

Supplementary materials:

Text S1: Compound 1 [pentacos-7-ene]: (500 mg) white waxy substance; R_f : 0.69 and 0.85 by using solvent systems (100% Light petroleum) and (Light petroleum: CH_2Cl_2 90:10), $^1\text{H-NMR}$ (CDCl_3 , 400 MHz) spectral analyses were carried out by: JEOL at 500 MHz and Bruker at 400 MHz and chemical shifts were given in ppm with the TMS as internal standard: δ_{H} 5.32 (2H, br s, H-7, H-8), 0.91(6H, t, H-1,25, $J= 7.2$ Hz), 1.00-1.30 (42H, m, H2-6 and H9-24); EI-MS that carried out on Jeol JMS-AX 500, 70 ev. displayed m/z (% relative abundance) as follow: 351 ($\text{M}^+\text{+H}$, 1), 350 (M^+ , 0.05), 322.28 ($\text{M}^+\text{+H-C}_2\text{H}_5$,1.5), 308 (322- CH_2 , 0.5), 295 (1.5), 294 (322- C_2H_4 , 0.3), 280 (322- C_3H_6 , 0.1), 266 (322- C_4H_8 , 0.07), 252 (322- C_5H_{10} , 0.1), 211 (1.6), 183 (2.2), 155 (2.3), 127 (3.4), 125 (23), 111 (42), 97 (65), 85 (87),71 (100).

Text S2: Compound 2 [Nonatriacontanoic acid]: (100 mg) white crystalline scales; R_f : 0.55 & 0.73 using solvent systems (Light petroleum: CH_2Cl_2 75:25) & (100% CH_2Cl_2); $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 2.33 (2H, t, H-2, $J=8.0\text{Hz}$), 1.63 (2H, m, H-3), 1.25 (66H, m, H-4-36), 1.25 (2H, m, H-37), 1.56 (2H, m, H-38), 0.86 (3H,t, H-39), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 176.94 (COOH), 33.64 (C-2), 32.08 (C-3), 29.85 (C-4 : C-36), 24.87 (C-37), 22.84 (C-38), 14.26 (C-39); EI-MS m/z (% relative abundance): 578 (M^+ ,1.62), 577 (3.83), 563 ($\text{M}^+\text{-CH}_3$, 1.84), 549 ($\text{M}^+\text{-C}_2\text{H}_5$, 1.02), 534 ($\text{M}^+\text{-COO}$, 1.16), 520 (534- CH_2 , 1.02), 506 (534- C_2H_4 , 0.06), 396 (5.59), 313 (32.81), 236 (47.94), 98 (91.02), 83 (82.21), 57 (C_4H_9 ,100.00).

Text S3: Compound 3 [Melissic acid (Triacontanoic acid)]: (400 mg) white amorphous substance; R_f : 0.53 [CH_2Cl_2 : MeOH (9.8: 0.2)], R_f : 0.50, 0.64 by using solvent systems (Light petroleum: CH_2Cl_2 75:25) & (100% CH_2Cl_2); $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 2.35 (2H, t, H-2, $J = 8.0$ Hz), 1.67 (2H, m, H-3), 1.65 (2H, m, H-4), 1.63 (2H, m, H-5, H-6), 1.60 (2H, m, H-7), 1.55

(2H, m, H-8), 1.25 (38H, m, H-9-27), 1.25 (2H, m, H-28), 1.25 (2H, m, H-29), 0.89 (3H, t, H-30, $J = 8.0$ Hz), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 176.82 (C=O), 33.65 (C-2), 24.89 (C-3), 29.22 (C-4), 29.39 (C-5, C-6), 29.51 (C-7), 29.59 (C-8), 29.85 (C-9 – C-27), 32.08 (C-28), 22.84 (C-29), 14.27 (C-30), 51.05 (C-O); EI-MS m/z (% relative abundance): 453.21 (M^+H , 0.5), 452 (M^+ , 4.17), 424 ($\text{M}^+\text{H-C}_2\text{H}_5$, 13.13), 408 ($\text{M}^+\text{H-COOH}$, 0.5), 396 ($424\text{-C}_2\text{H}_4$, 71.90), 354 ($396\text{-C}_3\text{H}_6$, 9.63), 353 (38.15), 297 ($354\text{-C}_4\text{H}_8\text{H}$, 32.87), 278 (10.42), 187 (7.86), 129 (61.82), 97 (58.93), 71 (63.26), 57 (100), 45 (77.19).

Text S4: Compound 4 [Cerotic acid (Hexacosanoic acid)]: (100 mg) white granules; R_f : 0.42, 0.54 by using solvent system (Light petroleum: CH_2Cl_2 75:25) & (100% CH_2Cl_2); $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 2.36 (2H, t, H-2, $J = 8.0$ Hz), 1.63 (2H, m, H-3), 1.25 (40H, m, H-4-24), 1.25 (2H, m, H-25), 0.86 (3H, t, H-26, $J = 8.0$ Hz), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 177.3 (C=O), 36.28 (C-2), 24.87 (C-3), 29.59 (C-4 – C-24), 22.85 (C-25), 14.27 (C-26), 51.05 (C-O) ; EI-MS m/z (% relative abundance): 396 (M^+ , 100), 381 ($\text{M}^+\text{-CH}_3$, 15.65), 354 (3.64), 288 (12.22), 255 (16.98), 213 (8.54), 161 (8.66), 147 (32.06), 95 (25.53), 85 (31.77), 69 (27.64), 57 (34.32).

Text S5: Compound 5 [Palmitic acid]: (200 mg) white crystalline scales; m.p. 63-64 °C; R_f 0.22 & 0.40 using solvent systems (Light petroleum: CH_2Cl_2 75:25 & (100% CH_2Cl_2); $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 2.33 (2H, t, H-2, $J = 8.0$ Hz), 1.63 (2H, m, H-3), 1.25 (20H, m, H-4-13), 1.25 (2H, m, H-14), 1.56 (2H, m, H-15), 0.88 (3H, t, H-16, $J = 8.0$ Hz), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 176.94 (C=O), 33.64 (C-2), 32.08 (C-3), 29.85 (C-4 - C-13), 24.87 (C-14), 22.84 (C-15), 14.26 (C-16); EI-MS m/z (relative abundance %): 256 (M^+ , 2.08), 227 ($\text{M}^+\text{-C}_2\text{H}_5$, 1.14), 213 ($\text{M}^+\text{-C}_3\text{H}_6\text{H}$, 2.62), 185 (2.84), 149 (7.25), 129 (7.80), 121 (9.61), 97 (11.27), 83 (16.02), 69 (72.17) and 40 (100).

Text S6: Compound 6 [Lupeol palmitate]: (200 mg) white needles; m.p. 110-112 °C, R_f 0.78 & 0.85 by using solvent system (CH_2Cl_2 :MeOH 99:1) & (CH_2Cl_2 :MeOH 98:2). $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 4.26 (1H, dd, H-3, $J = 6.0, 12.0$ Hz), 5.8 (1H, m, H-9'), 5.01 (1H, m, H-10'), 4.69, 4.57 (2H, s, H-29a, H-29b), 2.39, 2.41 (2H, t, H-2', $J = 7.2$ Hz), 2.31 (1H, t, H-19, $J = 1.8$ Hz), 1.68 (3H, s, H-30), 1.57 (2H, m, H-3'), 0.94 (3H, s, H-23), 0.89 (3H, s, H-24), 0.83 (3H, s, H-25), 0.79 (3H, s, H-26), 0.76 (3H, s, H-27), 0.76 (3H, t, H-16'), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 38.89 (C-1), 23.90 (C-2), 79.17 (C-3), 37.33 (C-4), 55.45 (C-5), 18.15 (C-6), 34.44 (C-7), 40.99 (C-8), 50.60 (C-9), 38.21 (C-10), 21.08 (C-11), 25.30 (C-12), 38.21 (C-13), 42.99 (C-14), 27.57 (C-15), 35.74 (C-16), 43.16 (C-17), 48.46 (C-18), 48.14 (C-19), 151.14 (C-20), 29.85 (C-21), 40.16 (C-22), 28.14 (C-23), 16.27 (C-24), 16.13 (C-25), 15.52 (C-26), 15.50 (C-27), 18.47 (C-28), 109.47 (C-29), 19.31 (C-30), 167.91 (C=O), 32.08 (C-1'), 25.30 (C-2'), 29.08 (C-3'), 29.51 (C-4'-C-8'), 131.03 (C-9'), 132.61 (C-10'), 30.51 (C-11'), 22.84 (C-12'), 33.97 (C-13'), 23.14 (C-14'), 22.84 (C-15'), 21.08 (C-16'); EI-MS m/z (relative abundance %): 662 [M^+ , 0.83], 647 ($\text{M}^+ - \text{CH}_3$, 0.5), 426 [$\text{M}^+ - \text{palmitic acid}$, 13.27], 411 (9.53), 316 (7.06), 267 (7.17), 239 (10.95), 219 (19.97), 218 (77.01), 189 (73.36), 147 (43.69), 135 (70.01), 109 (85.65), 95 (100), 81 (84.14), 69 (81.63), 54 (61.05).

Text S7: Compound 7 [Lupeol acetate]: (500 mg) white needles; m.p: 216-218 °C and R_f values 0.43 and 0.55 by using solvent systems (CH_2Cl_2 :MeOH 99:1) & (CH_2Cl_2 :MeOH 98:2). $^1\text{H-NMR}$ (CDCl_3 , 400 MHz): δ_{H} 4.46 (1H, m, H-3), 4.68, 4.56 (2H, s, H-29b, H-29a, respectively), 2.05 (3H, s, CH_3OCO), 2.04 (1H, m, H-19), 1.68 (3H, s, H-30), 1.03 (3H, s, H-25), 0.94 (3H, s, H-28), 0.85 (3H, s, H-23), 0.84 (3H, s, H-24), 0.83 (3H, s, H-26), 0.79 (3H, s, H-27), $^{13}\text{C-NMR}$ (CDCl_3 , 100MHz); δ_{C} 38.62 (C-1), 21.56 (C-2), 81.13 (C-3), 38.20 (C-4), 55.54 (C-5), 18.36 (C-6), 34.37 (C-7), 41 (C-8), 50.50 (C-9), 37.3 (C-10), 21.10 (C-11), 23.85 (C-12), 36.95 (C-13), 42.98 (C-14),

25.26 (C-15), 35.73 (C-16), 43.16 (C-17), 48.44 (C-18), 48.19 (C-19), 151.13 (C-20), 29.99 (C-21), 40.15 (C-22), 27.59 (C-23), 16.89 (C-24), 16.65 (C-25), 16.34 (C-26), 14.66 (C-27), 18.16 (C-28), 109.51 (C-29), 19.45 (C-30), 171.18 (C=O), 28.19 (CH₃ of acetyl group); EI-MS *m/z* (relative abundance %): 468[M⁺, 7.33], 426 [M⁺-acetate, 2.1], 393(5.27), 257(3.96), 229(5.27), 218(100), 203(57.32), 189(49.82), 147 (15.91), 121(22.55), 107 (24.29), 81 (32.94), 69 (43.28) and 43 (59.46).

Text S8: Compound 8 [β -sitosterol]: (500 mg) colorless crystals; m.p. 136-140°C; R_f 0.65 and 0.83 using solvent systems (CH₂Cl₂:MeOH 95:5) & (CH₂Cl₂:MeOH 85:15); ¹H-NMR (CDCl₃, 400 MHz): δ_{H} 3.52 (1H, m, H-3), 5.34 (1H, t, H-6) *J*=5.2 Hz, 1.03 (3H, s, H-18), 0.91 (3H, d, *J*= 8.0 Hz, H-21), 0.89 (3H, t, *J*= 7.2 Hz, H-24), 0.84 (3H, t, *J*= 7.2 Hz, H-29), 0.82 (3H, d, *J*= 7.2 Hz, H-27), 0.80 (3H, d, *J*= 7.2 Hz, H-26), 0.68 (3H, s, H-19), 0.68 (2H, m, H-28), ¹³C-NMR (CDCl₃, 100MHz); δ_{C} 37.41 (C-1), 31.82 (C-2), 71.97 (C-3), 42.46 (C-4), 140.91 (C-5), 121.88 (C-6), 32.06 (C-7, C-8), 50.29 (C-9), 36.66 (C-10), 21.24 (C-11), 39.93 (C-12), 42.47 (C-13), 56.92 (C-14), 26.22 (C-15), 28.40 (C-16), 56.21 (C-17), 36.30 (C-18), 19.18 (C-19), 34.10 (C-20), 24.46 (C-21), 45.99 (C-22), 23.22 (C-23), 12.13 (C-24), 29.30 (C-25), 19.97 (C-26), 19.55 (C-27), 18.93 (C-28), 12.01 (C-29); EI-MS *m/z* (relative abundance %): 414 (M⁺, 59.88%), 396 (M⁺-H₂O, 32.87%), 381 (M⁺-H₂O-CH₃, 28.23%), 329 (91.62%), 303 (90.25%), 273 (59.75%), 255 (59.65%), 213 (49.10%), 145(62.34%), 107(87.64%), 105(72.50%), 79(66.87%), 55(74.61%) and 43(100%).

