

Supporting Information

A low cost Y-type zeolite/carbon porous composite from coal gasification fine slag and its application in phenol remove in wastewater: Fabrication, characterization, equilibrium, and kinetic studies

Zhen Chai^a, Peng Lv^{a, *}, Yonghui Bai^{a, *}, Jiaofei Wang^a, Xudong Song^a,

Weiguang Su^a, Guangsuo Yu^{a, b}

^a *State Key Laboratory of High-Efficiency Utilization of Coal and Green Chemical Engineering,*

Ningxia University, Yinchuan 750021, China

^b *Institute of Clean Coal Technology, East China University of Science and Technology, Shanghai*

200237, China

**Corresponding author. Tel: 0951-2062008,*

E-mail: yhbai@nxu.edu.cn, lypeng@nxu.edu.cn

Table. S1 Proximate and ultimate analyses of CGFS

Sample	Proximate analysis (wt./%)			Ultimate analysis, d (wt./%)				
	V _d	A _d	Fc _d	C	H	O*	N	S
CGFS	6.96	80.61	12.43	16.33	0.71	0.76	0.08	1.51

Note: V_d: Volatile; A: Ash; Fc_d: Fixed carbon; d: dry basis; *by difference

Table. S2 Pore structure parameters of samples

Sample	S _{BET} (m ² /g)	S _{t-plot} (m ² /g)	V _{total} (cm ³ /g)	V _{micro} (cm ³ /g)	V _{meso} (cm ³ /g)
CGFS	111.09	18.89	0.14	0.01	0.13
Y-type zeolite/carbon composite	439.68	234.45	0.35	0.12	0.23

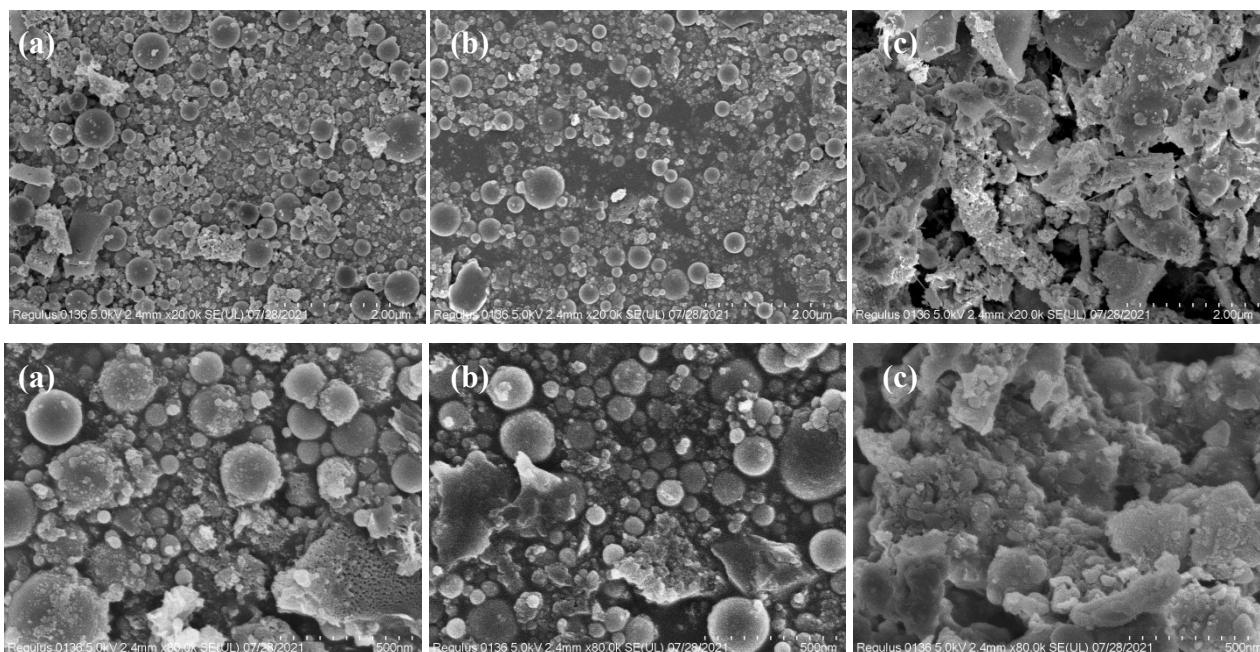


Fig. S1 SEM images of CGFS (a), HCGFS (b), HCGFS-HH (c)

