Supporting Information

Stable cycling of Prussian blue/Zn battery in a nonflammable aqueous/organic hybrid electrolyte

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Table S1. Comparison of cycling performance of various battery systems.

Cathode//Anode	Electrolyte	Current density	Initial capacity (mAh g^{-1})	Cycle number	Capacity retention	Reference
FeHCF//Zn	**Aqueous 1 m NaTfO and 0.1 m $Zn(TfO)_2$ in ACN/H_2O**	**1 A g^{-1}**	**69.1**	**19000**	**51.4%**	**This work**
CoHCF//Zn	4 M $Zn(BF_4)_2$ in in-situ poly(1,3-dioxolane, DOL) SPE	0.5 A g^{-1}	~160	20000	~66%	[1]
CoHCF//Zn	PVDF-HFP-PEO SPE with 2 M $Zn(BF_4)_2$/ [EMIM]BF_4 ionic liquid	2 A g^{-1}	103.4	30000	90%	[2]
MnHCF//Al	Aqueous 1 M $Al(TfO)_3$	0.5 A g^{-1}	82.1	100	69.5%	[3]
KCuHCF//Zn	Aqueous 0.5 M $Zn(TfO)_2$ in TEP/H_2O	1C	73.3	1000	74%	[4]
Graphite//Zn	1 M $Zn(TFSI)_2$ in ACN	1 A g^{-1}	42	4000	82%	[5]
δ-MnO_2//Zn	0.3 M $ZnCl_2$ in ChCl/urea	0.1 A g^{-1}	72	150	~50%	[6]

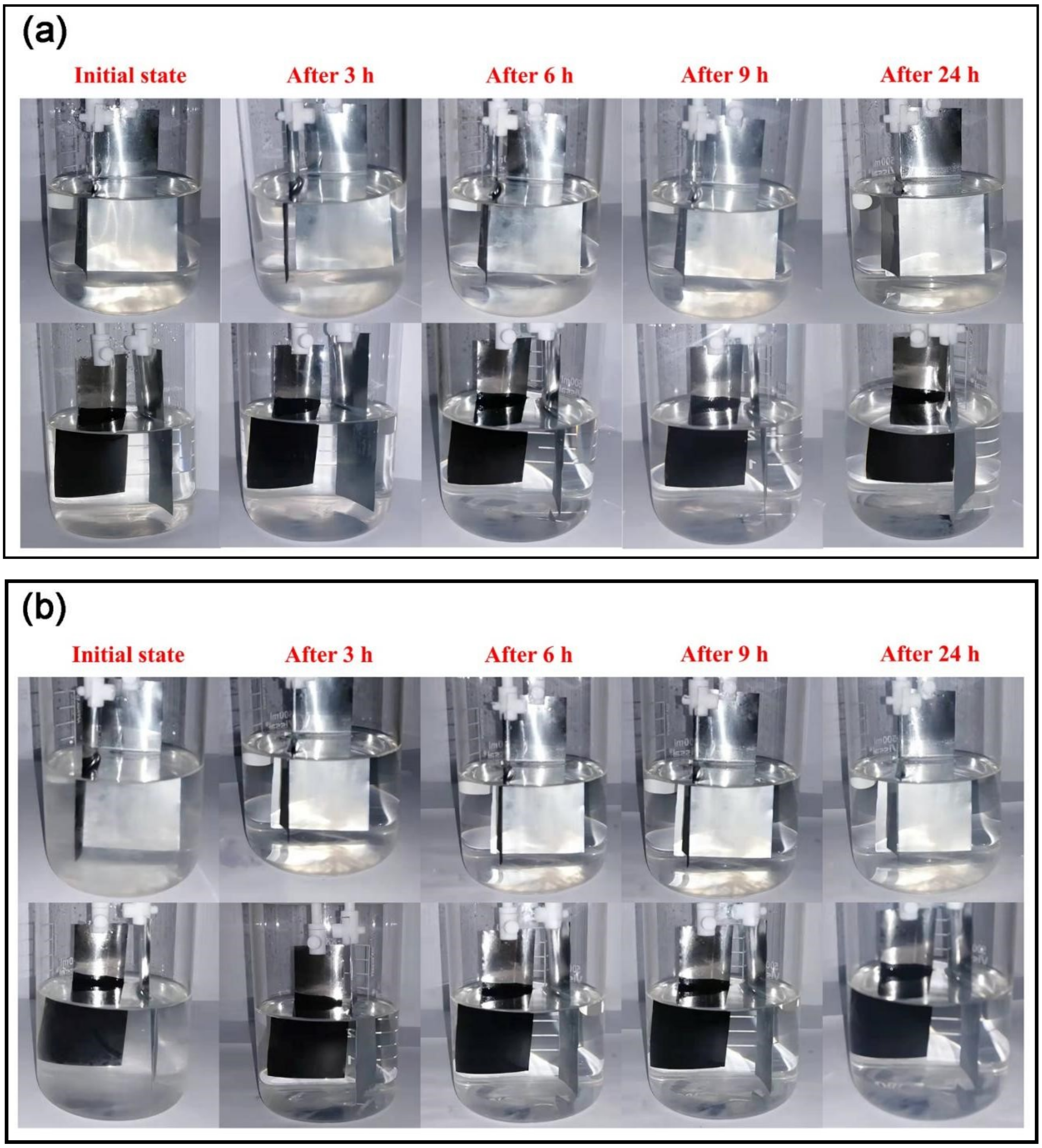

Fig. S1. Gas evolution experiments of the aqueous battery in the idle mode using ACN-50 electrolyte. (a) Images of the aqueous battery after rest for various durations after charge to 1.2 V. (b) Images of the aqueous battery after rest for various duration after subsequent discharge to 0.7 V.

References

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