

Electronic Supporting Information

Supported copper on a diamide-diacid-bridged PMO: An efficient hybrid catalyst for the cascade oxidation of benzyl alcohols/Knoevenagel condensation

*Ehsan Valiey, Mohammad G. Dekamin**

*^aPharmaceutical and Heterocyclic Compounds Research Laboratory, Department of Chemistry,
Iran University of Science and Technology, Tehran, 16846-13114, Iran.*

**E-mail: mdekamin@iust.ac.ir*

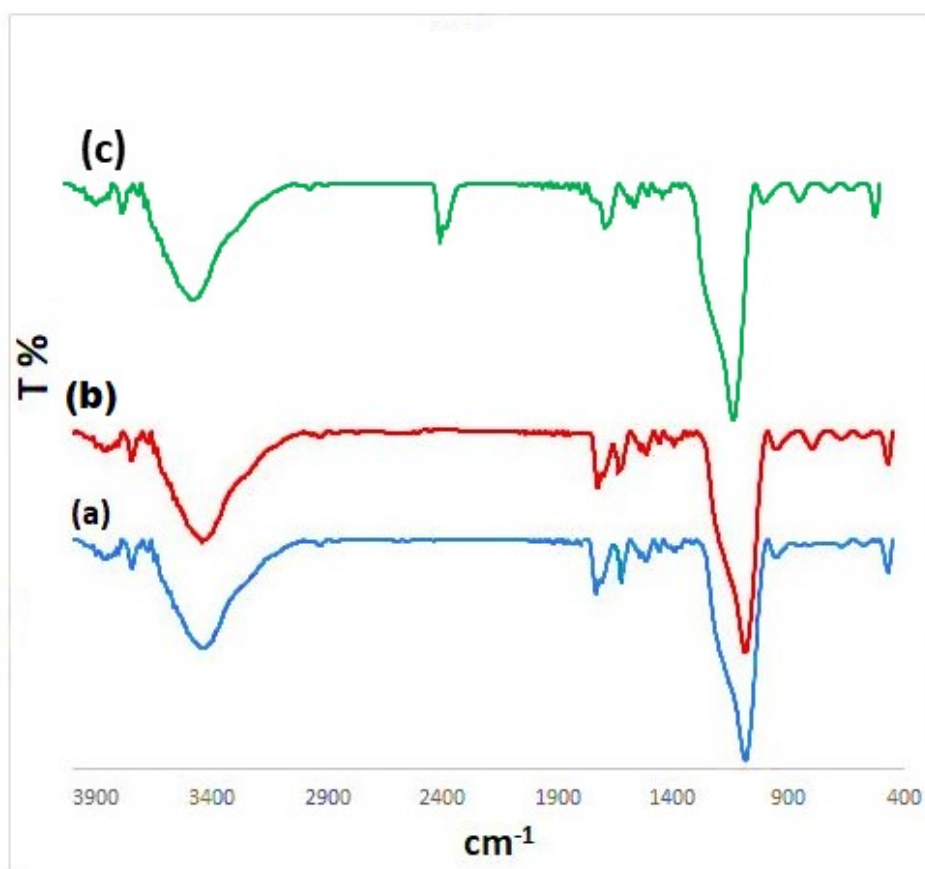


Fig. S1. FTIR spectra of the EDTAD-PMO (a), fresh Cu@EDTAD-PMO mesoporous catalyst (**1**, b) and the recycled catalyst after six consecutive runs (c).