Text S9: Compound 9 [β -sitosterol-3-O- β -D-glucopyranoside]: (2 g) white powder; R_f values 0.33 & 0.48 by using solvent systems (CH₂Cl₂: MeOH 95:5) & (CH₂Cl₂: MeOH 85:15); ¹H-NMR (CDCl₃, 400 MHz): δ_{H} 5.34 (1H, m, H-6) *J*=5.2 Hz, 4.44 (2H, dd, *J*= 2.5, 11.77, H-6'), 4.22 (1H, m, H-1'), 3.64 (1H, m, H-3), 2.36 (2H, m, H-4) *J*=5.2 Hz, 2.13 (2H, m, H-15), 1.95 (1H, m, H-14), 1.81 (2H, m, H-7), 1.64 (4H, m, H-2, H-12), 1.64 (1H, m, H-25), 1.48 (2H, m, H-11), 1.43

(1H, m, H-20), 1.30 (4H, m, H-1, H-28), 1.30 (1H, m, H-8), 1.24 (4H, m, H-16, H-23), 1.16 (1H, m, 17), 0.66 (3H, s, H-18), 0.92 (3H, s, H-19), 0.96 (3H, m, H-21), 0.90 (1H, m, H-24), 0.97 (3H, t, $J = 7.2$ Hz, H-29), 0.90 (3H, d, $J = 8$ Hz, H-27), 0.83 (3H, d, $J = 8$ Hz, H-26), 0.84 (1H, m, H-9), 2.91-3.36 (sugar protons; ESI-MS in positive mode MS m/z (relative abundance %): 577 ($M^+ + 1$, 100.00) and 415 ($M^+ + H - 162$, 16.23).

Text S10: Identification of phenolic acids and phenolic acid glycosides:

Compounds 1 and **3** showed molecular ion peaks at m/z 137, and 153 $[M-H]^+$ and base peak fragments at m/z 93 $[M-H-COO]$ and 109 were suggested to be hydroxybenzoic acid and protocatecheic acid, respectively.¹ In addition, **compound 6** was identified as coumaric acid from MS¹ at m/z 163 $[M-H]^+$ and MS² at m/z 119 due to loss of carboxylic group.² Finally, **compound 22** exhibited $[M-H]^-$ at m/z 301 and MS² base peak fragment ion at m/z 139 (100%) in addition to fragments at m/z 256 (35%) and 228 (30%), comparing these results with those previously published data, compound **22** was identified as ellagic acid.² All the above compounds were identified for the first time in *M. macroura*.

The molecular ion $[M-H]^+$ of **compound 23** was observed at m/z 355, and the base peak at m/z 193.3 was detected due to the loss of glucoside unit (162 amu) in the MS² spectrum. Hence, this compound was identified as ferulic acid-*O*-glucoside.² Additionally, **compound 29** showed a molecular ion peak at m/z 329 $[M-H]^+$, MS² at m/z 287 attributed to $[M-H-COCH_3]$, which suggested to be phloracetophenone-4-*O*-glucoside,³ and it was isolated as a pure compound in the current work. However, To our knowledge, ferulic acid-*O*-glucoside and phloracetophenone-4-*O*-glucoside, were identified for the first time in the family Moraceae.

Text S11: Identification of other phenolic compounds

Compound 4 showed a molecular ion peak at m/z 339 $[M-H]^+$ and fragment ion at m/z 177 (loss of 163 amu), besides other diagnostic peaks for morachalcone A at m/z 175, 161, 135, 109.⁴ Moreover, compound **47** showed a molecular ion peak at m/z 391 $[M+H]^+$ and significant MS² fragment ion at m/z 279 that attributed to dibutyl phthalate, 167 corresponding to phthalic acid and base peak at m/z 149, this compound was suggested to be bis (2-ethyl hexyl) phthalate,⁵ which was isolated as a pure compound in this work. All compounds are the first report in *M. macroura*, while bis (2-ethyl hexyl) phthalate is identified for the first time in the family Moraceae.

Text S12: Identification of flavonoid aglycones

The deprotonated molecular ion at m/z 301 was detected for compounds **25** and **26**, where, **peak 25** which showed MS² fragmentation at m/z 179, 151, and 121, was suggested to be quercetin. However, MS² fragment ions at m/z 256, 151, 107 were matched with morin (**26**).² Thus, these compounds were identified for the first time in *M. macroura*.

Text S13: Identification of flavonoid O- and C-glycosides:

Five peaks **5**, **10**, **15**, and **24** at R_t 6.27, 7.62, 8.53, 8.96, and 12.17 min, respectively, showed the same molecular ion peak at m/z 431 $[M-H]^-$. Where peaks **5**, **15** were identified as apigenin-*O*-glucoside and its isomer depending upon MS² fragmentation that gave a base peak at m/z 269 by the loss of glucosyl residue (162 amu).⁶ While MS² of compound **10** gave fragment ions at m/z 269.0 $[M-H-162]$, 341 $[M^+-H-90]$, and 311 $[M^+-H-120]$ that are characteristic features for mono-*C*-glucoside flavonoid. Moreover, the absence of fragment at $[M-H-18]$ indicated the presence of 8-*C*-glucoside; therefore, peak **10** was confirmed to be apigenin-8-*C*-glucoside.⁷ Additionally, compound **24** was suggested to be kaempferol-*O*-rhamnoside (afzelin) through MS² fragment ions at m/z 285, 284 [Loss of sugar moiety], 255, 227.⁸ Moreover, compounds **8**, **16**, **17**, and **20**, displayed at retention time 7.26, 8.71, 9.24, 9.24, and 11.14 min, respectively, have the same $[M-$

H]⁻ at m/z 447 with different MS² fragmentation patterns. Where, MS² of compound 8 exhibited fragment ions at m/z 285 due to loss of glucosyl moiety (162 amu), 257, and 242.5,⁶ hence it was identified as luteolin-7-*O*-glucoside. Furthermore, compound **16** showed MS² fragment ion at m/z 285 (100%) and 284; therefore, this compound was confirmed to be kaempferol-3-*O*-glucoside.² Also, compound 17 showed MS² fragment ion at m/z 301 [M-H-rham] and 343 [M-H-104] for [aglycone + 41], which is a characteristic feature of mono-*C*-rhamnoside flavonoid. Moreover, the absence of fragment at [M-H-18] indicated the presence of 8-*C*-rhamnoside,⁷ so it was identified as quercetin-8-*C*-rhamnoside. However, compound **20** was identified as quercetin-3-*O*-rhamnoside (quercitrin), depending on the MS² fragmentation pattern with characteristic ions at m/z 301 and 255.⁹ **Compound 18** (R_t , 10.95 min) was suggested to be hesperidin which was confirmed by the [M+H]⁺ ion at m/z 611 and MS² fragments at m/z 303 [M-H-rutinosyl moiety, 308 amu].¹⁰ In addition, three **compounds**, **11**, **13**, and **14**, have the same molecular ion peak at m/z 463 [M-H]⁺, **compound 11** was identified as morin-3-*O*-glucoside from MS² fragmentation that showed fragment ions at m/z 301, 257. **Compound 13** was identified as quercetin-3-*O*-glucoside from MS² that showed a base peak at m/z 301, indicating the loss of glucose moiety leaving quercetin aglycone.² Additionally, compound **14** was suggested to be quercetin-3-*O*-galactoside (Hyperoside) according to MS² fragmentation that displayed ion at m/z 301.¹¹ In addition, compound 38 was identified as chrysoeriol-uronic acid according to the molecular ion peak at m/z 475 [M-H]⁻ and MS² base peak fragment ion at m/z 299 [Chrysoeriol aglycone].⁶ **Compound 39** was identified as naringenin-7-*O*-glucoside,¹² with a molecular ion peak at m/z 433 [M-H]⁺ and MS² at m/z 271, indicating naringenin moiety. Finally, compound **53** showed a molecular ion peak at m/z 535 [M+H]⁻ along with MS² data which showed the loss of 218 amu for acetyl glucuronide from the fragment signal at m/z 317 [Isorhamnetin aglycone+H]⁺ and 153,¹³ so this compound

suggested to be isorhamnetin 3-*O*-acetyl glucuronide. All compounds are the first report in *M. macroura* and the family Moraceae.

Text S14: Identification of prenylated flavonoids:

Compound 34, displayed at R_t 17.83 min, was suggested to be albanin A, which was confirmed by $[M+H]^+$ ion at m/z 355 and MS^2 fragment ions at m/z 299, 69.¹⁴ Finally, **compound 55** at R_t 24.95 min was suggested to be euchrenone a7 from MS^1 at m/z 339 $[M-H]^+$ and MS^2 fragment ion at m/z 163 (100%).¹⁵ These prenylated flavonoids were identified for the first time in *M. macroura*.

Text S15: Identification of flavonoid derivatives

Compound 7 was identified as naringenin derivatives from MS^1 at m/z 463 $[M-H]^+$ and MS^2 at m/z 270.7, indicating naringenin moiety).⁷ **Also, compound 38** that detected at at m/z 475 $[M-H]^+$ together with a base peak at m/z 299 was identified as Chrysoeriol-uronic acid.⁶ Moreover, **compound 51** (R_t – 24.14 min) exhibited molecular ion peak at m/z 607 $[M+H]^+$ and gave base peak fragment ion in MS^2 at m/z 286, indicating the presence of kaempferol aglycone moiety. Therefore, it was tentatively assigned as australisine A.¹⁶ Thus, all compounds were identified for the first time in *M. macroura*.

Text S16: Identification of tannin compounds

Compounds 27 and **28** were suggested to be gallicocatechin glycoside and epigallocatechin glycoside,² respectively, based on MS^1 at m/z 467 $[M-H]^+$ and MS^2 fragments at m/z 305 due to loss of glycosyl moiety. Thus, these compounds represent the first report in the family Moraceae.

Text S17: Identification of coumarin compounds

Coumarin compounds were detected in the plant under investigation at different retention times as in compound **2** that identified as aesculin depending upon MS^1 at m/z 339 $[M-H]^+$ and MS^2 at m/z 177, which attributed to cleavage of glycosidic linkage and loss of glycosyl moiety.² In

addition, **compound 9** that displayed MS¹ at m/z 355 [M+H]⁺ and MS² at m/z 193 due to loss of 162 amu was identified as scopolin.¹⁷ Also, **compound 33**, which detected m/z 355 [M+H]⁺ with MS² fragment ion at m/z 203 (loss of side chain, 152), was identified as epoxybergamottin.¹⁸ All compounds are the first report in *M. macroura*.

Text S18: Identification of anthocyanin derivatives compounds:

Compound 12 showed a molecular ion peak at m/z 449 [M+H]⁺ with MS² fragment ion at m/z 287 due to loss of glycoside; it has been suggested to be cyanidin-3-*O* glucoside.¹ Additionally, **compound 19** showed a molecular ion peak at m/z 610 [M+H]⁺ and characteristic MS² fragment at m/z 317.5 due to loss of two molecules of rhamnose (292 amu), 256.8, 151.2, which suggested to be petunidin dirhamnoside.¹⁹ Furthermore, **compound 36** which detected at m/z 591 [M-H]⁺ with MS² at m/z 287 due to cyanidin moiety, was identified as cyanidin cinnamoyl glucuronide.²⁰ Also, **compound 42** that identified as malvidin diglucuronide showed MS¹ at m/z 683 [M+H]⁺ and MS² at m/z 331 attributed to loss of two moieties of glucuronic acid (352 amu).²¹ **Compound 44** was identified as cyanidin derivative from MS¹ at m/z 683 [M+H]⁺ and MS² at m/z 597 (loss of 86), 435 (loss of glucose), 287.²² Moreover, **compound 48** exhibited protonated molecular ion at m/z 593 [M+H]⁺ with MS² at m/z 287 due to loss of rutinose, was identified as cyanidin-3-*O*-rutinoside from **Jin et al. (2017)**.¹ Additionally, **compound 54** was confirmed as petunidin 3-*O*-acetyl glucuronide based on MS¹ at m/z 535 [M+H]⁺ and MS² at m/z 317 due to loss of 218 amu attributed to loss of acetyl group and glucuronic moiety (42, 176 amu), 153.²³ Moreover, malvidin-3-*O*-acetyl glucoside was designated for **compound 57** based on molecular ion at m/z 535 [M+H]⁺ and MS² at m/z 331 due to loss of 204 amu.²⁴ In addition, **compound 58** that showed a molecular ion at m/z 683 [M+H]⁺ with MS² at m/z 597 (loss of 86 malonyl moiety), 435 (loss of glucose), 287.7 was proposed to be cyanidin-3-*O*-malonyl glucoside derivative.²⁵ Also, cyanidin-3-*O*-

malonyl glucoside was suggested for **compound 59** depending on MS¹ at m/z 535 [M+H] and MS² at m/z 287. **Compound 61** concluded to be cyanidin-3-*O*-hexosyl hexoside according to MS¹ at m/z 611 [M+H]⁺ and MS² at m/z 449 (loss of glucose).¹ Finally, **compound 62** showed a molecular ion peak at m/z 475 [M-H]⁺ along with MS² ions at m/z 429 (loss of 46), 299.8 (loss of 176), hence identified as peonidin-3-*O*-glucuronide.²⁶ All anthocyanin compounds identified for the first time in *M. macroura*.

