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Fine-tuning of pharmacological potentials of novel thiazolium ionic liquids by anion alteration

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1. Materials and instrumentation

Chemicals were obtained from the following suppliers and used without further purification: toluene, cumene, tert-butylbenzene, 4-methylthiazole, sodium tetrafluoroborate, lithium bis(trifluoromethanesulfonimide), and anhydrous MgCl₂ (Sigma–Aldrich); Zinc iodide (Acros).

Melting points were measured using a BÜCHI Melting point B-540 apparatus; all melting points were measured in open glass capillaries and are uncorrected. Elemental analyses for C, H, N, were performed with a Perkin-Elmer 263 elemental analyzer. FT-IR spectra were recorded on a BRUKER Tensor-37 FT-IR spectrophotometer in the range 400-4000 cm⁻¹ as KBr disc in the 4000-550 cm^{-1} region with 2 cm^{-1} resolution or with an ATR (attenuated total reflection) unit (Platinum ATR-QL, diamond). For signal intensities the following abbreviations were used: br (broad), sh (sharp), w (weak), m (medium), s (strong), vs (very strong). UV/Vis spectra were measured at 25 °C in ethanol (10⁻⁵ mol/L) on a Shimadzu UV-2450 spectrophotometer using quartz cuvettes (1 cm). NMR-spectra were obtained with a Bruker Avance DRX200 (200 MHz for ¹H) or Bruker Avance DRX500 (125, 97 and 470 MHz for ¹³C, ¹¹B and ¹⁹F respectively) spectrometer with calibration to the residual proton solvent signal in DMSO-d₆ (¹H NMR: 2.52 ppm, ¹³C NMR: 39.5 ppm), CDCl₃ (¹H NMR: 7.26 ppm, ¹³C NMR: 77.16 ppm) against TMS ($\delta = 0.00$ ppm) for ¹H and ¹³C, 85% phosphoric acid ($\delta = 0.00$ ppm) for ³¹P and CFCl₃ ($\delta = 0.00$ ppm) for ¹⁹F NMR. Multiplicities of the signals were specified s (singlet), d (doublet), t (triplet), q (quartet) or m (multiplet). The ESI-MS of the synthesized compounds were acquired in the linear mode for positive ions on a UHR-QTOF maXis 4G (Bruker Daltonics) and BRUKER Ultraflex MALDI-TOF instrument equipped with a 337 nm nitrogen laser pulsing at a repetition rate of 10 Hz. The 2+ charge assignment of ions in HR-ESI-MS was confirmed by the m/z = 0.5 difference between the isotope peaks (x, x+1, x+2). Peaks with chlorine showed the isotope ratio $^{35/37}$ Cl = 75.8:24.2.

2. Synthesis of 1-alkyl-4-(chloromethyl)benzene (1a-c)

General procedure: A flask was charged with 5 mol% of ZnI_2 (1.3 mmol), chlorosulfonicacid (31 mmol) and CH_2Cl_2 (30 mL), followed by dropwise addition of dimethoxymethane (31 mmol) at -10 °C. After stirring the reaction mixture at -10°C for 30min, the aromatic compound (26 mmol) was slowly added. The resulting mixture was then stirred at 5-10 °C for the time 0.5 - 2 h. The reaction was monitored by TLC analysis. After

completion, the reaction was quenchedby addition of water (10 mL) in an ice bath. After extractionwith CH_2Cl_2 (3 x 20 mL), the organic phase was washedwith 5% sodium carbonate solution (2 x 10 mL), water (2 x10 mL) and brine (2 x 20 mL), then evaporated to drynessunder reduced pressure. The residue was purified by flashcolumn chromatography on a silica gel using petroleumether (boiling range: 60-90°C) and ethyl acetate as eluents to give the desired product.

1-(Chloromethyl)-4-ethylbenzene (**1a**) [24]: Colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ (ppm): δ H 7.41 (d, *J* = 8.1 Hz, 2H), 7.30 (d, *J* = 8.1 Hz, 2H), 4.67 (s, 2H), 2.68 (s, 3H); 13C NMR (125 MHz, CDCl₃) δ (ppm): 141.8, 131.5, 128.8, 127.9, 46.2, 28.4.

1-(Chloromethyl)-4-isopropylbenzene (**1b**) [24]: Colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ (ppm): 7.41 (d, *J* = 8.1 Hz, 2H), 7.32 (d, *J* = 8.1Hz, 2H), 4.66 (s, 2H), 3.05–2.93 (m, 1H), 1.35 (d, *J* = 7.4 Hz, 6H); 13C NMR (125 MHz, CDCl₃) δ (ppm): 146.3, 138.7, 128.6, 126.7, 46.0, 33.9, 23.9.

1-(tert-Butyl)-4-(chloromethyl)benzene (**1c**) [24]: Colorless liquid. ¹H NMR (500 MHz, CDCl₃) δ (ppm): δ H 7.46 (d, *J* = 8.4 Hz, 2H), 7.42–7.39 (m, 2H), 4.64 (s, 2H), 1.40 (s, 9H); 13C NMR (125 MHz, CDCl₃) δ (ppm):151.4, 134.5, 128.3, 125.6, 46.0, 34.5, 31.3.

3. Figures



Fig. S2: FTIR spectrum of (3b)



Fig. S4: FTIR spectrum of (4a)



Fig. S6: FTIR spectrum of (4c)



Fig. S8: ¹³C NMR of **2a** (125 MHz, DMSO-*d*₆)



Fig. S10: ¹³C NMR of 2b (125 MHz, CDCl₃)







100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175 -180 -185 -190 -195 f1 (ppm)





-100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 -160 -165 -170 -175 -180 -185 -190 -195 -20 f1 (ppm)





Fig. S18: ¹³C NMR of **3c** (125 MHz, DMSO-*d*₆)







Fig. S22: ¹⁹F NMR of 4a (565 MHz, DMSO-*d*₆)



Fig. S24: ¹³C NMR of **4b** (125 MHz, DMSO-*d*₆)



--81.68







Fig. S29: The effect of different doses (0 – 200 μM) of TILs (3a-c) and (4a-c) on the viability of (A,B) ovarian carcinoma cell lines (SKOV-3) and (C,D) human skin fibroblast (HSF) cells after 24 h of treatment as compared to the clinical antitumor drug (cisplatin, CDDP).