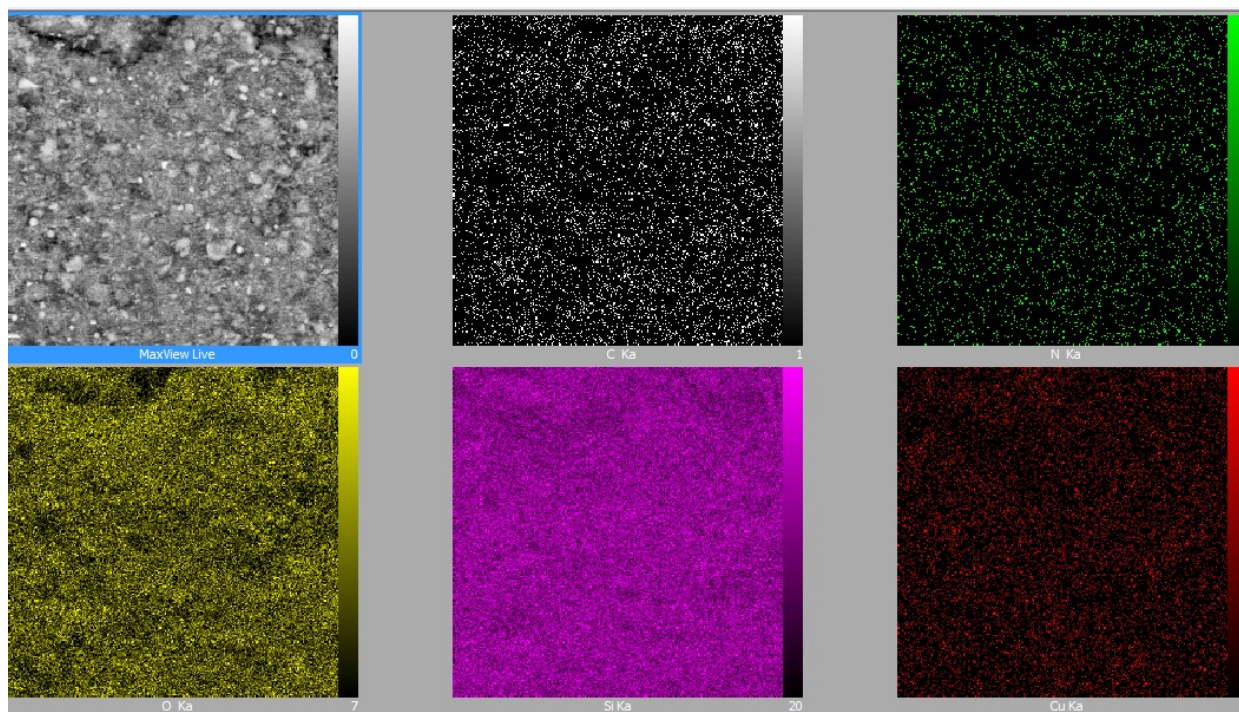
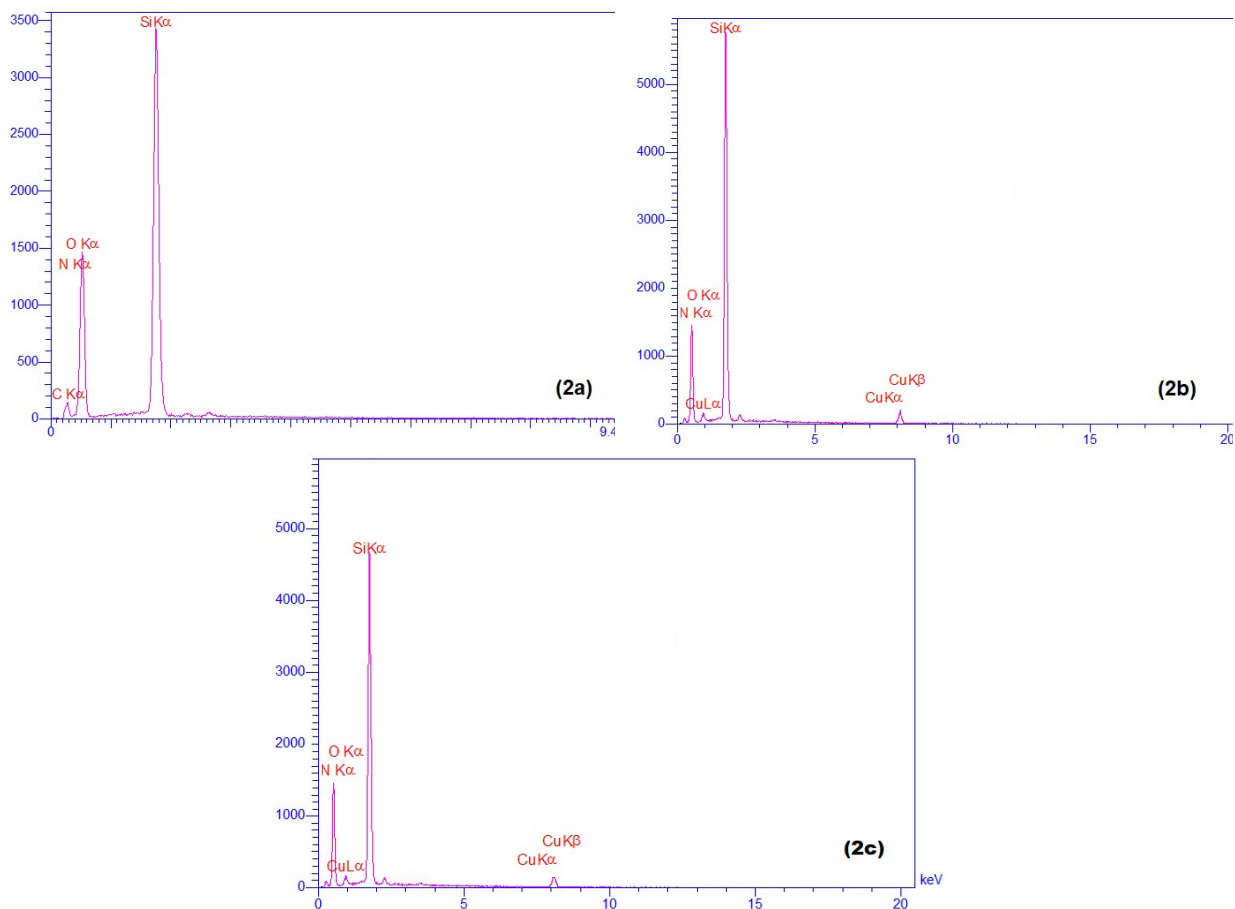


Fig. S2. EDX spectra of the EDTAD-PMO (**2a**), fresh Cu@EDTAD-PMO (**1**, **2b**) and the recycled Cu@EDTAD-PMO catalyst (**1**, **2c**) along with the elemental mapping of the fresh Cu@EDTAD-PMO (**1**).

