

## Supplementary Information

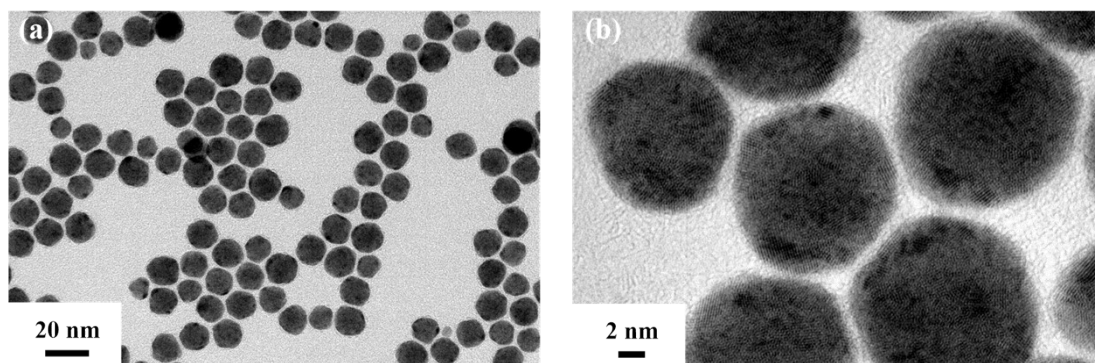
### **Core-Shell Pd-P@Pt-Ni Nanoparticles with Enhanced Activity and Durability as Anode Electrocatalyst for Methanol Oxidation Reaction**

*Jiangbin Guo,<sup>a</sup> Man Zhang,<sup>b</sup> Jing Xu,<sup>\*a</sup> Jun Fang,<sup>a</sup> Shuiyuan Luo,<sup>a</sup> and Chaolong Yang<sup>\*c</sup>*

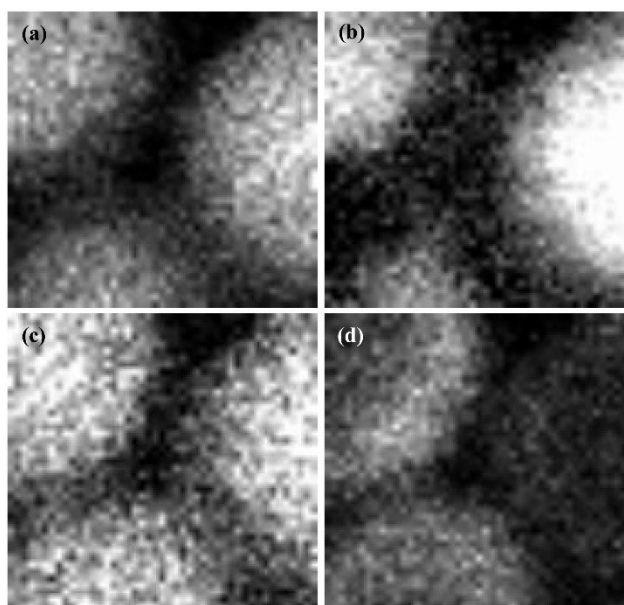
*<sup>a</sup>College of Chemical Engineering and Materials Science, Quanzhou Normal University, Quanzhou 362000, Fujian Province, PR China. E-mail: jingxu@qztc.edu.cn*

*<sup>b</sup>State Key Laboratory of Molecular Engineering of Polymers, Department of Macromolecular Science, Fudan University, Shanghai 200433, PR China*

*<sup>c</sup>School of Materials Science and Engineering, Chongqing University of Technology, Chongqing 400054, PR China. E-mail: yclzjun@163.com*