Text S19: Identification of fatty acids:

Cerotic acid (hexacosanoic acid) was designated for compound **21** according to protonated molecular ion at m/z 397 [M+H]⁺ and MS² at m/z 351 [M⁺-COOH].²⁷ Cerotic acid was isolated as a pure compound from the plant species under investigation (**Compound 4**). However, compound **32** was identified as nonatriacontanoic acid which displayed MS¹ at m/z 579 [M+H]⁺ and it was isolated as a pure compound in the current work (**Compound 2**).²⁸ Moreover, compound **35** was suggested to be hydroxy octadecatrienoic acid based on MS¹ at m/z 293 [M-H]⁺ and MS² at m/z 220.8 [M⁺-CH₂-CH₂-COOH].²⁹ Furthermore, compound **43** was identified as melissic acid (triacontanoic acid) depending upon MS¹ at m/z 451 [M-H]⁺ and MS² fragmentation at m/z 249.8, 390.5 [M⁺-COOH-CH₃], 406.7 [M⁺-COOH], 435.1 [M⁺-H-CH₃],³⁰ and it was isolated as a pure compound in our work (**Compound 3**). **Compound 45** showed a molecular ion peak at m/z 255 [M-H]⁺ which suggested to be palmitic acid,²⁸ and it was isolated as a pure compound in the current work (**Compound 5**). Oleic acid was designated for compound **46** which showed a molecular ion peak at m/z 281 [M-H]⁺.²⁸ Finally, compound **52** was proposed to be octadecanoic acid according to MS¹ at m/z 283 [M-H]⁺ and MS² at m/z 239 [M-H-COO]. All compounds identified for the first time in *M. macroura*. Nonatriacontanoic, hydroxyl octadecatrienoic and melissic acids are first report in Moraceae family.

Text S20: Identification of fatty acids derivatives

Compound 41 suggested to be palmitic acid ester based on MS¹ at m/z 475 [M-H]⁺ besides MS² fragment ion at m/z 255.²⁸ Also, compound **50** was identified as oleic acid ester which displayed molecular ion at m/z 395 [M-H]⁺ and MS² at m/z 281.²⁸ All these fatty acid esters are identified for the first time in *M. macroura*.

Identification of steroid and triterpenoid compounds

Compound 40 that exhibited a molecular ion peak at m/z 381 [M+H]⁻ with distinctive fragment ions at m/z 318, 163, 139, 71, 69, was identified as brasicasteol.³¹ Also, **compound 56** which showed a molecular ion at m/z 414 [M+H]⁻ along with characteristic fragment ion at m/z 254.9 was identified as β -sitosterol. Moreover, 24-Methylene ergosta-5-en-3 β -Ol was designated as compound **63** which showed MS¹ at m/z 398 and MS² at m/z 381 [M-H₂O+H]⁺, 281 [M-C₇H₁₃-H₂O-2H]⁺.³¹ Also, compound **64** showed a molecular ion peak at m/z 455 [M-H]⁻ which suggested to be oleanolic acid.²⁸ All compounds identified for the first time in *M. macroura*. Cholesterol, brasicasteol and 24-Methylene ergosta-5-en-3 β -Ol were reported for the first time in family Moraceae.

Text S21: Identification of steroid and triterpenoid derivatives

β -sitosterol-3-*O*-D-glucoside,³² was assigned for compound **30**, based on deprotonated molecular ion at m/z 575 [M-H]⁻ and it was isolated as a pure compound in this work (**Compound 10**). Finally compounds **31** and **60** were identified as lupeol acetate and lupeol palmitate,³³ based on MS¹ at m/z 467 [M-H]⁻ and 663 [M+H]⁻ and both compounds were isolated in pure form during the current study and designated as (**Compound 7**) and (**Compound 6**), respectively. All compounds identified for first time in *M. macroura*.

Text S22: Identification of hydrocarbons and sugars

Compound 49 was identified as pentacos-7-ene, ³⁴from MS¹ at *m/z* 349 [M-H]⁻ and it was isolated as a pure compound in the current work and is identified for the first time in family Moraceae.

In conclusion, the major compounds in DCM-L fraction are cyanidin-3-*O*-malonyl glucoside and its derivatives, cyanidin rutinoside, chlorogenic acid, quercetin-*O*-glucoside, kaempferol-*O*-glucoside, 24-Methylene-ergosta-5-en-3 β -Ol, isorhamnetin-3-*O*-acetyl glucuronide, bis (2-ethyl hexyl) phthalate and morin-*O*- β -glucoside. On the other side, cyanidin-3-*O*-malonyl glucoside, melissic acid, phloracetophenone-4-*O*-glucoside, 24-methylene-ergosta-5-en-3 β -Ol, cerotic acid, and bis (2-ethyl hexyl) phthalate are the most abundant constituents in DCM-S fraction. The major flavonoids that present in the DCM -L and absent in the DCM-S are quercetin-*O*-glucoside, kaempferol-*O*-glucoside, morin-*O*- β -glucoside, naringenin derivatives, apigenin-8-C-glucoside, luteolin-7-*O*-glucoside, quercetin-8-C-rhamnoside, and quercetin-3-*O*- β -rhamnoside. These results agree with **Jin et al. (2015)**, ³⁵and could give insights into the possible potentials of these fractions as an anti-depressant and cardioprotective against ISO induced post-MI depression rat model.

References

1. Q. Jin, J.L. Yang, L. Mab, D. Wen, F. Chen, J. Lia, Identification of polyphenols in mulberry (genus *Morus*) cultivars by liquid chromatography with time-of-flight mass spectrometer, *J Food Compost Anal.*, 2017, **63**, 55–64.
2. M.M. Natić, D. Dabic, C.A. Papetti, M.M.F. Aksic, V. Ognjanov, M. Ljubojevic, Z.L. Tesic, Analysis and characterization of phytochemicals in mulberry (*Morus alba* L.) fruits grown in Vojvodina, North Serbia, *Food Chem.*, 2015, **171**, 128-136.
3. Y. Wang, M. Yang, X. Wang, T. Li, L. Kong, Bioactive metabolites from the endophytic fungus *Alternaria alternata*, *Fitoterapia*, 2014, **99**, 153–158.
4. H. Wei, J.J. Zhu, X.Q. Liu, W.H. Feng, Z.M. Wang, L.H. Yan, Review of bioactive compounds from root barks of *Morus* plants (Sang-Bai-Pi) and their pharmacological effects, *Cogent Chem.*, 2016, **2**, 1212320.

5. D. Hamdan, R.A. El-Shiekh, M. El-Sayed, H.M. Khalil, M. Mousa, A. Al-Gendy, A. El-Shazly, Phytochemical characterization and anti-inflammatory potential of Egyptian *Murcott mandarin* cultivar waste (stem, leaves and peel), *Food Funct.*, 2020, **11**, 8214–8236.
6. A. Plazonić, F. Bucar, Z. Males, A. Mornar, B. Nigovic, N. Kujundzic, Identification and quantification of flavonoids and phenolic acids in burr parsley (*Caucalis platycarpos* L.), using high-performance liquid chromatography with diode array detection and electrospray ionization mass spectrometry, *Molecules*, 2009, **14**, 2466-2490.
7. A. Brito, J.E. Ramirez, C. Areche, B. Sepulveda, M.J. Simirgiotis, HPLC-UV-MS profiles of phenolic compounds and antioxidant activity of fruits from three citrus species consumed in Northern Chile, *Molecules*, 2014, **19**, 17400-17421.
8. G. Jang, H. Kim, M. Lee, S. Jeong, R. Bak, D. Lee, J. Kim, Characterization and quantification of flavonoid glycosides in the Prunus genus by UPLC-DADQTOF/MS, *Saudi J. Biol. Sci.*, 2016, **25**, 1-10.
9. A. Alberti-Der, LC-ESI-MS/MS methods in profiling of flavonoid glycosides and phenolic acids in traditional medicinal plants: *Sempervivum tectorum* L. and *Corylus avellana* L, Ph. D. Dissertation , Semmelweis University, 2013.
10. R.M. Ibrahim, A. El-Halawany, D. Saleh, E. El Naggar, A. El-Shabrawy, S. El-Hawary, HPLC-DAD-MS/MS profiling of phenolics from *Securigera securidaca* flowers and its anti-hyperglycemic and anti-hyperlipidemic activities, *Rev. bras. farmacogn.*, 2015, **25**, 134-141.
11. J. Yang, X. Liu, Q. Zhang, Q. Jin, J. Li, Phenolic Profiles, Antioxidant Activities, and Neuroprotective Properties of Mulberry (*Morus atropurpurea* Roxb.) Fruit Extracts from Different Ripening Stages, *J. Food Sci.*, 2016, **81**, C2439- C2446.
12. G. Le Gall, M.S. DuPont, F.A. Mellon, A.L. Davis, G.J. Collins, M.E. Verhoeven, Colquhoun IJ. Characterization and content of flavonoid glycosides in genetically modified tomato (*Lycopersicon esculentum*) fruits, *J. Agric. Food Chem.*, 2003, **51**, 2438-46.
13. S.H. Im, Z. Wang, S.S. Lim, O.H. Lee, I.J. Kang, Bioactivity-guided Isolation and identification of anti-adipogenic compounds from *Sanguisorba officinalis*, *Pharm. Biol.*, 2017, **55**, 2057-2064.
14. T.K. Lim, Edible Medicinal and Non-Medicinal Plants. Volume 4, Fruits: Springer Dordrecht Heidelberg London New York, 1- 436, *Trends Food Sci Technol.*, 2012, **19**, 505-512.
15. F. Ferlinahayati, Y.M. Syah, L.D. Juliawaty, E.H. Hakim, Flavanones from the wood of *Morus nigra* with cytotoxic activity, *Indones. J. Chem.*, 2013, **131**, 205-208.
16. Q.J. Zhang, Y.B. Tang, R.Y. Chen, D.Q. Yu, Three New Cytotoxic Diels–Alder-Type Adducts from *Morus australis*, *Chem. Biodivers.*, 2007, **4**, 1533-1540.

17. K. Doi, T. Kojima, M. Makino, Y. Kimura, Y. Fujimoto, Studies on the constituents of the leaves of *Morus alba* L, *Chem. Pharm. Bull.*, 2001, **49**, 151-153.
18. J.A. Manthey, HPLC-MS analysis of coumarins and furanocoumarin dimers in immature grape fruit, *Proc. Fla. State Hort. Soc.*, 2005, **118**, 429-436.
19. M.J. Simirgiotis, J. Bórquez, G. Schmeda-Hirschmann, Antioxidant capacity, polyphenolic content and tandem HPLC–DAD–ESI/MS profiling of phenolic compounds from the South American berries *Luma apiculata* and *L. chequén*, *Food Chem.*, 2013, **139**, 289-299.
20. C.D. Kay, G. Mazza, B.J. Holub, J. Wang, Anthocyanin metabolites in human urine and serum, *Br. J. Nutr.*, 2004, **91**, 933-942.
21. A. Decendit, M. Mamani-Matsuda, V. Aumont, P. Waffo-Teguo, D. Moynet, K. Boniface, J.M. Mérillon, Malvidin-3-O- β -glucoside, major grape anthocyanin inhibits human macrophage-derived inflammatory mediators and decreases clinical scores in arthritic rats, *Biochem. Pharmacol.*, 2013, **86**, 1461-1467.
22. N. Hassimotto, M. Genovese, F. Lajolo, Identification and characterisation of anthocyanins from wild mulberry (*Morus nigra* L.) growing in Brazil, *Food Sci Technol Int.*, 2007, **13**, 17-25.
23. H. Zhang, Z. Ma, X. Luo, X. Li, Effects of mulberry fruit (*Morus alba* L.) consumption on health outcomes: A mini-review, *Antioxidants*, 2018, **7**, 69.
24. A.S. Frey, LC/MS method development for the separation of anthocyanins and anthocyanin-derived pigments in red wines (Doctoral dissertation, Lincoln University), 2015, 1-91.
25. K. Schütz, M. Persike, R. Carle, A. Schieber, Characterization and quantification of anthocyanins in selected artichoke (*Cynara scolymus* L.) cultivars by HPLC–DAD–ESI–MSn, *Anal. Bioanal. Chem.*, 2006, **384**, 1511-1517.
26. J.M. Cooney, D.J. Jensen, T.K. McGhie, LC-MS identification of anthocyanins in boysen berry extract and anthocyanin metabolites in human urine following dosing, *J. Sci. Food Agric.*, 2004, **84**, 237-245.
27. Z.T. Murathan, M. Zarifikhosroshahi, N.E. Kafkas, Determination of fatty acids and volatile compounds in fruits of rosehip (*Rosa* L.) species by HS-SPME/GC-MS and Im-SPME/GC-MS techniques, *Turk J Agric For.*, 2016, **40**, 269-279.
28. A. Ali, M. Jameel, M. Ali, Fatty acids analysis of *Ficus religiosa* stem bark by Gas Chromatography-Mass spectrometry, *IJAPMBS.*, 2017, **112**, 1-6.
29. N.Y. Yang, Y.F. Yang, K. Li, Analysis of Hydroxy Fatty Acids from the Pollen of *Brassica campestris* L. var. *oleifera* DC. by UPLC-MS/MS, *Pharmaceutics*, 2012, **2013**, 847-852.

