## **Supplementary information**

## Alkali-modified heterogeneous Pd-catalyzed synthesis of acids, amides and esters from aryl halides using formic acid as the CO precursor

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## **General Considerations**

Unless otherwise stated, all chemicals were purchased from Sigma-Aldrich and Associated Chemical Enterprises and used as received. The <sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (125 MHz) spectra were recorded on a Bruker-500 MHz spectrometer, with reported values relative to tetramethylsilane ( $\delta$  0.0) as the internal standard. Multiplets were assigned as singlet (s), doublet (d), triplet (t), quartet (q) and double doublet (dd). Unless otherwise stated, isolated yields were determined by <sup>1</sup>H-NMR spectroscopy and GC equipped with FID. All other measurements were performed at room temperature unless otherwise stated.

Isolated yield: 76%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  8.71 (dddd, J = 8.2, 1.7, 1.3, 0.3 Hz, 2H), 7.57 (tt, J = 7.2, 1.3 Hz, 1H), 6.42 (dddd, J = 8.2, 7.3, 1.2, 0.3 Hz, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  178.2, 149.5, 146.6, 146.9, 145.8.

Isolated yield: 69%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): δ 8.21 (ddd, J = 8.4, 1.3, 0.3 Hz, 2H), 6.57 (ddd, J = 8.4, 1.3, 0.3 Hz, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 178.3, 165.3, 161.2, 148.5, 143.6.

Isolated yield: 77%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  8.41 (dd, J = 8.3, 1.3 Hz, 2H), 8.22 (dd, J = 8.3, 1.1 Hz, 2H), 3.36 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  178.8, 169.3, 149.2, 138.7, 136.3, 59.1, 57.2, 55.1.



Isolated yield: 80%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  7.81 (dd, J = 8.2, 1.5 Hz, 2H), 7.51 (dd, J = 8.2, 1.3 Hz, 2H), 2.62 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  173.5, 142.4, 141.7, 138.2, 22.6, 16.9.

Isolated yield: 55%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  8.28 (dd, J = 8.2, 1.5 Hz, 2H), 7.21 (dd, J = 8.2, 1.3 Hz, 2H), 2.46 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  175.2, 157.9, 153.4, 143.5, 35.9.

Isolated yield: 66%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): δ 7.35 (dd, J = 8.2, 1.5 Hz, 2H), 7.24 (dd, J = 8.2, 1.3 Hz, 2H), 2.52 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 177.9, 159.4, 157.2, 143.6, 122.3, 118.5, 39.3.

Isolated yield: 74%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  7.71 (ddq, J = 2.0, 1.7, 0.4 Hz, 1H), 7.40-7.46 (dddt, J = 7.5, 1.7, 1.6, 0.4 Hz, 2H), 7.23 (ddd, J = 8.3, 2.0, 0.4 Hz, 2H), 7.06-7.18 (dddd, J = 7.5, 6.9, 2.5, 0.4 Hz, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  176.3, 144.5, 143.2, 137.6, 136.7, 114.7.



Isolated yield: 22%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  8.45 (ddd, J = 8.4, 1.7, 0.4 Hz, 2H), 8.27 (ddd, J = 8.4, 1.2, 0.4 Hz, 2H), 7.67-7.84 (dddd, J = 7.3, 1.2, 1.0, 0.4 Hz, 5H, tdd, J = 7.4, 1.7, 0.4 Hz), 4.18 (s, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  176.6, 148.1, 141.3, 134.3, 132.7, 40.2.



Isolated yield: 43%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): δ 8.17 (ddd, J = 8.2, 1.6, 0.4 Hz, 2H), 7.53-7.75 (dddd, J = 7.3, 1.2, 1.0, 0.4 Hz, 5H, tdd, J = 7.4, 1.7, 0.4 Hz, tt, J = 7.3, 1.2 Hz), 2.47 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 174.1, 146.7, 141.2, 138.1, 132.1, 129. 1, 58.2, 31.3.



Isolated yield: 48%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): δ 8.74 (dddd, J = 8.2, 1.7, 1.3, 0.2 Hz, 2H), 7.72 (tt, J = 7.3, 1.2 Hz, 1H), 7.43 (dddd, J = 8.2, 7.0, 1.2, 0.3 Hz, 2H), 4.18 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 174.5, 130.3, 52.5.

OMe 4b  $\mathbf{C}^{\dagger}$ 

Isolated yield: 35%

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  7.64 (dd, J = 8.2, 1.3 Hz, 2H), 7.37 (dd, J = 8.2, 1.3 Hz, 2H), 3.76 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  172.3, 162.3, 132.9, 130.8, 56.4.



Fig. S1 TEM image of Co<sub>3</sub>O<sub>4</sub>.



Fig. S2 TGA spectra of the catalyst.



Fig. S3 TGA spectra of the catalyst.



Fig. S4 FT-IR spectra of the catalyst.



Fig. S5 GC-TCD spectra of the HCOOH decomposition. Reaction conditions:  $B_1$  (1 equiv.),  $X^*$  (0.5 g), HCOOH (3 mL), Me-THF (3 mL), 110-130 °C and 16 h.



**Fig. S6** 3% Pd/Co<sub>3</sub>O<sub>4</sub>-Li catalyzed carbonylation reaction with formic acid in an open system experiment to show the formation of CO. Reaction conditions: Room temperature with all reactants in place (**A**), and 110-130 °C, 12-16 h (**B**). The inflated balloon in (**B**) shows the evidence of the CO release.



Fig. S7 Hot filtration test for the 3% Pd/Co $_3O_4$ -Li catalyst.







210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)



 $\overbrace{6.63}^{8.42}$ 

 $\xleftarrow{}^{2.13}_{2.10}$ 



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210	200	190	180	170	160	150	140	130	120	110	100 f1 (ppm	90 )	80	70	60	50	40	30	20	10	0	-10	1



























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