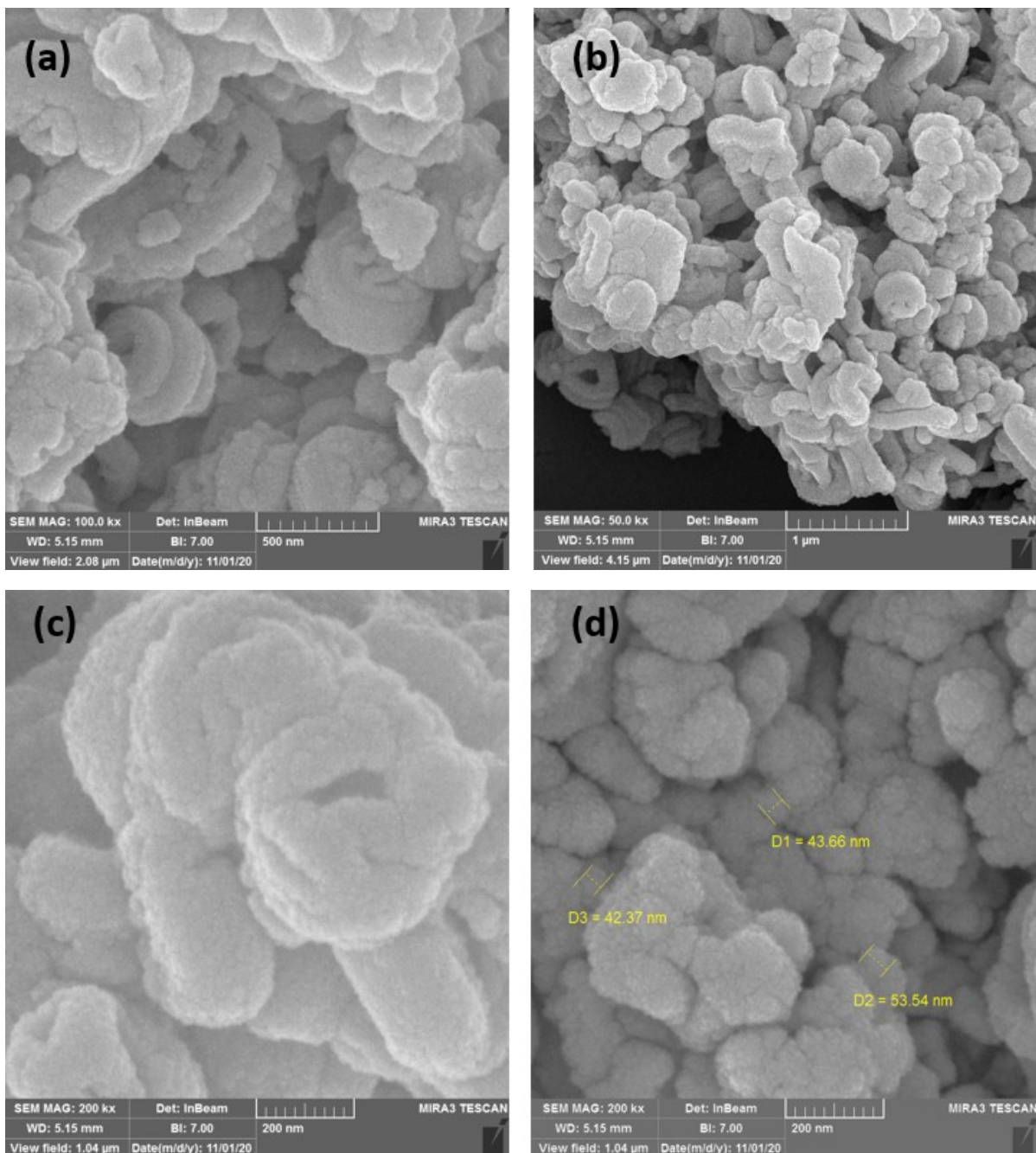


Fig. S3. FESEM images of the Cu@EDTAD-PMO nano-ordered catalyst (1).

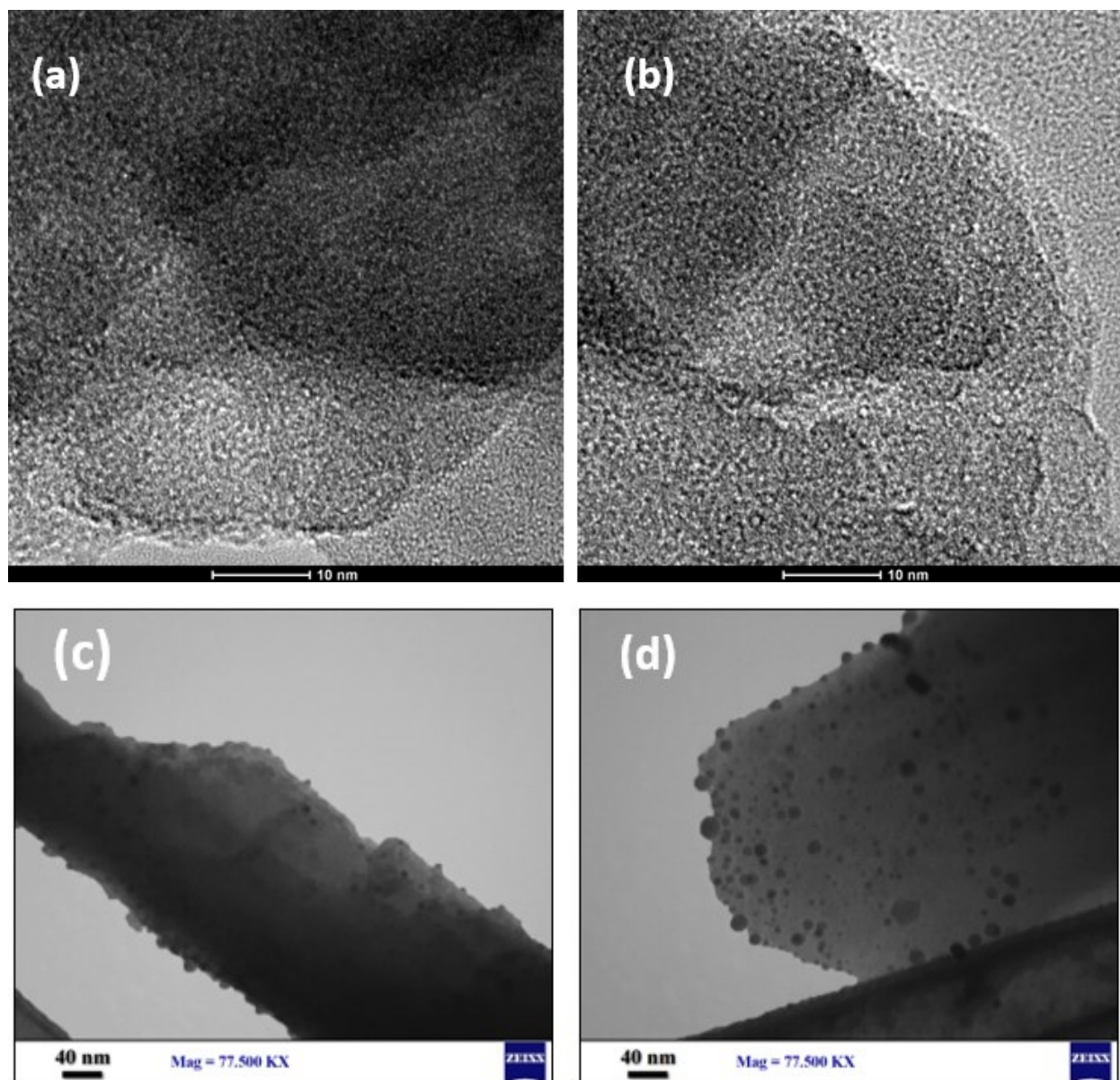


Fig. S4. HRTEM images of the Cu@EDTAD-PMO nano-ordered catalyst (1).