30. M. Jameel, A. Ali, M. Ali, Identification of new compounds from *Fumariaparviflora* Lam., *J App Pharm Sci.*, 2017, **7**, 053-060.
31. S. Panawan, C. Watcharapong, M. Sugunya L. Suwaporn, T. Somsuda L. Vijittra, Structures of phytosterols and triterpenoids with potential anti-cancer activity in bran of black non-glutinous rice, *Nutrients*, 2015, **7**, 1672-1687.
32. P. Tania, H.K. Kar, Isolation and Characterization of β -sitosterol-3-O- β -D-glucoside from the Extract of the Flowers of *Viola odorata*, *J. Pharm. Res. Int.*, 2017, **16**, 1-8.
33. C.G. Magalhães, L.P. Duarte, W.D. Mussel, A.L. Ruiz, L. Shiozawa, J.E. Carvalho, S.A.Vieira Filho, Lupeol and its ester: NMR, powder XRD data and in vitro evaluation of cancer cell growth, *Braz. J. Pharm. Sci.*, 2017, **53**, 1-10.
34. J. Huang, R. Tang, H. Wu, X. Wang, GC-MS Analysis of Essential Oil from the Flowers of *Musa basjoo*, *Chem. Nat. Compd.*, 2016, **52**, 334-335.
35. Q. Jin, J. Yang, L. Ma, J. Cai, J. Li, Comparison of Polyphenol Profile and Inhibitory Activities Against Oxidation and α -Glucosidase in Mulberry (Genus *Morus*) Cultivars from China, *J. Food Sci.*, 2015, **80**, 2440-2451.

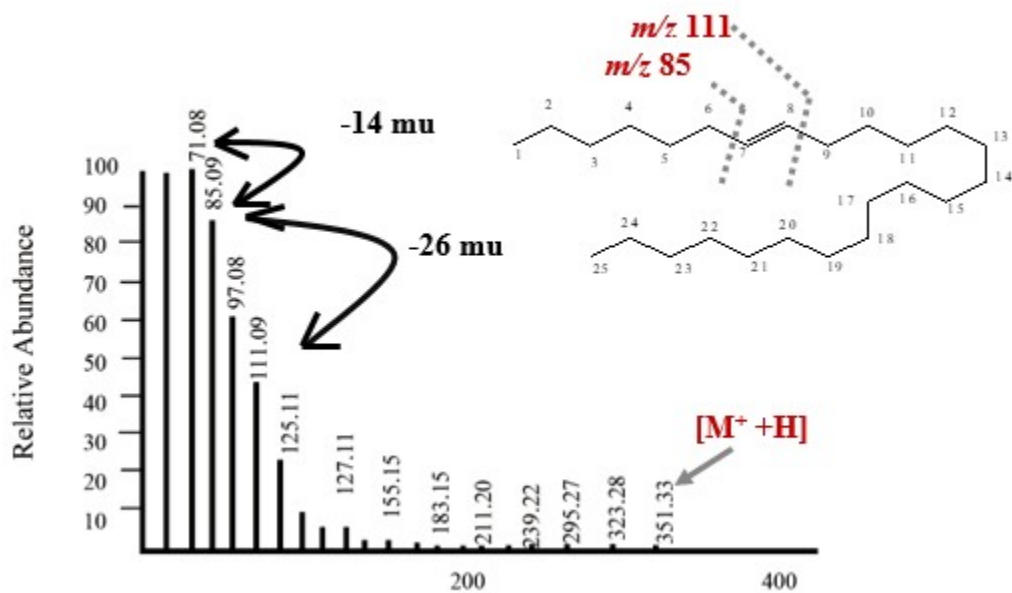


Figure (S1): EI-MS spectrum of compound "1".

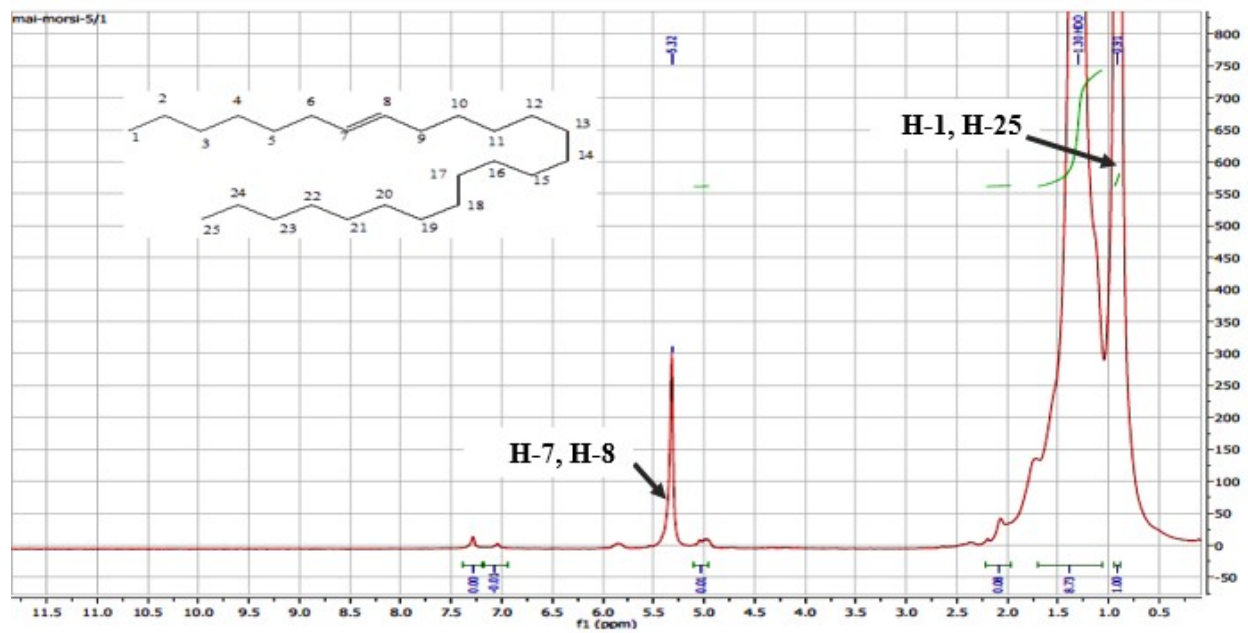


Figure (S2): $^1\text{H-NMR}$ spectrum of compound "1".

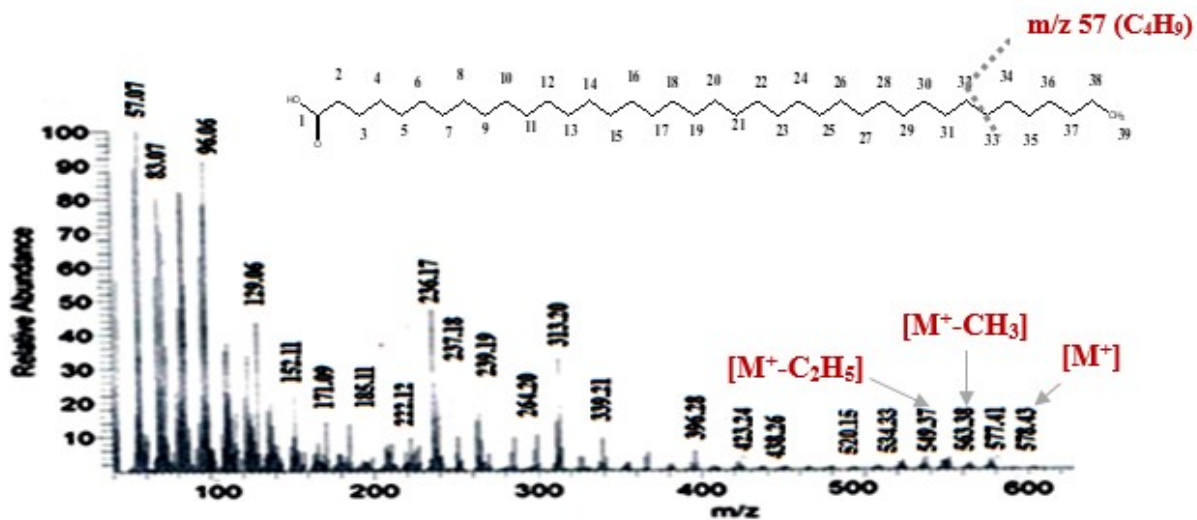


Figure (S3): EI-MS spectrum of compound "2".

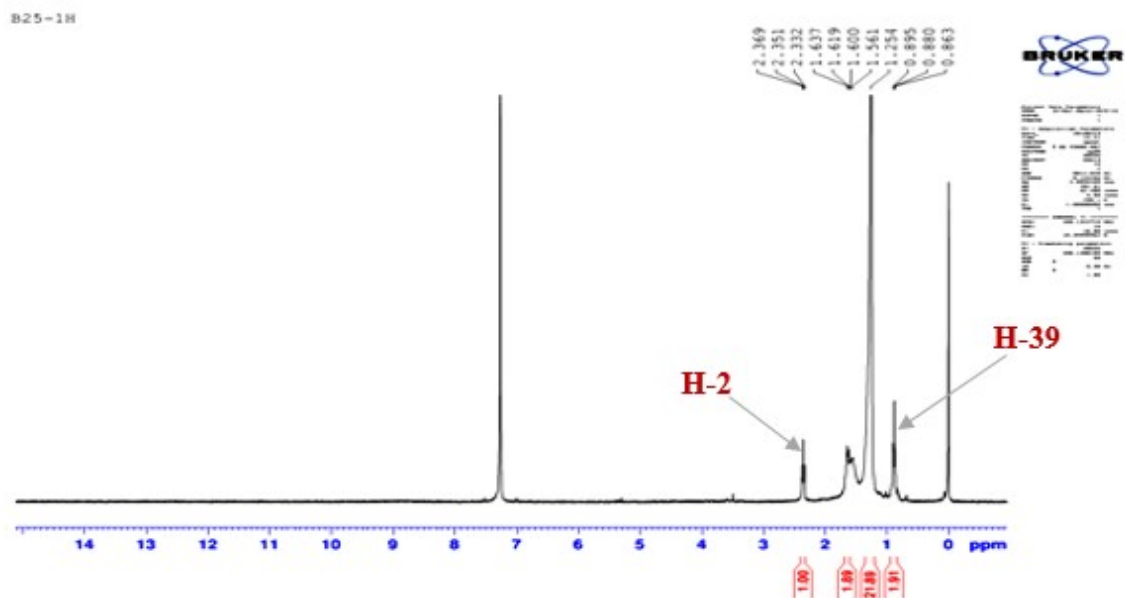


Figure (S4): 1H -NMR spectrum of compound "2".

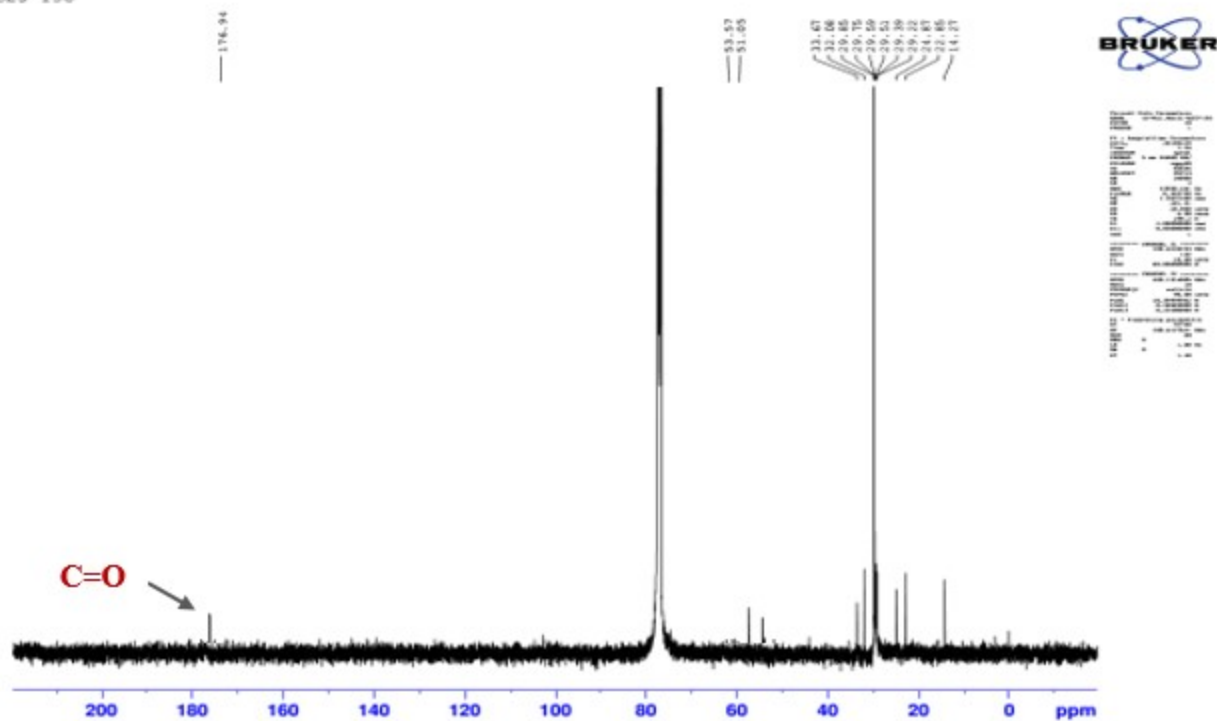


Figure (S5): ^{13}C -NMR spectrum of compound “2”.