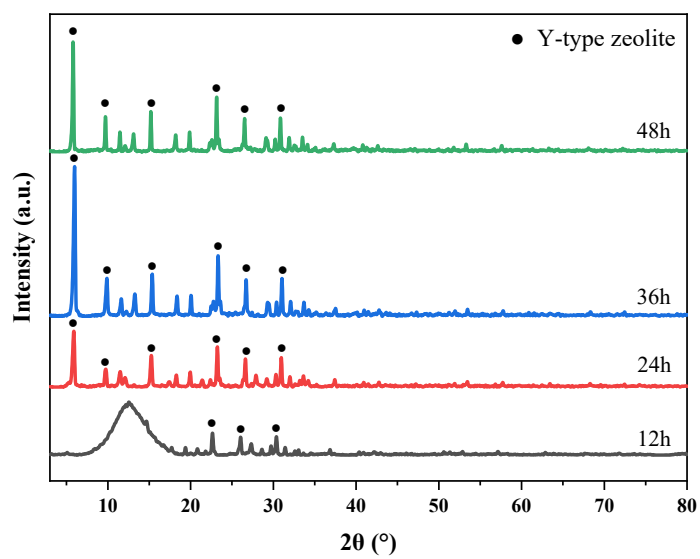


Fig. S2 XRD patterns of materials with different crystallization times

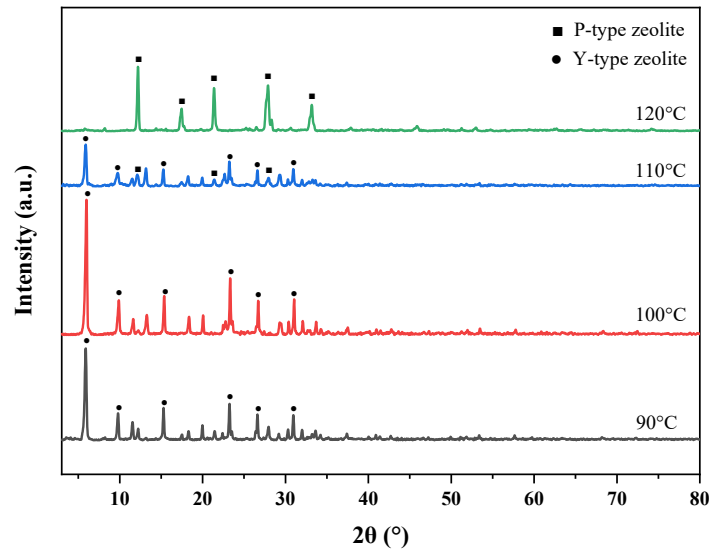


Fig. S3 XRD patterns of materials prepared at different crystallization temperatures