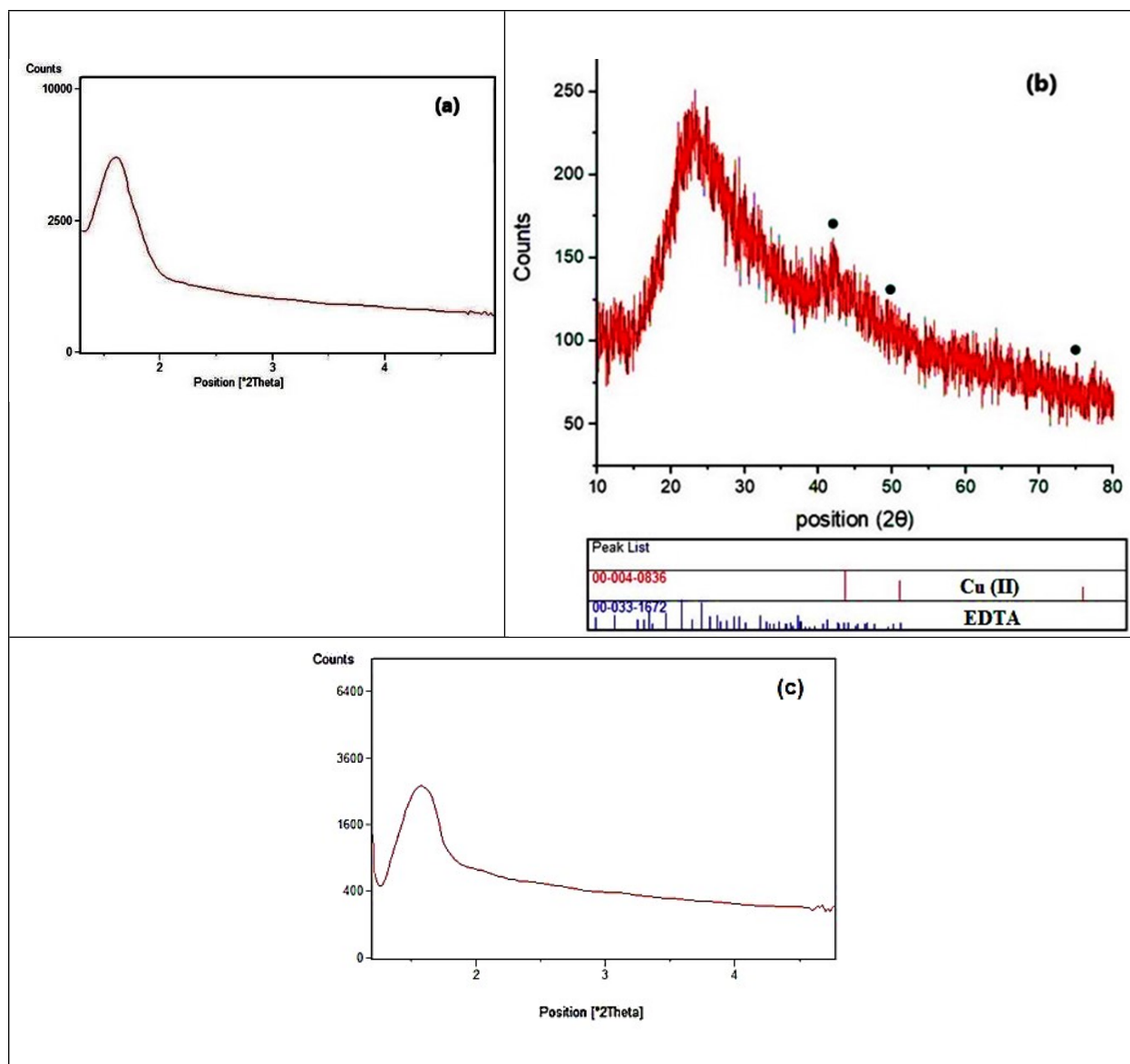


Fig. S5. Low-angle (**5a**) and wide-angle (**5b**) XRD patterns of the fresh Cu@EDTAD-PMO catalyst (**1**); low-angle XRD pattern (**5c**) of the recycled Cu@EDTAD-PMO catalyst (**1**) from the model reaction after six consecutive runs.