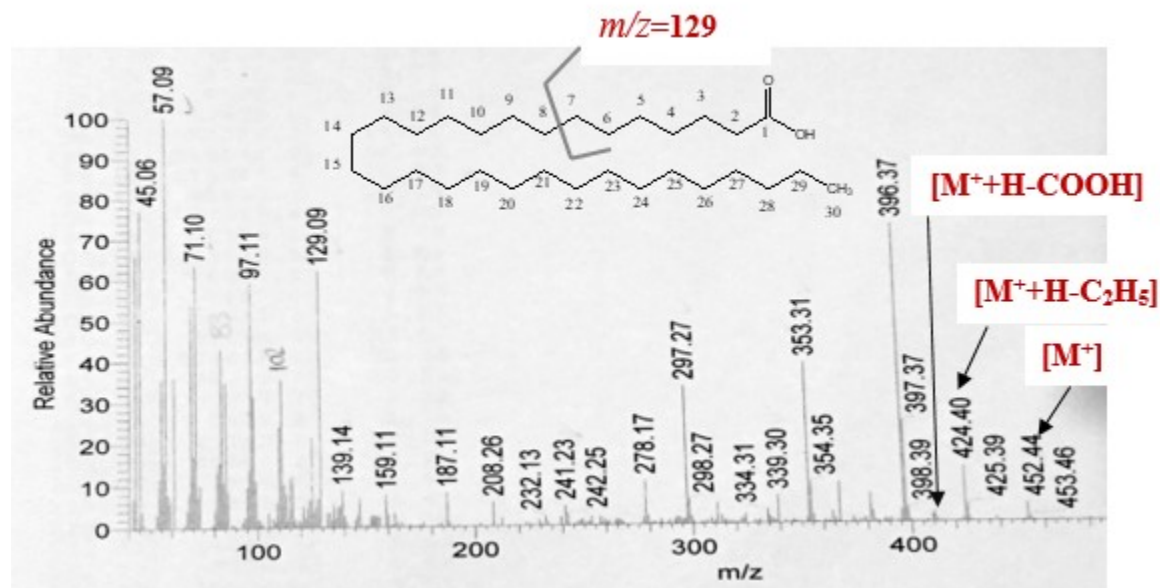
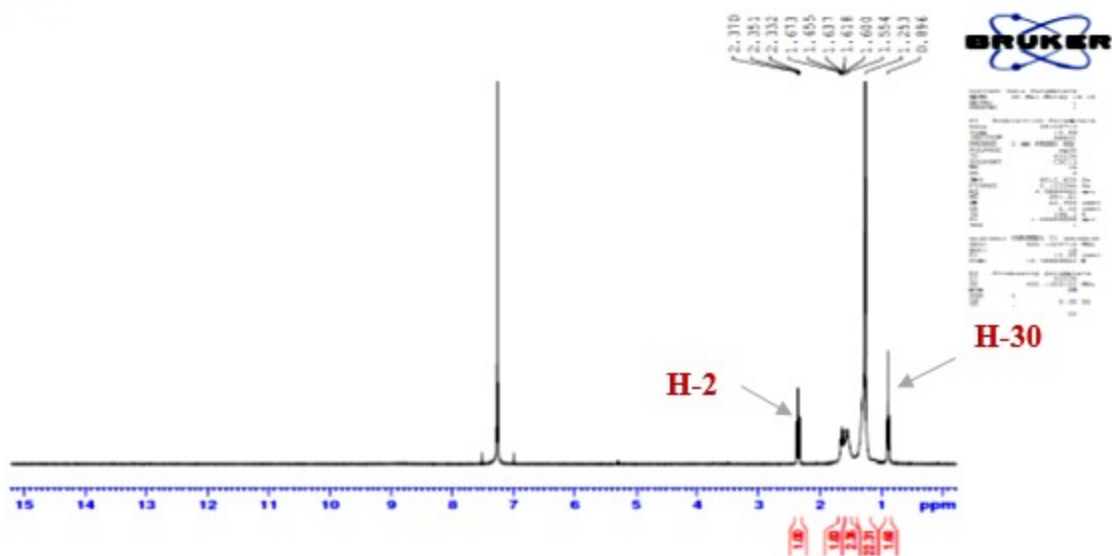


Figure (S6): EI-MS spectrum of compound “3”.

19-1H



re (S7): ¹H-NMR spectrum of compound "3".

Figur

19-13C

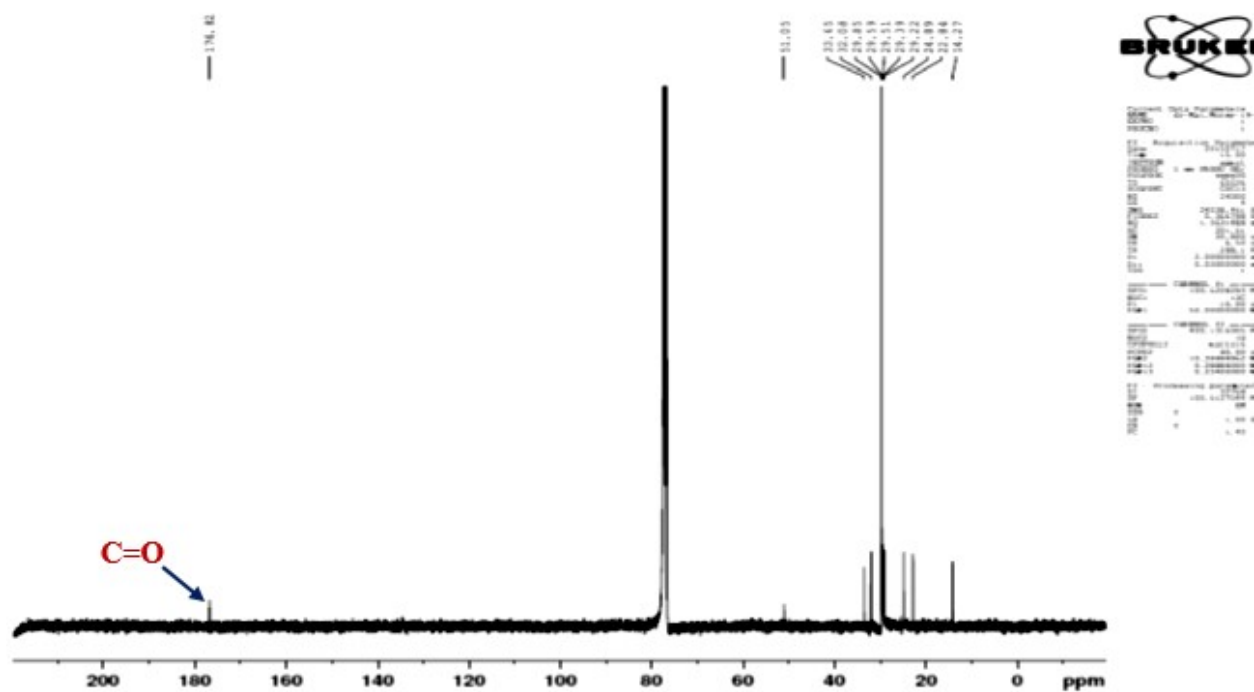


Figure (S8): ¹³C-NMR spectrum of compound "3".

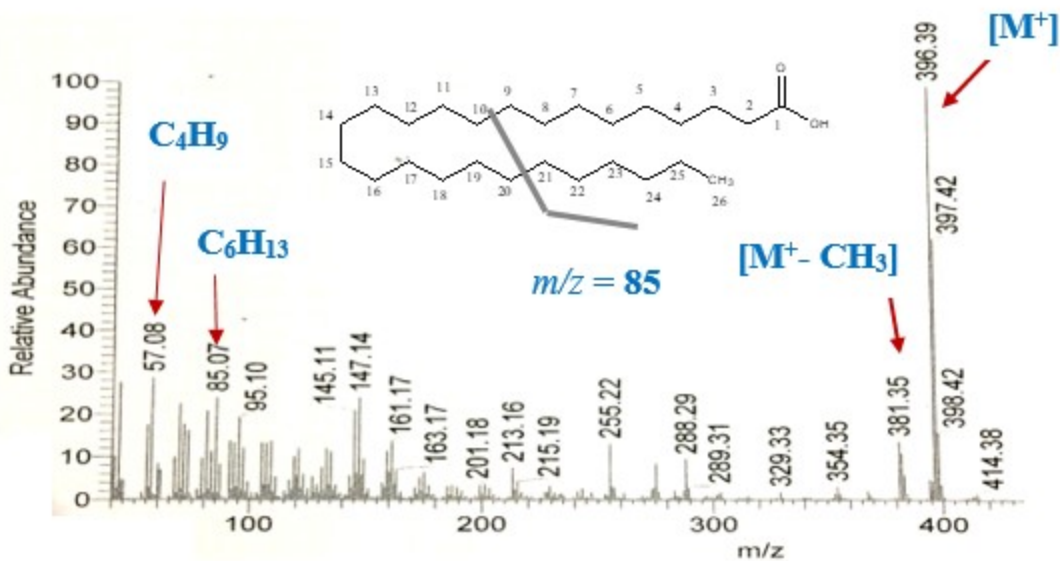


Figure (S9): EI-MS spectrum of compound "4".

24-1B

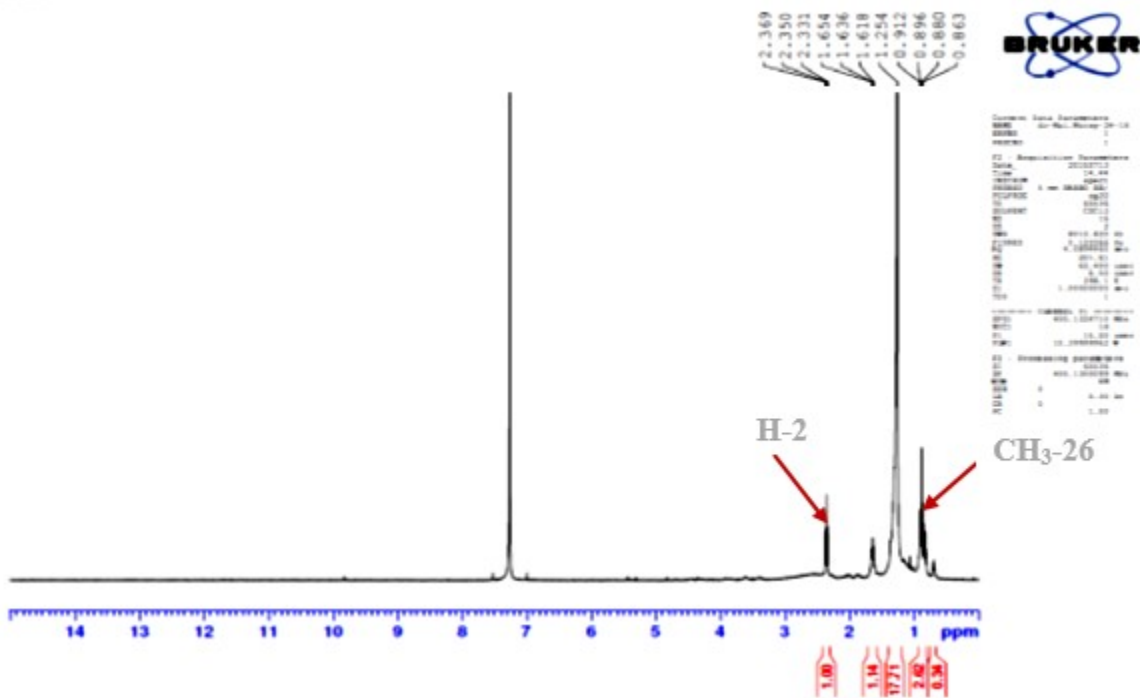


Figure (S10): ¹H-NMR spectrum of compound "4".