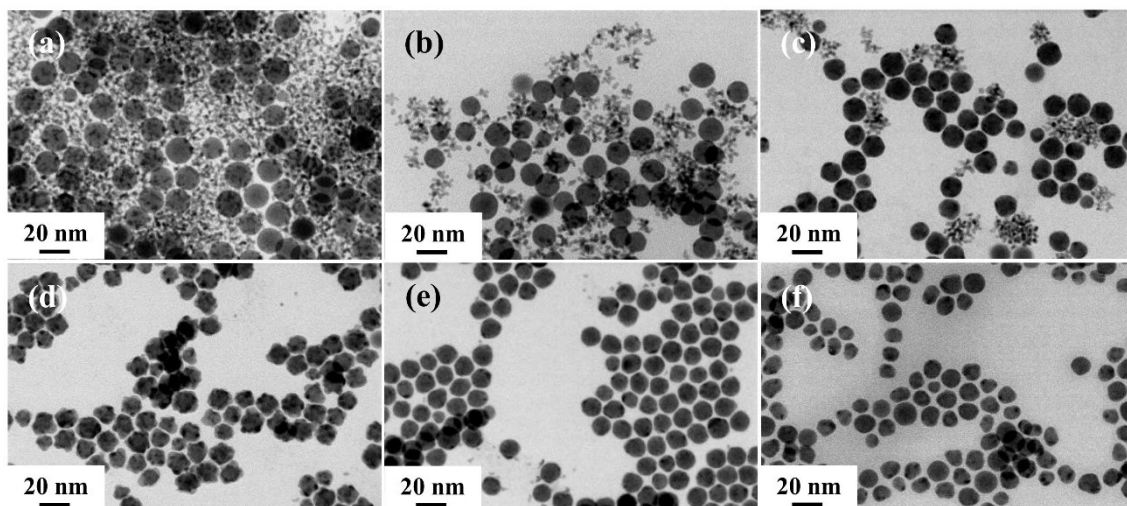
**Figure S1.** (a) Representative TEM and (b) HRTEM images of Pd-P@Pt-Ni NPs.



**Figure S2.** STEM-EDS images of (a) Pd, (b) P, (c) Ni and (d) Pt in Pd-P@Pt-Ni NPs.

**Table S1.** The amounts of Pt(acac)<sub>2</sub> and OAm corresponding to the TEM images in Figure S3.

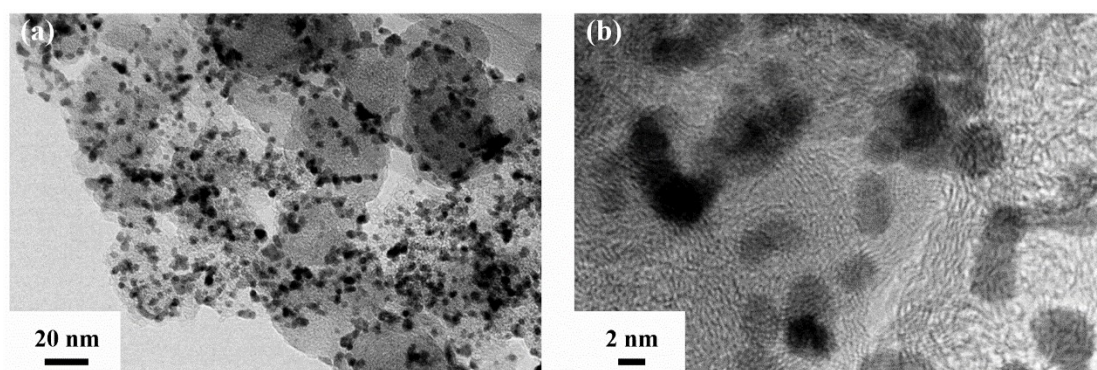
Sample	Pt(acac) <sub>2</sub> /mmol	OAm/mL	TEM image
1	0.20	10	Figure S3(a)
2	0.15	10	Figure S3(b)
3	0.10	10	Figure S3(c)
4	0.20	17	Figure S3(d)
5	0.15	17	Figure S3(e)
6	0.13	17	Figure S3(f)