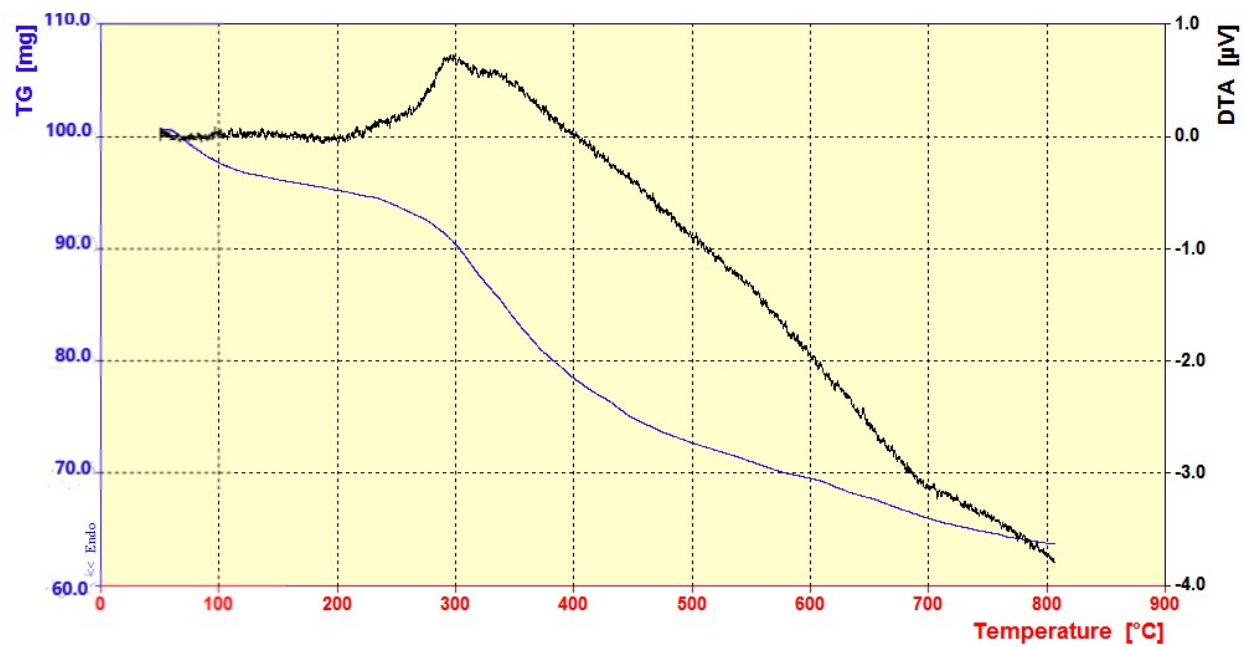


Fig. S6. TGA and DTA curves for the Cu@EDTAD-PMO nanomaterial (1).

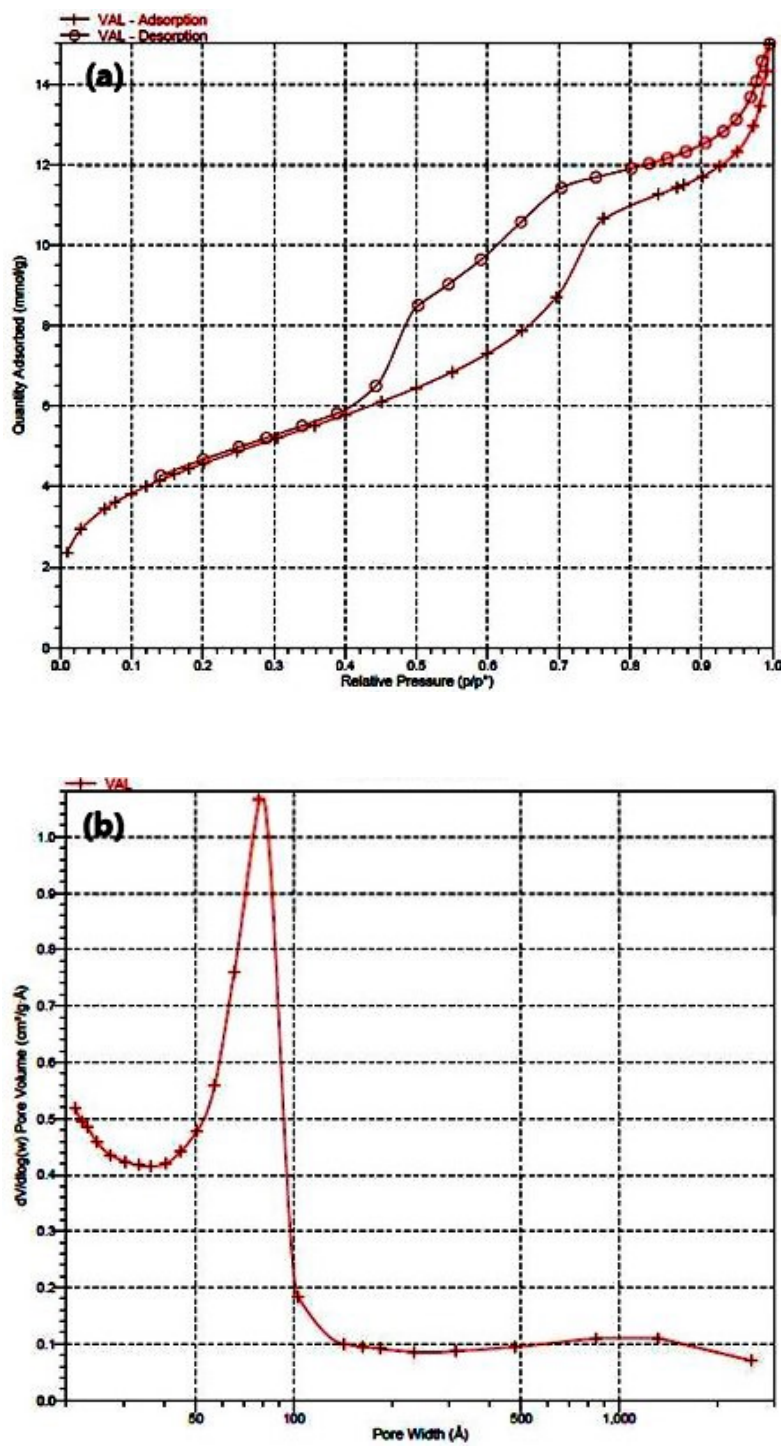


Fig. S7. (a) N_2 adsorption–desorption, and (b) pore size distribution isotherms of the Cu@EDTAD-PMO mesoporous material (1).

Spectral characterization of compounds 2c and 2g

2-(4-methylbenzylidene) malononitrile (2c)

FTIR (KBr, cm^{-1}): 3034, 2962, 2223, 1581, 1554, 1485; ^1H NMR (500MHz, CDCl_3): δ (ppm) 7.82 (d, $J = 8.0$ Hz, 2H), 7.73 (s, 1H), 7.34 (d, $J = 8.0$ Hz, 2H), 2.47 (Me, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ (ppm) 146.4, 130.9, 130.3, 128.4, 128.3, 114, 112.8, 81.