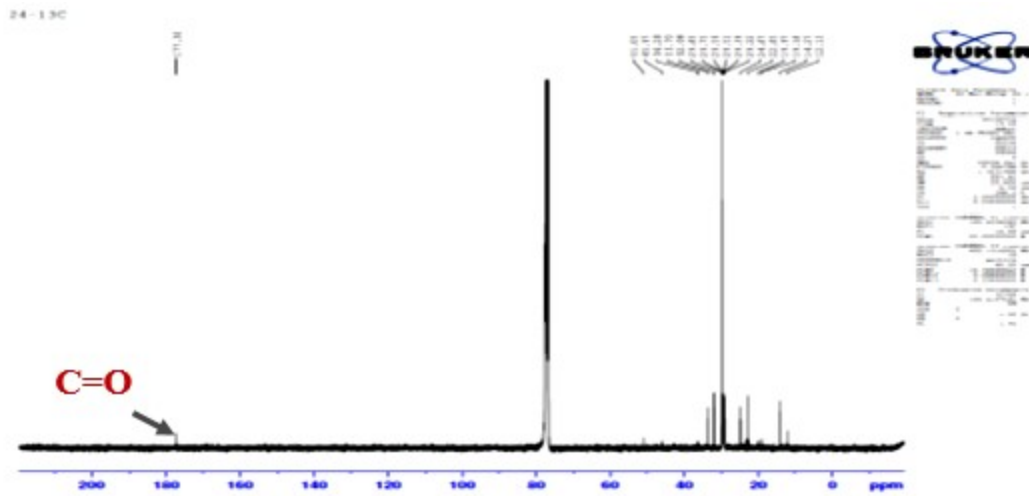


Figure (S11): ¹³C-NMR spectrum of compound “4”.

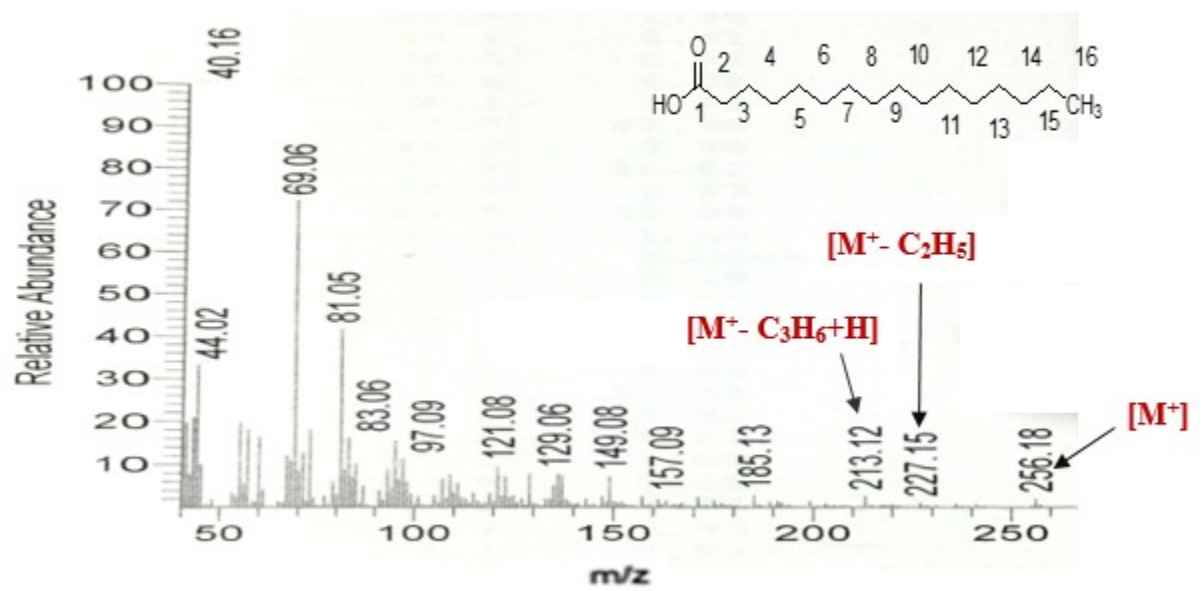


Figure (S12): EI-MS spectrum of compound “5”.

B23-1H

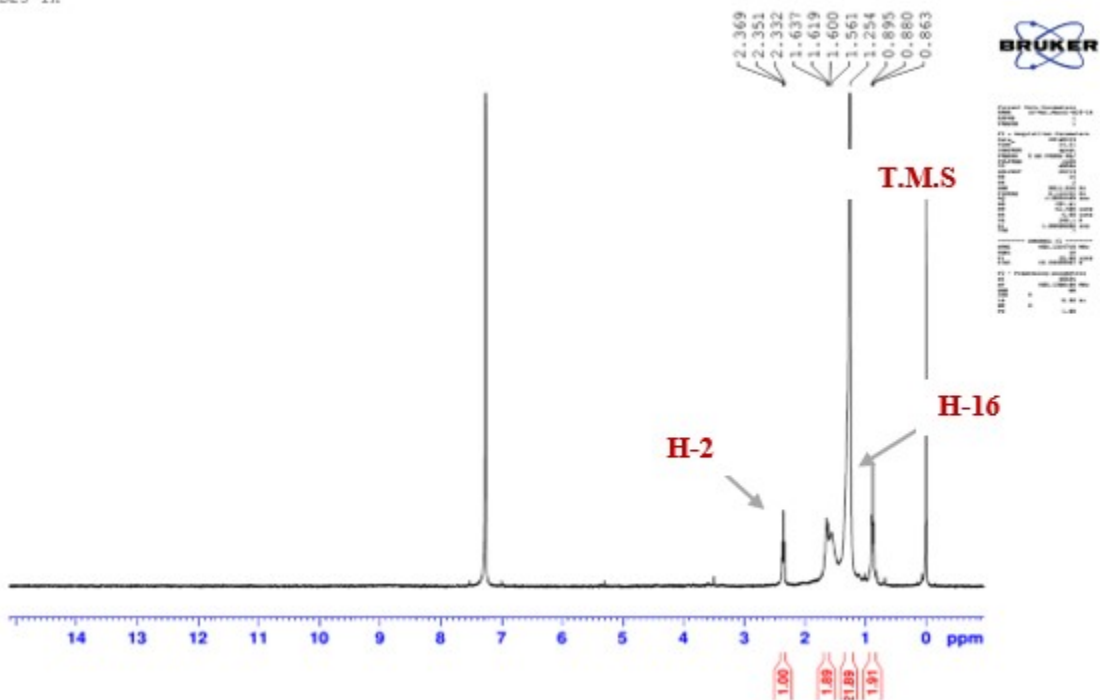


Figure (S13): ¹H-NMR spectrum of compound "5".

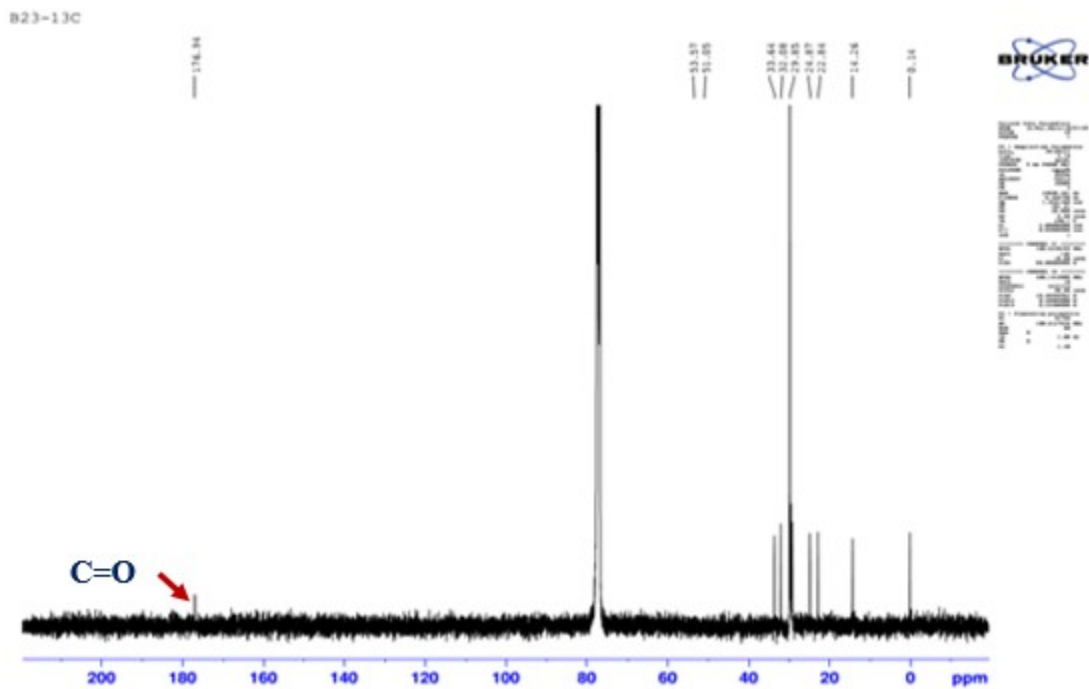


Figure (S14): ¹³C-NMR spectrum of compound "5".

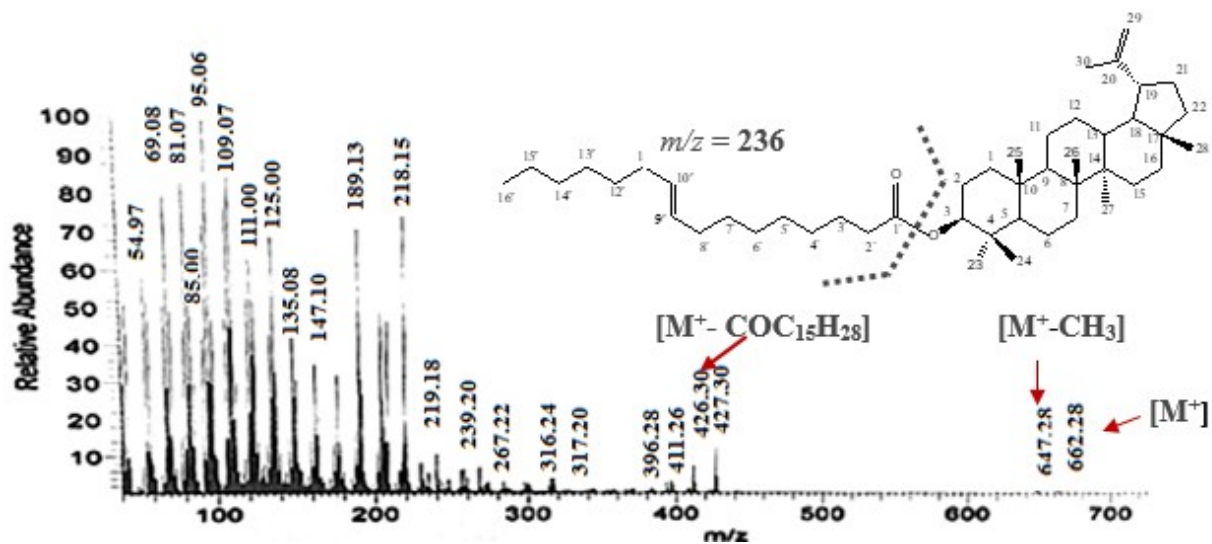


Figure (S15): EI-MS spectrum of compound "6".

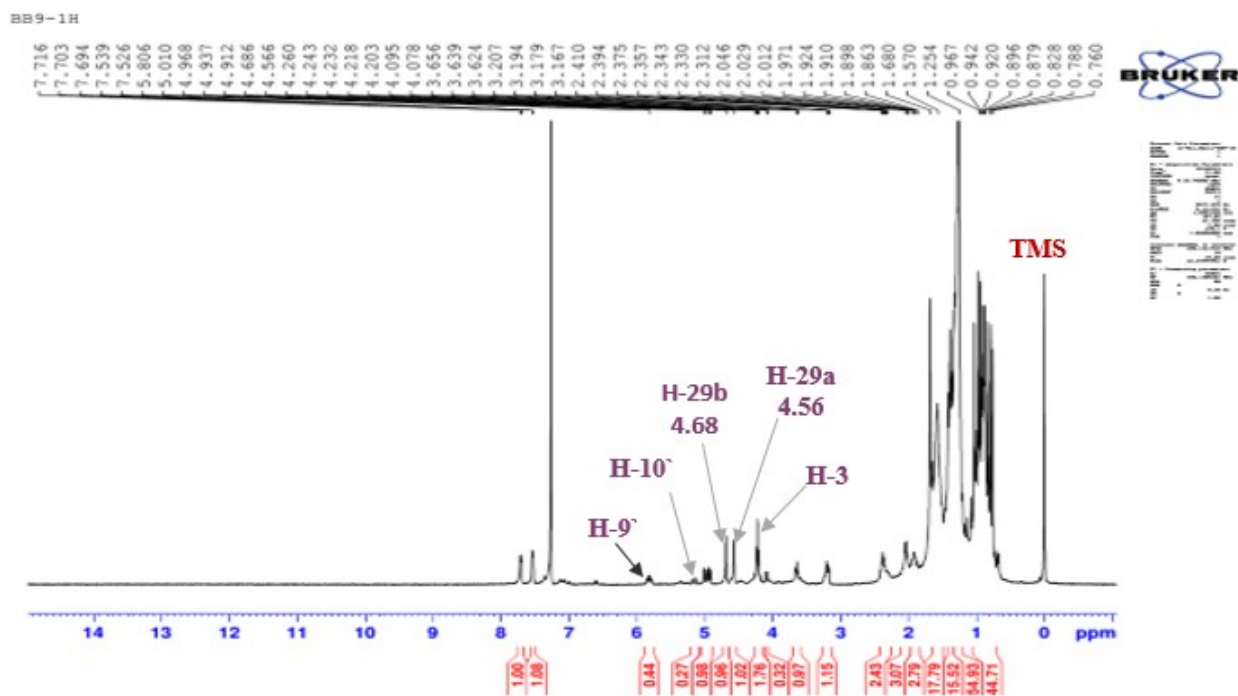


Figure (S16): 1H -NMR spectrum of compound "6".