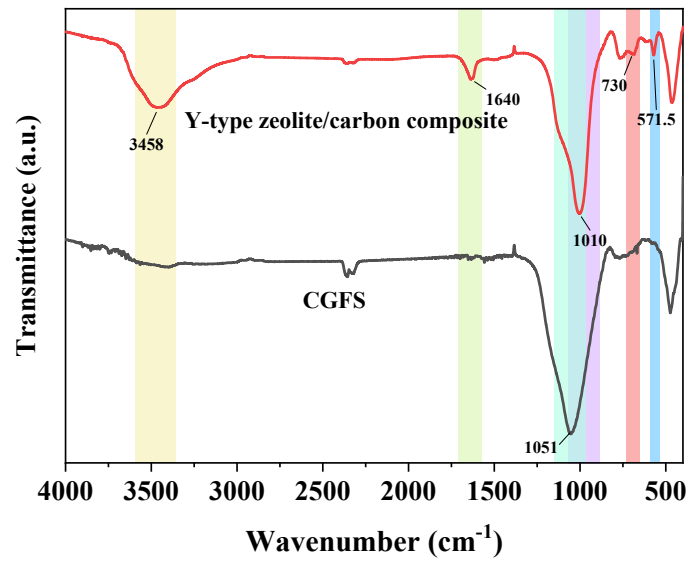


Fig. S4 FT-IR spectra of samples

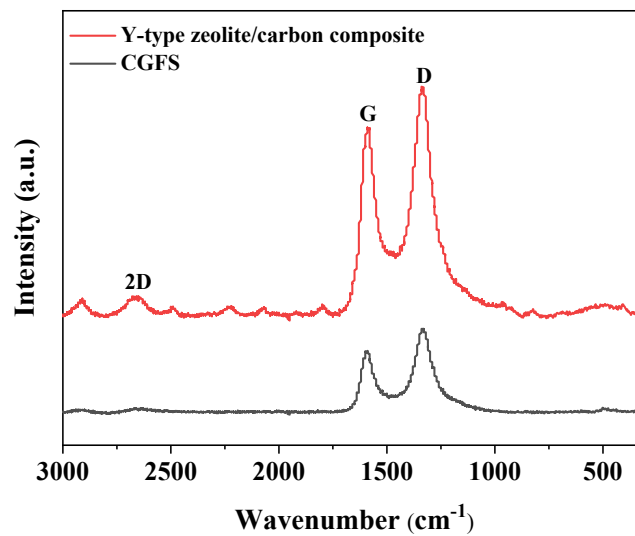


Fig. S5 Raman spectrum of the samples