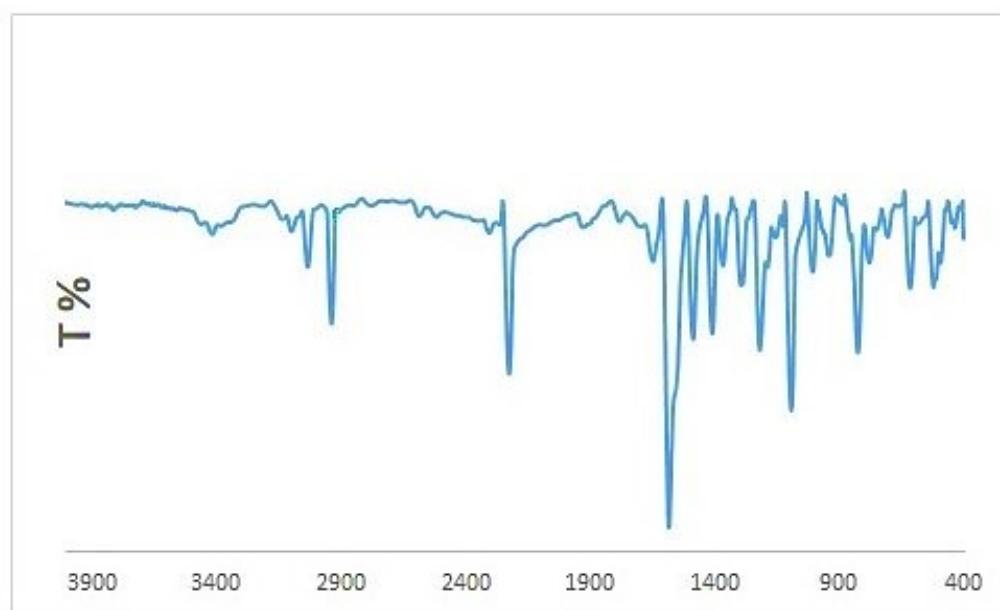


Fig. S8. FTIR of the 2-(4-methylbenzylidene) malononitrile (**2c**).

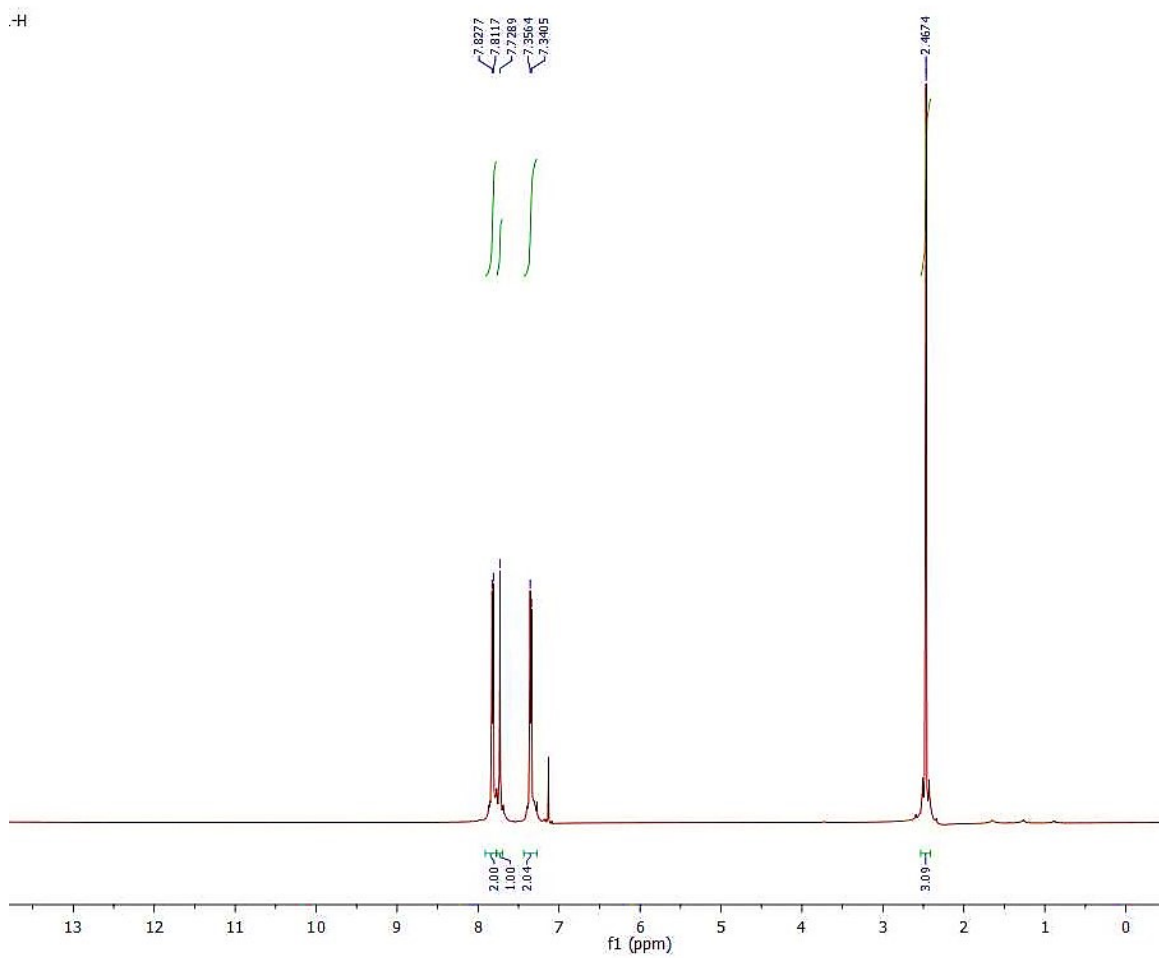


Fig. S9. ¹H NMR of the 2-(4-methylbenzylidene)malononitrile (**2c**).