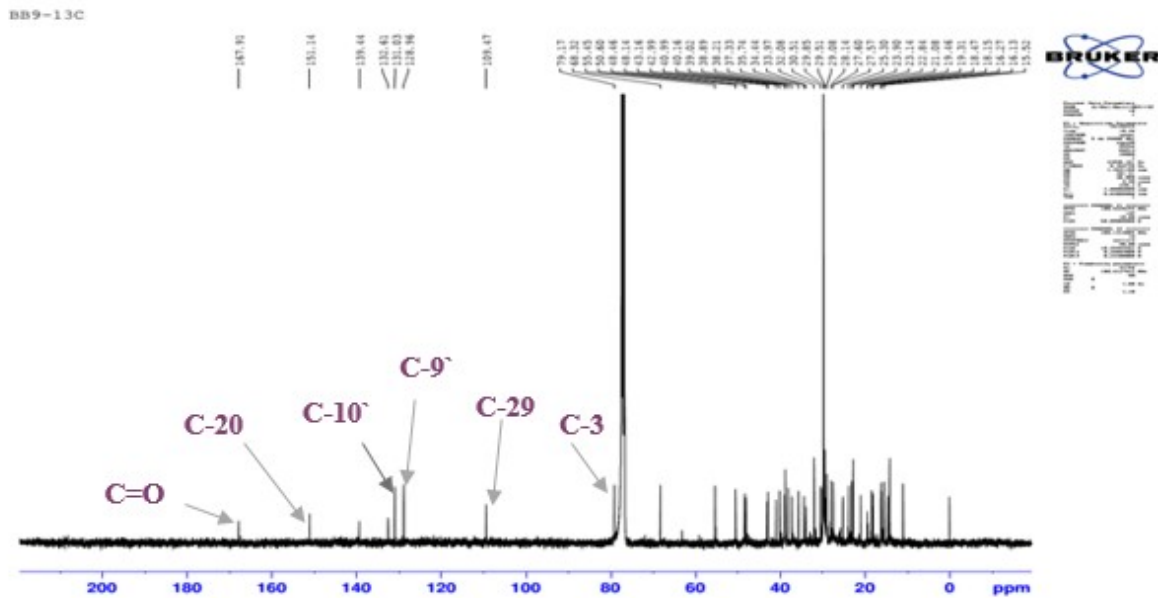


Figure (S17): ¹³C-NMR spectrum of compound "6".

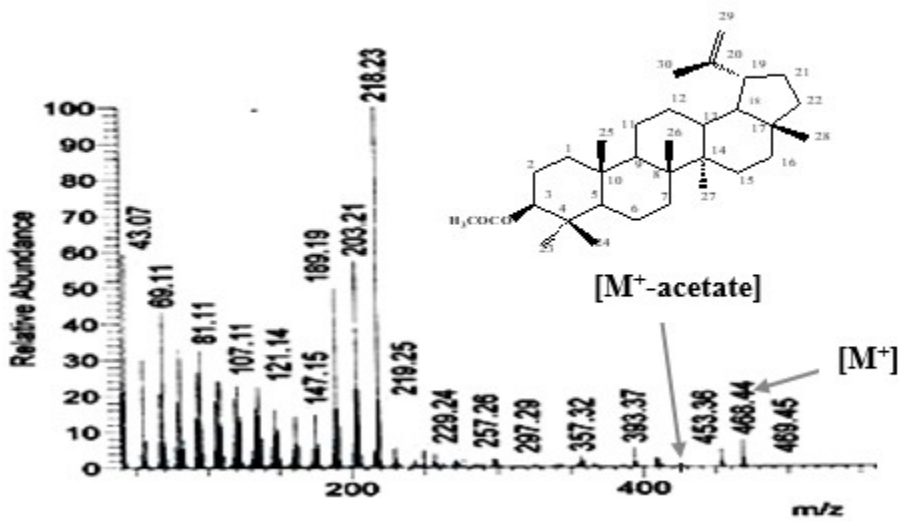


Figure (S18): EI-MS spectrum of compound "7".

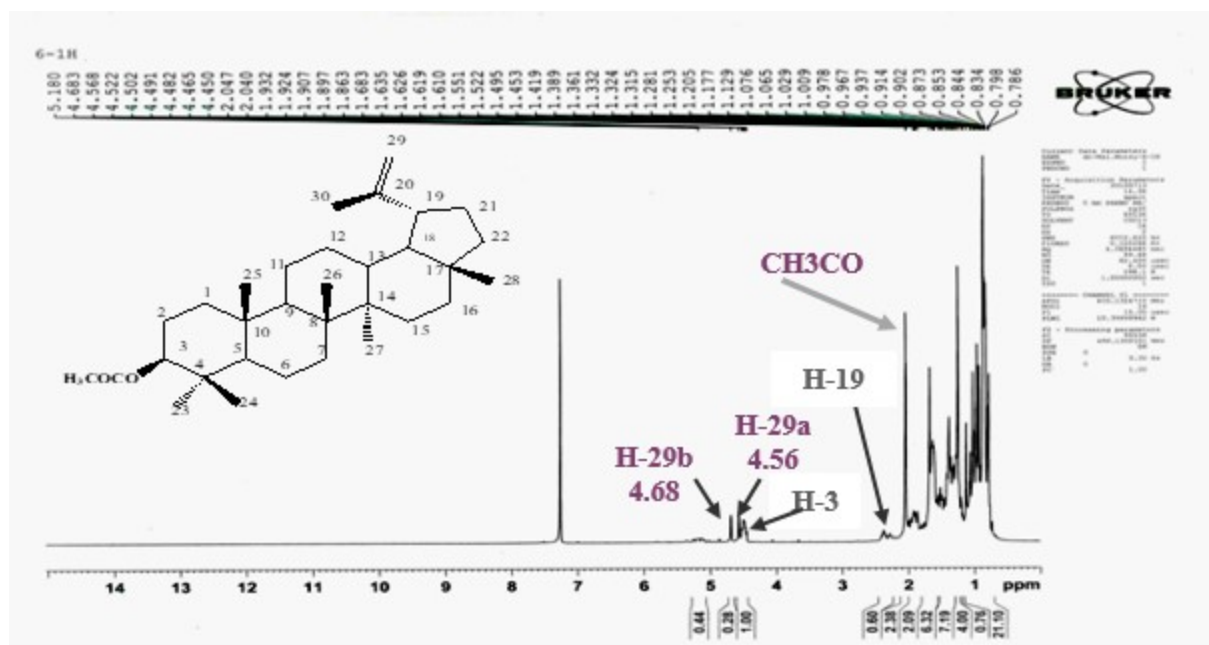


Figure (S19): $^1\text{H-NMR}$ spectrum of compound "7".

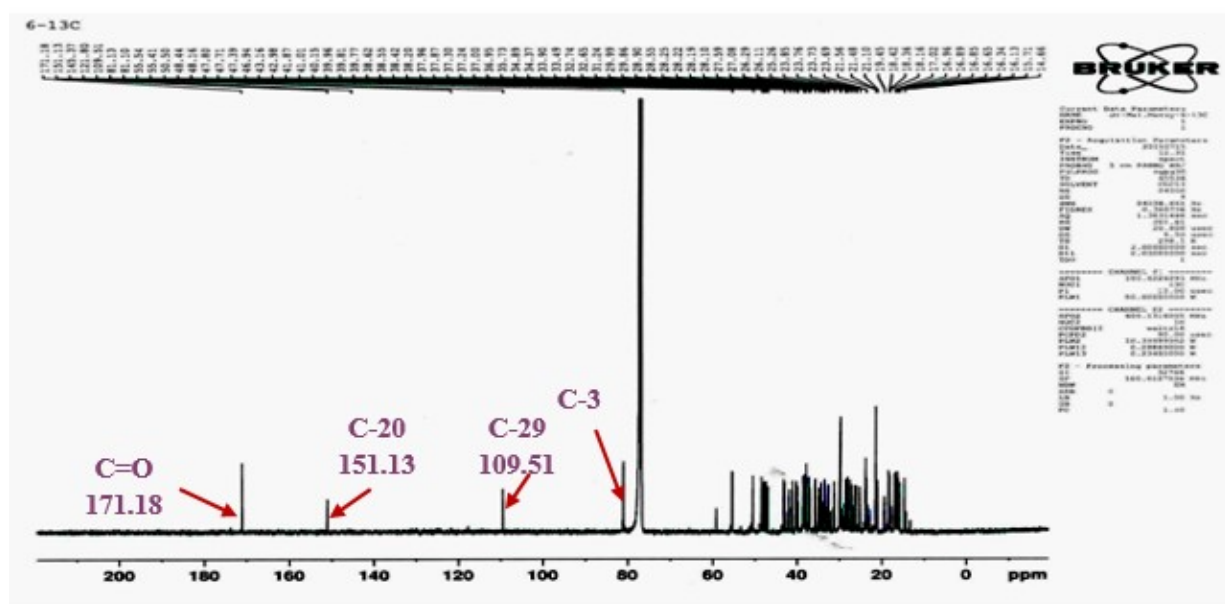


Figure (S20): $^{13}\text{C-NMR}$ spectrum of compound "7".

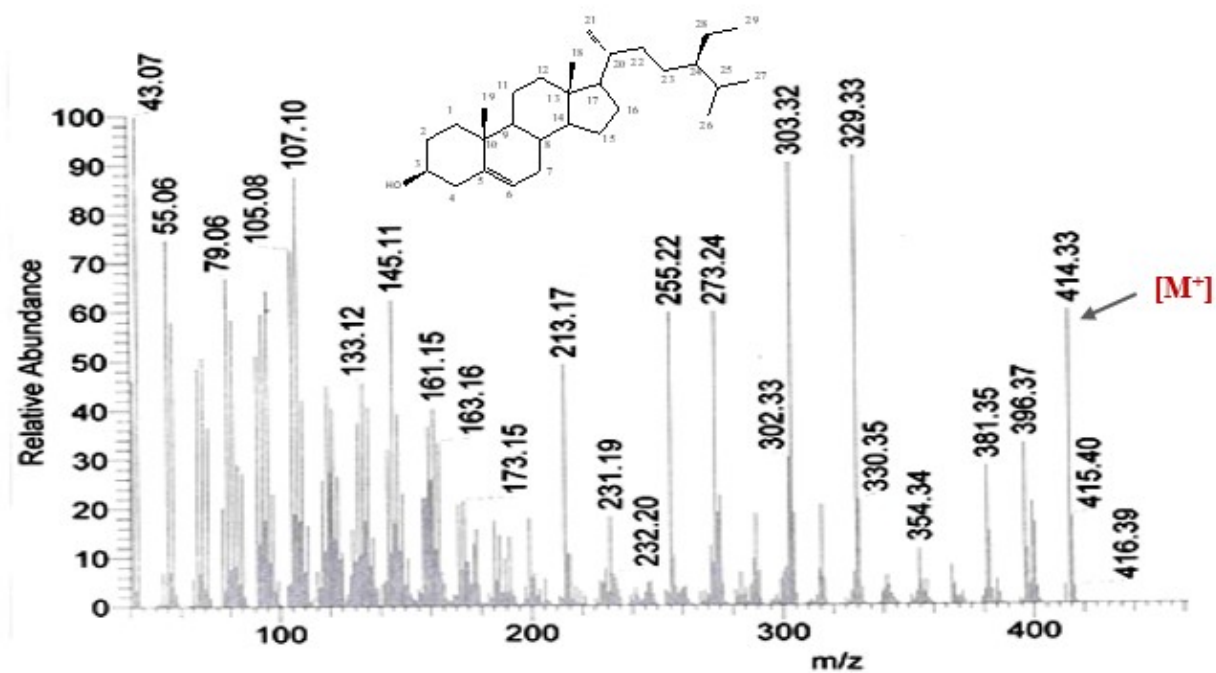


Figure (S21): EI-MS spectrum of compound "8".

