

**Rice husk biochar modified-CuCo₂O₄ as an efficient
peroxymonosulfate activator for non-radical degradation of organic
pollutants from aqueous environment**

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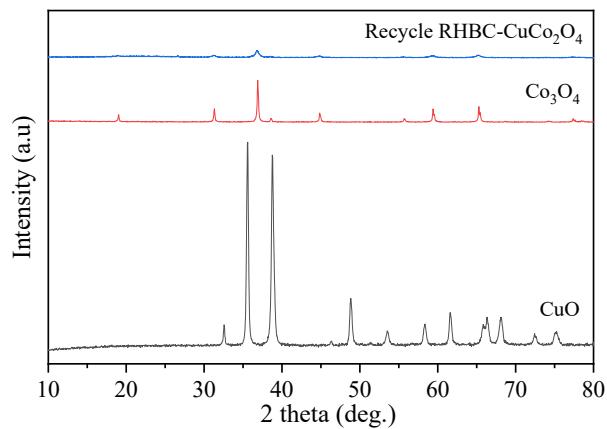


Fig. S1. XRD patterns of CuO, Co₂O₃, and recycle RHBC-CuCo₂O₄.

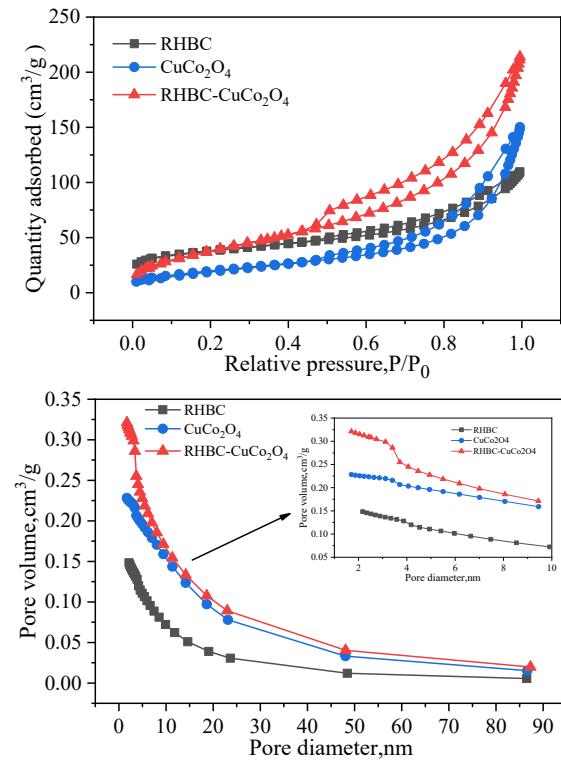


Fig. S2. N_2 sorption isotherms and pore size distributions of different materials (RHBC, CuCo_2O_4 , RHBC- CuCo_2O_4).