V1-C

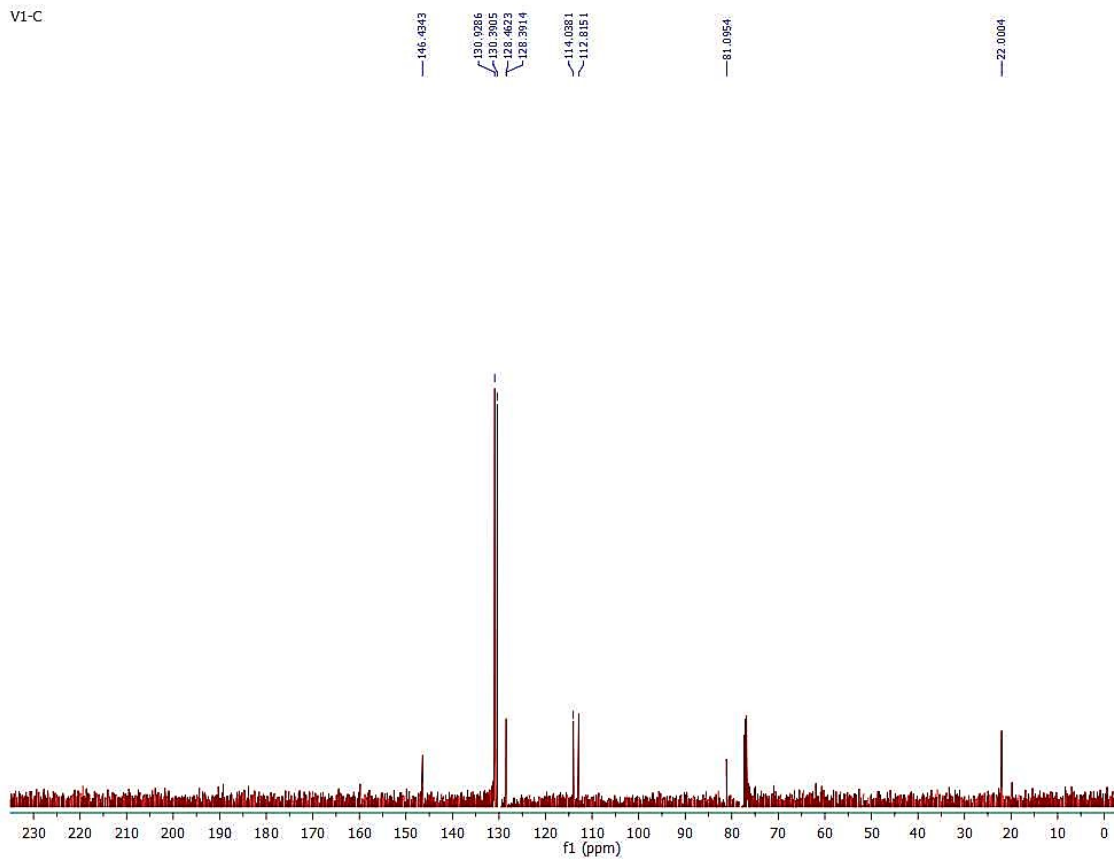


Fig. S10. ^{13}C NMR of the 2-(4-methylbenzylidene)malononitrile (**2c**).

2-(2,4-dichlorobenzylidene)malononitrile (2g)

FTIR (KBr, cm^{-1}): 3097, 3043, 2227, 1577, 1546, 1461; ^1H NMR (500MHz, CDCl_3): δ (ppm) 8.19 (s, 1H), 8.15 (d, $J = 8.5$ Hz, 1H), 7.5 (s, 1H), 7.4 (d, $J = 8.5$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3) δ (ppm) 154.6, 141, 137.1, 130.7, 130.2, 128.3, 127.5, 113, 111.8, 85.9.

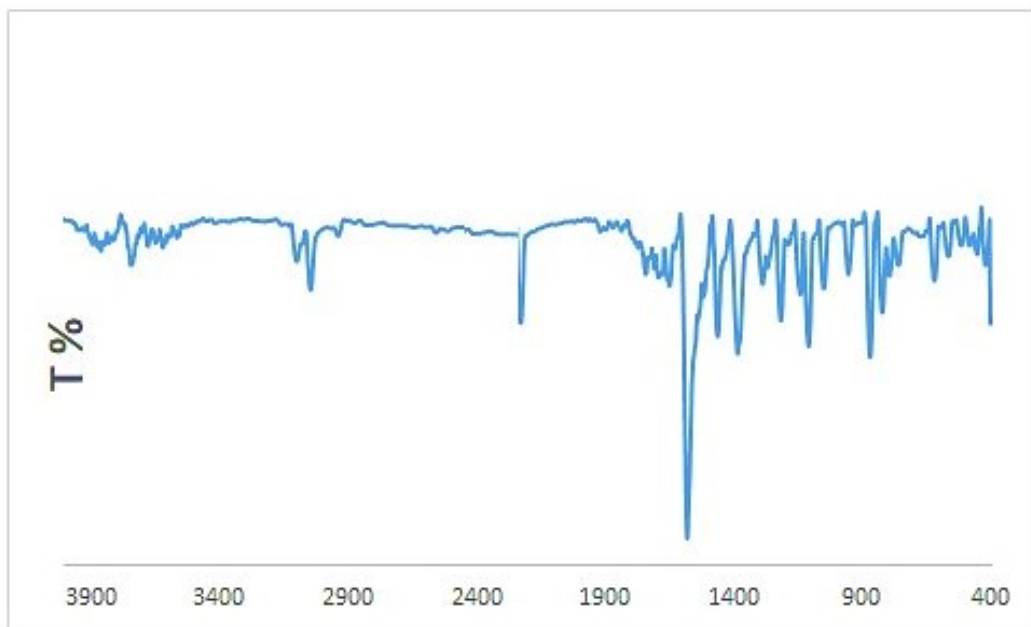


Fig. S11. FTIR of the 2-(2,4-dichlorobenzylidene)malononitrile (**2g**).