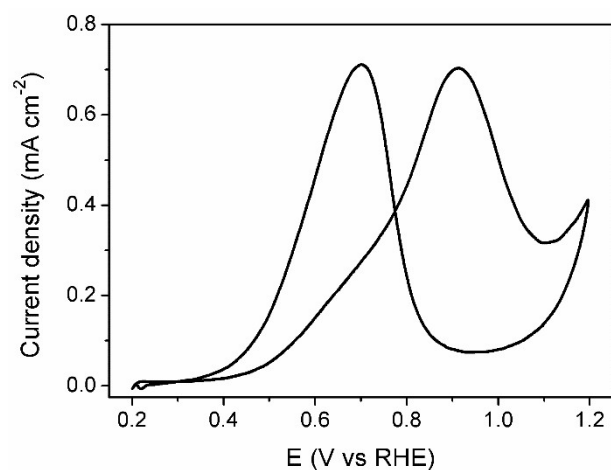
**Figure S3.** TEM images of products fabricated with different amounts of  $\text{Pt}(\text{acac})_2$  and OAm. The detailed information is summarized in Table S1.

**Table S2.** Peak position ( $2\theta$ ),  $d$ -spacing and lattice parameter collected from XRD data.

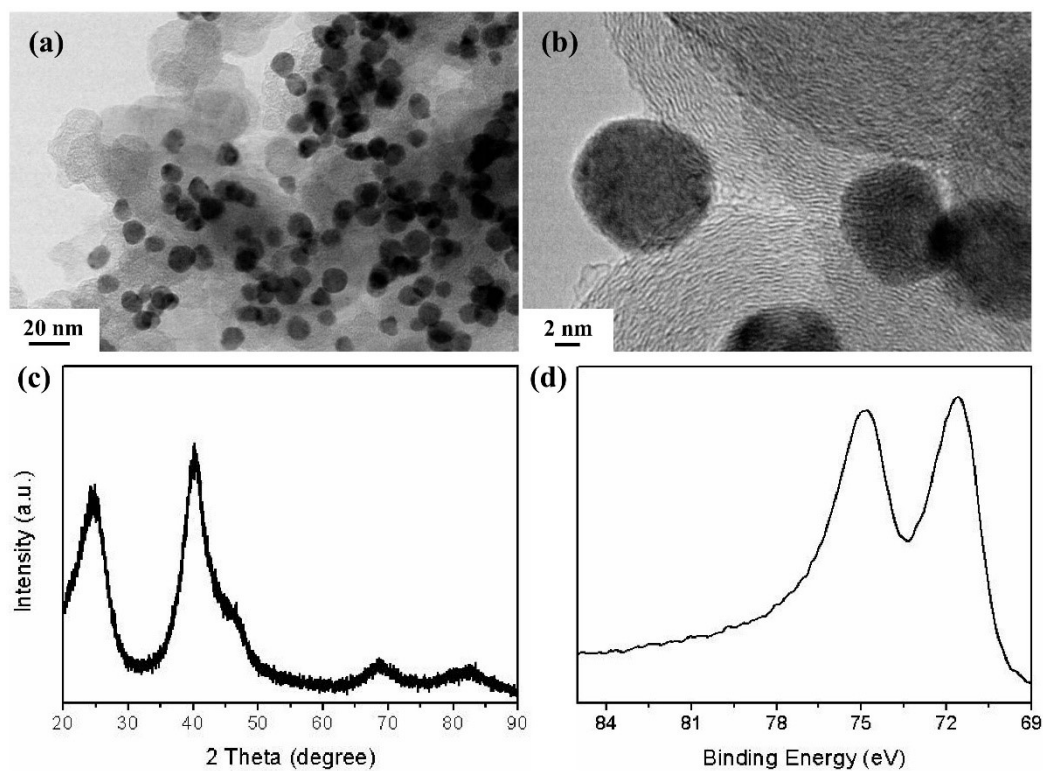
materials	plane (111)			plane (220)	
	$2\theta$	$d$ -spacing	lattice parameter	$2\theta$	lattice parameter
	(degree)	(nm)	(nm)	(degree)	(nm)
Pd-P@Pt-Ni/C	40.13	0.2244	0.3887	68.67	0.3861
commercial Pt/C	39.72	0.2267	0.3927	67.55	0.3917
Ni powder	44.61	0.2029	0.3514	76.48	0.3519