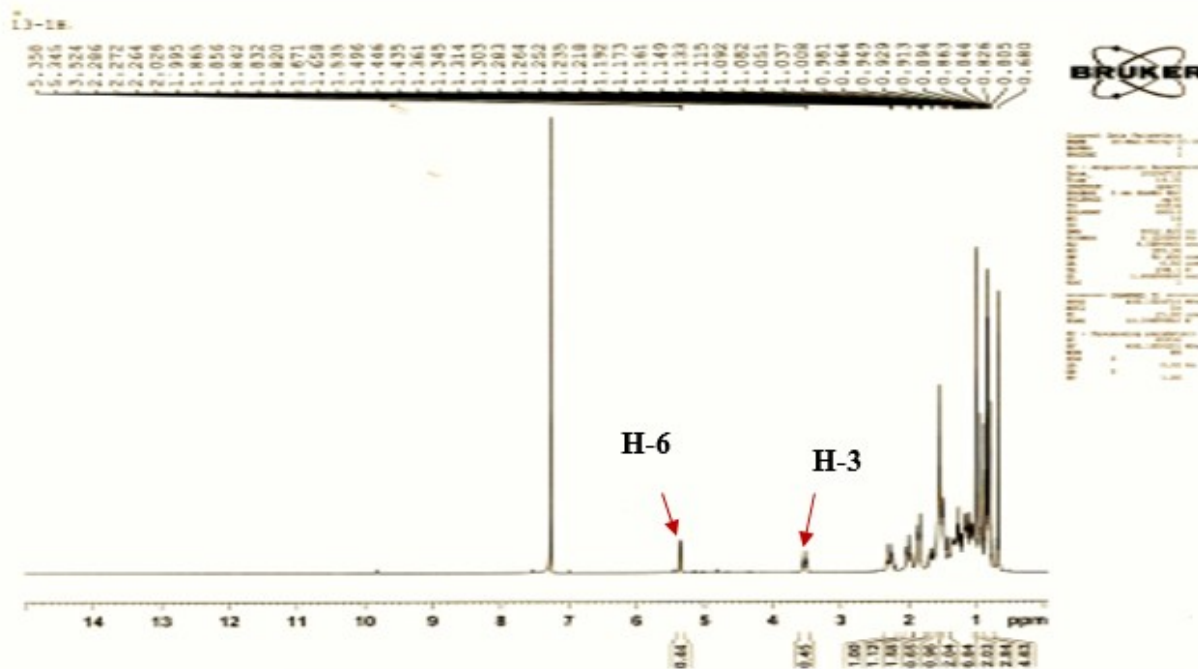


Figure (S22): ¹H-NMR spectrum of compound "8".

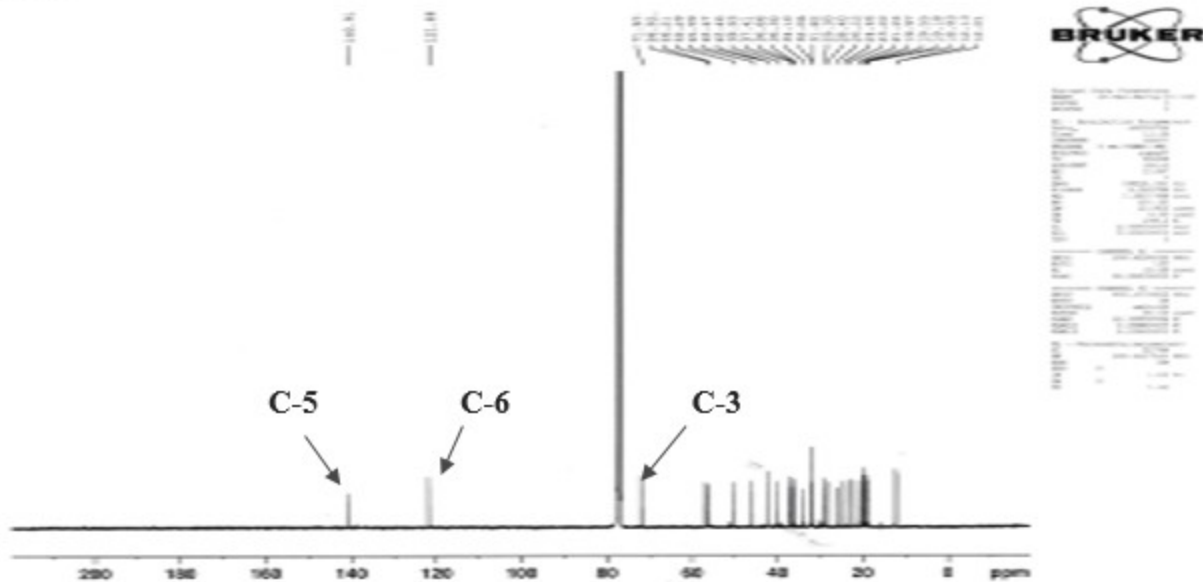


Figure (S23): ¹³C-NMR spectrum of compound "8".

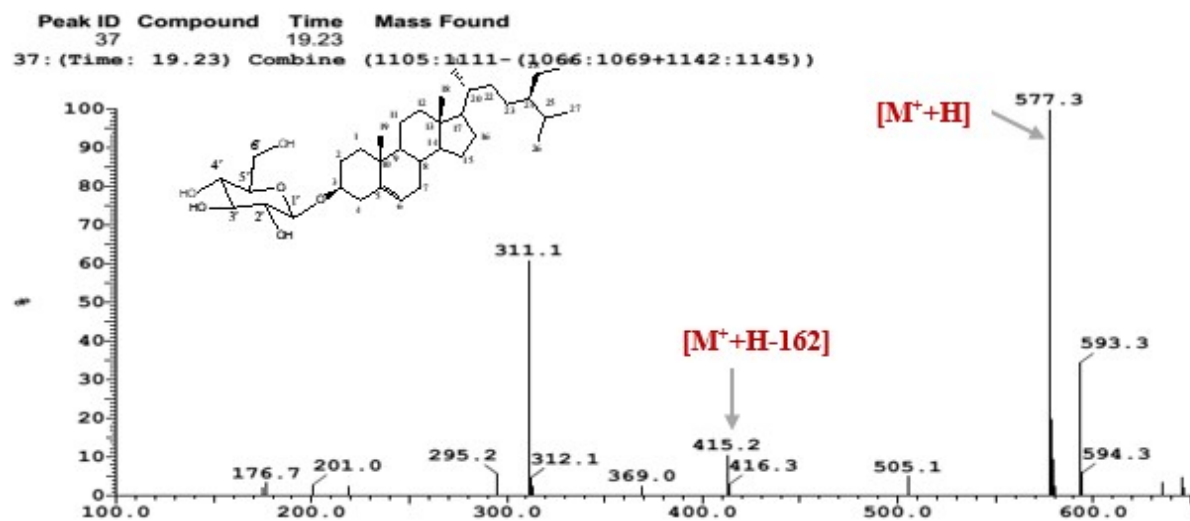


Figure (S24): ESI-MS spectrum (positive mode) of compound "9".

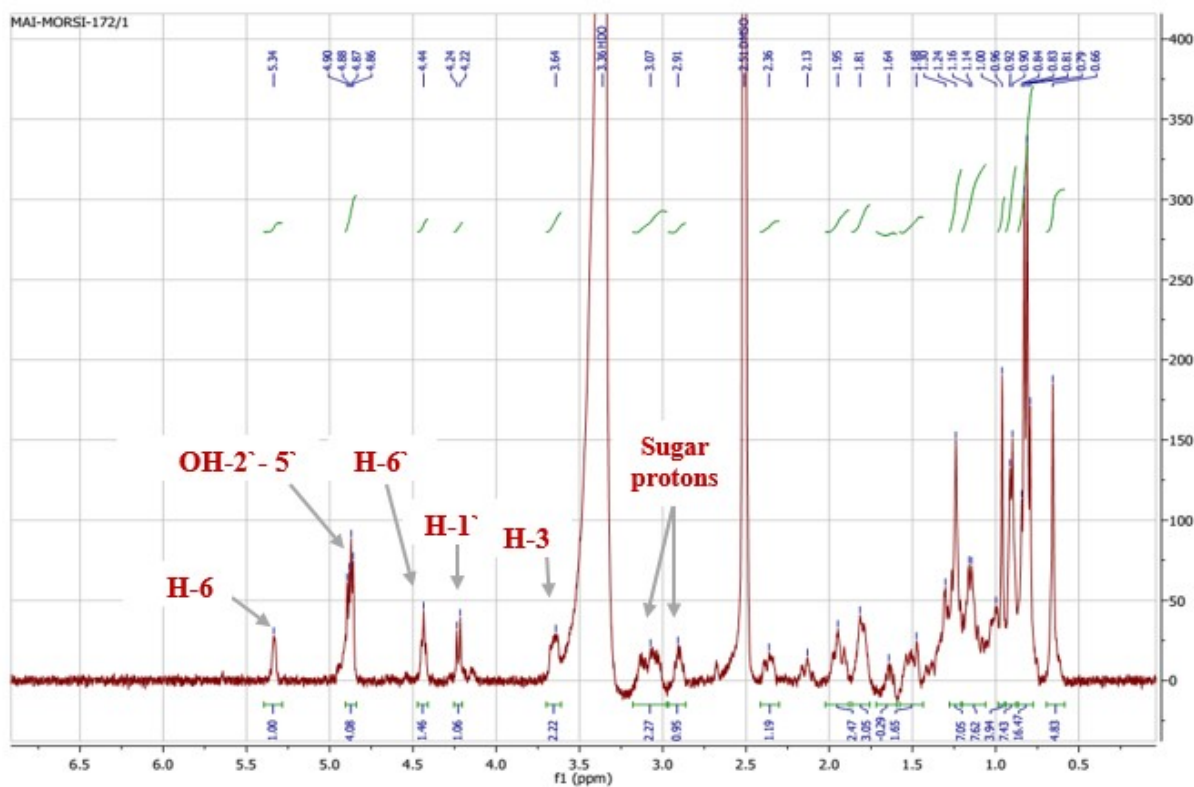
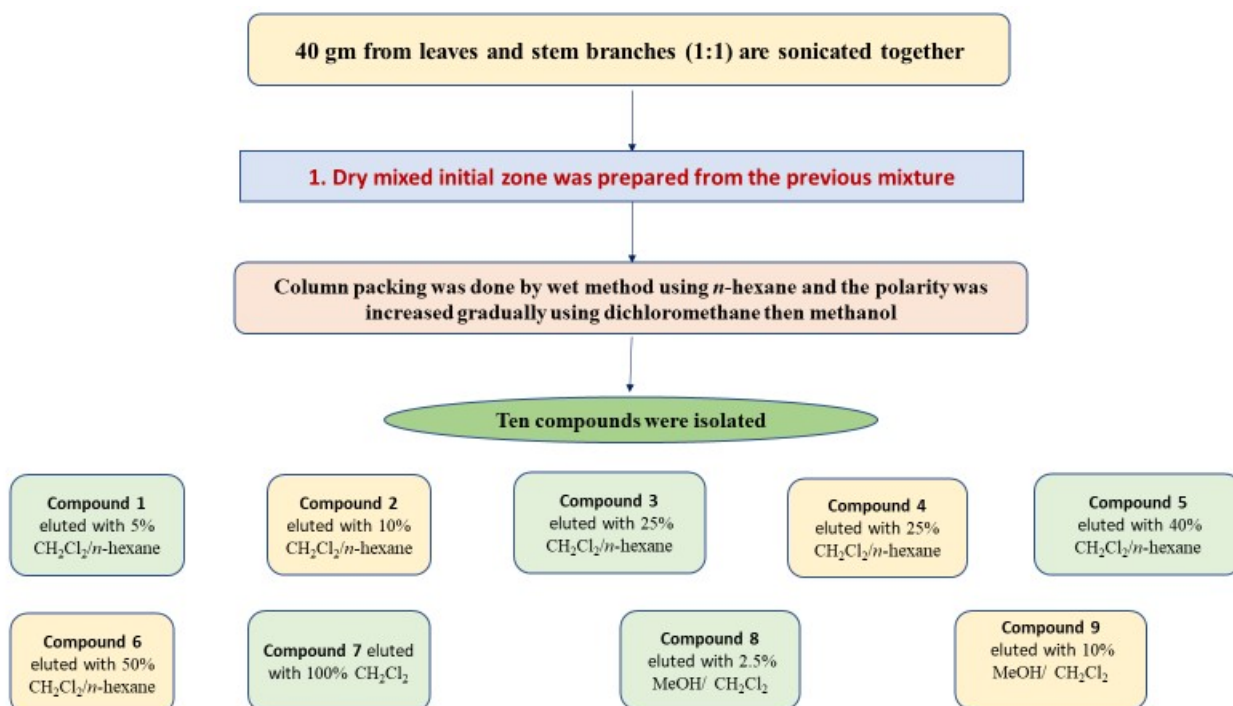


Figure (S25): ¹H-NMR spectrum of compound "9".



Scheme S1: Column chromatography for isolation of secondary metabolites from DCM fractions mixture of *M. macroura* leaves and stem branches.