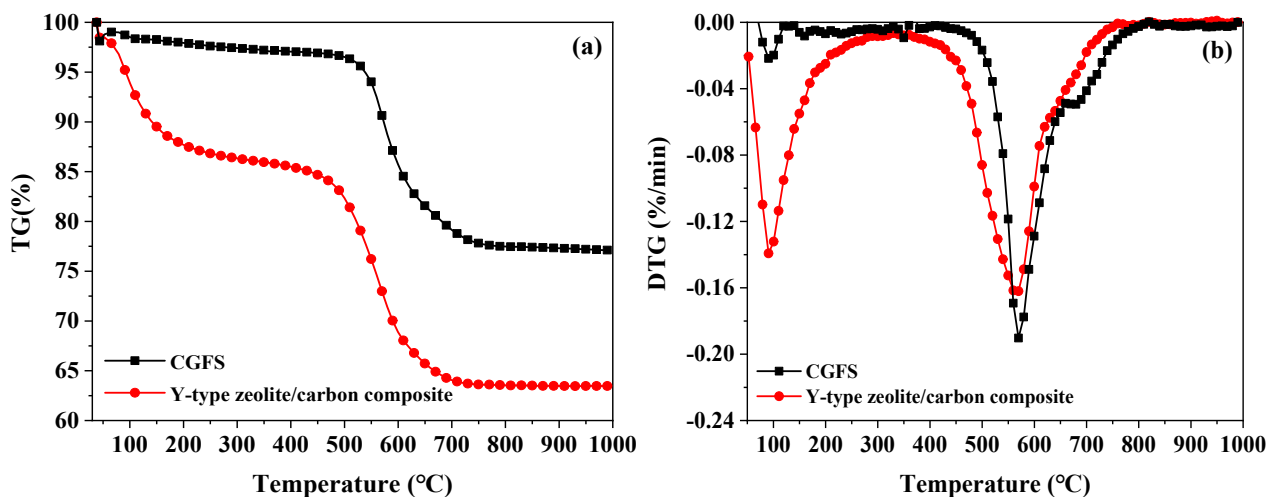


Fig. S6 TG (a) and DTG (b) curves of samples

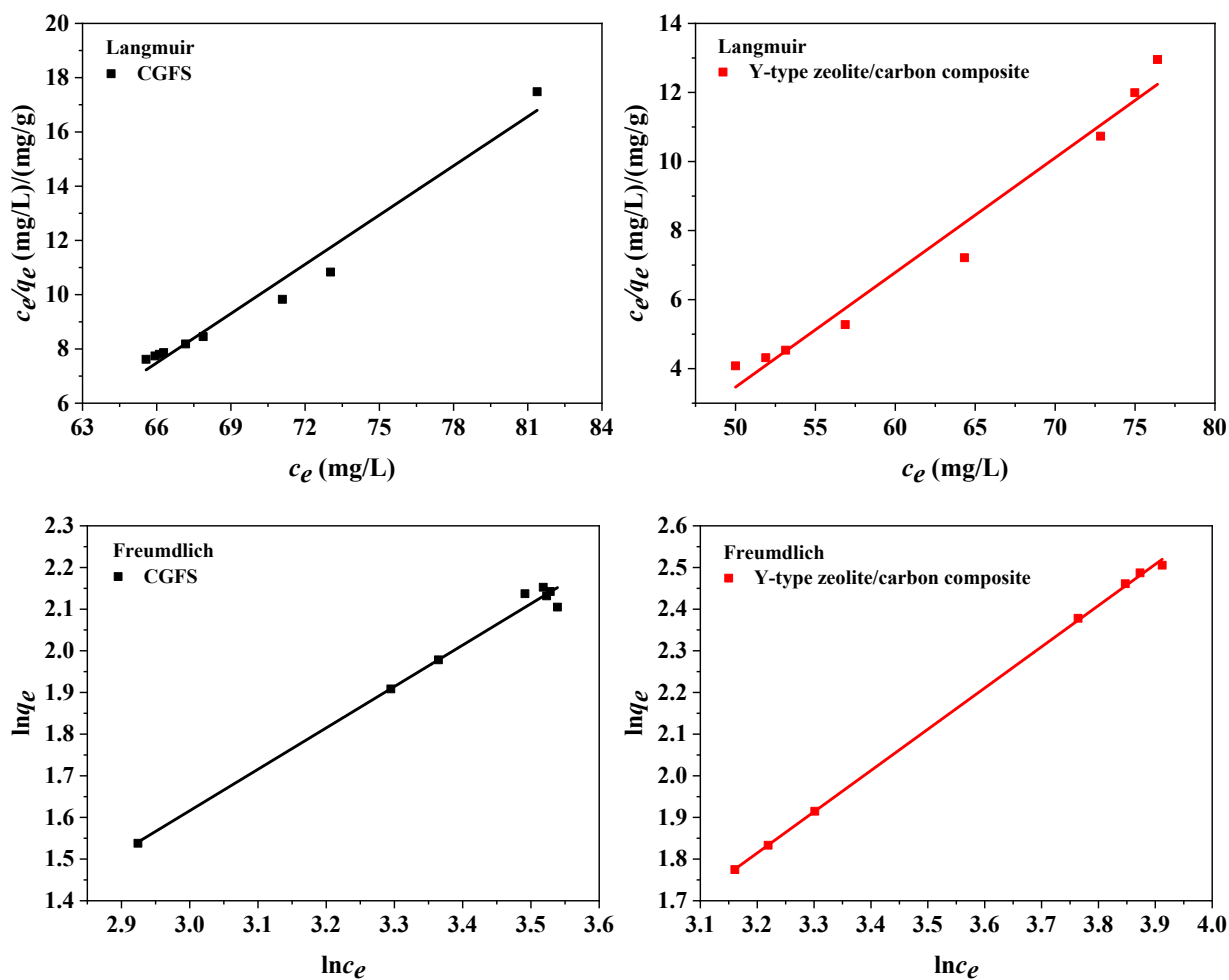


Fig. S7 Adsorption isotherms of phenol onto samples