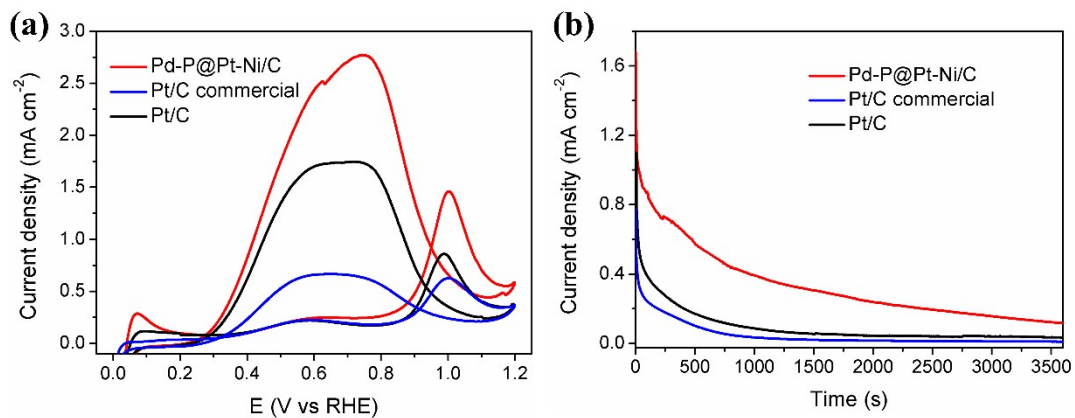
**Figure S4.** (a) TEM and (b) HRTEM images of the as-prepared Pt/C NPs.



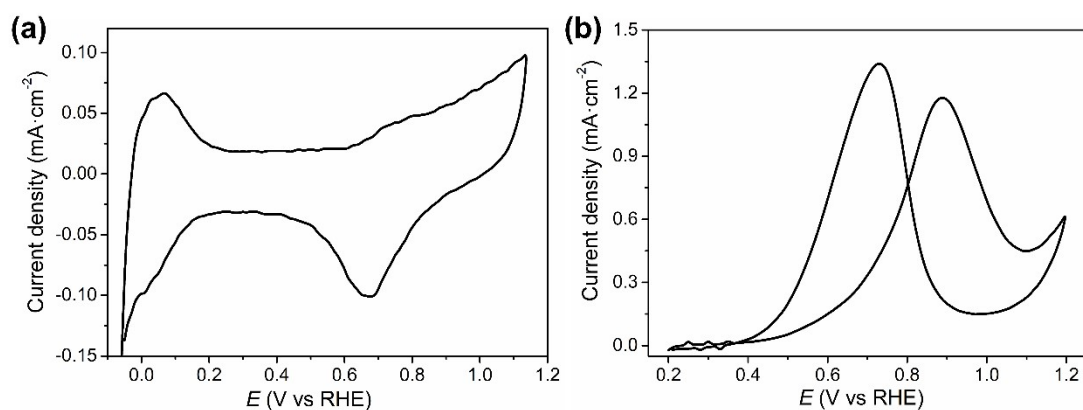
**Figure S5.** CV curve of Pd-P@Pt-Ni/C NPs in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 10 M CH<sub>3</sub>OH solution. The sweep rate is 50 mV·s<sup>-1</sup>.



**Figure S6.** (a)-(b) TEM and HRTEM images, (c) XRD pattern and (d) XPS spectrum of Pt 4f in Pd-P@Pt-Ni/C after the stability test.



**Figure S7.** (a) CV curves of Pd-P@Pt-Ni/C NPs, commercial Pt/C and the as-prepared Pt/C in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 0.5 M HCOOH solution. The sweep rate is 50 mV·s<sup>-1</sup>. (b) Chronoamperometry curves of Pd-P@Pt-Ni/C NPs, commercial Pt/C and the as-prepared Pt/C in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 0.5 M HCOOH solution at 0.95 V.



**Figure S8.** Cyclic voltammetry profiles of Pd-P@Pt/C NPs (a) in N<sub>2</sub>-saturated 0.1 M HClO<sub>4</sub> solution and (b) in 0.5 M H<sub>2</sub>SO<sub>4</sub> + 1 M CH<sub>3</sub>OH solution. The sweep rate is 50 mV·s<sup>-1</sup>.