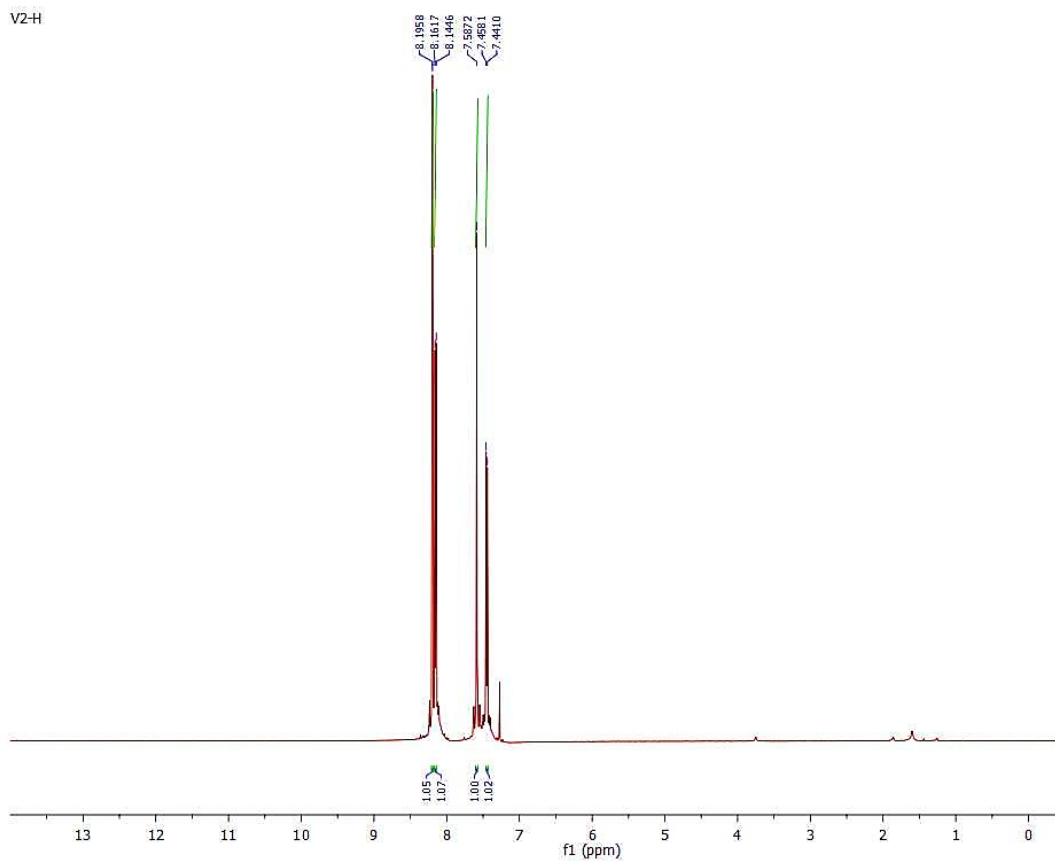


Fig. S12. ^1H NMR of the 2-(2,4-dichlorobenzylidene)malononitrile (**2g**).

V2-C

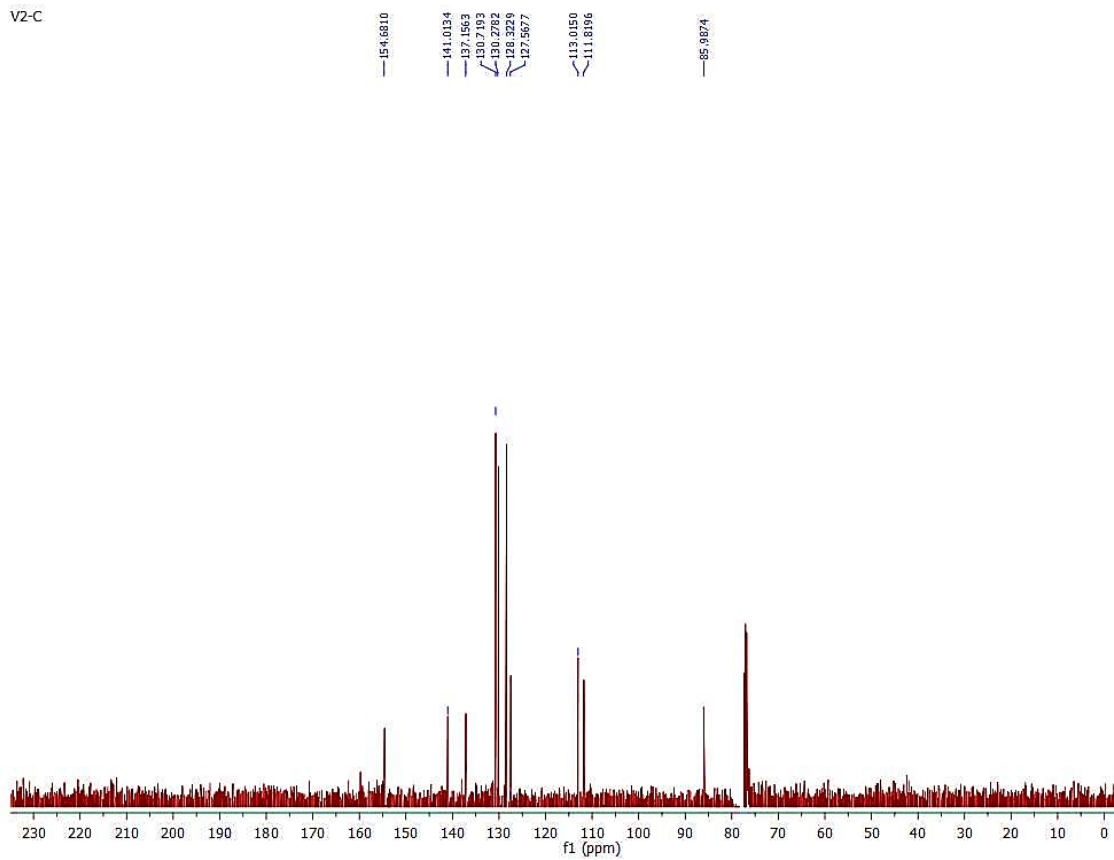


Fig.S13. ^{13}C NMR of the 2-(2,4-dichlorobenzylidene)malononitrile (**2g**).