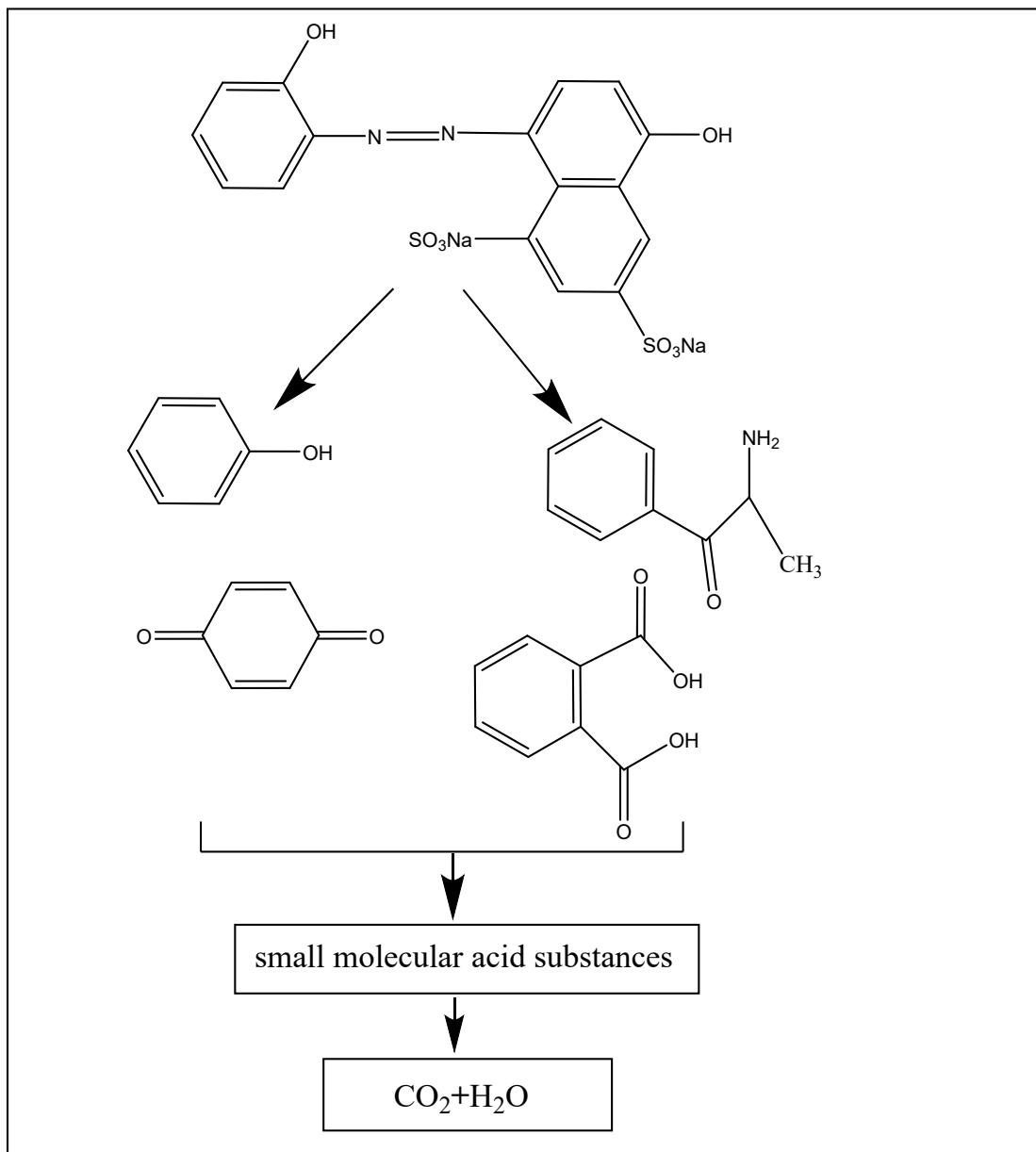


Fig S3. A possible pathway of OG oxidation degradation.

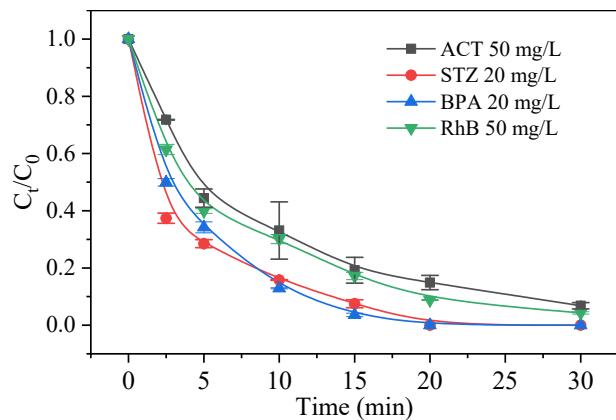


Fig S4. Biodegradation efficiency of RHBC-CuCo₂O₄ for different pollutant. Condition: [ACT] = 50 mg/L, [STZ] = 20 mg/L, [BPA] = 20 mg/L, [RhB] = 50 mg/L, [catalyst] = 100 mg/L, [PMS] = 307 mg/L, and T = 25 °C.

Table S1. Surface porosity of various materials.

Catalysts	SSA(m^2/g)	Pore size (nm)	Pore volume (cm^3/g)
RHBC	128.6	5.12	0.165
CuCo_2O_4	72.6	12.01	0.218
RHBC- CuCo_2O_4	142.9	8.81	0.315

Table S2. Degradation of pollutants by different catalysts.

Preparation method	Catalysts	Pollutant	Oxidants	Initial pH	Ros	Ref.
Sol-gel method	CuCo ₂ O ₄ @kaolin (0.1 g/L)	SIZ (10 mg/L)	PMS (1 mM)	7.0	SO ₄ ²⁻ , •OH, O ₂ ^{•-} , ¹ O ₂	[1]
pyrolysis method	RHBC-CuCo ₂ O ₄ (0.307 g/L)	STZ (20 mg/L)	PMS (1 mM)	3.4	SO ₄ ²⁻ , •OH, O ₂ ^{•-} , ¹ O ₂	This study
hydro-thermal method	GO-CuCo ₂ O ₄ (0.05 g/L)	BPA (22.83 mg/L)	PMS (0.2 mM)	7.0	SO ₄ ²⁻ , •OH	[2]
pyrolysis method	RHBC-CuCo ₂ O ₄ (0.1 g/L)	BPA (20 mg/L)	PMS (1 mM)	3.4	SO ₄ ²⁻ , •OH, O ₂ ^{•-} , ¹ O ₂	This study
solvothermal method	AC-CuCo ₂ O ₄ (0.2 g/L)	3BF (25.63 mg/L)	PMS (0.4 mM)	10.0	SO ₄ ²⁻ , •OH, O ₂ ^{•-} , ¹ O ₂	[3]
pyrolysis method	RHBC-CuCo ₂ O ₄ (0.1 g/L)	OG (50 mg/L)	PMS (1 mM)	3.4	SO ₄ ²⁻ , •OH, O ₂ ^{•-} , ¹ O ₂	This study

References

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