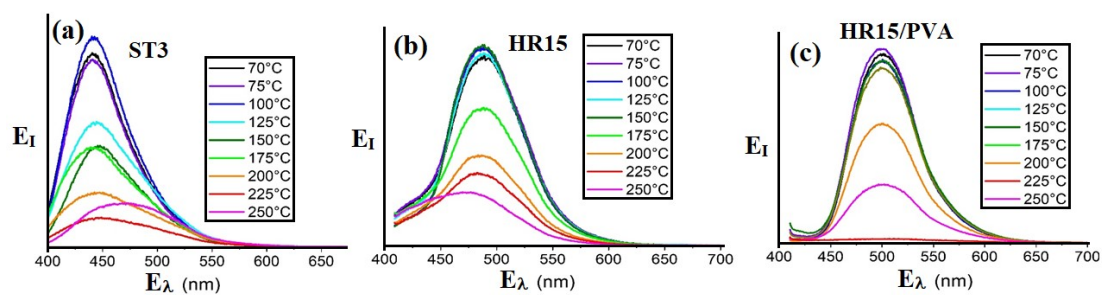


Figure S5. E_I - E_λ plots of ST3 (a), HR15 (b) and HR15/PVA (c) at different temperatures.